

SUPPORTING INFORMATION

Controllable redox reaction cycle enabled by multifunctional Ru-containing polyoxometalates-based catalysts

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Experimental

Materials

All reagents were purchased from Sigma-Aldrich and TCI without any further purification.

Characterization

The compounds **1–3** powder was characterized by inductively coupled plasma optical emission spectrometry (ICP-OES, PerkinElmer Optima 2100 DV inductively coupled plasma), Fouriertransform infrared spectroscopy (FTIR, Bruker VERTEX 70), Powder X-ray diffraction (XRD, Bruker D8 Advance diffractometer), Ultraviolet-visible spectrophotometry (UV-vis, Hitachi U-4100 spectrophotometer), Thermogravimetric analysis (TGA, NETZSCH STA449F5/QMS403D), X-ray photoelectron spectra (XPS, Kratos, Manchester, UK), and EPR spectra (Bruker EMX580 spectrometer). Ultraviolet photoemission spectroscopy (UPS, ESCALAB 250Xi, Thermo Fisher Scientific) was performed to evaluate the energy band structure using the He I UPS spectrum from the excitation energy (21.22 eV). The reaction mixture was stirred and heated at the prescribed conditions on the parallel reactor of WATTCAS-1020 H. The photocatalytic reactions were performed on WATTCAS Parallel Light Reactor (WP-TEC-1020HSL) with 10 W COB LED. GC analyses were performed on a Bruker 450-GC with a flame ionization detector equipped with a 30 m column (GsBP-1, 0.25 mm internal diameter and 0.25 µm film thickness) with nitrogen as the carrier gas. ¹H NMR and ¹³C NMR spectra were taken on Bruker 500 MHz Liquid State NMR Spectrometer (¹H 500 MHz, ¹³C 126 MHz). Chemical shifts of ¹H NMR spectra and ¹³C NMR spectra were reported using residual solvent signal of CDCl₃ (δ = 7.26 ppm) and CDCl₃ (δ = 77.0 ppm) as internal standard, respectively.

Preparation of 2

Compound **2** was synthesized by using Na₂WO₄·2H₂O (4.0 g, 12 mmol) and Sb₂O₃ (0.35 g, 1.2 mmol) to 10 mL H₂O with stirring. After dissolution the pH was adjusted to 6.0 by adding 4 M HCl dropwise. RuCl₃ (0.25 g, 1.2 mmol) was added successively, followed by the adjustment pH = 4.5 with 4.0 M HCl (aq). This mixture was stirred until a black solution was formed. Then, CsCl (0.2 g, 1.2 mmol) was dissolved in above solution with stirred for 2 h at room temperature. The black solution was sealed in a Teflon-lined autoclave and heated at 200 °C for 72 h. After cooling down to room temperature, the filtrate was kept in a 25 mL beaker to allow slow evaporation at room temperature. After 2 weeks, black block crystals suitable for X-ray crystallography were isolated, filtrated, and air-dried to give 256 mg of **2** (7.8 % yield, based on W). Anal. Calcd (%) for **2**: Na, 0.89; Cs, 1.29; Ru, 0.98; W, 63.52; Sb, 3.56. Found: Na, 1.04; Cs, 1.11; Ru, 0.94; W, 64.14; Sb, 3.50. IR (KBr pellet) for **2**: λ = 3477 (s), 1620 (m), 972 (s), 906 (s), 840 (m) and 681 (m) cm⁻¹.

Preparation of 3

Compound **3** was synthesized by using Na₂WO₄·2H₂O (4.0 g, 12 mmol) and Sb₂O₃ (0.35 g, 1.2 mmol) to 10 mL H₂O with stirring. After dissolution the pH of the mixture was adjusted to 4.2 with 4.0 M HCl (aq) under stirring continuously. The yellow solution was sealed in a Teflon-lined autoclave and heated at 200 °C for 72 h. After cooling down to room temperature, the filtrate was kept in a 25 mL beaker to allow slow evaporation at room temperature. After 2 weeks, yellow block crystals suitable for X-ray crystallography were isolated, filtrated, and air-dried to give 528 mg of **3** (16.6 % yield, based on W). Anal. Calcd (%) for **3**: Na, 1.28; W, 68.72; Sb, 4.08. Found: Na, 1.42; W, 69.98; Sb, 4.24. IR (KBr pellet) for **3**: λ = 3477 (s), 1620 (m), 971 (s), 906 (s), 840 (m) and 685 (m) cm⁻¹.

X-ray crystallography

Suitable quality crystals of **1–3** were directly mounted in a cooled nitrogen gas stream on a Bruker D8 VENTURE PHOTON II CCD diffractometer with graphite-monochromated Mo K α radiation (the value of λ is 0.71073 Å). The structure was solved with the ShelXT structure solution program using intrinsic phasing and further refined by the full-matrix least-squares method on F² using the ShelXL refinement package within Olex-2. During the refinement, the command ‘omit –3 50.2’ was used for compounds **1–3** to remove the weak reflections above 50 degrees. In the last refinement cycles, all the atoms of the polyoxoanions **1–3** (Sb, Ru, W, Na, Cs, Cl and O) were refined anisotropically. All lattice water molecules were positioned using Fourier maps and established by TG analyses. The crystallographic data and structural determination of **1–3** are listed in Table S1. Further details on the crystal structure investigations quoting the depository numbers CCDC 2172137–2172139.

Heat treatment experiments

Compounds **1** and **3** (1.0 g) was transferred to a temperature-programmed tube furnace under N₂ flow (40–60 mL min⁻¹) and subsequently heat treated at 325 °C for 4 h with a heating rate of 5 °C min⁻¹. The sample was then cooled to room temperature naturally under Ar flow to obtain **1-HT** and **3-HT**.

Recycling experiments

The products of AOB, AB, NSB, NB and AN are soluble in the corresponding reaction system, but can successfully be separated by extraction with normal hexane without interfering with the catalyst. The experimental procedure was as follows: After the reaction is completed, an excess of normal hexane is added to the reaction system. After the intense shock stratification, the catalyst is retained in the solvent phase, and the product is retained in the normal hexane system. The two phases are separated, and the recovered catalyst was obtained by rotary evaporation in the solvent phase. The reusability was tested by three consecutive times for model substrates after the initial reaction (Figure 5).

Analysis of data

Compounds **1–3** were synthesized through a simple control composition strategy and characterized by FTIR, XRD, UV-vis, TGA, ICP and XPS analyses. The phase purity of compounds **1–3** are identified by the conformity between the experimental PXRD pattern and the simulated pattern on the strength of structural analyses (Figure 2a).¹ The characteristic vibration patterns of the lacunary Keggin-type POM framework dominate the IR spectra of **1–3** (Figure 2b). Several same characteristic bands were observed at 976, 906, 840 and 685 cm⁻¹ which are assigned to the characteristic $\nu_{as}(Sb-O_a)$, $\nu_{as}(W-O_t)$, $\nu_{as}(W-O_b)$, and $\nu_{as}(W-O_c)$.² The ultraviolet-visible (UV-vis) spectra in Figure S3 show that **1** and **2** exhibited one main absorption band with maxima at 400 nm, which could be assigned to the charge-transfer transition of O→Ru.^{3,4} The determined number of crystal water molecules in **1–3** was 43, 113 and 56, respectively, as determined by the weight loss of TGA between 30–500 °C (Figures 4a, S1 and S2). Besides, the weight loss between 200 and 400 °C indicates the coordination of 18, 13 and 9 water molecules in **1–3**, which is agreement with the result by bond valence sum calculations. Elemental analyses were conducted by ICP-OES, confirming that the atomic ratio of W:Ru was 96.9:3.1 and 98.5:1.5 for **1** and **2**, respectively.

Supporting results

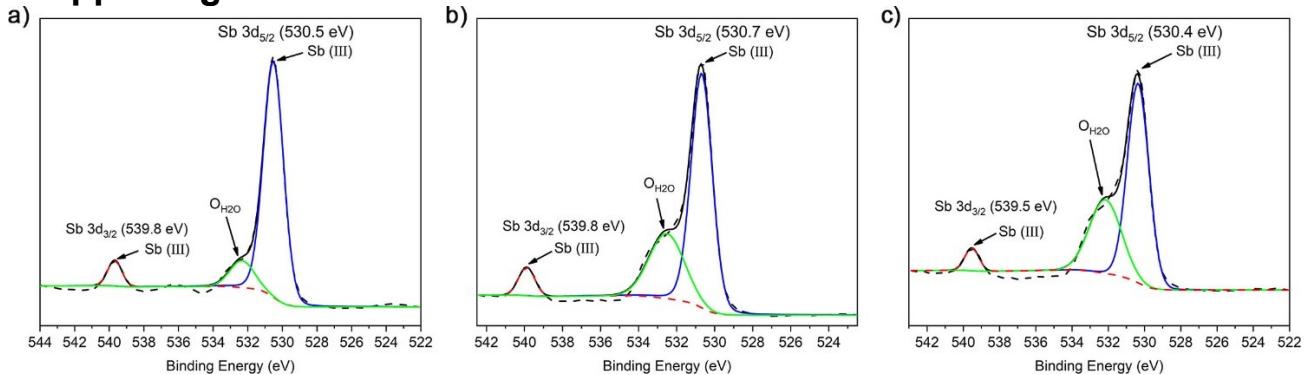


Figure S1. The XPS for Sb 3d for a) 1; b) 2; c) 3.

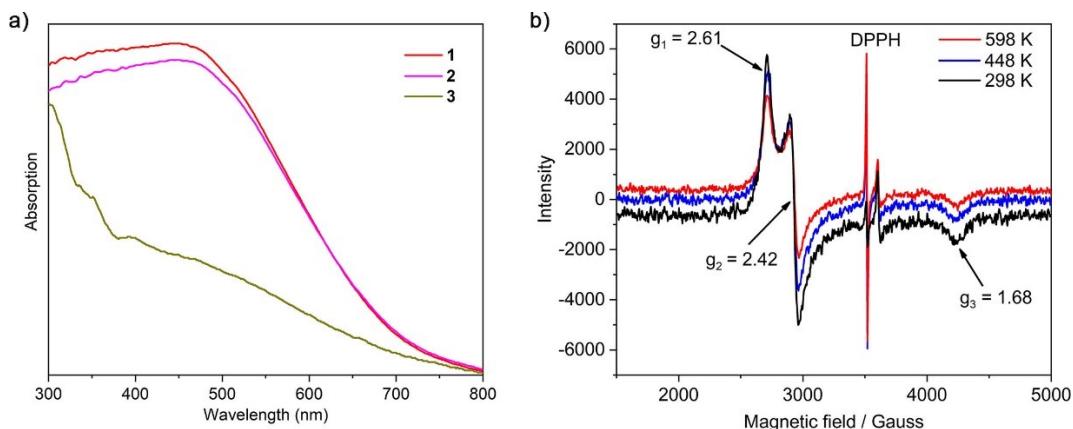


Figure S2. a) UV-vis DRS of compounds 1-3;b) the EPR spectra of heat treatment of 1. DPPH = di(phenyl)-(2,4,6-trinitrophenyl)iminoazanium radical was used as a standard.

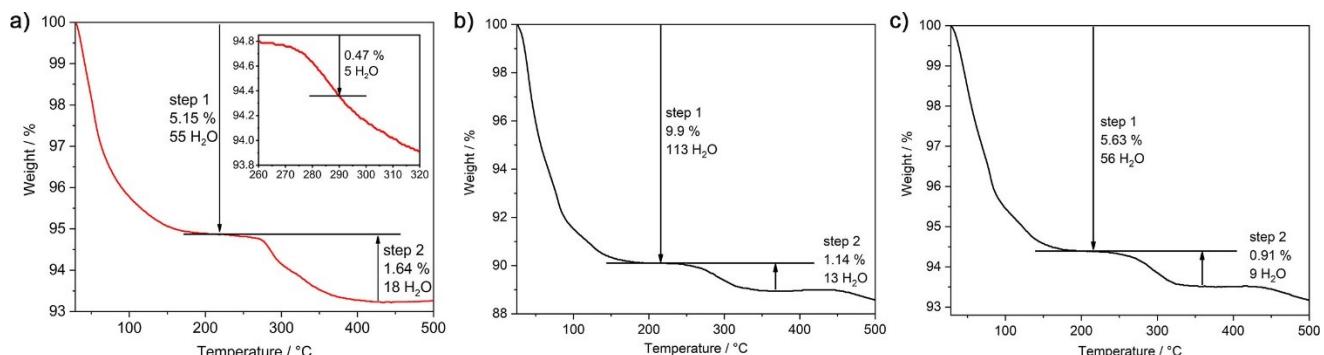


Figure S3. Thermogravimetric analysis curve for a) 1; b) 2; c) 3.

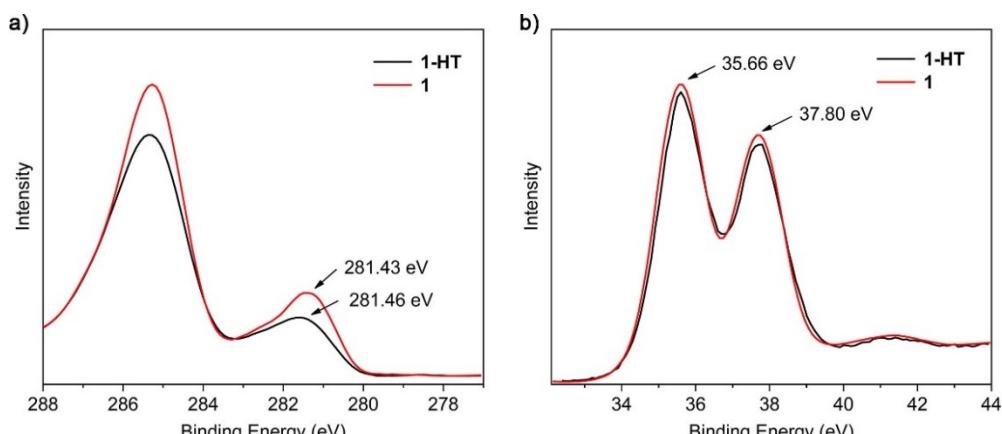


Figure S4. XPS spectra of a) Ru3d; b) W4f.

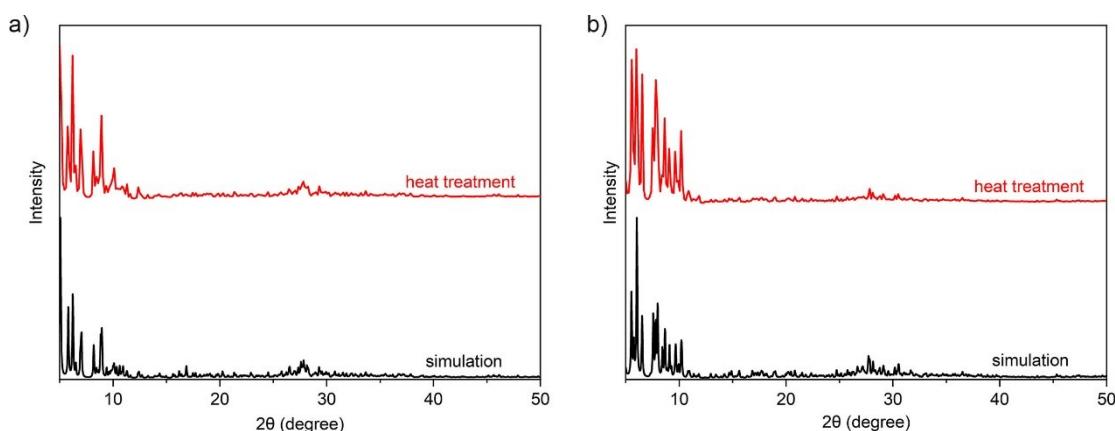
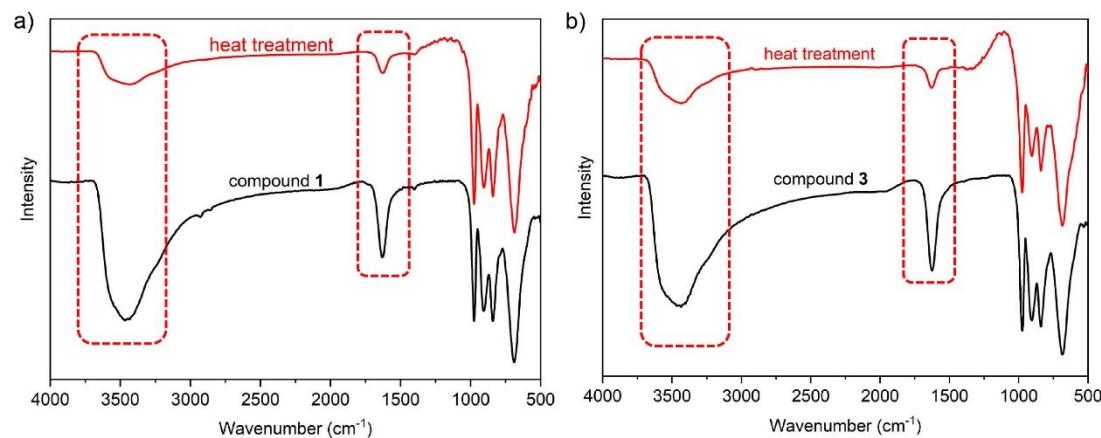
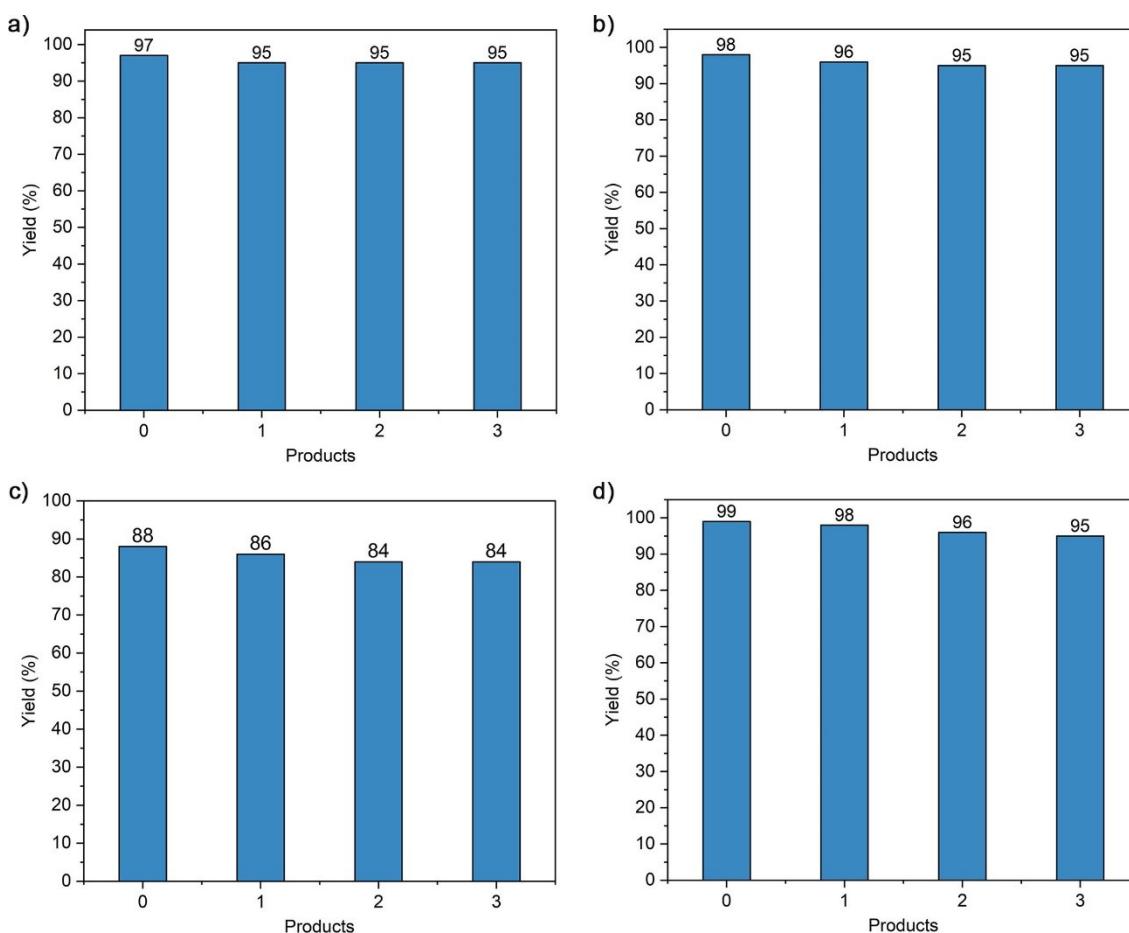


Figure S5. IR spectra after heat treatment of a) **1** and b) **3**.

Figure S6. XRD spectra after heat treatment of a) **1**, and b) **3**.

Figure S7. Recyclability tests for oxidation of AN to AOB, AB, NSB, NB in three consecutive runs, respectively.



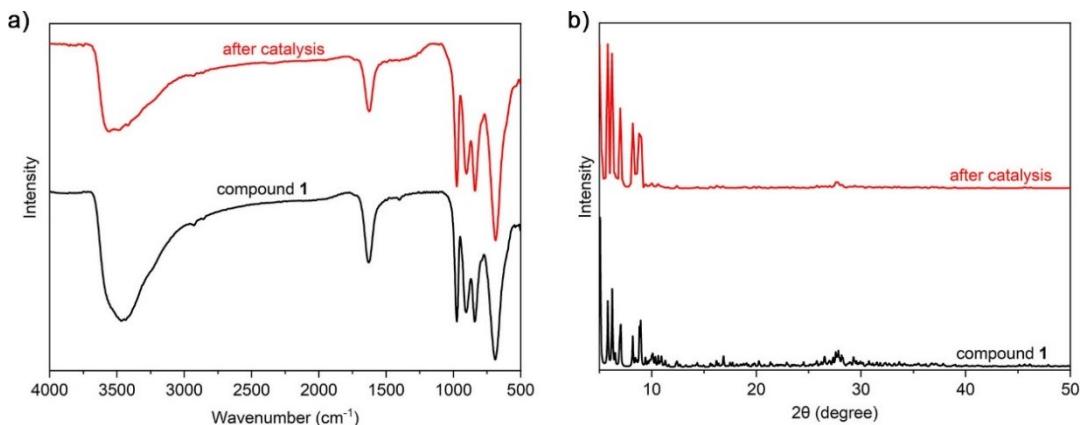


Figure S8. a) The IR spectra of **1** after catalysis; b) The XRD spectra of **1** after catalysis.

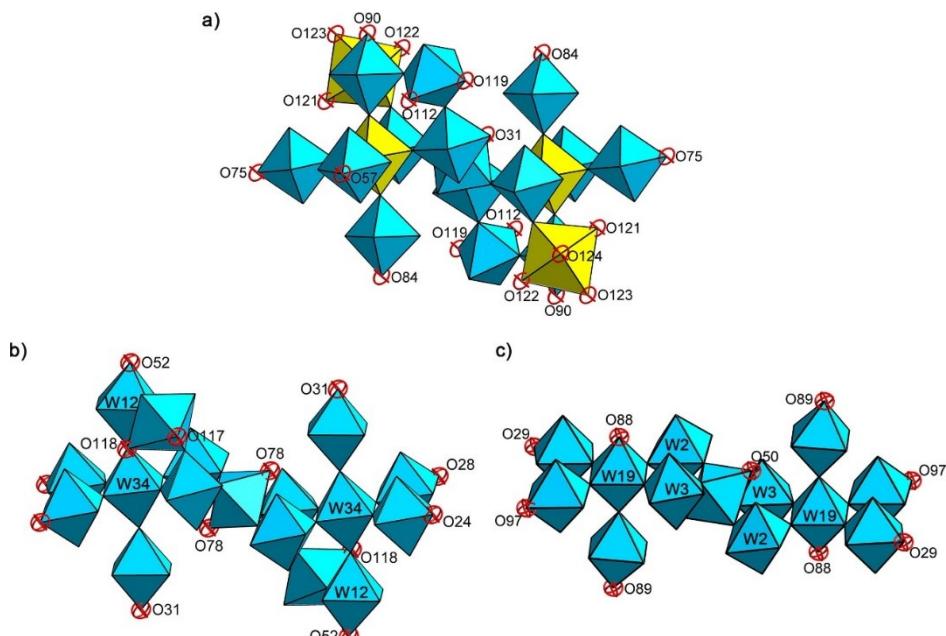


Figure S9. a) The position of water molecules for **1**; b) The position of water molecules for **2**; c) The position of water molecules for **3**. (RuO₆, yellow polytope; WO₆, sky blue polytope)

The structural differences of the three compounds are mainly reflected in the intermediate metallic oxygen clusters. Two WO₆ hexahedrons connect two W19 and W3 polyhedra in compound **3**, respectively, then, the anion carries two ruthenium atoms on the structure outside to form compound **2**. Four W12 and W34 atoms in compound **2** are replaced by Ru atoms to form compound **1**. In addition, the detail explanation of structural differences about three compounds also should be provided in the revised supplement information.

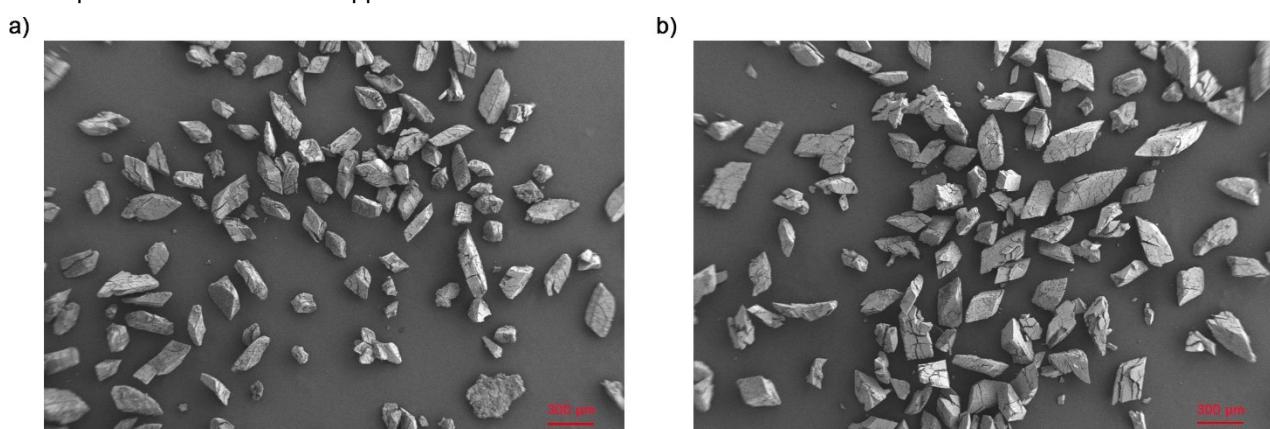
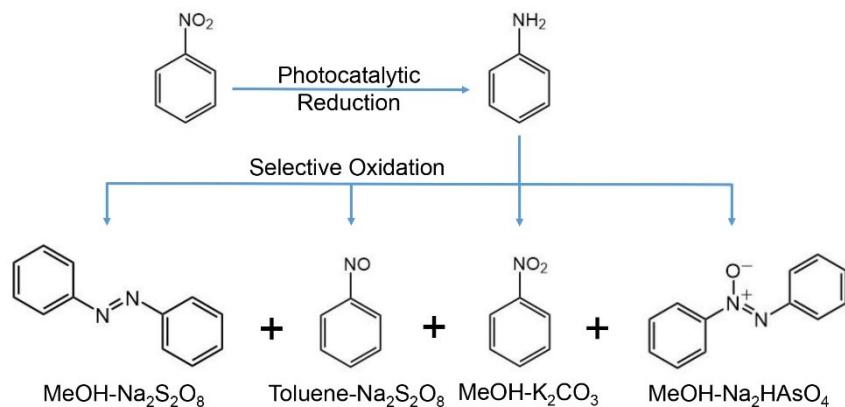
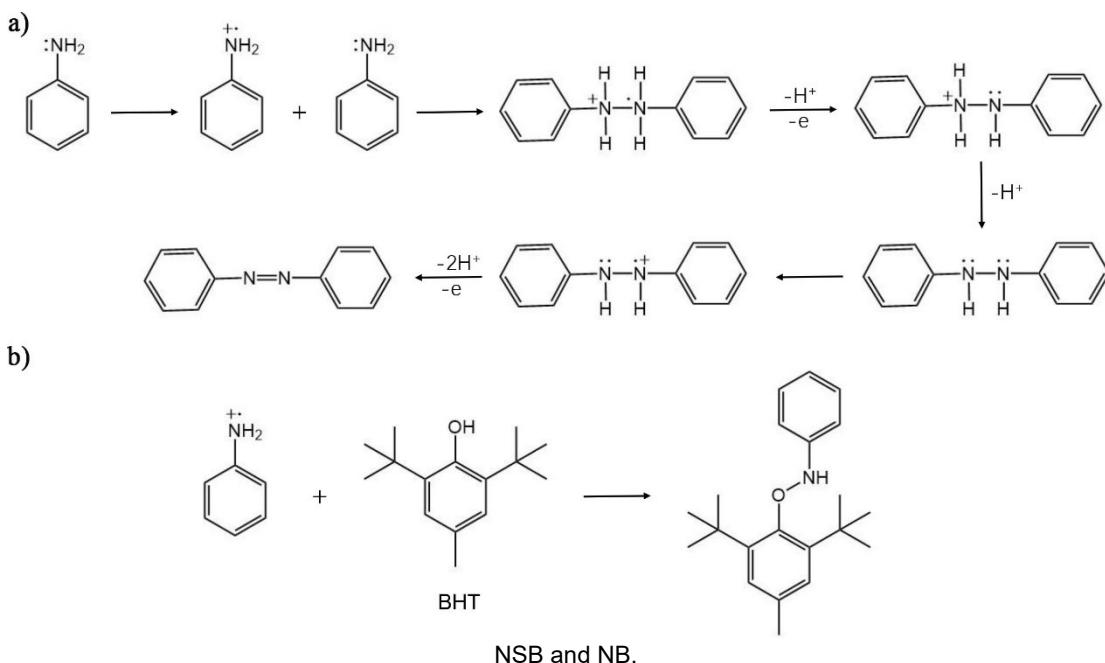


Figure S10. SEM images of a) **1** and b) **2**.

This work:



Scheme S1. The reaction systems of photocatalytic reduction of NB to anilines and selective oxidation of anilines to AOB, AB,



Scheme S2. a) The synthesis mechanism of azobenzene from aniline in oxygen atmosphere proposed by Corma; b) Spin trapping mechanism of aniline radical with BHT. The reaction mixture was analyzed by GC-MS after cooling to room temperature.

Scheme S3. Reaction mechanism of NB reduction to anilines proposed by Haber.

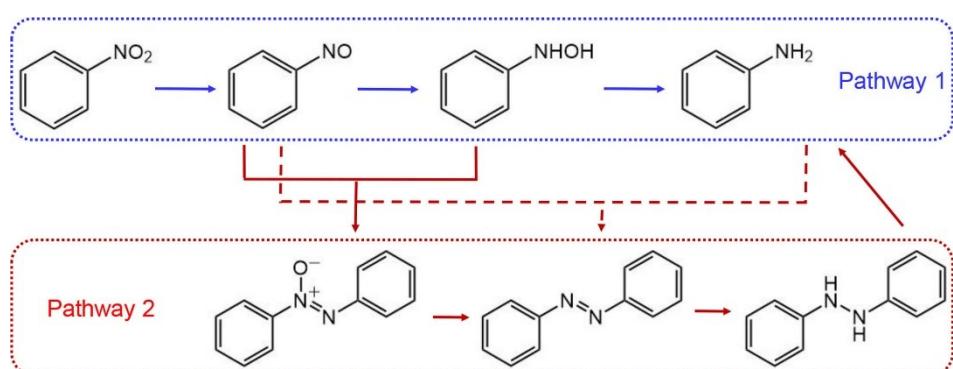


Table S1. Crystal data and structure refinement for compounds **1-3**.

Compound	1	2	3
Formula	$\text{Na}_4\text{Cs}_2\text{Sb}_6\text{O}_{273}\text{Ru}_4\text{W}_{69}\text{Cl}_3$	$\text{Na}_8\text{Cs}_2\text{Sb}_6\text{O}_{296}\text{Ru}_2\text{W}_{71}\text{Cl}_6$	$\text{Na}_{10}\text{Sb}_6\text{O}_{278}\text{W}_{67}$
<i>M</i>	18652.56	19384.43	17726.35
<i>λ/Å</i>	0.71073	0.71073	0.71073
<i>T/K</i>	150	150	150
Crystal dimensions/mm	$0.11 \times 0.08 \times 0.07$	$0.18 \times 0.16 \times 0.12$	$0.1 \times 0.05 \times 0.04$
Crystal system	triclinic	triclinic	triclinic
Space group	P-1	P-1	P-1
<i>a/Å</i>	20.0315 (5)	20.3555 (7)	18.065 (9)
<i>b/Å</i>	20.3566 (4)	20.4304 (6)	20.373 (9)
<i>c/Å</i>	21.2690 (5)	23.8990 (10)	24.517 (13)
α (°)	98.9870 (10)	79.8480 (10)	76.462 (14)
β (°)	115.7630 (10)	64.8990 (10)	73.843 (13)
γ (°)	92.0170 (10)	63.0760 (10)	64.497 (15)
<i>V/Å³</i>	7663.0 (3)	8023.2 (5)	7754 (7)
<i>Z</i>	1	1	1
$D_v/\text{Mg m}^{-3}$	4.042	4.012	3.796
μ/mm^{-1}	26.844	26.301	25.356
<i>F</i> (000)	7977.0	8316.0	7598.0
2θ Range/°	4.332 to 50.198	4.342 to 50.2	4.386 to 50.198
Data/restraints/parameters	27229/47/1666	28494/79/1765	27614/87/1639
$R_1(I > 2\sigma(I))^a$	$R_1 = 0.0463$	$R_1 = 0.0535$	$R_1 = 0.0527$
wR_2 (all data) ^b	$wR_2 = 0.1365$	$wR_2 = 0.1611$	$wR_2 = 0.1617$
Goodness-of-fit on F^2	1.025	1.038	1.021

^a $R_1 = \sum |F_o| - |F_c| / \sum |F_o|$. ^b $wR_2 = \{\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]\}^{1/2}$

Table S2. Selected bond distances of compound 1.

Bond	Bond length	Bond	Bond length	Bond	Bond length
Ru1–O3	2.043 (11)	Ru1–O11	1.893 (11)	Ru1–O12	1.926 (11)
Ru1–O22	2.072 (11)	Ru1–O24	1.905 (11)	Ru1–O41	2.043 (10)
Ru2–O92	2.068 (10)	Ru2–O121	2.430 (13)	Ru2–O122	2.18 (3)
Ru2–O123	2.21 (3)	Ru2–O124	2.25 (3)	Ru2–O108	2.114 (11)
Ru3–Cl1	2.363 (10)	Ru3–Cl2	2.348 (12)	Ru3–Cl3	2.361 (11)
Ru3–O89	2.032 (12)	Ru3–O96	1.991 (13)	Ru3–O97	2.017 (12)
Sb1–O4	1.994 (10)	Sb1–O13	2.011 (9)	Sb1–O40	2.003 (10)
Sb2–O59	1.990 (9)	Sb2–O66	1.985 (10)	Sb2–O27	1.994 (9)
Sb3–O53	2.020 (10)	Sb3–O55	1.972 (10)	Sb3–O46	2.001 (10)
W1–O1	2.074 (10)	W1–O2	1.897 (9)	W1–O3	1.755 (10)
W1–O10	1.920 (9)	W1–O19	2.220 (10)	W1–O47	1.781 (10)
W2–O5	2.116 (10)	W2–O6	1.941 (9)	W2–O92	1.771 (10)
W2–O41	1.764 (10)	W2–O17	1.911 (9)	W2–O52	2.231 (9)
W3–O52	1.739 (10)	W3–O1	1.839 (10)	W3–O70	1.975 (10)
W3–O13	2.203 (9)	W3–O30	1.915 (10)	W3–O36	1.959 (10)
W4–O5	1.818 (9)	W4–O14	1.983 (10)	W4–O18	1.901 (10)
W4–O19	1.769 (10)	W4–O25	1.949 (10)	W4–O27	2.225 (10)
W5–O59	2.276 (10)	W5–O61	1.960 (11)	W5–O6	1.900 (10)
W5–O18	1.925 (10)	W5–O26	1.916 (10)	W5–O99	1.708 (11)
W6–O20	2.088 (10)	W6–O22	1.755 (11)	W6–O87	1.851 (10)
W6–O90	2.264 (11)	W6–O34	2.002 (10)	W6–O108	1.766 (10)
W7–O50	1.964 (10)	W7–O59	2.284 (10)	W7–O61	1.920 (10)
W7–O21	1.882 (10)	W7–O28	1.935 (10)	W7–O105	1.722 (11)
W8–O7	1.904 (10)	W8–O13	2.291 (11)	W8–O16	1.910 (10)
W8–O80	1.924 (10)	W8–O36	1.896 (10)	W8–O106	1.715 (11)
W9–O48	1.950 (11)	W9–O4	2.295 (10)	W9–O10	1.927 (10)
W9–O91	1.704 (12)	W9–O30	1.906 (10)	W9–O38	1.913 (9)
W10–O55	2.256 (11)	W10–O81	1.991 (11)	W10–O29	1.910 (9)
W10–O93	1.722 (11)	W10–O33	1.947 (10)	W10–O42	1.897 (11)
W11–O14	1.884 (11)	W11–O27	2.280 (10)	W11–O35	1.951 (11)
W11–O98	1.728 (11)	W11–O39	1.902 (10)	W11–O43	1.916 (10)
W12–O63	1.996 (11)	W12–O64	1.961 (10)	W12–O74	1.935 (10)
W12–O20	1.831 (10)	W12–O88	1.734 (11)	W12–O46	2.275 (10)
W13–O55	2.305 (11)	W13–O74	1.903 (11)	W13–O32	1.951 (10)
W13–O34	1.856 (10)	W13–O100	1.713 (12)	W13–O42	1.990 (11)
W14–O62	1.951 (11)	W14–O64	1.913 (12)	W14–O8	1.919 (9)
W14–O85	1.927 (11)	W14–O103	1.709 (11)	W14–O46	2.260 (10)
W15–O51	1.915 (11)	W15–O70	1.902 (10)	W15–O13	2.314 (9)
W15–O80	1.937 (11)	W15–O86	1.921 (11)	W15–O111	1.707 (12)
W16–O56	1.913 (11)	W16–O68	1.900 (11)	W16–O25	1.924 (11)
W16–O27	2.320 (9)	W16–O39	1.945 (10)	W16–O113	1.704 (11)
W17–O66	2.229 (10)	W17–O69	1.937 (11)	W17–O76	1.945 (12)
W17–O23	1.907 (9)	W17–O97	1.724 (12)	W17–O45	1.882 (10)
W18–O56	1.891 (11)	W18–O59	2.315 (10)	W18–O78	1.912 (11)
W18–O26	1.919 (11)	W18–O28	1.952 (10)	W18–O107	1.702 (11)
W19–O53	2.319 (10)	W19–O67	1.949 (11)	W19–O81	1.879 (10)
W19–O82	1.939 (11)	W19–O83	1.920 (10)	W19–O104	1.707 (12)
W20–O53	2.232 (11)	W20–O54	1.889 (10)	W20–O77	1.926 (11)
W20–O83	1.938 (10)	W20–O89	1.714 (11)	W20–O37	1.902 (10)
W21–O67	1.885 (11)	W21–O55	2.338 (10)	W21–O79	1.907 (10)

W21–O32	1.945 (11)	W21–O33	1.902 (10)	W21–O101	1.726 (11)
W22–O48	1.898 (11)	W22–O49	1.944 (10)	W22–O4	2.276 (10)
W22–O71	1.964 (12)	W22–O15	1.926 (10)	W22–O102	1.699 (12)
W23–O49	1.922 (11)	W23–O51	1.893 (11)	W23–O4	2.295 (9)
W23–O73	1.858 (11)	W23–O38	1.956 (11)	W23–O226	1.728 (11)
W24–O50	1.884 (10)	W24–O58	1.914 (12)	W24–O66	2.333 (10)
W24–O76	1.922 (11)	W24–O78	1.938 (11)	W24–O120	1.721 (10)
W25–O58	1.958 (11)	W25–O66	2.346 (11)	W25–O68	1.935 (11)
W25–O69	1.919 (11)	W25–O95	1.687 (11)	W25–O35	1.913 (11)
W26–O63	1.901 (12)	W26–O79	1.902 (10)	W26–O85	1.942 (11)
W26–O94	1.900 (12)	W26–O115	1.717 (12)	W26–O46	2.317 (10)
W27–O53	2.300 (10)	W27–O62	1.894 (11)	W27–O77	1.966 (11)
W27–O82	1.917 (11)	W27–O94	1.931 (13)	W27–O110	1.712 (10)
W28–O54	1.900 (10)	W28–O75	2.173 (11)	W28–O21	1.979 (10)
W28–O23	1.876 (9)	W28–O24	1.809 (11)	W28–O29	1.952 (10)
W29–O60	1.935 (11)	W29–O65	1.946 (11)	W29–O9	1.897 (10)
W29–O96	1.765 (13)	W29–O40	2.235 (11)	W29–O44	1.895 (10)
W30–O60	1.913 (12)	W30–O72	1.928 (11)	W30–O16	1.922 (10)
W30–O86	1.923 (11)	W30–O40	2.332 (11)	W30–O109	1.729 (12)
W31–O7	1.925 (10)	W31–O9	1.886 (10)	W31–O12	1.816 (11)
W31–O84	2.169 (12)	W31–O43	1.930 (10)	W31–O45	1.896 (9)
W32–O57	2.159 (11)	W32–O8	1.930 (10)	W32–O11	1.829 (11)
W32–O15	1.933 (11)	W32–O37	1.891 (10)	W32–O44	1.882 (10)
W33–O65	1.894 (12)	W33–O71	1.878 (12)	W33–O72	1.949 (11)
W33–O73	1.984 (12)	W33–O40	2.313 (10)	W33–O114	1.706 (12)
W34–O87	1.936 (11)	W34–O112	2.424 (16)	W34–O117	1.746 (16)
W34–O118	1.741 (16)	W34–O119	2.097 (11)	W34–O47	2.128 (10)
W35–O2	1.908 (9)	W35–O17	1.892 (9)		

Table S3. Selected bond distances of compound 2.

Bond	Bond length	Bond	Bond length	Bond	Bond length
Ru1–Cl1	2.335 (13)	Ru1–Cl2	2.309 (9)	Ru1–Cl3	2.322 (16)
Ru1–O54	2.013 (14)	Ru1–O87	1.987 (13)	Ru1–O94	2.002 (13)
Sb1–O50	1.956 (12)	Sb1–O63	2.020 (12)	Sb1–O64	1.994 (11)
Sb2–O61	2.000 (13)	Sb2–O25	2.021 (12)	Sb2–O34	2.008 (11)
Sb3–O62	1.993 (12)	Sb3–O69	1.997 (11)	Sb3–O19	1.992 (12)
W1–O55	1.793 (11)	W1–O3	2.072 (11)	W1–O4	1.851 (11)
W1–O13	2.226 (11)	W1–O20	1.946 (10)	W1–O21	1.756 (11)
W2–O71	1.918 (11)	W2–O13	1.757 (11)	W2–O79	1.994 (12)
W2–O25	2.205 (12)	W2–O44	1.931 (12)	W2–O45	1.809 (11)
W3–O27	1.767 (12)	W3–O32	1.717 (12)	W3–O33	1.901 (11)
W3–O37	1.884 (11)	W3–O42	2.214 (13)	W3–O45	2.120 (12)
W4–O63	2.193 (12)	W4–O3	1.826 (11)	W4–O15	1.942 (11)
W4–O42	1.767 (13)	W4–O46	1.917 (12)	W4–O43	1.987 (11)
W5–O63	2.290 (12)	W5–O8	1.948 (11)	W5–O72	1.912 (12)
W5–O12	1.898 (11)	W5–O109	1.701 (13)	W5–O43	1.907 (12)
W6–O68	1.975 (13)	W6–O71	1.912 (12)	W6–O89	1.909 (12)
W6–O33	1.940 (11)	W6–O34	2.246 (12)	W6–O98	1.741 (14)
W7–O51	1.938 (12)	W7–O67	1.942 (14)	W7–O18	1.915 (11)
W7–O25	2.289 (12)	W7–O108	1.719 (13)	W7–O44	1.888 (12)
W8–O1	1.959 (12)	W8–O64	2.282 (12)	W8–O11	1.879 (12)
W8–O75	1.691 (12)	W8–O20	1.909 (11)	W8–O46	1.896 (12)
W9–O50	2.239 (12)	W9–O16	1.893 (13)	W9–O23	1.856 (13)
W9–O87	1.751 (13)	W9–O91	1.952 (12)	W9–O92	1.982 (12)
W10–O58	1.851 (13)	W10–O64	2.310 (11)	W10–O11	2.008 (12)
W10–O17	1.928 (13)	W10–O82	1.912 (13)	W10–O107	1.728 (12)
W11–O69	2.258 (12)	W11–O14	1.843 (11)	W11–O90	1.958 (13)
W11–O35	1.942 (12)	W11–O39	1.990 (14)	W11–O106	1.699 (14)
W12–O52	2.256 (12)	W12–O2	1.732 (14)	W12–O66	1.783 (11)
W12–O14	2.073 (12)	W12–O22	1.964 (11)	W12–O115	1.871 (14)
W13–O54	1.750 (14)	W13–O59	1.878 (12)	W13–O61	2.210 (14)
W13–O73	1.974 (13)	W13–O77	1.899 (13)	W13–O83	1.949 (13)
W14–O63	2.300 (12)	W14–O72	1.937 (12)	W14–O15	1.924 (11)
W14–O82	1.907 (12)	W14–O30	1.879 (12)	W14–O102	1.716 (12)
W15–O53	1.947 (12)	W15–O60	1.857 (12)	W15–O9	1.887 (12)
W15–O19	2.222 (13)	W15–O94	1.731 (13)	W15–O38	1.949 (12)
W16–O62	2.275 (12)	W16–O5	1.896 (12)	W16–O6	1.977 (13)
W16–O86	1.875 (14)	W16–O97	1.939 (12)	W16–O111	1.702 (13)
W17–O62	2.275 (13)	W17–O7	1.921 (14)	W17–O22	1.889 (11)
W17–O86	2.007 (12)	W17–O90	1.876 (13)	W17–O100	1.697 (14)
W18–O50	2.361 (12)	W18–O8	1.904 (12)	W18–O92	1.902 (12)
W18–O30	1.973 (12)	W18–O96	1.708 (13)	W18–O41	1.932 (12)
W19–O57	1.967 (12)	W19–O68	1.904 (12)	W19–O81	1.938 (13)
W19–O34	2.272 (12)	W19–O104	1.712 (14)	W19–O47	1.876 (11)
W20–O69	2.272 (11)	W20–O70	1.911 (13)	W20–O35	1.919 (11)
W20–O105	1.718 (13)	W20–O40	1.913 (13)	W20–O48	1.946 (12)
W21–O49	1.867 (12)	W21–O50	2.360 (11)	W21–O58	1.973 (12)
W21–O91	1.891 (14)	W21–O99	1.716 (13)	W21–O41	1.928 (12)
W22–O49	1.992 (12)	W22–O1	1.924 (12)	W22–O64	2.281 (12)
W22–O17	1.983 (12)	W22–O85	1.730 (13)	W22–O29	1.864 (11)
W23–O74	1.880 (14)	W23–O76	1.878 (13)	W23–O81	1.928 (14)

W23–O89	1.939 (14)	W23–O34	2.320 (11)	W23–O119	1.729 (12)
W24–O65	1.940 (13)	W24–O6	1.871 (13)	W24–O80	1.913 (13)
W24–O19	2.325 (12)	W24–O38	1.909 (13)	W24–O112	1.723 (14)
W25–O53	1.933 (12)	W25–O80	1.960 (13)	W25–O19	2.336 (12)
W25–O93	1.928 (15)	W25–O95	1.701 (15)	W25–O40	1.912 (13)
W26–O51	1.935 (12)	W26–O56	1.924 (14)	W26–O76	1.916 (13)
W26–O79	1.897 (13)	W26–O25	2.305 (11)	W26–O114	1.727 (13)
W27–O69	2.324 (12)	W27–O93	1.895 (15)	W27–O101	1.707 (13)
W27–O103	1.889 (13)	W27–O39	1.911 (14)	W27–O48	1.935 (12)
W28–O62	2.317 (12)	W28–O65	1.873 (13)	W28–O7	1.971 (13)
W28–O97	1.939 (13)	W28–O103	1.920 (13)	W28–O113	1.709 (13)
W29–O57	1.862 (12)	W29–O61	2.329 (12)	W29–O73	1.888 (15)
W29–O74	1.964 (15)	W29–O88	1.917 (14)	W29–O120	1.732 (14)
W30–O56	1.924 (14)	W30–O61	2.329 (13)	W30–O67	1.896 (13)
W30–O83	1.929 (15)	W30–O88	1.941 (13)	W30–O116	1.721 (14)
W31–O12	1.936 (12)	W31–O16	1.879 (13)	W31–O77	1.876 (12)
W31–O18	1.920 (12)	W31–O26	1.969 (12)	W31–O31	2.037 (11)
W32–O59	1.899 (12)	W32–O5	1.944 (12)	W32–O9	1.889 (12)
W32–O84	1.939 (12)	W32–O24	2.064 (11)	W32–O47	1.965 (12)
W33–O60	1.906 (12)	W33–O10	1.897 (13)	W33–O70	1.932 (13)
W33–O23	1.915 (13)	W33–O28	2.128 (13)	W33–O29	1.984 (12)
W34–O66	2.089 (11)	W34–O10	1.836 (13)	W34–O21	2.108 (11)
W34–O84	1.828 (13)	W34–O26	1.864 (12)	W34–O27	2.087 (12)
W35–O55	2.112 (11)	W35–O115	1.907 (14)	W35–O117	2.102 (12)
W35–O118	2.393 (16)	W35–O121	1.734 (17)	W35–O122	1.730 (19)
W36–O4	1.925 (12)	W36–O78	1.700 (13)	W36–O37	1.923 (12)

Table S4. Selected bond distances of compound 3.

Bond	Bond length	Bond	Bond length	Bond	Bond length
Sb1–O66	2.016 (12)	Sb1–O40	1.996 (13)	Sb1–O48	1.994 (12)
Sb2–O14	2.019 (12)	Sb2–O20	1.996 (12)	Sb2–O34	1.980 (11)
Sb3–O9	2.010 (15)	Sb3–O83	1.994 (14)	Sb3–O109	1.990 (13)
W1–O2	1.815 (13)	W1–O14	2.208 (11)	W1–O18	1.985 (12)
W1–O22	1.946 (12)	W1–O30	1.743 (12)	W1–O32	1.957 (11)
W2–O26	1.930 (11)	W2–O28	1.799 (11)	W2–O93	1.734 (13)
W2–O6	1.877 (10)	W2–O62	2.217 (13)	W2–O2	2.177 (11)
W3–O4	2.165 (11)	W3–O8	1.925 (12)	W3–O12	1.878 (12)
W3–O102	1.737 (11)	W3–O42	1.758 (11)	W3–O30	2.224 (12)
W4–O62	1.765 (12)	W4–O64	1.942 (13)	W4–O66	2.210 (11)
W4–O4	1.805 (11)	W4–O16	1.922 (14)	W4–O38	1.980 (11)
W5–O52	1.890 (14)	W5–O20	2.275 (12)	W5–O22	1.887 (12)
W5–O26	1.918 (13)	W5–O94	1.971 (13)	W5–O103	1.714 (14)
W6–O74	1.957 (11)	W6–O76	1.939 (14)	W6–O77	1.709 (13)
W6–O14	2.283 (11)	W6–O24	1.898 (13)	W6–O32	1.894 (12)
W7–O8	1.928 (11)	W7–O16	1.895 (14)	W7–O82	1.966 (14)
W7–O90	1.739 (13)	W7–O36	1.893 (13)	W7–O40	2.265 (12)
W8–O54	1.928 (12)	W8–O76	1.932 (13)	W8–O14	2.309 (13)
W8–O18	1.904 (11)	W8–O110	1.739 (14)	W8–O46	1.896 (13)
W9–O70	1.874 (12)	W9–O98	1.987 (13)	W9–O34	2.279 (13)
W9–O100	1.872 (13)	W9–O101	1.922 (15)	W9–O107	1.669 (16)
W10–O67	1.975 (15)	W10–O72	1.870 (13)	W10–O11	1.971 (13)
W10–O43	1.720 (14)	W10–O44	1.879 (13)	W10–O48	2.260 (12)
W11–O66	2.307 (12)	W11–O13	1.899 (16)	W11–O91	1.912 (18)
W11–O38	1.902 (12)	W11–O108	1.894 (13)	W11–O47	1.747 (13)
W12–O61	1.679 (16)	W12–O64	1.905 (12)	W12–O66	2.284 (13)
W12–O68	1.944 (14)	W12–O10	1.892 (11)	W12–O91	1.957 (14)
W13–O52	1.996 (12)	W13–O54	1.896 (12)	W13–O81	1.716 (15)
W13–O20	2.312 (13)	W13–O96	1.845 (14)	W13–O112	1.918 (13)
W14–O60	1.849 (13)	W14–O82	1.902 (15)	W14–O86	1.983 (13)
W14–O87	1.704 (16)	W14–O40	2.251 (14)	W14–O111	1.979 (13)
W15–O58	1.845 (15)	W15–O7	1.981 (13)	W15–O20	2.274 (12)
W15–O27	1.692 (14)	W15–O94	1.914 (13)	W15–O112	1.975 (14)
W16–O63	1.943 (13)	W16–O71	1.917 (15)	W16–O78	1.912 (12)
W16–O84	1.903 (15)	W16–O85	1.720 (14)	W16–O109	2.283 (13)
W17–O74	1.880 (11)	W17–O34	2.328 (12)	W17–O101	1.889 (17)
W17–O104	1.942 (13)	W17–O45	1.728 (14)	W17–O46	1.941 (14)
W18–O13	1.911 (16)	W18–O86	1.871 (16)	W18–O31	1.695 (14)
W18–O36	1.972 (12)	W18–O106	1.854 (13)	W18–O40	2.339 (12)
W19–O56	1.763 (13)	W19–O88	2.205 (12)	W19–O92	1.754 (15)
W19–O28	2.070 (11)	W19–O42	2.122 (11)	W19–O113	1.764 (12)
W20–O3	1.715 (14)	W20–O7	1.896 (14)	W20–O96	1.971 (15)
W20–O98	1.872 (14)	W20–O34	2.332 (12)	W20–O104	1.934 (14)
W21–O68	1.886 (15)	W21–O11	1.923 (12)	W21–O79	1.932 (17)
W21–O25	1.716 (15)	W21–O108	1.949 (13)	W21–O48	2.285 (13)
W22–O65	1.956 (14)	W22–O9	2.257 (14)	W22–O95	1.864 (14)
W22–O99	1.865 (13)	W22–O35	1.698 (18)	W22–O39	1.908 (15)
W23–O55	1.939 (12)	W23–O59	1.981 (15)	W23–O83	2.314 (17)
W23–O37	1.704 (18)	W23–O105	1.889 (12)	W23–O41	1.867 (14)
W24–O48	2.345 (15)	W24–O5	1.702 (18)	W24–O67	1.891 (13)

W24–O79	1.909 (16)	W24–O106	1.970 (12)	W24–O111	1.866 (14)
W25–O53	1.922 (16)	W25–O57	1.894 (14)	W25–O1	1.735 (19)
W25–O65	1.887 (15)	W25–O71	1.932 (15)	W25–O109	2.275 (15)
W26–O53	1.930 (17)	W26–O59	1.861 (16)	W26–O63	1.896 (14)
W26–O73	1.930 (15)	W26–O21	1.742 (16)	W26–O109	2.307 (14)
W27–O55	1.938 (16)	W27–O69	1.831 (13)	W27–O73	1.939 (14)
W27–O75	1.883 (19)	W27–O17	1.686 (16)	W27–O83	2.335 (14)
W28–O49	1.758 (14)	W28–O51	1.94 (2)	W28–O9	2.241 (13)
W28–O80	2.052 (13)	W28–O23	1.730 (15)	W28–O95	2.001 (14)
W29–O57	1.961 (14)	W29–O9	2.323 (15)	W29–O75	1.908 (18)
W29–O80	1.847 (13)	W29–O33	1.672 (18)	W29–O39	1.934 (15)
W30–O51	1.90 (2)	W30–O69	2.079 (14)	W30–O15	1.732 (15)
W30–O19	1.792 (16)	W30–O83	2.223 (13)	W30–O41	2.010 (17)
W31–O56	2.204 (13)	W31–O70	1.901 (12)	W31–O10	1.962 (11)
W31–O24	1.950 (13)	W31–O89	1.769 (13)	W31–O44	1.904 (13)
W32–O56	1.612 (15)	W32–O70	1.969 (17)	W32–O10	1.962 (15)
W32–O24	1.989 (19)	W32–O89	2.358 (15)	W32–O44	1.922 (18)
W33–O60	2.007 (13)	W33–O72	1.916 (13)	W33–O78	1.895 (12)
W33–O92	2.120 (15)	W33–O97	2.052 (13)	W33–O99	1.954 (13)
W34–O58	1.996 (15)	W34–O84	1.878 (15)	W34–O29	2.035 (12)
W34–O100	1.903 (13)	W34–O105	1.964 (13)	W34–O113	2.098 (12)
W35–O50	1.726 (11)	W35–O6	1.921 (10)	W35–O12	1.930 (12)

Table S5. BVS values of selected O atoms of compound 1.

Atom	BVS	Atom	BVS	Atom	BVS	Atom	BVS
O1	1.89	O2	1.69	O3	2.03	O4	2.16
O5	1.89	O6	1.98	O7	2.01	O8	1.96
O9	2.14	O10	1.97	O11	1.99	O12	1.97
O13	2.07	O14	1.93	O15	1.93	O16	2.01
O17	2.05	O18	2.02	O19	1.93	O20	1.89
O21	1.94	O22	1.99	O23	1.81	O24	2.03
O25	2.05	O26	1.90	O27	2.20	O28	1.86
O29	1.93	O30	2.04	O31	0.32	O32	1.54
O33	1.96	O34	1.97	O35	1.92	O36	1.95
O37	2.11	O38	1.91	O39	2.14	O40	2.01
O41	1.99	O42	1.88	O43	1.97	O44	2.16
O45	2.16	O46	2.04	O47	2.01	O48	1.97
O49	1.92	O50	1.97	O51	2.07	O52	2.05
O53	2.00	O54	2.13	O55	2.07	O56	2.08
O57	0.52	O58	1.90	O59	20.4	O60	1.96
O61	1.88	O62	1.98	O63	1.85	O64	1.90
O65	1.99	O66	2.04	O67	2.01	O68	2.00
O69	1.94	O70	1.90	O71	1.99	O72	1.89
O73	2.01	O74	1.99	O75	0.50	O76	1.91
O77	1.85	O78	1.96	O79	2.07	O80	1.93
O81	1.93	O82	1.94	O83	1.94	O84	0.51
O85	1.91	O86	1.97	O87	2.15	O88	1.64
O89	2.22	O90	0.39	O91	1.78	O92	1.96
O93	1.69	O94	2.01	O95	1.86	O96	2.06
O97	2.20	O98	1.67	O99	1.76	O100	1.74
O101	1.68	O102	1.80	O103	1.75	O104	1.76
O105	1.69	O106	1.73	O107	1.79	O108	1.88
O109	1.66	O110	1.74	O111	1.76	O112	0.25
O113	1.78	O114	1.77	O115	1.72	O116	1.67
O117	1.59	O118	1.61	O119	0.61	O120	1.70
O121	0.15	O122	0.25	O123	0.17	O124	0.16

Table S6. BVS values of selected O atoms of compound **2**.

Atom	BVS	Atom	BVS	Atom	BVS	Atom	BVS
O1	1.87	O2	1.65	O3	1.94	O4	2.11
O5	1.99	O6	1.98	O7	1.85	O8	1.96
O9	2.16	O10	2.30	O11	1.89	O12	2.00
O13	1.97	O14	1.88	O15	1.92	O16	2.18
O17	1.81	O18	2.00	O19	2.04	O20	1.95
O21	2.14	O22	1.96	O23	2.18	O24	0.67
O25	2.05	O26	2.02	O27	2.13	O28	0.57
O29	1.99	O30	1.97	O31	0.72	O32	1.72
O33	1.98	O34	2.04	O35	1.93	O37	2.12
O38	1.94	O39	1.84	O40	2.02	O41	1.93
O42	1.95	O43	1.88	O44	2.04	O45	1.92
O46	2.06	O47	2.00	O48	1.88	O49	1.96
O50	2.07	O51	1.90	O52	0.40	O53	1.88
O54	2.09	O55	1.99	O56	1.96	O57	2.03
O58	2.05	O59	2.16	O60	2.21	O61	2.04
O62	2.05	O63	2.08	O64	2.04	O65	2.07
O66	2.06	O67	1.99	O68	1.89	O69	2.05
O70	1.98	O71	2.01	O72	1.96	O73	1.94
O74	1.99	O75	1.84	O76	2.11	O77	2.17
O78	0.36	O79	1.87	O80	1.90	O81	1.92
O82	2.04	O83	1.89	O84	2.21	O85	1.66
O86	1.90	O87	2.12	O88	1.94	O89	1.96
O90	2.01	O91	1.98	O92	1.88	O93	2.03
O94	2.19	O95	1.79	O96	1.76	O97	1.88
O98	1.61	O99	1.72	O100	1.81	O101	1.76
O102	1.72	O103	2.07	O104	1.74	O105	1.71
O106	1.80	O107	1.67	O108	1.71	O109	1.79
O111	1.79	O112	1.69	O113	1.75	O114	1.67
O115	2.16	O116	1.70	O117	0.60	O118	0.28
O119	1.66	O120	1.65	O121	1.64	O122	1.66

Table S7. BVS values of selected O atoms of compound **3**.

Atom	BVS	Atom	BVS	Atom	BVS	Atom	BVS
O1	1.64	O2	1.81	O3	1.73	O4	1.87
O5	1.79	O6	2.10	O7	1.90	O8	1.95
O9	2.05	O10	2.84	O11	1.85	O12	2.14
O13	2.07	O14	2.06	O15	1.65	O16	2.05
O17	1.87	O18	1.87	O19	1.41	O20	2.04
O21	1.60	O22	2.01	O23	1.66	O24	1.88
O25	1.72	O26	1.96	O27	1.84	O28	2.04
O29	0.73	O30	2.04	O31	1.82	O32	1.96
O33	1.94	O34	2.01	O35	1.81	O36	1.93
O37	1.78	O38	1.88	O39	1.98	O40	2.06
O41	1.92	O42	2.11	O43	1.70	O44	2.09
O45	1.67	O46	2.00	O47	1.58	O48	2.02
O49	1.54	O50	0.33	O51	1.99	O52	1.88
O53	1.95	O54	2.03	O55	1.89	O56	1.98
O57	1.95	O58	2.02	O59	2.00	O60	1.99
O61	1.90	O62	1.95	O63	1.99	O64	1.97
O65	1.98	O66	2.06	O67	1.93	O68	2.01
O69	1.91	O70	1.99	O71	1.96	O72	2.14
O73	1.91	O74	2.00	O75	2.12	O76	1.90
O77	1.75	O78	2.07	O79	1.98	O80	1.90
O81	1.72	O82	1.92	O83	2.05	O84	2.15
O85	1.70	O86	1.97	O87	1.78	O88	0.46
O89	0.30	O90	1.62	O91	1.91	O92	2.13
O93	1.64	O94	1.87	O95	1.95	O96	2.08
O97	0.69	O98	1.96	O99	2.06	O100	2.17
O101	1.90	O102	1.63	O103	1.73	O104	1.89
O105	1.96	O106	2.05	O107	1.95	O108	1.98
O109	2.06	O110	1.62	O111	1.99	O112	1.85
O113	2.13						

Table S8. Screening of solvents for aniline oxidation over **1** in H₂O₂ system.^a

Entry	Solvent	Conv. %	Sel. of AOB % ^b	Sel. of AB % ^b	Sel. of NSB % ^b	Sel. of NB % ^b
1	Methanol	86	63	35	2	0
2	Ethanol	83	62	31	1	5
3	Acetonitrile	82	51	43	2	4
4	Acetone	79	60	33	5	2
5	Acetic acid	76	57	42	1	0
6	Diethyl ether	79	51	44	4	1
7	Petroleum ether	86	63	36	1	0
8	N-hexane	78	52	42	5	1
9	Cyclohexane	71	56	48	1	1
10	Toluene	64	34	31	31	3
11	Dichloromethane	77	49	46	3	1
12	Mesitylene	58	26	40	31	3

^a Reaction conditions: aniline (1 mmol), catalyst (0.1 mol%), solvent (1 mL), H₂O₂ (30%, w/w, 2.0 equiv.), 30 °C, 12 h. ^b Determined by GC and naphthalene as internal standard.

