Electronic Supporting Information

Enhancement of catalytic hydrolysis activity for organophosphates by the Metal-Organic Framework MOF-808-NH₂ via post-synthetic modification.

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Abstract: Metal-organic frameworks (MOFs) necessitate buffers or basic amine moieties for high activity and turnover in the hydrolysis of organophosphates. While polymeric amine buffers can be integrated with a MOF, all solid-state formula-tions suffer from active site poisoning of the secondary building units (SBUs) within MOFs which inhibits further cat-alytic turnover. Herein, we developed a simple soaking procedure with a basic aqueous solution that reactivates the active sites of spent MOFs after organophosphate catalysis. Moreover, we develop amine functionalized MOF-808 de-rivatives through de novo synthesis with H3-BTC-NH2 (BTC = 1,3,5-benzenetricarboxylic acid) and post-synthetic mod-ification (PSM) that are highly active for organophosphate hydrolysis under non-buffered aqueous conditions.

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Experimental Procedures

All reagents were purchased from commercial sources and used without further purification. H₃-BTC-NH₂ was synthesized according to published procedures.¹

¹H-NMR spectra were recorded on Varian FT-NMR spectrometer (300 MHz) and data were analysed with Mestre Nova software. Samples (~10 mg) were digested using ~60 μ L of D₂SO₄ and 600 μ Lof DMSO-d₆.

³¹P-NMR spectra were recorded on Varian FT-NMR spectrometer (400 MHz) and data were analysed with Mestre Nova software.

Powder X-ray diffraction (PXRD) data were measured at room temperature on aPXRD data were also measured on a Rigaku Miniflex 600 diffractometer at 30kV, 15mA (CuK α 1 radiation, $\lambda = 1.54056$ Å) with a scan speed of 5°/min and a step size of 0.05 in 2 θ at room temperature.

Nitrogen isotherm measurements were carried out on a Micromeritics ASAP 2420 instrument at 77 K. Samples were activated at specified temperatures under vacuum on a Micromeritics Smart VacPrep instrument until an outgas rate below 0.05 mmHg/min was achieved.

Thermogravimetric analyses were performed under a stream of air using a TA Instrument Q600 SDT running from room temperature to 600 °C with a scan rate of 5 °C/min.

pH measurements were measured with EMD Millipore[™] MColorpHast[™] 1.09535.0001 Non-Bleeding pH-Indicator Strips.

MOF-808 synthesis, activation and SALI modification

Zr-MOF-808-PA: ZrCl₄ (117 mg, 0.5 mmol), BTC (35 mg, 0.168 mmol) and propanoic acid (3.7 mL, 49 mmol) were mixed in 10 mL of DMF in an 8-dram vial and ultrasonically dissolved. The clear solution was incubated in an oven at 120 °C for 24 h. After cooling down to room temperature, the white polycrystalline material was isolated by centrifuge (5 min, 7500 rpm) and solvent exchanged with fresh DMF three times (10 mL each) followed by methanol three times (10 mL). Zr-MOF-808-PA was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

The as synthesized MOF-808-PA was suspended in 12 mL DMF or acetone and 0.5 mL of 4 M aqueous HCl was added to an 8-dram vial and heated in an oven at 100°C or 50°C (DMF and acetone, respectively) for 18 h. After cooling to room temperature, the powder was isolated by centrifugation and washed with DMF three times (10 mL each) and acetone three times (10 mL each). MOF-808-act was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h, and then activated at 120 °C for 18 h using Micromeritics Smart VacPrep instrument as described above.

Hf-MOF-808-PA: HfCl₄ (160 mg, 0.5 mmol), BTC (35 mg, 0.168 mmol) and propanoic acid (3.7 mL, 49 mmol) were mixed in 10 mL of DMF in an 8-dram vial and ultrasonically dissolved. The clear solution was incubated in an oven at 120 °C for 24 h. After cooling down to room temperature, the white polycrystalline material was isolated by centrifuge (5 min, 7500 rpm) and solvent exchanged with fresh DMF three times (10 mL each) followed by methanol three times (10 mL). Hf-MOF-808-PA was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

The as synthesized Hf-MOF-808-PA was suspended in 12 mL DMF or acetone and 0.5 mL of 4 M aqueous HCl was added to an 8-dram vial and heated in an oven at 100°C or 50°C (DMF and acetone, respectively) for 18 h. After cooling to room temperature, the powder was isolated by centrifugation and washed with DMF and/or acetone three

times (10 mL each). Hf-MOF-808-HCl-act was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

