

Electronic supplementary information (ESI)

Synthesis of Gallic Acid Grafted Epoxidized Natural Rubber and its Role in Self-Healable Flexible Temperature Sensor

Aparna Guchait, Simran Sharma, Santanu Chattopadhyay*, Titash Mondal*

Rubber Technology Centre, Indian Institute of Technology Kharagpur, Kharagpur, India-721302

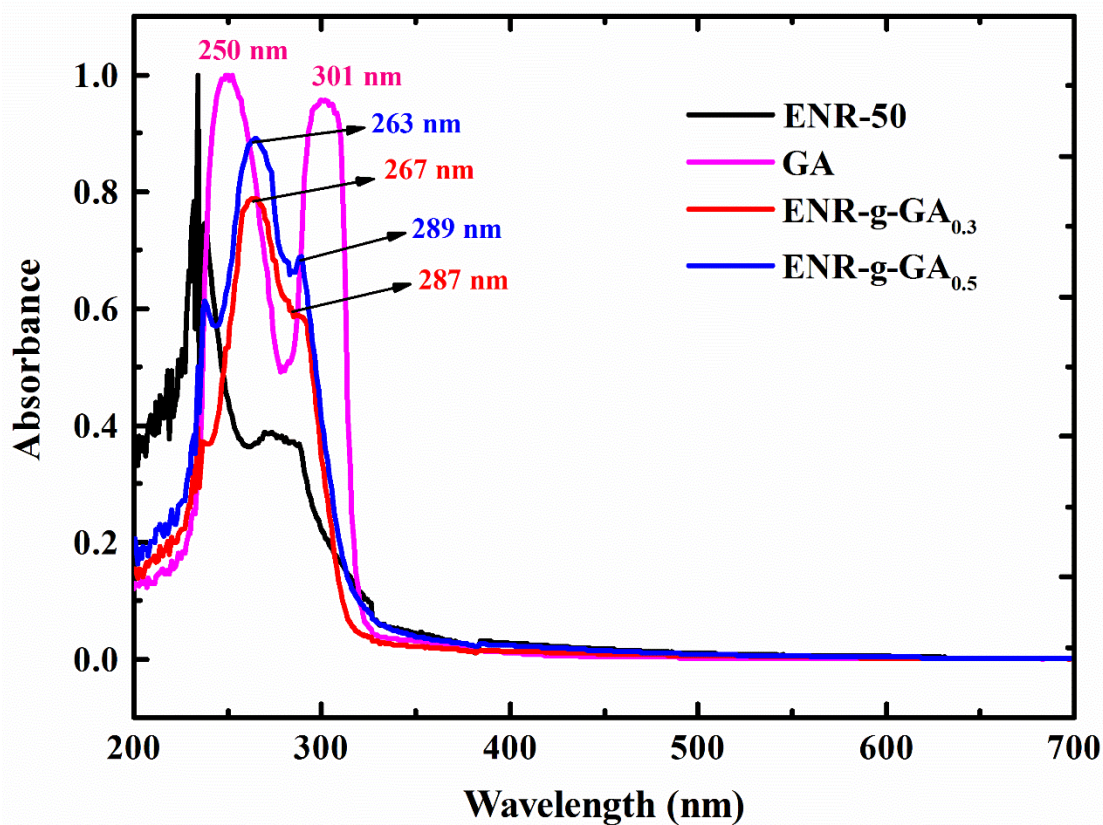


Fig. S1: UV-vis spectra of unmodified ENR-50, GA, and ENR-g-GA.

