

Amorphous Carbon Coating Enhances Activity of High Rate CO₂ Electroreduction to CO

Yiwen Ma, Wenzhe Niu, Wenjuan Shi, Xiaoxiong Huang, Yi Liu,
Junfeng Chen, Liangyao Xue and Bo Zhang*

State Key Laboratory of Molecular Engineering of Polymers, Department
of Macromolecular Science, Fudan University, Shanghai 200438,

*E-mail addresses: bozhang@fudan.edu.cn

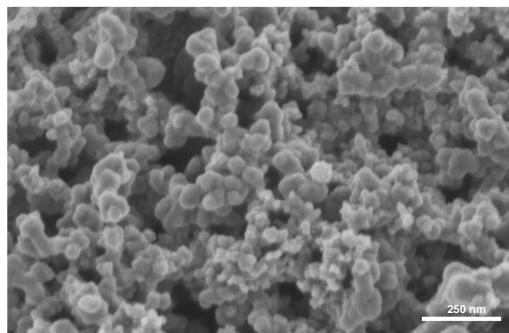
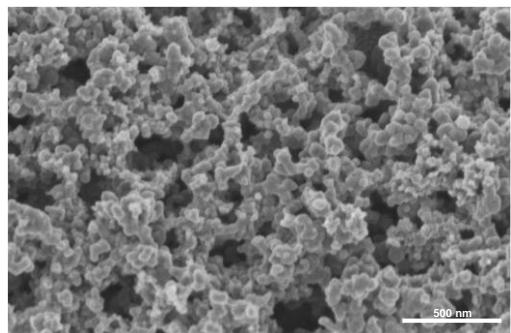


Fig. S1. The scanning electron microscopy images of Ag/C.

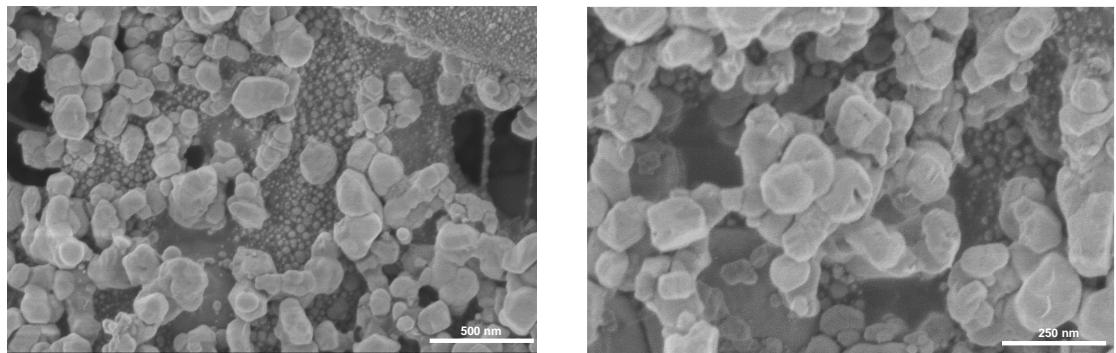


Fig. S2. The scanning electron microscopy images of Ag.

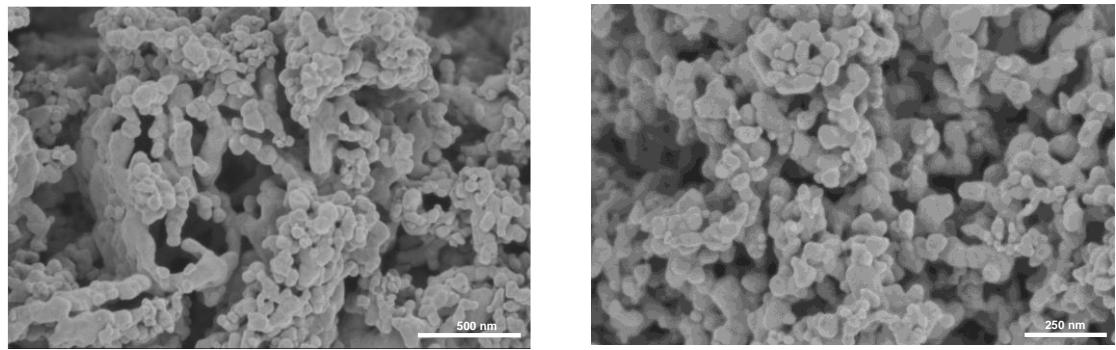


Fig. S3. The scanning electron microscopy images of Ag nano.