Table S9. Screening of additives for aniline oxidation over **1** in H₂O₂ system.^a

Entry	Additive (equiv)	Conv. %	Sel. of AOB % ^c	Sel. of AB % ^c	Sel. of NSB % ^c	Sel. of NB % ^c
1	As ₂ O ₃	88	76	22	2	0
2	NaAsO ₂	93	73	26	1	0
3	Na ₂ HAsO ₄	95	96	3	1	0
4	K ₃ PO ₄	90	67	28	4	1
5	Na ₂ SO ₃	74	52	47	1	0
6	Na ₂ HSO ₃	82	53	43	4	0
7	Na ₂ SO ₄	88	41	58	1	0
8	Na ₂ S ₂ O ₃	74	58	37	5	0
9	Na ₂ S ₂ O ₄	76	51	45	4	0
10	Na ₂ S ₂ O ₈	99	20	79	1	0
11 ^b	Na ₂ S ₂ O ₈	99	0	12	88	0
12	K ₂ CO ₃	82	14	32	3	51
13	Na ₂ CO ₃	71	28	36	6	31
14	NaHCO ₃	71	25	46	3	27

^a Reaction conditions: aniline (1 mmol), catalyst (0.1 mol%), methanol (1 mL), H₂O₂ (30%, w/w, 2.0 equiv.), 30 °C, 12 h. ^b Toluene (1 mL). ^c Determined by GC and naphthalene as internal standard.

Table S10. Optimization of the reaction conditions for oxidation of aniline to AOB over **1** in H₂O₂ system.^a

Entr y	Catalyst (mol%)	T/ °C	Time (h)	H ₂ O ₂ (equiv)	Solvent (mL)	Additive (equiv)	Conv. %	Sel. of AOB % ^b
1	0.1	40	12	2.0	1	Na ₂ HAsO ₄ (0.075)	99	98
2	0.1	40	12	2.0	1	Na ₂ HAsO ₄ (0.05)	99	97
3	0.1	40	12	2.0	1	Na ₂ HAsO ₄ (0.025)	99	96
4	0.15	40	12	2.0	1	Na ₂ HAsO ₄ (0.05)	99	97
5	0.05	40	12	2.0	1	Na ₂ HAsO ₄ (0.05)	99	96
6	0.1	30	12	2.0	1	Na ₂ HAsO ₄ (0.05)	98	95
7	0.1	50	12	2.0	1	Na ₂ HAsO ₄ (0.05)	99	95
8	0.1	40	12	2.5	1	Na ₂ HAsO ₄ (0.05)	99	89
9	0.1	40	12	1.5	1	Na ₂ HAsO ₄ (0.05)	98	91
10	0.1	40	24	2.0	1	Na ₂ HAsO ₄ (0.05)	99	97
11	0.1	40	6	2.0	1	Na ₂ HAsO ₄ (0.05)	98	91
12	0.1	40	12	2.0	2	Na ₂ HAsO ₄ (0.05)	97	80
13	0.1	40	12	2.0	0.5	Na ₂ HAsO ₄ (0.05)	99	74

^a Reaction conditions: aniline (1 mmol), solvent (methanol). ^b Determined by GC and naphthalene as internal standard.

Table S11. Optimization of the reaction conditions for oxidation of aniline to AB over **1** in H₂O₂ system.^a

Entry	Catalyst (mol%)	T/ °C	Time (h)	H ₂ O ₂ (equiv)	Solvent (mL)	Additive (equiv)	Conv. %	Sel. of AB % ^b
1	0.1	40	24	1.6	3	Na ₂ S ₂ O ₈ (0.075)	99	98
2	0.1	40	24	1.6	3	Na ₂ S ₂ O ₈ (0.05)	99	97
3	0.1	40	24	1.6	3	Na ₂ S ₂ O ₈ (0.025)	95	96
4	0.15	40	24	1.6	3	Na ₂ S ₂ O ₈ (0.05)	99	97
5	0.05	40	24	1.6	3	Na ₂ S ₂ O ₈ (0.05)	90	94
6	0.1	30	24	1.6	3	Na ₂ S ₂ O ₈ (0.05)	79	87
7	0.1	50	24	1.6	3	Na ₂ S ₂ O ₈ (0.05)	98	98
8	0.1	40	24	1.8	3	Na ₂ S ₂ O ₈ (0.05)	99	91
9	0.1	40	24	1.4	3	Na ₂ S ₂ O ₈ (0.05)	84	98
10	0.1	40	36	1.6	3	Na ₂ S ₂ O ₈ (0.05)	99	99
11	0.1	40	18	1.6	3	Na ₂ S ₂ O ₈ (0.05)	88	93
12	0.1	40	24	1.6	4	Na ₂ S ₂ O ₈ (0.05)	84	98
13	0.1	40	24	1.6	2	Na ₂ S ₂ O ₈ (0.05)	99	82

^a Reaction conditions: aniline (1 mmol), solvent (methanol). ^b Determined by GC and naphthalene as internal standard.

Table S12. Optimization of the reaction conditions for oxidation of aniline to NSB over **1** in H₂O₂ system.^a

Entry	Catalyst (mol%)	T/ °C	Time (h)	H ₂ O ₂ (equiv)	Solvent (mL)	Additive (equiv)	Conv. %	Sel. of NSB % ^b
1	0.025	30	6	1.4	1	Na ₂ S ₂ O ₈ (0.075)	99	87
2	0.025	30	6	1.4	1	Na ₂ S ₂ O ₈ (0.05)	99	88
3	0.025	30	6	1.4	1	Na ₂ S ₂ O ₈ (0.025)	91	89
4	0.03	30	6	1.4	1	Na ₂ S ₂ O ₈ (0.05)	99	91
5	0.02	30	6	1.4	1	Na ₂ S ₂ O ₈ (0.05)	89	86
6	0.025	25	6	1.4	1	Na ₂ S ₂ O ₈ (0.05)	81	89
7	0.025	40	6	1.4	1	Na ₂ S ₂ O ₈ (0.05)	99	73
8	0.025	30	6	1.6	1	Na ₂ S ₂ O ₈ (0.05)	99	73
9	0.025	30	6	1.2	1	Na ₂ S ₂ O ₈ (0.05)	72	89
10	0.025	30	8	1.4	1	Na ₂ S ₂ O ₈ (0.05)	99	63
11	0.025	30	4	1.4	1	Na ₂ S ₂ O ₈ (0.05)	83	70
12	0.025	30	6	1.4	2	Na ₂ S ₂ O ₈ (0.05)	67	51
13	0.025	30	6	1.4	0.5	Na ₂ S ₂ O ₈ (0.05)	98	43

^a Reaction conditions: aniline (1 mmol), solvent (toluene). ^b Determined by GC and naphthalene as internal standard.**Table S13.** Optimization of the reaction conditions for oxidation of aniline to NB over **1** in H₂O₂ system.^a

Entry	Catalyst (mol%)	T/ °C	Time (h)	H ₂ O ₂ (equiv)	Solvent (mL)	Additive (equiv)	Conv. %	Sel. of NB % ^b
1	0.05	40	2	5.0	1	K ₂ CO ₃ (0.075)	99	99
2	0.05	40	2	5.0	1	K ₂ CO ₃ (0.05)	99	99
3	0.05	40	2	5.0	1	K ₂ CO ₃ (0.025)	92	93
4	0.06	40	2	5.0	1	K ₂ CO ₃ (0.05)	99	99
5	0.03	40	2	5.0	1	K ₂ CO ₃ (0.05)	91	96
6	0.05	30	2	5.0	1	K ₂ CO ₃ (0.05)	74	89
7	0.05	50	2	5.0	1	K ₂ CO ₃ (0.05)	99	99
8	0.05	40	2	6.0	1	K ₂ CO ₃ (0.05)	99	97
9	0.05	40	2	4.0	1	K ₂ CO ₃ (0.05)	92	93
10	0.05	40	3	5.0	1	K ₂ CO ₃ (0.05)	99	99
11	0.05	40	1	5.0	1	K ₂ CO ₃ (0.05)	86	94
12	0.05	40	2	5.0	2	K ₂ CO ₃ (0.05)	84	89
13	0.05	40	2	5.0	0.5	K ₂ CO ₃ (0.05)	99	83

^a Reaction conditions: aniline (1 mmol), solvent (methanol). ^b Determined by GC and naphthalene as internal standard.**Table S14.** The catalytic results of anilines oxidation reaction in Methanol-Na₂HAsO₄ system using different catalysts.^a

Eentry	Catalyst	Conv. %	Sel of AOB % ^b	Sel. of AB % ^b	Sel. of NSB % ^b	Sel. of NB % ^b
1	none	0	0	0	0	0
2	Na ₂ WO ₄	18	58	42	0	0
3	Sb ₂ O ₃	0	0	0	0	0
4	RuCl ₃	0	0	0	0	0
5	1	99	96	2	2	0
6	1-HT	92	95	1	1	0
7	2	87	91	9	0	0
8	3	49	85	15	0	0
9	3-HT	50	84	16	0	0

^a Catalyst (0.1 mol%), **2a** (1.0 mmol), methanol (1.0 mL), 30% H₂O₂ (2.0 equiv), Na₂S₂O₈ (0.05 equiv), 40 °C, 12 h. ^b Determined by GC and naphthalene as internal standard.

Table S15. The catalytic results of anilines oxidation reaction in Methanol-Na₂S₂O₈ system using different catalysts.^a

Entry	Catalyst	Conv. %	Sel. of AOB % ^b	Sel. of AB % ^b	Sel. of NSB % ^b	Sel. of NB % ^b
1	none	0	0	0	0	0
2	Na ₂ WO ₄	29	23	77	0	0
3	Sb ₂ O ₃	0	0	0	0	0
4	RuCl ₃	0	0	0	0	0
5	1	99	0	98	2	0
6	1-HT	92	3	95	2	0
7	2	90	7	93	0	0
8	3	56	6	94	2	0
9	3-HT	57	18	82	0	0

^a Catalyst (0.1 mol%), **2a** (1.0 mmol), methanol (3.0 mL), 30% H₂O₂ (1.5 equiv), Na₂S₂O₈ (0.05 equiv), 40 °C, 24 h. ^b Determined by GC and naphthalene as internal standard.

Table S16. The catalytic results of anilines oxidation reaction in Toluene-Na₂S₂O₈ system using different catalysts.^a

Entry	Catalyst	Conv. %	Sel. of AOB % ^b	Sel. of AB % ^b	Sel. of NSB % ^b	Sel. of NB % ^b
1	none	0	0	0	0	0
2	Na ₂ WO ₄	24	11	38	51	0
3	Sb ₂ O ₃	0	0	0	0	0
4	RuCl ₃	0	0	0	0	0
5	1	99	0	12	88	0
6	1-HT	93	12	13	75	0
7	2	94	9	9	82	0
8	3	44	32	14	54	0
9	3-HT	44	30	11	59	0

^a Catalyst (0.025 mol%), **2a** (1.0 mmol), Toluene (1.0 mL), 30% H₂O₂ (1.5 equiv), Na₂S₂O₈ (0.05 equiv), 30 °C, 1 h. ^b Determined by GC and naphthalene as internal standard.

Table S17. The catalytic results of anilines oxidation reaction in Methanol-K₂CO₃ system using different catalysts.^a

Entry	Catalyst	Conv. %	Sel. of AOB % ^b	Sel. of AB % ^b	Sel. of NSB % ^b	Sel. of NB % ^b
1	none	0	0	0	0	0
2	Na ₂ WO ₄	26	6	42	0	53
3	Sb ₂ O ₃	0	0	0	0	0
4	RuCl ₃	0	0	0	0	0
5	1	99	0	0	0	99
6	1-HT	92	0	0	0	99
7	2	92	0	0	0	99
8	3	36	16	33	4	47
9	3-HT	37	19	31	4	46

^a Catalyst (0.05 mol%), **2a** (1.0 mmol), methanol (1.0 mL), 30% H₂O₂ (5.0 equiv), K₂CO₃ (0.05 equiv), 40 °C, 2 h. ^b Determined by GC and naphthalene as internal standard.

Table S18. Experiments on investigating the reaction mechanism over **1** in H₂O₂ system.

Entry	Substrate	Solvent/mL	H ₂ O ₂	Additive	Conv%	Sel. of	Sel. of	Sel. of	Sel. of
			Equiv.			AOB % ^a	AB % ^a	NSB % ^a	NB % ^a
1 ^b	Ph-NH ₂ +BHT	CH ₃ OH/1 mL	2.0	NaHAsO ₄	99	96	0	4	0
2 ^c	Ph-NH ₂ +BHT	CH ₃ OH/3 mL	1.6	Na ₂ S ₂ O ₈	99	1	97	2	0
3 ^d	Ph-NH ₂ +BHT	Toulene/1 mL	1.4	Na ₂ S ₂ O ₈	95	16	3	81	0
4 ^e	Ph-NH ₂ +BHT	CH ₃ OH/1 mL	5.0	K ₂ CO ₃	99	2	0	0	98
5 ^f	Ph-NHOH	CH ₃ OH/1 mL	2.0	no	99	98	0	2	0
6 ^g	Ph-NHOH	CH ₃ OH/3 mL	1.6	Na ₂ S ₂ O ₈	86	0	0	99	0
7 ^h	Ph-NHOH	CH ₃ OH/3 mL	1.6	no	43	90	2	8	0
8 ⁱ	Ph-NHOH	Toulene/1 mL	1.4	Na ₂ S ₂ O ₈	99	18	0	82	0
9 ^j	Ph-NHOH	Toulene/1 mL	1.4	no	46	16	1	83	0
10 ^k	Ph-NHOH	CH ₃ OH/1 mL	5.0	K ₂ CO ₃	99	2	0	1	97
11 ^l	Ph-NHOH+Ph-NO	CH ₃ OH/1 mL	0	no	99	99	0	1	0
12 ^m	Ph-NHOH+Ph-NO	CH ₃ OH/3 mL	0	Na ₂ S ₂ O ₈	36	3	1	96	0
13 ⁿ	Ph-NHOH+Ph-NO	CH ₃ OH/3 mL	0	no	99	99	0	1	0
14 ^o	Ph-NHOH+Ph-NO	Toulene/1 mL	0	Na ₂ S ₂ O ₈	21	20	2	78	0
15 ^p	Ph-NHOH+Ph-NO	CH ₃ OH/1 mL	0	K ₂ CO ₃	25	80	3	17	0
16 ^q	Ph-NH ₂ +Ph-NO	CH ₃ OH/1 mL	0	NaHAsO ₄	33	0	78	22	0
17 ^r	Ph-NH ₂ +Ph-NO	CH ₃ OH/3 mL	0	Na ₂ S ₂ O ₈	99	1	96	2	0
18 ^s	Ph-NH ₂ +Ph-NO	Toulene/1 mL	0	Na ₂ S ₂ O ₈	15	0	77	23	0
19 ^t	Ph-NH ₂ +Ph-NO	CH ₃ OH/1 mL	0	K ₂ CO ₃	22	0	89	11	0
20 ^u	Ph-NH ₂ +Ph-NO	CH ₃ OH/1 mL	0	no	97	0	91	9	0
21 ^v	Ph-NO	CH ₃ OH/1 mL	2.0	K ₂ CO ₃	99	0	0	0	99
22 ^w	Ph-NO	CH ₃ OH/1 mL	2.0	no	99	0	0	0	99

^a Determined by GC and naphthalene as internal standard. ^b Reaction conditions: Ph-NH₂ (1 mmol), BHT (1.2 mmol), catalyst (0.1 mol% based on Ph-NH₂), additive (0.05 equiv.), 40°C, 12 h. ^c Reaction conditions: Ph-NH₂ (1 mmol), BHT (1.2 mmol), catalyst (0.1 mol% based on Ph-NH₂), additive (0.05 equiv.), 40°C, 24 h. ^d Reaction conditions: Ph-NH₂ (1 mmol), BHT (1.2 mmol), catalyst (0.025 mol% based on Ph-NH₂), additive (0.05 equiv.), 30°C, 1 h. ^e Reaction conditions: Ph-NH₂ (1 mmol), BHT (1.2 mmol), catalyst (0.05 mol% based on Ph-NH₂), additive (0.05 equiv.), 40°C, 2 h. ^f Reaction conditions: Ph-NHOH (1 mmol), catalyst (0.1 mol% based on Ph-NHOH), 40°C, 30 min. ^g Reaction conditions: Ph-NHOH (1 mmol), catalyst (0.1 mol% based on Ph-NHOH), additive (0.5 equiv.), 40°C, 24 h. ^h Reaction conditions: Ph-NHOH (1 mmol), catalyst (0.1 mol% based on Ph-NHOH), 40°C, 24 h. ⁱ Reaction conditions: Ph-NHOH (1 mmol), catalyst (0.025 mol% based on Ph-NHOH), additive (0.05 equiv.), 30°C, 1 h. ^j Reaction conditions: Ph-NHOH (1 mmol), catalyst (0.025 mol% based on Ph-NHOH), 30°C, 1 h. ^k Reaction conditions: Ph-NHOH (1 mmol), catalyst (0.05 mol% based on Ph-NHOH), additive (0.05 equiv.), 40°C, 2 h. ^l Reaction conditions: Ph-NHOH (0.5 mmol), Ph-NO (0.5 mmol), catalyst (0.1 mol% based on Ph-NHOH+Ph-NO), 40°C, 30 min. ^m Reaction conditions: Ph-NHOH (0.5 mmol), Ph-NO (0.5 mmol), catalyst (0.1 mol% based on Ph-NHOH+Ph-NO), additive (0.05 equiv.), 40°C, 24 h. ⁿ Reaction conditions: Ph-NHOH (0.5 mmol), Ph-NO (0.5 mmol), catalyst (0.1 mol% based on Ph-NHOH+Ph-NO), 40°C, 30 min. ^o Reaction conditions: Ph-NHOH (0.5 mmol), Ph-NO (0.5 mmol), catalyst (0.025 mol% based on Ph-NHOH+Ph-NO), 30°C, 1 h. ^p Reaction conditions: Ph-NHOH (0.5 mmol), Ph-NO (0.5 mmol), additive (0.05 equiv.), catalyst (0.05 mol% based on Ph-NHOH+Ph-NO), 40°C, 2 h. ^q Reaction conditions: Ph-NH₂ (0.5 mmol), Ph-NO (0.5 mmol), additive (0.05 equiv.), catalyst (0.1 mol% based on Ph-NH₂+Ph-NO), 40°C, 12 h. ^r Reaction conditions: Ph-NH₂ (0.5 mmol), Ph-NO (0.5 mmol), additive (0.05 equiv.), catalyst (0.1 mol% based on Ph-NH₂+Ph-NO), 40°C, 30 min. ^s Reaction conditions: aniline (0.5 mmol), Ph-NO (0.5 mmol), additive (0.05 equiv.), catalyst (0.025 mol% based on Ph-NH₂+Ph-NO), 30°C, 1 h. ^t Reaction conditions: Ph-NH₂ (0.5 mmol), Ph-NO (0.5 mmol), additive (0.05 equiv.), catalyst (0.05 mol% based on Ph-NH₂+Ph-NO), 40°C, 2 h. ^u Reaction conditions: Ph-NH₂ (0.5 mmol), Ph-NO (0.5 mmol), catalyst (0.05 mol% based on Ph-NH₂+Ph-NO), 40°C, 2 h. ^v Reaction conditions: Ph-NO (1 mmol), additive (0.05 equiv.), catalyst (0.05 mol% based on Ph-NO), 40°C, 30 min. ^w Reaction conditions: Ph-NO (1 mmol), (0.05 mol% based on Ph-NO), 40°C, 30min.

Table S19. Photocatalytic activity with different catalysts and control experiment.^a

Entry	Catalyst	t(h)	Conv. % ^c	Sel. % ^c
1	1	4	99	99
2	2	6	99	99
3	3	6	43	99
4	Na ₂ WO ₄	6	24	99
5	Sb ₂ O ₃	6	24	99
6	RuCl ₃	6	31	99
7	RuCl ₃ / 3	6	40	99
8	1 ^b	6	trace	trace
9	None	6	trace	trace

^a Reaction conditions: substrate (0.5 mmol), catalyst (0.05 mol%), methanol (0.5 mL), N₂H₄·H₂O (2 mmol), 10 W LED lamp, rt. ^b Dark.^c Determined by GC and naphthalene as internal standard.**Table S20.** Photocatalytic activity of photocatalyst **1** with different solvents.^a

Entry	Solvent	Conv. % ^b	Sel. % ^b
1	CH ₃ CN	45	98
2	CH ₃ CH ₂ OH	56	99
3	CH ₂ Cl	57	99
4	DMF	58	99
5	Toluene	46	99
6	Hexane	75	99
7	CH ₃ OH	99	99

^a Reaction conditions: substrate (0.5 mmol), catalyst (0.05 mol%), solvent (0.5 mL), N₂H₄·H₂O (2 mmol), 10 W LED lamp, rt. ^b Determined by GC and naphthalene as internal standard.**Table S21.** Photocatalytic activity of photocatalyst **1** with different substrates for reaction pathway study.^a

Entry	Substrate	t(h)	Conv. % ^c	Sel. % ^e
1	Nitrobenzene	4	99	99
2	Nitrosobenzene	1	99	99
3	Hydroxylamine	0.5	99	99
4	Azobenzene	4	0	0
5	Azoxybenzene	4	25	0
6	Nitrobenzene+CCl ₄ ^b	4	11	99
7	Nitrobenzene ^c	4	0	0
8	Nitrobenzene+BHT ^d	4	99	99

^a Reaction conditions: substrate (0.5 mmol), catalyst (0.05 mol%), methanol (0.5 mL), N₂H₄·H₂O (2 mmol), 10 W LED lamp, rt. ^b CCl₄ (0.5 mmol).^c H₂ (2 atm) replace N₂H₄·H₂O. ^d BHT (0.6 mmol). ^e Determined by GC and naphthalene as internal standard.

Table S22. The summary of aniline oxidation reaction.

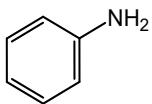
Catalyst	Solvent	Oxidizing agent	t	T	Conversion (%)	Yield				Ref
						AB	AOB	NSB	NB	
Material catalysts										
Au/TiO ₂	toluene	O ₂ (5 bars)	22 h	100 °C	100	98	0.5	1.5	--	5
Ti superoxide	MeOH	H ₂ O ₂	0.5 h	25 °C	100	--	--	1.82	98.18	6
CuBr	toluene	air	20 h	60 °C	n.p.	96	--	--	--	7
Meso-Mn ₂ O ₃	toluene	air	8 h	110 °C	99	99	--	--	--	8
Na ₁₂ [$\text{L}-\text{Ln}_4$]	C ₂ H ₄ Cl ₂	H ₂ O ₂	9 h	RT	100	--	100	--	--	9
Au/CeO ₂	toluene	O ₂ (5 bars)	10 h	100 °C	100	93	--	--	--	10
RuO ₂ /Cu ₂ O NPS	MeCN	air	16 h	85 °C	n.p.	94	--	--	--	11
Ag/C-KOH	DMSO	air (1 atm)	24 h	60 °C	n.p.	97	--	--	--	12
Ag/WO ₃	CH ₃ CN	H ₂ O ₂	24 h	25 °C	87	--	79	--	--	13
Nb ₂ O ₅ /FeOOH	propanol	H ₂ O ₂	24 h	25 °C	100	--	70	--	--	14
Zr(OH) ₄	methanol	H ₂ O ₂	2 h	25 °C	98	--	96	3	--	15
Zr(OH) ₄	acetic acid	TBHP	24 h	40 °C	96	90	3	1	--	15
Zr(OH) ₄	mesitylene	H ₂ O ₂	4 h	25 °C	96	0	12	84	--	15
WO _x MCN _x	CH ₃ CN	H ₂ O ₂	12 h	90 °C	89.7	71.3	13	6.3	--	16
CuCr ₂ O ₄	1,4-dioxane	H ₂ O ₂	10 h	70 °C	78	--	71.8	--	--	17
AuPd@C	DMSO	O ₂	12 h	25 °C	55	54	--	--	--	18
CuCl	CH ₃ CN	air	48 h	25 °C	n.p.	33	--	--	--	19
Ag _{1-x} Ni _x	DMSO	air	24 h	60 °C	94.4	94	--	--	--	20
POM catalysts										
Mo ₆	methanol	H ₂ O ₂	24 h	60 °C	99	99	--	--	--	21
Mo ₆	MTBE	H ₂ O ₂	24 h	50 °C	99	--	93	--	--	21
Ru-POM	CH ₃ OH	H ₂ O ₂	48 h	25 °C	n.p.	--	94	--	--	22
Rh-POM	CH ₃ OH	H ₂ O ₂	48 h	60 °C	n.p.	--	95	--	--	23
1	methanol	H ₂ O ₂	12	40 °C	99	--	97	--	--	This work
1	methanol	H ₂ O ₂	24	40 °C	99	98	--	--	--	This work
1	toluene	H ₂ O ₂	1	60 °C	99	--	--	88	--	This work
1	methanol	H ₂ O ₂	2	60 °C	99	--	--	--	99	This work

Note: n.p = not provided

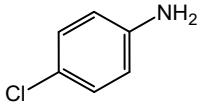
Table S23. Summary of the yield of the reduction of nitrobenzene with different catalysts.

Catalyst	T/°C	Light source	Hydrogen source	t/h	Yield/%	Ref.
Material catalysts						
Pd-Au NRs	25	λ> 320nm	HCOOH/HCOONa	5	>99	24
PbBiO ₂ Br	RT	440 nm	TEOA	6	>99	25
Cu-CoFe ₂ O ₄	RT	Visible light	H ₂ O	1.5	99	26
Ag-rGO/g-C ₃ N ₄	RT	λ> 400 nm	methanol	4	98	27
EY	RT	LED (525 nm)	TEOA	24	>99	28
3.0 wt% CQDs/ZnIn ₂ S ₄	/	LED irradiation	TEOA/MeOH (1:1)	16	>99	29
Co@CN/SiO ₂ -500	70	/	N ₂ H ₄ •H ₂ O	1.2	>99.5	30
Pd/C	80	/	N ₂ H ₄ •H ₂ O	4	100	31
MMC-Fe ₂ O ₃	80	/	N ₂ H ₄ •H ₂ O	2	100	32
Rh-Fe ₂ O ₃	80	/	N ₂ H ₄ •H ₂ O	1	99	33
NC-90	90	/	N ₂ H ₄ •H ₂ O	1	100	34
Zr ₁₂ -TPDC-Co	110	/	NaBEt ₃ H/H ₂	72	100	35
POM catalysts						
ZnW-TPT	RT	LED irradiation	N ₂ H ₄ •H ₂ O	6	99	36
Rh-POM	RT	10 W LED lamp	N ₂ H ₄ •H ₂ O	6	96	23
CoW-TPT	RT	365 nm	TEOA	12	95	37
1	RT	Visible light	N ₂ H ₄ •H ₂ O	6	97	This work

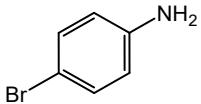
NMR data of products



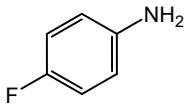
Aniline (2a) ^1H NMR (500 MHz, CDCl_3) δ 7.06 (t, $J=8.36$ Hz, 2H), 6.67 (t, $J=7.36$ Hz, 1H), 6.55–6.52 (dd, $J=1.04$, 0.92 Hz, 2H), 3.47 (s, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 146.73, 129.48, 118.60, 115.29 ppm.



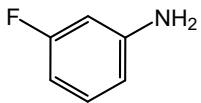
4-chloroaniline (2b) ^1H NMR (500 MHz, CDCl_3) δ 7.12 (d, $J=8.72$ Hz, 2H), 6.60 (d, $J=8.72$ Hz, 2H), 3.64 (s, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 145.13, 129.15, 123.02, 116.31 ppm.



4-bromoaniline (2c) ^1H NMR (500 MHz, CDCl_3) δ 7.29 (d, $J=8.55$ Hz, 2H), 6.62 (d, $J=8.70$ Hz, 2H), 3.64 (s, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 145.46, 132.04, 116.74, 110.21 ppm.



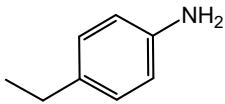
4-fluoroaniline (2d) ^1H NMR (500 MHz, CDCl_3) δ 6.93 (t, $J=8.73$ Hz, 2H), 6.67 (dd, $J=4.5$, 4.5 Hz, 2H), 3.60 (s, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 157.96, 154.85, 142.71, 115.84 ppm.



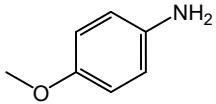
3-fluoroaniline (2e) ^1H NMR (500 MHz, CDCl_3) δ 7.14 (t, $J=7.98$ Hz, 1H), 6.83 (d, $J=7.98$ Hz, 1H), 6.71 (t, $J=2.07$ Hz, 1H), 6.61 (dd, $J=2.16$, 1.5 Hz, 1H), 3.76 (s, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 147.89, 134.83, 130.49, 118.43, 114.99, 113.38 ppm.



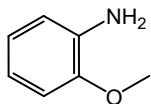
2-fluoroaniline (2f) ^1H NMR (500 MHz, CDCl_3) δ 7.13–7.01 (m, 2H), 6.87–6.76 (m, 2H), 3.79 (s, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 153.43, 150.27, 134.78, 134.61, 124.60, 118.73, 117.11, 115.43, 115.18 ppm.



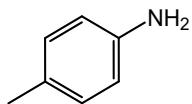
4-ethylaniline (2g) ^1H NMR (500 MHz, CDCl_3) δ 7.19 (d, $J=8.1$ Hz, 2H), 6.79 (d, $J=8.31$ Hz, 2H), 3.65 (s, 2H), 2.78 (dd, $J=7.56$, 7.59 Hz, 2H), 1.42 (t, $J=7.56$ Hz, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 144.37, 134.45, 128.75, 115.45, 28.19, 16.17 ppm.



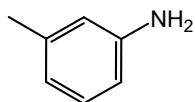
4-methoxyaniline (2h) ^1H NMR (500 MHz, CDCl_3) δ 6.82 (d, $J=8.85$ Hz, 2H), 6.71 (d, $J=8.85$ Hz, 2H), 3.79 (s, 3H), 3.42 (s, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 152.83, 140.02, 116.48, 114.86, 55.78 ppm.



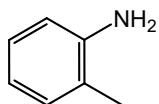
2-methoxyaniline (2i) ^1H NMR (500 MHz, CDCl_3) δ 6.94–6.82 (m, 4H), 3.95 (s, 3H), 3.89 (s, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 147.45, 136.44, 121.25, 118.49, 115.14, 110.63, 55.52 ppm.



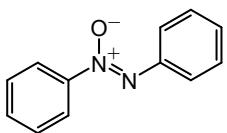
p-toluidine (2j) ^1H NMR (500 MHz, CDCl_3) δ 7.13 (d, $J=8.07$ Hz, 2H), 6.74 (d, $J=8.28$ Hz, 2H), 3.61 (s, 2H), 2.40 (s, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 144.08, 129.89, 127.76, 115.40, 20.60 ppm.



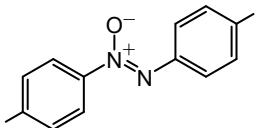
m-toluidine (2k) ^1H NMR (500 MHz, CDCl_3) δ 7.22 (t, $J=7.71$ Hz, 1H), 6.78 (d, $J=7.59$ Hz, 1H), 6.65 (d, $J=6.42$ Hz, 2H), 3.70 (s, 2H), 2.44 (s, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 146.66, 139.21, 129.32, 119.52, 116.09, 112.43, 21.60 ppm.



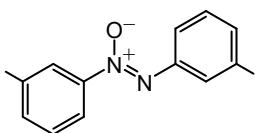
o-toluidine (2l) ^1H NMR (500 MHz, CDCl_3) δ 7.26 (t, $J=7.32$ Hz, 2H), 6.94 (t, $J=7.35$ Hz, 1H), 6.85 (d, $J=7.86$ Hz, 1H), 3.71 (s, 2H), 2.35 (s, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 144.90, 130.65, 127.18, 122.49, 118.75, 115.14, 17.52 ppm.



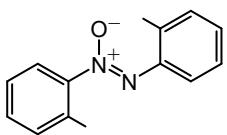
(Z)-Diphenyldiazene N-oxide (3a) ^1H NMR (500 MHz, CDCl_3) δ 8.34 (d, $J=8.7$ Hz, 2H), 8.22 (d, $J=8.3$ Hz, 2H), 7.56–7.41 (m, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 148.26, 143.94, 131.50, 129.54, 128.71, 125.47, 122.27 ppm.



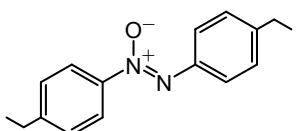
(Z)-1,2-Di-p-tolyldiazene 1-oxide (3b). ^1H NMR (500 MHz, CDCl_3) δ 8.20 (d, $J = 8.25$ Hz, 2H), 8.14 (d, $J = 7.1$ Hz, 2H), 7.30 (d, $J = 8.4$ Hz, 4H), 2.44 (d, $J = 11.0$ Hz, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 146.21, 141.87, 139.95, 129.25, 125.61, 122.10, 21.50, 21.24 ppm.



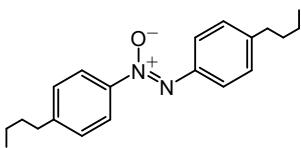
(Z)-1,2-Di-m-tolyldiazene 1-oxide (3c). ^1H NMR (500 MHz, CDCl_3) δ 8.10 (t, $J = 15.2$ Hz, 2H), 8.00 (d, $J=8.35$ Hz, 2H), 7.39 (t, $J = 19.85$ Hz, 3H), 7.23 (d, $J = 7.1$ Hz, 1H), 2.48 (s, 3H), 2.44 (s, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 148.40, 144.03, 138.95, 138.42, 132.25, 130.33, 128.56, 128.47, 125.98, 122.76, 122.48, 119.48, 21.45, 21.37 ppm.



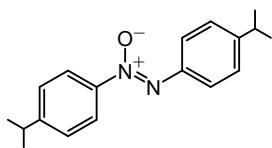
(Z)-1,2-Bis(2-methylphenyl)diazene 1-oxide (3d). ^1H NMR (500 MHz, CDCl_3) δ 8.10 (d, $J = 7.75$ Hz, 1H), 7.94 (d, $J = 8.65$ Hz, 1H), 7.70 (d, $J = 7.75$ Hz, 1H), 7.38–7.30 (m, 5H), 2.53 (s, 3H), 2.39 (s, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 142.79, 134.10, 131.77, 130.77, 130.00, 128.56, 126.58, 126.04, 125.66, 123.58, 121.53, 18.42, 17.22 ppm.



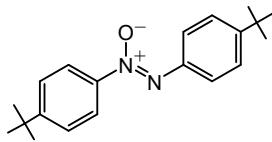
(Z)-1,2-bis(4-ethylphenyl)diazene 1-oxide (3e) ^1H NMR (500 MHz, CDCl_3) δ 8.22 (d, $J = 8.7$ Hz, 2H), 8.15 (d, $J = 8.5$ Hz, 2H), 7.32 (d, $J = 8.2$ Hz, 4H), 2.71–2.75 (m, 4H), 1.30–1.27 (m, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 148.14, 146.21, 142.05, 128.06, 125.71, 122.25, 28.61, 15.28 ppm.



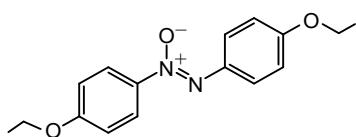
(Z)-1,2-bis(4-butylphenyl)diazene 1-oxide (3f) ^1H NMR (500 MHz, CDCl_3) δ 8.21 (d, $J = 8.5$ Hz, 2H), 8.15 (d, $J = 12.6$ Hz, 2H), 7.30 (d, $J = 8.6$ Hz, 4H), 2.72–2.66 (m, 4H), 1.65 (t, $J = 5.9$ Hz, 4H), 1.39 (d, $J = 7.6$ Hz, 4H), 0.95 (t, $J = 7.4$ Hz, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 146.84, 144.94, 142.01, 128.60, 125.62, 122.15, 35.59, 33.30, 22.23, 13.89 ppm.



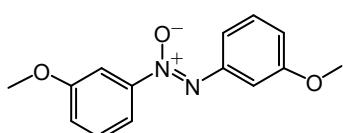
(Z)-1,2-bis(4-isopropylphenyl)diazene 1-oxide (3g) ^1H NMR (500 MHz, CDCl_3) δ 8.22–8.20 (d, $J = 8.75$ Hz, 2H), 8.15–8.14 (d, $J = 8.6$ Hz, 2H), 7.34 (m, 4H), 3.01 (m, 2H), 1.29 (m, 12H). ^{13}C NMR (126 MHz, CDCl_3) δ 151.73, 149.77, 145.43, 141.13, 125.71, 124.71, 121.28, 32.98, 22.78 ppm.



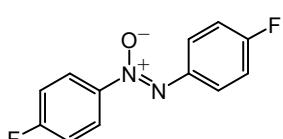
(Z)-1,2-bis(4-tert-butylphenyl)diazene 1-oxide (3h) ^1H NMR (500 MHz, CDCl_3) δ 8.23 (d, $J = 8.95$ Hz, 2H), 8.16 (d, $J = 8.85$ Hz, 2H), 7.52–7.50 (m, 4H), 1.38 (d, $J = 2.8$ Hz, 18H). ^{13}C NMR (126 MHz, CDCl_3) δ 155.00, 152.96, 146.06, 141.75, 125.53, 121.95, 34.94, 31.17 ppm.



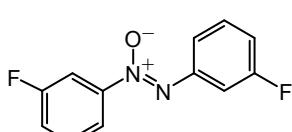
(Z)-1,2-bis(3-methoxyphenyl)diazene 1-oxide (3i) ^1H NMR (500 MHz, CDCl_3) δ 8.28 (dd, $J = 9.25, 9.2$ Hz, 4H), 6.92 (dd, $J = 9.2, 9.2$ Hz, 4H), 4.08 (dd, $J = 4.25, 4.25$ Hz, 4H), 1.44 (dd, $J = 7.0, 7.0$ Hz, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 161.18, 159.55, 141.48, 137.82, 127.75, 123.66, 114.13, 63.63, 14.70 ppm.



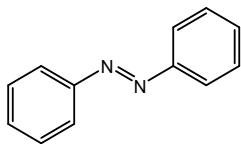
(Z)-1,2-bis(3-methoxyphenyl)diazene 1-oxide (3j) ^1H NMR (500 MHz, CDCl_3) δ 7.90 (d, $J = 8.15$ Hz, 1H), 7.84 (t, $J = 2.3$ Hz, 1H), 7.80 (t, $J = 2.05$ Hz, 1H), 7.74 (d, $J = 7.95$ Hz, 1H), 7.42 (dd, $J = 8.2, 8.15$ Hz, 2H), 7.12 (d, $J = 11.35$ Hz, 1H), 6.98 (d, $J = 10.85$ Hz, 1H), 3.91 (d, $J = 16.75$ Hz, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 159.95, 149.56, 145.05, 129.49, 118.40, 116.32, 114.69, 110.06, 107.55, 55.45 ppm.



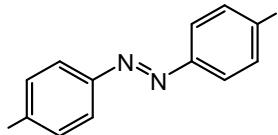
(Z)-1,2-bis(4-fluorophenyl)diazene 1-oxide (3k) ^1H NMR (500 MHz, CDCl_3) δ 8.30 (dd, $J = 4.8, 4.8$ Hz, 2H), 7.26 (dd, $J = 5.35, 5.35$ Hz, 2H), 7.20–7.15 (m, 4H). ^{13}C NMR (126 MHz, CDCl_3) δ 162.48, 143.24, 139.25, 127.00, 123.55, 114.78, 114.55 ppm.



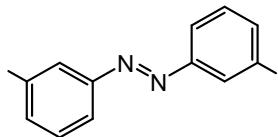
(Z)-1,2-bis(3-fluorophenyl)diazene 1-oxide (3l) ^1H NMR (500 MHz, CDCl_3) δ 8.13 (dd, $J = 1.55, 1.55$ Hz, 1H), 8.08–8.03 (m, 2H), 7.86 (d, $J = 8.15$ Hz, 1H), 7.50–7.45 (m, 2H), 7.31–7.28 (m, 1H), 7.15–7.11 (m, 1H). ^{13}C NMR (126 MHz, CDCl_3) δ 163.42, 161.40, 149.31, 144.83, 130.07, 129.72, 122.10, 119.08, 118.13, 116.76, 112.14, 110.20 ppm.



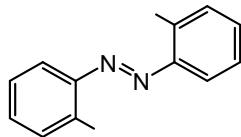
(E)-1,2-Diphenyldiazene (4a) ^1H NMR (500 MHz, CDCl_3) δ 8.05 (d, $J = 8.35$ Hz, 4H), 7.60–7.53 (m, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 152.79, 131.10, 129.20, 123.01 ppm.



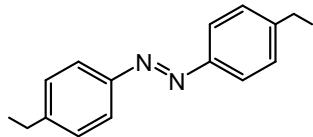
(E)-1,2-Di-p-tolyldiazene (4b) ^1H NMR (500 MHz, CDCl_3) δ 7.82 (d, $J = 8.3$ Hz, 4H), 7.31 (d, $J = 8.05$ Hz, 4H), 2.43 (s, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 150.83, 141.19, 129.69, 122.71, 21.47 ppm.



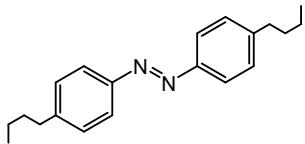
(E)-1,2-Di-m-tolyldiazene (4c) ^1H NMR (500 MHz, CDCl_3) δ 7.73 (s, 4H), 7.41 (t, $J = 7.7$ Hz, 2H), 7.30 (d, $J = 7.5$ Hz, 2H), 2.47 (s, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 152.83, 138.98, 131.70, 128.91, 122.88, 120.48, 21.38 ppm.



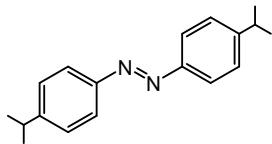
(E)-1,2-Di-o-tolyldiazene (4d) ^1H NMR (500 MHz, CDCl_3) δ 7.64 (d, $J = 7.9$ Hz, 2H), 7.37 (d, $J = 7.35$ Hz, 4H), 7.29 (d, $J = 11.7$ Hz, 2H), 2.75 (s, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 151.10, 137.99, 131.25, 130.66, 126.35, 115.84, 17.60 ppm.



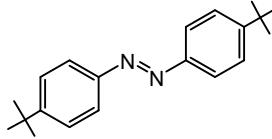
(E)-1,2-bis(4-ethylphenyl)diazene (4e) ^1H NMR (500 MHz, CDCl_3) δ 7.92 (d, $J = 8.2$ Hz, 4H), 7.39 (d, $J = 8.1$ Hz, 4H), 2.78 (d, $J = 7.65$ Hz, 4H), 1.33 (t, $J = 7.6$ Hz, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 151.14, 147.50, 128.57, 122.91, 28.89, 15.50 ppm.



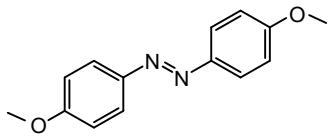
(E)-1,2-bis(4-butylphenyl)diazene (4f) ^1H NMR (500 MHz, CDCl_3) δ 7.83 (d, $J = 8.35$ Hz, 4H), 7.32 (d, $J = 8.4$ Hz, 4H), 2.69 (t, $J = 7.65$ Hz, 4H), 1.68–1.62 (m, 4H), 1.41–1.36 (m, 4H), 0.95 (t, $J = 7.35$ Hz, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 151.04, 146.17, 129.05, 122.69, 35.56, 33.45, 22.33, 13.93 ppm.



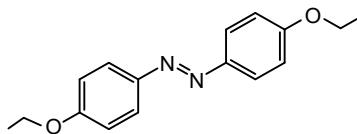
(E)-1,2-bis(4-isopropylphenyl)diazene (4g) ^1H NMR (500 MHz, CDCl_3) δ 7.86 (d, $J = 8.4$ Hz, 4H), 7.38 (d, $J = 8.3$ Hz, 4H), 1.31 (d, $J = 6.95$ Hz, 12H). ^{13}C NMR (126 MHz, CDCl_3) δ 151.99, 151.17, 127.05, 122.78, 34.12, 23.86 ppm.



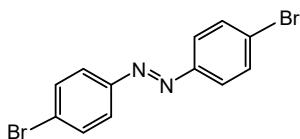
(E)-1,2-bis(4-tert-butylphenyl)diazene (4h) ^1H NMR (500 MHz, CDCl_3) δ 7.85 (d, $J = 8.65$ Hz, 4H), 7.54 (d, $J = 8.65$ Hz, 4H), 1.38 (s, 18H). ^{13}C NMR (126 MHz, CDCl_3) δ 154.22, 150.76, 125.96, 122.43, 34.97, 31.28 ppm.



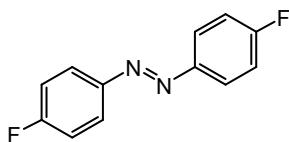
(E)-1,2-bis(4-methoxyphenyl)diazene (4i) ^1H NMR (500 MHz, CDCl_3) δ 7.89 (d, $J = 8.95$ Hz, 4H), 7.01 (d, $J = 8.95$ Hz, 4H), 3.89 (s, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 161.57, 147.11, 124.35, 114.17, 55.57 ppm.



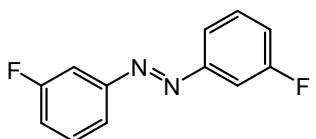
(E)-1,2-bis(4-ethoxyphenyl)diazene (4j) ^1H NMR (500 MHz, CDCl_3) δ 7.87 (d, $J = 9$ Hz, 4H), 7.00 (d, $J = 9$ Hz, 4H), 4.13 (dd, $J = 7, 6.95$ Hz, 4H), 1.45 (t, $J = 7$ Hz, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 160.97, 147.00, 124.32, 114.65, 63.78, 14.80 ppm.



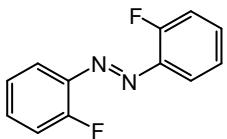
(E)-1,2-bis(4-bromophenyl)diazene (4k) ^1H NMR (500 MHz, CDCl_3) δ 7.80 (d, $J = 8.7$ Hz, 4H), 7.66 (d, $J = 8.75$ Hz, 4H). ^{13}C NMR (126 MHz, CDCl_3) δ 151.15, 132.40, 124.41 ppm.



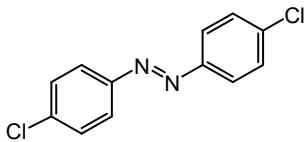
(E)-1,2-bis(4-fluorophenyl)diazene (4l) ^1H NMR (500 MHz, CDCl_3) δ 7.94–7.91 (m, 4H), 7.21–7.18 (m, 4H). ^{13}C NMR (126 MHz, CDCl_3) δ 165.36, 148.96, 124.75, 116.08 ppm.



(E)-1,2-bis(3-fluorophenyl)diazene (4m) ^1H NMR (500 MHz, CDCl_3) δ 7.78 (dt, $J = 1.7, 1$ Hz, 2H), 7.62 (dt, $J = 4.35, 4.35$ Hz, 2H), 7.53 (dt, $J = 8.1, 8.05$ Hz, 2H), 7.21–7.19 (m, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 162.29, 153.76, 130.30, 120.79, 118.29, 108.23 ppm.



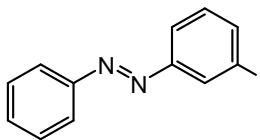
(E)-1,2-bis(2-fluorophenyl)phenyl (4n) ^1H NMR (500 MHz, CDCl_3) δ 7.85 (t, $J = 6.15$ Hz, 2H), 7.55–7.50 (m, 2H), 7.34–7.28 (m, 4H). ^{13}C NMR (126 MHz, CDCl_3) δ 161.39, 159.34, 140.83, 133.02, 124.34, 117.90 ppm.



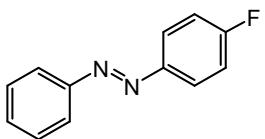
(E)-1,2-bis(4-chlorophenyl)diazene (4o) ^1H NMR (500 MHz, CDCl_3) δ 7.87 (d, $J = 8.7$ Hz, 4H), 7.50 (d, $J = 8.7$ Hz, 4H). ^{13}C NMR (126 MHz, CDCl_3) δ 150.80, 137.23, 129.40, 124.18 ppm.



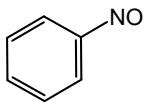
(E)-1-(4-methylphenyl)-2-phenyldiazene (4p) ^1H NMR (500 MHz, CDCl_3) δ 7.93 (t, $J = 11.25$ Hz, 2H), 7.84 (t, $J = 11.15$ Hz, 2H), 7.54–7.51 (m, 3H), 7.33 (t, $J = 6$ Hz, 2H), 2.46 (d, $J = 3.7$ Hz, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 152.73, 150.83, 141.16, 130.66, 129.68, 129.02, 122.84, 122.70, 21.45 ppm.



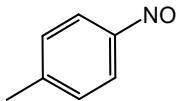
(E)-1-(3-methylphenyl)-2-phenyldiazene (4q) ^1H NMR (500 MHz, CDCl_3) δ 7.93 (t, $J = 5.85$ Hz, 2H), 7.73 (d, $J = 5.55$ Hz, 2H), 7.53–7.51 (m, 2H), 7.48–7.40 (m, 2H), 2.47 (s, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 152.74, 139.00, 131.69, 130.99, 129.10, 122.86, 120.47, 21.38 ppm.



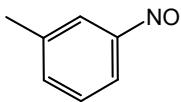
(E)-1-(4-fluorophenyl)-2-phenyldiazene (4r) ^1H NMR (500 MHz, CDCl_3) δ 7.97–7.91 (m, 4H), 7.53–7.48 (m, 3H), 7.22–7.19 (m, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 164.34, 151.45, 148.12, 130.00, 128.07, 123.79, 121.78, 114.91 ppm.



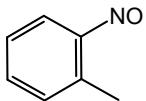
Nitrosobenzene (5a) ^1H NMR (500 MHz, CDCl_3) δ 7.81 (d, $J = 7.9$ Hz, 2H), 7.61 (t, $J = 7.3$ Hz, 1H), 7.52 (d, $J = 2.9$ Hz, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 164.77, 134.54, 128.21, 119.76 ppm.



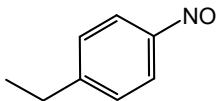
4-MethylNitrosobenzene (5b) ^1H NMR (500 MHz, CDCl_3) δ 8.20 (dd, $J = 8.6, 8.45$ Hz, 2H), 7.29 (d, $J = 8.25$ Hz, 2H), 2.44 (d, $J = 13.25$ Hz, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 165.56, 146.23, 129.28, 122.13, 21.52 ppm.



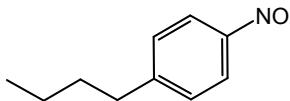
3-MethylNitrosobenzene (5c) ^1H NMR (500 MHz, CDCl_3) δ 8.11–7.96 (m, 1H), 7.78 (d, $J = 7.25$ Hz, 1H), 7.64–7.52 (m, 1H), 7.50–7.36 (m, 1H), 2.50 (d, $J = 17.5$ Hz, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 166.25, 139.44, 136.21, 129.08, 120.66, 119.01, 21.14 ppm.



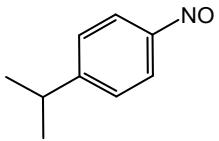
2-MethylNitrosobenzene (5d) ^1H NMR (500 MHz, CDCl_3) δ 7.59 (d, $J = 8.2$ Hz, 1H), 7.55 (d, $J = 7.6$ Hz, 1H), 7.15 (t, $J = 7.95$ Hz, 1H), 6.29 (d, $J = 8.15$ Hz, 1H), 3.35 (s, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 163.97, 141.28, 135.22, 131.91, 124.64, 106.31, 16.19 ppm.



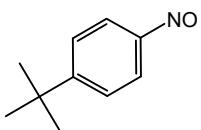
4-ethylphenylnitrosobenzene (5e) ^1H NMR (500 MHz, CDCl_3) δ 7.84 (d, $J = 8$ Hz, 2H), 7.42 (d, $J = 8.05$ Hz, 2H), 2.76 (dd, $J = 7.6, 7.55$ Hz, 2H), 1.29 (t, $J = 7.65$ Hz, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 165.69, 153.14, 128.53, 121.35, 29.16, 14.84 ppm.



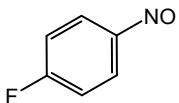
4-butylphenylnitrosobenzene (5f) ^1H NMR (500 MHz, CDCl_3) δ 7.83 (d, $J = 8.2$ Hz, 2H), 7.40 (d, $J = 8.5$ Hz, 2H), 2.70 (t, $J = 7.7$ Hz, 2H), 1.65 (t, $J = 7.75$ Hz, 2H), 1.38 (t, $J = 7.45$ Hz, 2H), 0.96 (d, $J = 7.35$ Hz, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 165.70, 152.02, 129.08, 121.31, 35.92, 32.95, 22.31, 13.84 ppm.



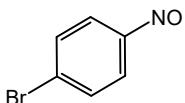
4-isopropylphenylnitrosobenzene (5g) ^1H NMR (500 MHz, CDCl_3) δ 7.85 (d, $J = 8.25$ Hz, 2H), 7.46 (d, $J = 8.35$ Hz, 2H), 3.01 (t, $J = 6.9$ Hz, 1H), 1.31 (d, $J = 6.85$ Hz, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 165.78, 157.60, 127.16, 121.43, 34.56, 23.44 ppm.



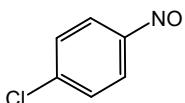
4-tert-butylphenylnitrosobenzene (5h) ^1H NMR (500 MHz, CDCl_3) δ 7.85 (d, $J = 8.5$ Hz, 2H), 7.63 (d, $J = 8.8$ Hz, 2H), 1.38 (s, 9H). ^{13}C NMR (126 MHz, CDCl_3) δ 164.21, 158.76, 125.05, 119.99, 34.60, 29.94 ppm.



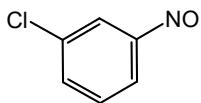
4-fluorophenylnitrosobenzene (5i) ^1H NMR (500 MHz, CDCl_3) δ 8.02 (dd, $J = 5.25, 5.2$ Hz, 2H), 7.33 (t, $J = 8$ Hz, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 166.89, 162.59, 122.97, 115.25 ppm.



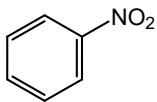
1-Bromo-4-nitrosobenzene (5j) ^1H NMR (500 MHz, CDCl_3) δ 7.78 (d, $J = 1.9$ Hz, 4H). ^{13}C NMR (126 MHz, CDCl_3) δ 163.82, 132.69, 131.69, 122.12 ppm.



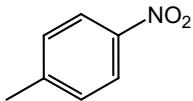
4-chlorophenylnitrosobenzene (5k) ^1H NMR (500 MHz, CDCl_3) δ 7.86 (d, $J = 8.65$ Hz, 2H), 7.60 (d, $J = 8.8$ Hz, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 163.72, 142.26, 129.66, 122.15 ppm.



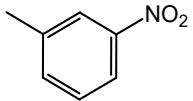
3-chlorophenylnitrosobenzene (5l) ^1H NMR (500 MHz, CDCl_3) δ 8.07 (d, $J = 7.75$ Hz, 1H), 7.70 (d, $J = 7.4$ Hz, 1H), 7.65 (t, $J = 7.9, 7.75$ Hz, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 165.03, 136.02, 134.96, 130.72, 121.44, 118.72 ppm.



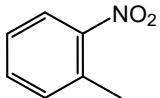
Nitrobenzene (6a) ^1H NMR (500 MHz, CDCl_3) δ 8.06 (d, $J = 8.8$ Hz, 2H), 7.59 (t, $J = 7.55$ Hz, 1H), 7.42 (t, $J = 9.1$ Hz, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 147.91, 134.57, 129.21, 123.09 ppm.



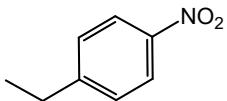
4-methylNitrobenzene (6b) ^1H NMR (500 MHz, CDCl_3) δ 7.93 (d, $J = 8.85$ Hz, 2H), 7.18 (d, $J = 8.8$ Hz, 2H), 2.31 (s, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 145.04, 144.96, 128.71, 122.24, 20.27 ppm.



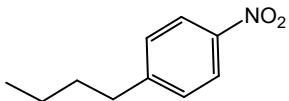
3-methylNitrobenzene (6c) ^1H NMR (500 MHz, CDCl_3) δ 7.96 (d, $J = 7.85$ Hz, 2H), 7.46 (d, $J = 7.65$ Hz, 1H), 7.38 (d, $J = 8$ Hz, 1H), 2.42 (d, $J = 6.75$ Hz, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 147.15, 138.75, 134.30, 128.00, 122.69, 119.52, 20.08 ppm.



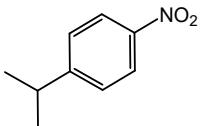
2-methylNitrobenzene (6d) ^1H NMR (500 MHz, CDCl_3) δ 7.96 (d, $J = 7.25$ Hz, 1H), 7.49 (d, $J = 8.75$ Hz, 1H), 7.33 (t, $J = 4.95$ Hz, 2H), 2.60 (s, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 148.28, 131.95, 131.71, 125.85, 123.59, 19.34 ppm.



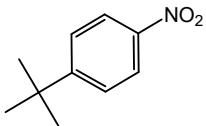
4-ethylphenylnitrobenzene (6e) ^1H NMR (500 MHz, CDCl_3) δ 8.08 (d, $J = 8.8$ Hz, 2H), 7.29 (d, $J = 8.85$ Hz, 2H), 2.71 (d, $J = 7.7$ Hz, 2H), 1.25 (d, $J = 7.65$ Hz, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 151.02, 145.16, 127.58, 122.49, 27.77, 13.92 ppm.



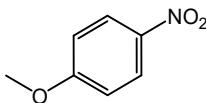
4-butylphenylnitrobenzene (6f) ^1H NMR (500 MHz, CDCl_3) δ 8.21 (dd, $J = 8.55, 8.6$ Hz, 2H), 7.30 (t, $J = 8.05$ Hz, 2H), 2.71 (d, $J = 7.55$ Hz, 2H), 1.64 (d, $J = 8.1$ Hz, 2H), 1.37 (t, $J = 7.45$ Hz, 2H), 0.95 (t, $J = 7.35$ Hz, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 149.75, 145.85, 128.09, 122.48, 34.50, 32.03, 21.22, 12.79 ppm.



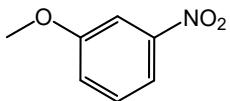
4-isopropylphenylnitrobenzene (6g) ^1H NMR (500 MHz, CDCl_3) δ 7.92 (d, $J = 8.85$ Hz, 2H), 7.17 (d, $J = 8.9$ Hz, 2H), 2.81 (t, $J = 6.95$ Hz, 1H), 1.09 (t, $J = 5.35$ Hz, 6H). ^{13}C NMR (126 MHz, CDCl_3) δ 155.50, 145.13, 126.18, 122.46, 33.13, 22.37 ppm.



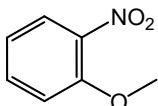
4-tert-butylphenylnitrobenzene (6h) ^1H NMR (500 MHz, CDCl_3) δ 7.92 (d, $J = 8.85$ Hz, 2H), 7.34 (d, $J = 8.9$ Hz, 2H), 1.19 (d, $J = 11.7$ Hz, 9H). ^{13}C NMR (126 MHz, CDCl_3) δ 157.71, 144.75, 125.12, 122.11, 34.17, 29.82 ppm.



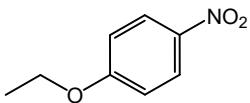
4-methoxyphenylnitrobenzene (6i) ^1H NMR (500 MHz, CDCl_3) δ 8.16 (d, $J = 9.3$ Hz, 2H), 6.93 (d, $J = 9.3$ Hz, 2H), 3.87 (s, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 163.61, 140.46, 124.83, 112.99, 54.93 ppm.



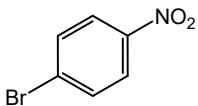
3-methoxyphenylnitrobenzene (6j) ^1H NMR (500 MHz, CDCl_3) δ 7.82 (d, $J = 8.1$ Hz, 1H), 7.72 (s, 1H), 7.42 (t, $J = 8.25$ Hz, 1H), 7.23 (d, $J = 8.3$ Hz, 1H), 3.89 (s, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 160.12, 149.23, 129.88, 121.24, 115.71, 108.10, 55.79 ppm.



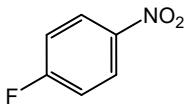
2-methoxyphenylnitrobenzene (6k) ^1H NMR (500 MHz, CDCl_3) δ 7.67 (d, $J = 8.1$ Hz, 1H), 7.39 (t, $J = 8.65$ Hz, 1H), 6.97 (d, $J = 8.5$ Hz, 1H), 6.87 (t, $J = 7.2$ Hz, 1H), 3.79 (s, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 152.86, 139.59, 134.34, 125.44, 120.23, 113.59, 56.40 ppm.



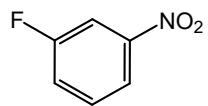
4-ethoxyphenylnitrobenzene (6l) ^1H NMR (500 MHz, CDCl_3) δ 8.13 (d, $J = 9.25$ Hz, 2H), 6.90 (d, $J = 9.25$ Hz, 2H), 4.10 (dd, $J = 7, 7$ Hz, 2H), 1.41 (t, $J = 7.05$ Hz, 3H). ^{13}C NMR (126 MHz, CDCl_3) δ 163.07, 140.24, 124.81, 113.36, 63.43, 13.49 ppm.