Zr-MOF-808-AA: ZrCl₄ (117 mg, 0.5 mmol), BTC (35 mg, 0.168 mmol) and acetic acid (2.8 mL, 49 mmol) were mixed in 10 mL of DMF in an 8-dram vial and ultrasonically dissolved. The clear solution was incubated in an oven at 120 °C for 24 h. After cooling down to room temperature, the white polycrystalline material was isolated by centrifuge (5 min, 7500 rpm) and solvent exchanged with fresh DMF three times (10 mL each) followed by methanol three times (10 mL). Zr-MOF-808-AA was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

Zr-MOF-808-TFA: ZrCl₄ (117 mg, 0.5 mmol), BTC (35 mg, 0.168 mmol) and trifluoroacetic acid (3.8 mL, 49 mmol) were mixed in 10 mL of DMF in an 8-dram vial and ultrasonically dissolved. The clear solution was incubated in an oven at 120 °C for 24 h. After cooling down to room temperature, the white polycrystalline material was isolated by centrifuge (5 min, 7500 rpm) and solvent exchanged with fresh DMF three times (10 mL each) followed by methanol three times (10 mL). MOF-808-P was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

Zr-MOF-808-SALI-DMP: Zr-MOF-808-HCl-act (43 mg, 0.032 mmol) and dimethylphosphate (3.6 uL, 0.04 mmol) were mixed in 7 mL of water in an 8-dram vial. The clear solution was incubated at room temperature (~35 °C) for 24 h. The material was solvent exchanged with water three times (10 mL each) followed by methanol three times (10 mL). Zr-MOF-808-SALI-DMP was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

Hf-MOF-808-SALI-DMP: Hf-MOF-808-HCl-act (30 mg, 0.016 mmol) and dimethylphosphate (2 uL, 0.02 mmol) were mixed in 7 mL of water in a 8-dram vial. The clear solution was incubated at room temperature (~35 °C) for 24 h. The material was solvent exchanged with water three times (10 mL each) followed by methanol three times (10 mL). Hf-MOF-808-SALI-DMP was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

Synthesis of Zr-MOF-808-SALI-[BA-morph]: Zr-MOF-808-HCl-acetone-act (55 mg, 0.04 mmol) and 4-(morpholinomethyl)benzoic acid (171 mg, 0.78 mmol) were mixed in 8 mL of DMF in an 8-dram vial and ultrasonically dissolved. The vial was incubated in an oven at 80 °C for 18 h. After cooling down to room temperature, material was isolated by centrifuge (5 min, 7500 rpm) and solvent exchanged with fresh DMF three times (10 mL each) followed by methanol three times (10 mL). MOF-808-SALI-[BA-Morph]₂[PA]₁ was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

Synthesis of Hf-MOF-808-SALI-[BA-morph]₃: Hf-MOF-808-HCl-DMF-act (158 mg, 0.087 mmol) and 4- (morpholinomethyl)benzoic acid (155 mg, 0.7 mmol) were mixed in 8 mL of DMF in an 8-dram vial and ultrasonically dissolved. The clear solution was incubated in an oven at 70 °C for 18 h. After cooling down to room temperature, material was isolated by centrifuge (5 min, 7500 rpm) and solvent exchanged with fresh DMF three times (10 mL each) followed by methanol three times (10 mL). Hf-MOF-808-SALI-[BA-morph]₃[PA]₁ was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

Synthesis of Zr-MOF-808-SALI-[BA-CH₂NH₂]₃[BA-CH2-NH₃⁺]₁[PA]₁: Zr-MOF-808-HCl-DMF-act (50 mg, 0.04 mmol) and 4-(aminomethyl)benzoic acid (107 mg, 0.7 mmol) were mixed in 8 mL of H₂O in an 8-dram vial and ultrasonically dissolved. The clear solution was incubated in an oven at 80 °C for 48 h. After cooling down to room temperature, material was isolated by centrifuge (5 min, 7500 rpm) and solvent exchanged with fresh H₂O three times (10 mL each) followed by methanol three times (10 mL). MOF-808-SALI-[BA-CH₂NH₂]₃[BA-CH₂-NH₃⁺]₁[PA]₁ was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