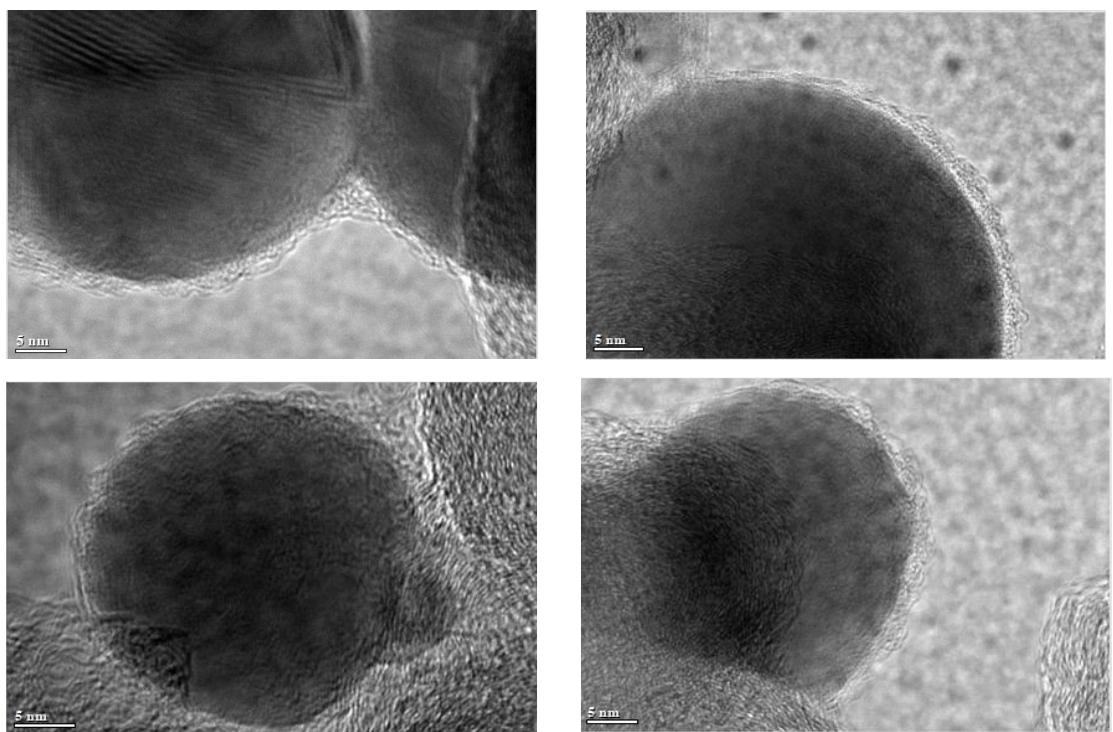


Fig. S4. The different high-resolution transmission electron microscopy images of the Ag/C.

