Electronic Supplementary Information for

Non-isothermal cold crystallization of liquid crystalline porphyrins

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Fig. S1 ¹H NMR of cyclohexyl 6-bromohexanoate.



Fig. S2 ¹³C NMR of cyclohexyl 6-bromohexanoate.







Fig. S4 ¹³C NMR of cyclohexyl 8-bromooctanoate.







Fig. S6 ¹³C NMR of cyclohexyl 10-bromodecanoate.











Fig. S9 ¹H NMR of 1.



Fig. S10 ¹³C NMR of 1.



Fig. S11 ¹H NMR of 2.



Fig. S12 ¹³C NMR of 2.



Fig. S13 ¹H NMR of 3.



Fig. S14 ¹³C NMR of 3.



Fig. S15 ¹H NMR of 4.



Fig. S16 ¹³C NMR of 4.



Fig. S17 MALDI-TOF MS of 1 – 4, panels a – d, respectively.



Fig. S18 TGA of all four molecules showing loss of cyclohexanol at 338 °C and alkanoic acid at 410 °C.



Fig. S19 MDSC of 1 showing modulated temperature (magenta), total heat flow (maroon), reversing heat flow for thermal events (green), and non-reversing heat flow for kinetic events (blue).



Fig. S20 MDSC of 2 showing modulated temperature (magenta), total heat flow (maroon), reversing heat flow for thermal events (green), and non-reversing heat flow for kinetic events (blue).



Fig. S21 1 – 4 cooling from isotropic transition: (a) 1 cooled for 30 min, (b) 2 cooled for 90 min, (c) 3 cooled for 1 min, 4 cooled for 3 min.



Fig. S22 (Left) HS-POM of isothermal cold crystallization of **1** at 90 °C. (Right) Disc radius vs. time for one disc (circled in 10 min micrograph); y = 1.1583 - 0.6329, $R^2 = 0.9994$.



Fig. S23. Kissinger plots: $E_a \mathbf{1} = 81 \pm 10 \text{ kJ/mol} (R^2 = 0.92), E_a \mathbf{2} = 108 \pm 7 \text{ kJ/mol} (R^2 = 0.97).$