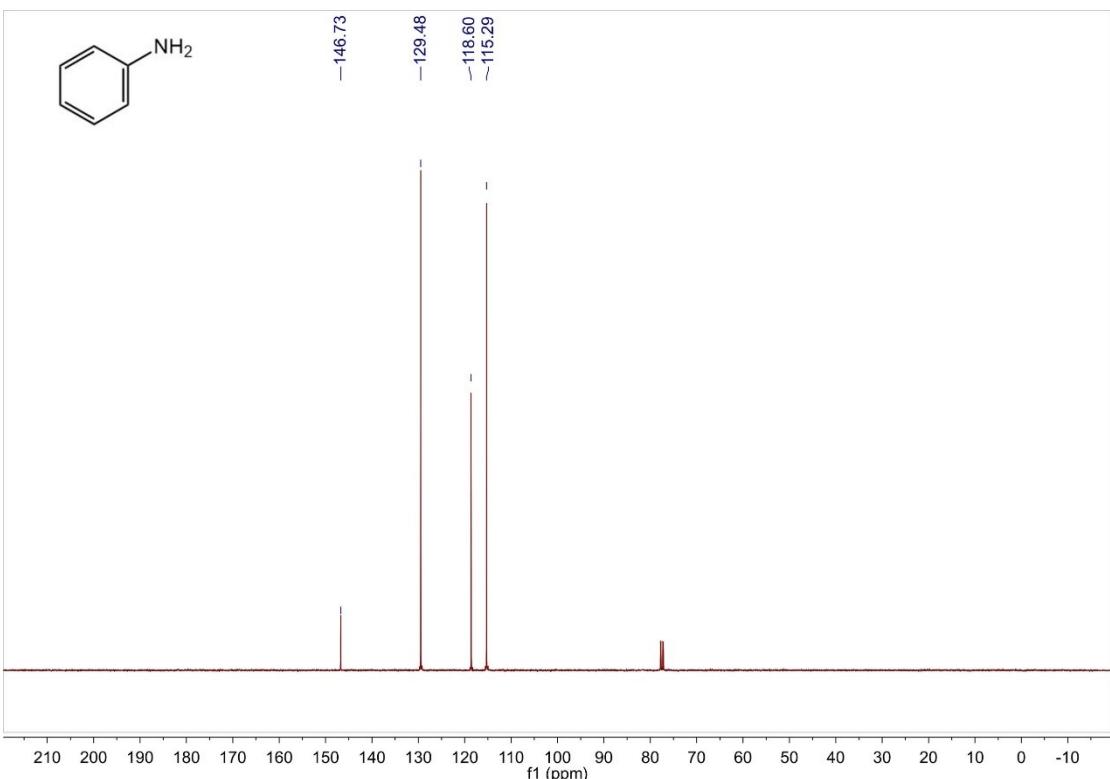
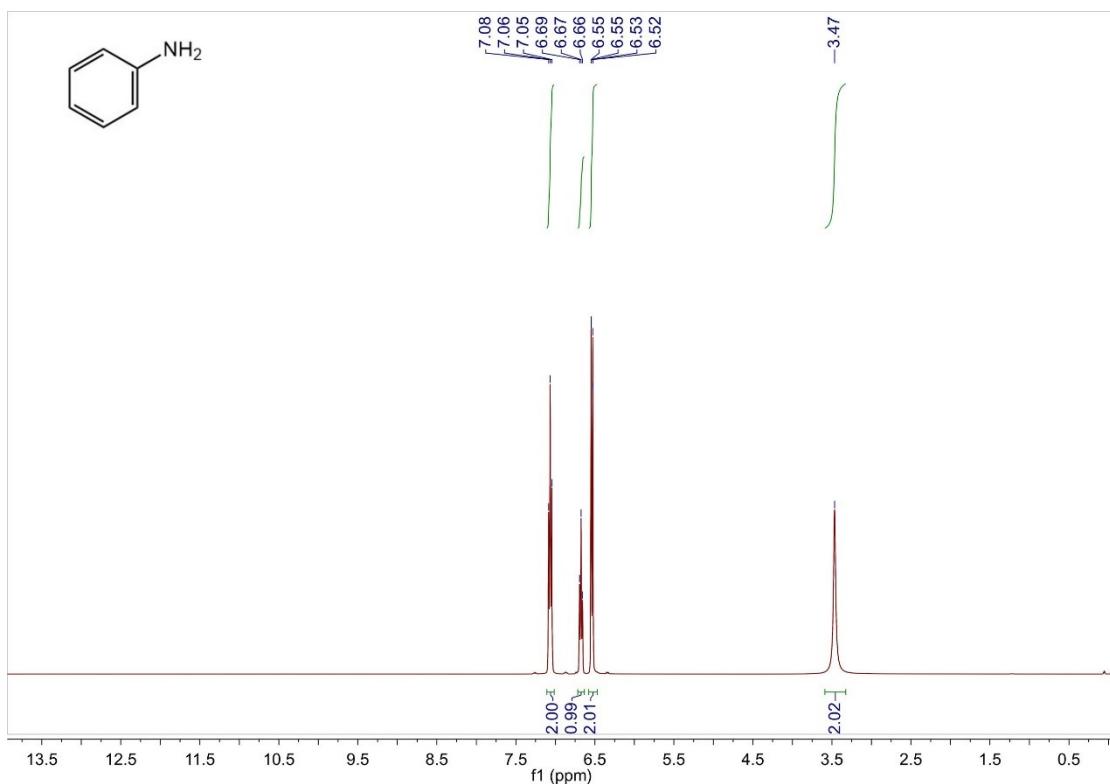
1-Bromo-4-nitrobenzene (6m) ^1H NMR (500 MHz, CDCl_3) δ 8.11 (d, $J = 11.25$ Hz, 2H), 7.70 (d, $J = 11.25$ Hz, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 147.03, 132.64, 130.00, 125.02 ppm.



4-Fluoronitrobenzene (6n) ^1H NMR (500 MHz, CDCl_3) δ 8.19 (d, $J = 11.25$ Hz, 2H), 7.52 (d, $J = 11.25$ Hz, 2H). ^{13}C NMR (126 MHz, CDCl_3) δ 146.52, 141.39, 129.60, 124.95 ppm.



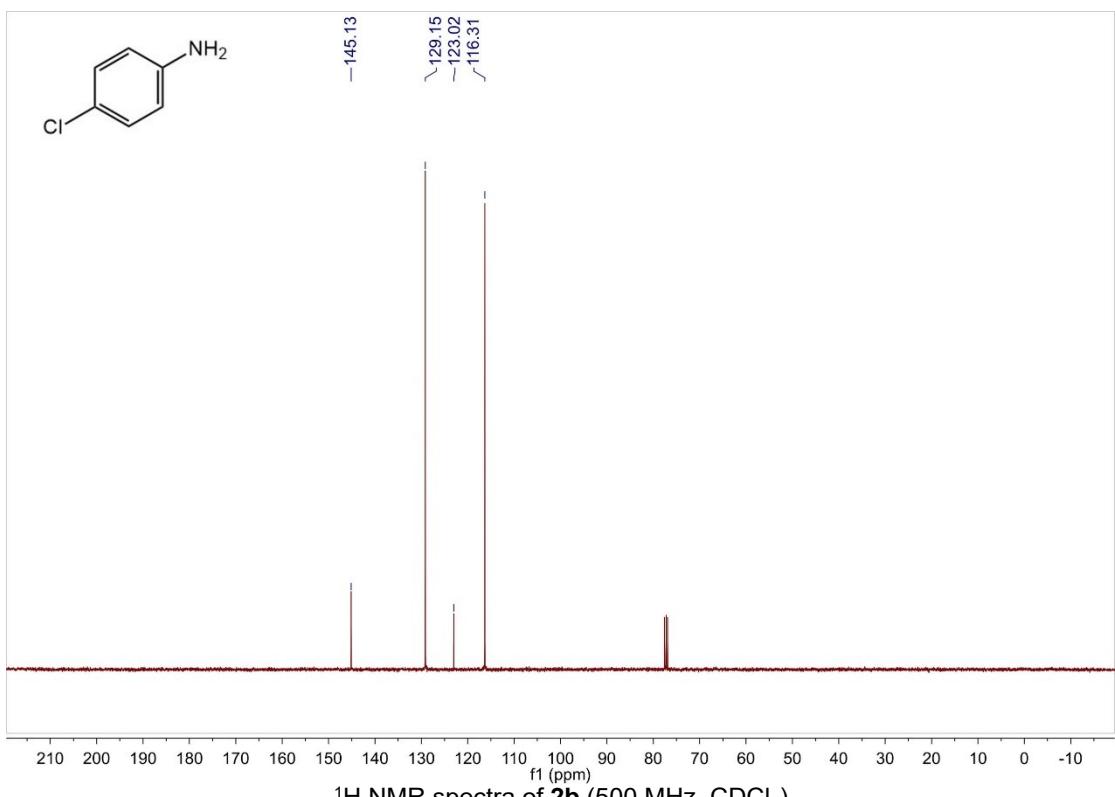
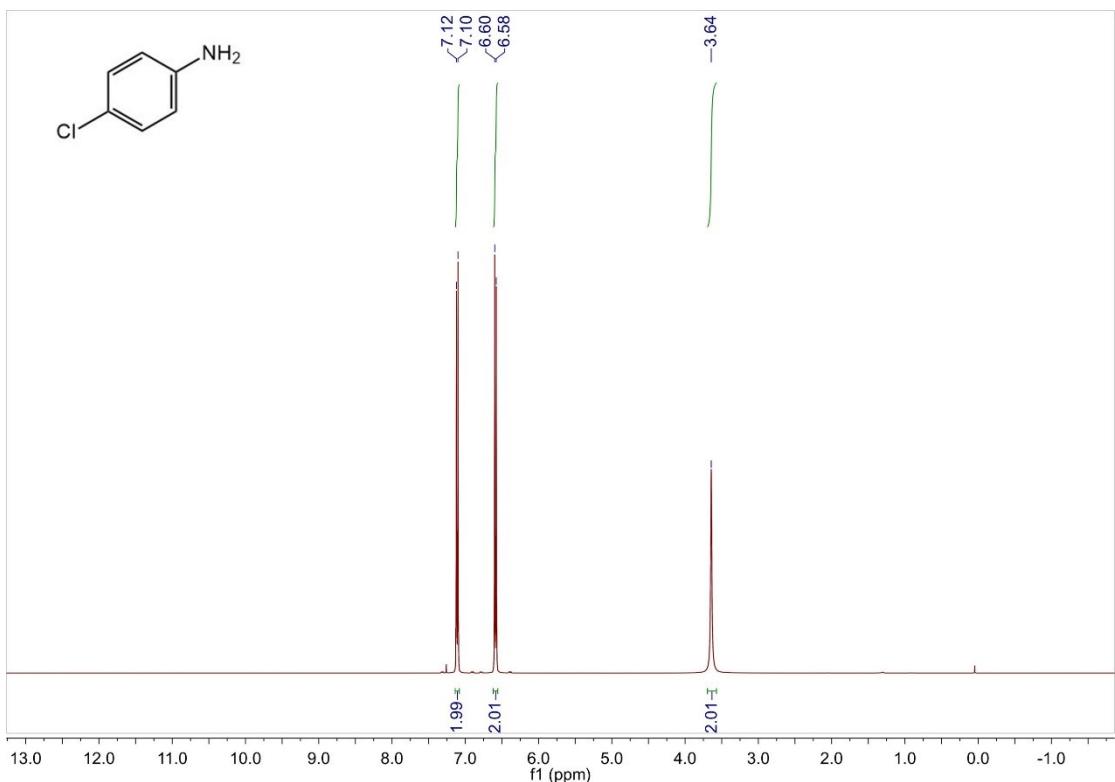
3-Fluoronitrobenzene (6o) ^1H NMR (500 MHz, CDCl_3) δ 8.02 (d, $J = 10.25$ Hz, 1H), 7.89–7.86 (m, 1H), 7.57–7.52 (m, 1H), 7.43–7.39 (m, 1H). ^{13}C NMR (126 MHz, CDCl_3) δ 163.58, 149.15, 130.83, 121.79, 119.28, 111.33 ppm.



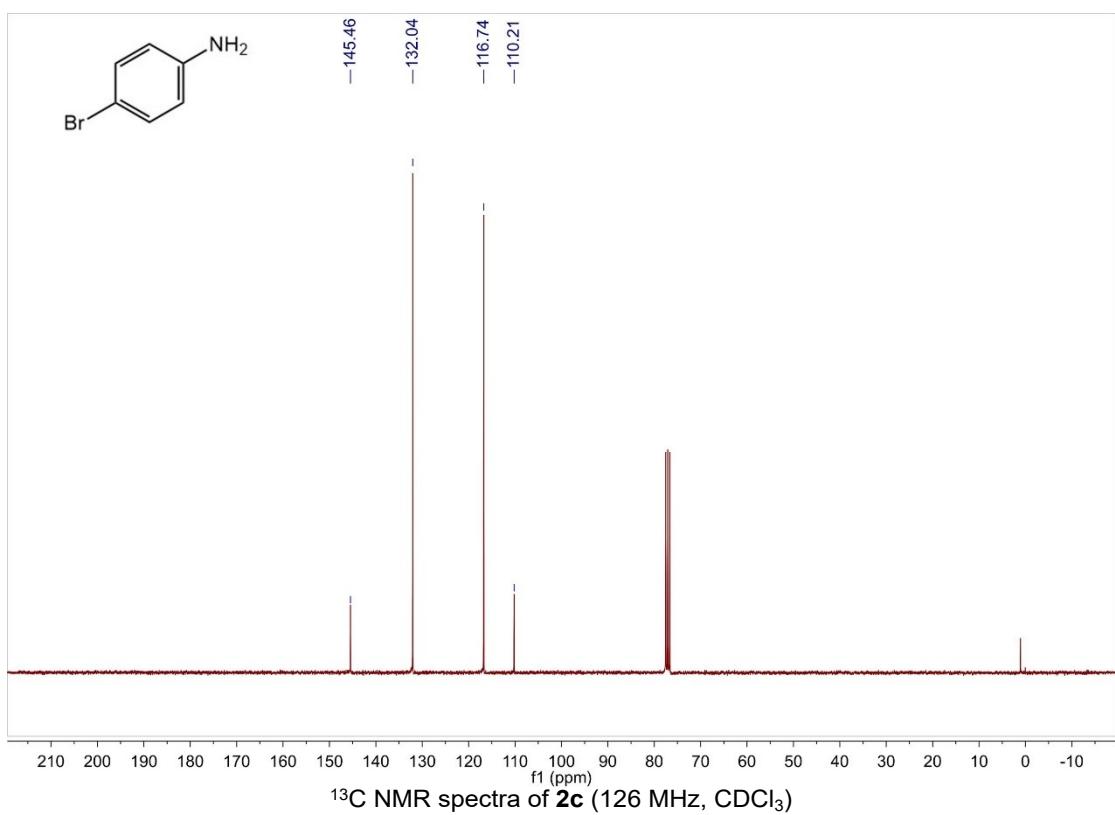
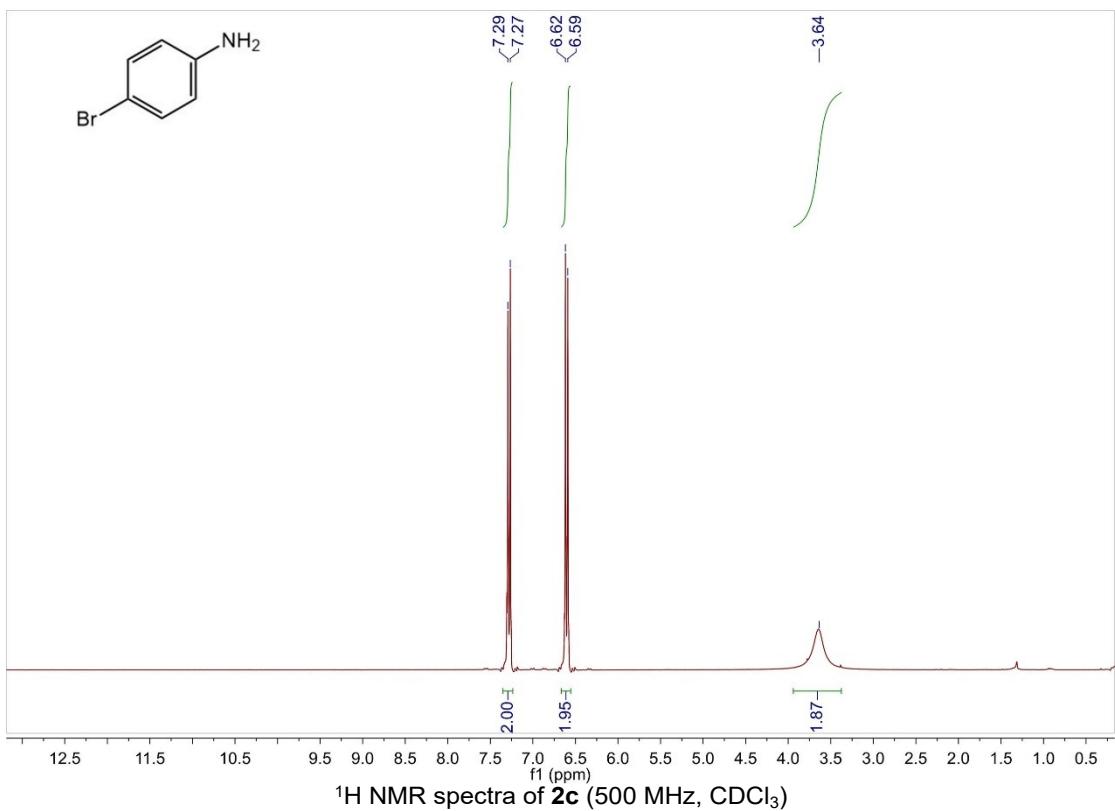
¹H NMR and ¹³C NMR spectra of spectra

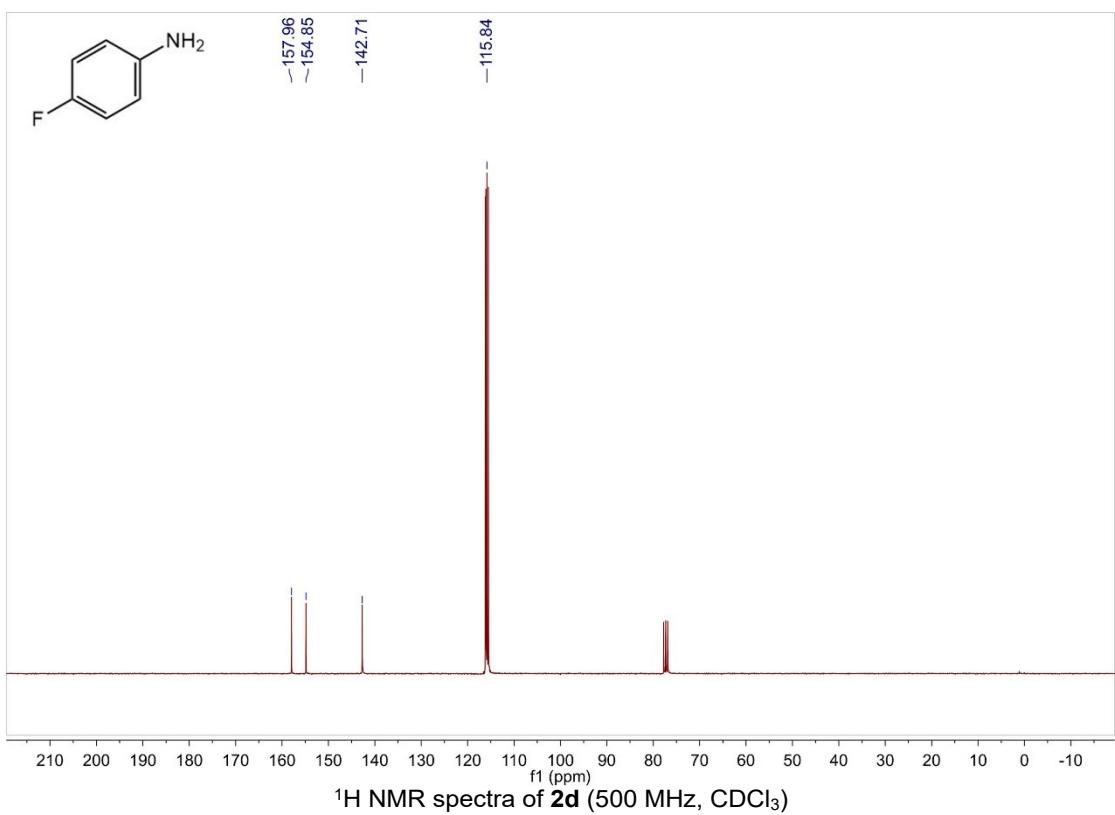
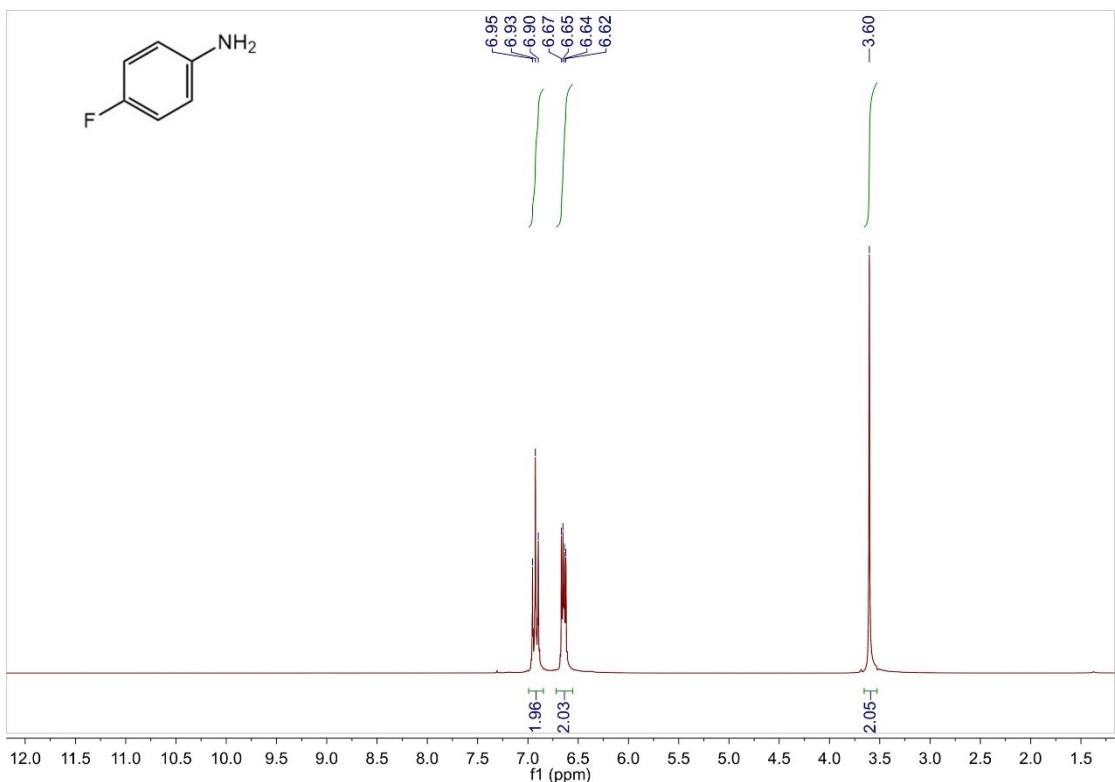
¹H NMR spectra of **2a** (500 MHz, CDCl₃)

¹³C NMR spectra of **2a** (126 MHz, CDCl₃)



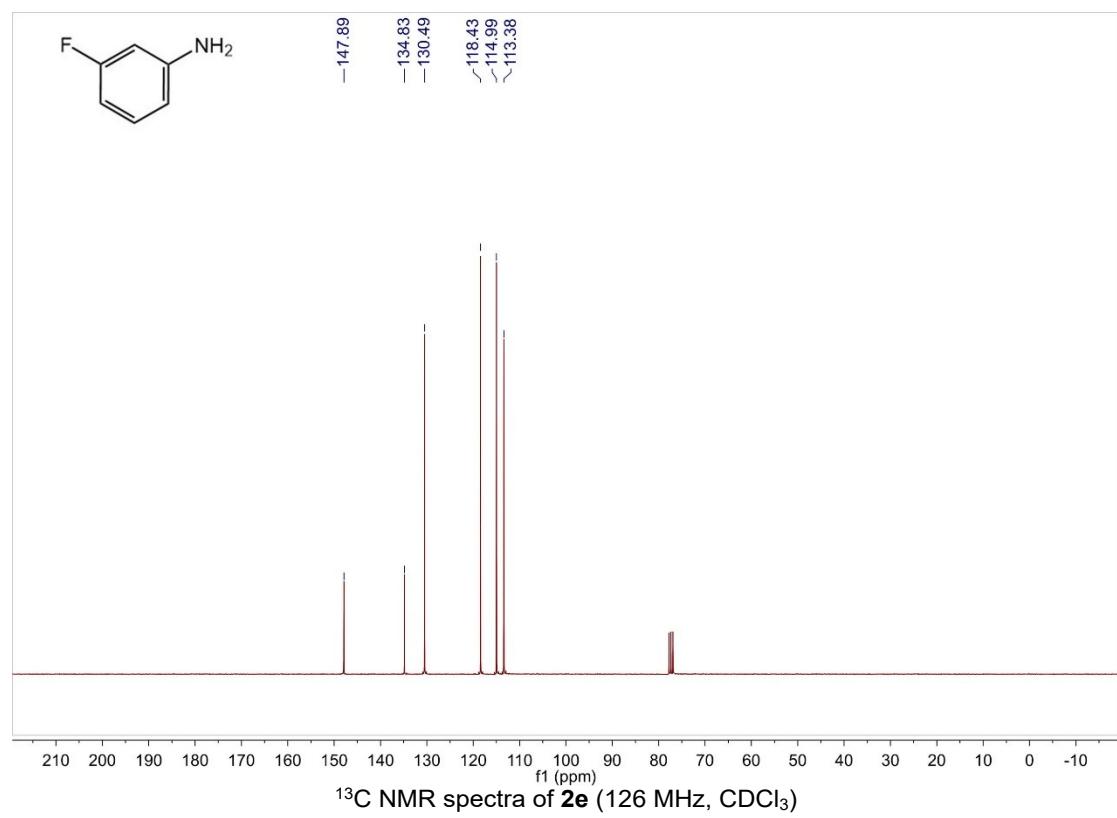
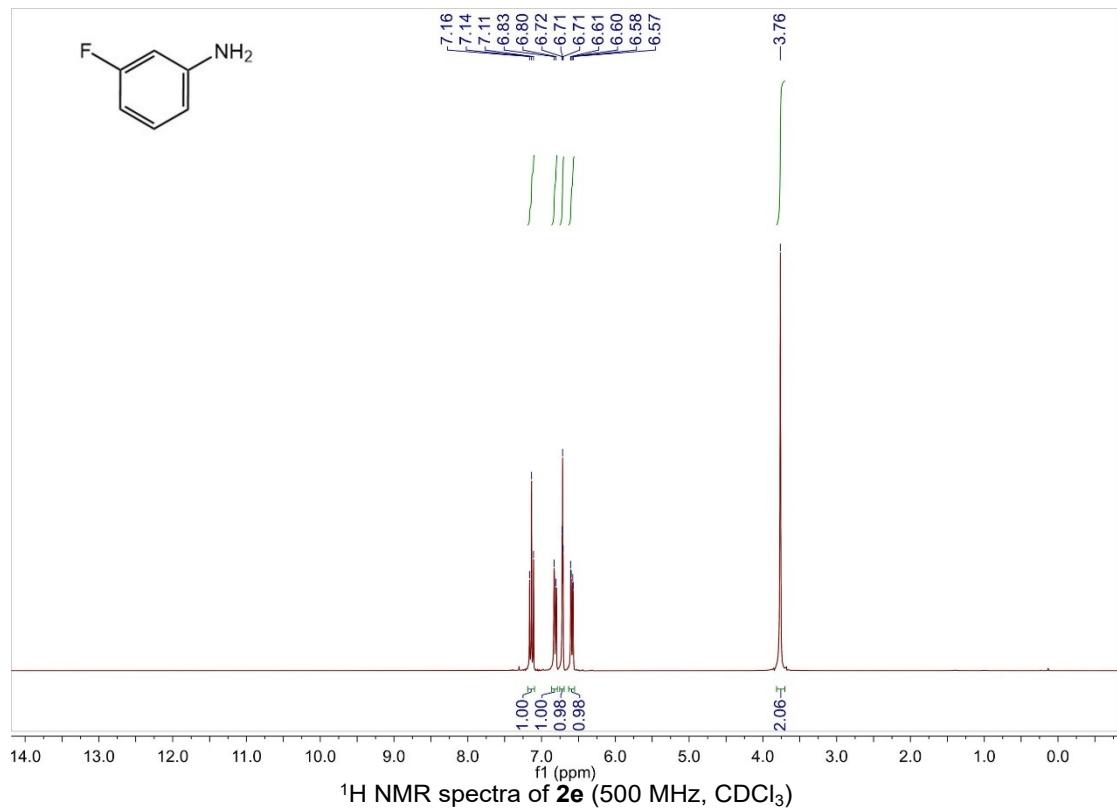
¹³C NMR spectra of **2b** (126 MHz, CDCl₃)

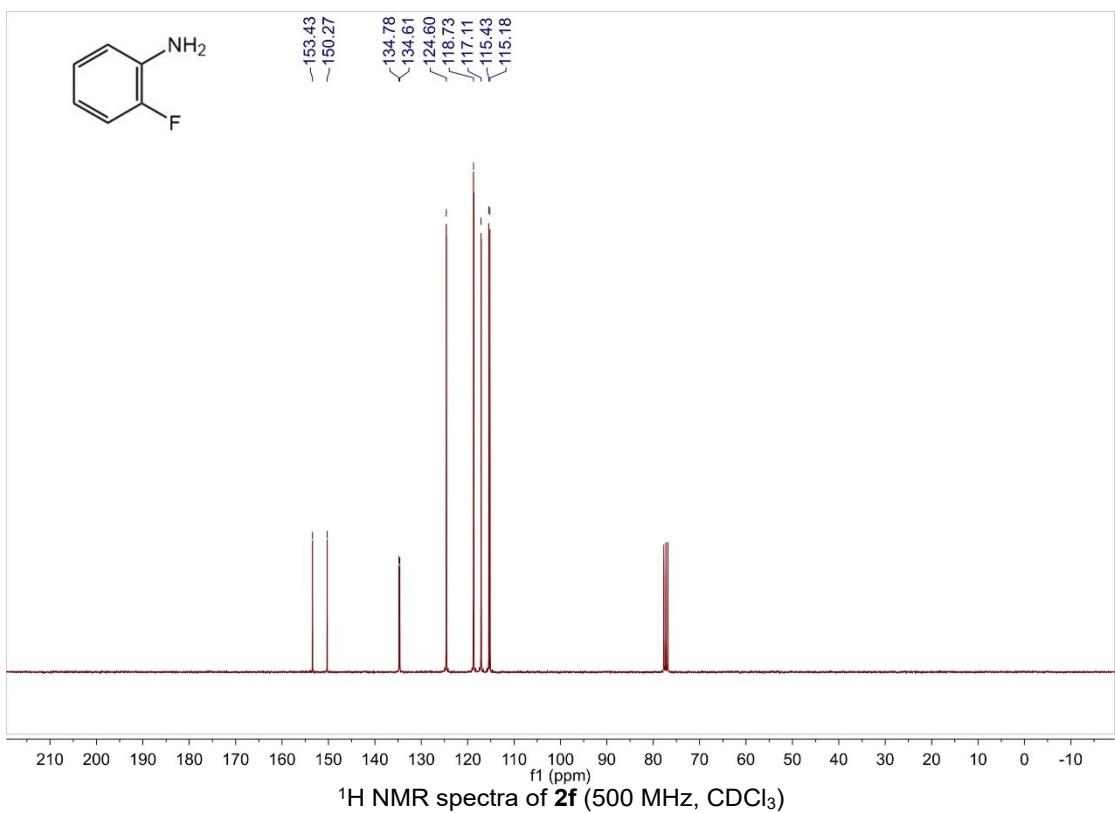
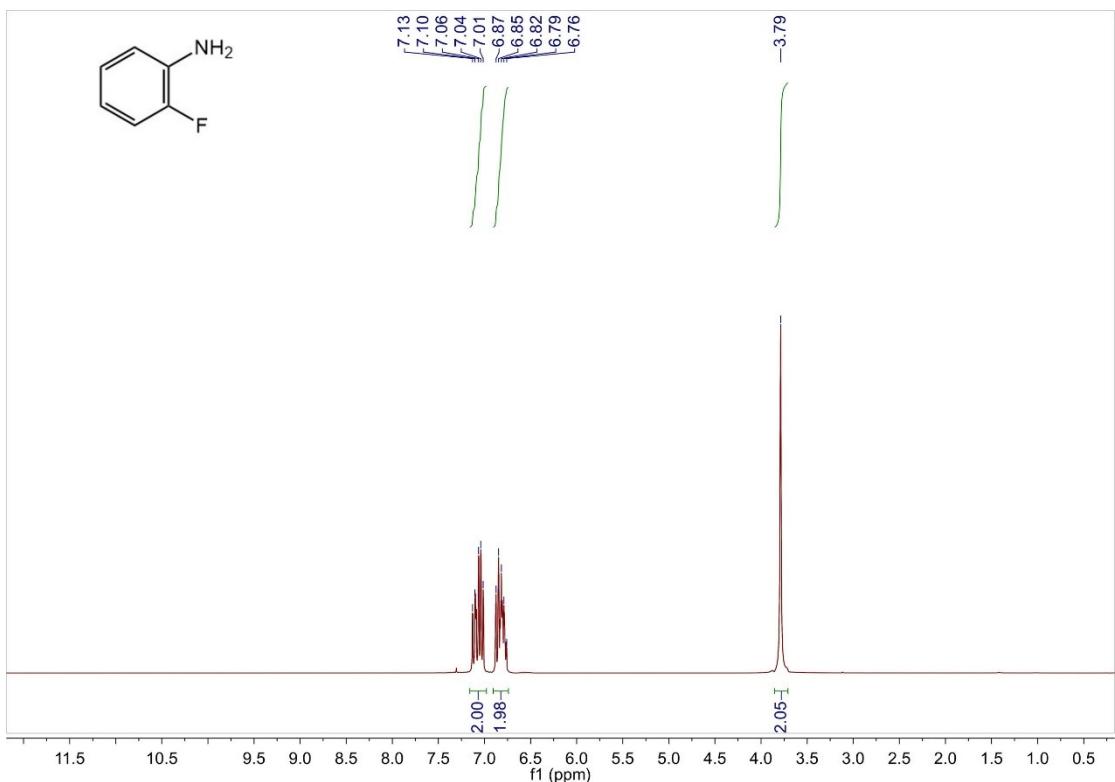




¹H NMR spectra of **2d** (500 MHz, CDCl₃)

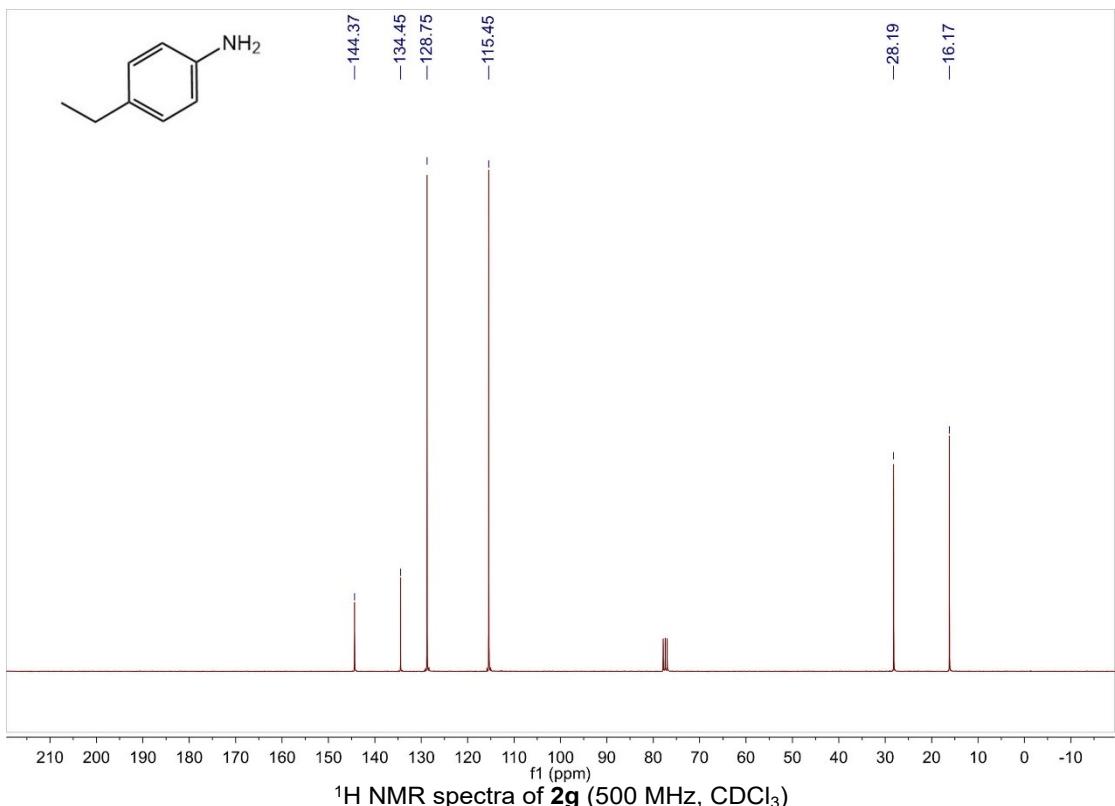
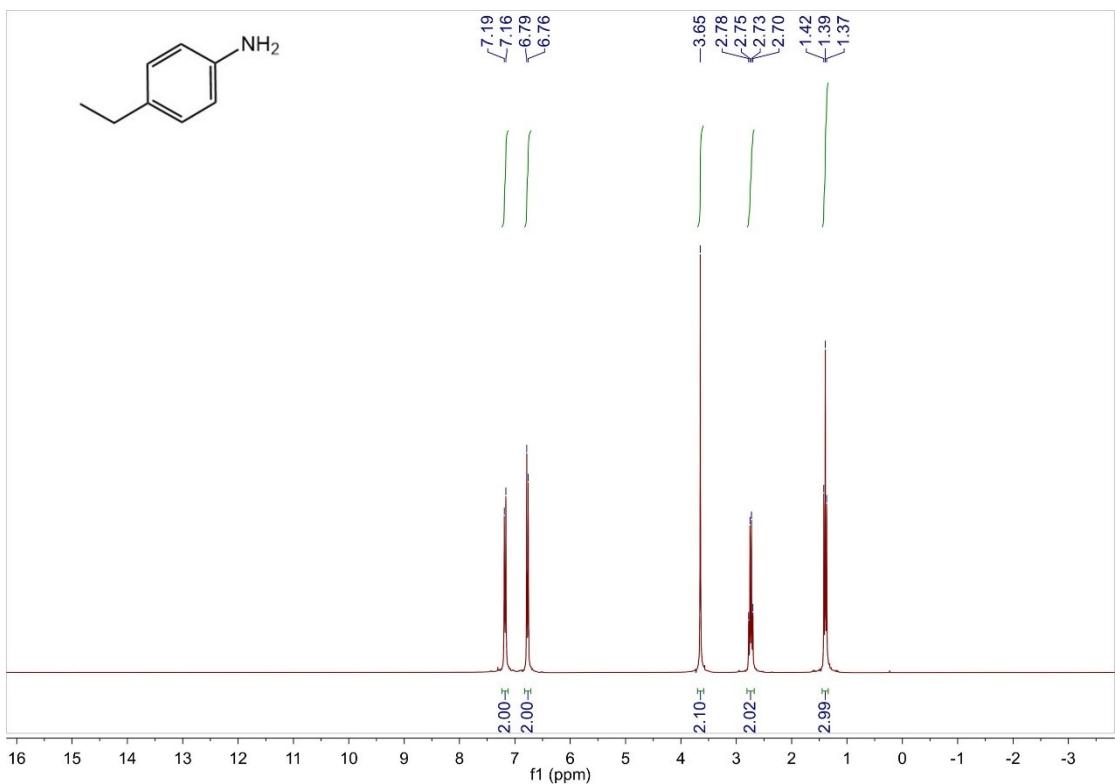
¹³C NMR spectra of **2d** (126 MHz, CDCl₃)





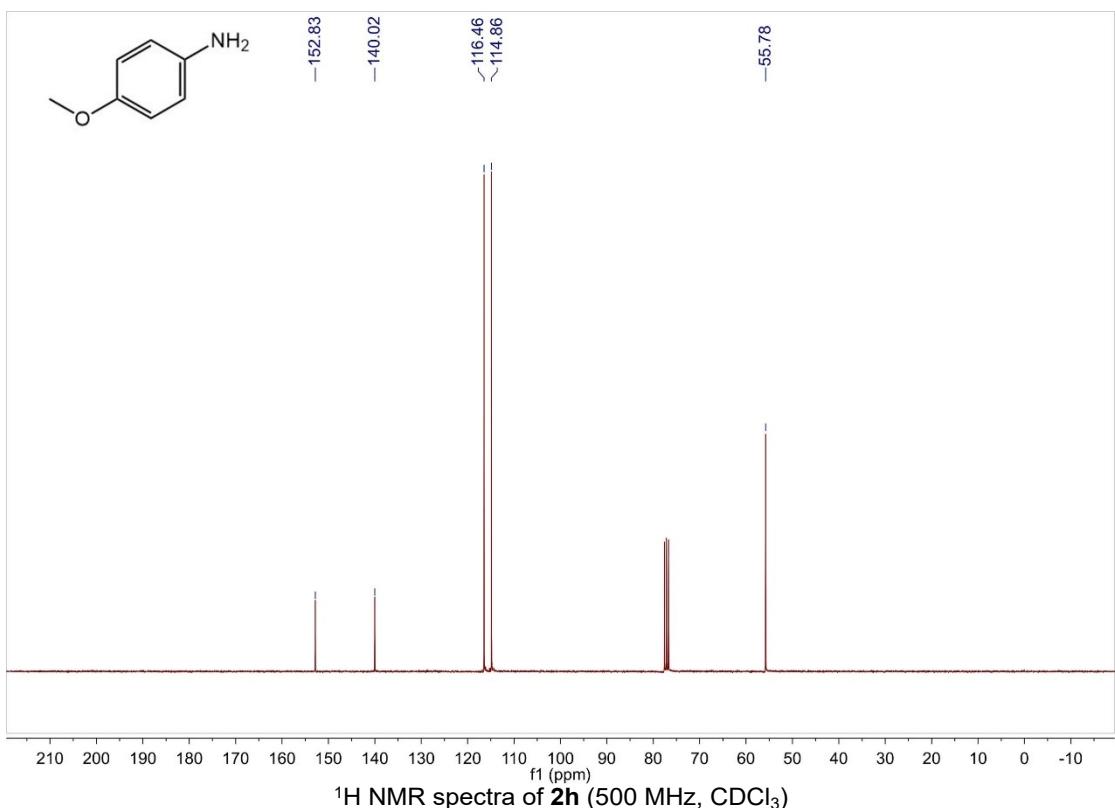
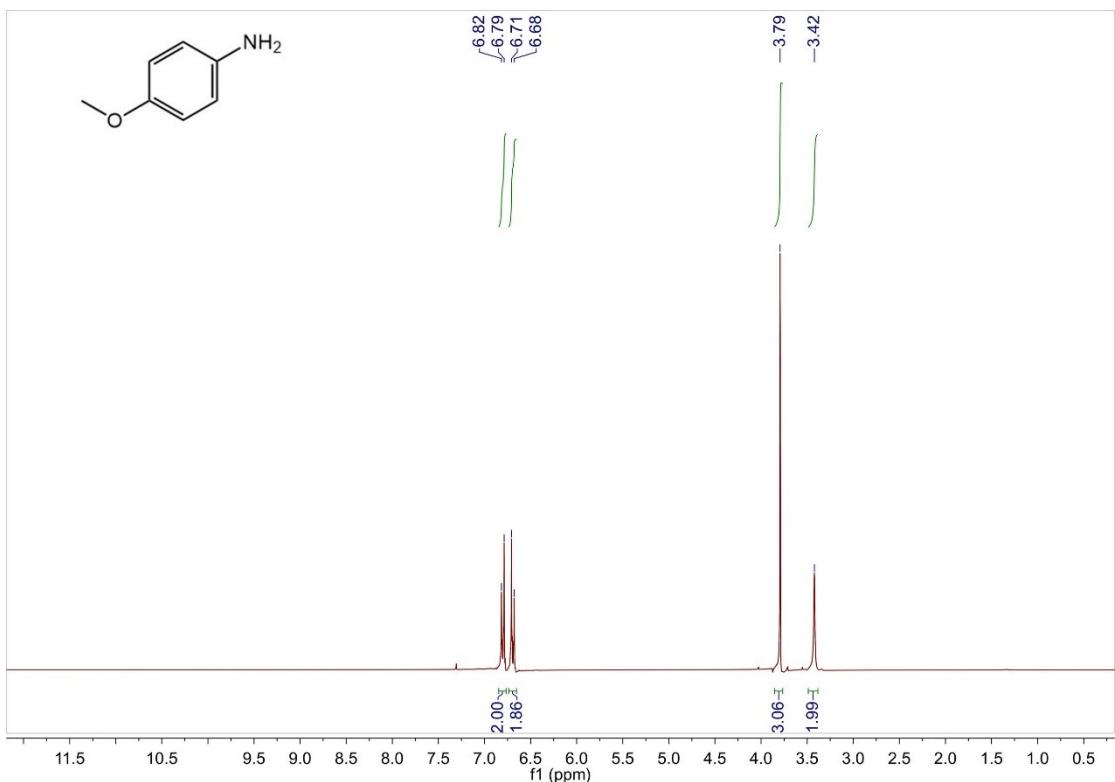
¹H NMR spectra of **2f** (500 MHz, CDCl₃)

¹³C NMR spectra of **2f** (126 MHz, CDCl₃)



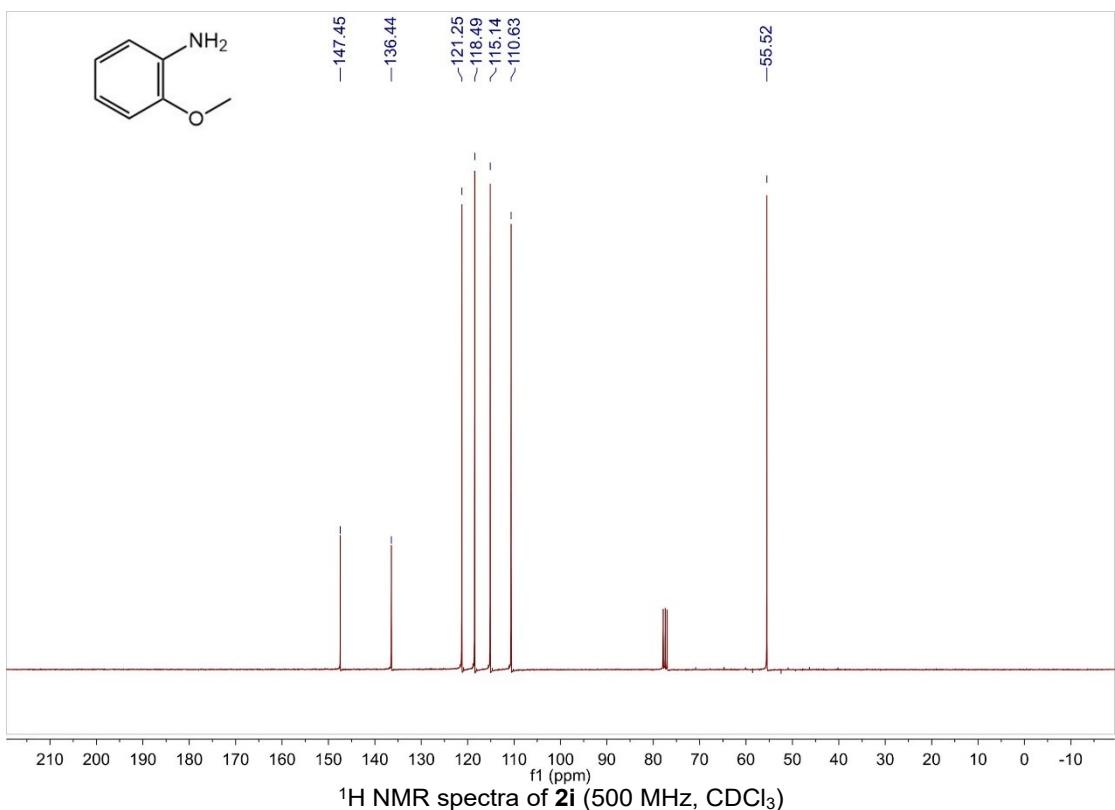
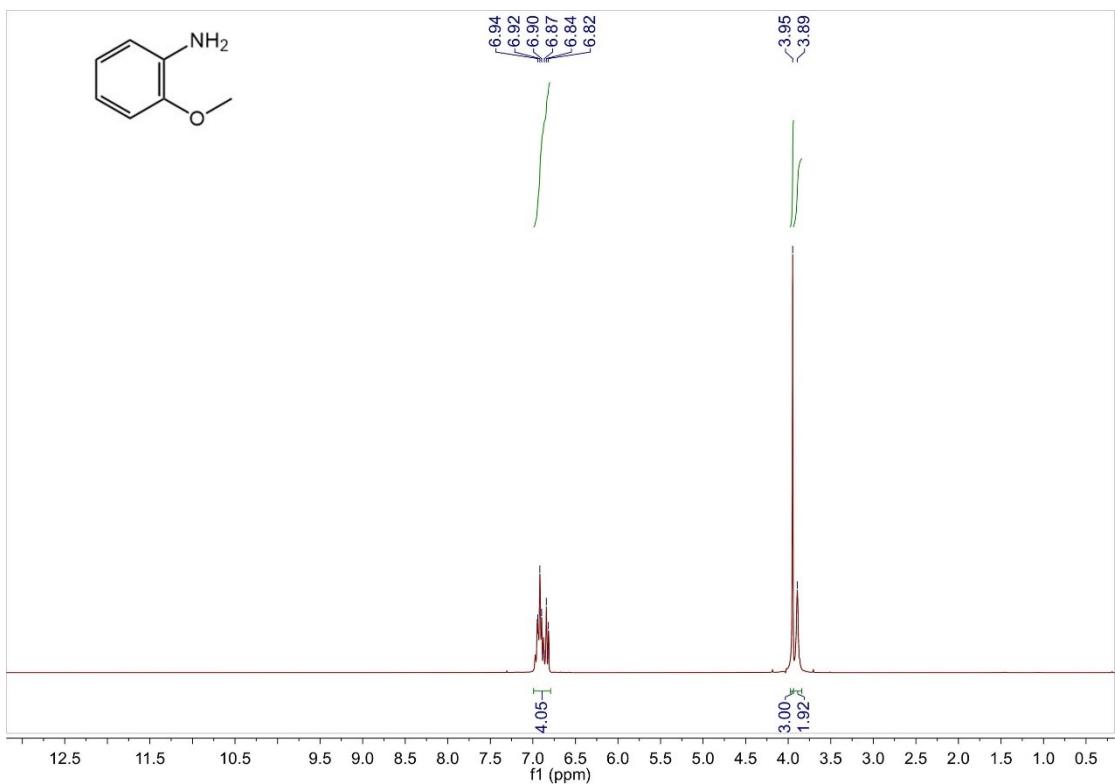
¹H NMR spectra of **2g** (500 MHz, CDCl₃)

¹³C NMR spectra of **2g** (126 MHz, CDCl₃)

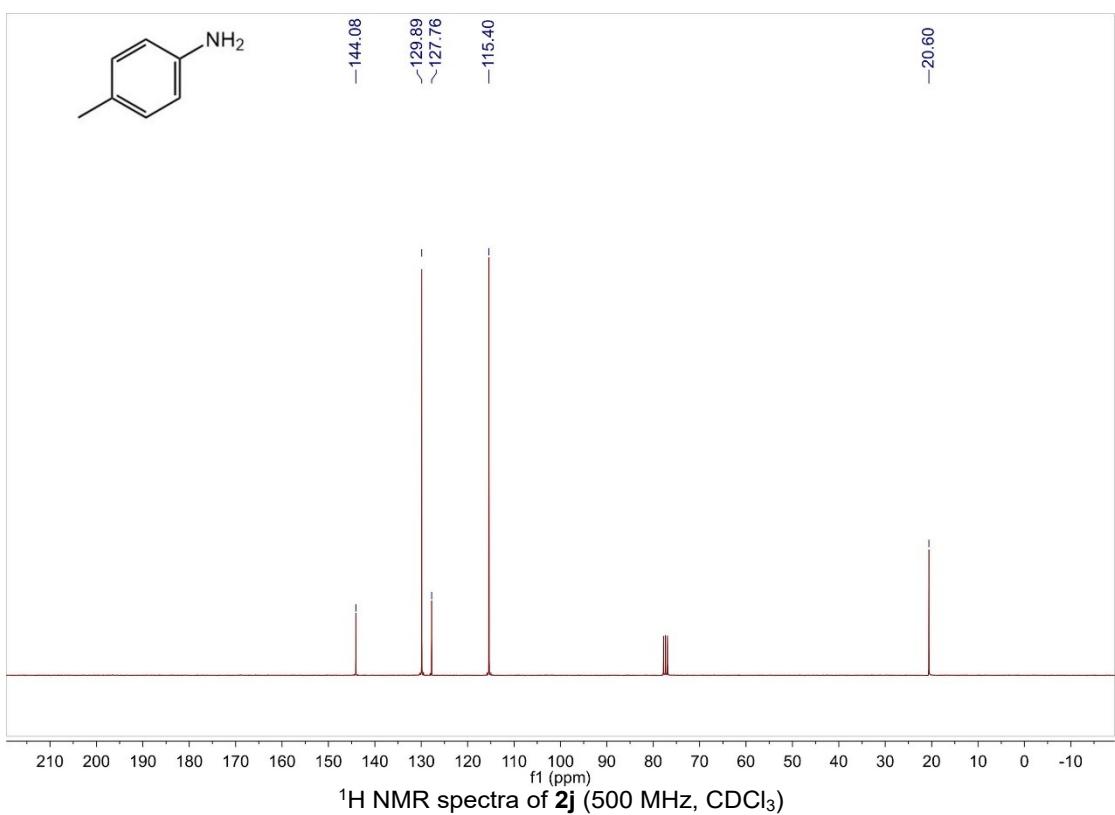
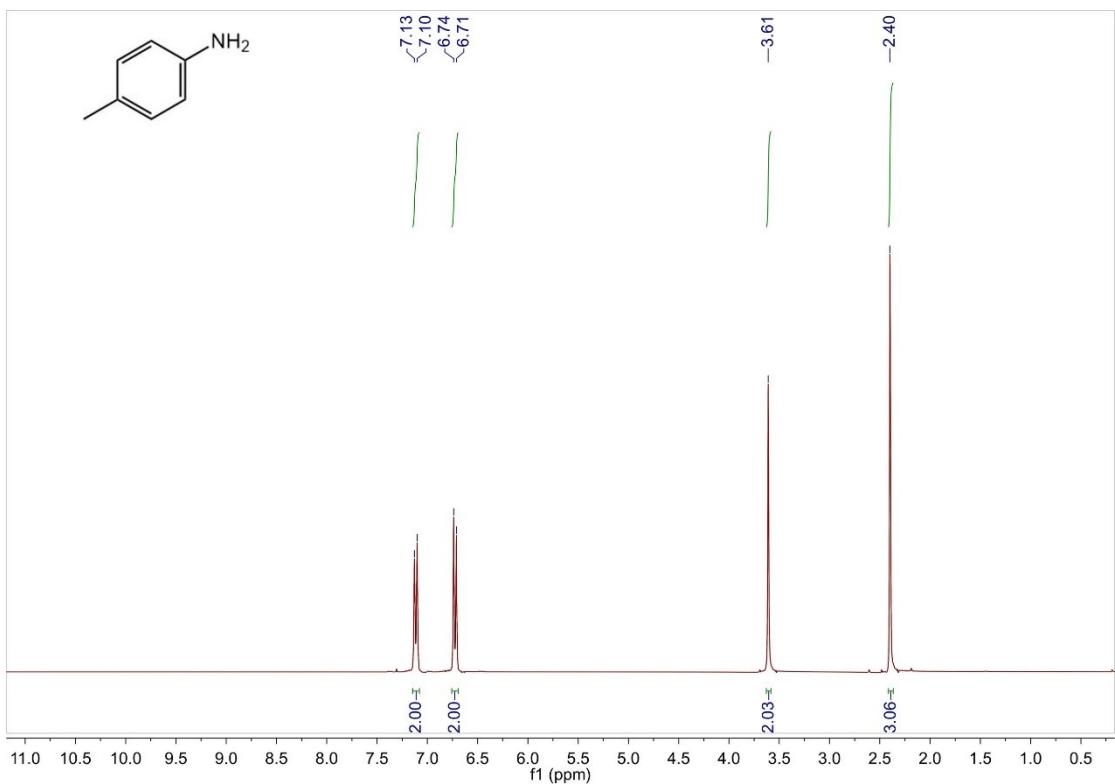


¹H NMR spectra of **2h** (500 MHz, CDCl₃)

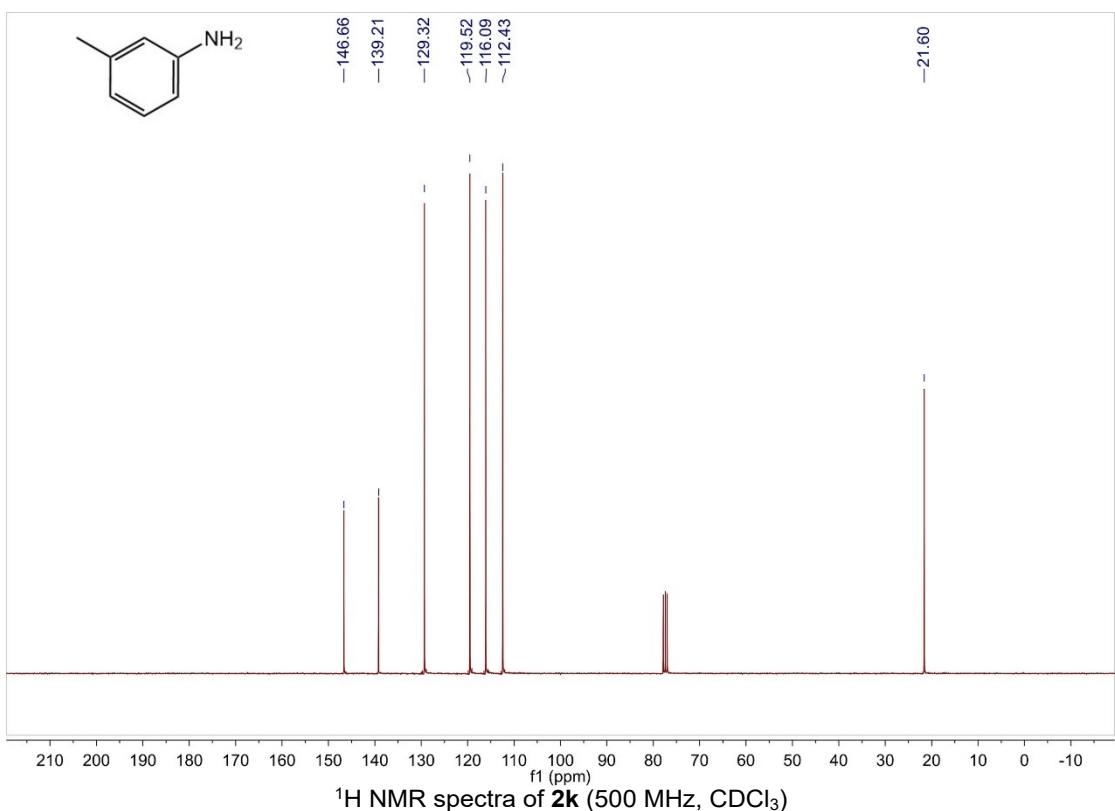
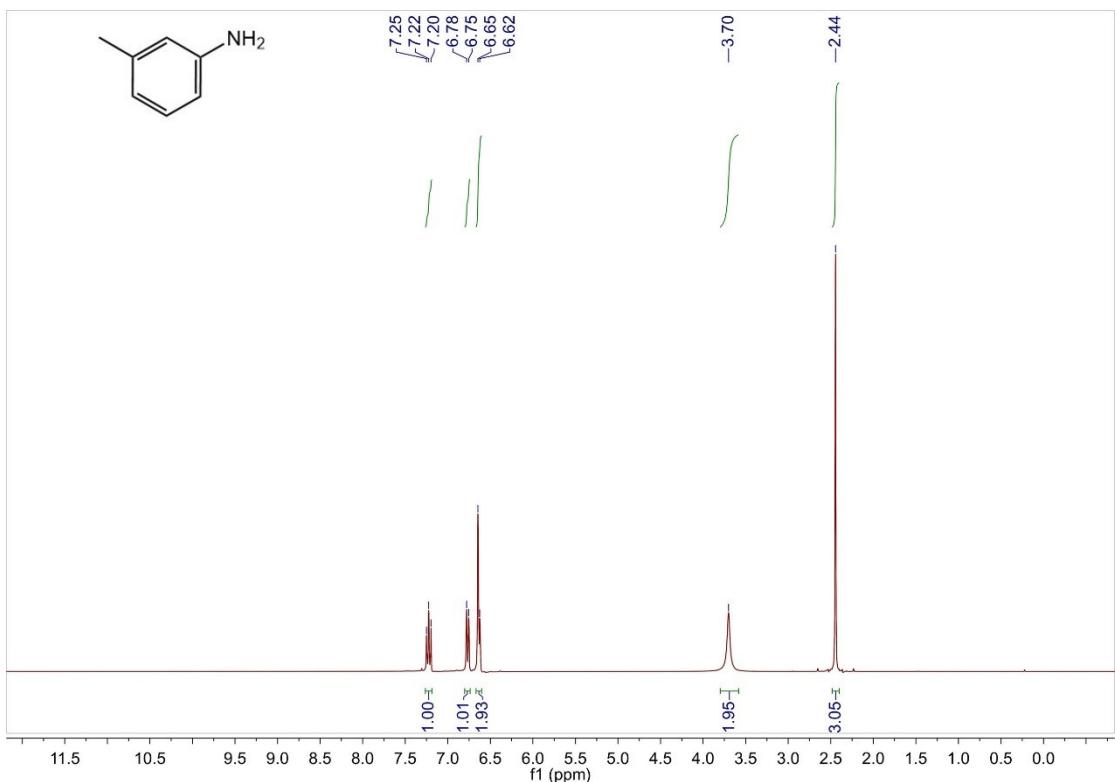
¹³C NMR spectra of **2h** (126 MHz, CDCl₃)



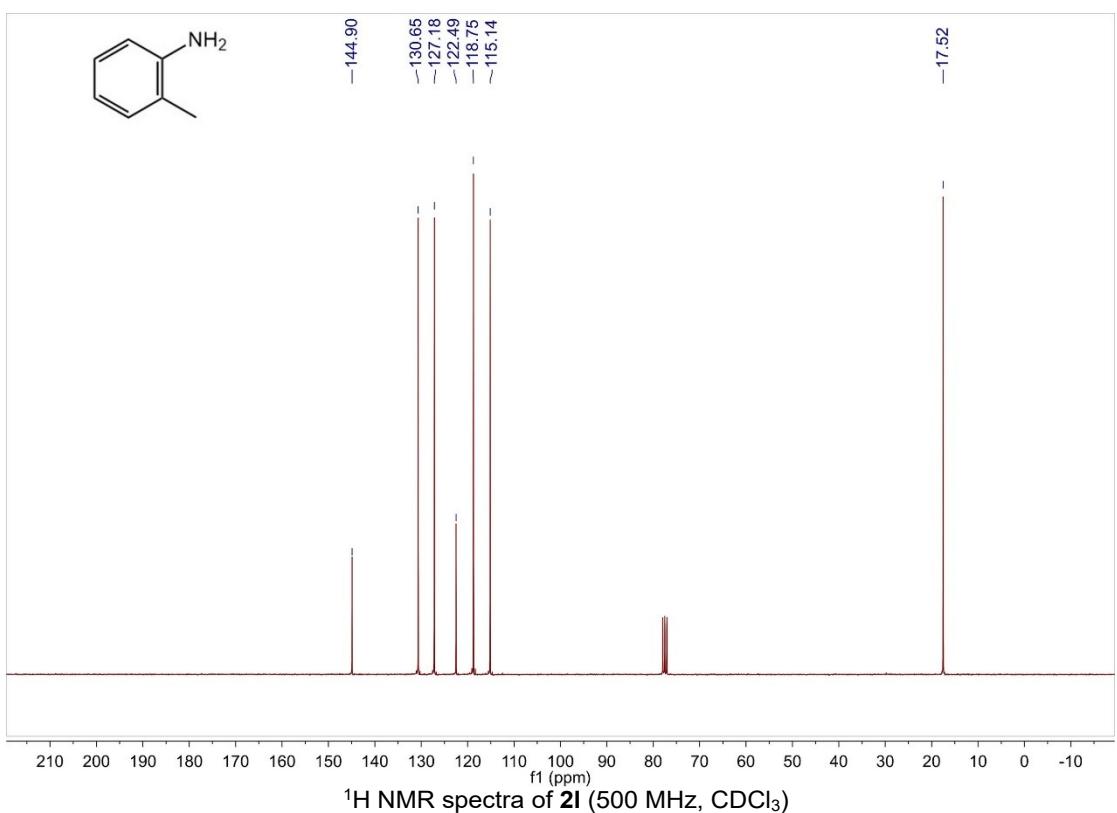
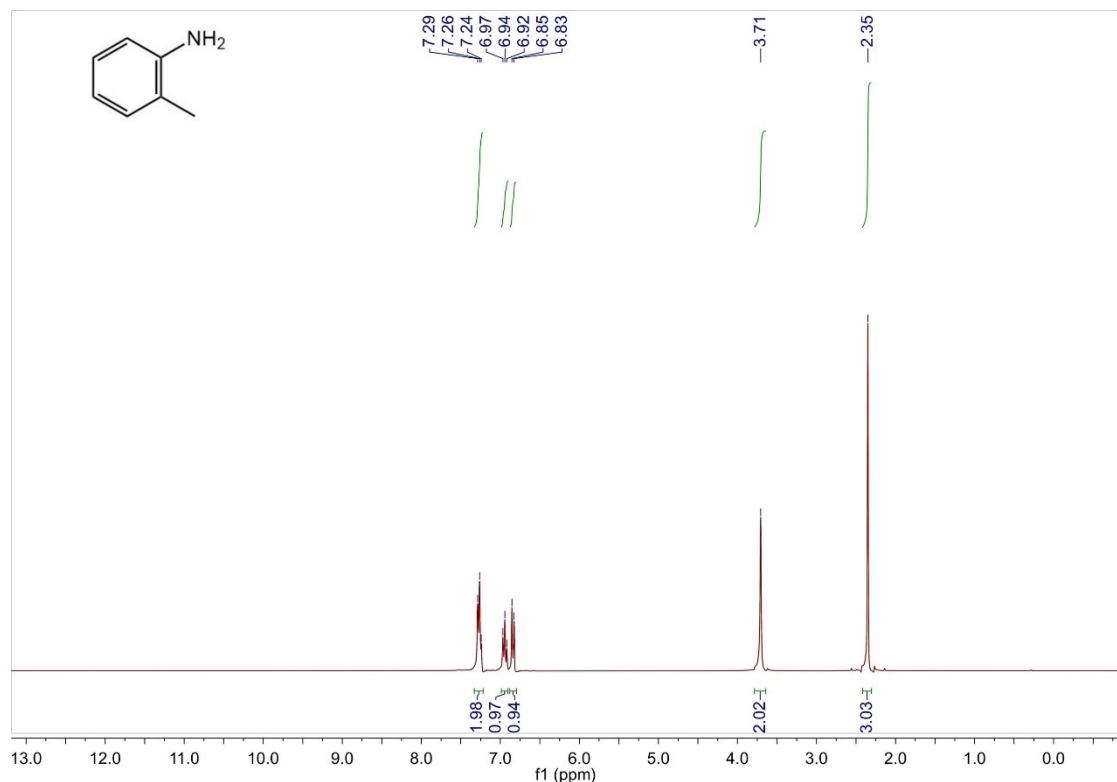
¹³C NMR spectra of **2i** (126 MHz, CDCl₃)