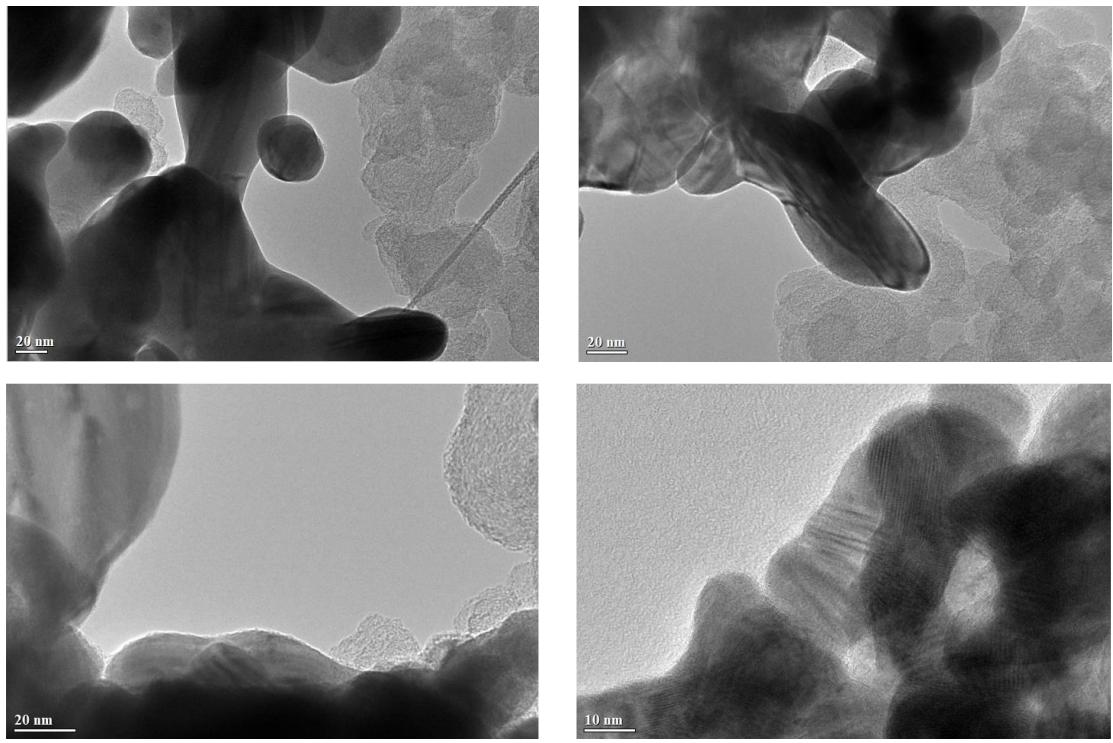


Fig. S5. The different high-resolution transmission electron microscopy images of the Ag-C.

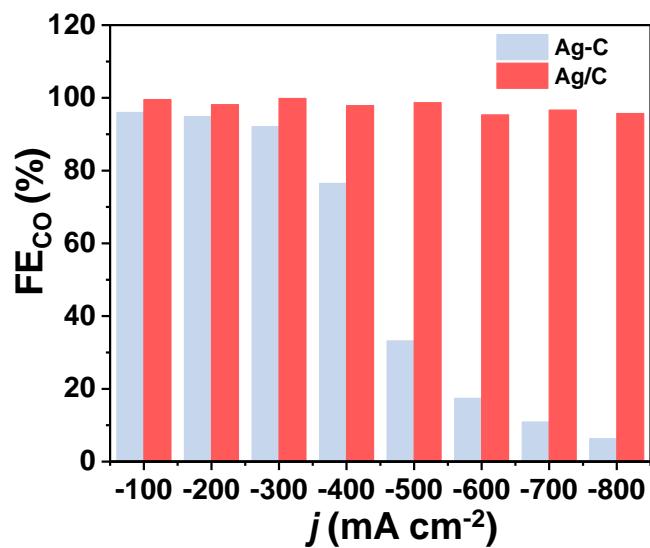


Fig. S6. Faradaic efficiency of CO of the Ag/C and Ag-C catalysts in 1 M KOH.

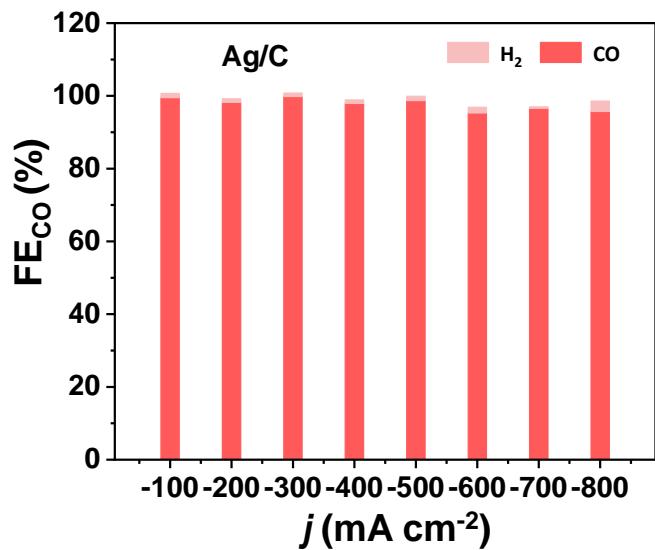


Fig. S7. Faradaic efficiencies of CO and H₂ on the Ag/C in 1 M KOH.