Hf-MOF-808-SALI-[BA-CH₂NH₂]₂[BA-CH₂-NH₃⁺]₂: Hf-MOF-808-HCl-DMF-act (127 mg, 0.07 mmol) and 4-(aminomethyl)benzoic acid (210 mg, 1.4 mmol) were mixed in 16 mL of H₂O in an 8-dram vial and ultrasonically dissolved. The clear solution was incubated in an oven at 70 °C for 18 h. After cooling down to room temperature, material was isolated by centrifuge (5 min, 7500 rpm) and solvent exchanged with fresh H₂O three times (10 mL each) followed by methanol three times (10 mL). MOF-808-SALI-[BA-CH₂NH₂]₂[BA-CH₂-NH₃⁺]₂[PA]₁ was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

Zr-MOF-808-SALI-[BA-AO]₂: Zr-MOF-808-HCl-acetone-act (55 mg, 0.04 mmol) and 4-(N'-Hydroxycarbamimidoyl)benzoic acid (140 mg, 0.78 mmol) were mixed in 8 mL of DMF in an 8-dram vial and ultrasonically dissolved. The clear solution was incubated in an oven at 80 °C for 18 h. After cooling down to room temperature, material was isolated by centrifuge (5 min, 7500 rpm) and solvent exchanged with fresh DMF three times (10 mL each) followed by methanol three times (10 mL). Zr-MOF-808-SALI-[BA-AO]₂[PA]₁ was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

Hf-MOF-808-SALI-[BA-AO]₃: Hf-MOF-808-HCl-DMF-act (150 mg, 0.082 mmol) and 4-(N'-Hydroxycarbamimidoyl)benzoic acid (148 mg, 0.82 mmol) were mixed in 8 mL of DMF in an 8-dram vial and ultrasonically dissolved. The clear solution was incubated in an oven at 70 °C for 18 h. After cooling down to room temperature, material was isolated by centrifuge (5 min, 7500 rpm) and solvent exchanged with fresh DMF three times (10 mL each) followed by methanol three times (10 mL). Hf-MOF-808-SALI-[BA-AO]₃[PA]₁ was collected by centrifugation and dried in a vacuum oven at 80 °C for 1 h.

Zr-MOF-808-NH-TFA: $ZrCl_4$ (117 mg, 0.5 mmol), H₃-BTC-NH₂ (38 mg, 0.168 mmol) and trifluoroacetic acid (1.8 mL, 23.5 mmol) were mixed in 10 mL of DMF in an 8-dram vial and ultrasonically dissolved. The clear solution was incubated in an oven at 120 °C for five days. After cooling down to room temperature, the white polycrystalline material was exchanged with fresh DMF three times (10 mL each) followed by methanol three times (10 mL). Zr-MOF-808-NH-TFA was dried in a vacuum oven at 80 °C for 1 h.

Zr-MOF-808-NH-TFA-HCl-act: The as synthesized Zr-MOF-808-NH-TFA (70 mg) was suspended in 12 mL acetone and 1 mL of 4 M aqueous HCl was added to an 8-dram vial and heated in an oven at 50 °C for 18 h. After cooling to room temperature, the powder was solvent exchanged with water three times (10 mL each) and acetone three times (10 mL each). Zr-MOF-808-NH-TFA-HCl-act was dried in a vacuum oven at 80 °C for 1 h.

Zr-MOF-808-NH₂-K₂CO₃-act: The as synthesized Zr-MOF-808-NH-TFA (62 mg, 0.034 mmol) was suspended in an aqueous K_2CO_3 solution (9.2 mM, 10 mL) was added to an 8-dram vial and heated in an oven at 90°C for 18 h. After cooling to room temperature, the powder was solvent exchanged with water three times (10 mL each) and

methanol three times (10 mL each). Zr-MOF-808-NH₂ was dried in a vacuum oven at 80 °C for 1 h, and then activated at 100 °C for 18 h.

Hf-MOF-808-NH-TFA: HfCl₄ (160 mg, 0.5 mmol), H₃-BTC-NH₂ (38 mg, 0.168 mmol) and trifluoroacetic acid (1.8 mL, 23.5 mmol) were mixed in 10 mL of DMF in an 8-dram vial and ultrasonically dissolved. The clear solution was incubated in an oven at 120 °C for five days. After cooling down to room temperature, the white polycrystalline material was exchanged with fresh DMF three times (10 mL each) followed by methanol three times (10 mL each). Hf-MOF-808-NH-TFA was dried in a vacuum oven at 80 °C for 1 h.