¹³C NMR spectra of **2j** (126 MHz, CDCl₃)

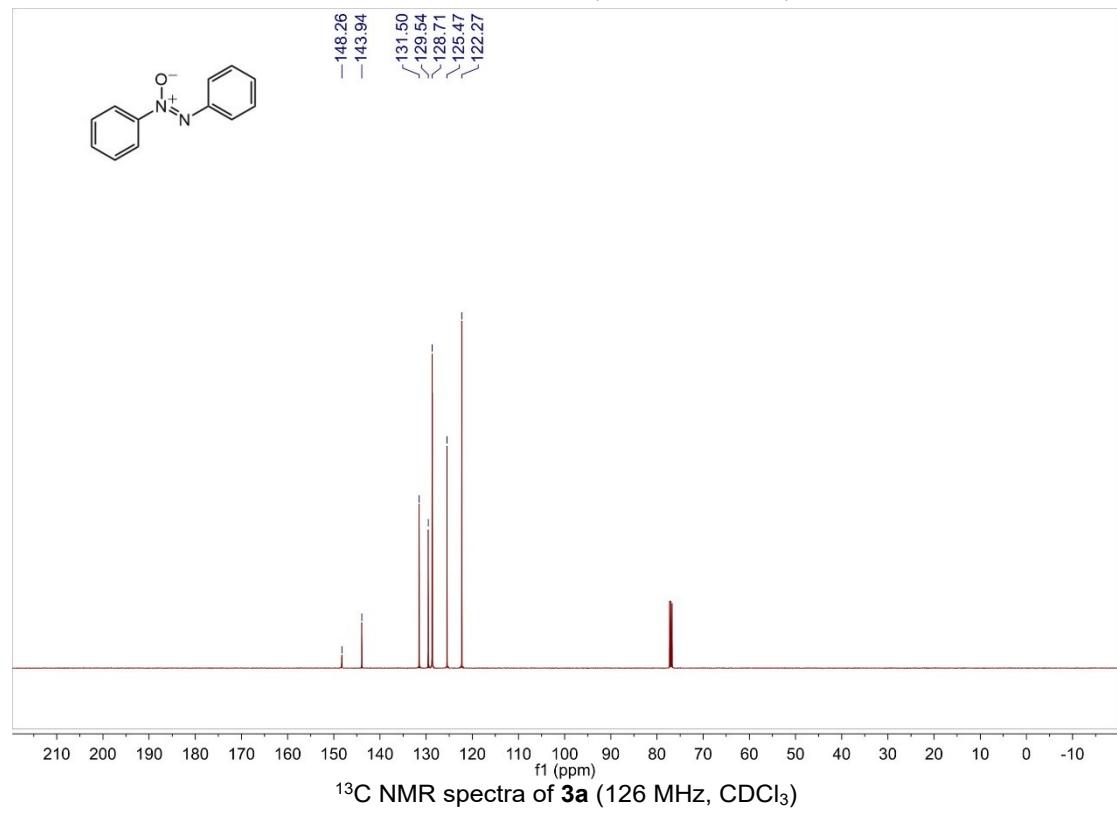
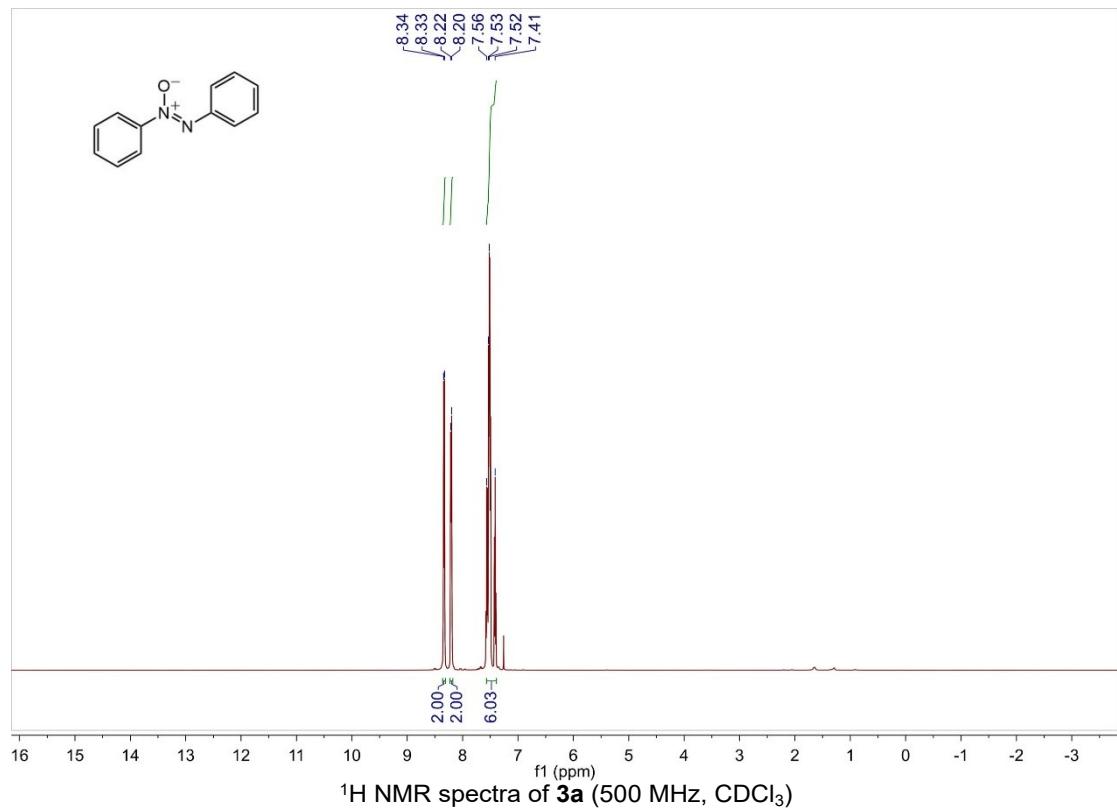


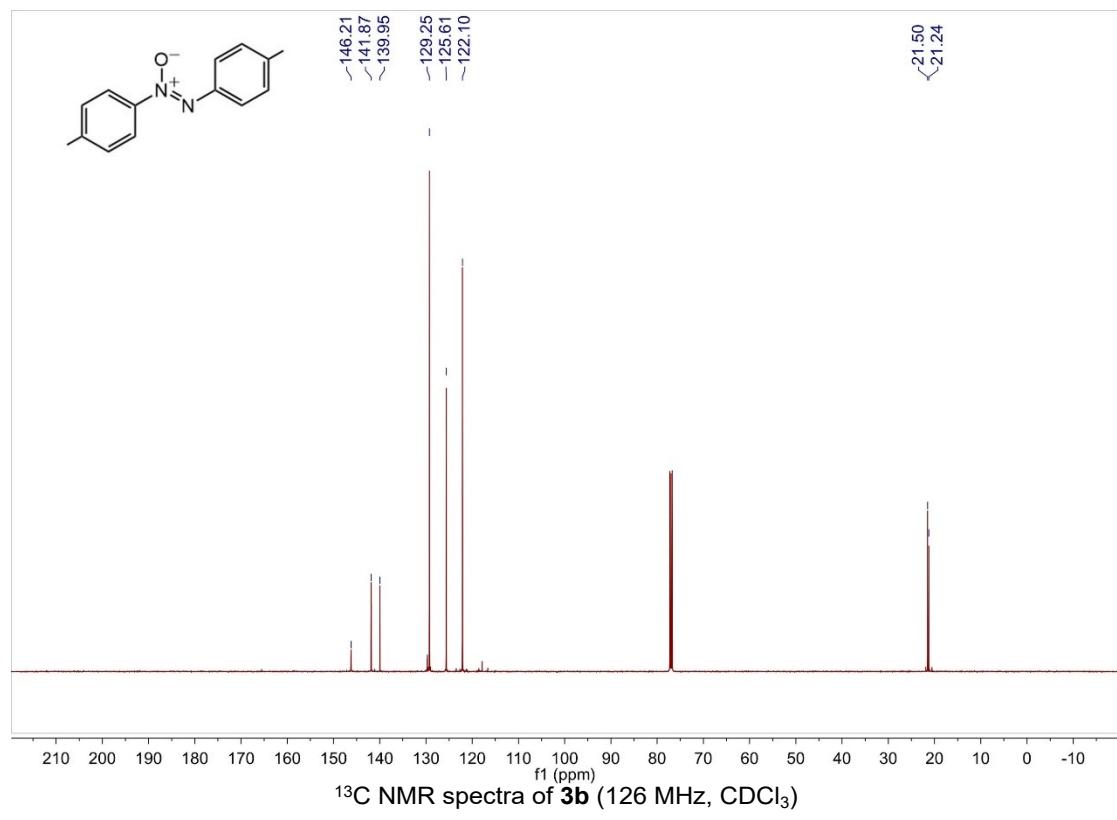
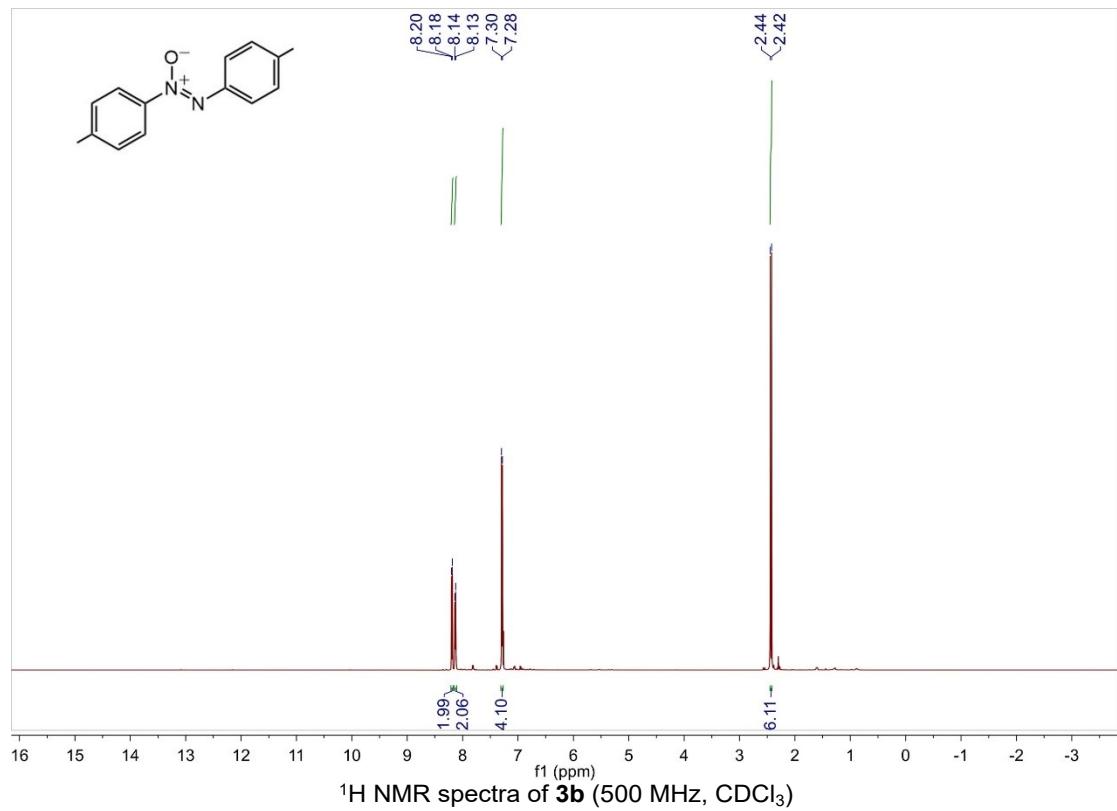
¹³C NMR spectra of 2k (126 MHz, CDCl₃)

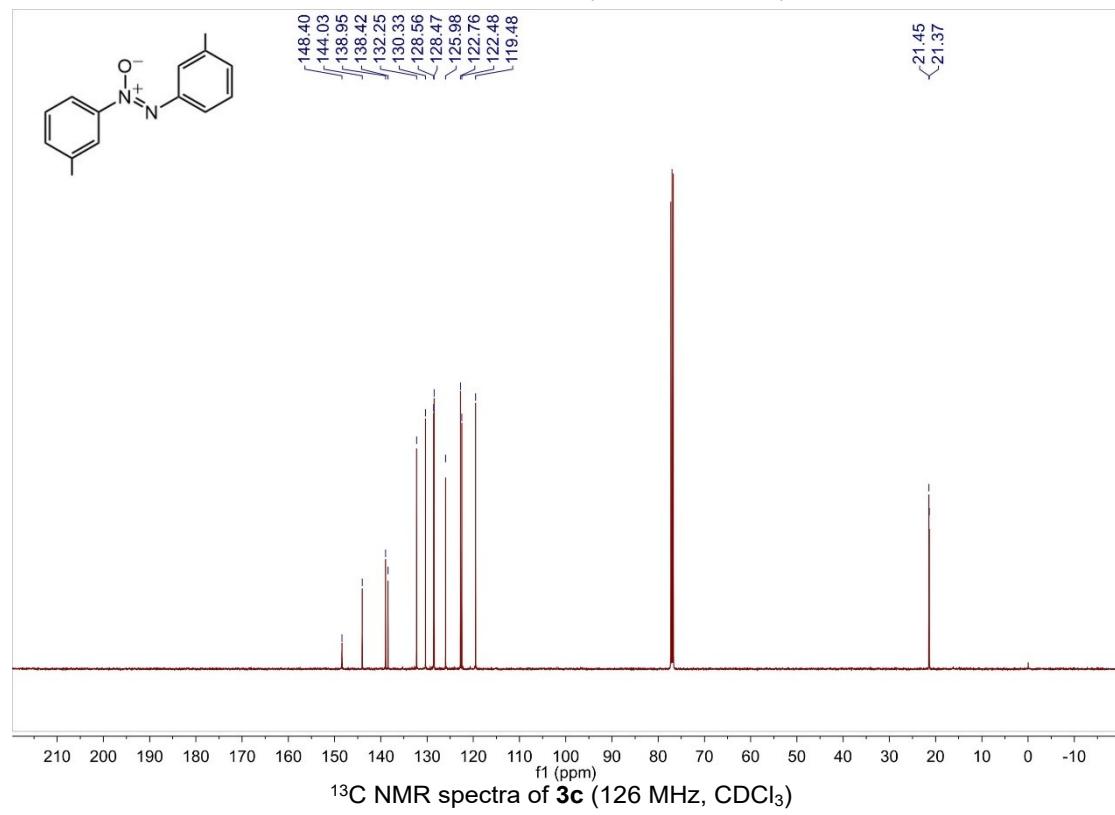
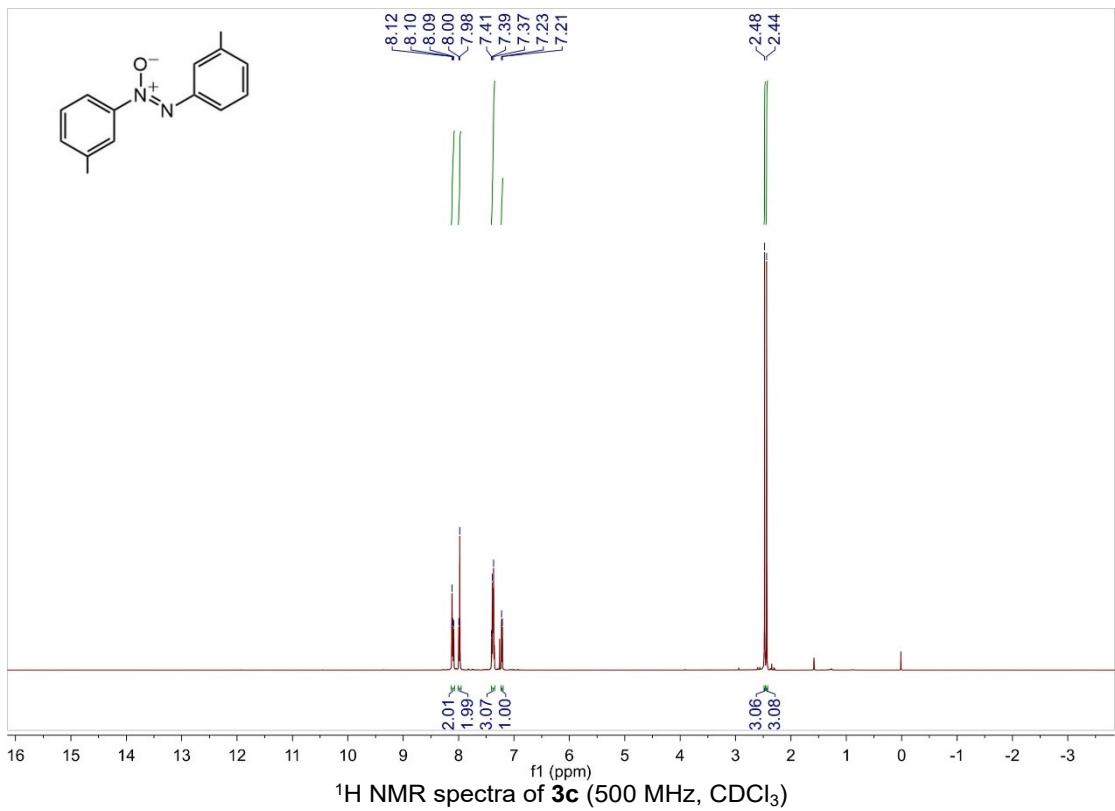


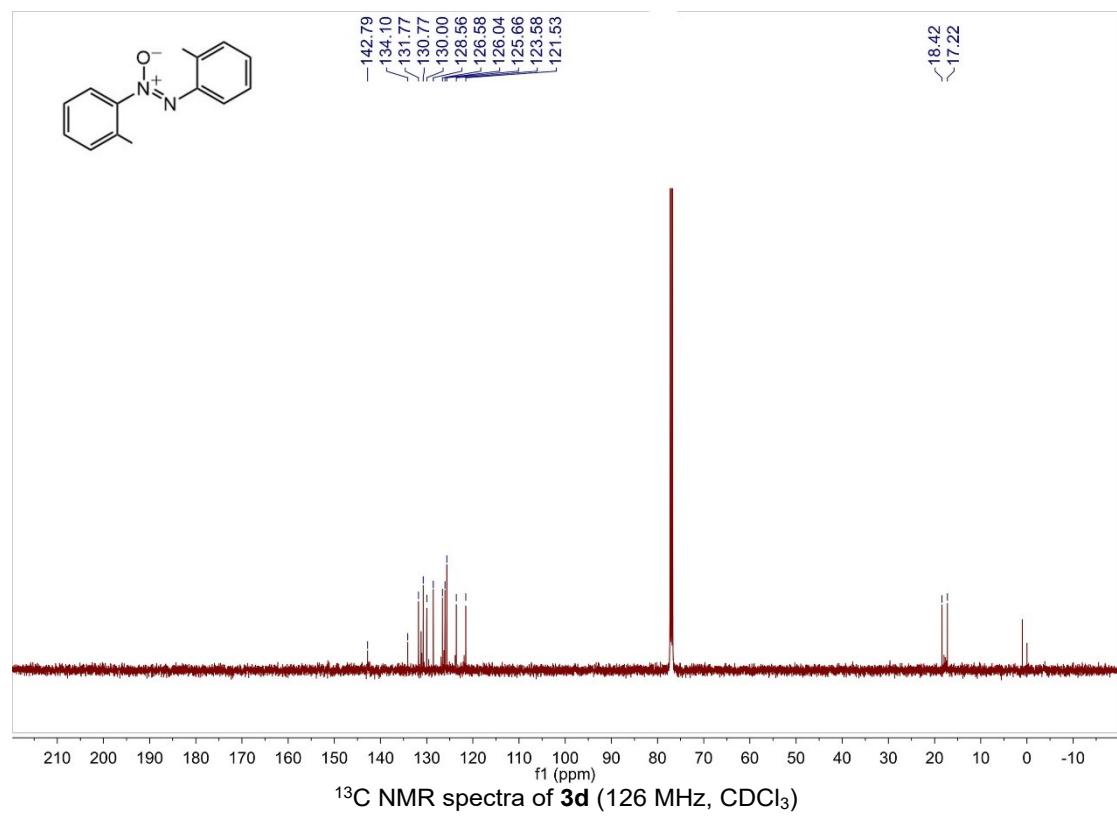
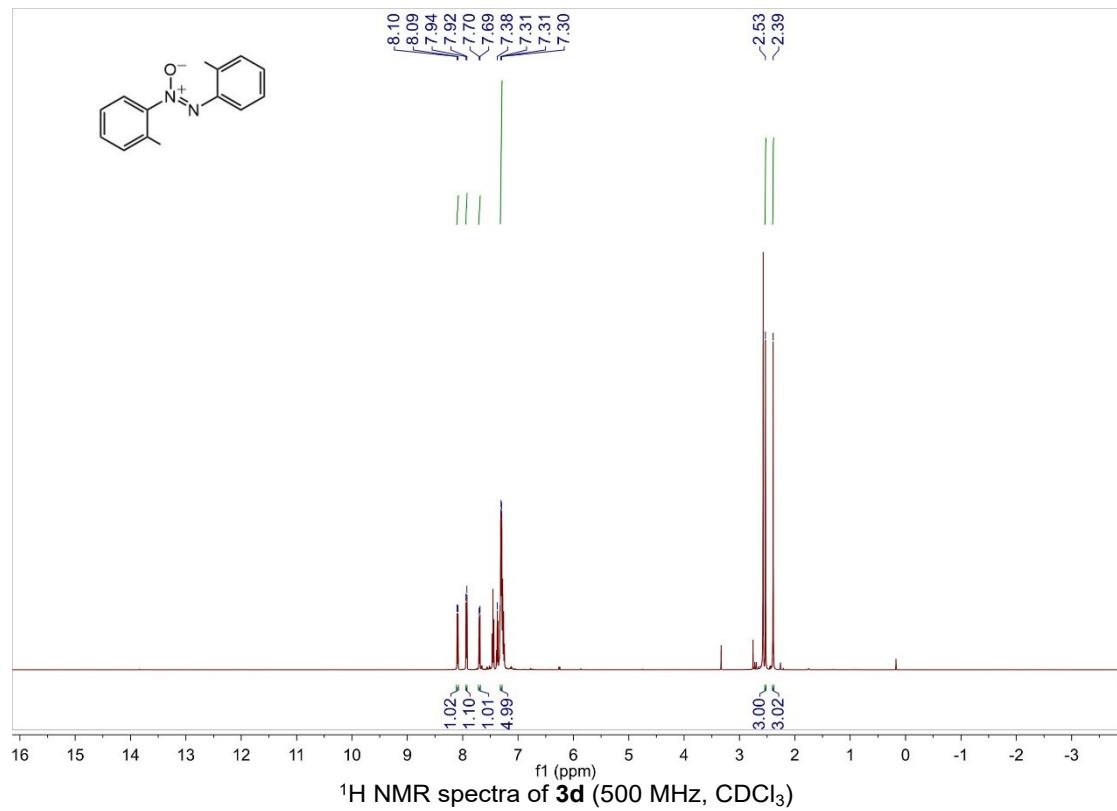
¹H NMR spectra of **2I** (500 MHz, CDCl₃)

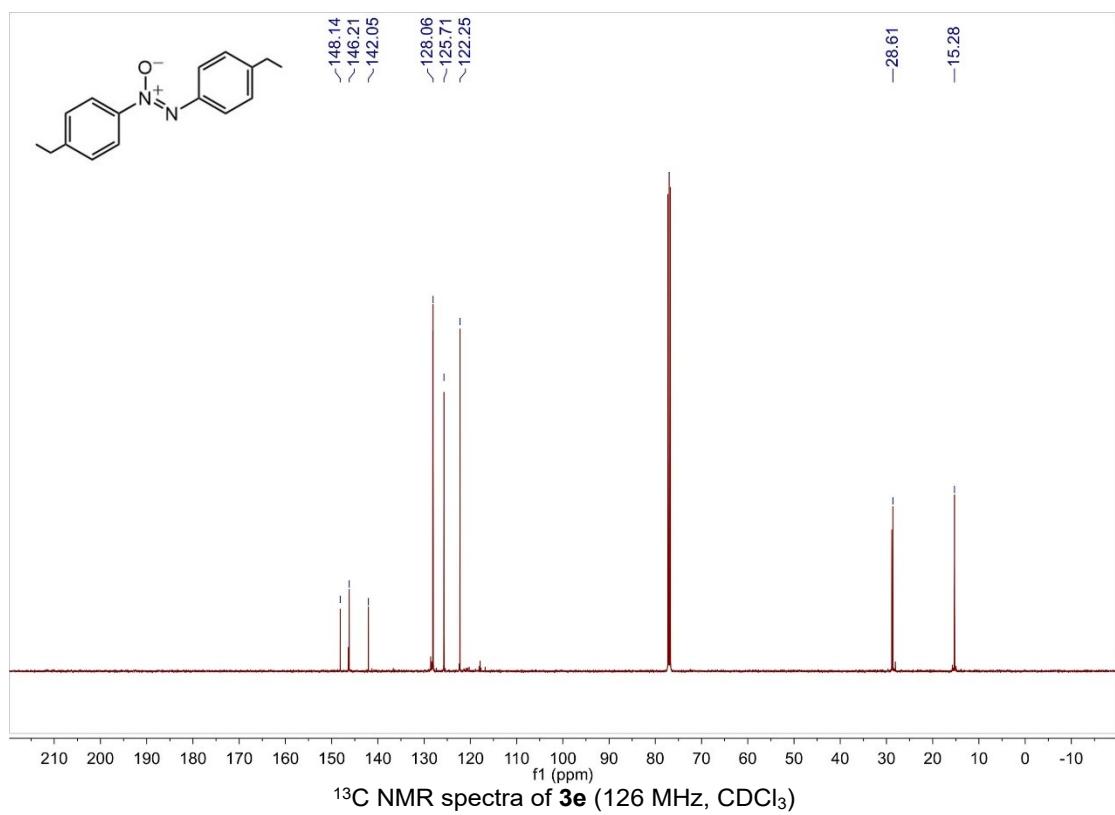
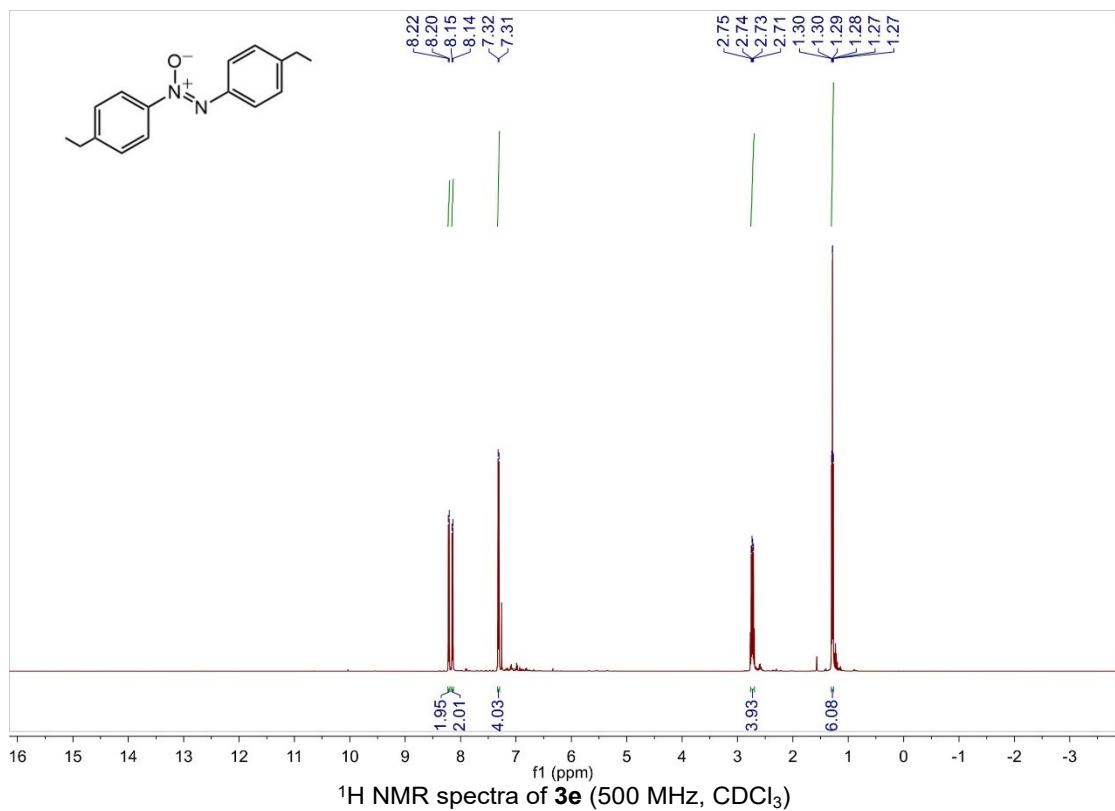
¹³C NMR spectra of **2I** (126 MHz, CDCl₃)

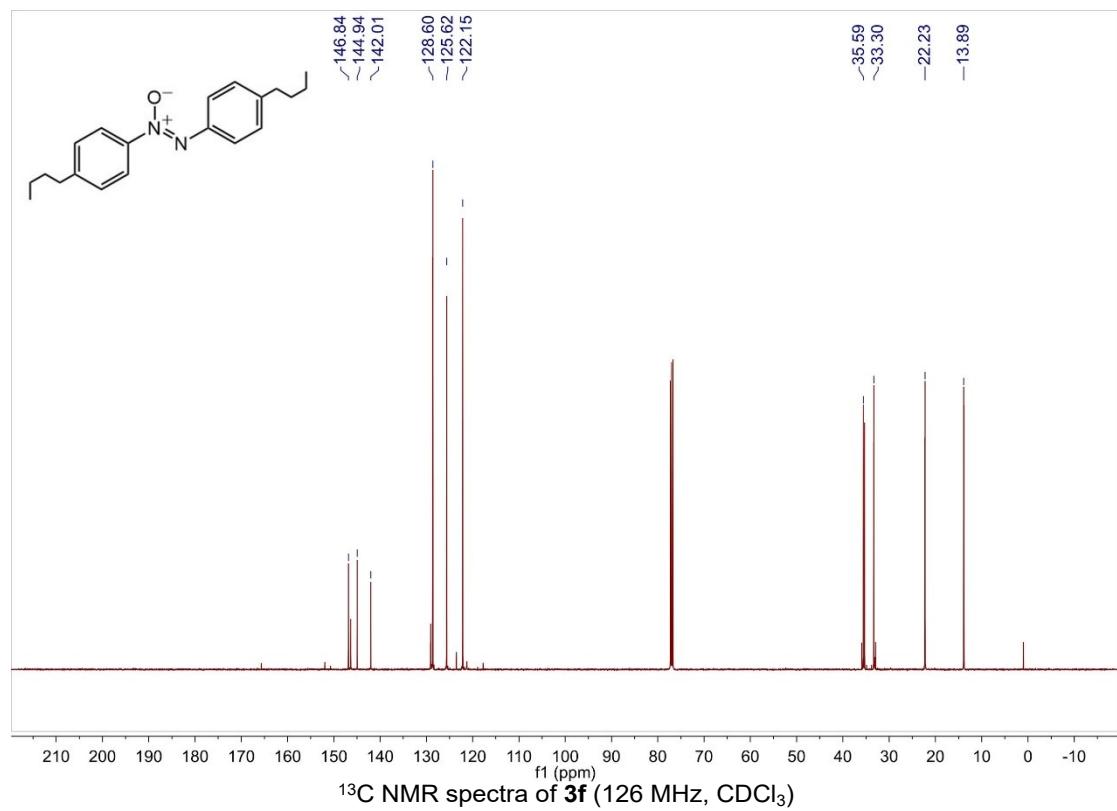
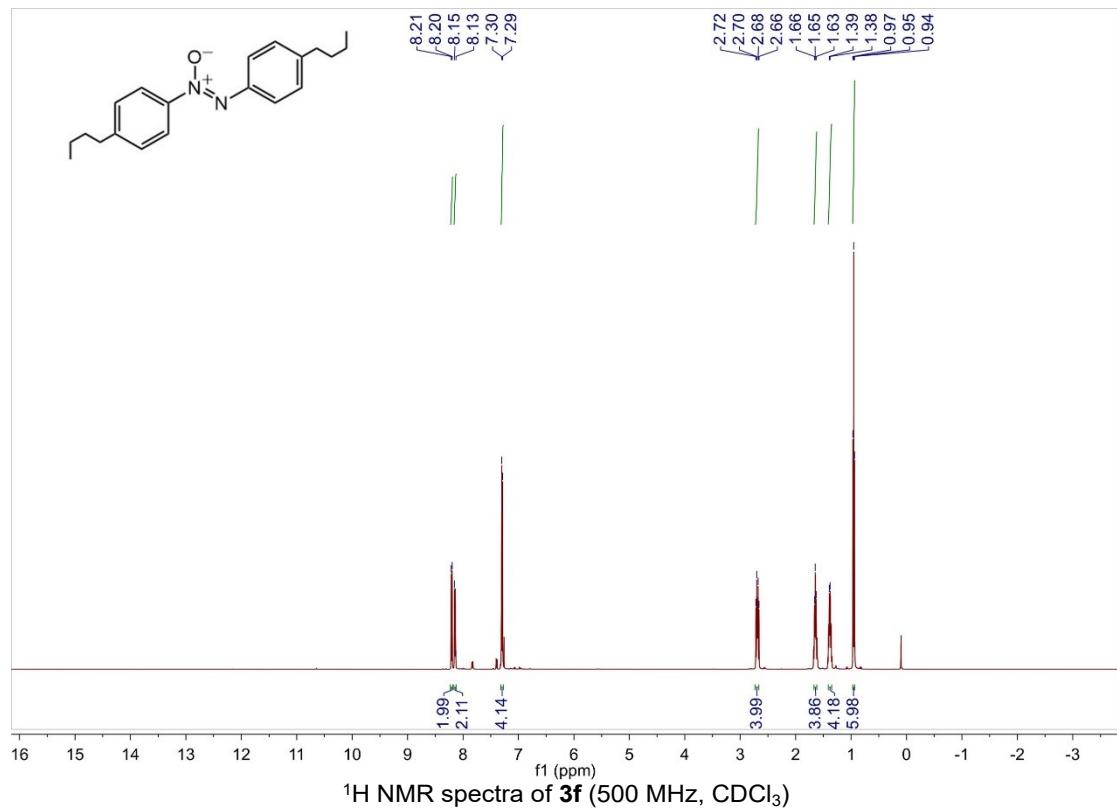


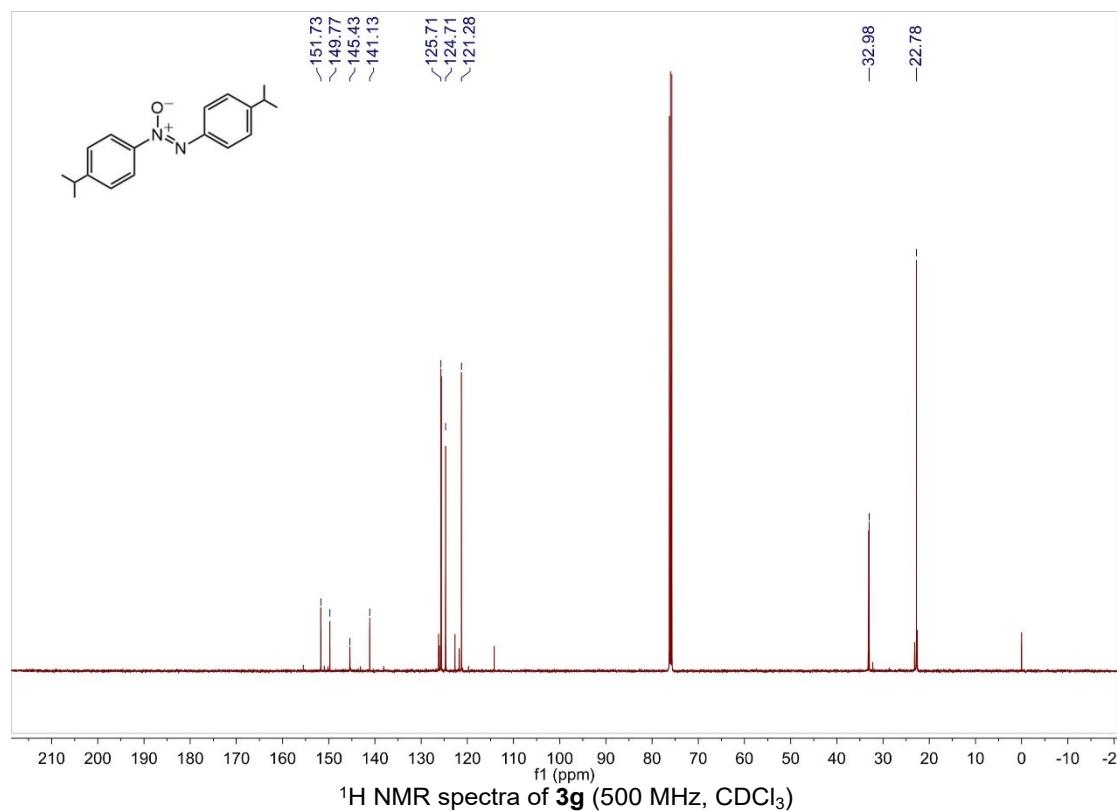
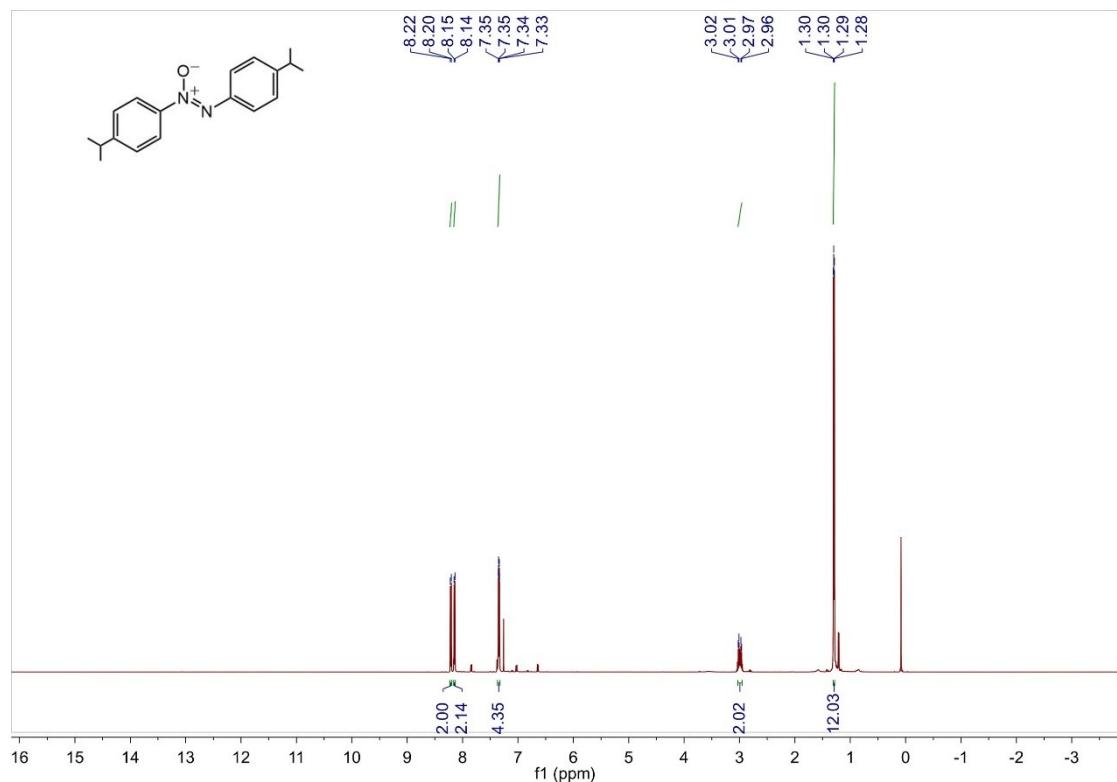




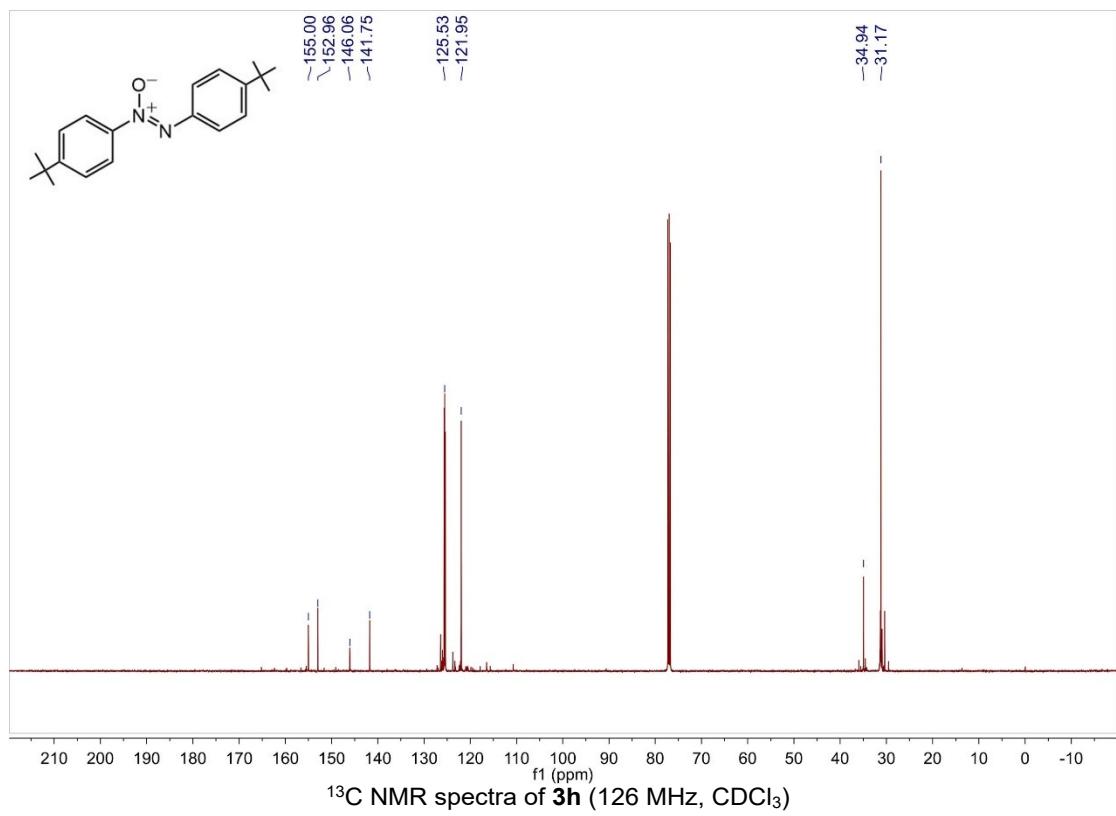
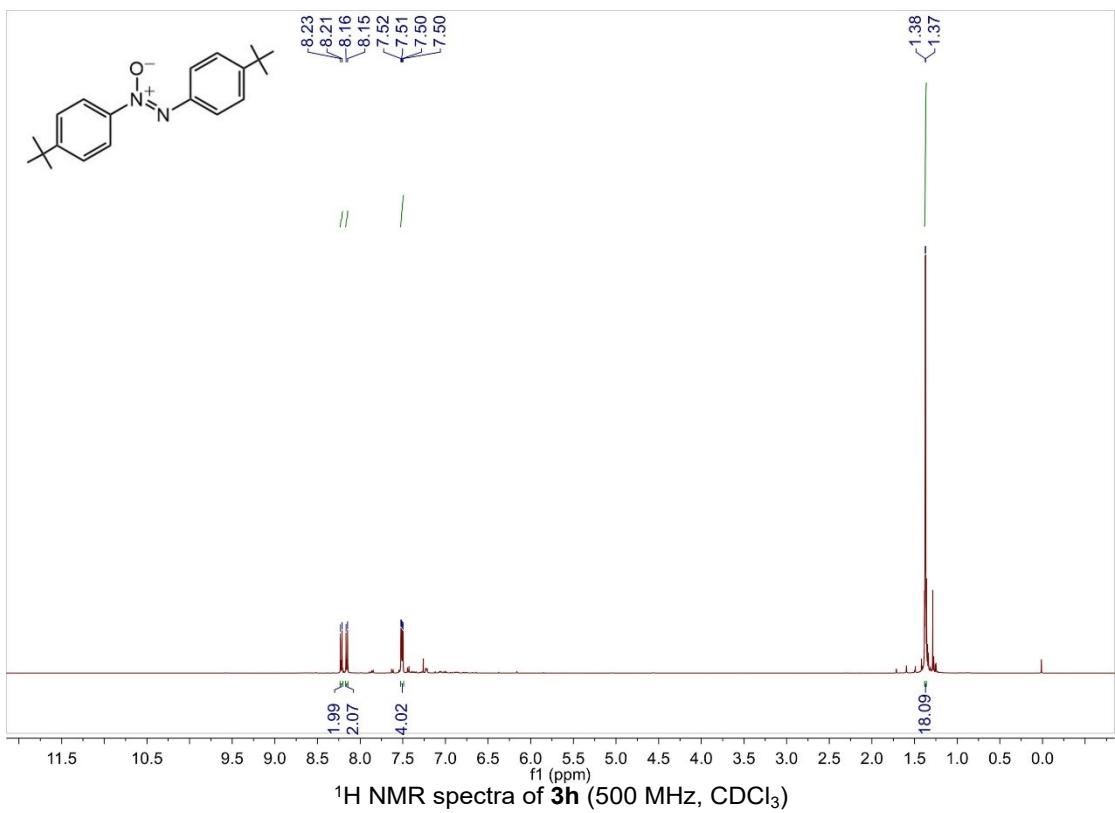


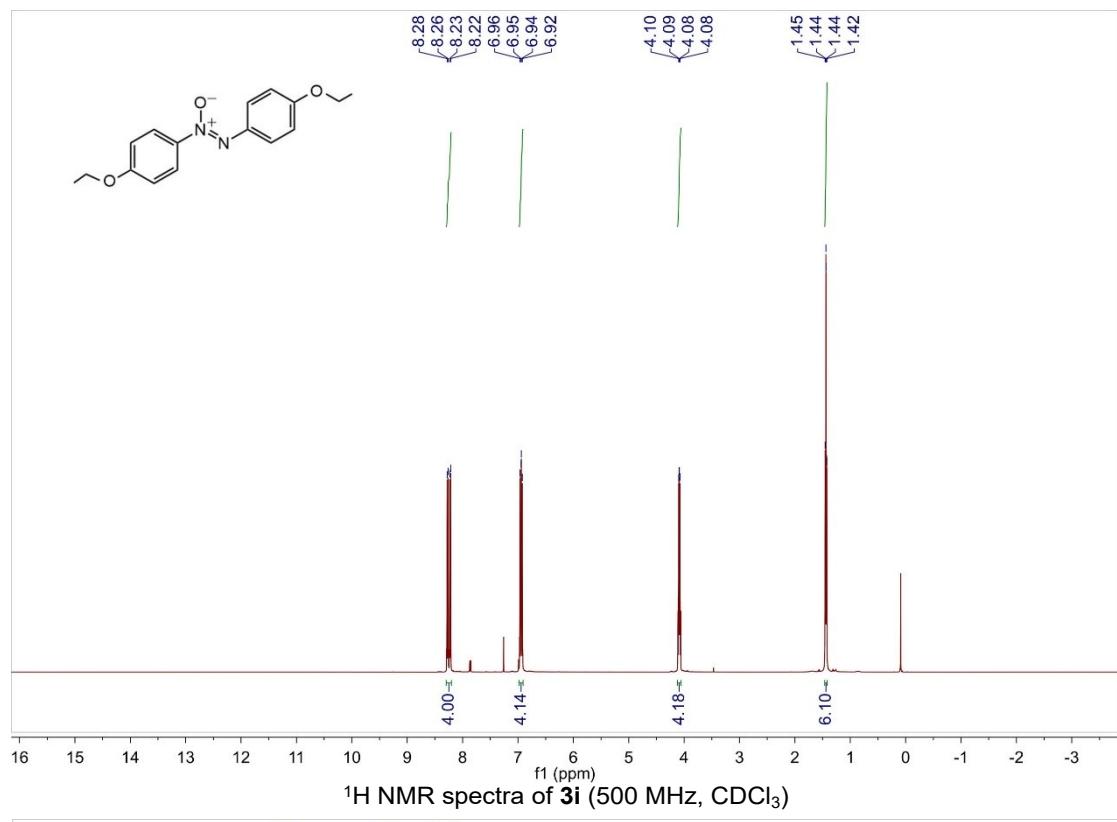


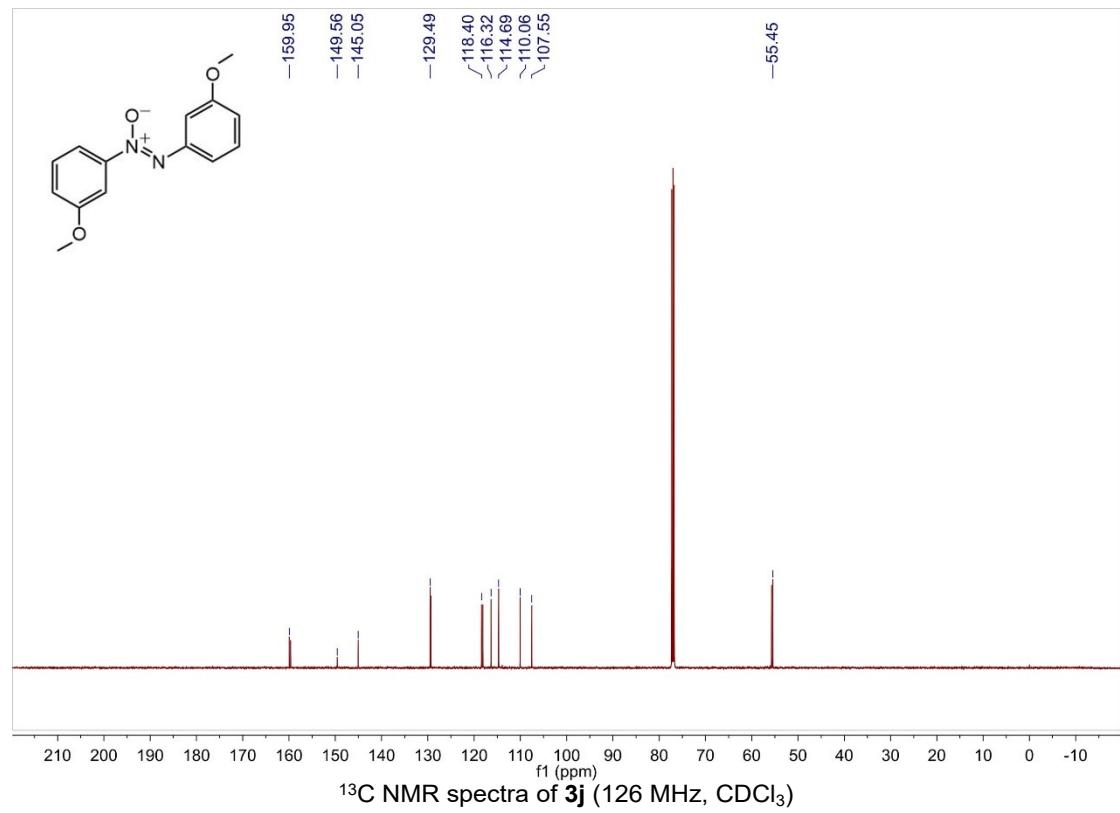
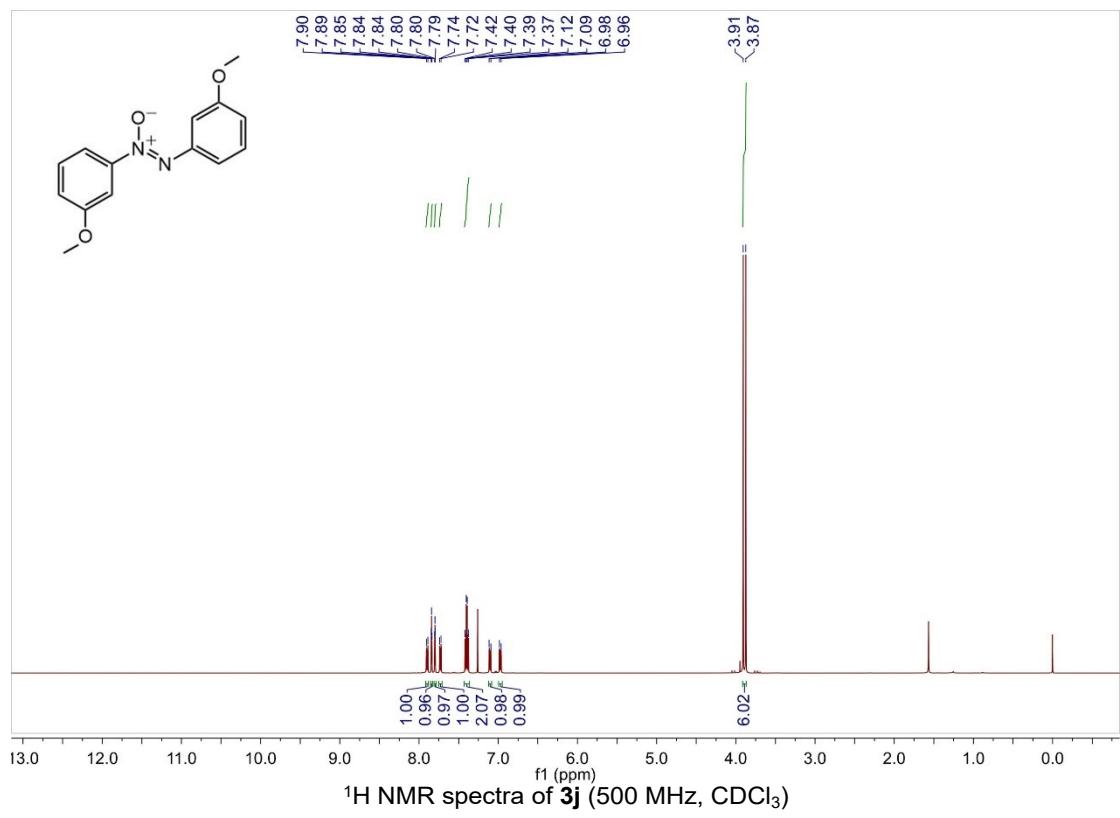


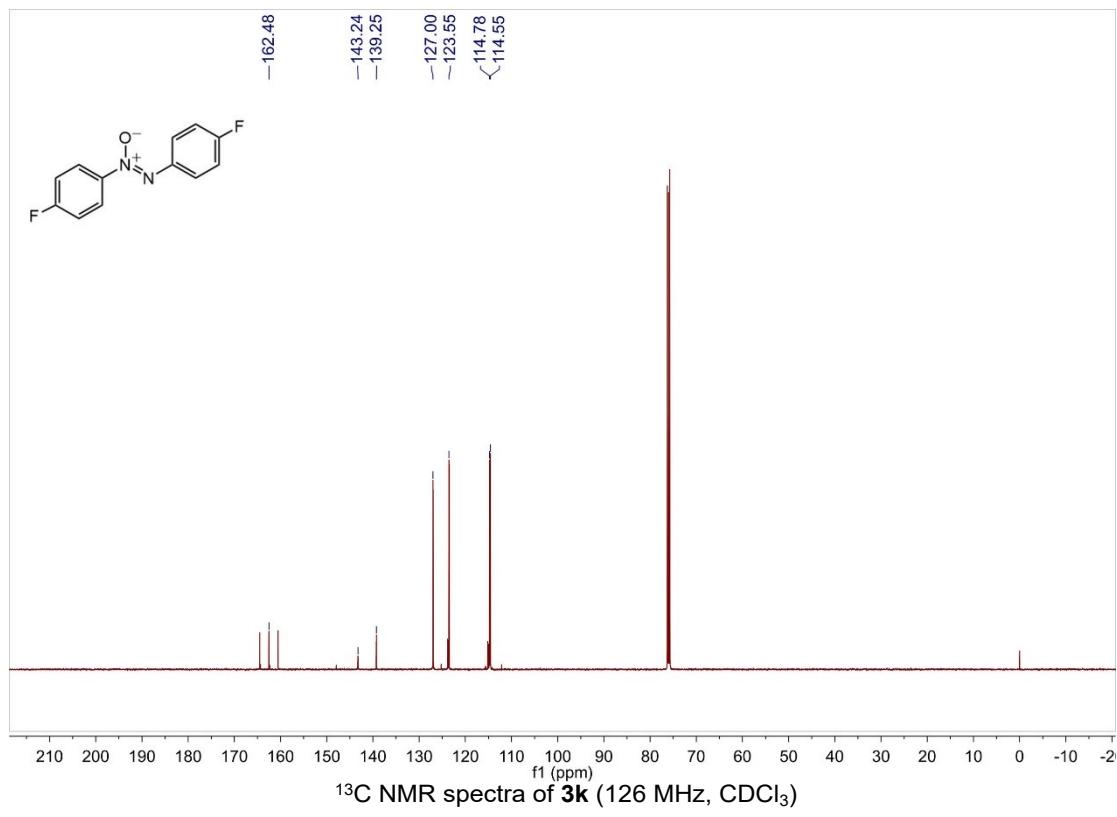
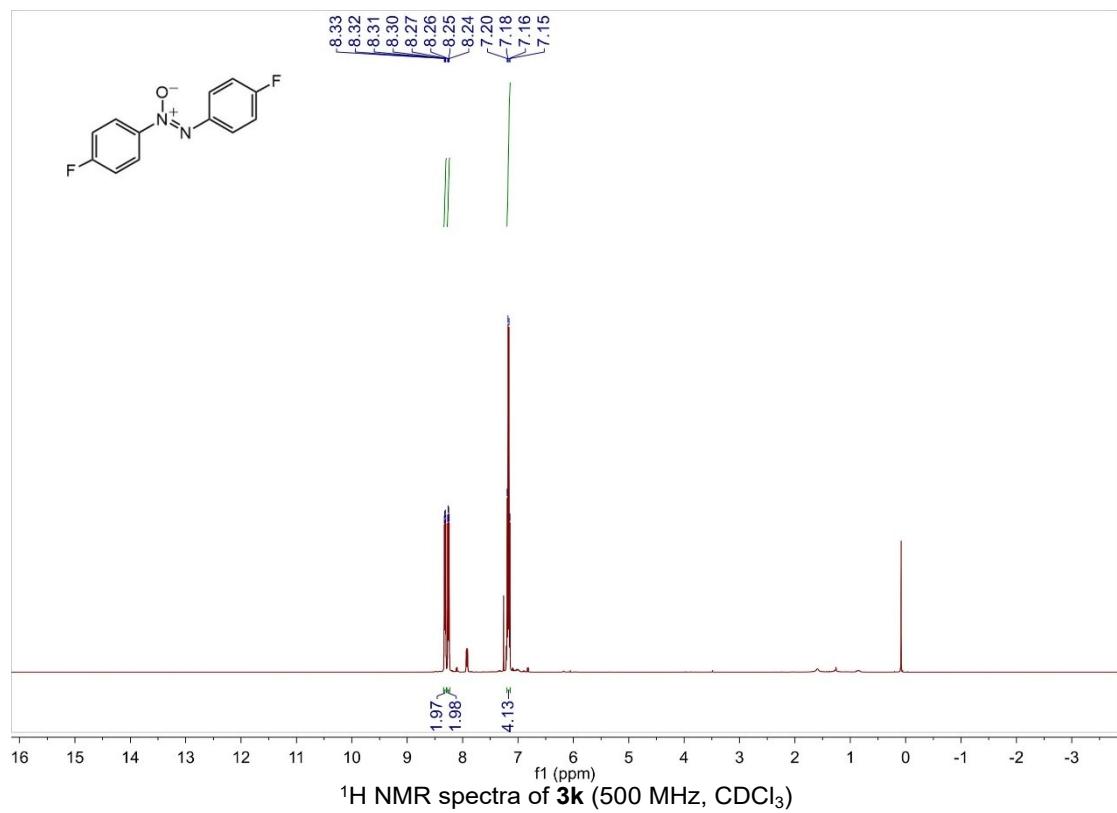