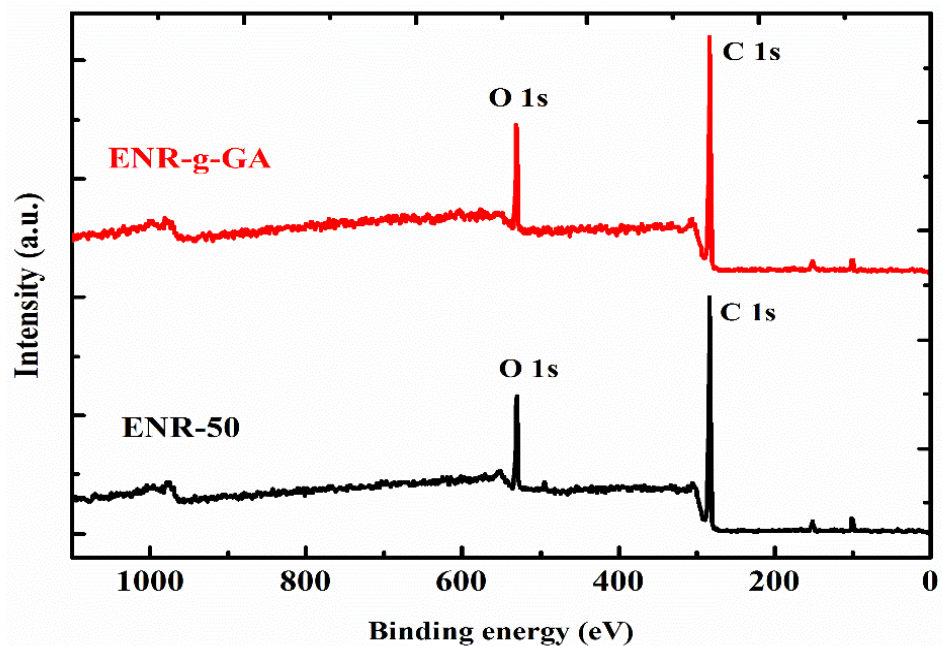


Fig. S2: XPS survey scan of pristine ENR-50 and ENR-g-GA

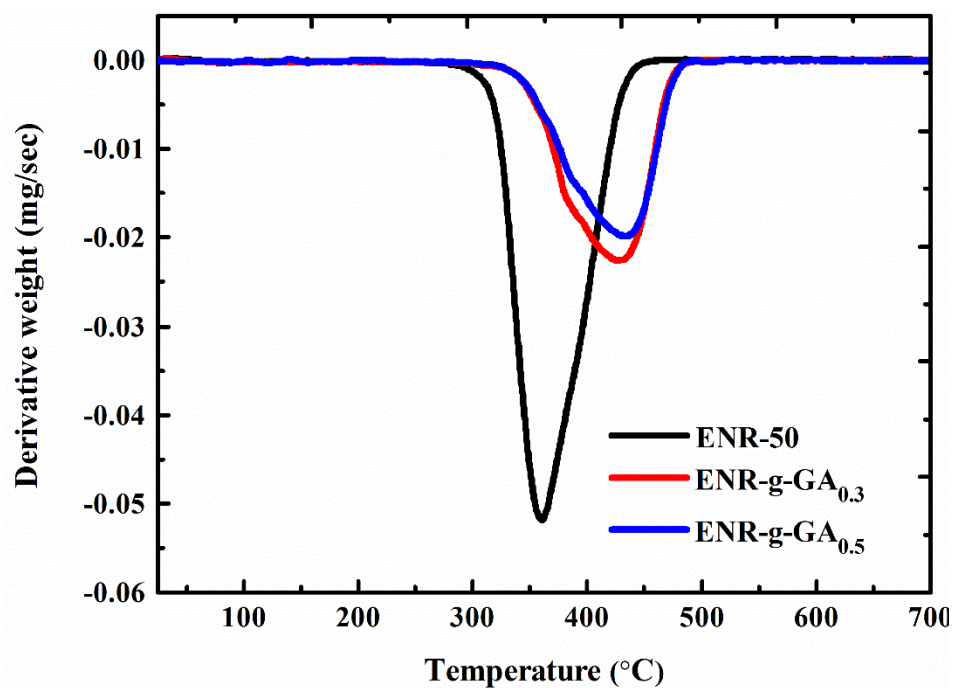


Fig. S3: DTG thermogram of ENR-50 and ENR-g-GA.

Table S1: Thermal degradation of ENR, ENR-g-GA_{0.3}, ENR-g-GA_{0.5}

Sample Name	T ₁₀ (°C)	T ₅₀ (°C)	T ₉₀ (°C)	Residue (%)
ENR-50	336	367	407	1.9
ENR-g-GA _{0.3}	365	414	451	0.3
ENR-g-GA _{0.5}	365	417	454	0.2

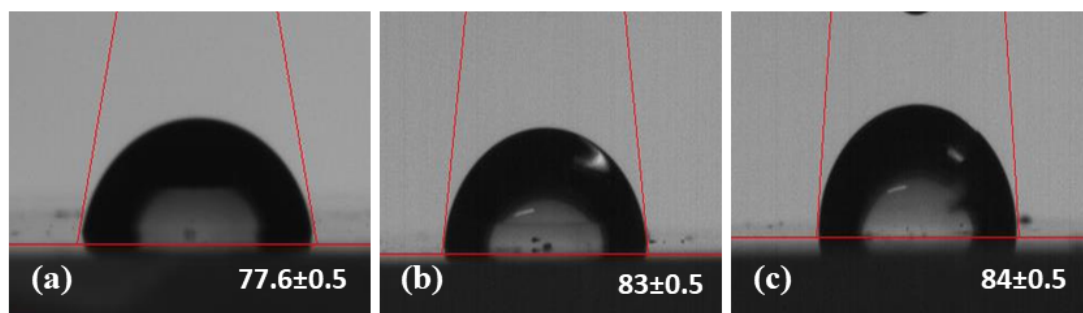


Fig. S4: Contact angle of pristine ENR-50 (a), ENR-g-GA_{0.3} (b) and ENR-g-GA_{0.5} (c)