Hf-MOF-808-NH₂-K₂CO₃-act: The as synthesized Hf-MOF-808-NH- TFA (240 mg, 0.094 mmol) was suspended in 10 mL water and an aqueous K_2CO_3 solution (9.2 mM, 10 mL) was added to an 8-dram vial and heated in an oven at 90°C for 18 h. After cooling to room temperature, the powder was solvent exchanged with water three times (10 mL each) and methanol three times (10 mL each). Hf-MOF-808-NH₂ was dried in a vacuum oven at 80 °C for 1 h, and then activated at 100 °C for 18 h.

Zr-MOF-808-NH-Morph: Zr-MOF-808-NH₂-K₂CO₃-act (90 mg, 0.068 mmol), 4-(2-chloroethyl)morpholine (14 mg, 0.1 mmol) and K₂CO₃ (13 mg, 0.092 mmol) were mixed in 10 mL of CH₃CN in an 8-dram vial. The solution was incubated in an oven at 75 °C for 18 h. After cooling down to room temperature, material was solvent exchanged with fresh H₂O three times (10 mL each) followed by methanol three times (10 mL). Zr-MOF-808-NH-Morph was dried in a vacuum oven at 80 °C for 1 h.



Figure S1. PXRD patterns of Zr-MOF-808-PA (black, bottom), after 24 h treatment with 0.0125 M piperidine aqueous solution (blue), 0.5 M piperidine aqueous solution (red), 0.1 M piperidine aqueous solution (yellow, top).



Figure S2. PXRD patterns of Hf-MOF-808-PA (black, bottom), after 24 h treatment with 0.0125 M piperidine aqueous solution (blue, middle), 0.1 M piperidine aqueous solution (red).



Figure S3. PXRD patterns of Zr-MOF-808-PA (bottom), Zr-MOF-808-DMP (middle) and Hf-MOF-808-DMP (top).



Figure S4. ¹H NMR spectrum of digested Zr-MOF-808-SALI-DMP in d_6 -DMSO.



Figure S5. ¹H NMR spectrum of digested Zr-MOF-808-SALI-DMP treated with piperidine in d_6 -DMSO.



Figure S6. ¹H NMR spectrum of digested Hf-MOF-808-SALI-DMP in d_6 -DMSO.



Figure S7. ¹H NMR spectrum of digested Hf-MOF-808-SALI-DMP treated with piperidine in d_6 -DMSO.



Figure S8. PXRD patterns of Zr-MOF-808-PA (bottom), and Hf-MOF-808-PA (top).



Figure S9. N_2 isotherms (left) and pore size distribution (right) of Hf-MOF-808-PA. Adsorption = filled, desorption = empty markers.



Figure S10. ¹H NMR spectrum of digested Hf-MOF-808-PA in *d*₆-DMSO.



Figure S11. ¹H NMR spectrum of digested Hf-MOF-808-PA after DMNP hydrolysis in *d*₆-DMSO.



Figure S12. N_2 isotherms (left) and pore size distribution (right) of Hf-MOF-808-HCl-DMF. Adsorption = filled, desorption = empty markers.



Figure S13. PXRD pattern of Hf-MOF-808-HCl-DMF.



Figure S14. ¹H NMR spectrum of digested Hf-MOF-808-HCl-DMF in *d*₆-DMSO.



Figure S15. N₂ isotherms (left) and pore size distribution (right) of Zr-MOF-808-AA. Adsorption = filled, desorption = empty markers.



Figure S16. TGA curve (black) and derivative weight in %/°C (red) of Zr-MOF-808-AA.



Figure S17. ¹H NMR spectrum of digested Zr-MOF-808-AA in *d*₆-DMSO.



Figure S18. ¹H NMR spectrum of digested Zr-MOF-808-AA after DMNP hydrolysis in *d*₆-DMSO.



Figure S19. N₂ isotherms (left) and pore size distribution (right) of Zr-MOF-808-TFA. Adsorption = filled, desorption = empty markers.



Figure S20. TGA curve (black) and derivative weight in %/°C (red) of Zr-MOF-808-TFA.