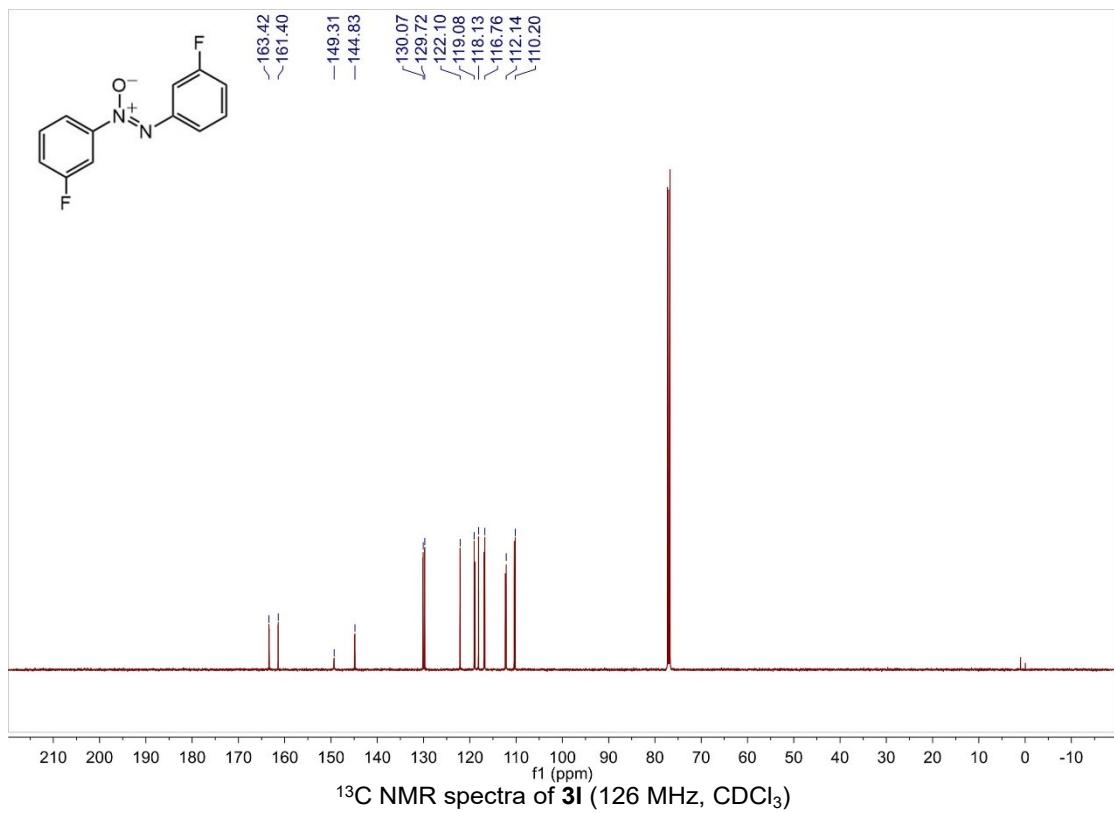
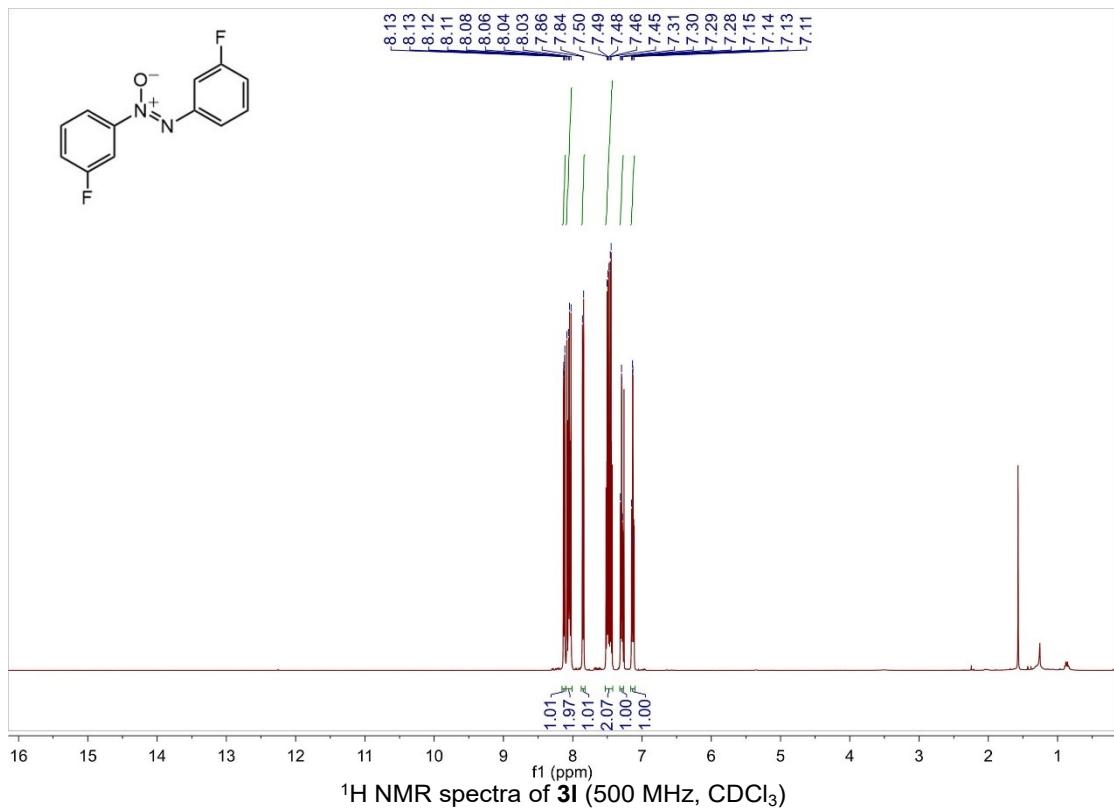
¹³C NMR spectra of **3g** (126 MHz, CDCl₃)

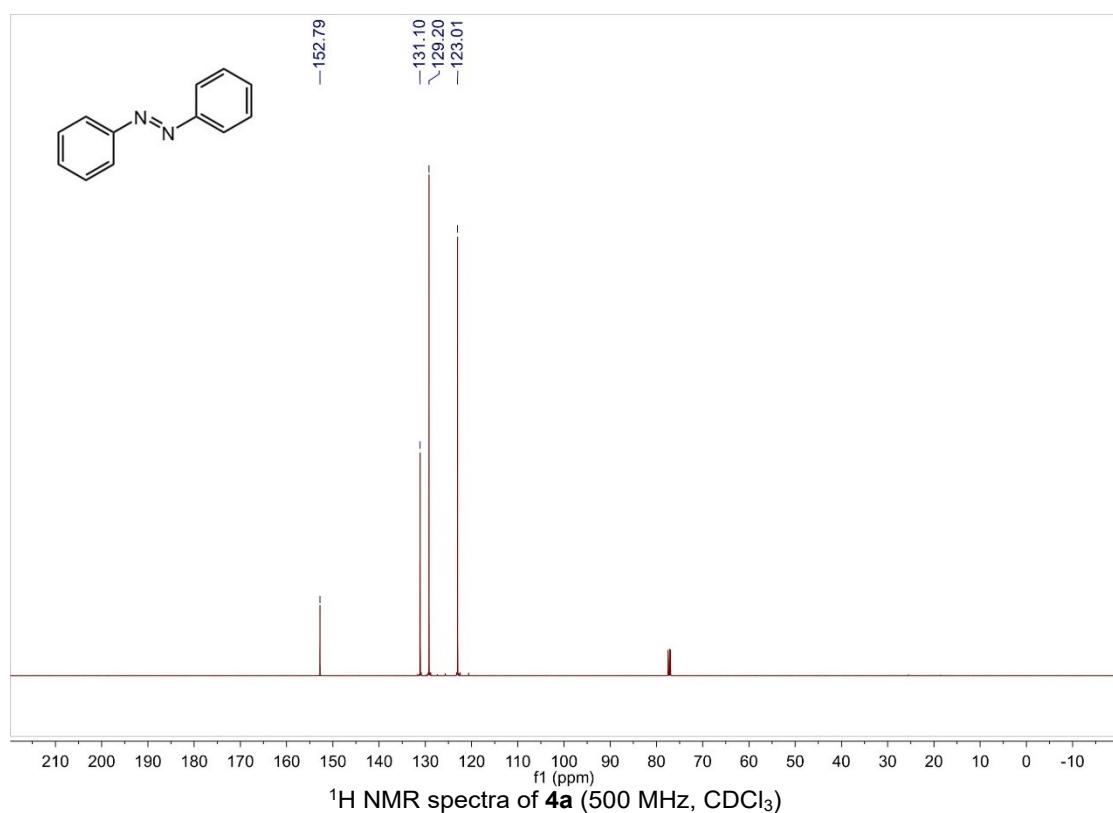
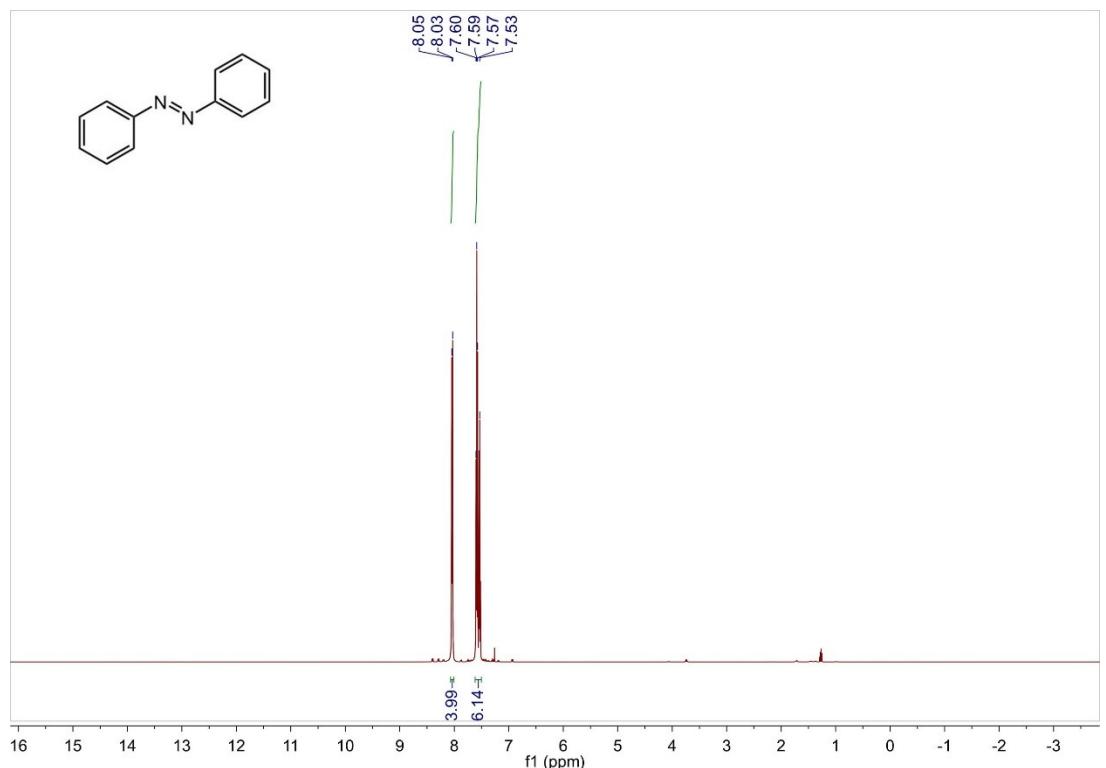




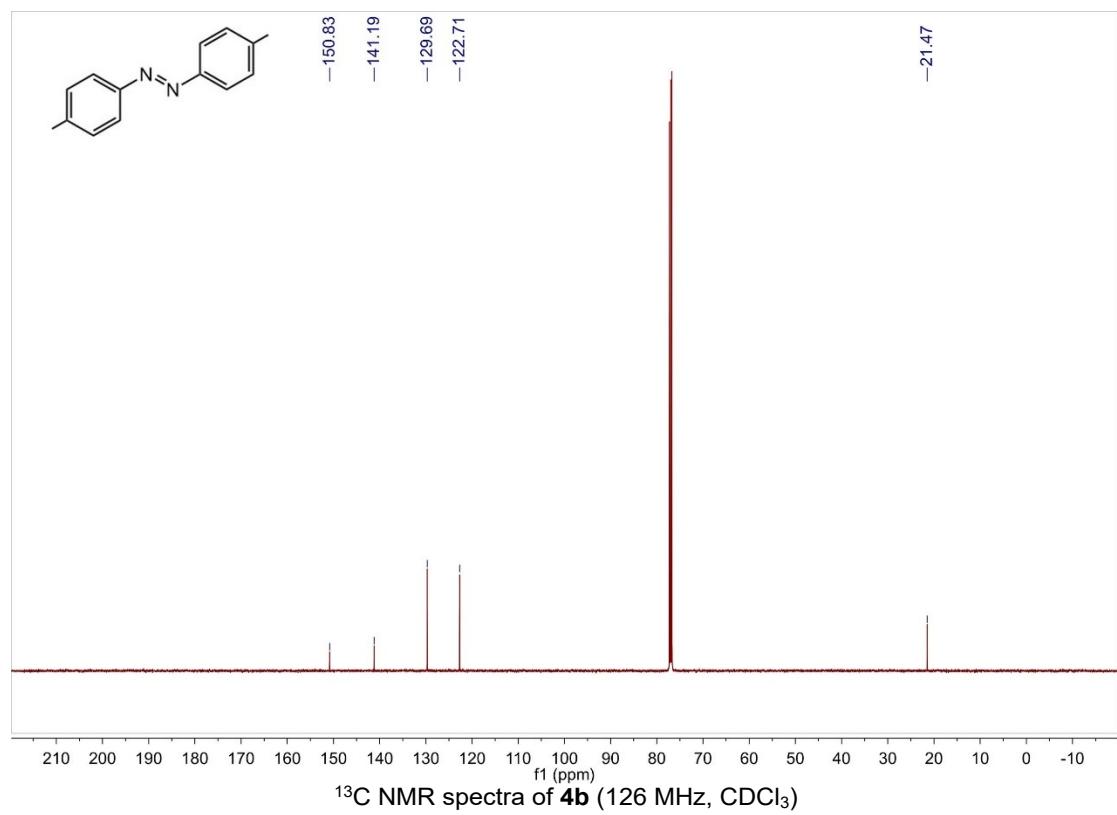
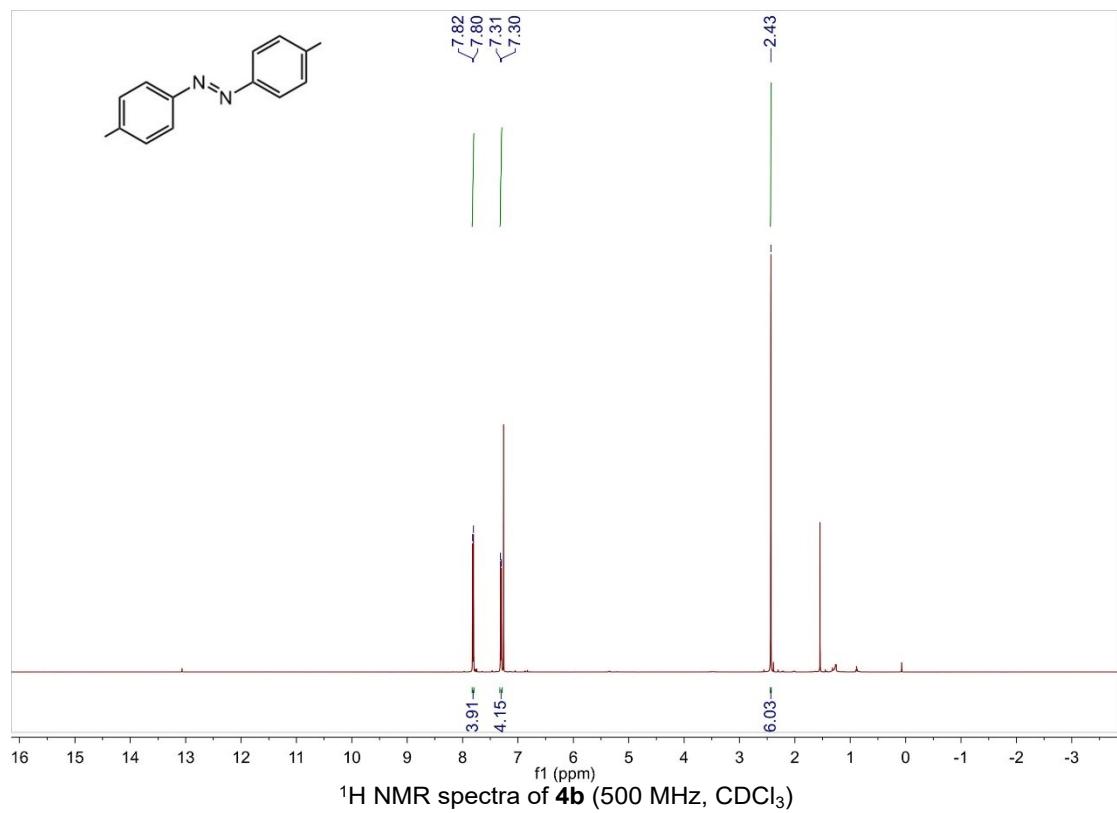






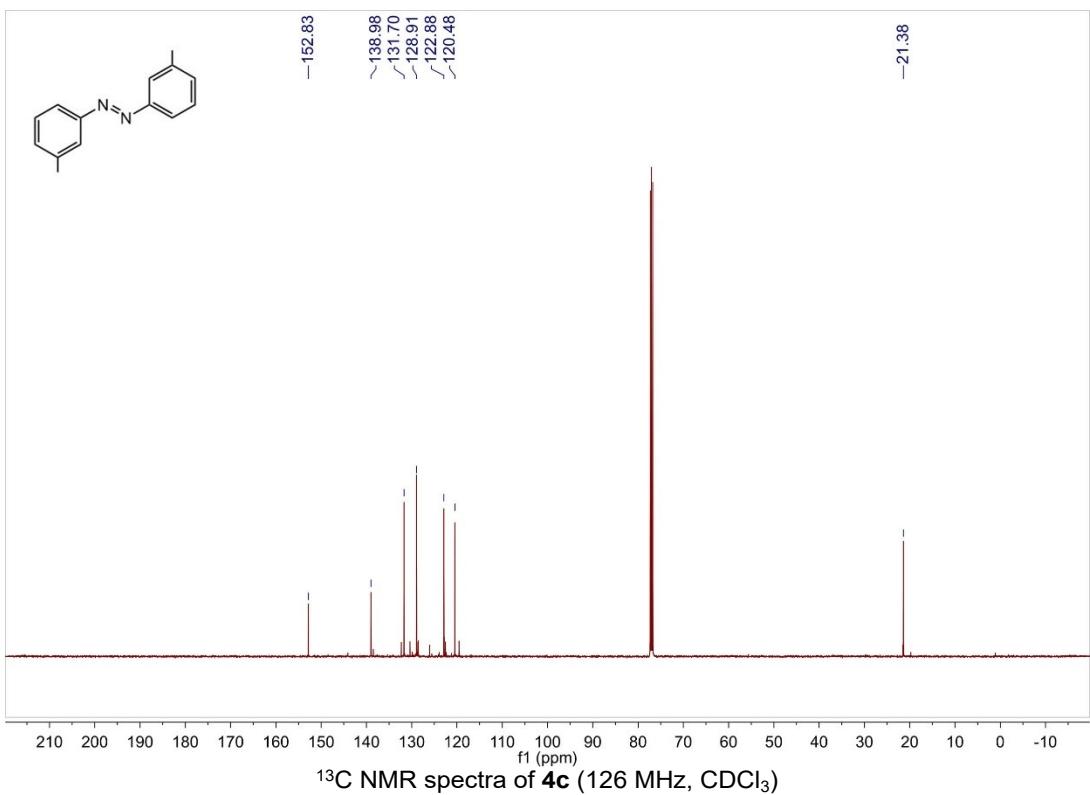


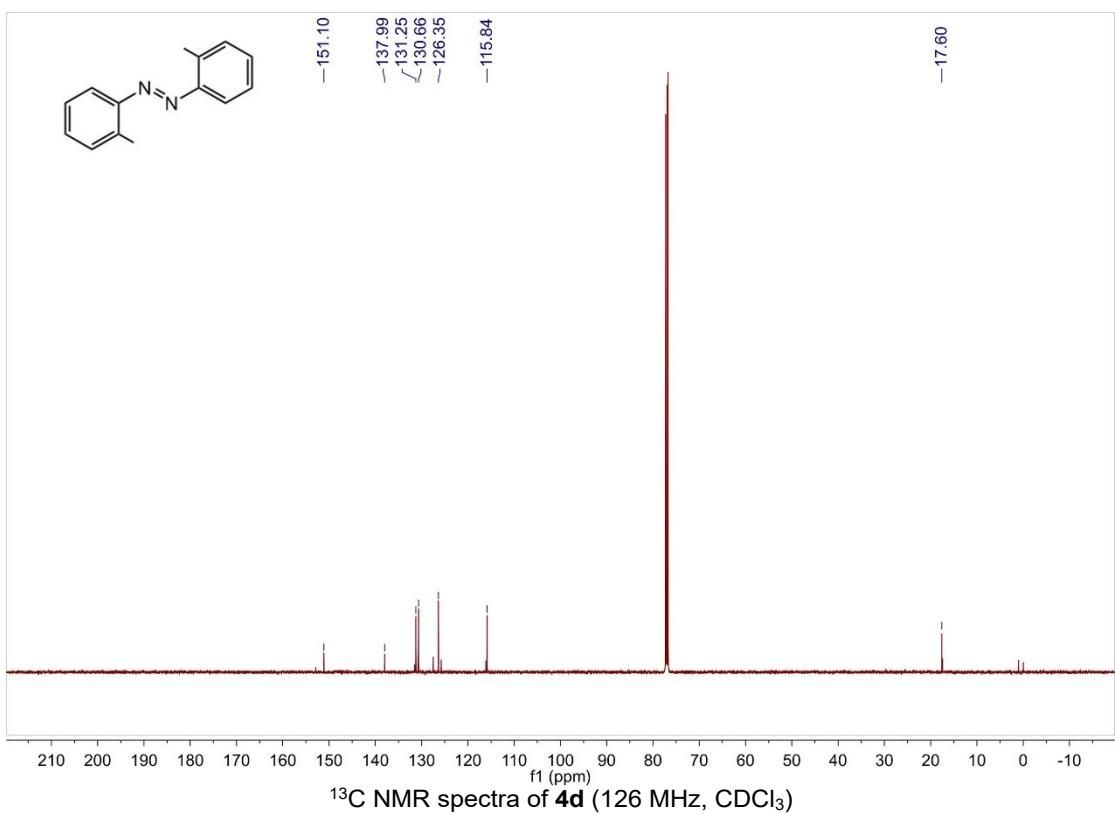
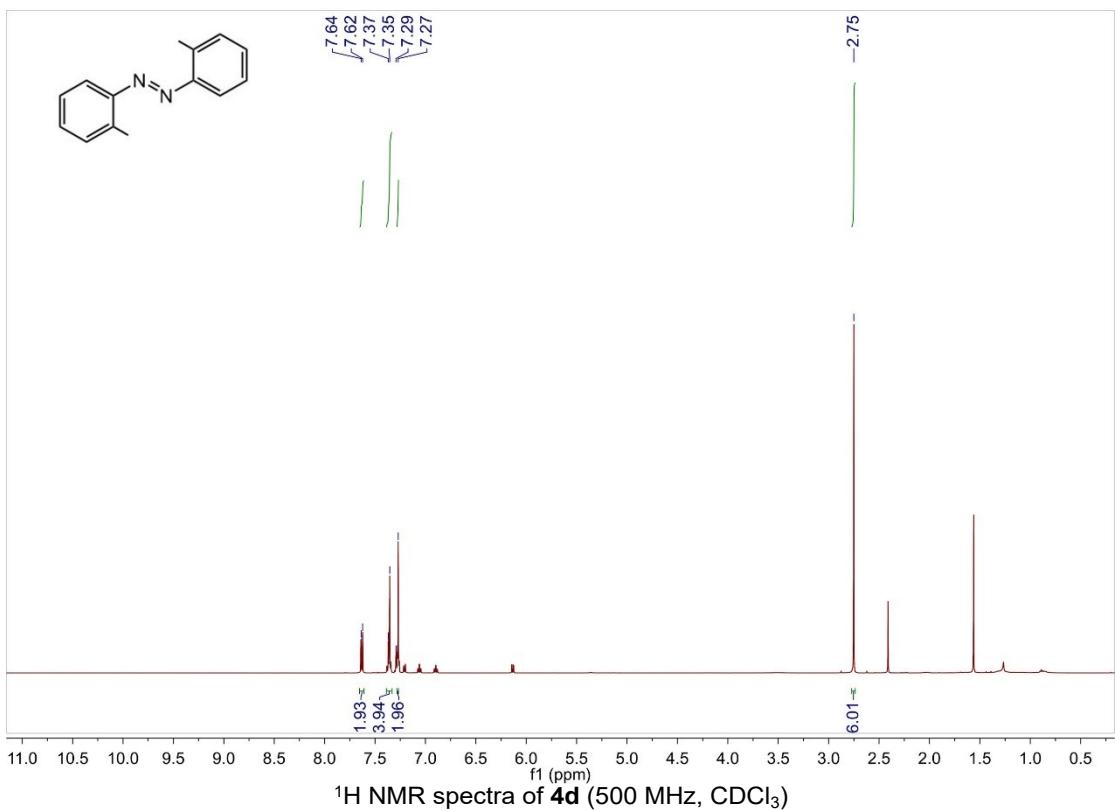
¹³C NMR spectra of **4a** (126 MHz, CDCl₃)

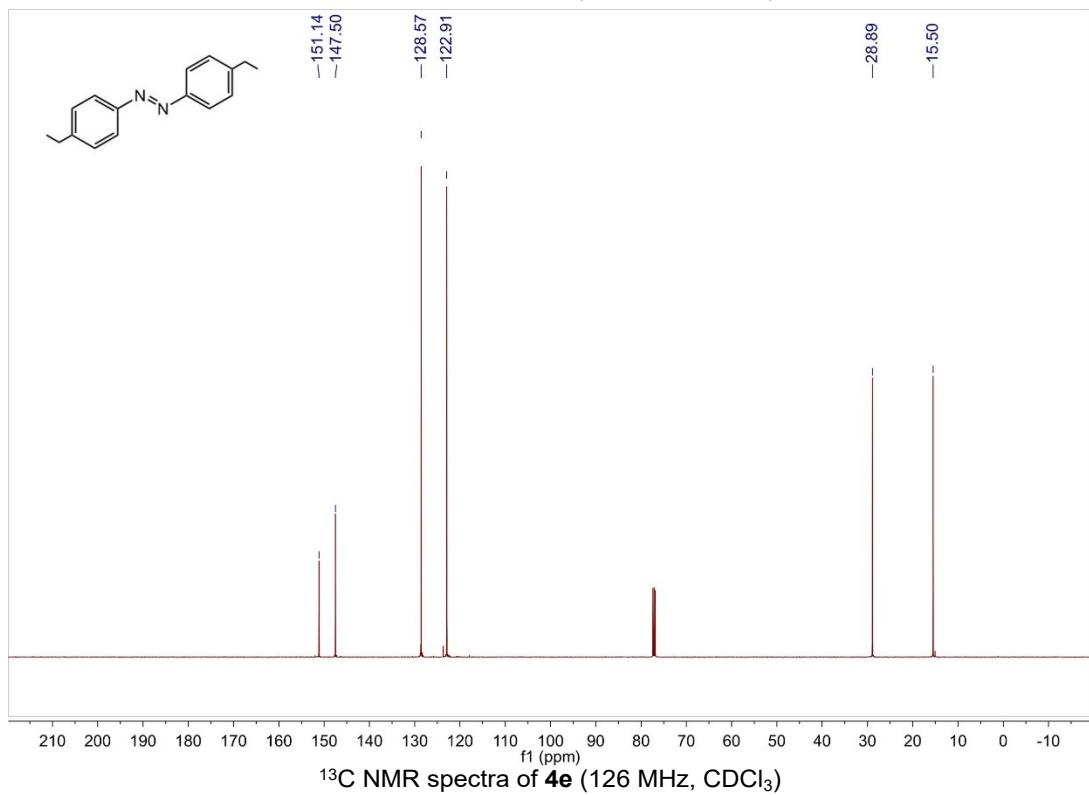
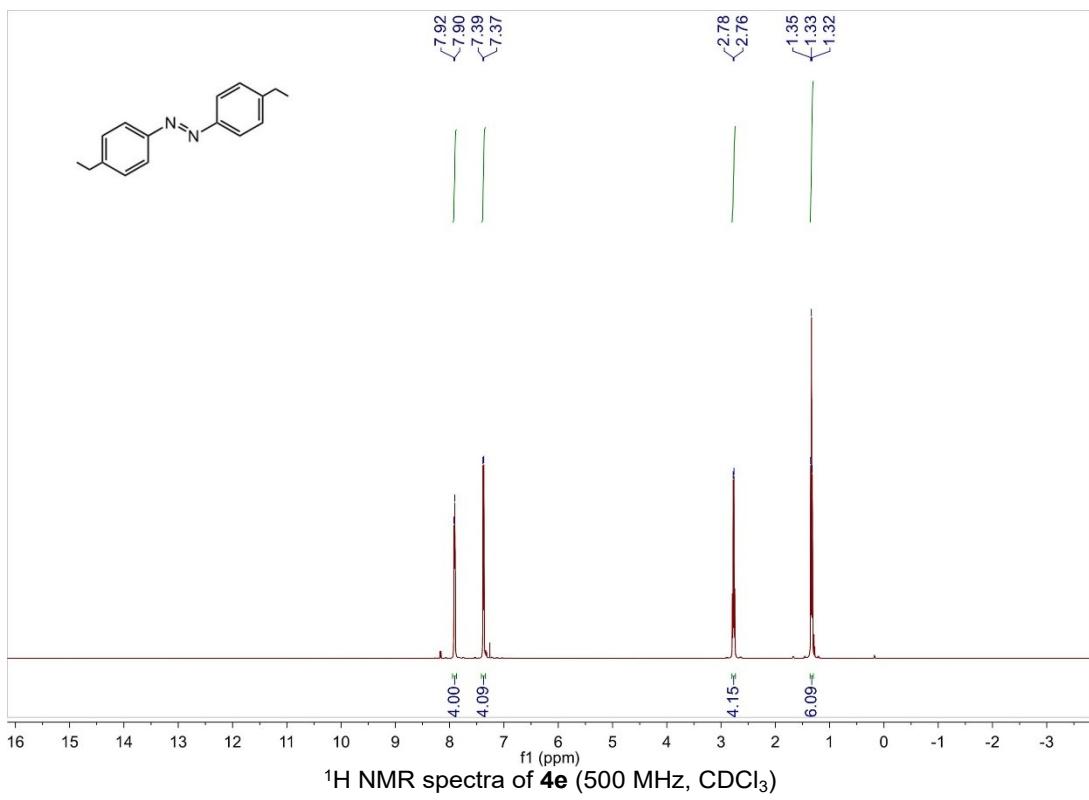


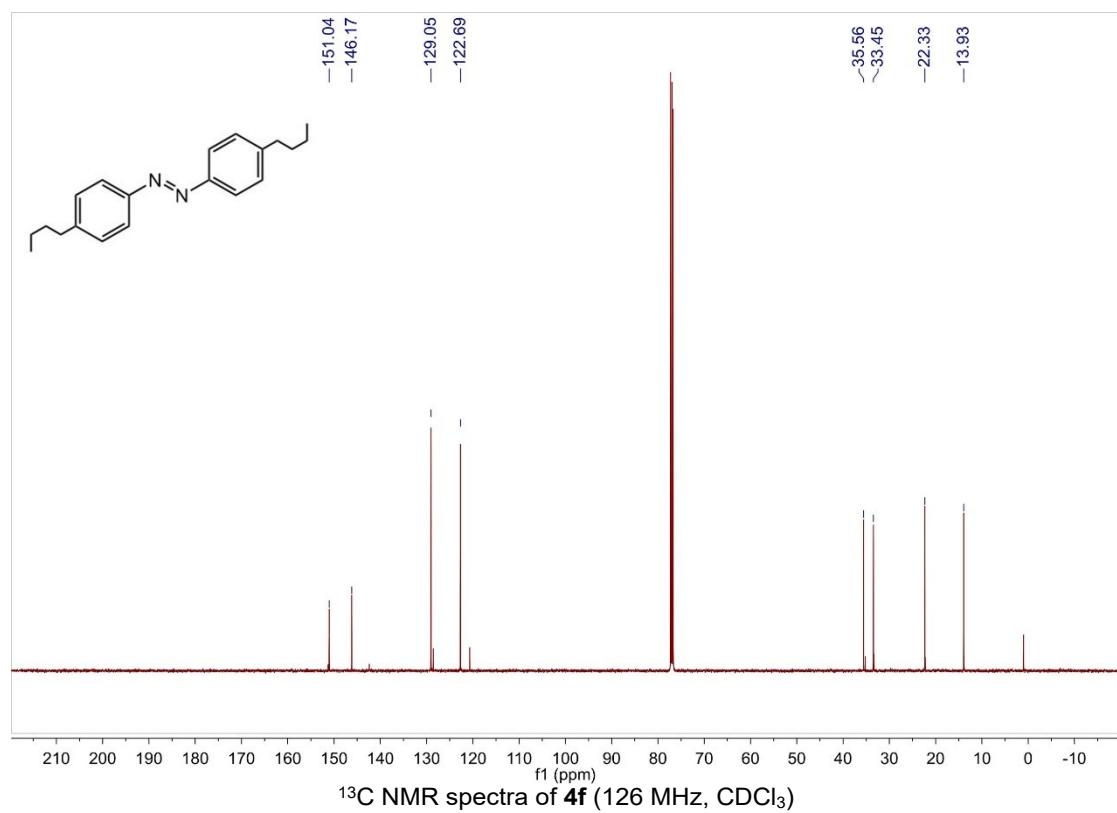
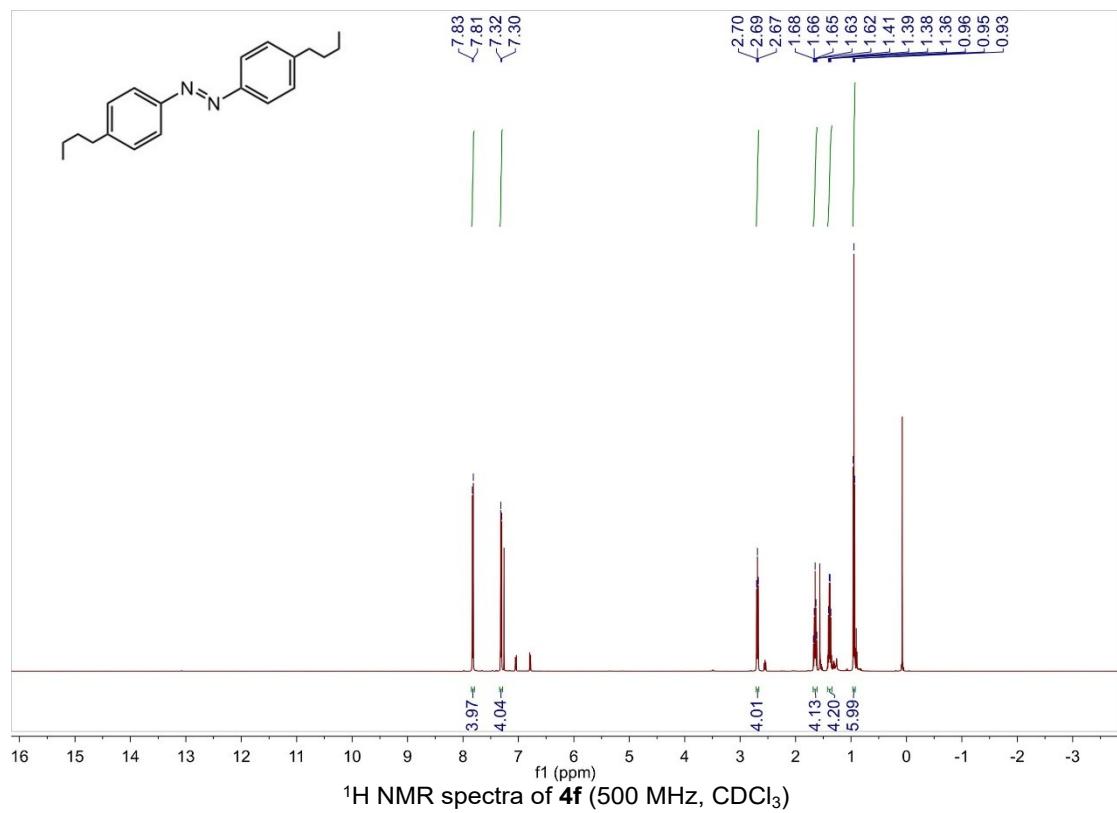


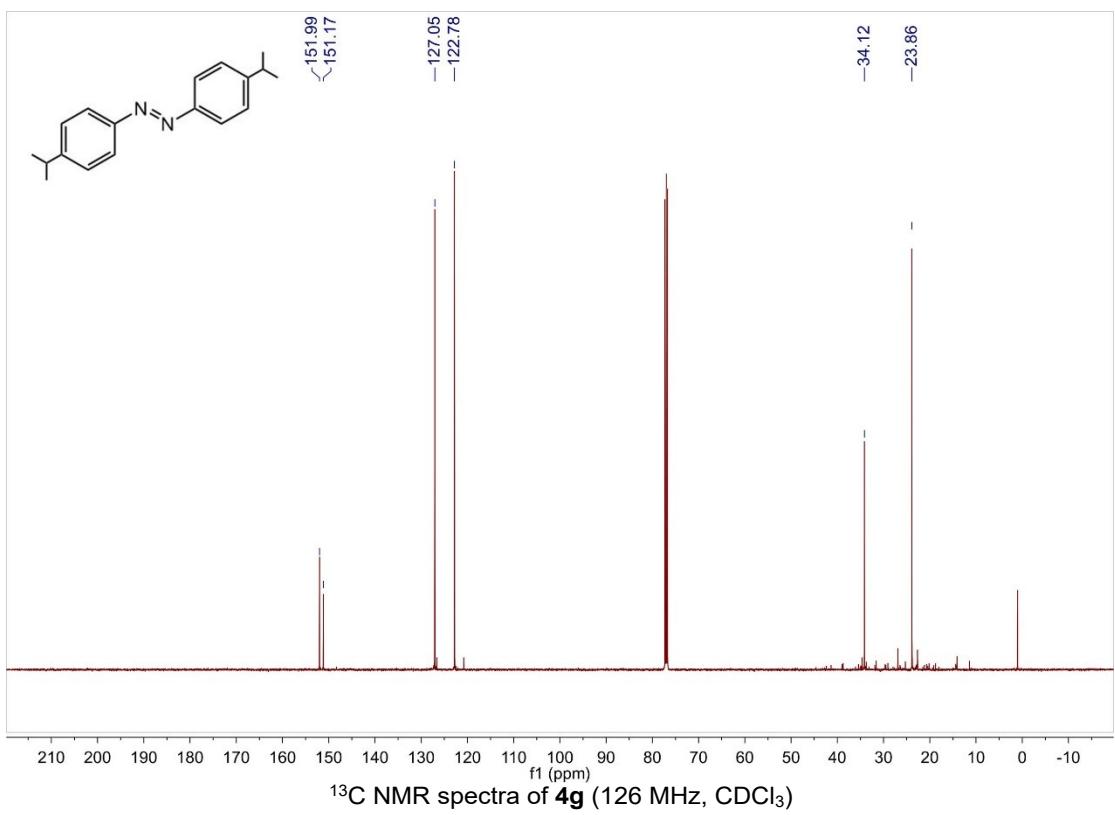
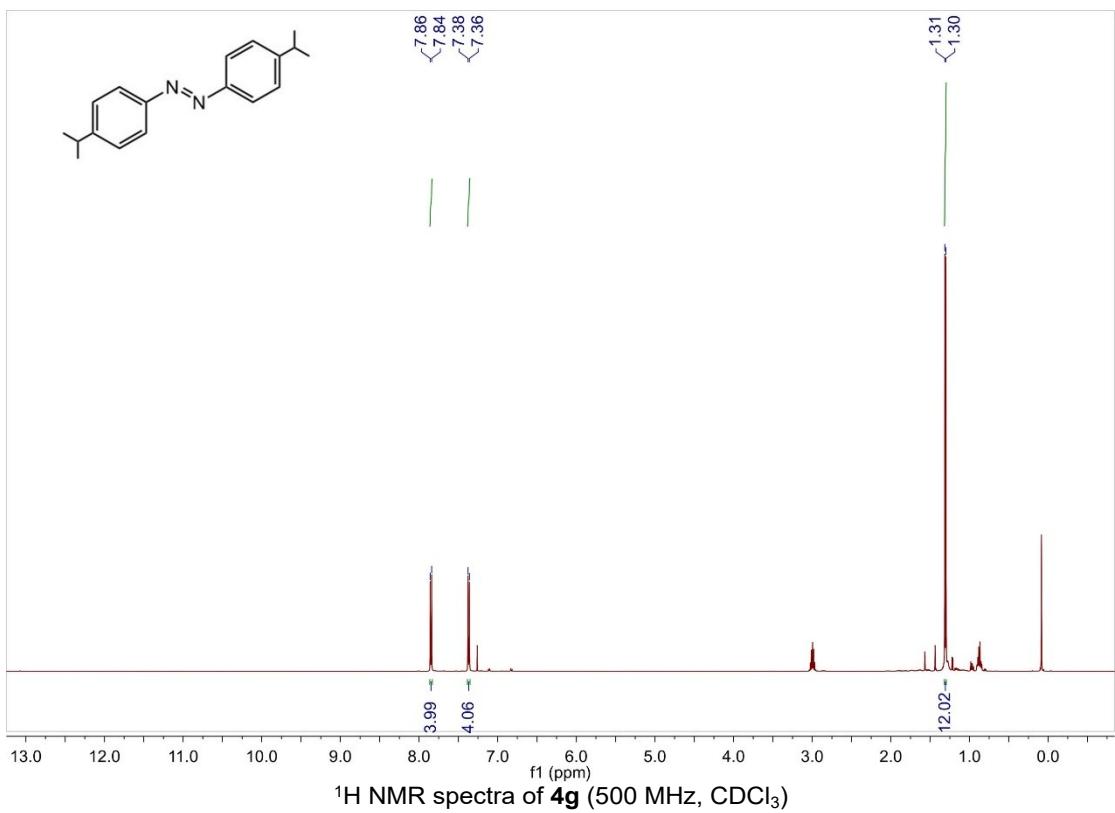
¹H NMR spectra of **4c** (500 MHz, CDCl₃)

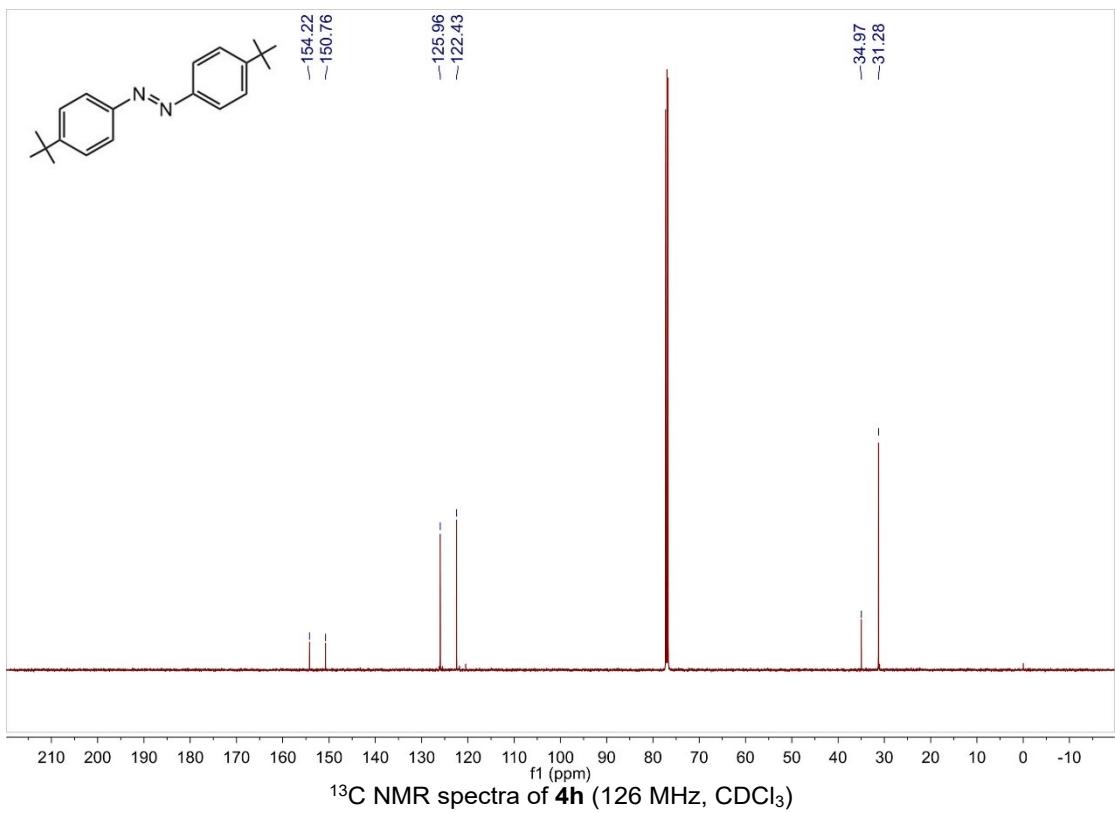
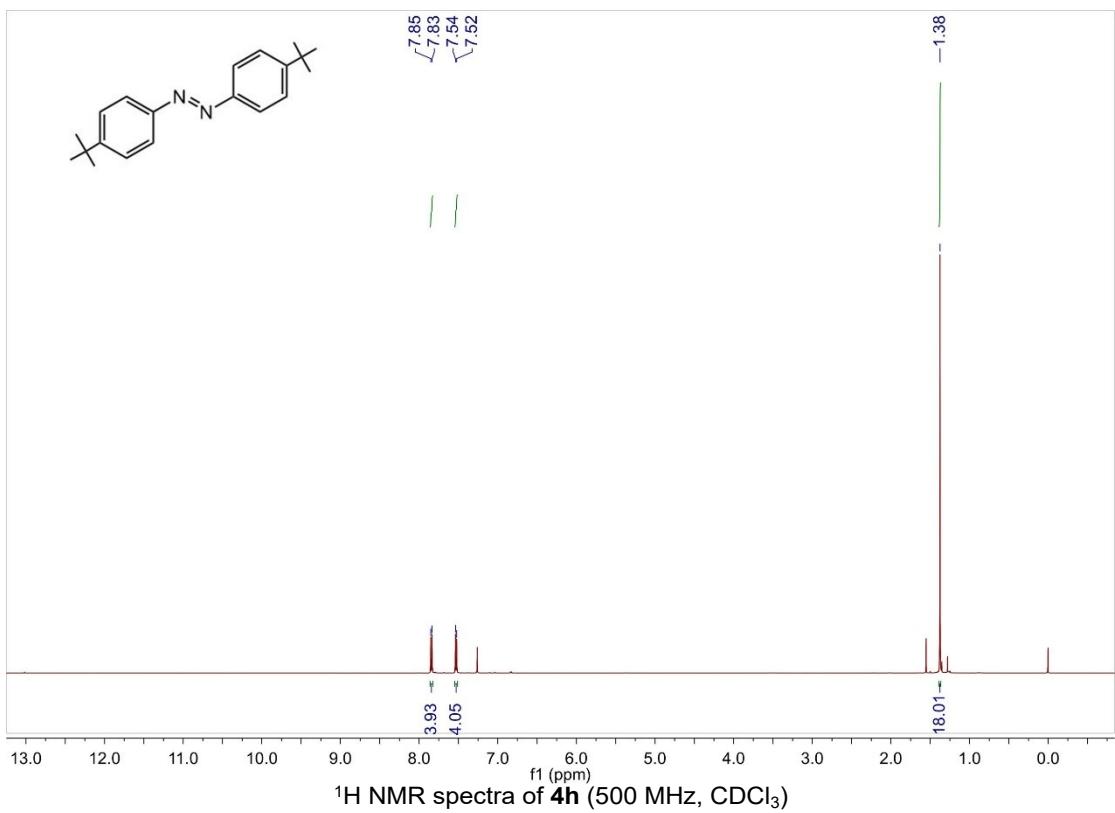


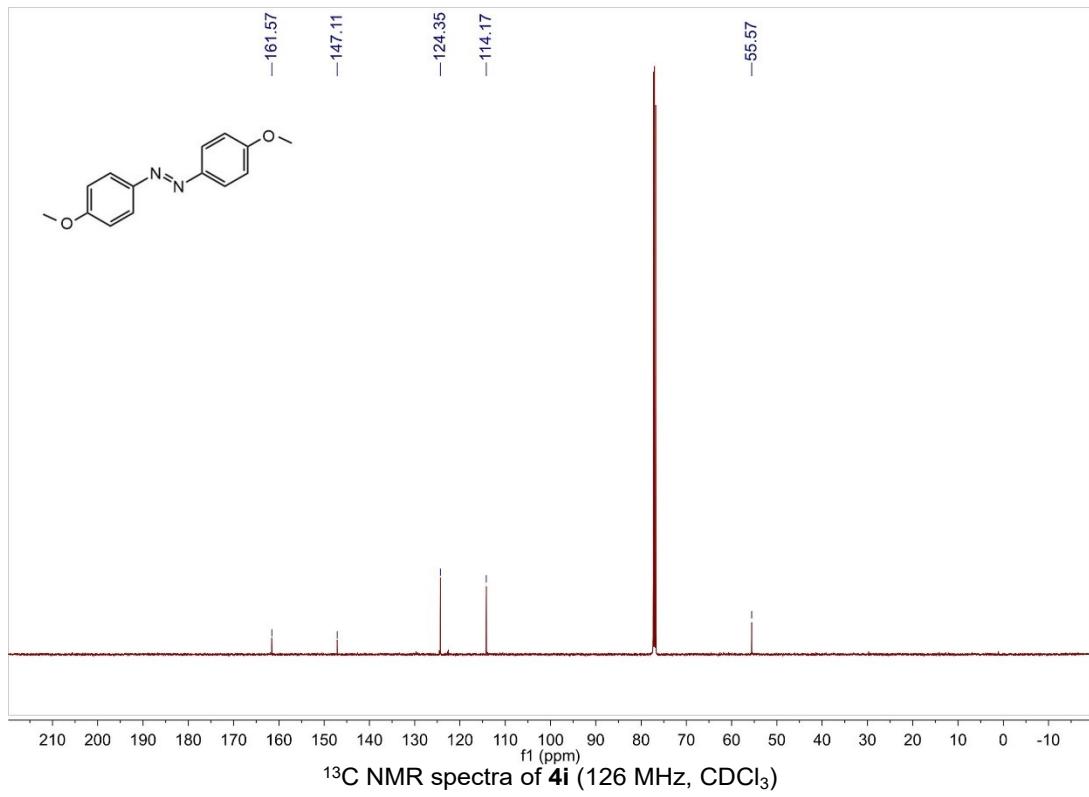
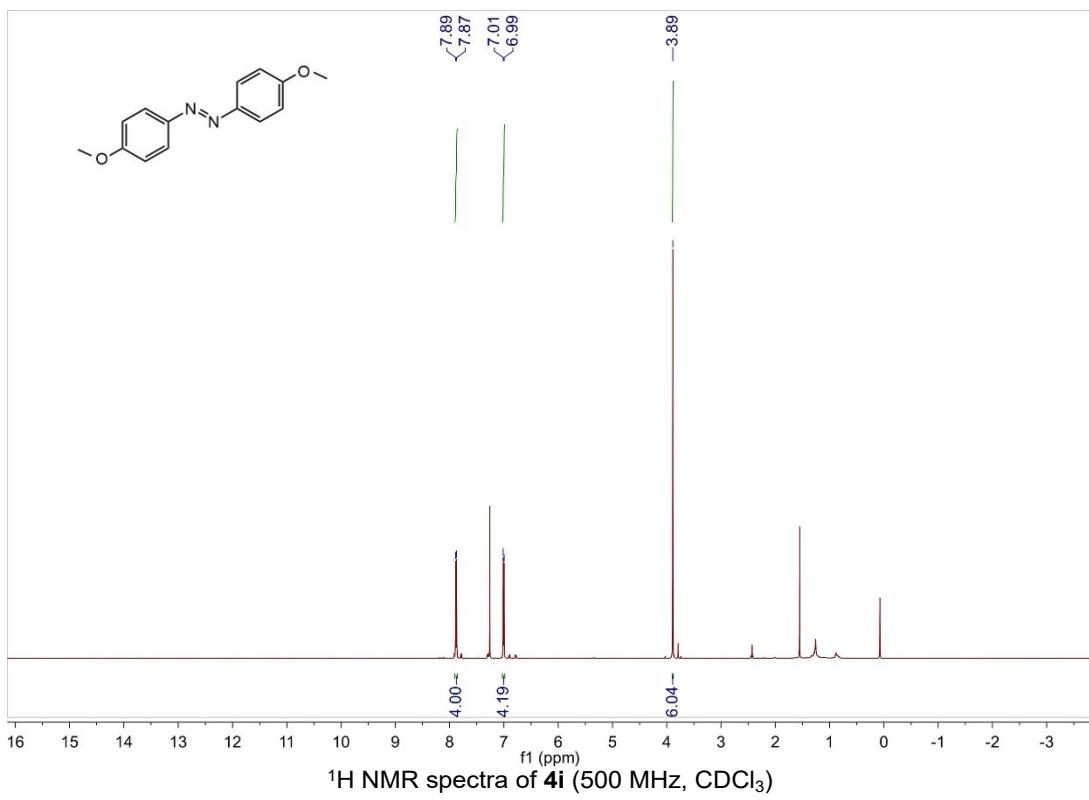


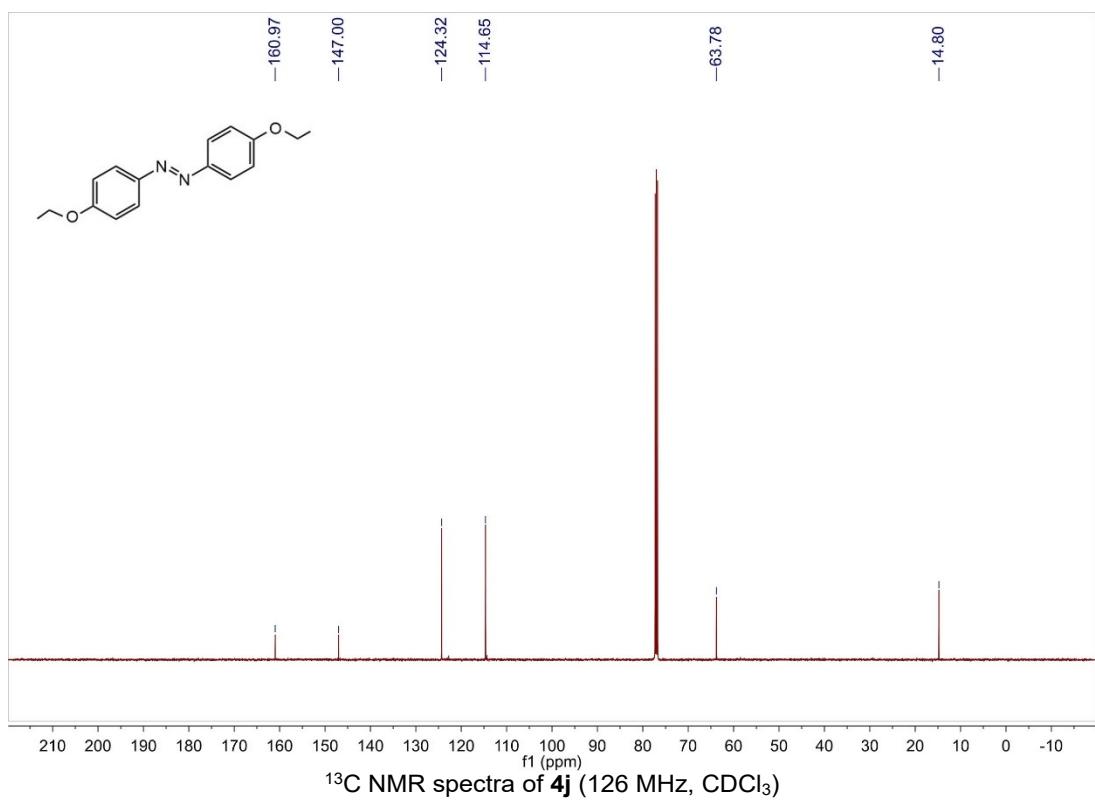
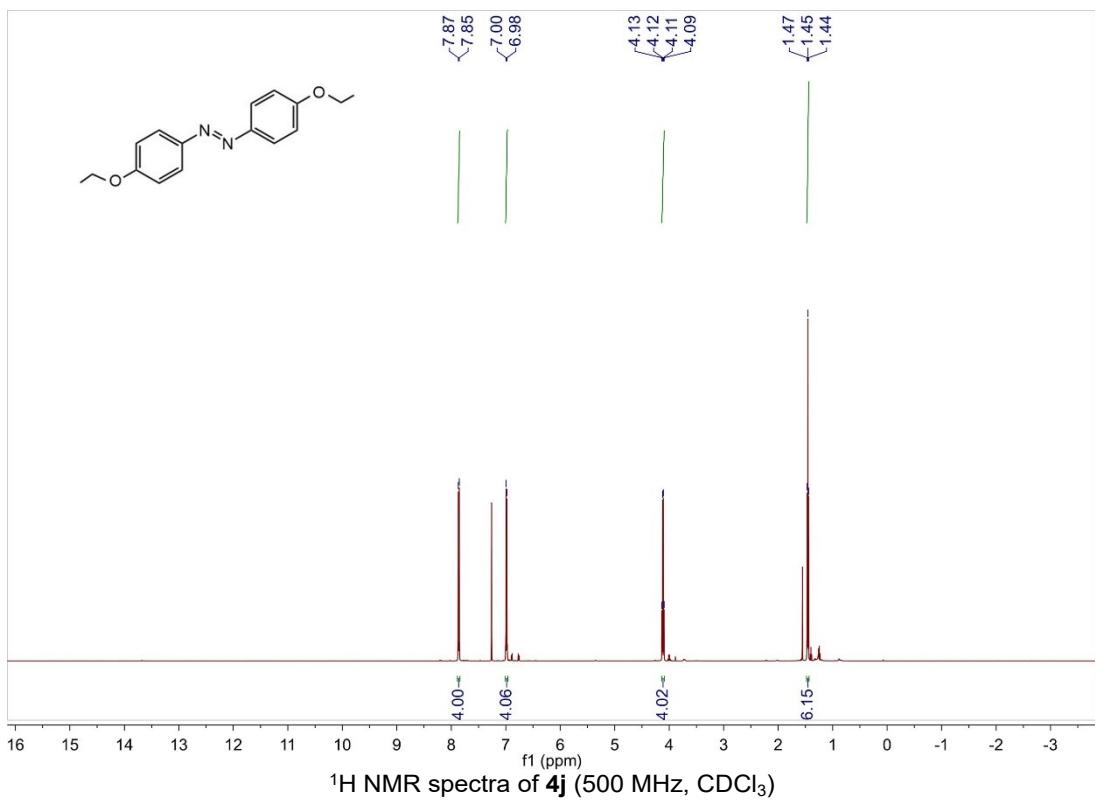


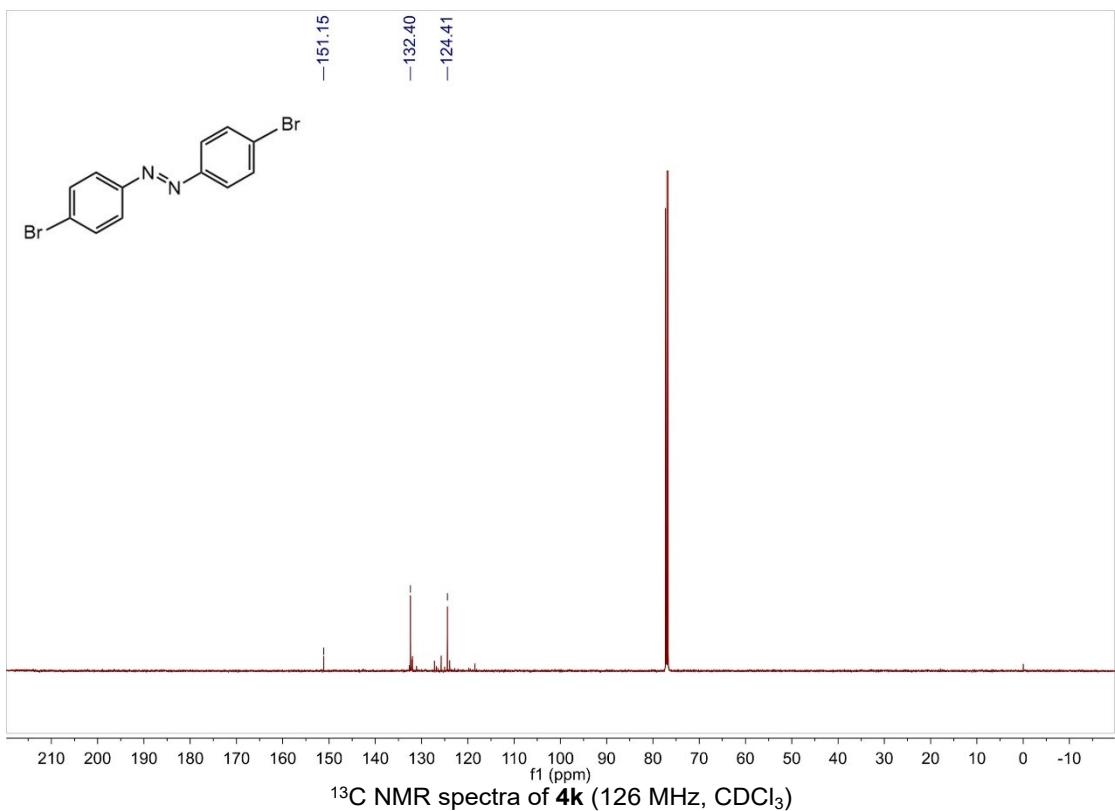
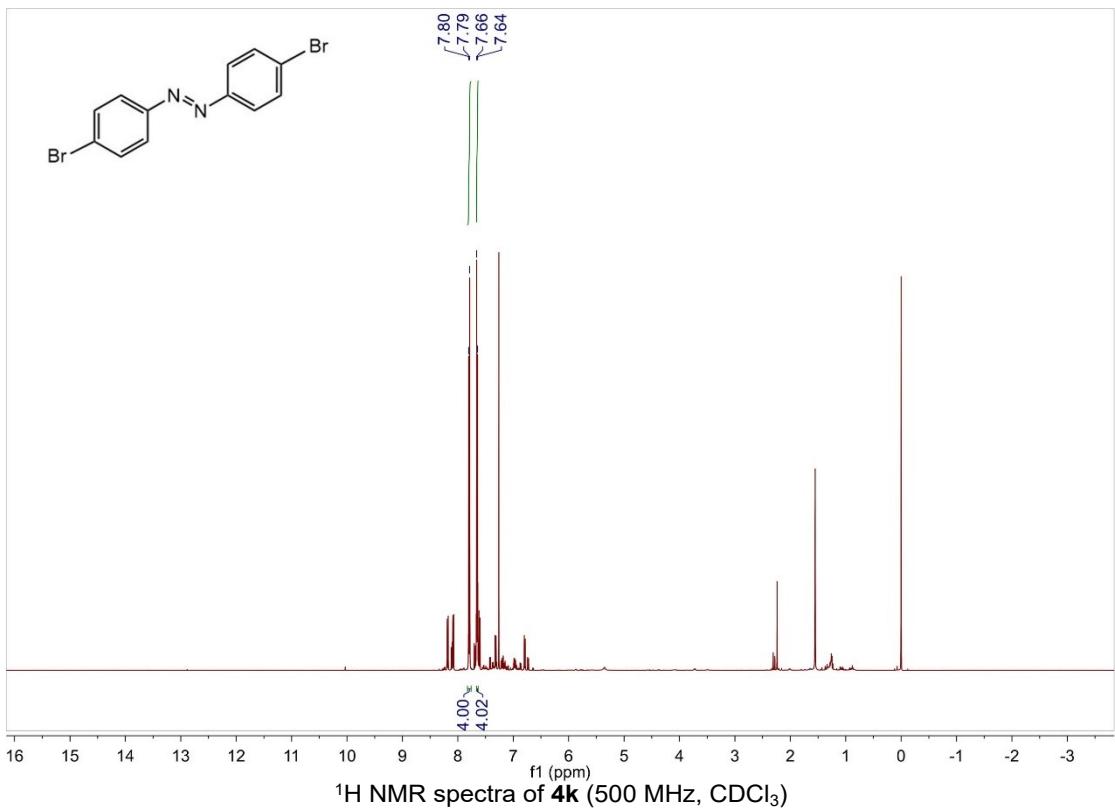