Table S2: Contact angle and work of adhesion value for pristine ENR-50, ENR-g-GA_{0.3} and ENR-g-GA_{0.5}

Parameter	Purified ENR-50	ENR-g-GA _{0.3}	ENR-g-GA _{0.5}
Contact angle (°)	77.6 ± 0.5	83 ± 0.5	84 ± 0.5
W _{SL} (J/m ²)	88.36	80.7	79.5

Table S3: Tensile strength and elongation value of the sample.

Sample name	Tensile strength (MPa)	Elongation at Break (%)
$\text{ENR}/\text{Fe}_{0.02}^{3+}$	2.73 ± 0.25	748 ± 37
$\text{ENR-g-GA}/\text{Fe}_{0.02}^{3+}$	4.42 ± 0.13	667 ± 25
$\text{ENR}/\text{Fe}_{0.05}^{3+}$	3.71 ± 0.05	561 ± 97
$\text{ENR-g-GA}/\text{Fe}_{0.05}^{3+}$	6.5 ± 0.34	451 ± 54
$\text{ENR}/\text{Fe}_{0.10}^{3+}$	5.1 ± 0.23	470 ± 59
$\text{ENR-g-GA}/\text{Fe}_{0.10}^{3+}$	19.6 ± 0.33	248 ± 13
$\text{ENR}/\text{Fe}_{0.15}^{3+}$	8.5 ± 0.85	429 ± 13
$\text{ENR-g-GA}/\text{Fe}_{0.15}^{3+}$	14.8 ± 0.58	12 ± 2.2

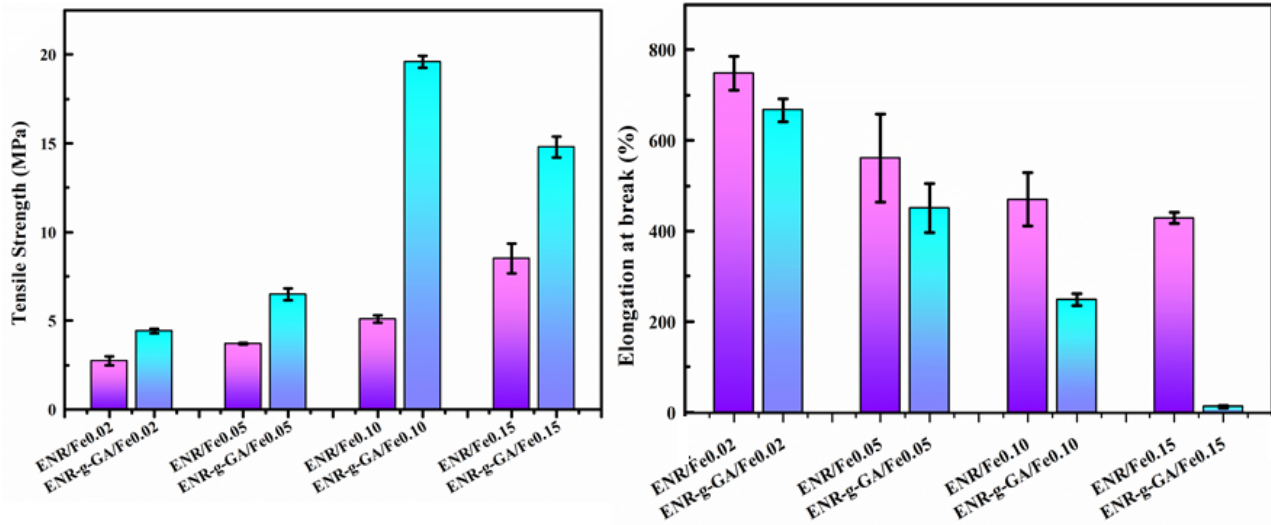


Fig. S5: Tensile strength and elongation at break.

Table S4: Tensile strength of ENR-g-GA/Fe with different Fe³⁺ content and CNT filled ENR-g-GA/Fe with different CNT loading.