13.0 12.5 12.0 11.5 11.0 10.5 10.0 9.5 9.0 8.5 8.0 7.5 7.0 6.5 6.0 5.5 5.0 4.5 4.0 3.5 3.0 2.5 2.0 1.5 1.0 0.5 0.0 fl (ppm)

Figure S21. ¹H NMR spectrum of digested Zr-MOF-808-TFA in *d*₆-DMSO.



Figure S22. PXRD pattern of Zr-MOF-808-TFA.



Figure S23. N_2 isotherms (left) and pore size distribution (right) of Hf-MOF-808-HCl-acetone. Adsorption = filled, desorption = empty markers.



Figure S24. PXRD pattern of Hf-MOF-808-HCl-acetone.



Figure S25. ¹H NMR spectrum of digested Hf-MOF-808-HCl-acetone in *d*₆-DMSO.



Figure S26. N_2 isotherms (left) and pore size distribution (right) of Hf-MOF-808-SALI-BA-Morph. Adsorption = filled, desorption = empty markers.



Figure S27. PXRD pattern of Hf-MOF-808-SALI-BA-Morph.



Figure S28. ¹H NMR spectrum of digested Hf-MOF-808-SALI-BA-Morph in *d*₆-DMSO.



Figure S29. ¹H NMR spectrum of digested Hf-MOF-808-SALI-BA-Morph after DMNP hydrolysis in *d*₆-DMSO.



Figure S30. ¹H NMR spectrum of digested Hf-MOF-808-SALI-BA-Morph after 0.013 mM aqueous piperidine treatment in d_6 -DMSO.



Figure S31. N₂ isotherms (left) and pore size distribution (right) of Hf-MOF-808-SALI-BA-AO. Adsorption = filled, desorption = empty markers.



Figure S32. PXRD pattern of Hf-MOF-808-SALI-BA-AO.



Figure S33. ¹H NMR spectrum of digested Hf-MOF-808-SALI-BA-AO in d_6 -DMSO.





Figure S34. ¹H NMR spectrum of digested Hf-MOF-808-SALI-BA-AO after DMNP hydrolysis in *d*₆-DMSO.



Figure S35. N_2 isotherms (left) and pore size distribution (right) of Hf-MOF-808-SALI-BA-CH₂NH₂. Adsorption = filled, desorption = empty markers.



Figure S36. PXRD pattern of Hf-MOF-808-SALI-BA-CH₂NH₂.



Figure S37. ¹H NMR spectrum of digested Hf-MOF-808-SALI-BA-CH₂NH₂ in *d*₆-DMSO.



Figure S38. ¹H NMR spectrum of digested Hf-MOF-808-SALI-BA-CH₂NH₂ after DMNP hydrolysis in *d*₆-DMSO.



Figure S39. N₂ isotherms (left) and pore size distribution (right) of Zr-MOF-808-PA. Adsorption = filled, desorption = empty markers.



Figure S40. PXRD pattern of Zr-MOF-808-PA.



Figure S41. ¹H NMR spectrum of digested Zr-MOF-808-PA in *d*₆-DMSO.



Figure S42. ¹H NMR spectrum of digested Zr-MOF-808-PA after DMNP hydrolysis in *d*₆-DMSO.



Figure S43. N_2 isotherms (left) and pore size distribution (right) of Zr-MOF-808-HCl-DMF. Adsorption = filled, desorption = empty markers.



Figure S44. PXRD pattern of Zr-MOF-808-HCl-DMF.



Figure S45. ¹H NMR spectrum of digested Zr-MOF-808-HCl-DMF in *d*₆-DMSO.



Figure S46. N_2 isotherms (left) and pore size distribution (right) of Zr-MOF-808-HCl-acetone. Adsorption = filled, desorption = empty markers.



Figure S47. PXRD pattern of Zr-MOF-808-HCl-acetone.



Figure S48. ¹H NMR spectrum of digested Zr-MOF-808-HCl-acetone in d_6 -DMSO.



Figure S49. N_2 isotherms (left) and pore size distribution (right) of Zr-MOF-808-SALI-BA-Morph. Adsorption = filled, desorption = empty markers.



Figure S50. PXRD pattern of Zr-MOF-808-SALI-BA-Morph.



Figure S51. ¹H NMR spectrum of digested Zr-MOF-808-SALI-BA-Morph in *d*₆-DMSO.



Figure S52. ¹H NMR spectrum of digested Zr-MOF-808-SALI-BA-Morph after DMNP hydrolysis in *d*₆-DMSO.