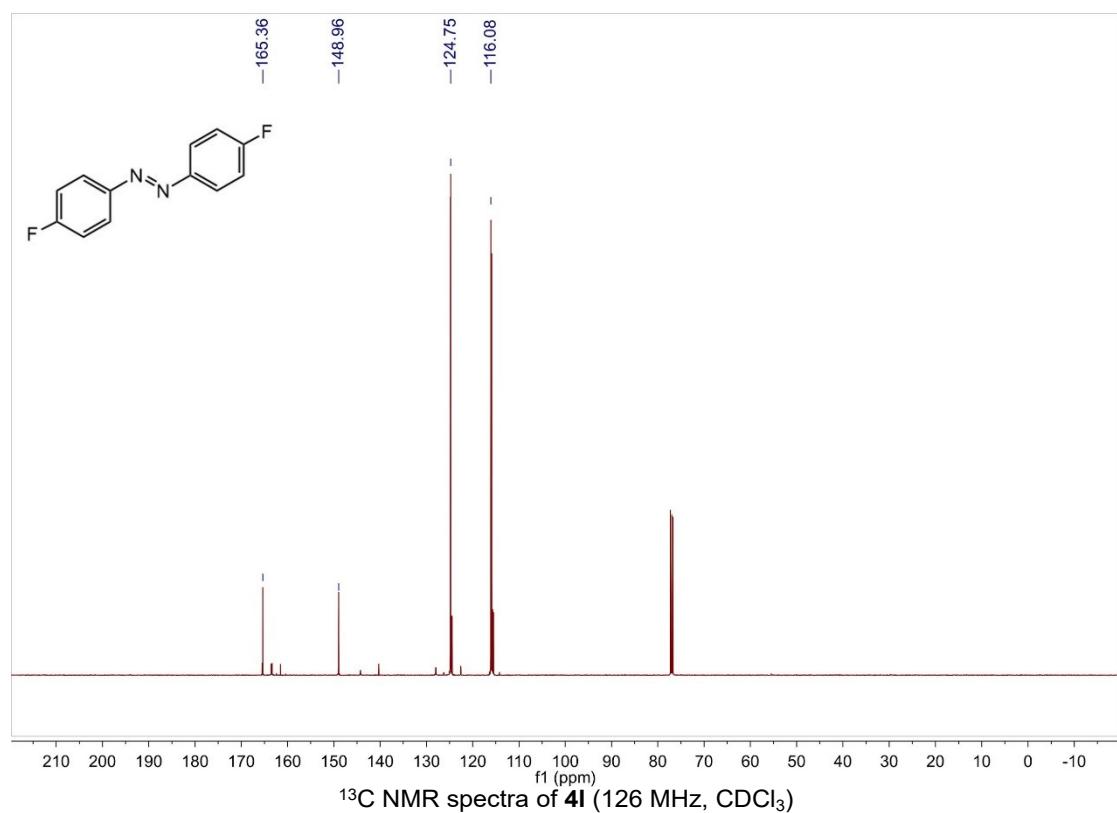
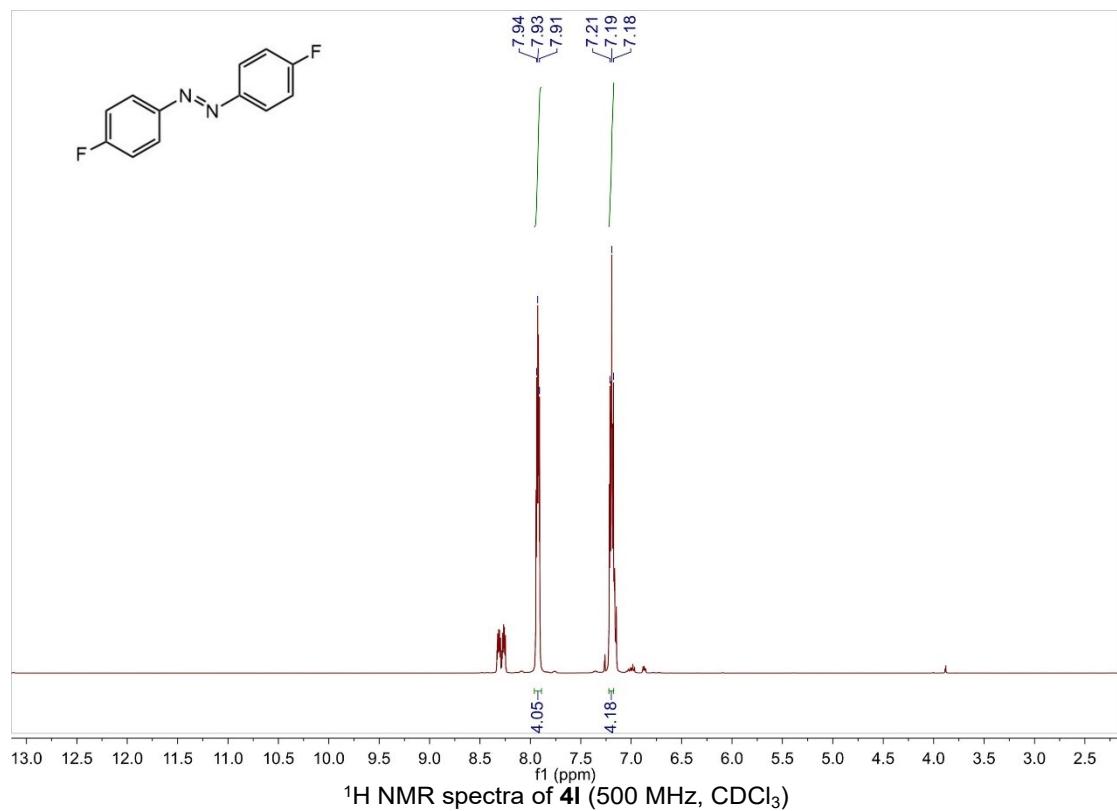


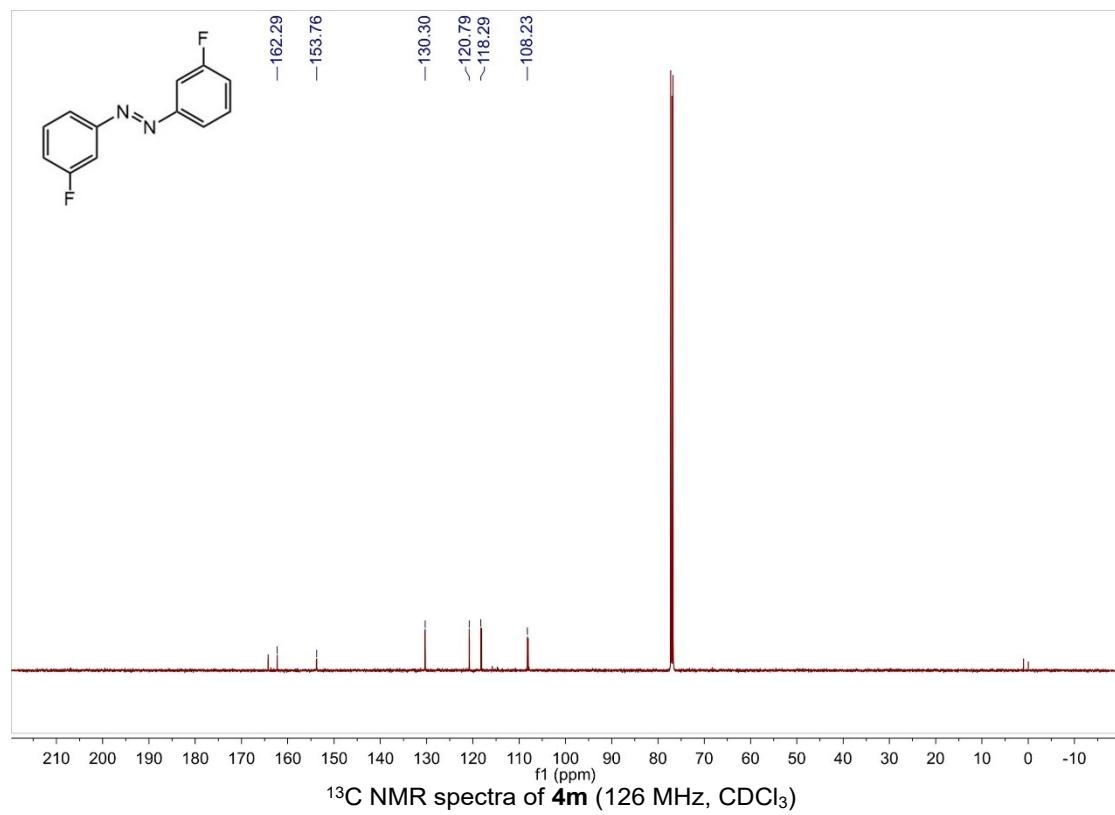
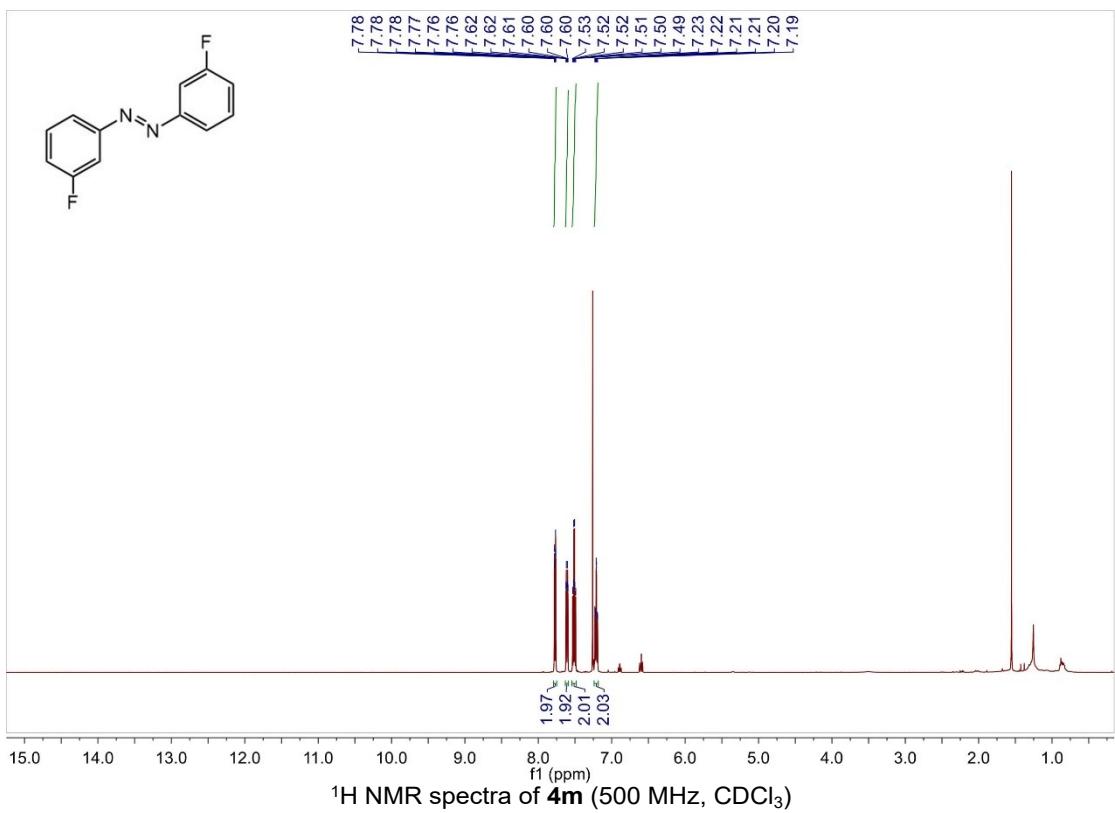


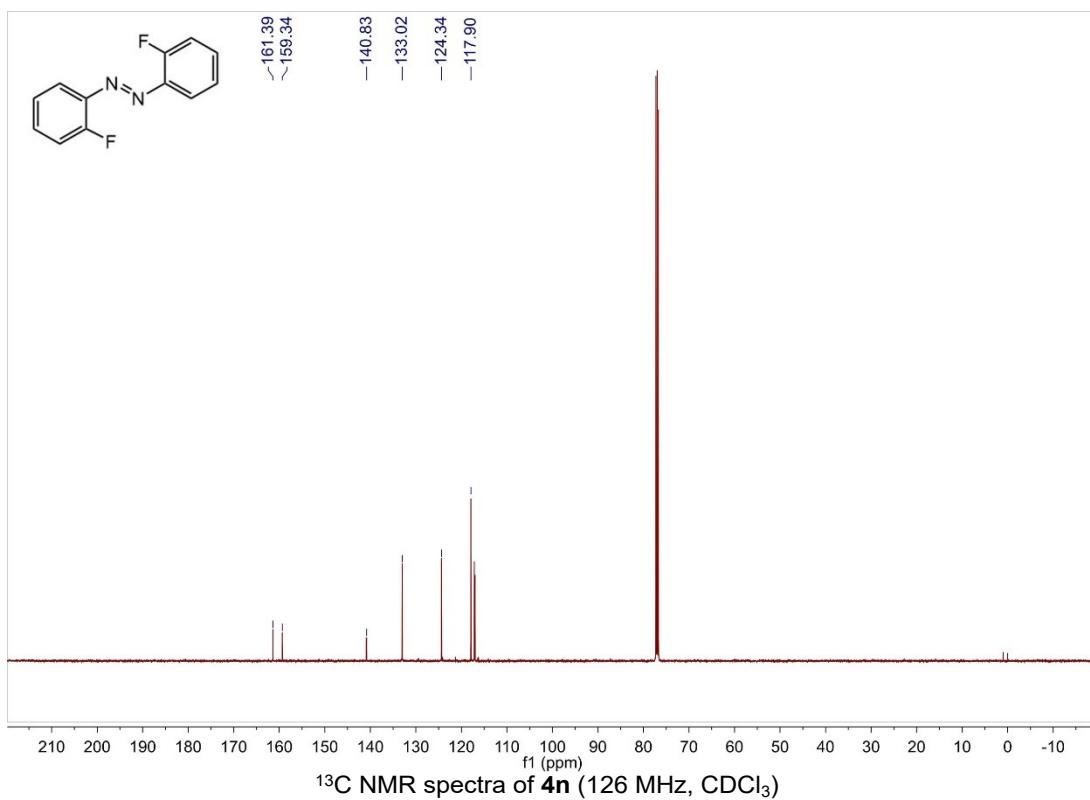
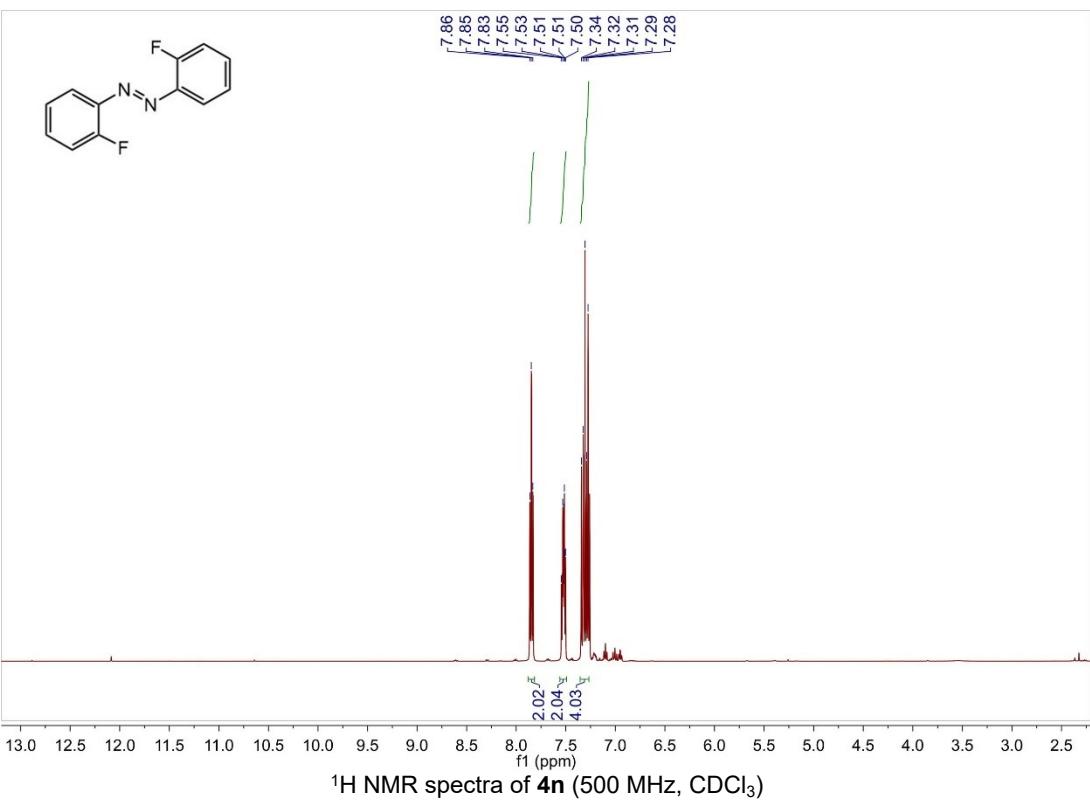


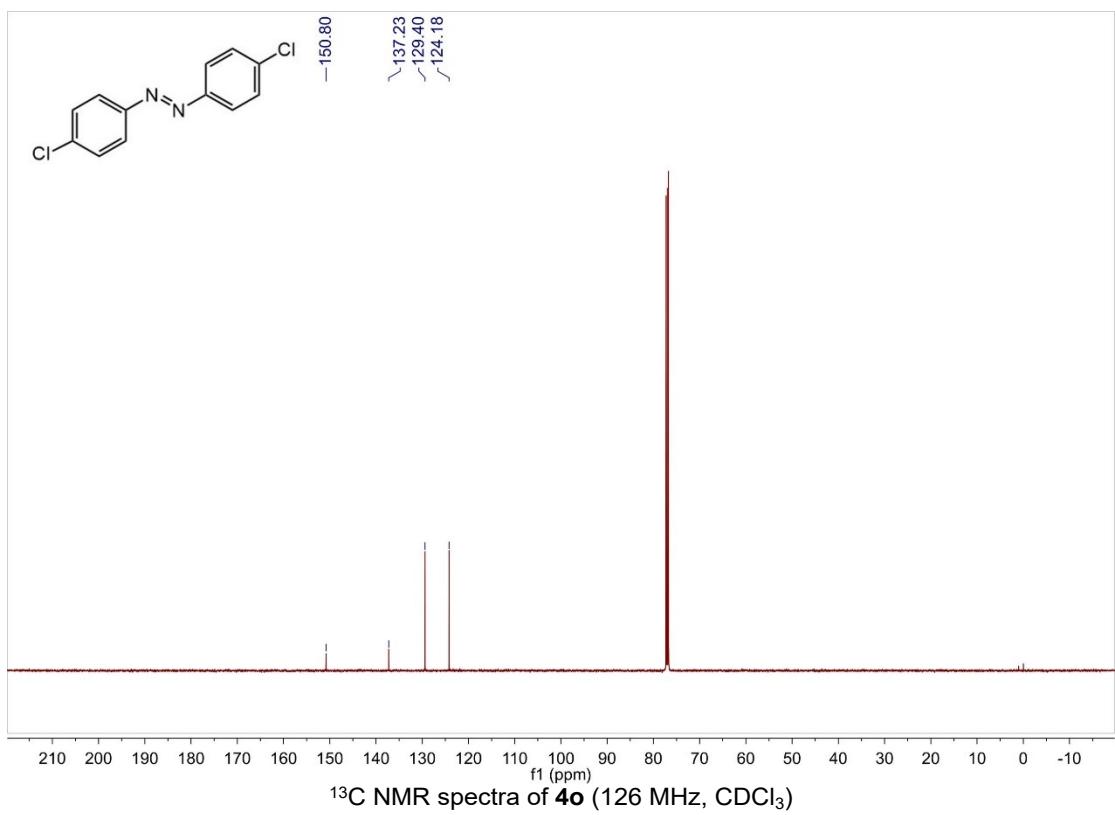
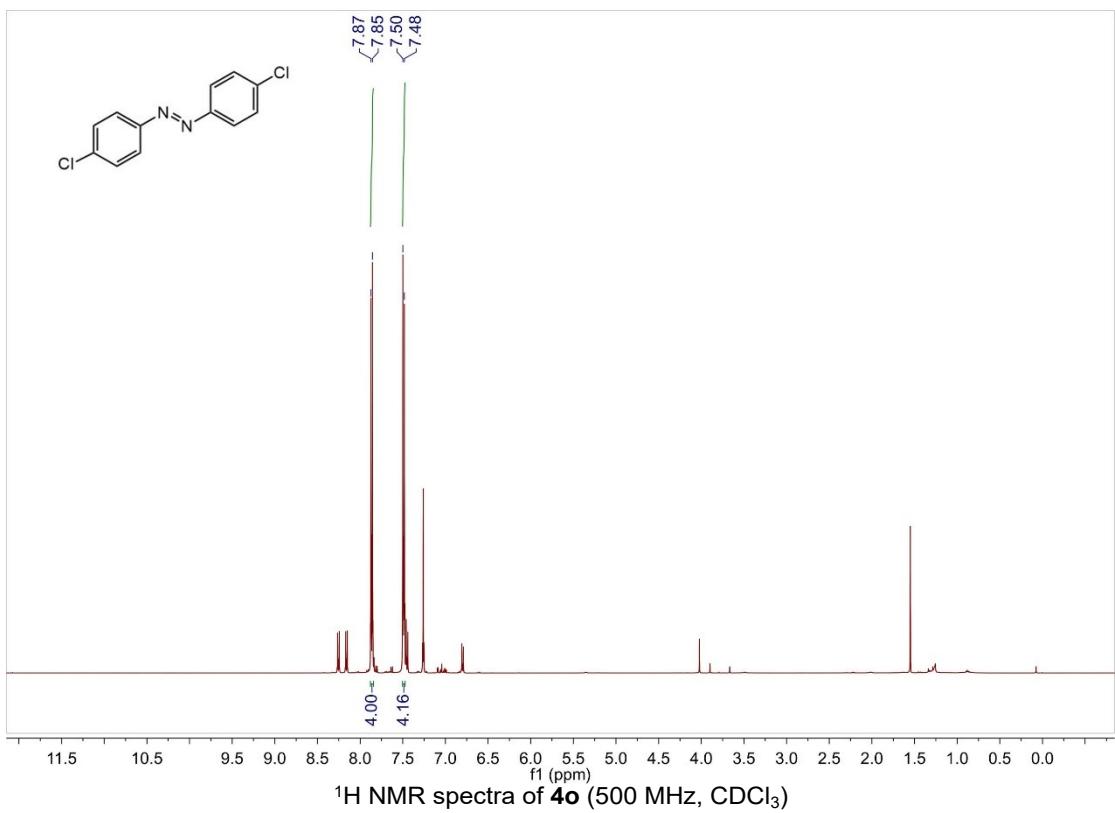


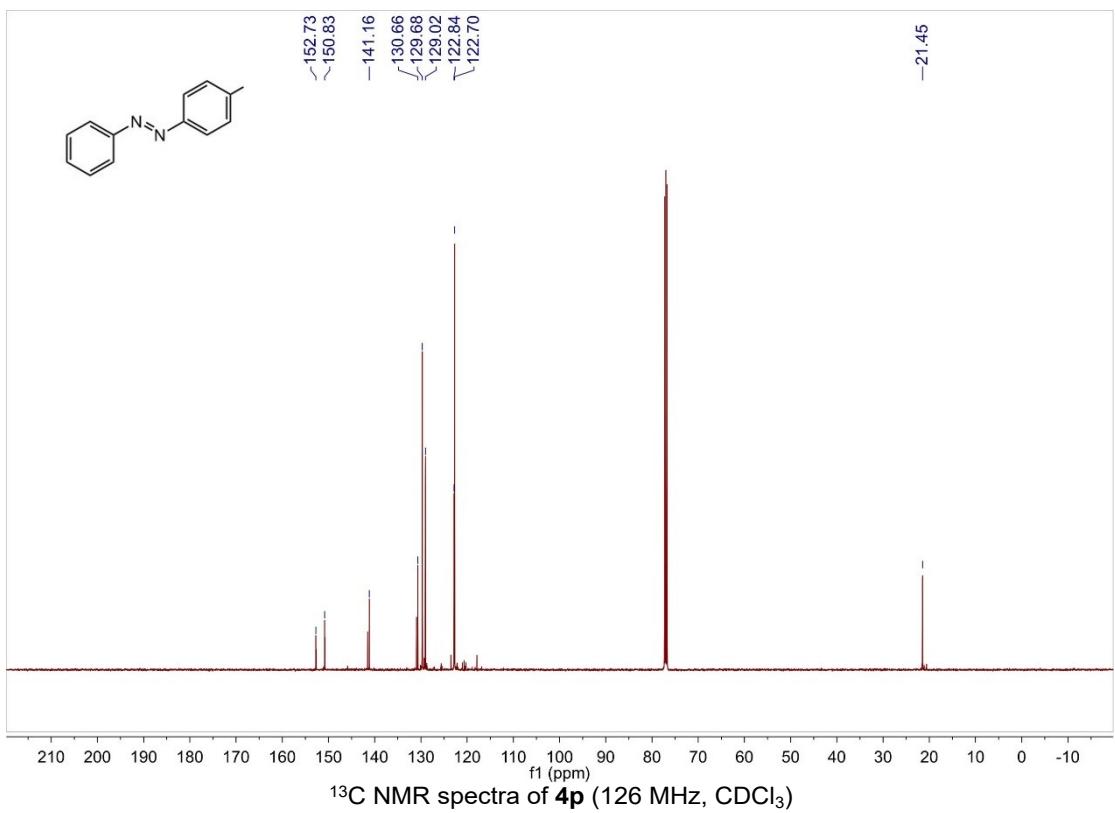
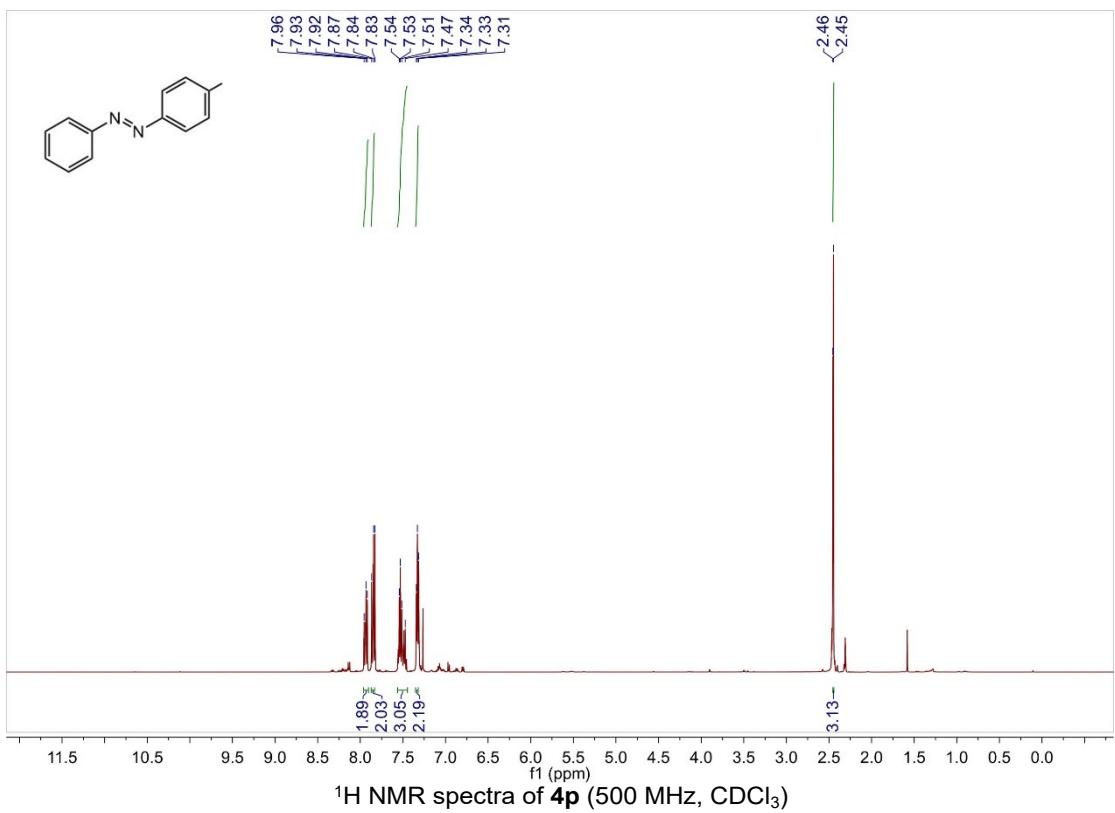


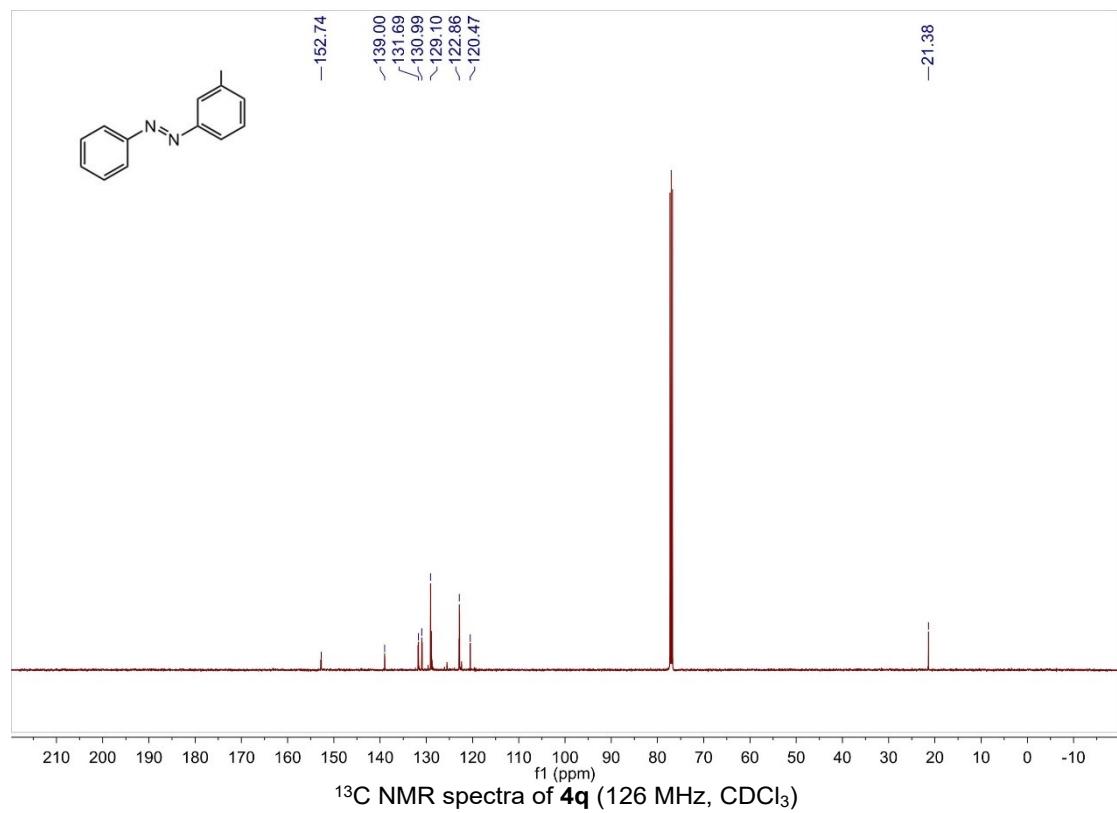
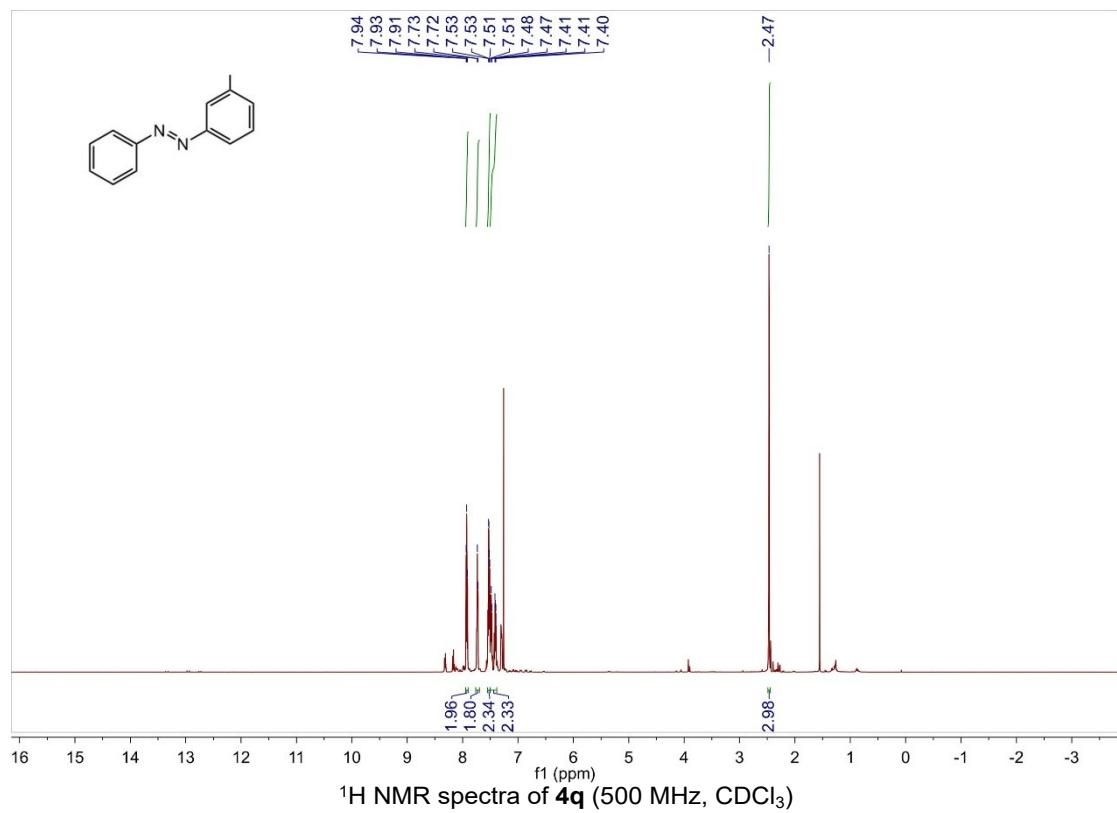


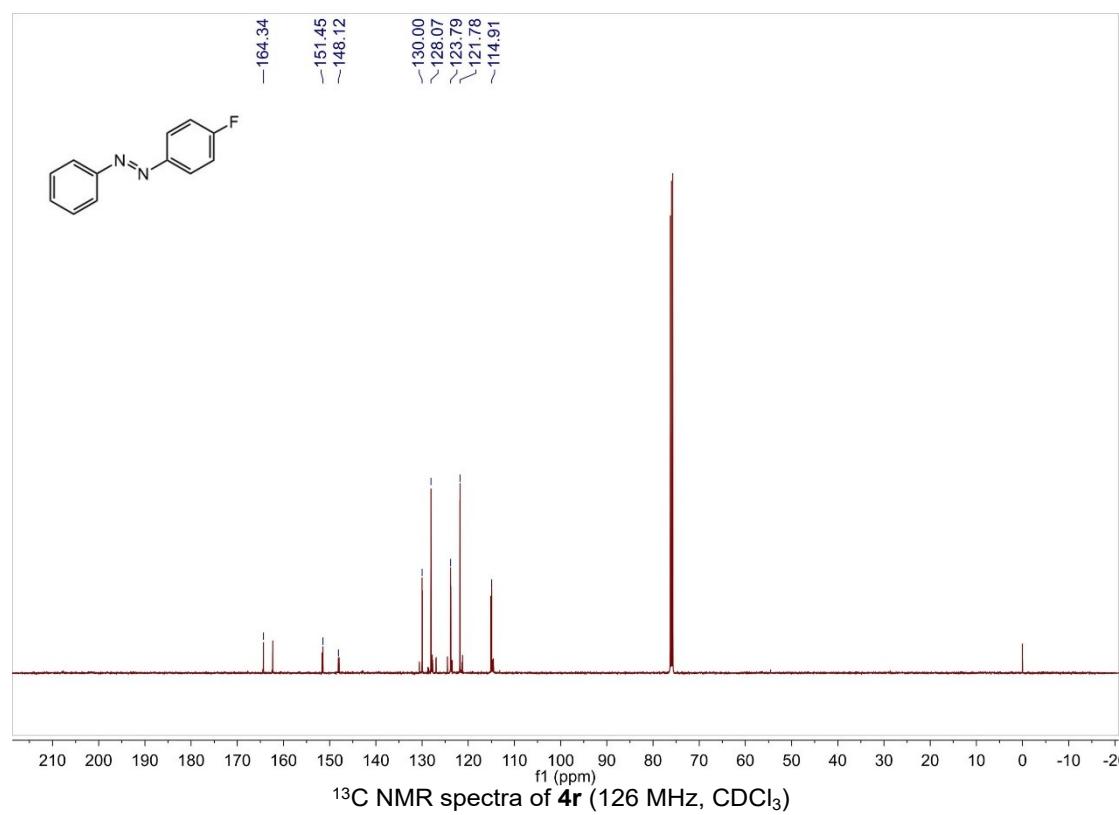
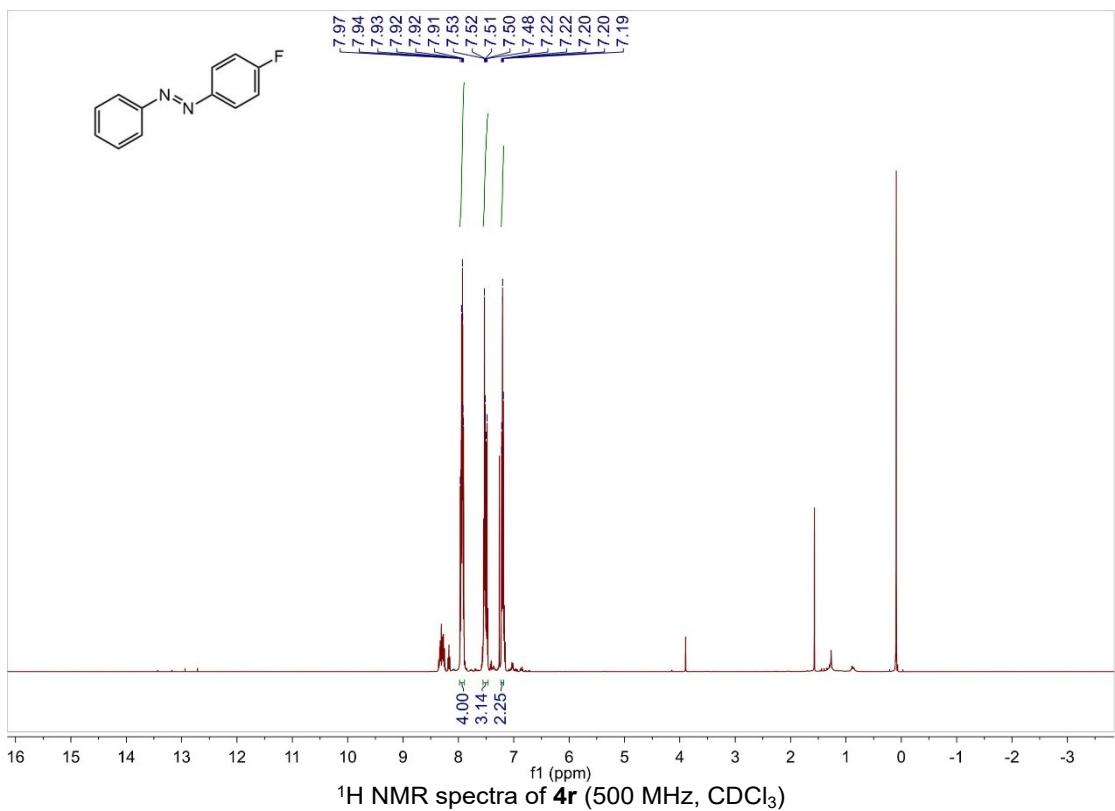


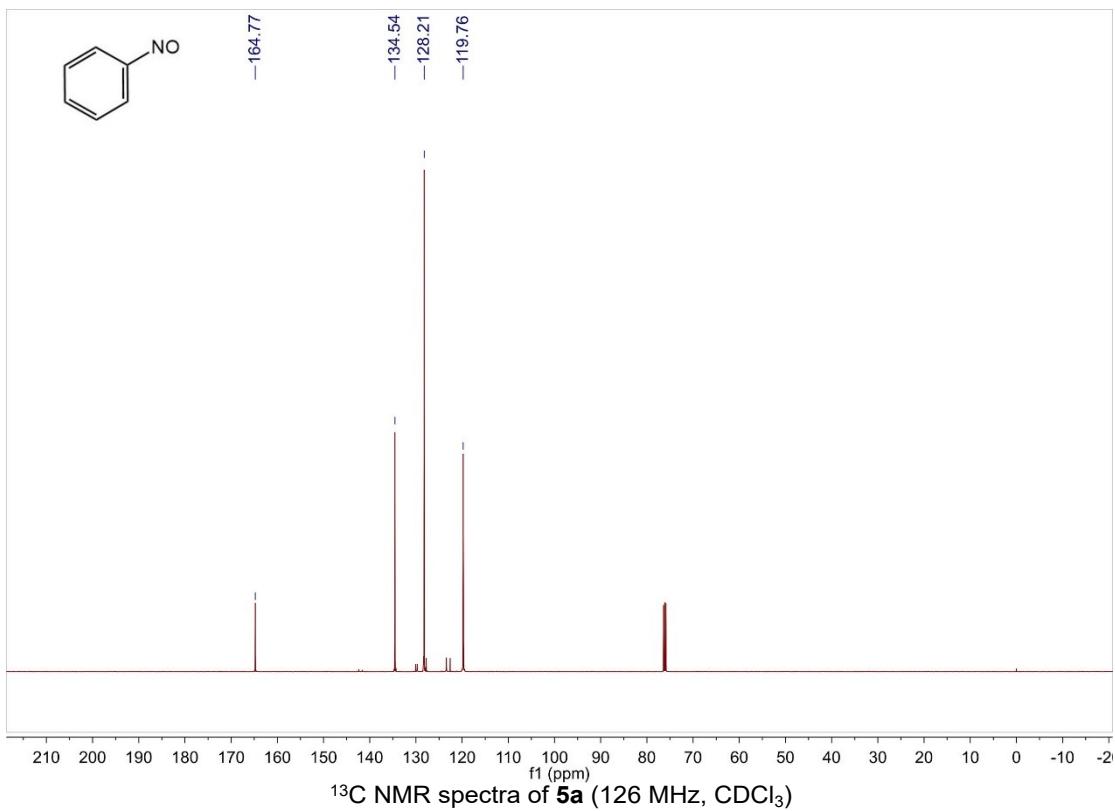
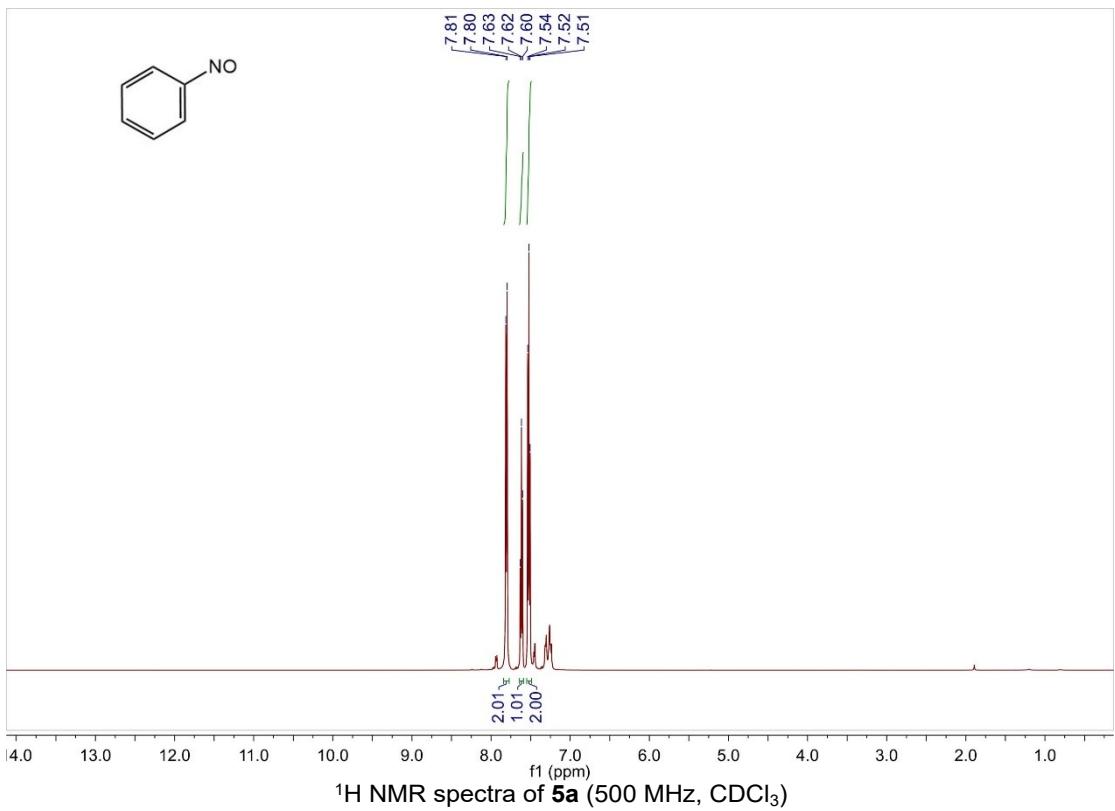


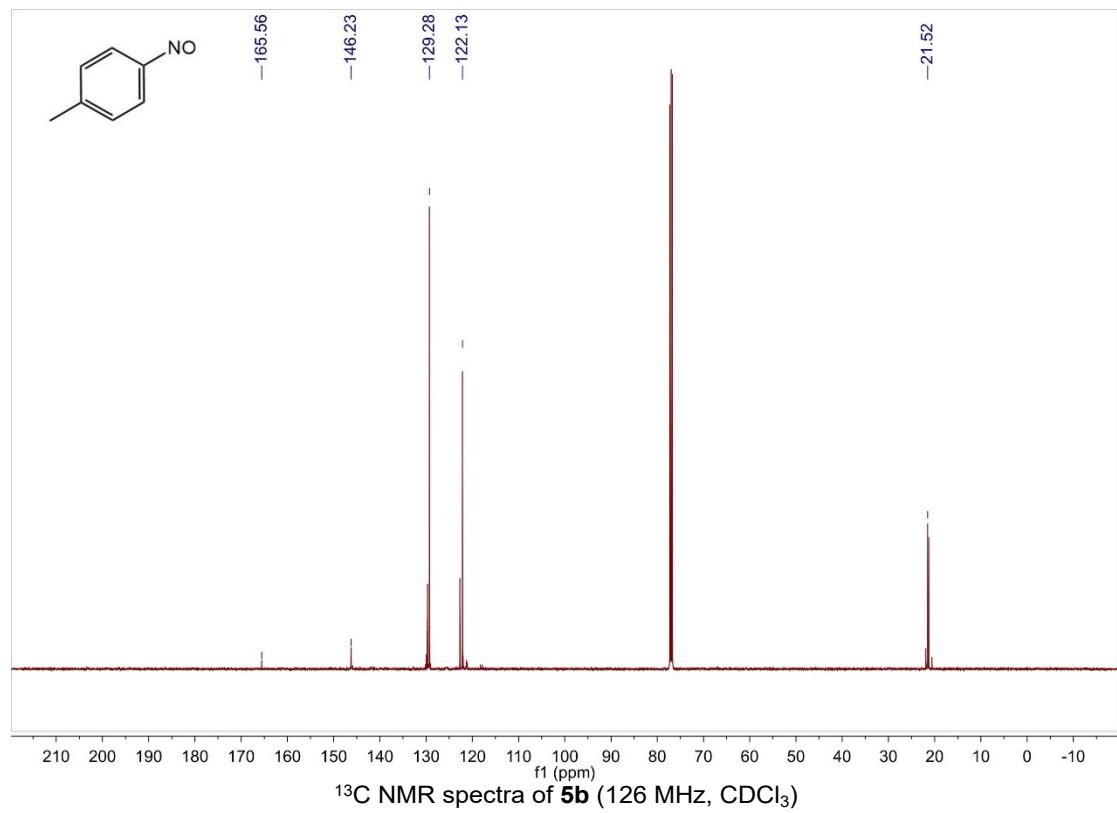
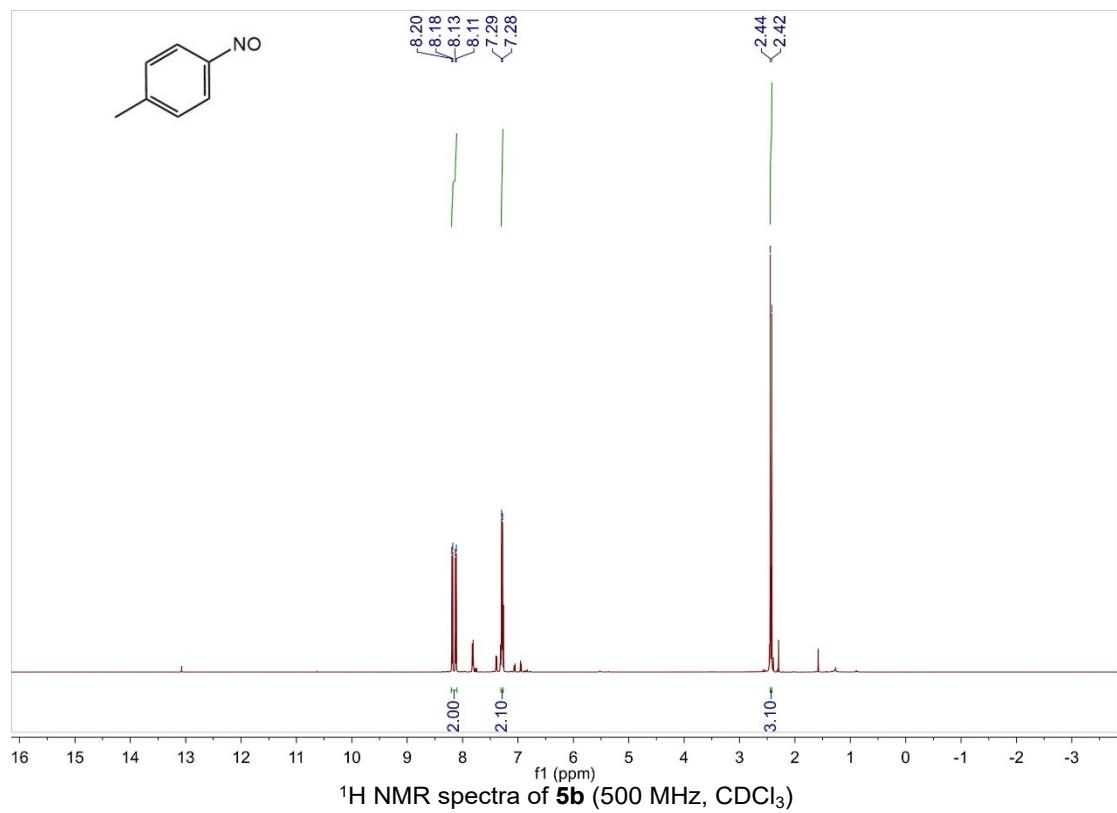


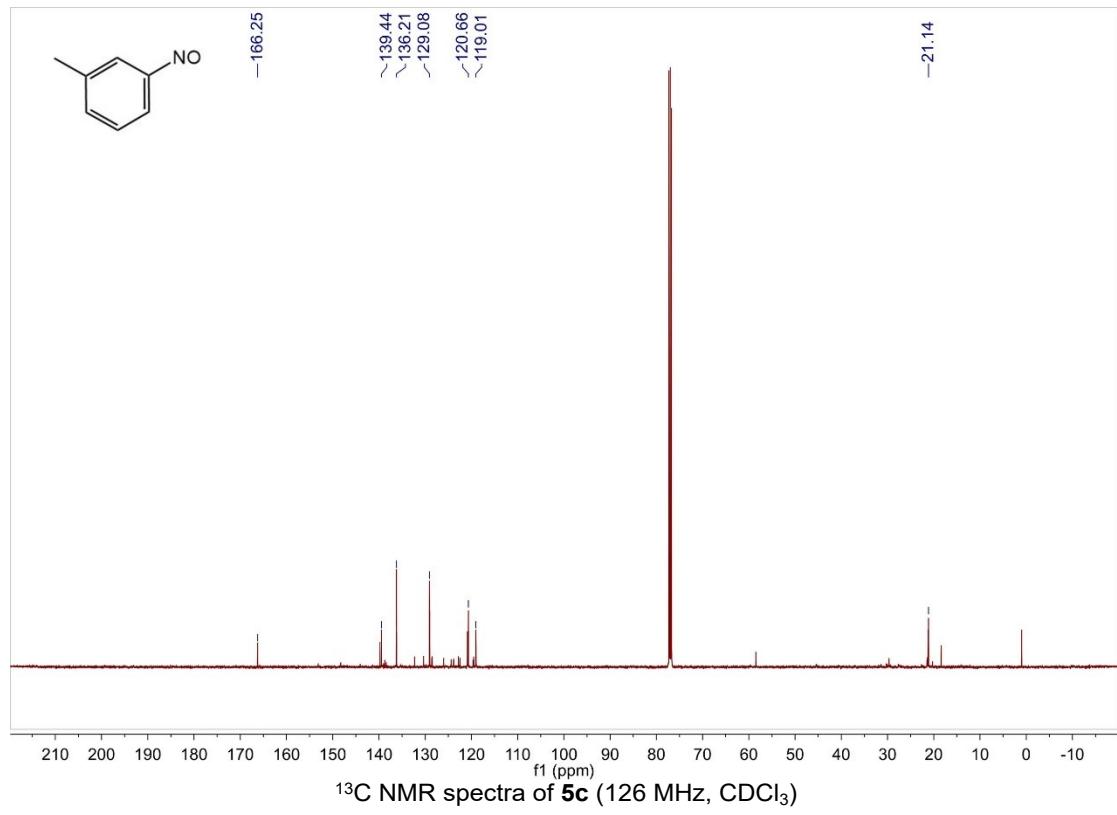
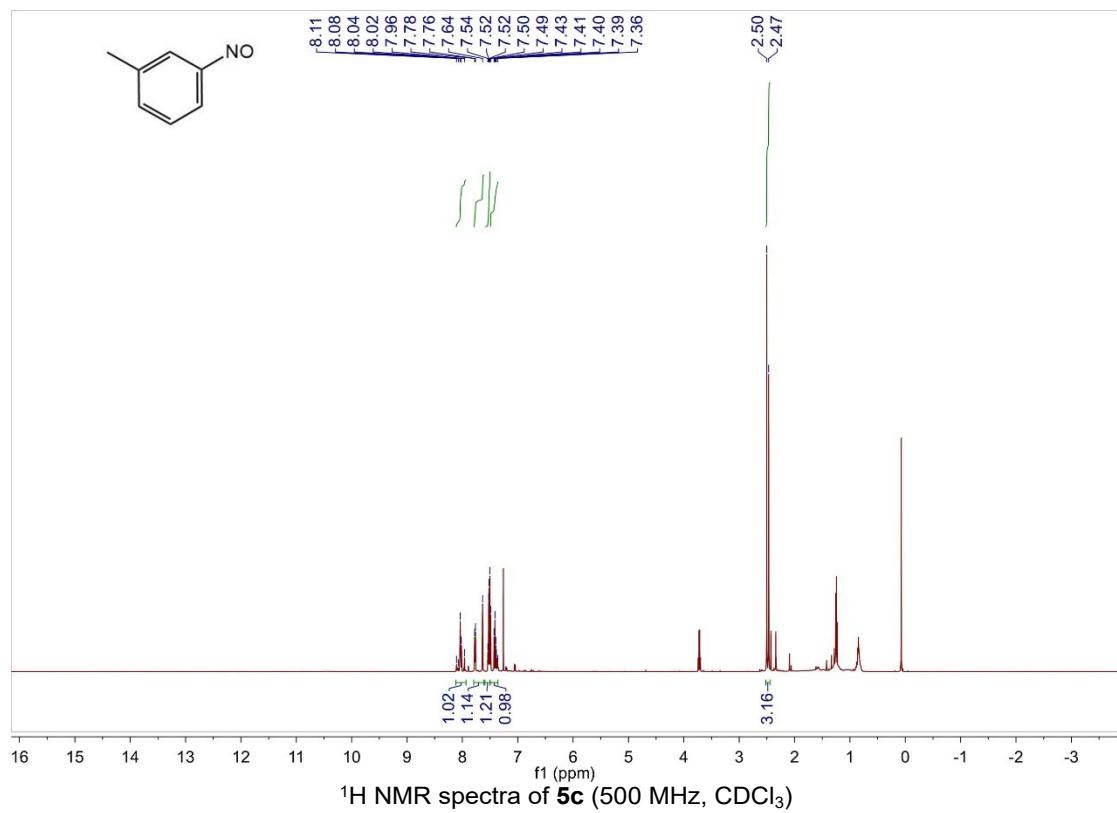


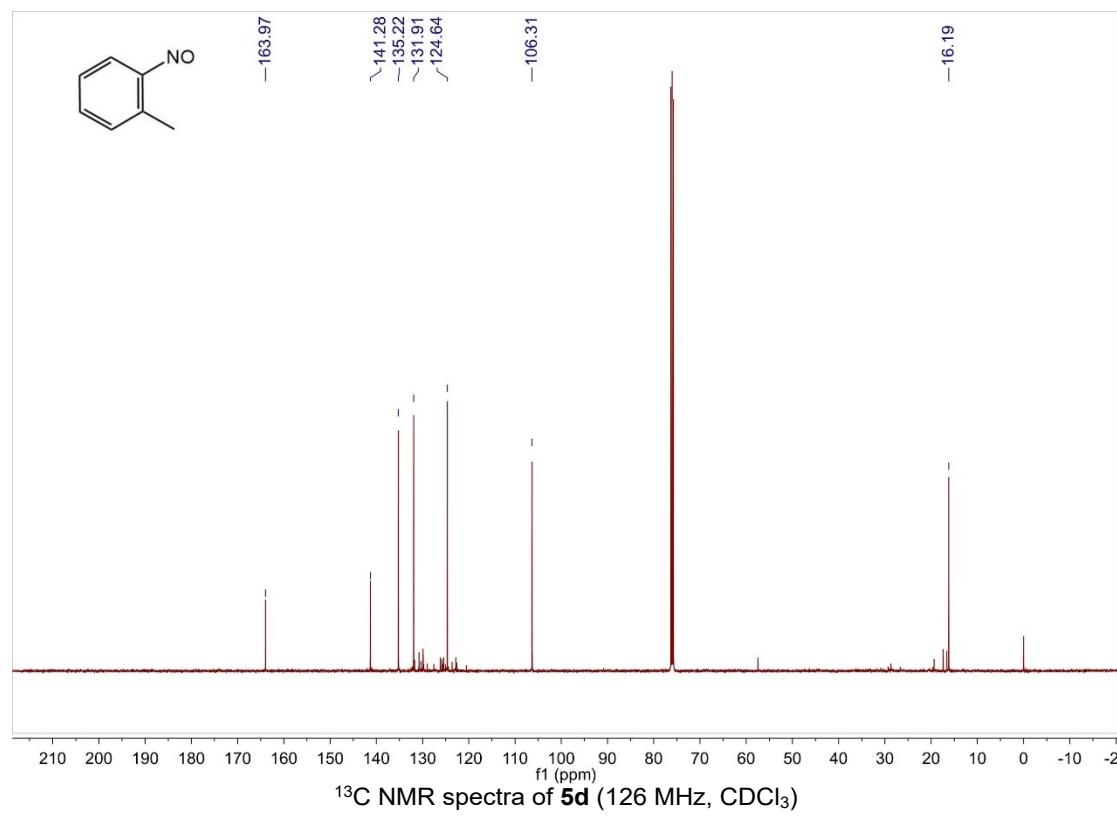
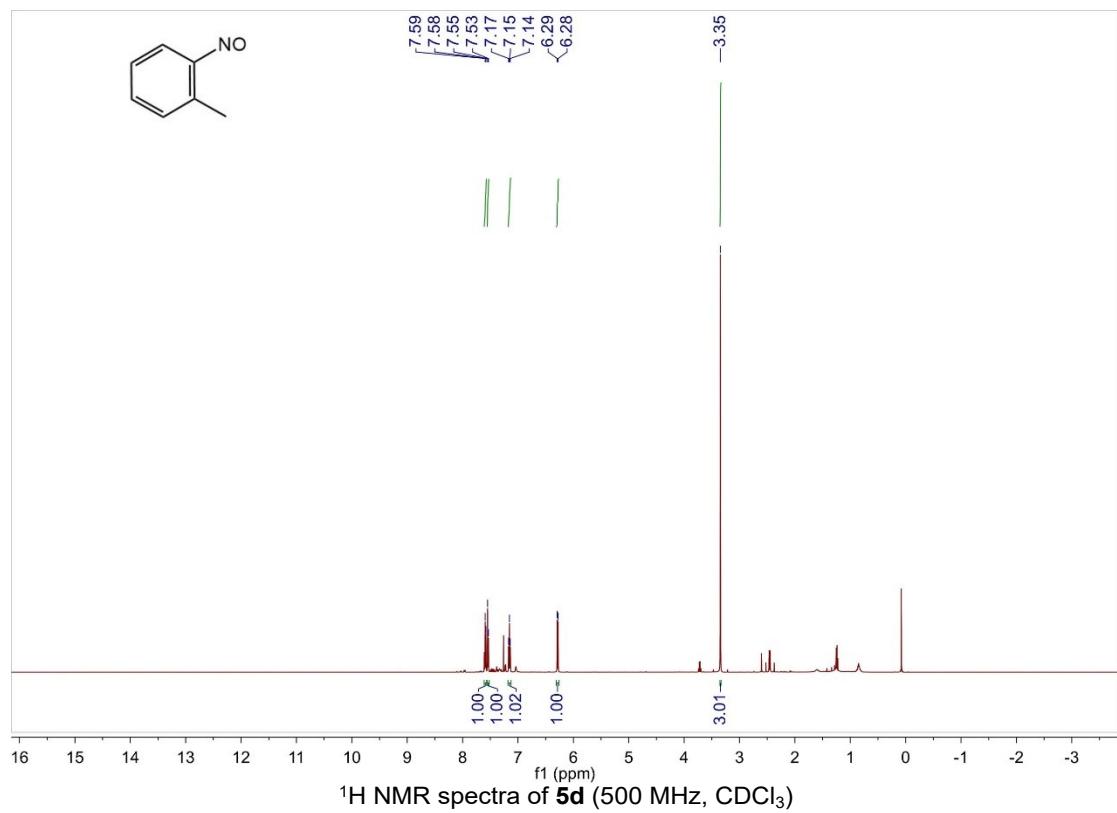


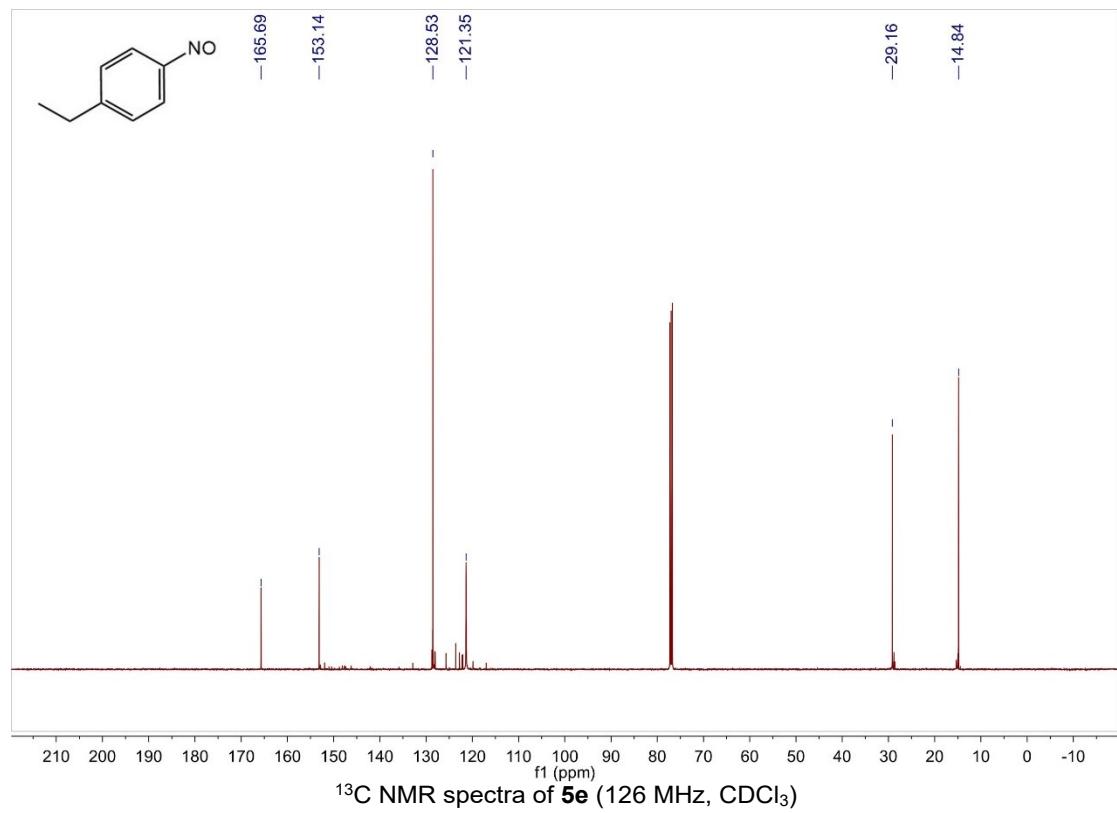
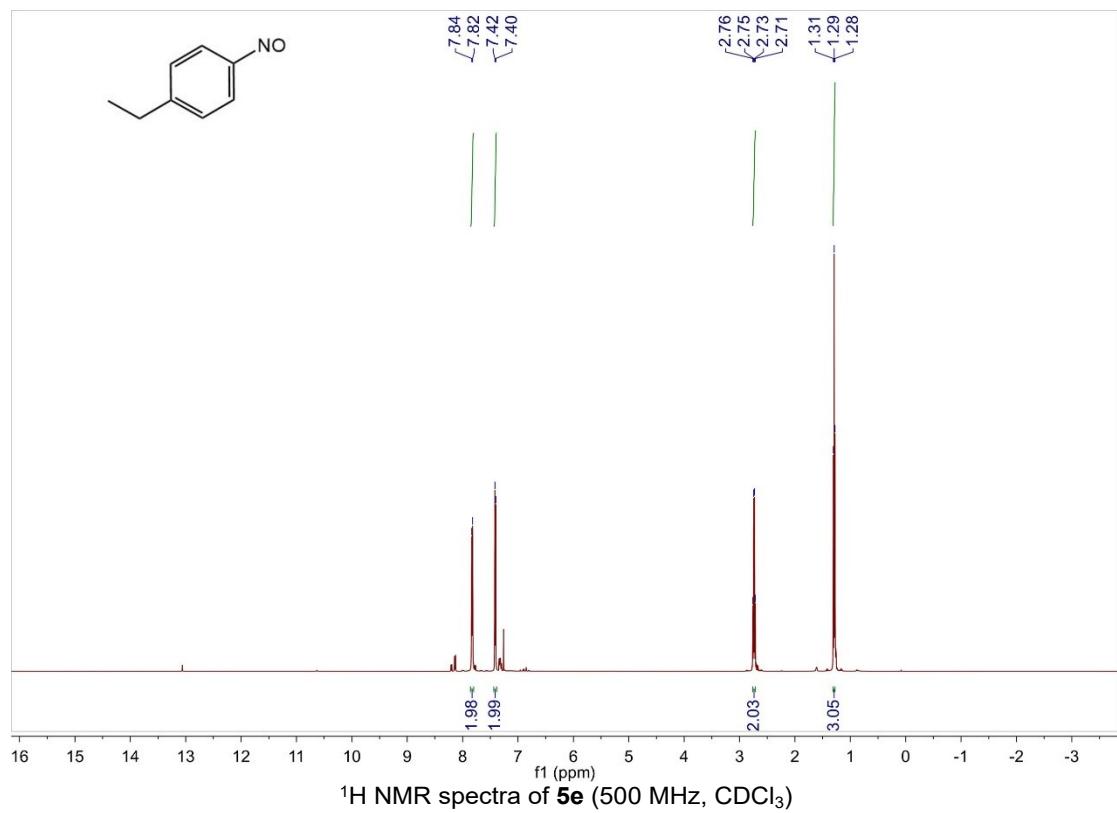