Sample	Tensile strength (MPa)	EAB	Efficiency of tensile strength	Efficiency of EAB
ENR-g-GA/Fe _{0.02} ³⁺	4.42	667		
ENR-g-GA/Fe _{0.02} ³⁺ H	1.90	323	42.9%	48.4%
ENR-g-GA/Fe _{0.05} ³⁺	6.5	451		
ENR-g-GA/Fe _{0.05} ³⁺ H	1.57	118	24.1%	26.1%
ENR-g-GA/Fe _{0.10} ³⁺	19.59	248		
ENR-g-GA/Fe _{0.10} ³⁺ H	1.33	87	6.7%	35%
ENR-g-GAC2/Fe _{0.02} ³⁺	3.2	617		
ENR-g-GAC2/Fe _{0.02} ³⁺ H	1.0	146	31.2%	23.6%

Table S5: Temperature sweep

	Temperature (° C)	Tan delta
ENR/Fe _{0.10} ³⁺	-4.5	0.508
ENR-g-GA/Fe _{0.02} ³⁺	11.75	1.27
ENR-g-GA/Fe _{0.05} ³⁺	14.99	0.687
ENR-g-GA/Fe _{0.15} ³⁺	20.8	0.583

Table S6: Frequency sweep

	Log frequency	Log E', E''
ENR/Fe _{0.10} ³⁺	-1.85	6.12
ENR-g-GA/Fe _{0.05} ³⁺	-1.64	6.257
ENR-g-GA/Fe _{0.10} ³⁺	-1.44	6.287

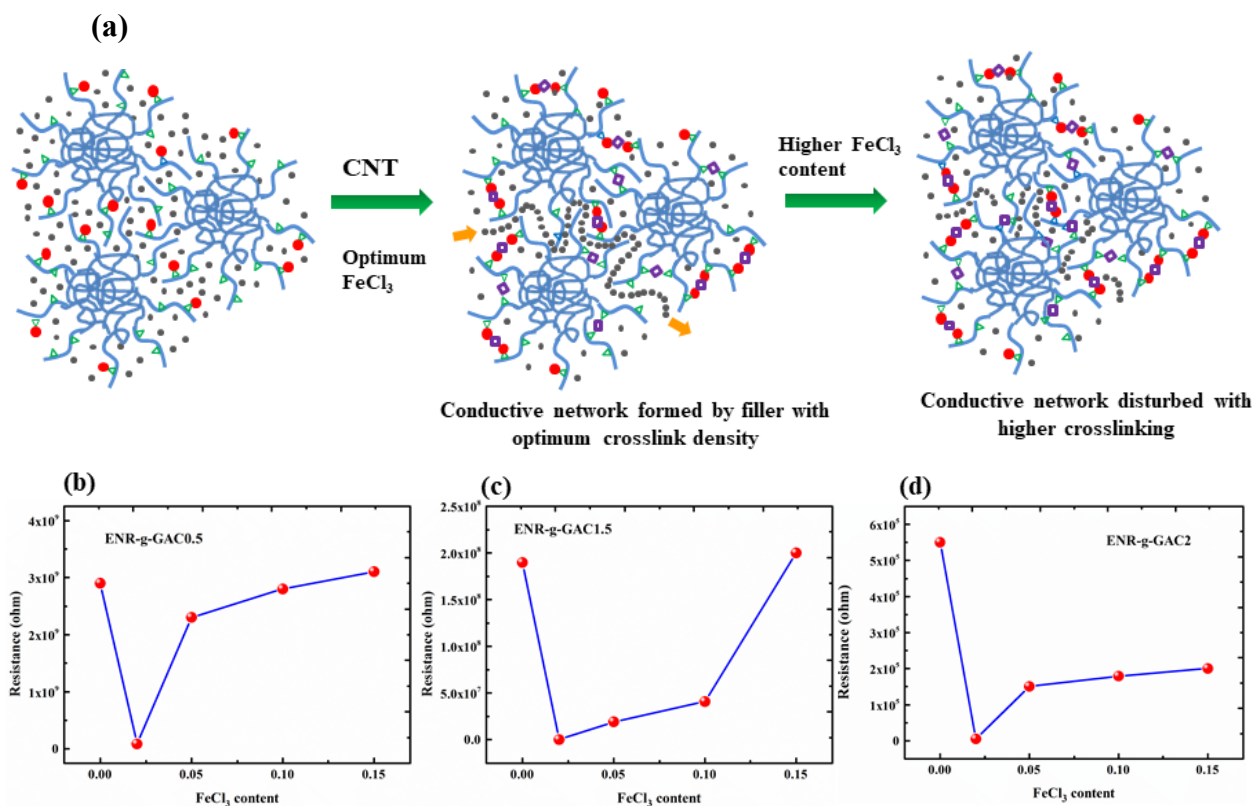


Fig. S6: (a) Schematic representation of effect of conductivity with FeCl_3 content. Resistance vs FeCl_3 content for ENR-g-GAC0.5 (CNT content 0.5%) (b), ENR-g-GAC1.5 (CNT content 1.5%) (c), ENR-g-GAC2 (CNT content 2%) (e).