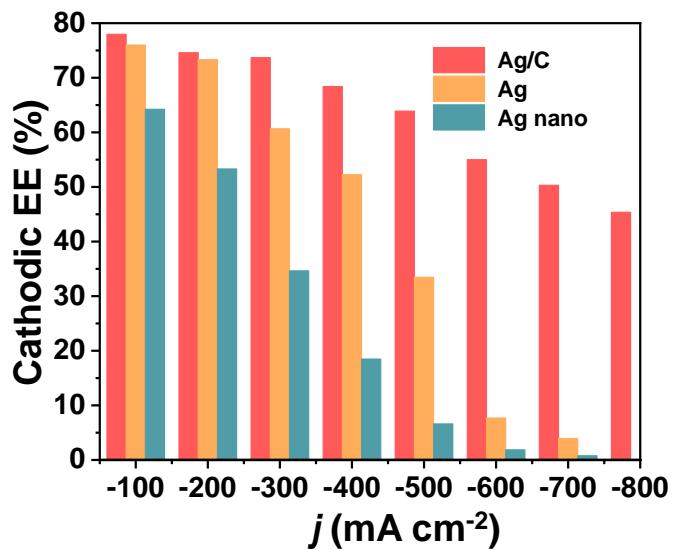


Fig. S8. Cathodic energy efficiency at 700 mA cm^{-2} of Ag nano, Ag, Ag/C in 1 M KOH.

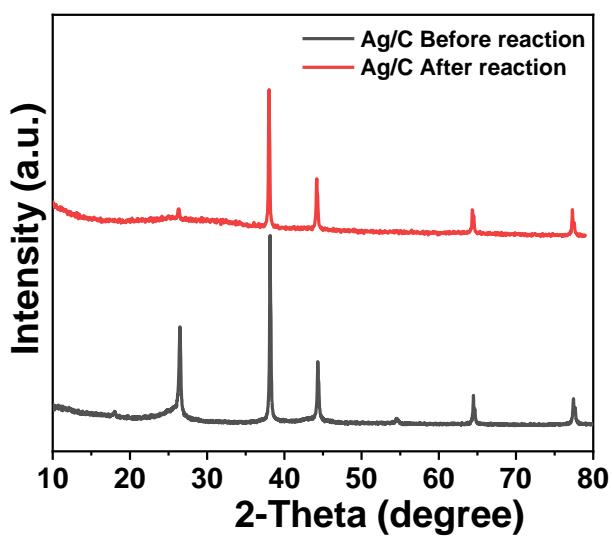


Fig. S9. X-ray diffraction patterns of the Ag/C before and after reaction.