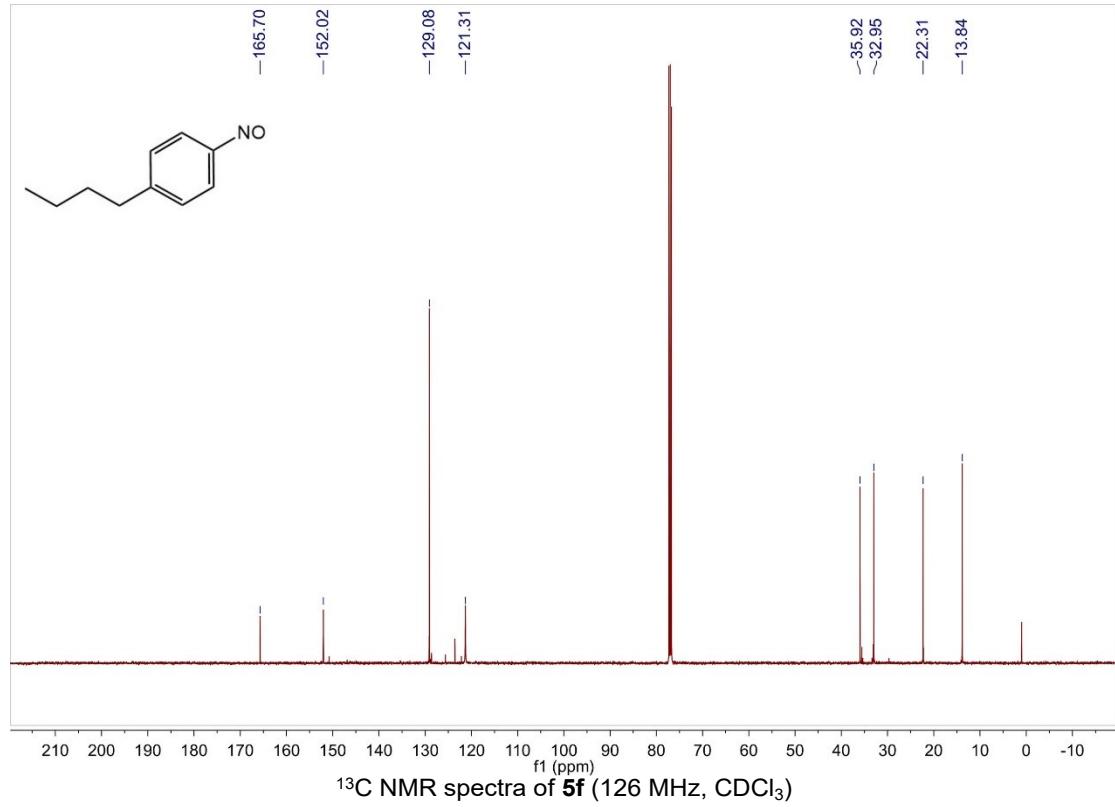
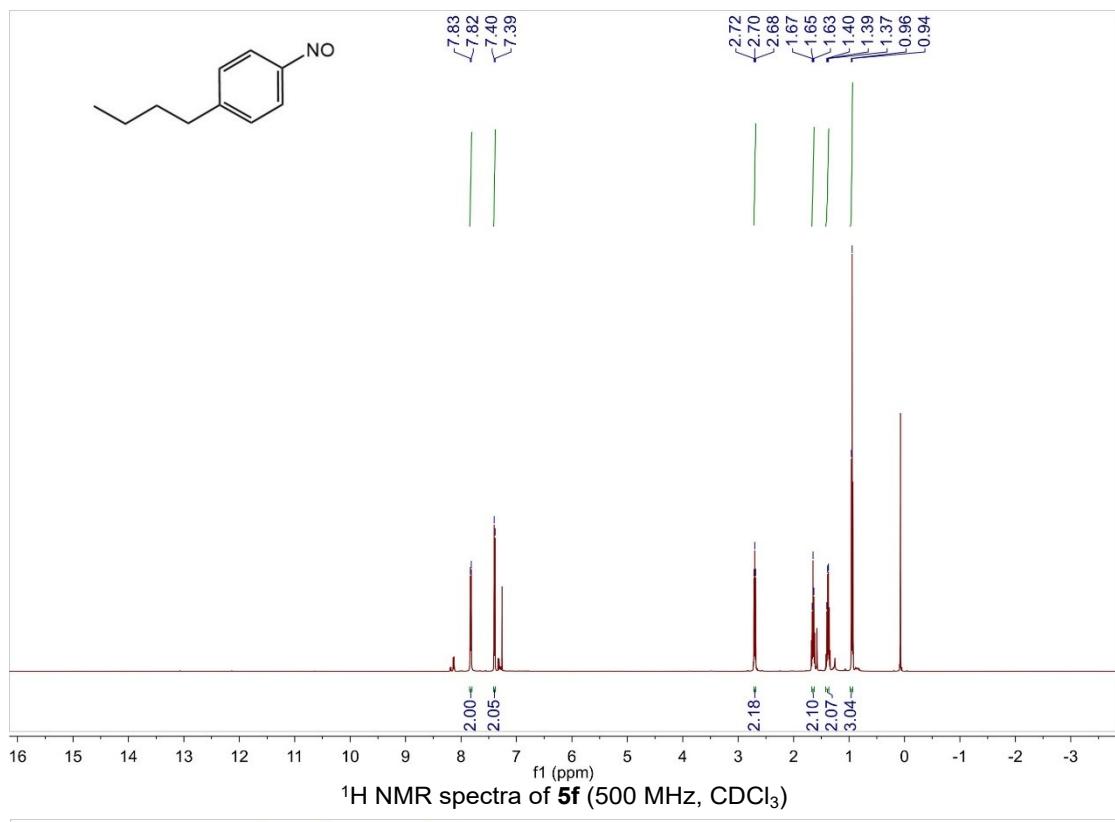


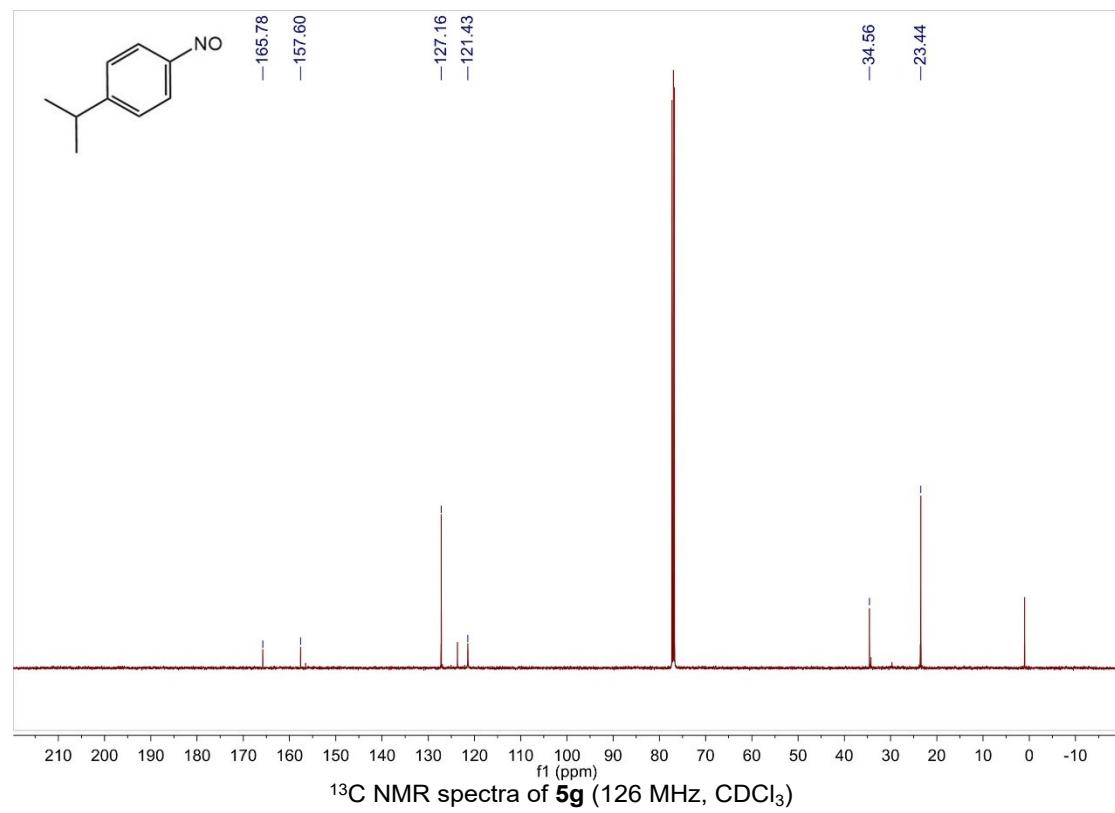
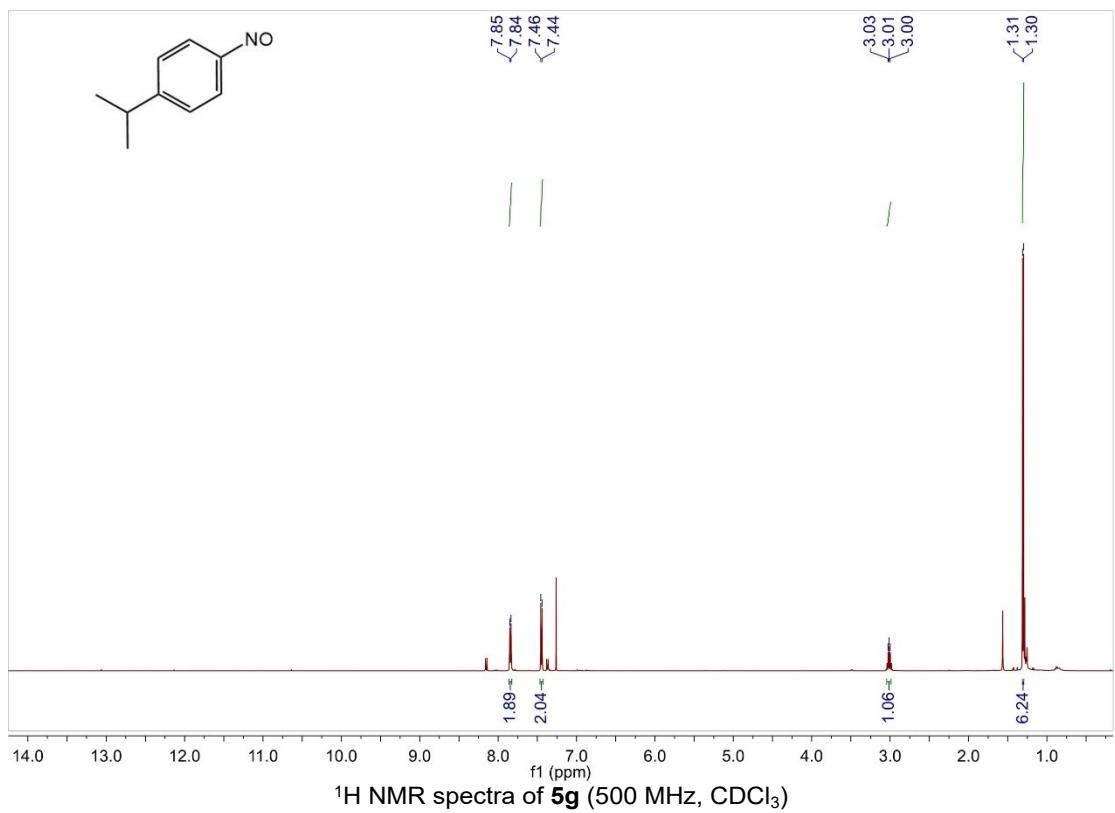


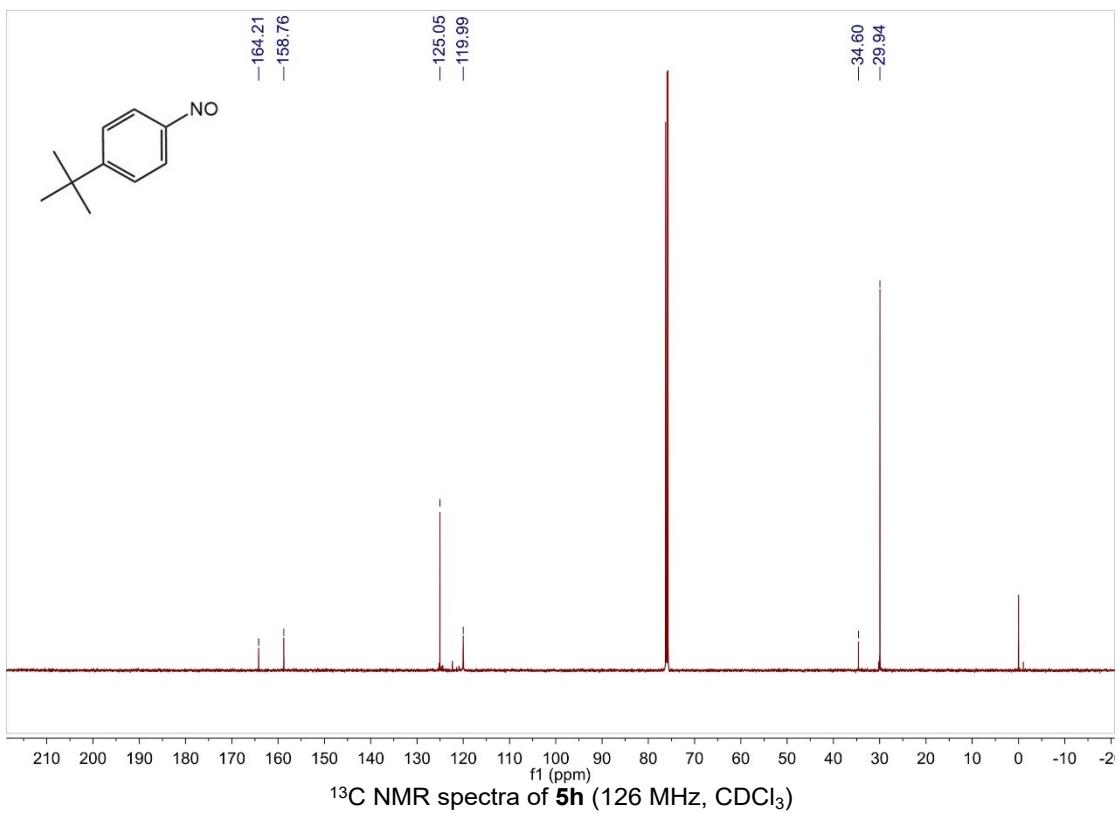
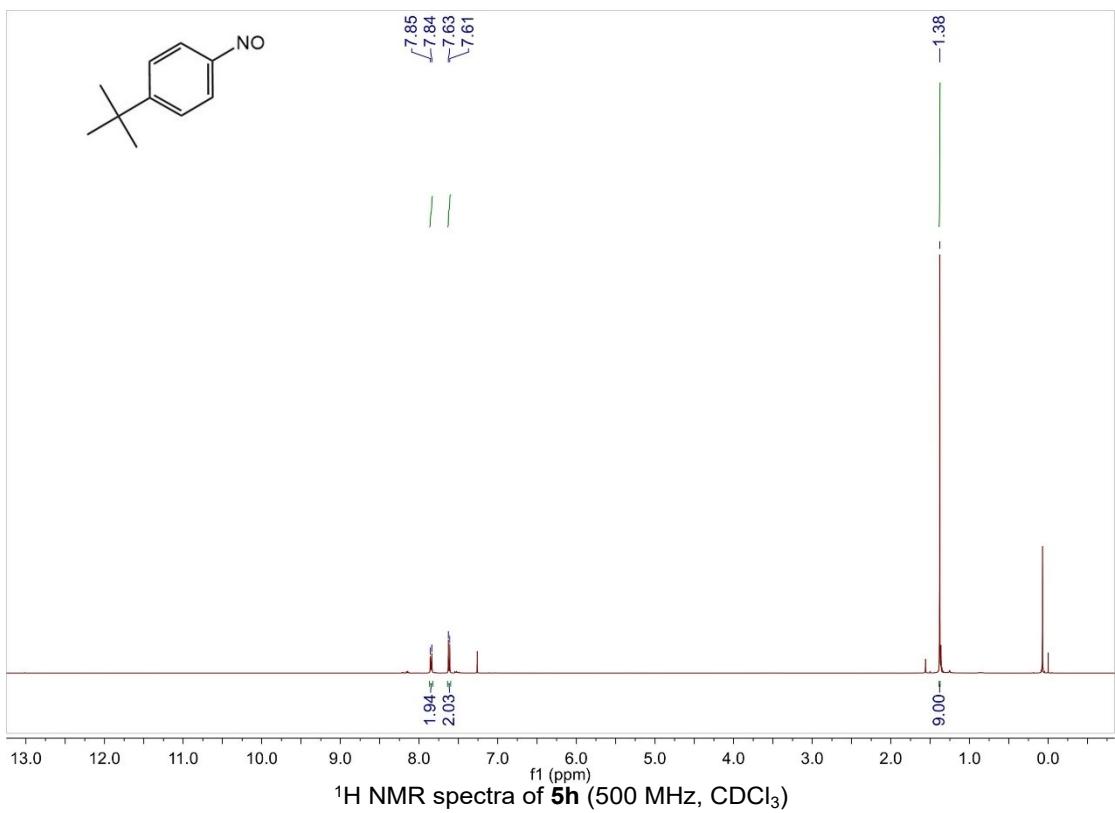


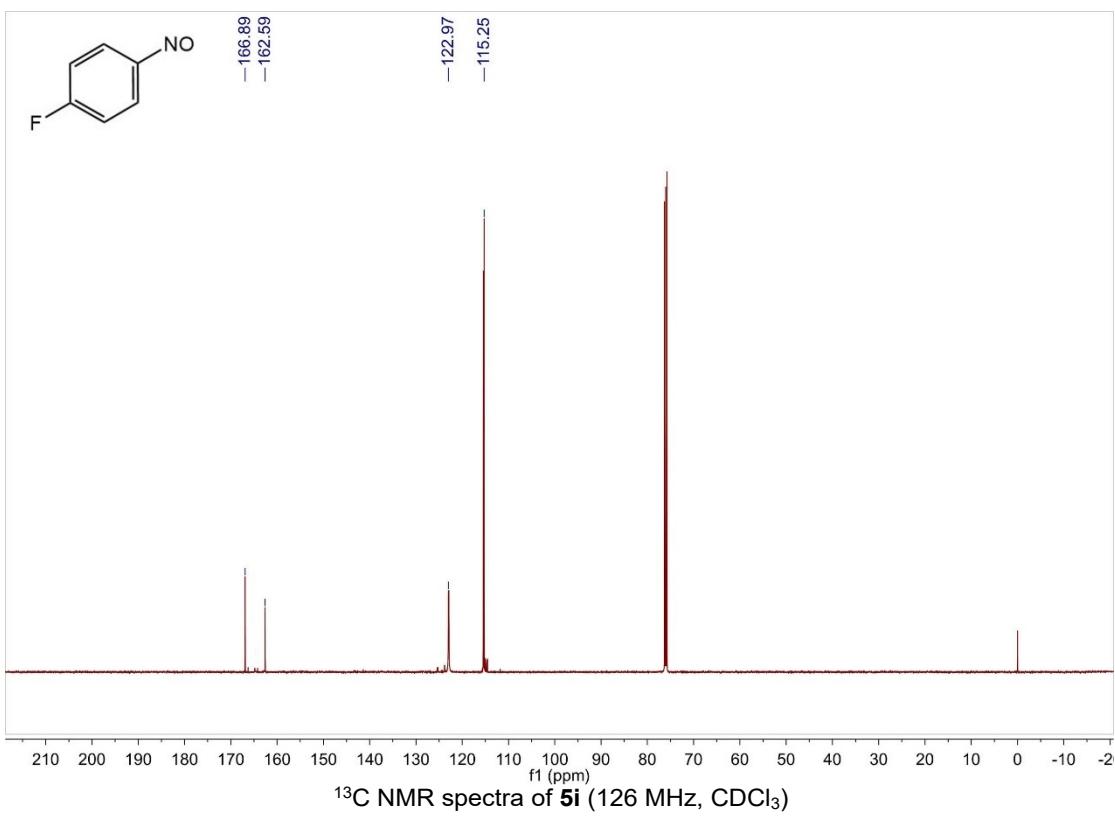
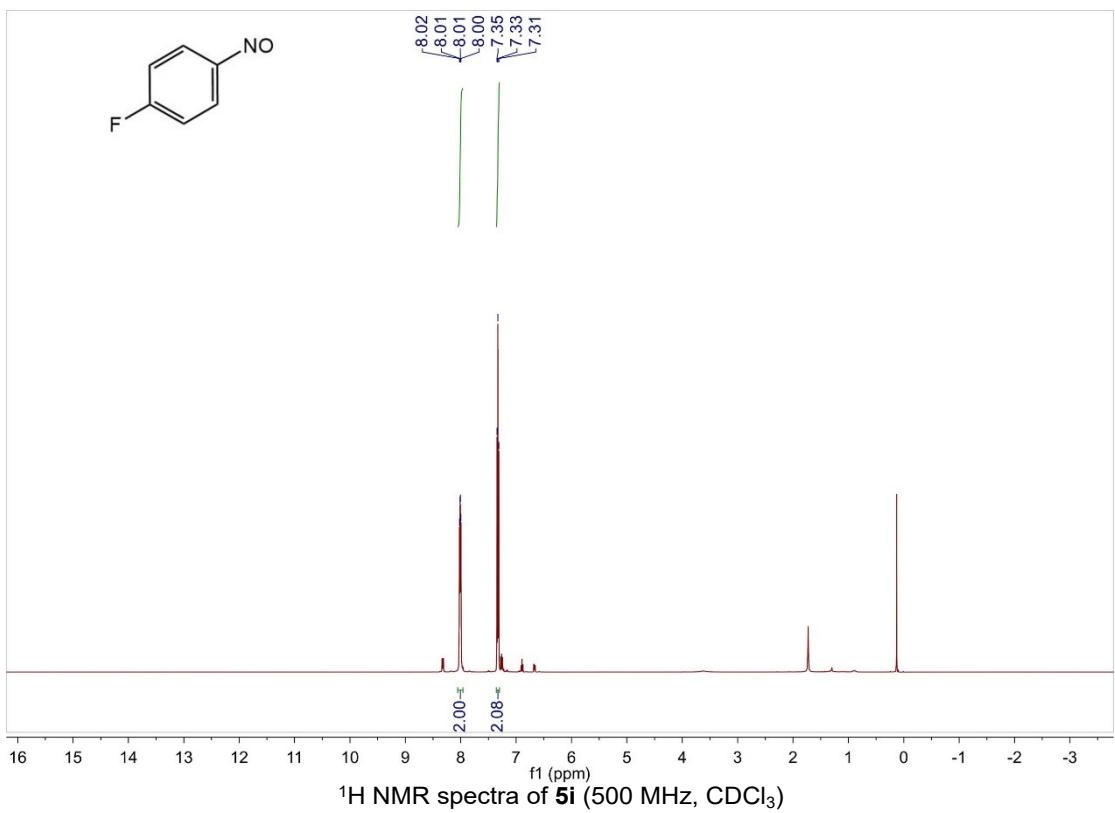


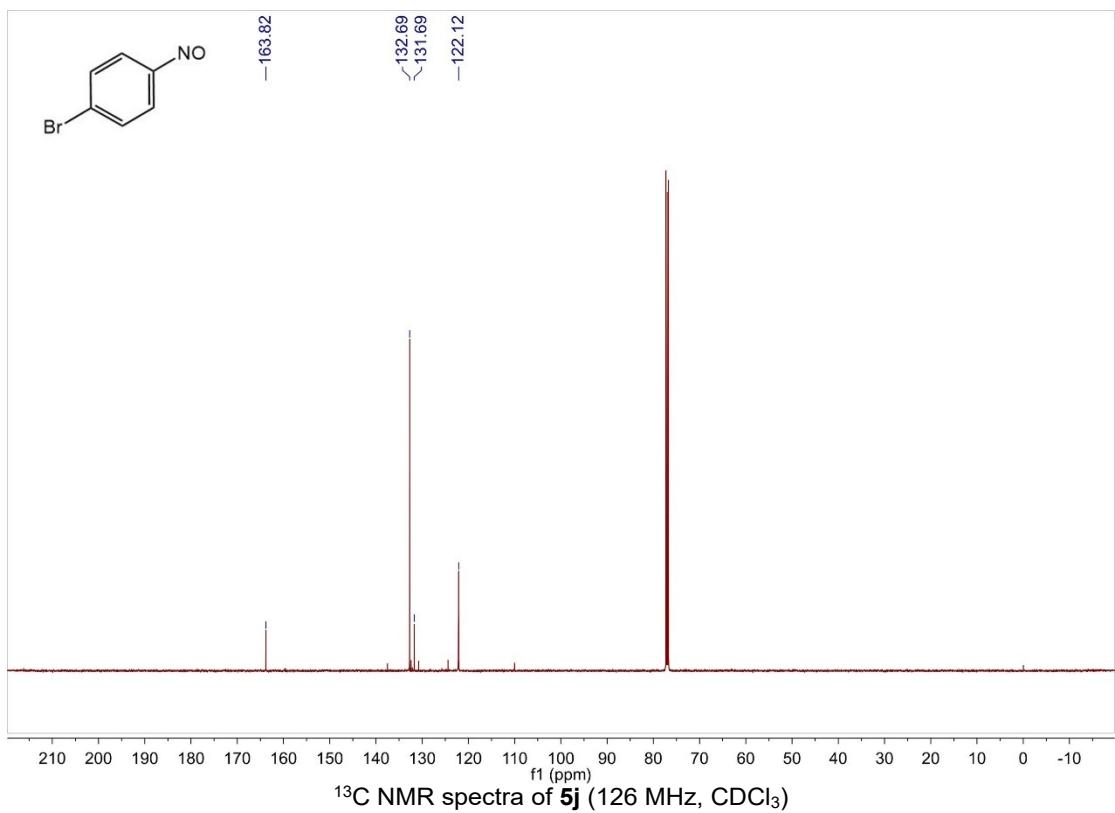
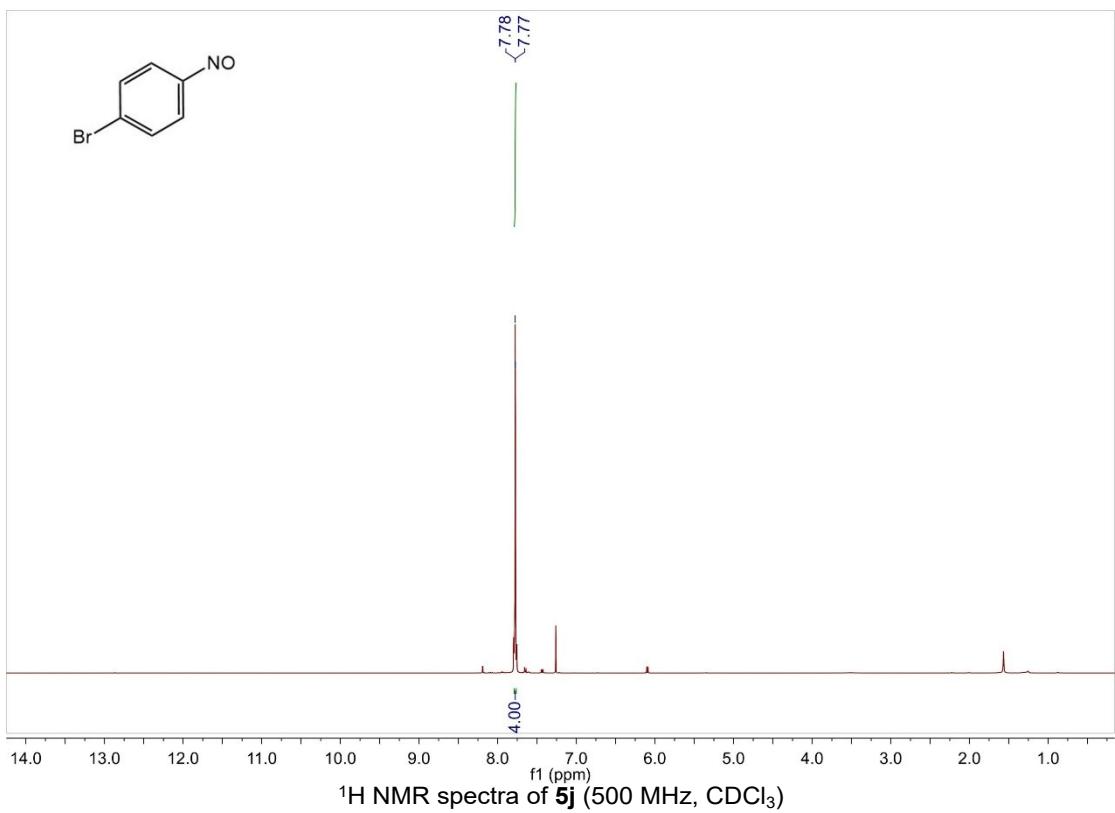


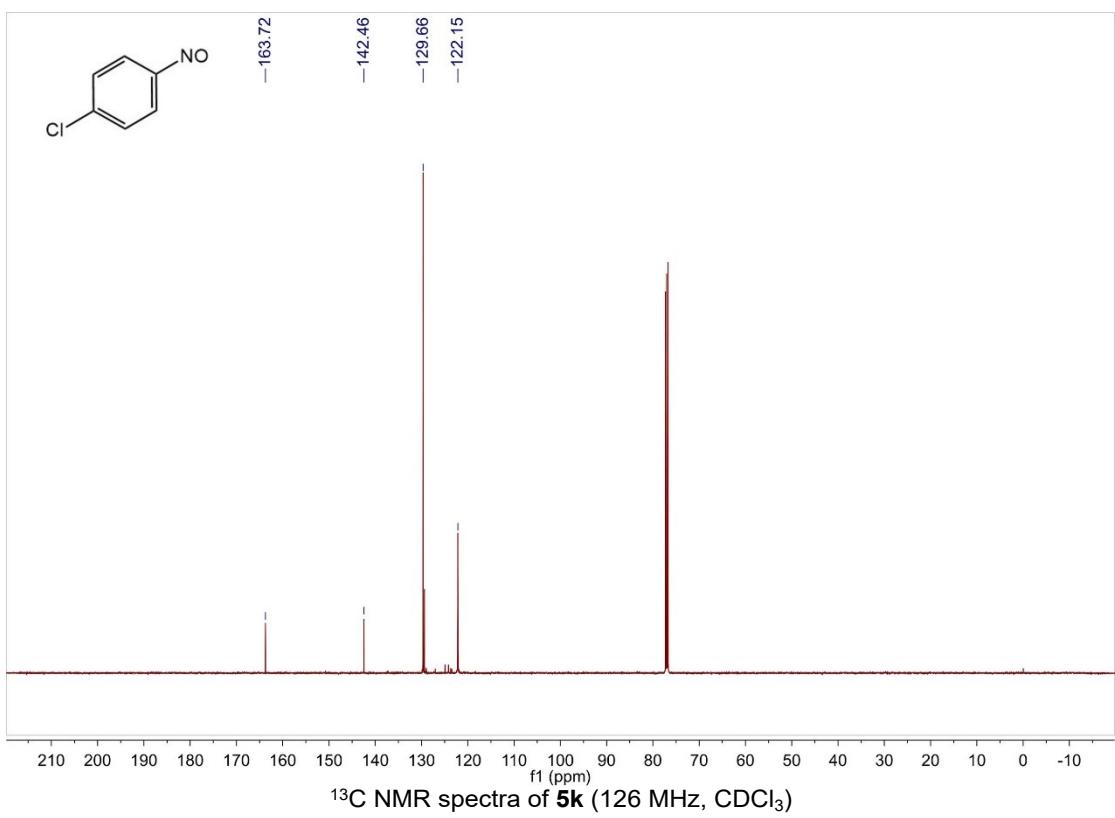
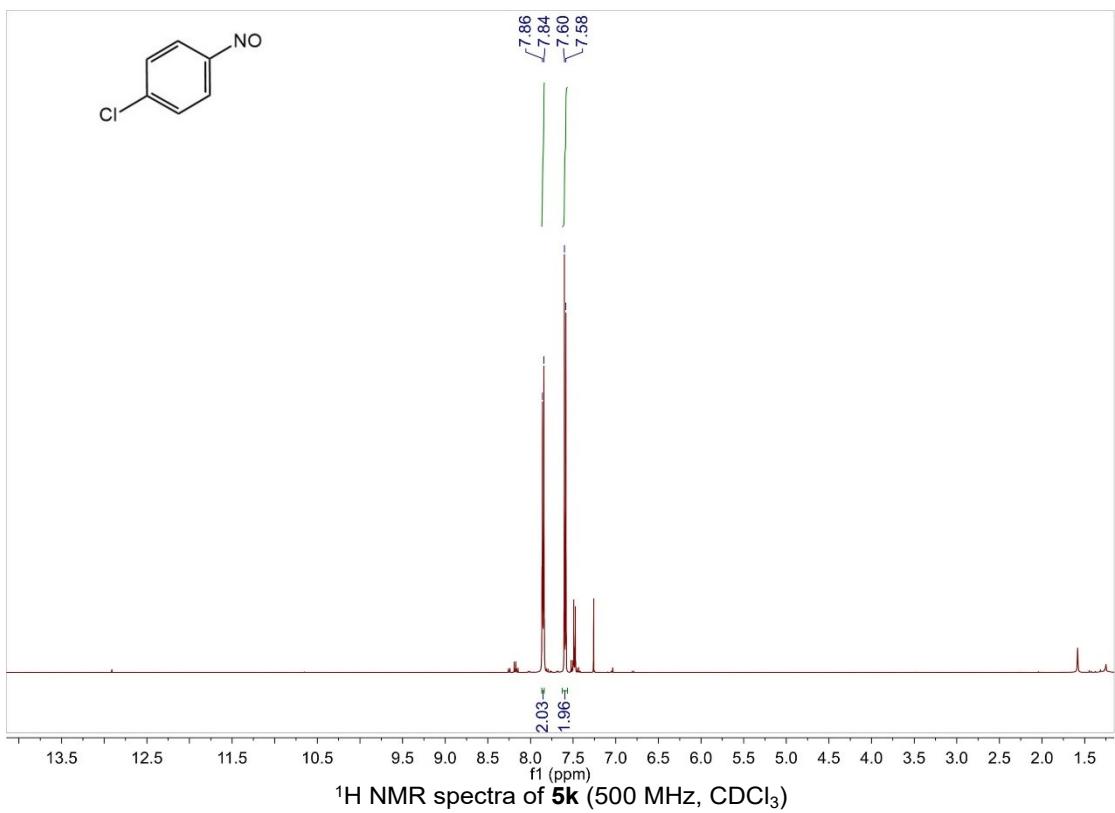


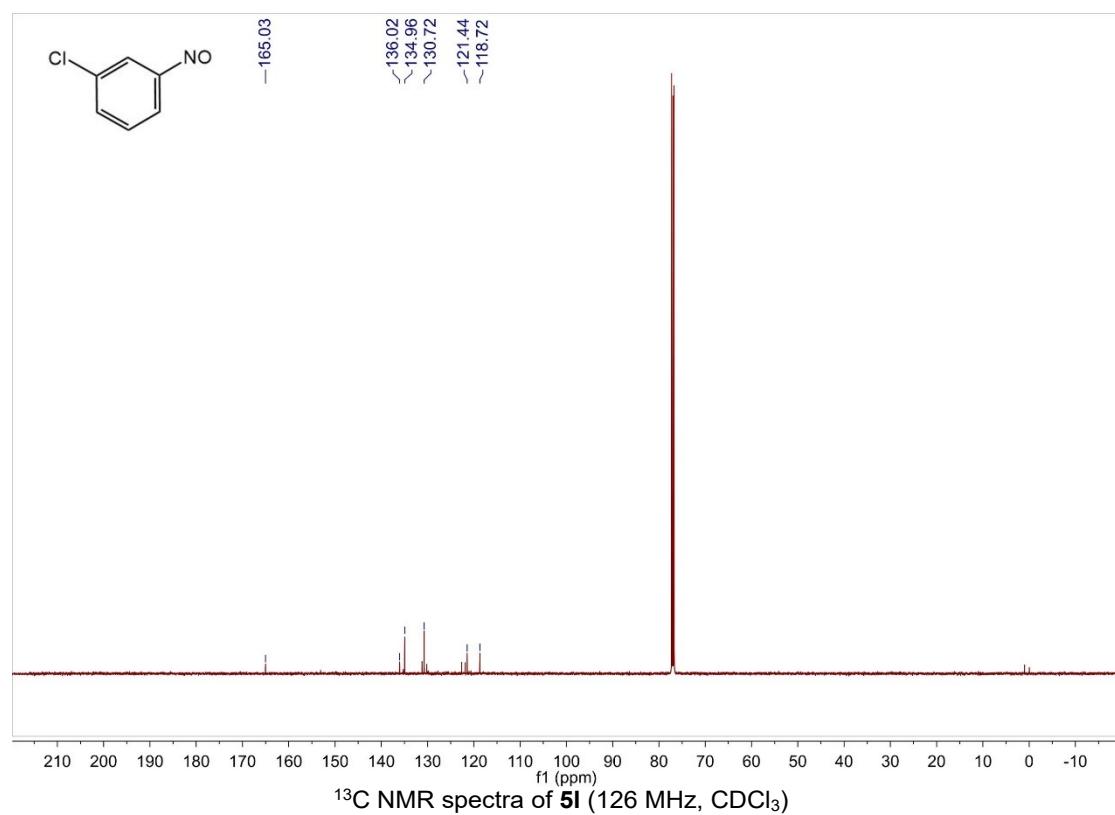
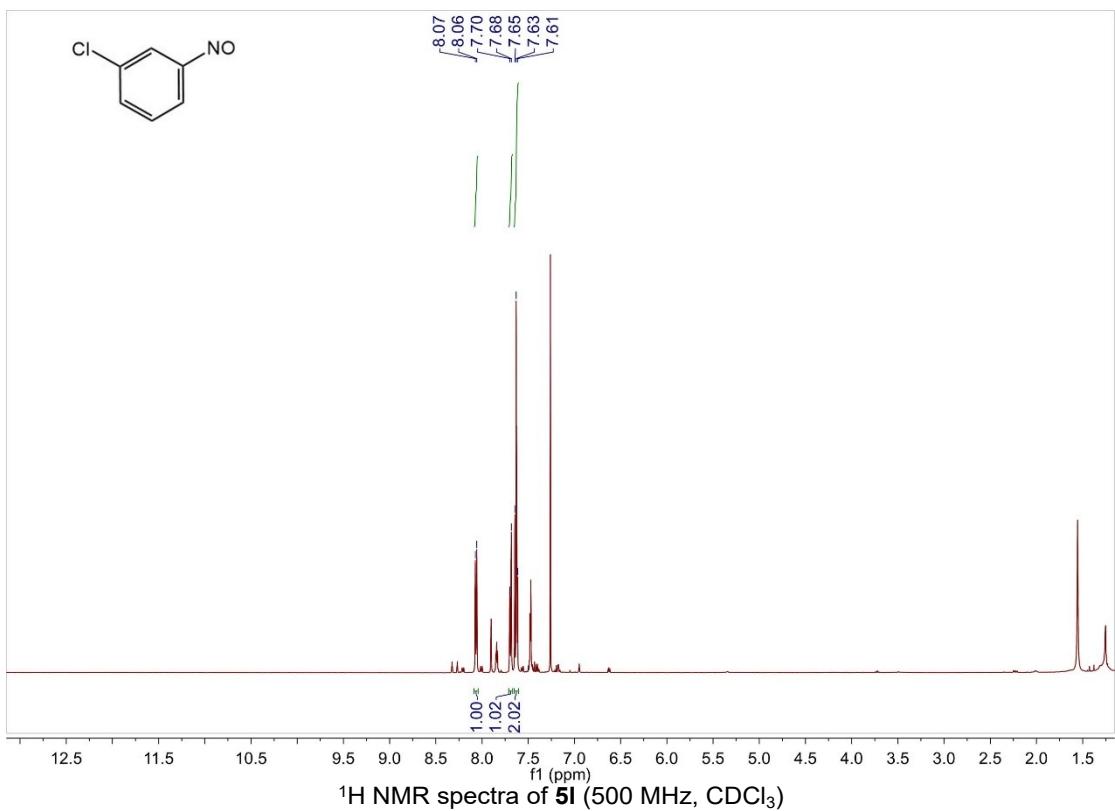


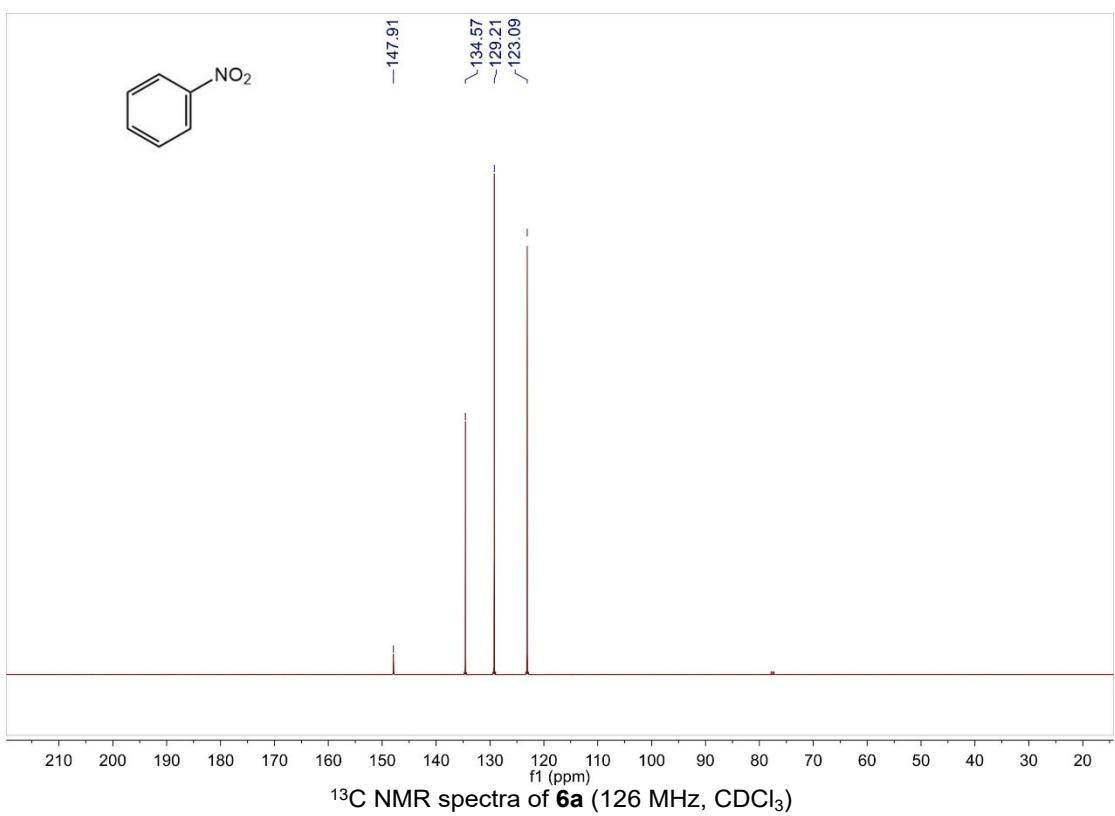
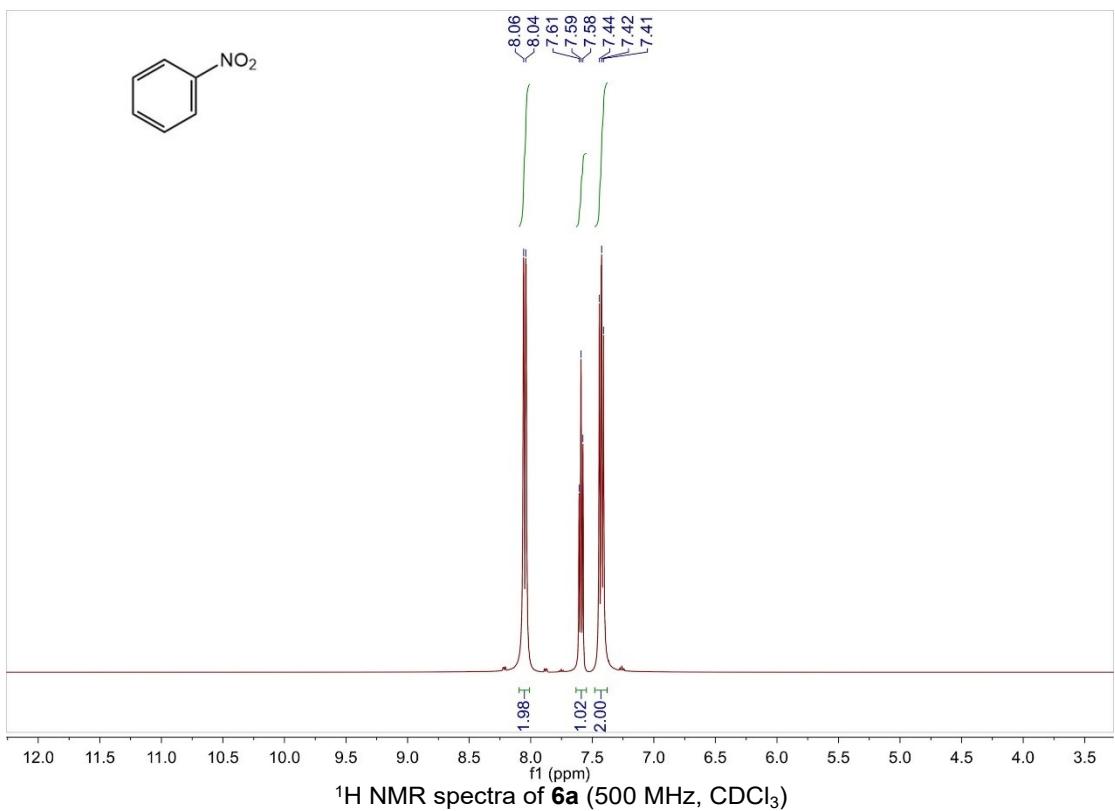


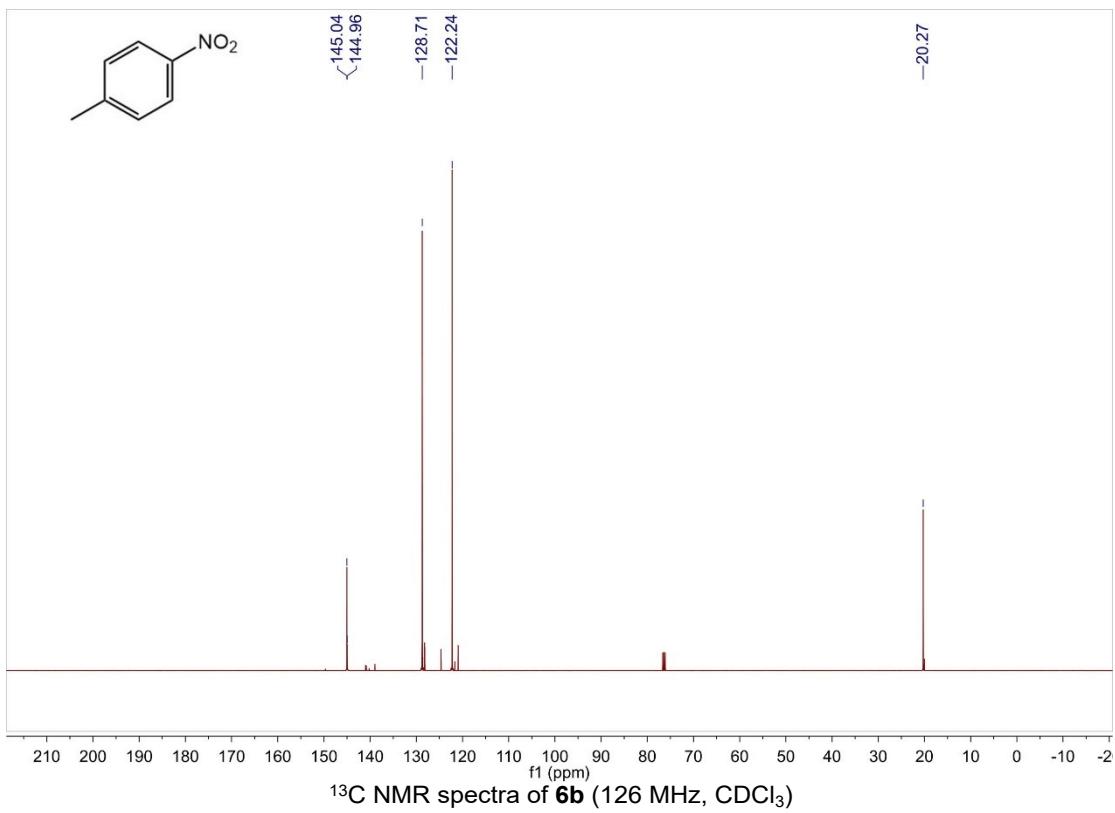
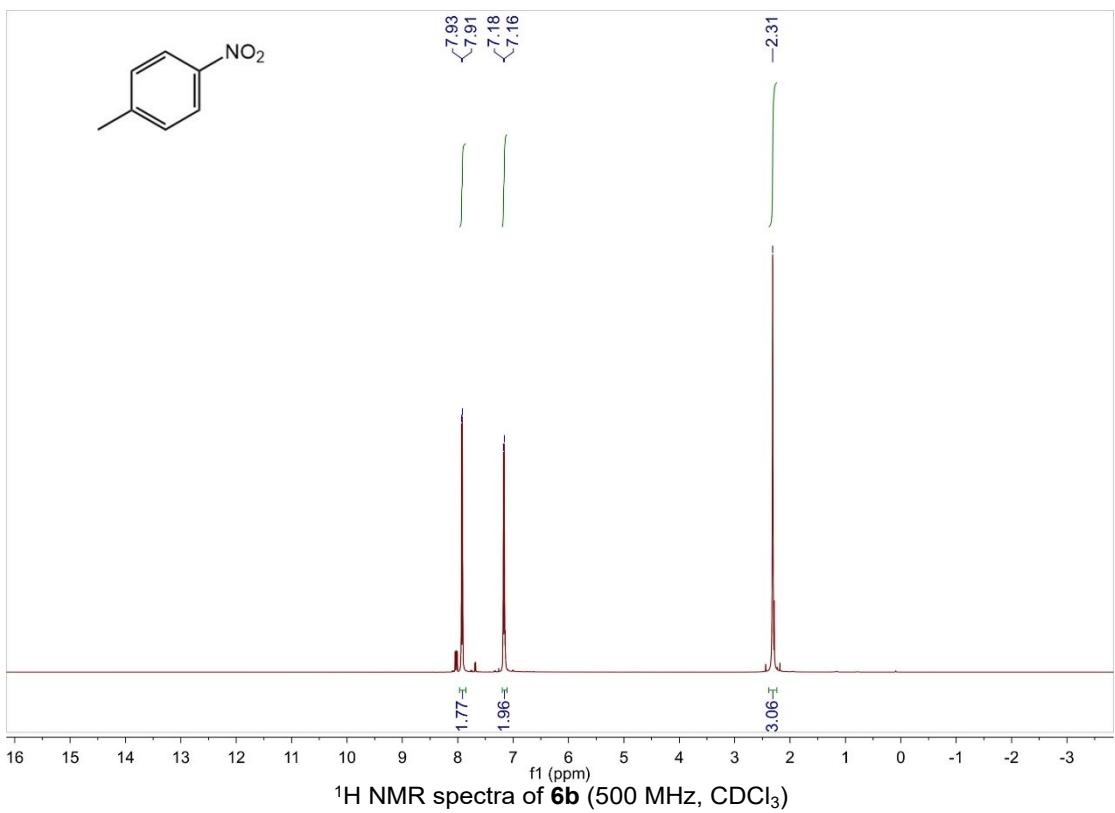


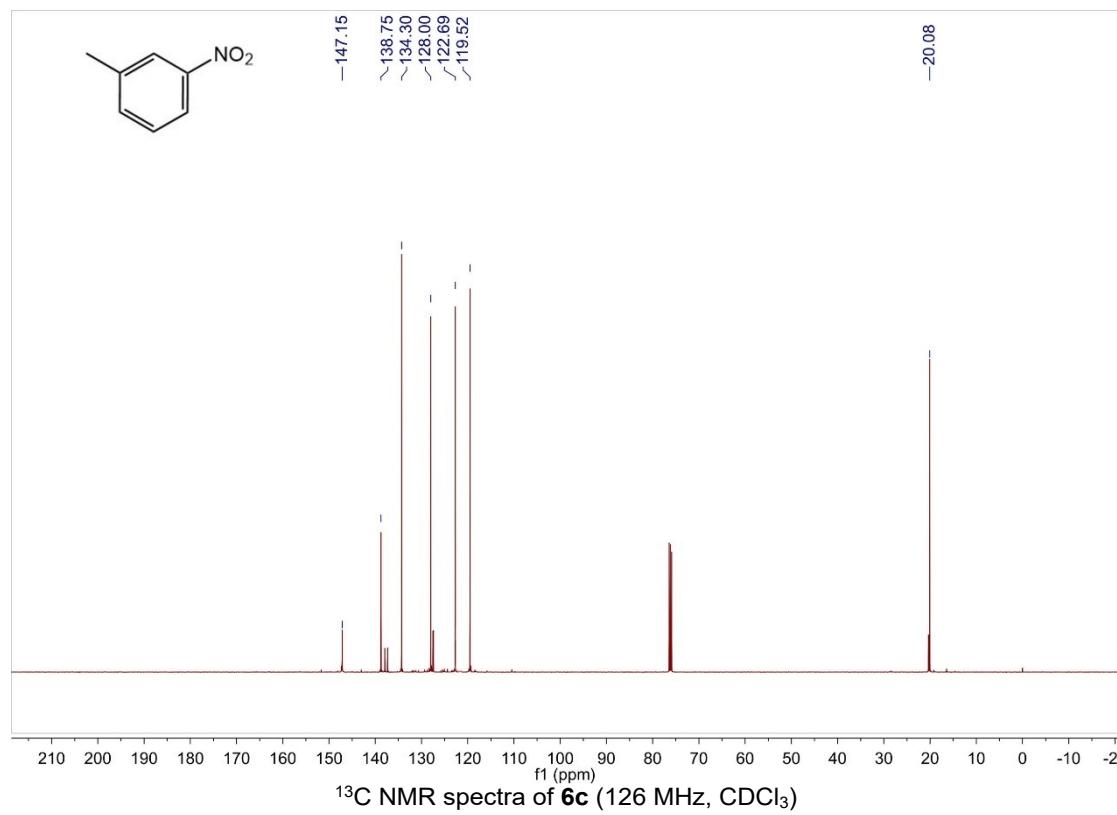
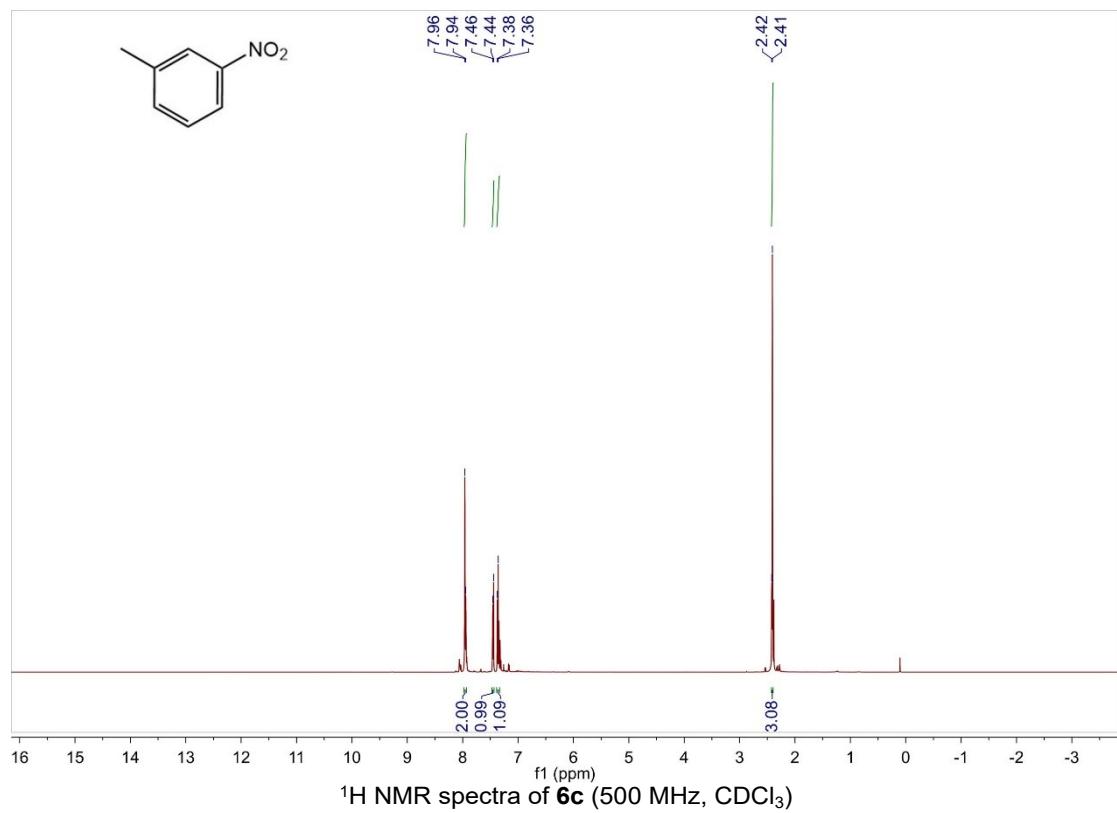


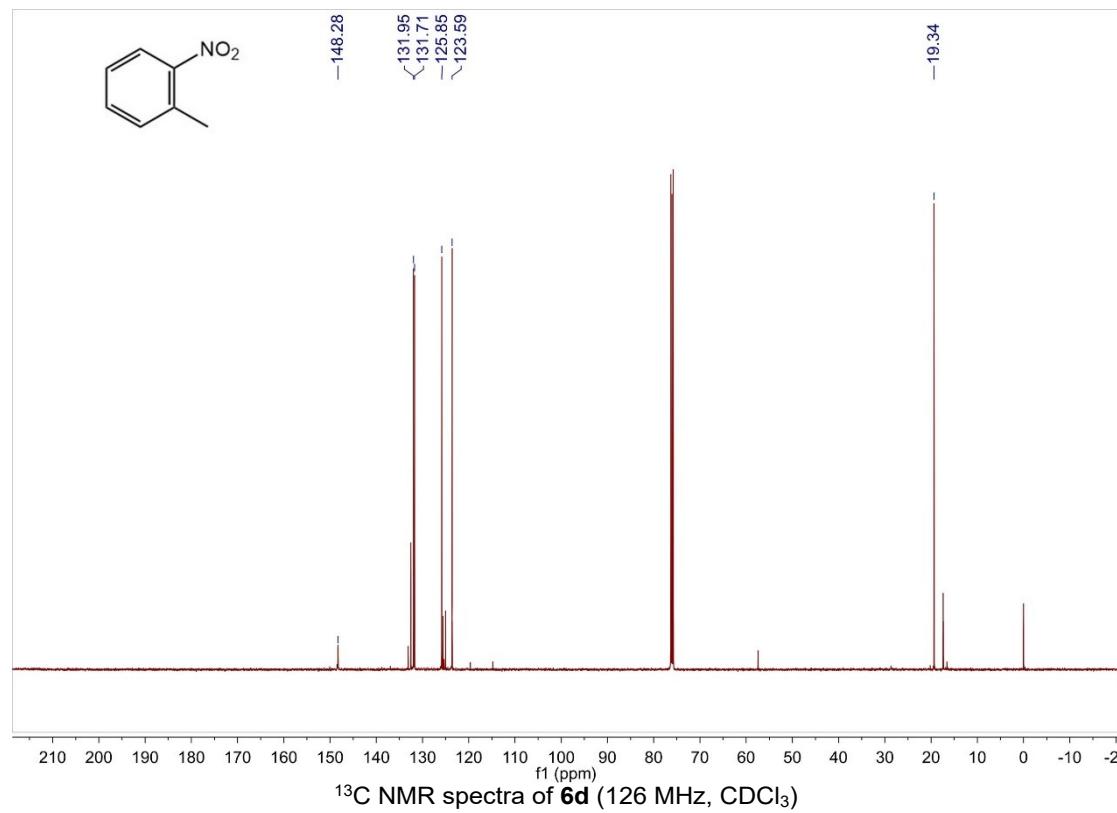
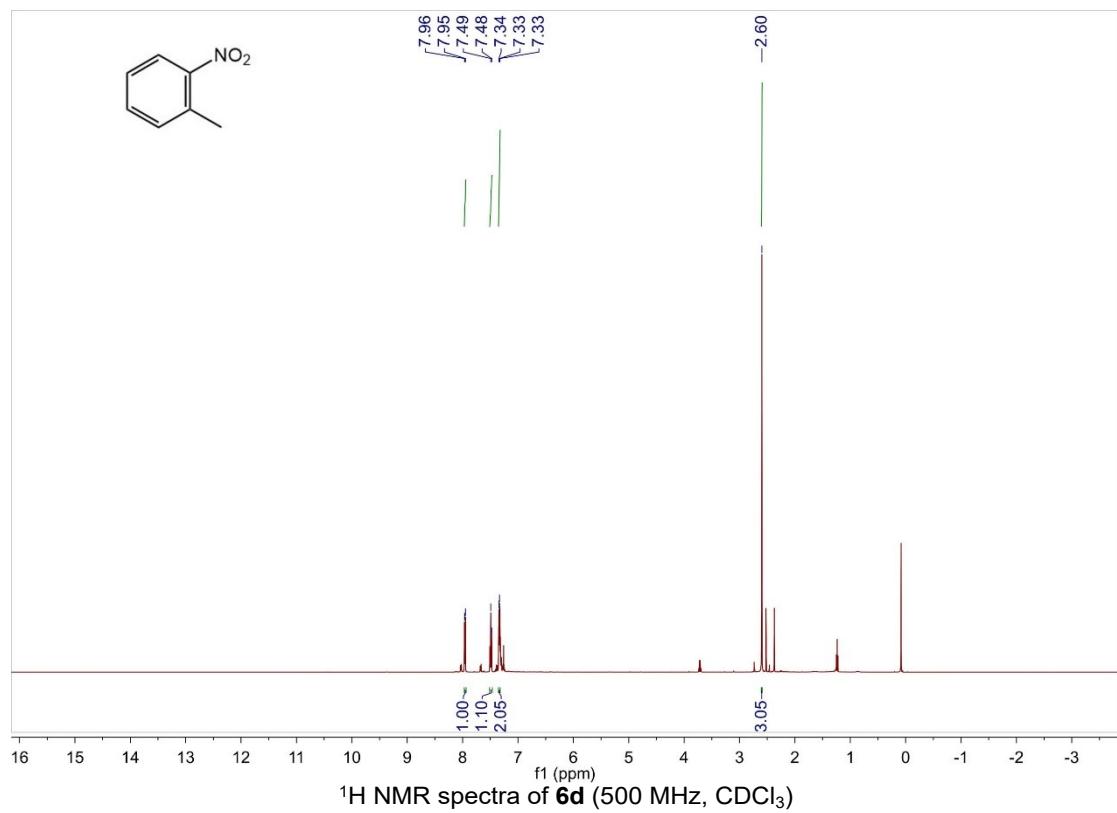