Figure S6a demonstrated the schematic diagram of effect of crosslink (with respect to FeCl_3 content) on the conductivity. Figure S6(b-d) represent the resistance value of the composite film with varying FeCl_3 content for CNT loading of 0.5%, 1.5%, 2%. All this figure showed a decrease in resistance for FeCl_3 0.02, then increases with increasing amount of FeCl_3 . This can be explained by the fact that, at an optimum FeCl_3 content, the conductive filler particles are forced to come in a close proximity. Therefore, a small amount of FeCl_3 form a conductive filler network which reduced the resistance of the composite film. However, with increasing FeCl_3 content the filler particles are bound in small area due to the increase in crosslink density, which imparts disconnect in the filler network. Therefore, with increasing FeCl_3 beyond an optimal value the resistance value increases.

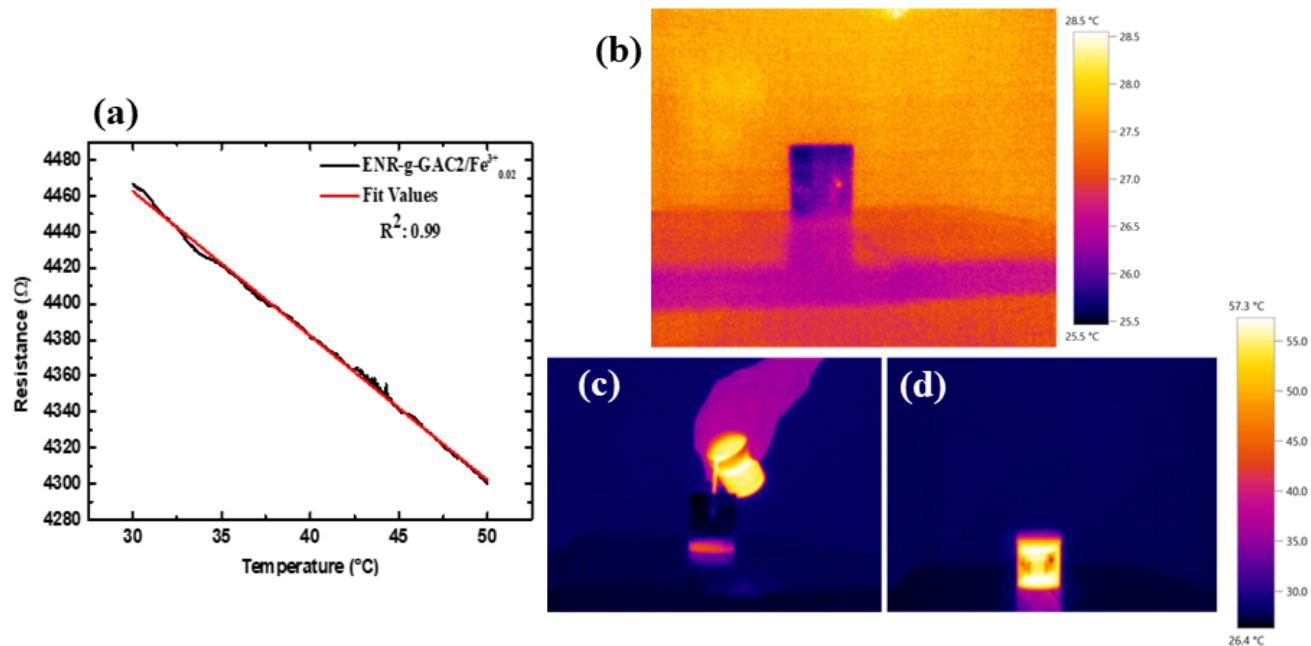


Fig. S7: (a) resistance vs temperature plot, (b-d) thermal camera image

Table S7: Tensile strength of the sensor and the heated sensor sample.

Sample	Tensile strength (MPa)	EAB	Efficiency of tensile strength	Efficiency of EAB
ENR-g-GAC2/ $Fe_{0.02}^{3+}$	3.2	617		
ENR-g-GAC2/ $Fe_{0.02}^{3+}H$	1.0	146	31.2%	23.6%