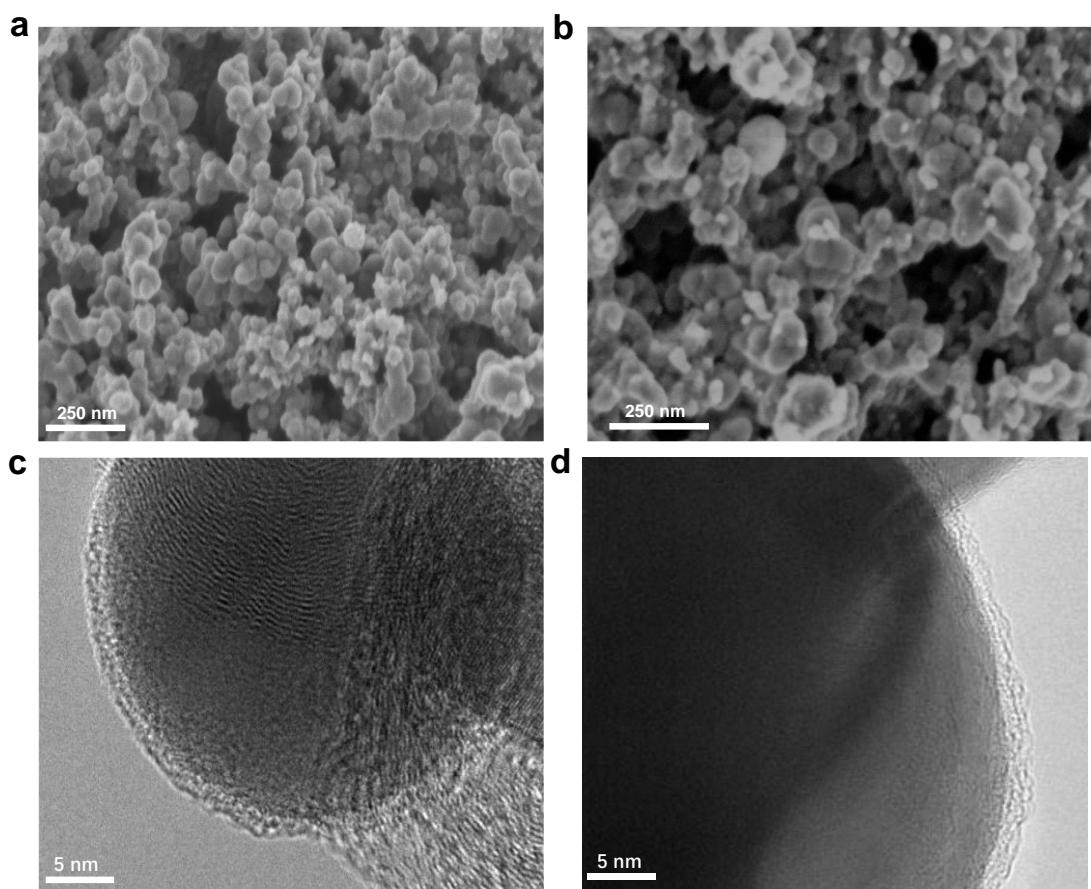


Fig. S10. (a) Scanning electron microscopy and (c) high-resolution transmission electron microscopy images of the Ag/C before reaction. (b) scanning electron microscopy and (d) high-resolution transmission electron microscopy images of the Ag/C after reaction.

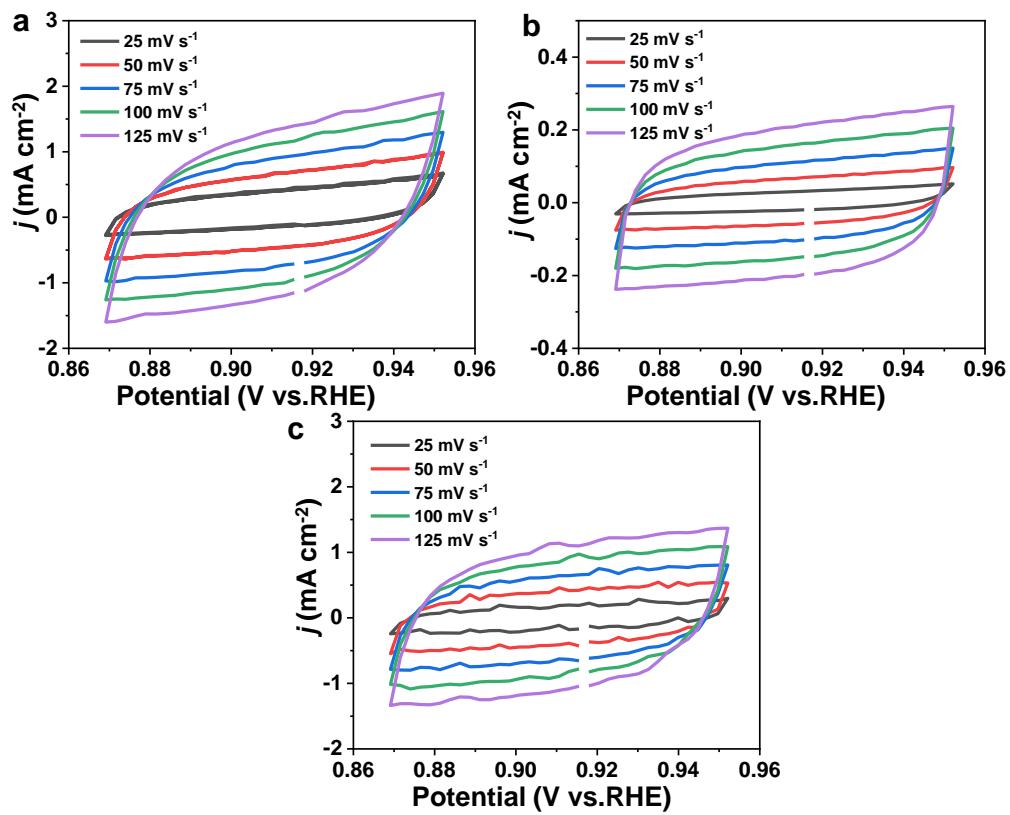


Fig. S11. Cyclic voltammetry curves of Ag nano (a), Ag (b), Ag/C (c) at different scan rates of $25, 50, 75, 100$ and 125 mV s^{-1} collected.

