

Supporting Information:

Coarse-grained molecular dynamics simulation of cation distribution profiles on negatively charged lipid membranes during phase separation

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Hydration state of sodium ion

The all-atomic molecular dynamics (MD) simulation of Na^+ and Cl^- ions solution consisting of 602 atoms in total is performed for 120 ps under three-dimensional periodic boundary conditions. An ion concentration is set at 217 mM. The third-order density-functional tight-binding (DFTB3) method is used with DFTB+^{S1} (version 20.1) open-source software. Parameter sets of 3ob^{S2,S3} and 3obw^{S4} are used with H5 correction.^{S5} The temperature is kept constant at 300 K by a Nose-Hoover thermostat^{S6} with a coupling strength of 3200 cm^{-1} . A pressure of 1 atm was controlled isotropically by a Berendsen barostat^{S7} with a damping time of 0.1 ps. The velocity-Verlet algorithm with a time step of 1 fs is used for the integration of the equations of motion.

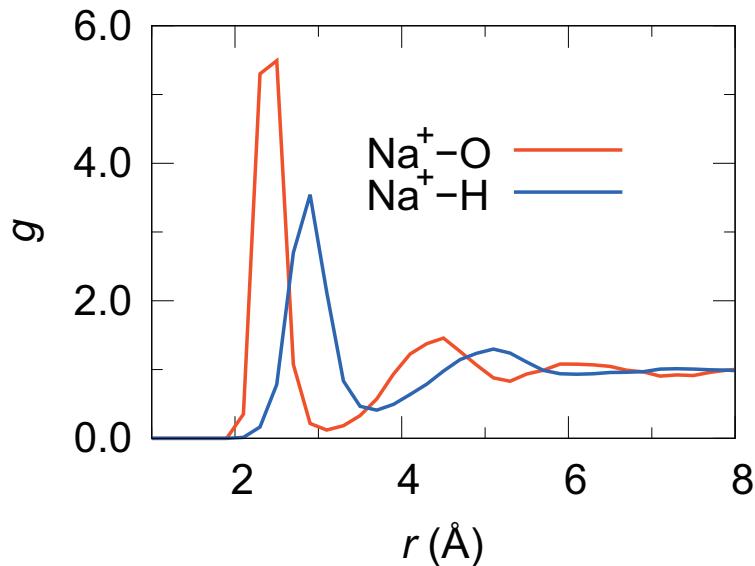


Figure S1: Radial distribution functions (RDFs), $g(r)$, of water molecules around a sodium ion by the all-atomic molecular dynamics (MD) simulation based on a third-order density functional tight-binding (DFTB3) method.

Surface roughness

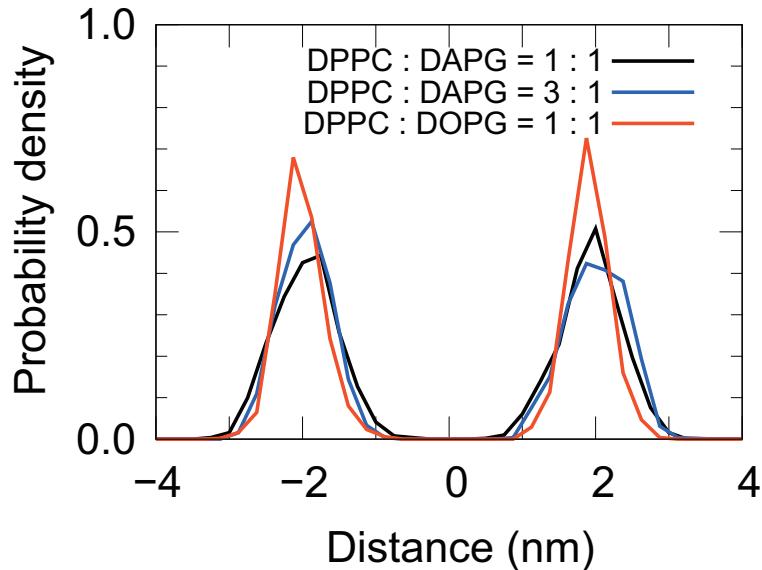


Figure S2: The probability density of the negatively charged PO_4^- group along z -direction (perpendicular to the membrane surface) at 5 μs for DPPC:DAPG = 1 : 1 mixture, DPPC:DAPG = 3 : 1, and DPPC:DAPG = 1 : 1 with a NaCl concentration of 100 mM.