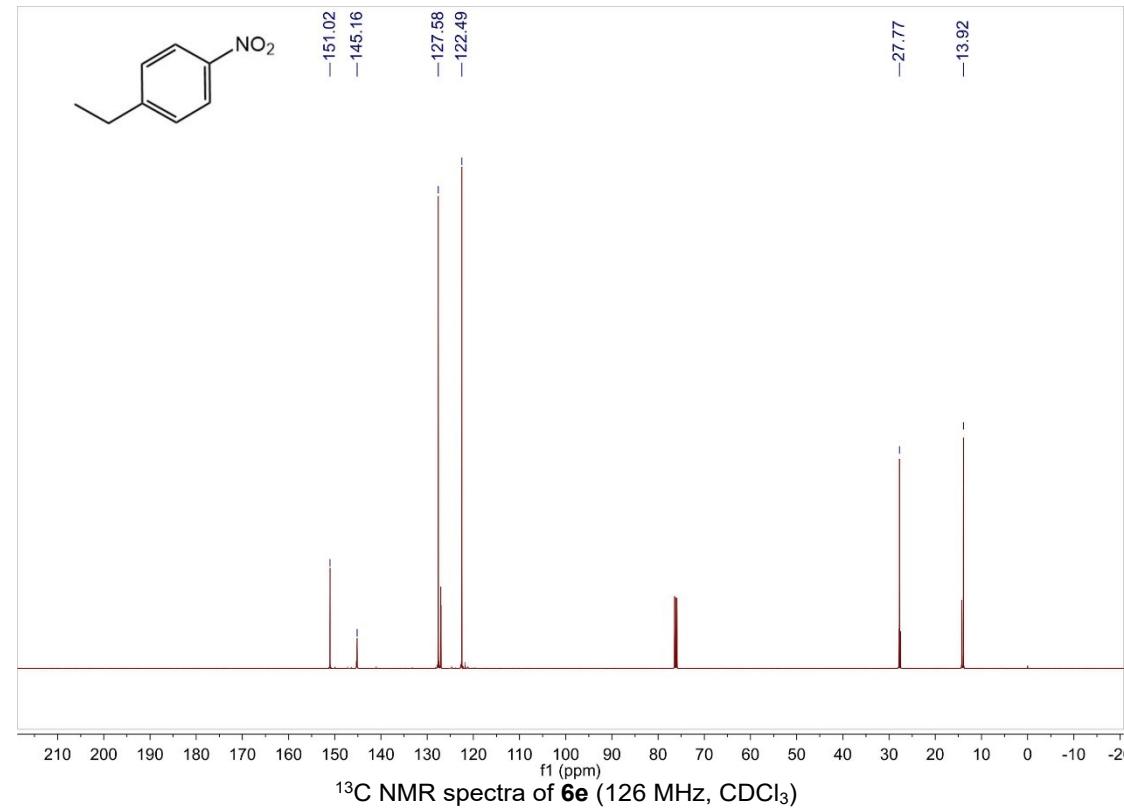
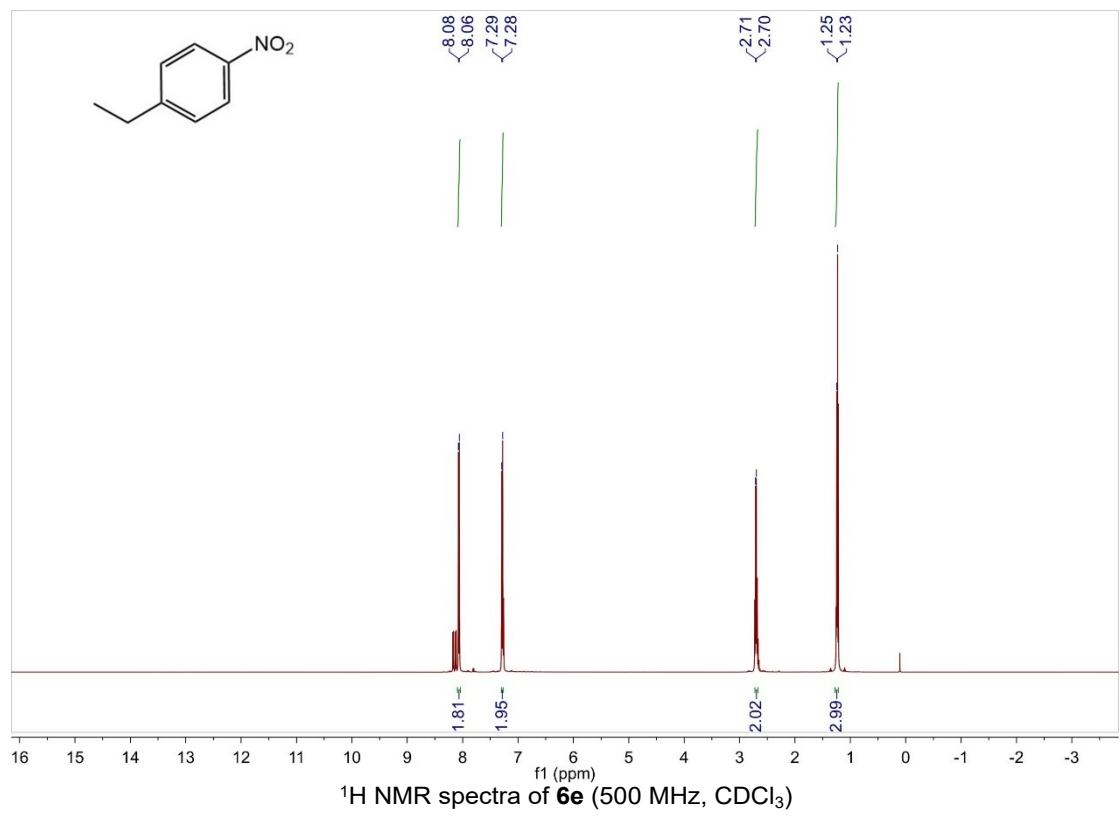


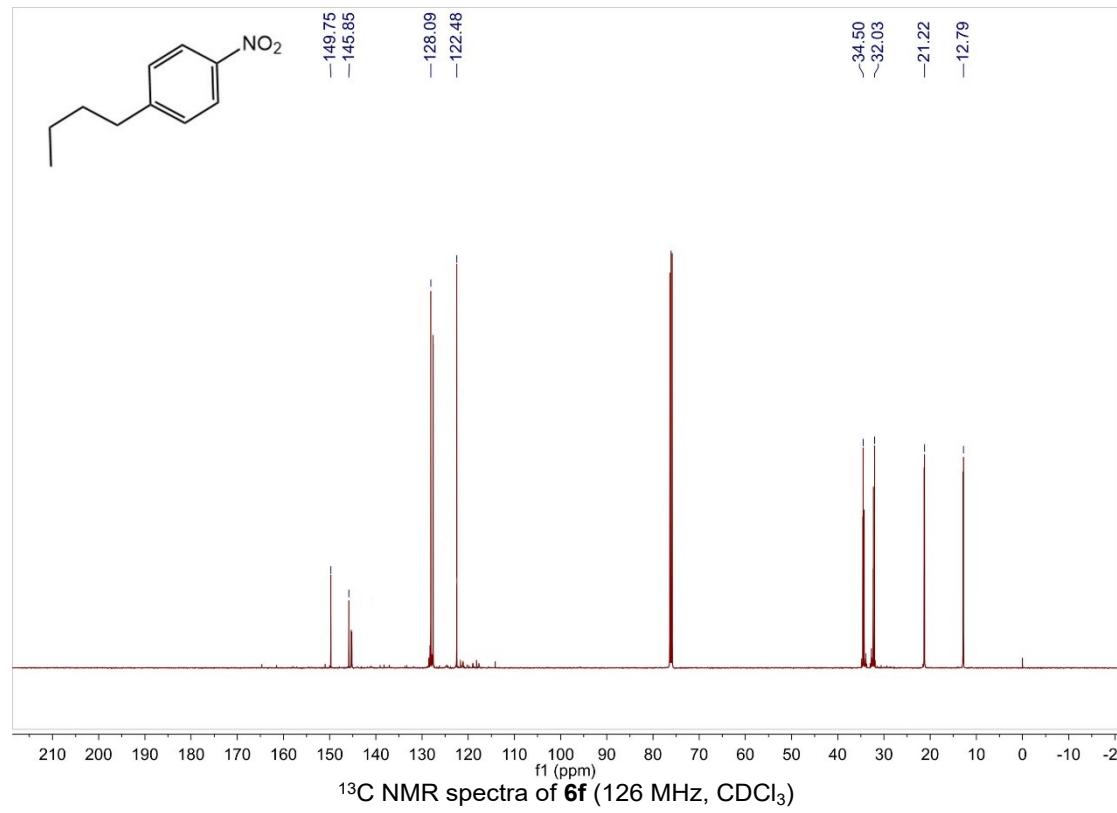
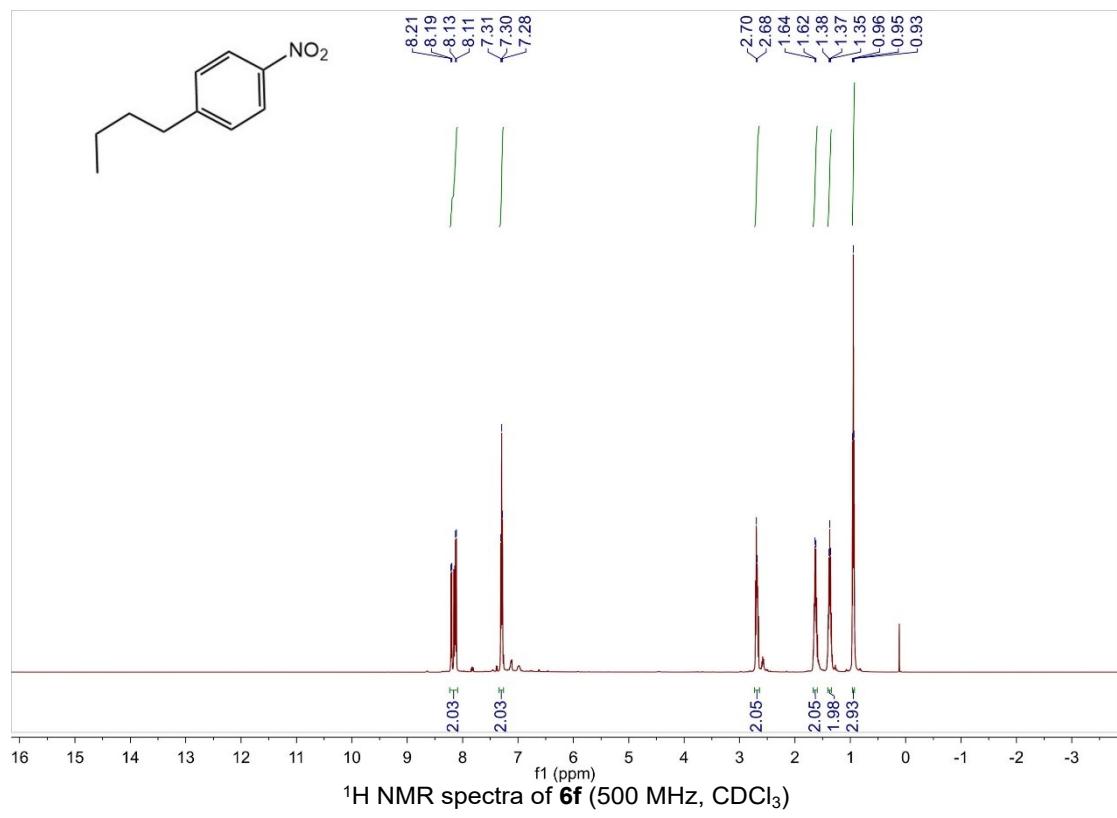


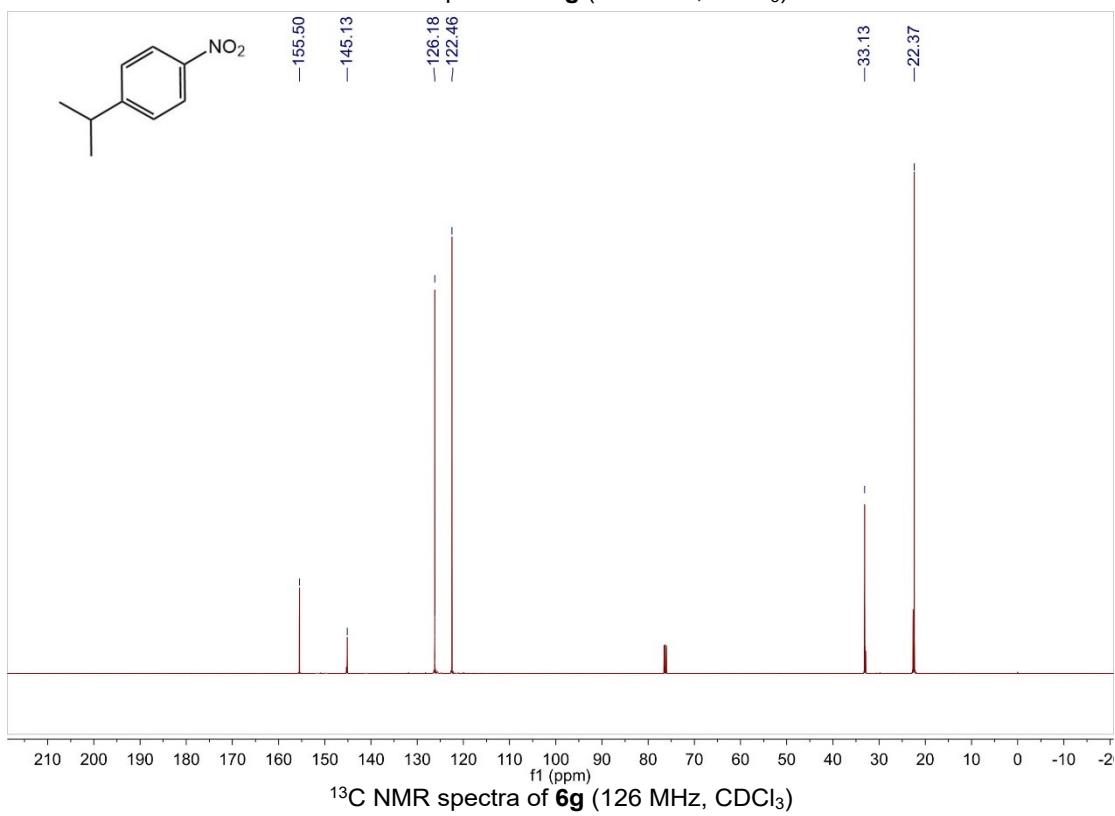
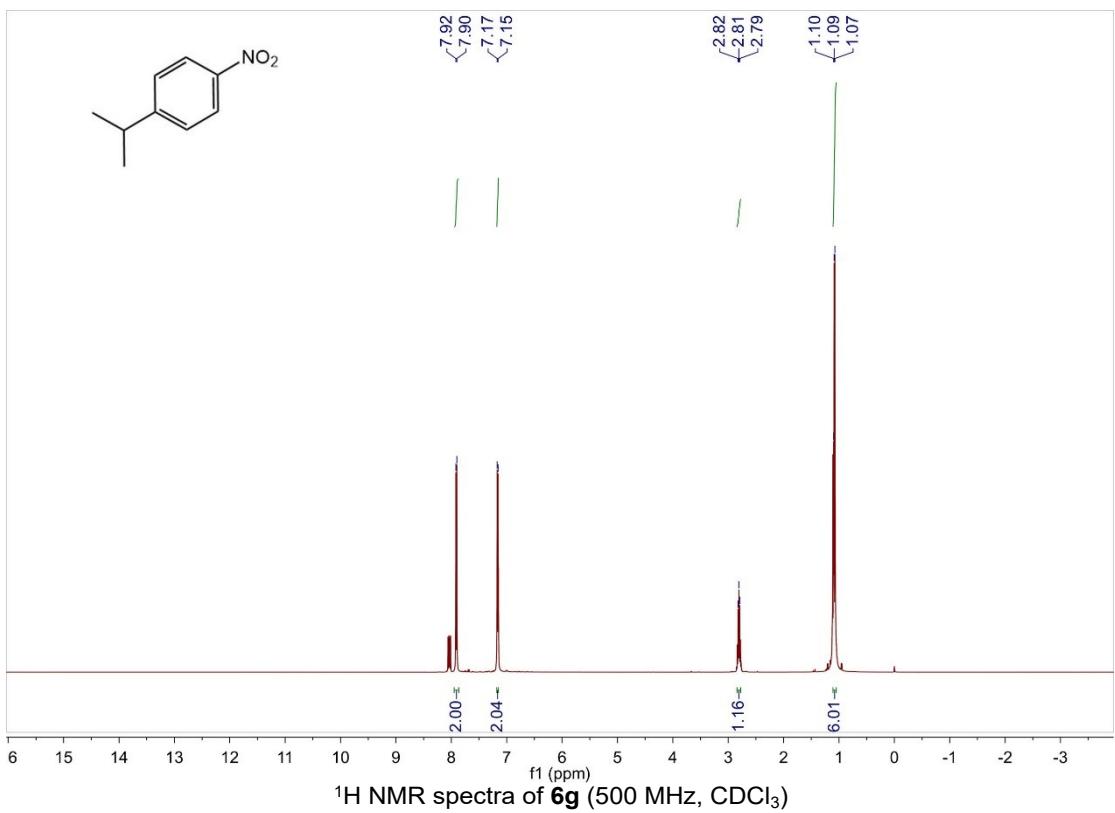


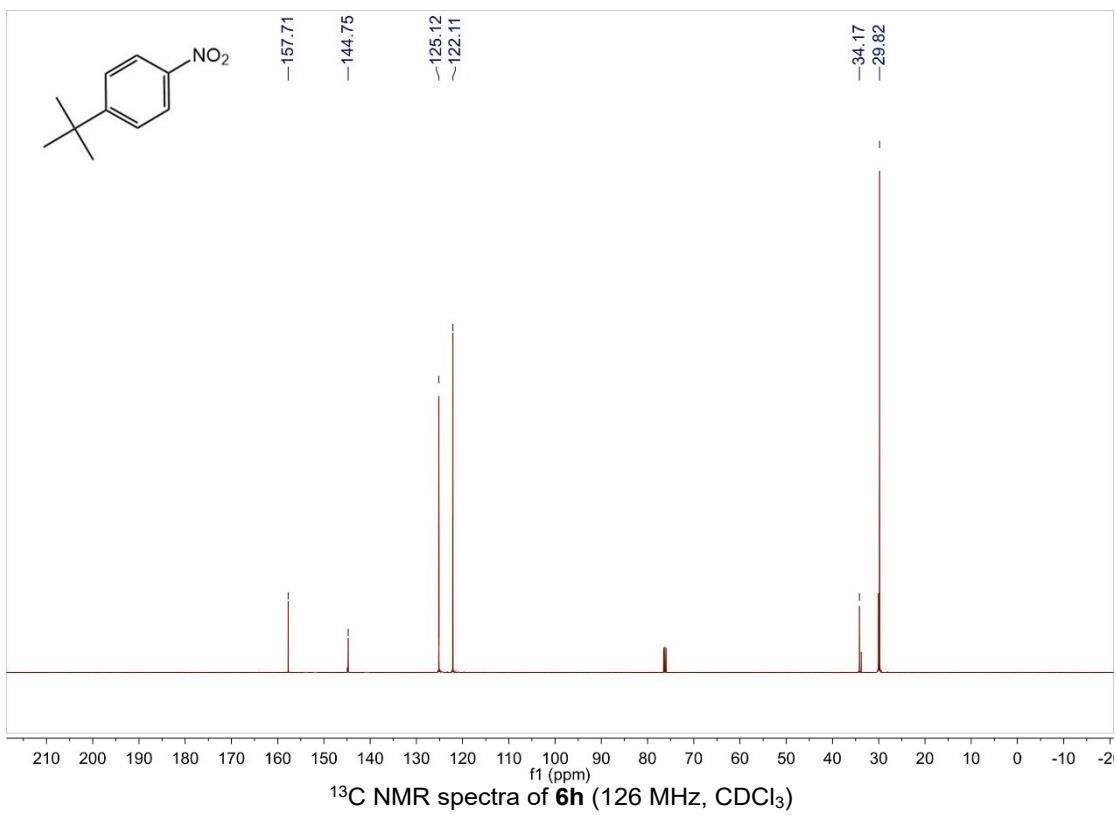
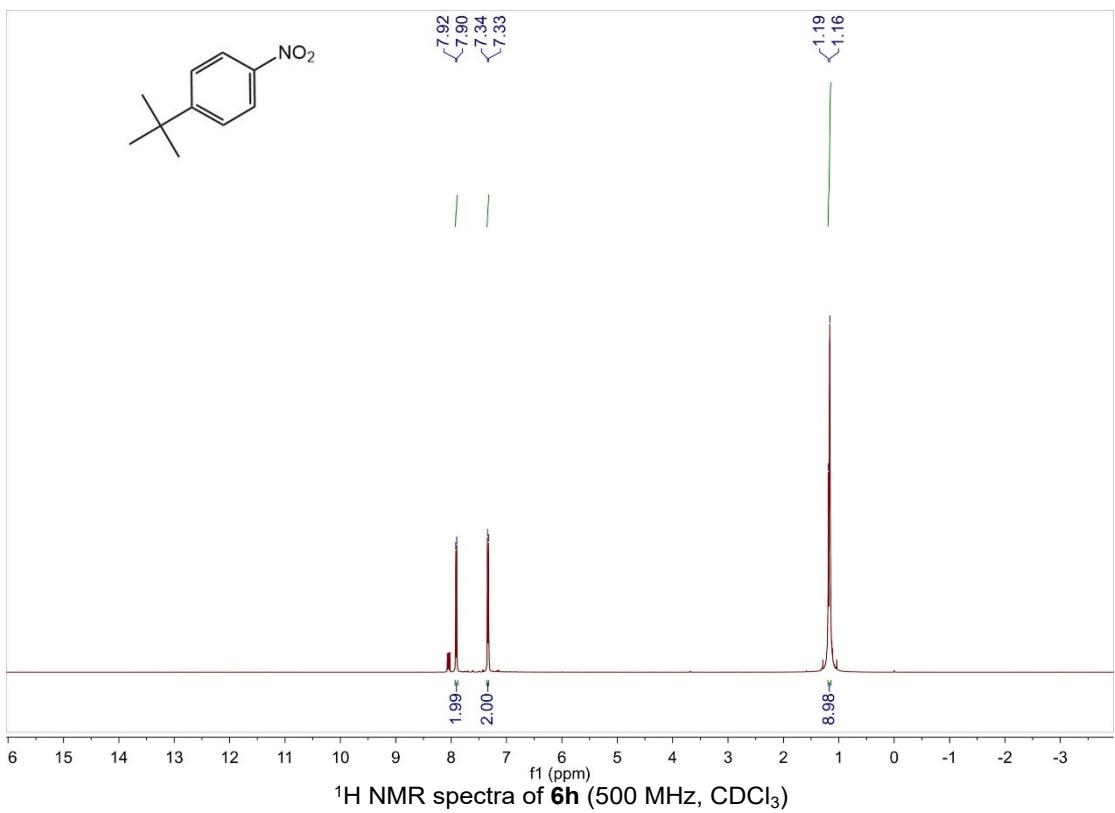


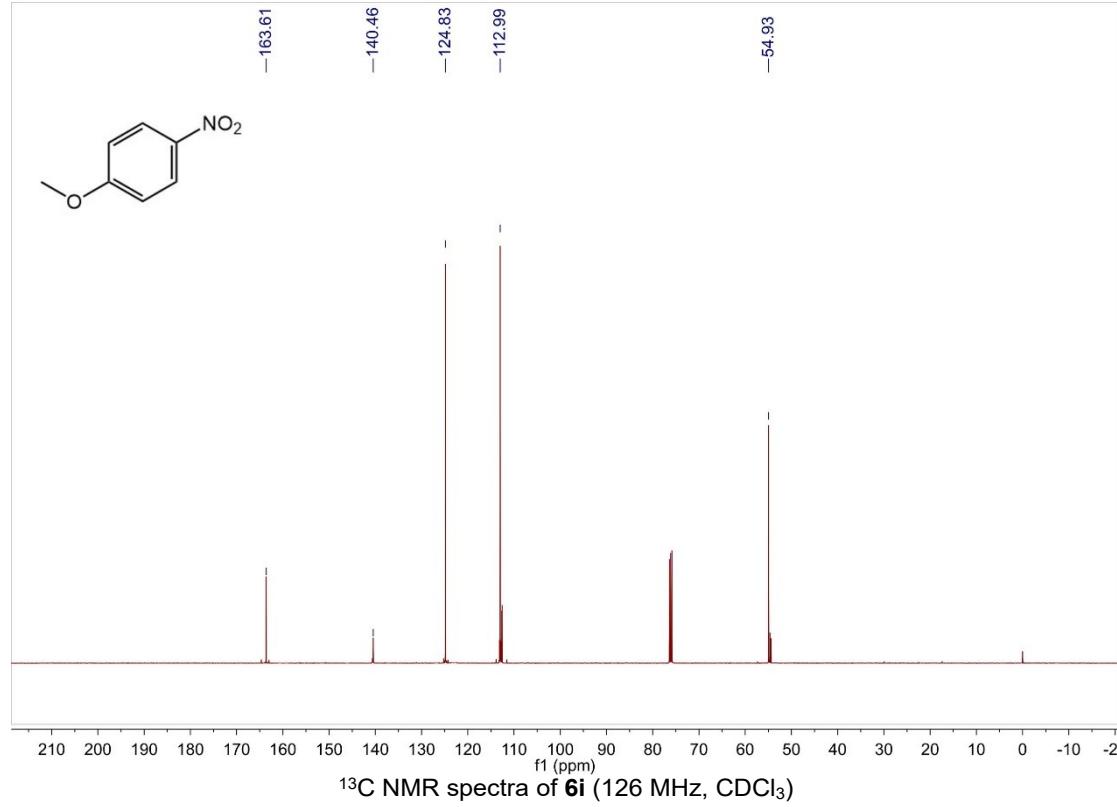
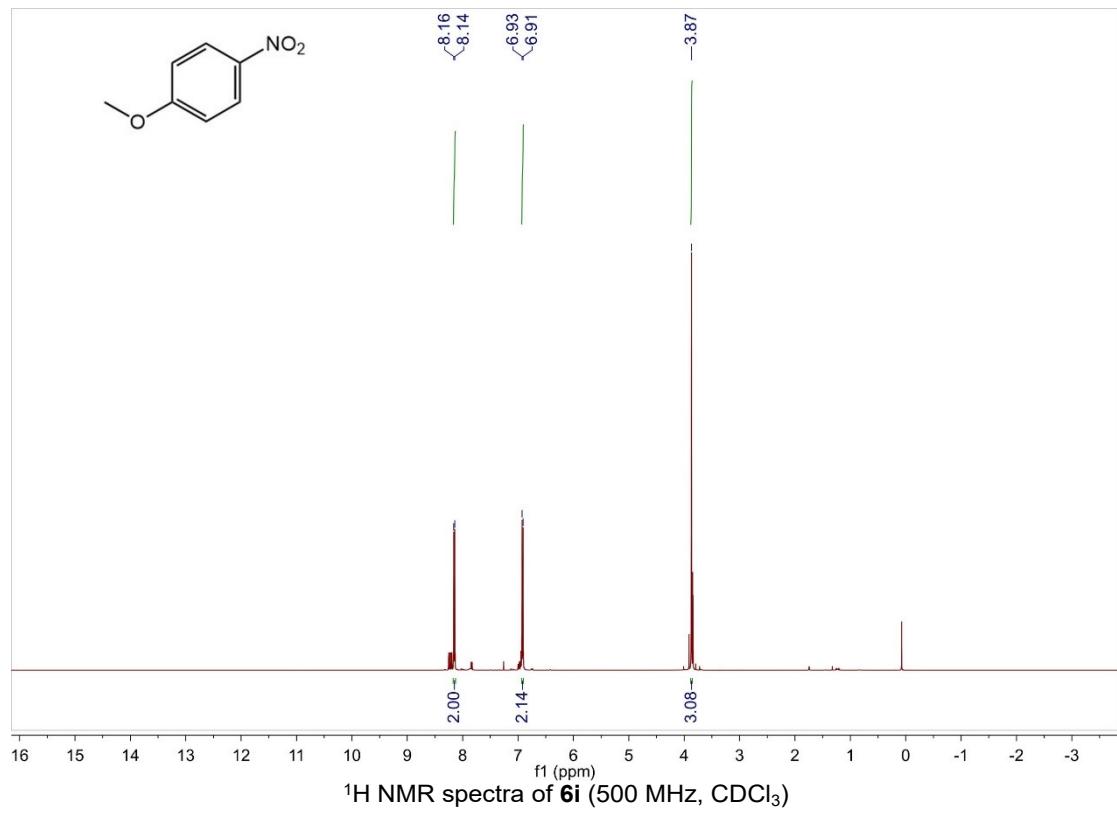


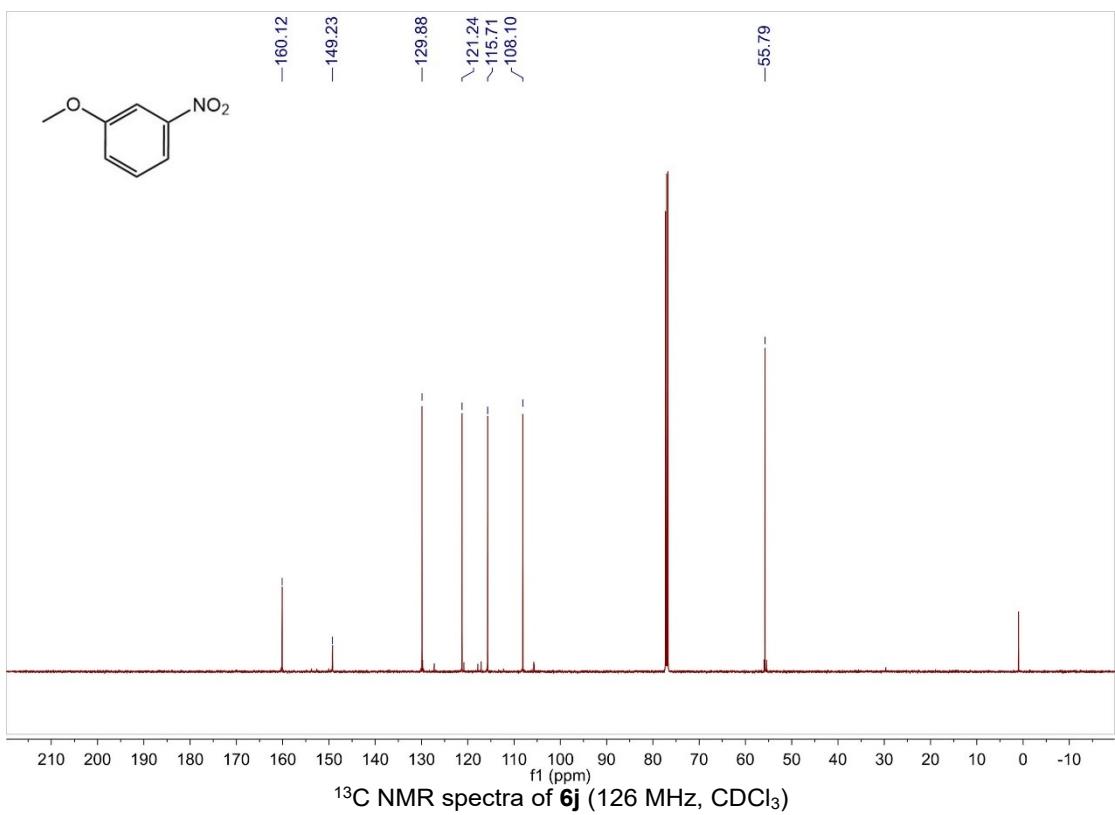
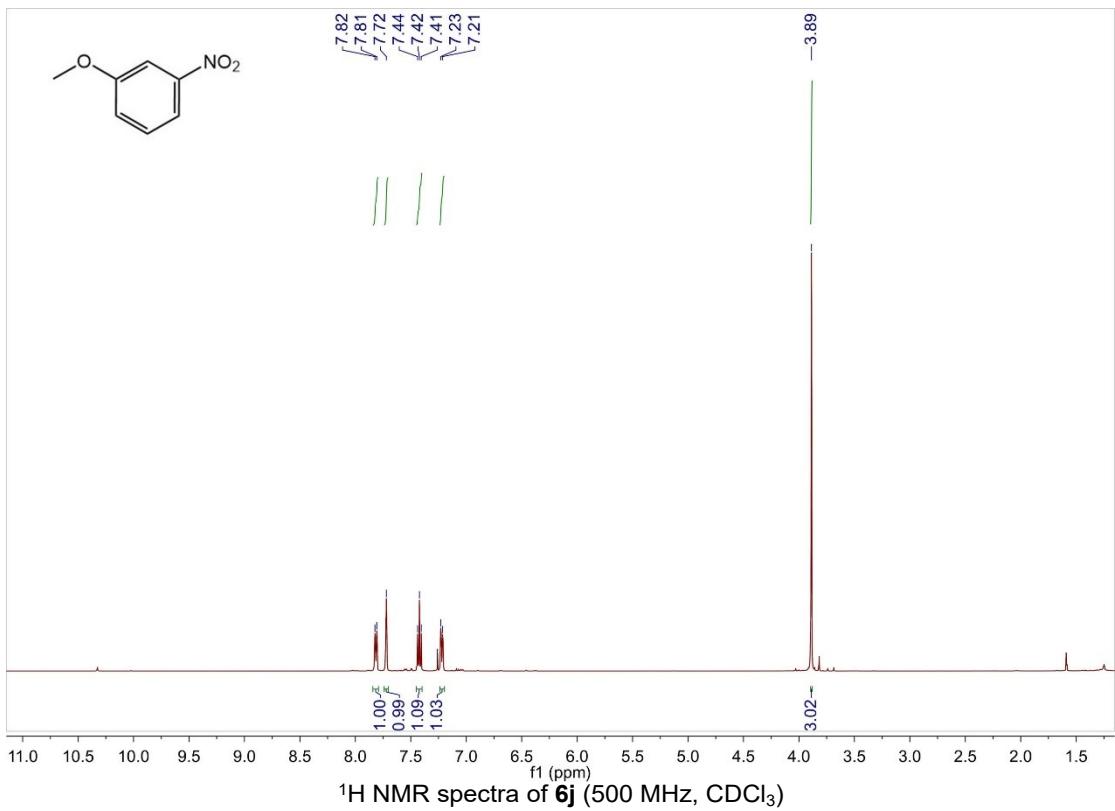


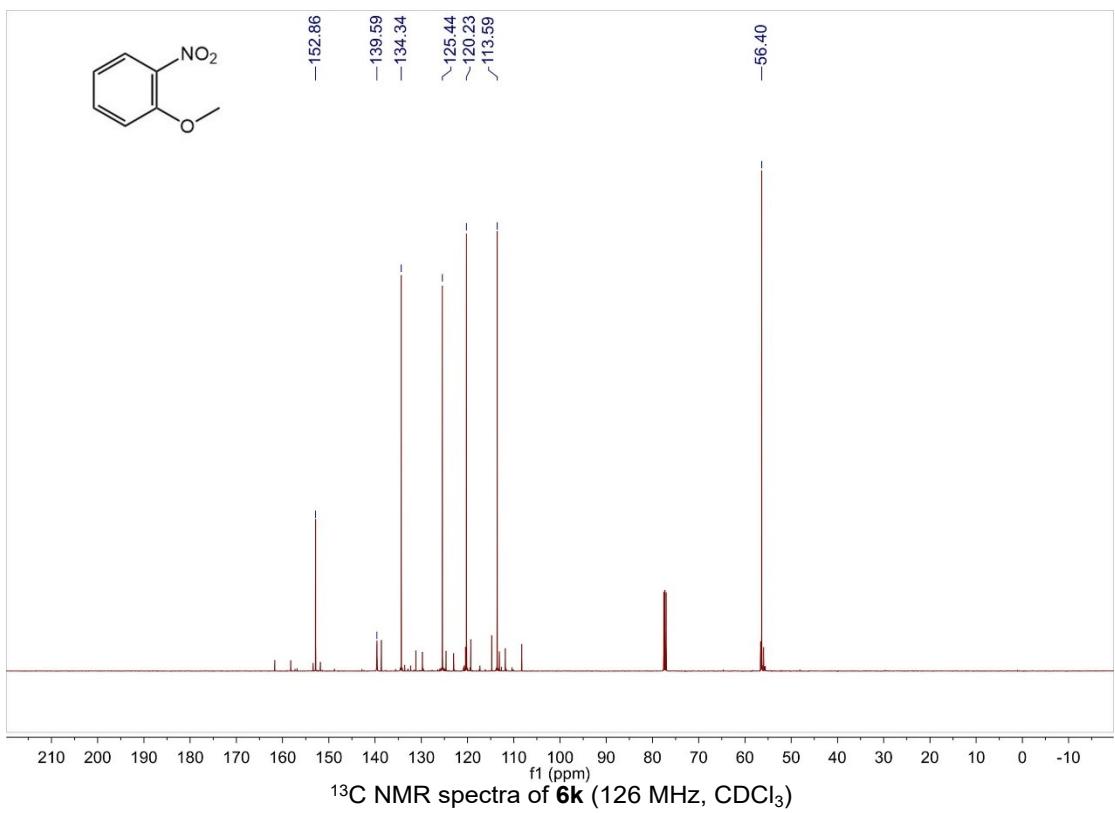
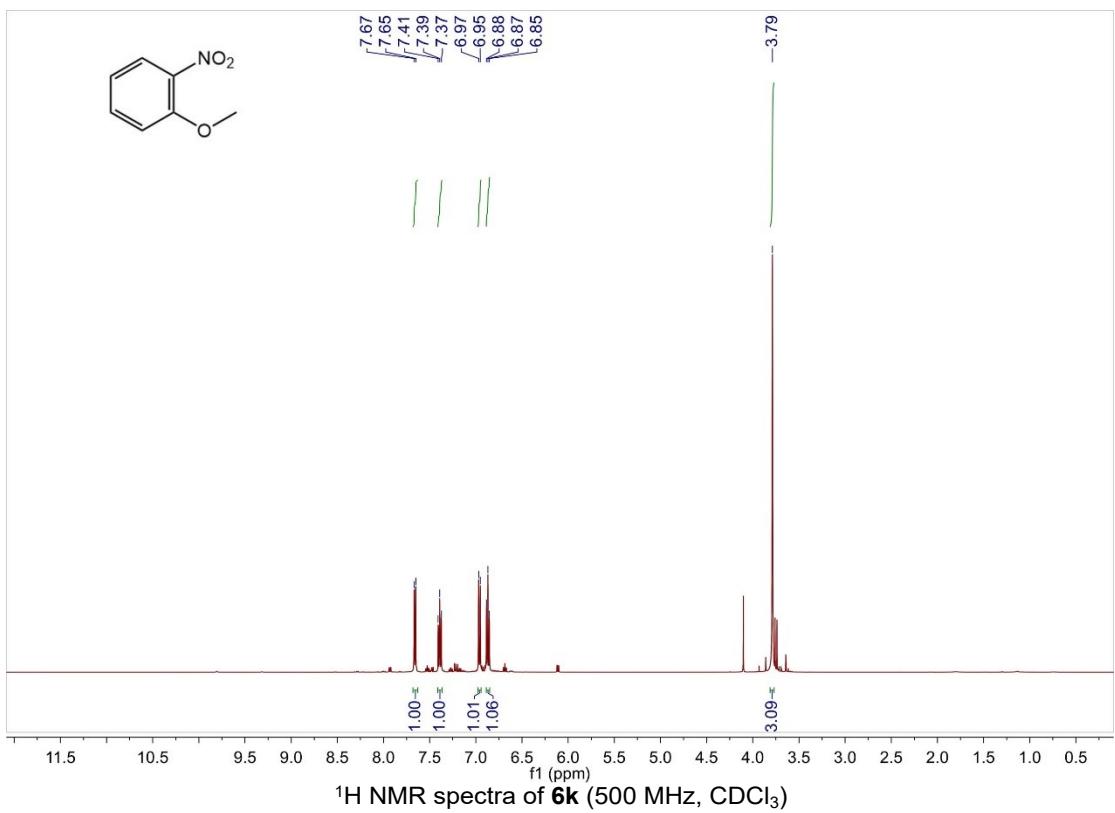


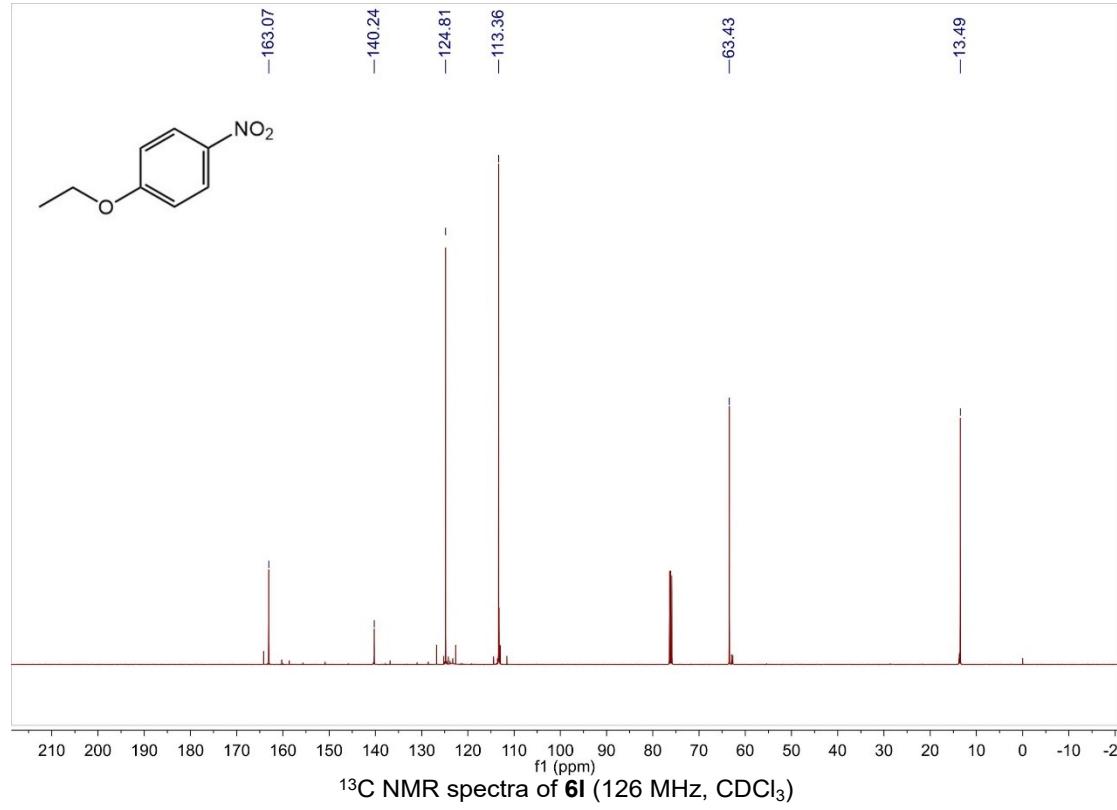
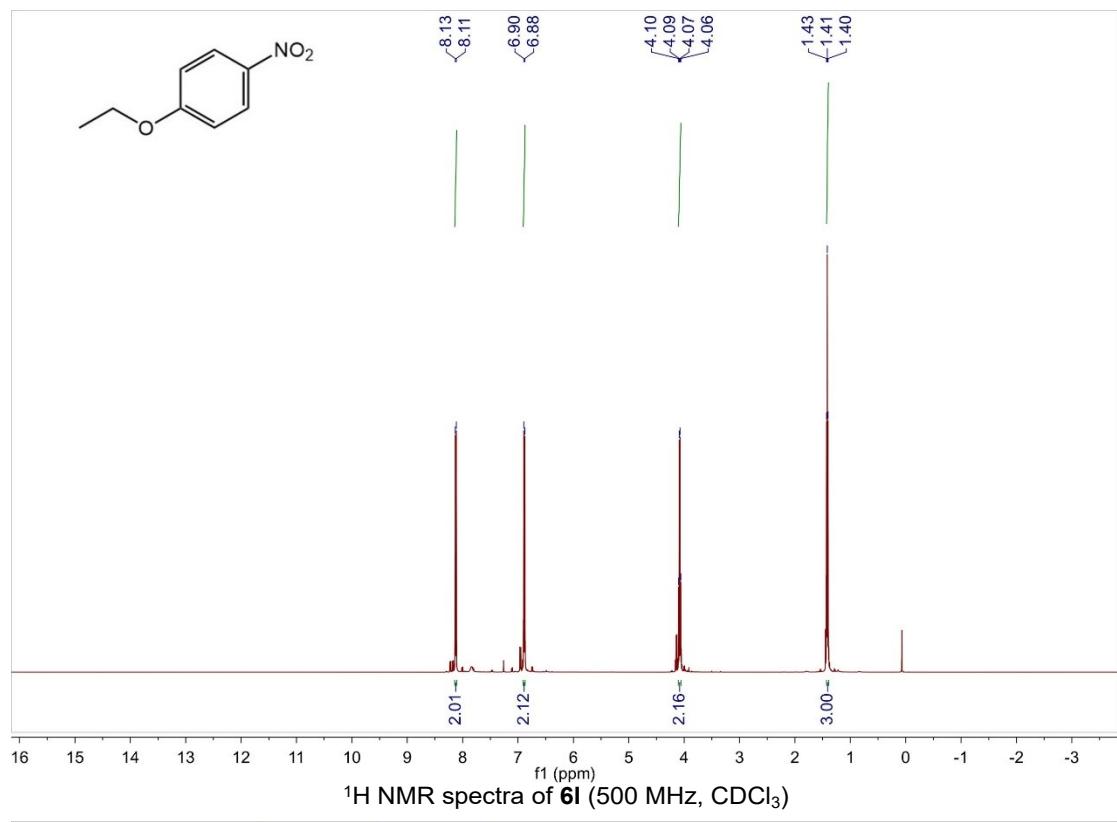


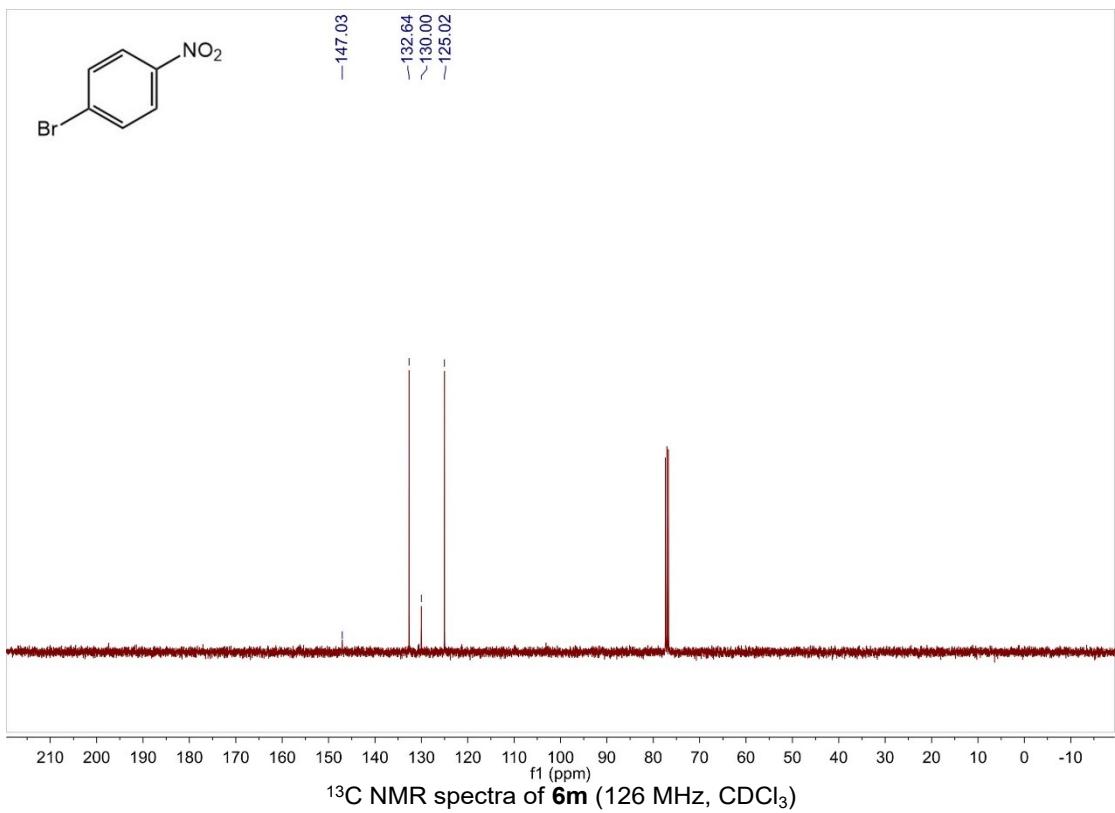
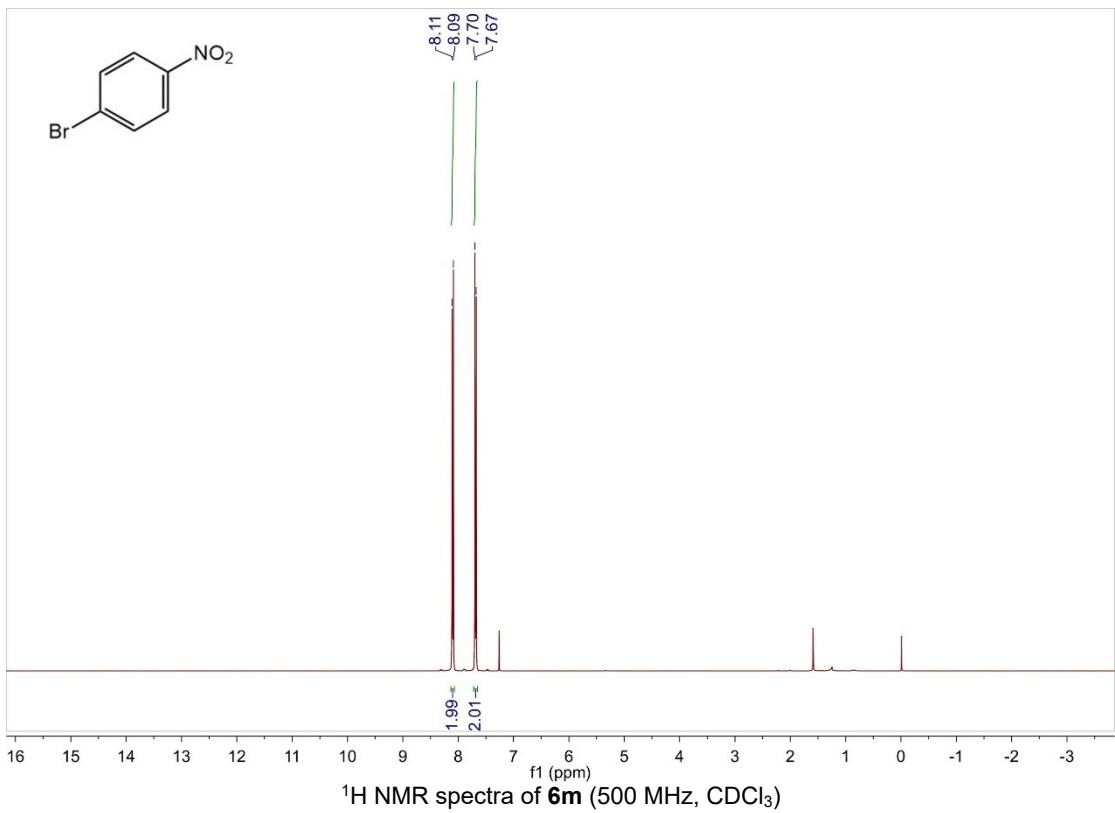


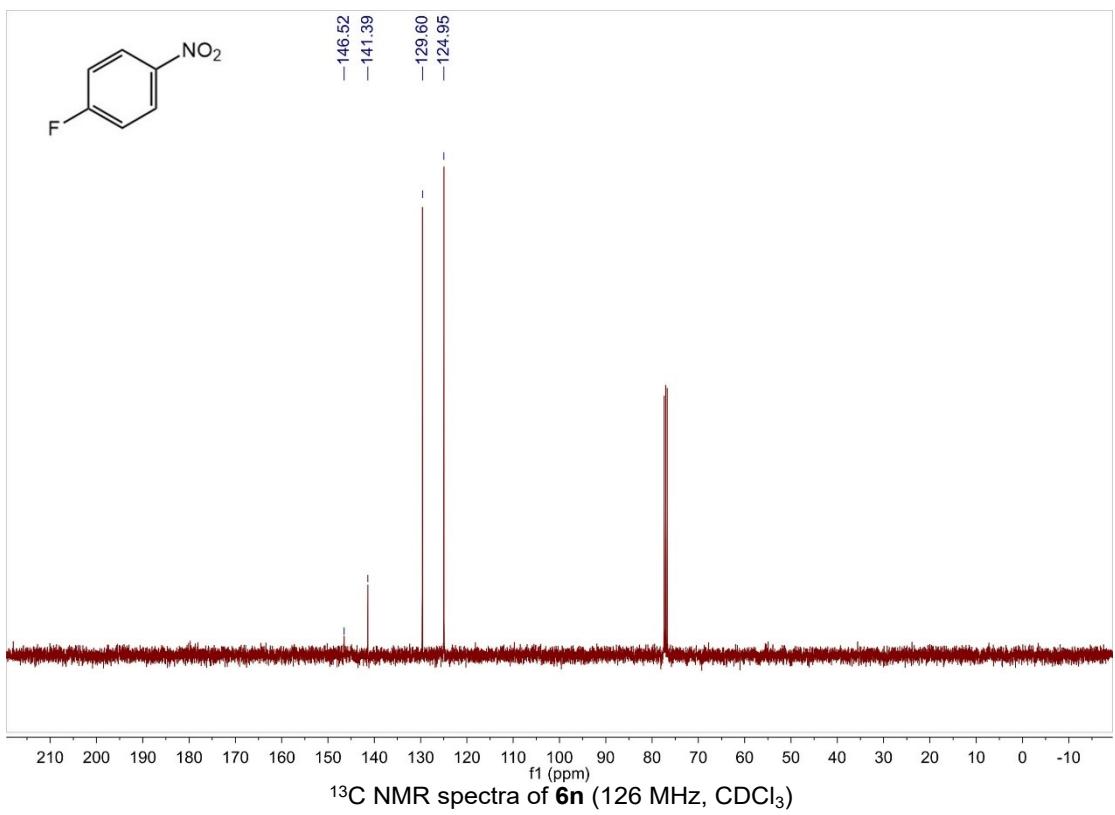
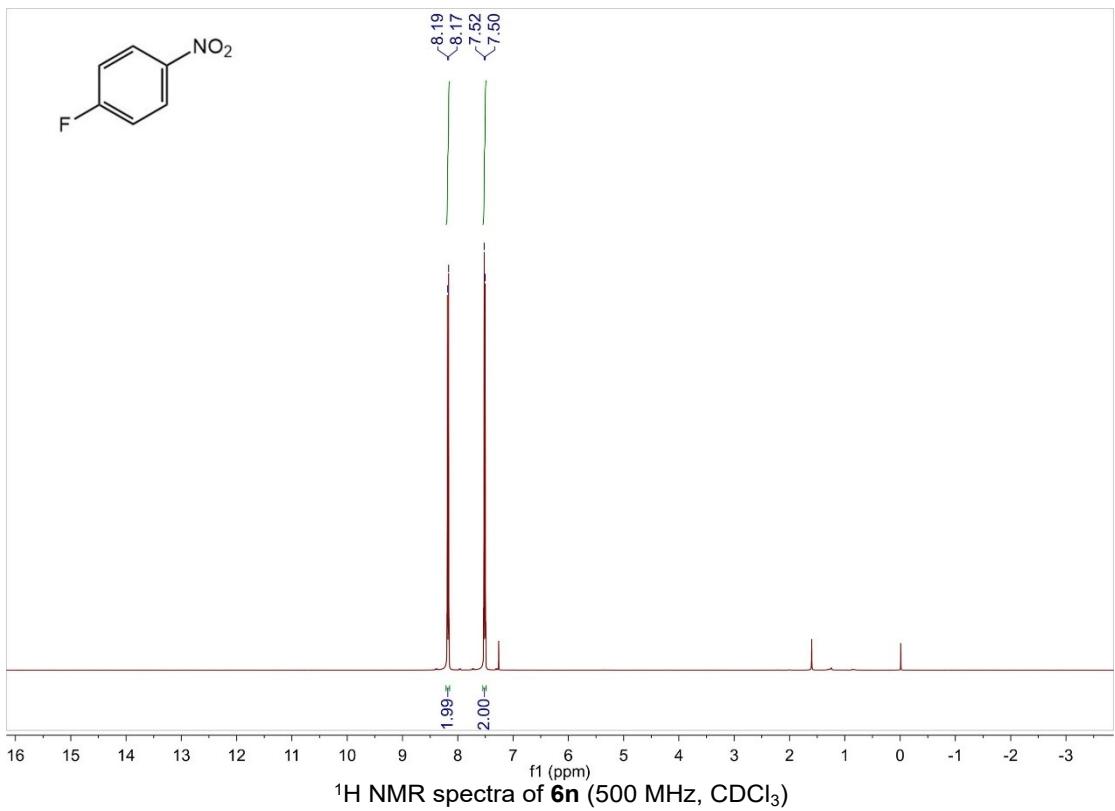


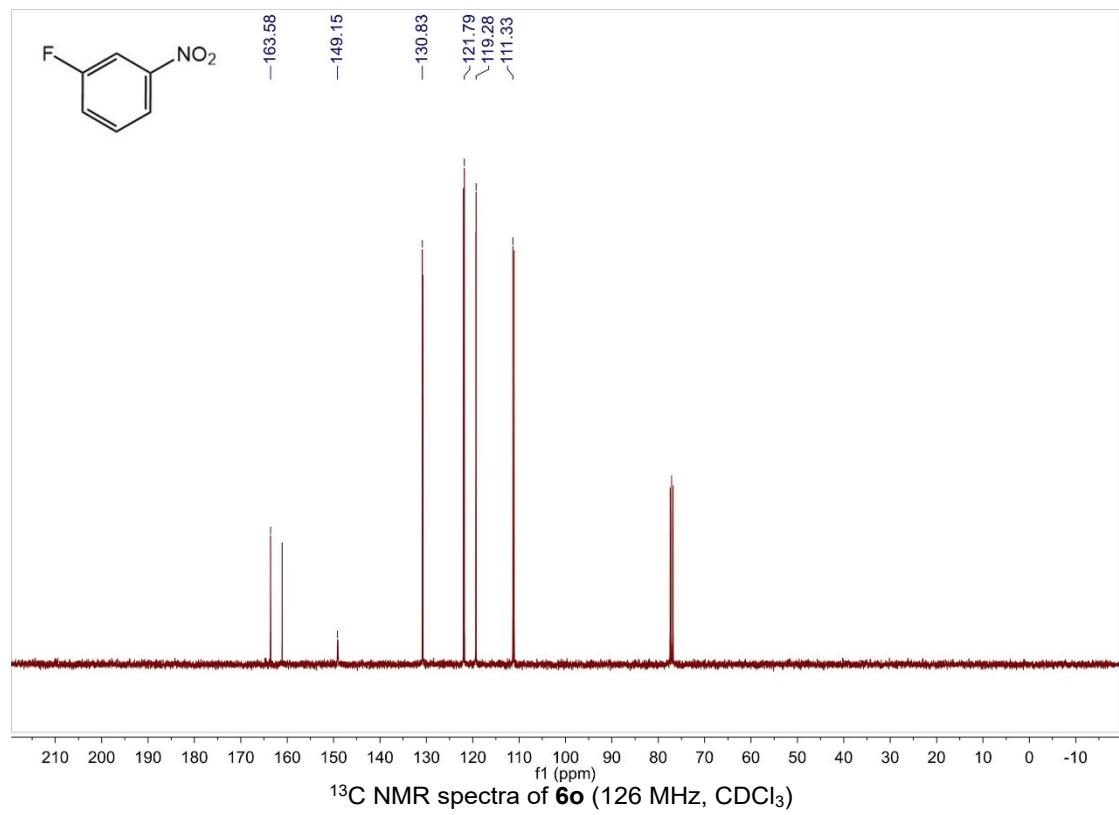
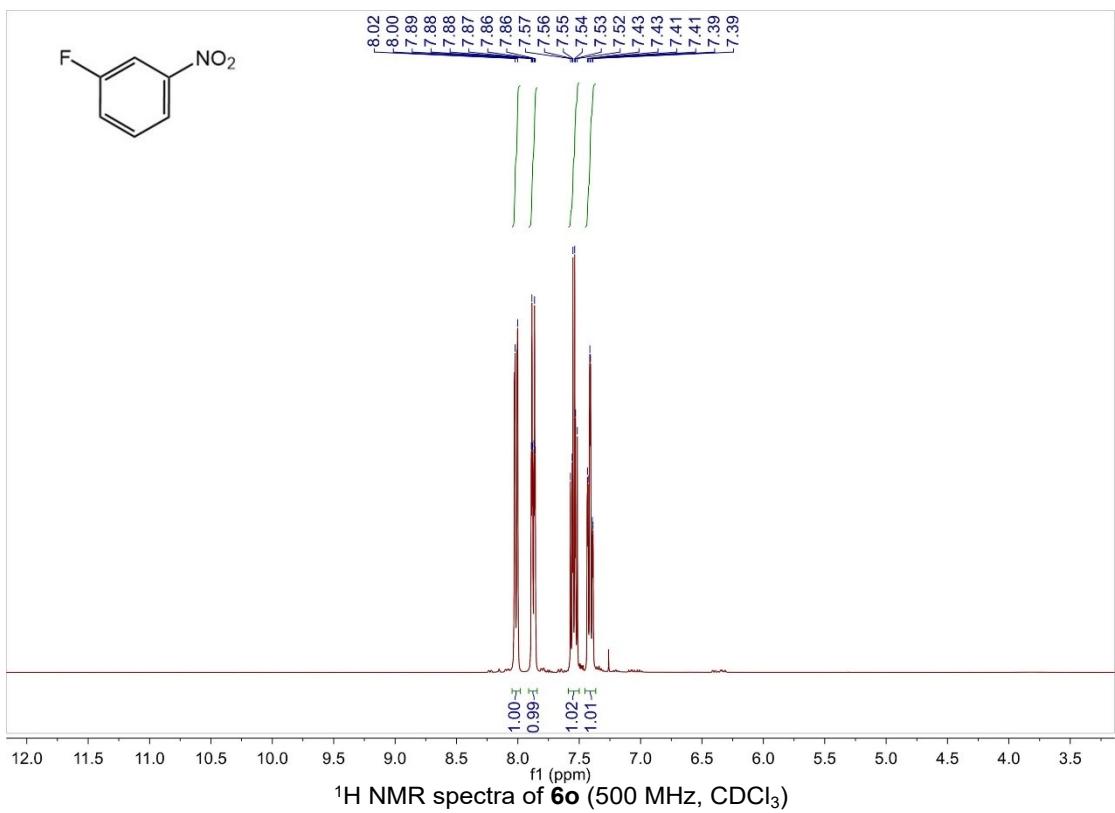












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