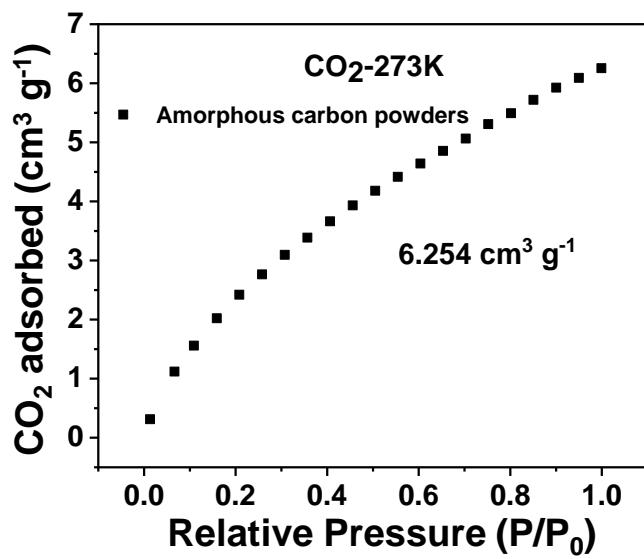


Fig. S12. CO₂ adsorption isotherm curve of amorphous carbon powders.

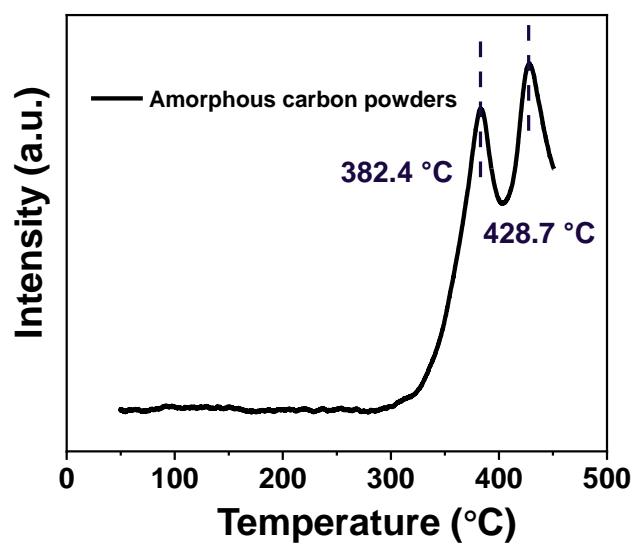


Fig. S13. Temperature programmed desorption of CO₂ measurement (CO₂-TPD) profiles of amorphous carbon powders.

Table S1. The crystal sizes and full width at half maxima (FWHM) values of the (111) plane of the catalysts.

Catalyst	Crystallite sizes(nm)	FWHM
Ag nano	27.1	0.348
Ag	45.3	0.270
Ag/C	47.5	0.222

Table S2. Parameter values of equivalent circuit components.

Catalyst	$R_s(\Omega)$	$R_{ct}(\Omega)$
Ag nano	7.189	5.629
Ag	6.802	1.681
Ag/C	6.884	0.644

Table S3. Performance of CO₂ reduced to CO on contrast catalysts.

Catalysts	j (mA cm ⁻²)	FEco (%)	References
Ag/C	-800	95	This work
	-700	96.6	
	-600	95.3	
Ag powder	-480	91.2	1 ¹
Ag/MPL-3C	-100	98.80	2 ²
sputtered Ag/PTFE	-180	89.70	3 ³
Ag/PTFE	-253	84.30	4 ⁴
Ag NP	-281	97.5	5 ⁵
Ag NP	-248	94.70	6 ⁶
Ag/C	-231	83.5	7 ⁷
Ag DAT	-109	93	8 ⁸
Ag-NOLI	-500	84	9 ⁹
AgSn	-200	100	10 ¹⁰
AgNP/MWCNT	-368	95	11 ¹¹
Ag-alloyed Zn	-400	72	12 ¹²
Au25/C	-600	90	13 ¹³
AuCu	-104	75	14 ¹⁴
AuCuB	-76	99	15 ¹⁵
MWNT/PyPBI/Au	-267	60	16 ¹⁶
h-NiNC	-513	90	17 ¹⁷
SbCu	-497	91	18 ¹⁸
ZnAg	-500	74	19 ¹⁹