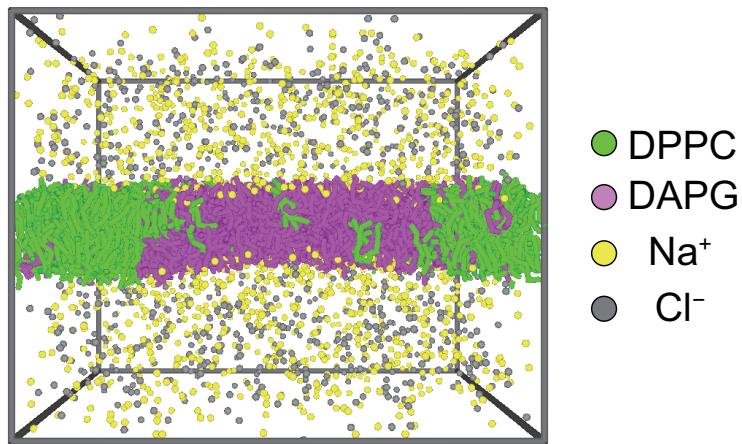


Figure S3: The snapshot of the side view of the membrane at 5 μs for DPPC:DAPG = 1 : 1 mixture with a NaCl concentration of 100 mM.

References

- (S1) Hourahine, B.; Aradi, B.; Blum, V.; Bonafé, F.; Buccheri, A.; Camacho, C.; Cevallos, C.; Deshaye, M. Y.; Dumitrică, T.; Dominguez, A.; Ehlert, S.; Elstner, M.; van der Heide, T.; Hermann, J.; Irle, S.; Kranz, J. J.; Köhler, C.; Kowalczyk, T.; Kubář, T.; Lee, I. S.; Lutsker, V.; Maurer, R. J.; Min, S. K.; Mitchell, I.; Negre, C.; Niehaus, T. A.; Niklasson, A. M. N.; Page, A. J.; Pecchia, A.; Penazzi, G.; Persson, M. P.; Řezáč, J.; Sánchez, C. G.; Sternberg, M.; Stöhr, M.; Stuckenber, F.; Tkatchenko, A.; Yu, V. W.-z.; Frauenheim, T. DFTB+, a software package for efficient approximate density functional theory based atomistic simulations. *J. Chem. Phys.* **2020**, *152*, 124101.
- (S2) Gaus, M.; Goez, A.; Elstner, M. Parametrization and Benchmark of DFTB3 for Organic Molecules. *J. Chem. Theory Comput.* **2013**, *9*, 338–354.
- (S3) Kubillus, M.; Kubář, T.; Gaus, M.; Řezáč, J.; Elstner, M. Parameterization of the DFTB3 Method for Br, Ca, Cl, F, I, K, and Na in Organic and Biological Systems. *J. Chem. Theory Comput.* **2015**, *11*, 332–342.
- (S4) Goyal, P.; Qian, H.-J.; Irle, S.; Lu, X.; Roston, D.; Mori, T.; Elstner, M.; Cui, Q. Molecular Simulation of Water and Hydration Effects in Different Environments: Challenges and Developments for DFTB Based Models. *J. Phys. Chem. B* **2014**, *118*, 11007–11027.
- (S5) Řezáč, J. Empirical Self-Consistent Correction for the Description of Hydrogen Bonds in DFTB3. *J. Chem. Theory Comput.* **2017**, *13*, 4804–4817.
- (S6) Martyna, G. J.; Tuckerman, M. E.; Tobias, D. J.; Klein, M. L. Explicit reversible integrators for extended systems dynamics. *Mol. Phys.* **1996**, *87*, 1117–1157.
- (S7) Berendsen, H. J. C.; Postma, J. P. M.; van Gunsteren, W. F.; DiNola, A.; Haak, J. R. Molecular dynamics with coupling to an external bath. *J. Chem. Phys.* **1984**, *81*, 3684–3690.