Figure S53. N_2 isotherms (left) and pore size distribution (right) of Zr-MOF-808-SALI-BA-AO. Adsorption = filled, desorption = empty markers.



Figure S54. PXRD pattern of Zr-MOF-808-SALI-BA-AO.



Figure S55. ¹H NMR spectrum of digested Zr-MOF-808-SALI-BA-AO in *d*₆-DMSO.



Figure S56. ¹H NMR spectrum of digested Zr-MOF-808-SALI-BA-AO after DMNP hydrolysis in *d*₆-DMSO.



Figure S57. N_2 isotherms (left) and pore size distribution (right) of Zr-MOF-808-SALI-BA-CH₂NH₂. Adsorption = filled, desorption = empty markers.



Figure S58. PXRD pattern of Zr-MOF-808-SALI-BA-CH₂NH₂.



Figure S59. ¹H NMR spectrum of digested Zr-MOF-808-SALI-BA-CH₂NH₂ in *d*₆-DMSO.



Figure S60. ¹H NMR spectrum of digested Zr-MOF-808-SALI-BA-CH₂NH₂ after DMNP hydrolysis in *d*₆-DMSO.



Figure S61. N_2 isotherms (left) and pore size distribution (right) of Zr-MOF-808-NH-TFA. Adsorption = filled, desorption = empty markers.



Figure S62. TGA curve (black) and derivative weight in %/°C (red) of Zr-MOF-808-NH-TFA.



Figure S63. N₂ isotherms (left) and pore size distribution (right) of Zr-MOF-808-NH-TFA (black squares) activated with acetone-HCl (red circles), aqueous K_2CO_3 (blue triangles). Adsorption = filled, desorption = empty markers.



Figure S64. ¹⁹F NMR spectrum of digested as synthesized Zr-MOF-808-NH-TFA (1), after acetone-HCl activation (2), and after aqueous K_2CO_3 activation (3) in d_6 -DMSO. Inset graph is the zoom in region of the ¹⁹F NMR spectra.



Figure S65. ¹H NMR spectrum of digested Zr-MOF-808-NH-TFA in *d*₆-DMSO.



Figure S66. TGA curve (black) and derivative weight in %/°C (red) of Zr-MOF-808-NH-TFA after aqueous K₂CO₃ activation at 90 °C for 2 hours.



Figure S67. ¹H NMR spectrum of digested Zr-MOF-808-NH-TFA (bottom), and after K_2CO_3 activation (top) in d_6 -DMSO.



Figure S68. N₂ isotherms (left) and pore size distribution (right) of Zr-MOF-808-NH-Morph. Adsorption = filled, desorption = empty markers.



Figure S69. PXRD pattern of Zr-MOF-808-NH-Morph.



Figure S70. ¹H NMR spectrum of digested Zr-MOF-808-NH-Morph in d_6 -DMSO.



Figure S71. TGA curve (black) and derivative weight in %/°C (red) of Zr-MOF-808-NH-Morph.



Figure S72. Comparison of the TGA curves and derivatives weight in %/°C (dashed lines) of as synthesized Zr-MOF-808-NH-TFA (black), Zr-MOF-808-NH-TFA after K₂CO₃ activation (red), and Zr-MOF-808-NH-Morph (blue).



Figure S73. N_2 isotherms (left) and pore size distribution (right) of Hf-MOF-808-NH-TFA. Adsorption = filled, desorption = empty markers.



Figure S74. PXRD pattern of Hf-MOF-808-NH-TFA.



Figure S75. TGA curve (black) and derivative weight in %/°C (red) of Hf-MOF-808-NH-TFA.



Figure S76. N_2 isotherms (left) and pore size distribution (right) of Hf-MOF-808-NH-TFA after K_2CO_3 activation. Adsorption = filled, desorption = empty markers.



Figure S77. PXRD pattern of Hf-MOF-808-NH-TFA after K₂CO₃ activation.



Figure S78. TGA curve (black) and derivative weight in %/°C (red) of Hf-MOF-808-NH-TFA after K₂CO₃ activation.