References

1. S. Verma, X. Lu, S. Ma, R. I. Masel and P. J. Kenis, *Phys Chem Chem Phys*, 2016, **18**, 7075-7084.
2. R. Wang, H. Haspel, A. Pustovarenko, A. Dikhtierenko, A. Russkikh, G. Sherk, D. Osadchii, S. Ould-Chikh, M. Ma, W. A. Smith, K. Takanabe, F. Kapteijn and J. Gascon, *ACS Energy Letters*, 2019, **4**, 2024-2031.
3. C.-T. Dinh, F. P. García de Arquer, D. Sinton and E. H. Sargent, *ACS Energy Letters*, 2018, **3**, 2835-2840.
4. C. M. Gabardo, A. Seifitokaldani, J. P. Edwards, C.-T. Dinh, T. Burdyny, M. G. Kibria, C. P. O'Brien, E. H. Sargent and D. Sinton, *Energy & Environmental Science*, 2018, **11**, 2531-2539.
5. B. Kim, F. Hillman, M. Ariyoshi, S. Fujikawa and P. J. A. Kenis, *Journal of Power Sources*, 2016, **312**, 192-198.
6. S. Ma, R. Luo, S. Moniri, Y. Lan and P. J. A. Kenis, *Journal of The Electrochemical Society*, 2014, **161**, F1124-F1131.
7. J. Hong, K. T. Park, Y. E. Kim, D. Tan, Y. E. Jeon, J. E. Park, M. H. Youn, S. K. Jeong, J. Park, Y. N. Ko and W. Lee, *Chemical Engineering Journal*, 2022, **431**.
8. C. E. Tornow, M. R. Thorson, S. Ma, A. A. Gewirth and P. J. Kenis, *J Am Chem Soc*, 2012, **134**, 19520-19523.
9. D. Kim, S. Yu, F. Zheng, I. Roh, Y. Li, S. Louisia, Z. Qi, G. A. Somorjai, H. Frei, L.-W. Wang and P. Yang, *Nature Energy*, 2020, **5**, 1032-1042.
10. C. Cai, B. Liu, K. Liu, P. Li, J. Fu, Y. Wang, W. Li, C. Tian, Y. Kang, A. Stefancu, H. Li, C. W. Kao, T. S. Chan, Z. Lin, L. Chai, E. Cortes and M. Liu, *Angew Chem Int Ed Engl*, 2022, **61**, e202212640.
11. S. Ma, R. Luo, J. I. Gold, A. Z. Yu, B. Kim and P. J. A. Kenis, *J. Mater. Chem. A*, 2016, **4**, 8573-8578.
12. S. Lamaison, D. Wakerley, J. Blanchard, D. Montero, G. Rousse, D. Mercier, P. Marcus, D. Taverna, D. Giaume, V. Mougel and M. Fontecave, *Joule*, 2020, **4**, 395-406.
13. B. Kim, H. Seong, J. T. Song, K. Kwak, H. Song, Y. C. Tan, G. Park, D. Lee and J. Oh, *ACS Energy Lett.*, 2019, **5**, 749-757.
14. Y. Han, Z. Wang, X. Han, W. Fang, Y. Zhou, K. Lei, B. You, H. S. Park and B. Y. Xia, *ACS Sustain. Chem. Eng.*, 2021, **9**, 2609-2615.
15. Y. Liu, Y. Fang, Q. Yuan, J. Lu and H. Wang, *Green Chem.*, 2023, **25**, 1339-1344.
16. H. M. Jhong, C. E. Tornow, C. Kim, S. Verma, J. L. Oberst, P. S. Anderson, A. A. Gewirth, T. Fujigaya, N. Nakashima and P. J. A. Kenis, *Chemphyschem*, 2017, **18**, 3274-3279.
17. Y. Chen, J. Zhang, J. Tian, Y. Guo, F. Xu, Y. Zhang, X. Wang, L. Yang, Q. Wu and Z. Hu, *Adv. Funct. Mater.*, 2023, 2214658.
18. J. Li, H. Zeng, X. Dong, Y. Ding, S. Hu, R. Zhang, Y. Dai, P. Cui, Z. Xiao, D. Zhao, L. Zhou, T. Zheng, J. Xiao, J. Zeng and C. Xia, *Nat Commun*, 2023, **14**, 340.
19. S. Lamaison, D. Wakerley, F. Kracke, T. Moore, L. Zhou, D. U. Lee, L. Wang, M. A. Hubert, J. E. Aviles Acosta, J. M. Gregoire, E. B. Duoss, S. Baker, V. A. Beck, A. M. Spormann, M. Fontecave, C. Hahn and T. F. Jaramillo, *Adv Mater*, 2022, **34**, e2103963.