DMNP hydrolysis with MOFs: MOF catalyst 12 mol% (3μ mol) was added to a 1 dram vial. 1 mL of a 10% v/v D₂O/H₂O (0.1 mL D₂O, 0.9 mL DI H₂O) solution was added to the MOF. The vial was capped and sonicated briefly (~1 min) to disperse the MOF. The mixture was then transferred to an NMR tube via pipette. DMNP (25

 μ mol, 4 μ L) was added to the inside of the tube via pipetter and capped. The NMR tube was quickly inverted thrice and placed into an NMR instrument. The hydrolysis was monitored by ³¹P NMR as described above.

Reactivation of spent MOFs and DMNP hydrolysis: The NMR tubes containing previously utilized MOF catalyst 12 mol% and 1 mL of a 10% v/v D₂O/H₂O (0.1 mL D₂O, 0.9 mL DI H₂O) solution were removed from the NMR instrument and MOF particles were allowed to settle on the bottom of the tube. The solution was carefully removed via pipette and 1 mL of 0.013M aqueous piperidine solution or an aqueous K₂CO₃ solution (9.2 mM, 10 mL) was added to the MOF within the tube. For K₂CO₃ activation, the NMR tube was capped and placed in a 90 °C oven for ~18 hours at room temperature. For piperidine activation, the NMR tube was capped and allowed to react with the MOF for ~18 hours. After removing the tube from the oven and allowing it to cool to room temperature, the solution was carefully removed via pipette and solvent exchanged with fresh H2O (3x 2mL). The reactivated MOF was then added1 mL of a 10% v/v D₂O/H₂O (0.1 mL D₂O, 0.9 mL DI H₂O) solution. DMNP (25 μ mol, 4 μ L) was added to the inside of the tube via pipetter and capped. The NMR tube was quickly inverted thrice and placed into an NMR instrument. The hydrolysis was monitored by ³¹P NMR as described above.



Figure S79. (Left) Plots of DMNP hydrolysis with Zr-MOF-808-PA at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Zr-MOF-808-PA at 12 mol% MOF loading.



Figure S80. (Left) Plots of DMNP hydrolysis with Hf-MOF-808-PA at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Hf-MOF-808-PA at 12 mol% MOF loading.



Figure S81. (Left) Plots of DMNP hydrolysis with Zr-MOF-808-AA at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Zr-MOF-808-AA at 12 mol% MOF loading.



Figure S82. (Left) Plots of DMNP hydrolysis with Zr-MOF-808-TFA at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Zr-MOF-808-TFA at 12 mol% MOF loading.



Figure S83. (Left) Plots of DMNP hydrolysis with HCl-DMF activated Zr-MOF-808 at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with HCl-DMF activated Zr-MOF-808 at 12 mol% MOF loading.



Figure S84. (Left) Plots of DMNP hydrolysis with HCl-acetone activated Zr-MOF-808 at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with HCl-acetone activated Zr-MOF-808 at 12 mol% MOF loading.



Figure S85. (Left) Plots of DMNP hydrolysis with HCl-DMF activated Hf-MOF-808 at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with HCl-DMF activated Hf-MOF-808 at 12 mol% MOF loading.



Figure S86. (Left) Plots of DMNP hydrolysis with HCl-acetone activated Hf-MOF-808 at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with HCl-acetone activated Hf-MOF-808 at 12 mol% MOF loading.



Figure S87. (Left) Reaction profile of DMNP hydrolysis with Hf-MOF-808-SALI-[BA-Morph]₂ at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Hf-MOF-808-SALI-[BA-Morph]₂.



Figure S88. (Left) Reaction profile of DMNP hydrolysis with spent Hf-MOF-808-SALI-[BA-Morph]₂ at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Hf-MOF-808-SALI-[BA-Morph]₂.



Figure S89. (Left) Reaction profile of DMNP hydrolysis with reactivated Hf-MOF-808-SALI-[BA-Morph]₂ at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with piperidine treated Hf-MOF-808-SALI-[BA-Morph]₂.



Figure S90. (Left) Reaction profile of DMNP hydrolysis with Hf-MOF-808-SALI-[BA-CH₂NH₂]₂ at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Hf-MOF-808-SALI-[BA-CH₂NH₂]₂.



Figure S91. (Left) Reaction profile of DMNP hydrolysis with Hf-MOF-808-SALI-[BA-AO]₂ at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Hf-MOF-808-SALI-[BA-AO]₂.



Figure S92. (Left) Reaction profile of DMNP hydrolysis with Zr-MOF-808-SALI-[BA-Morph]₂ at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Zr-MOF-808-SALI-[BA-Morph]₂.



Figure S93. (Left) Reaction profile of DMNP hydrolysis with spent Zr-MOF-808-SALI-[BA-Morph]₂ at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Zr-MOF-808-SALI-[BA-Morph]₂.



Figure S94. (Left) Reaction profile of DMNP hydrolysis with Zr-MOF-808-SALI-[BA-CH₂NH₂]₂ at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Zr-MOF-808-SALI-[BA-CH₂NH₂]₂.



Figure S95. (Left) Reaction profile of DMNP hydrolysis with Zr-MOF-808-SALI-[BA-AO]₂ at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Zr-MOF-808-SALI-[BA-AO]₂.



Figure S96. (Left) Reaction profile of DMNP hydrolysis with Zr-MOF-808-NH-TFA at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Zr-MOF-808-NH-TFA.



Figure S97. (Left) Reaction profile of DMNP hydrolysis with Zr-MOF-808-NH-TFA after HCl activation at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Zr-MOF-808-NH-TFA after HCl activation.



Figure S98. (Left) Reaction profile of DMNP hydrolysis with Zr-MOF-808-NH-TFA after K_2CO_3 activation at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Zr-MOF-808-NH-TFA after K_2CO_3 activation.



Figure S99. (Left) Reaction profile of DMNP hydrolysis with Zr-MOF-808-NH-Morph after at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Zr-MOF-808-NH-Morph.



Figure S100. (Left) Reaction profile of DMNP hydrolysis with Hf-MOF-808-NH-TFA at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Hf-MOF-808-NH-TFA.



Figure S101. (Left) Reaction profile of DMNP hydrolysis with Hf-MOF-808-NH-TFA after K_2CO_3 activation at 12 mol% MOF loading. (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Hf-MOF-808-NH-TFA after K_2CO_3 activation.



Figure S102. (Top Left) SEM image of as synthesized amine functionalized Zr-MOF-808 at 10 μ m magnification. (Top Right) SEM image of as synthesized amine functionalized Zr-MOF-808 at 3 μ m magnification. (Bottom Left) SEM image of as synthesized amine functionalized Zr-MOF-808 after K₂CO₃ activation at 10 μ m magnification. (BottomRight) SEM image of as synthesized amine functionalized Zr-MOF-808 after K₂CO₃ activation at 10 μ m control at 3 μ m magnification.





Figure S103. (Top Left) SEM image of Zr-MOF-808-PA at 10 µm magnification. (Top Right) SEM image of Zr-MOF-808 at 3 µm magnification. (Bottom Left) SEM image of Zr-MOF-808-5X at 5 µm magnification. (Bottom Right) SEM image of Zr-MOF-808-5x at 3 µm magnification

Schiff-base test: The MOFs (50 mg) were added to a 8 dram vial and 10 mL of acetone was added to the MOF. The vial was capped and allowed to stand at room temperature for ~18 hours. The acetone was carefully removed from the vial via pipette and the uncapped vial was heated briefly ~80 C.

K₂CO₃ activated amine functionalized Zr-MOF-808 before acetone exposure (Zr-MOF-808-NH2) As synthesized amine functionalized Zr-MOF-808 (Zr-MOF-808-NH-TFA)



K₂CO₃ activated amine functionalized Zr-MOF-808 after acetone exposure (Zr-MOF-808-NH2) As synthesized amine functionalized Zr-MOF-808 (Zr-MOF-808-NH-TFA)



Figure S104. (Left) K_2CO_3 activated Zr-MOF-808-NH-TFA (055-1) and as synthesized Zr-MOF-808-NH-TFA (028). (Right) K_2CO_3 activated Zr-MOF-808-NH-TFA and as synthesized Zr-MOF-808-NH-TFA after acetone treatment.



(1) SEM-EDX combined map

FOV: 5.16 μm, Mode: 15kV - Map, Detector: BSD Full, Time: FEB 27 2020 06:19 (2)

Element	Element	Element	Atomic	Weight
Number	Symbol	Name	Conc.	Conc.
40	Zr	Zirconium	39.81	80.72

7	N	Nitrogen	59.21	18.43
19	К	Potassium	0.98	0.85



Disabled elements: C, O

Figure S105. (1) Combined SEM-EDX mapping of K_2CO_3 activated amine functionalized Zr-MOF-808. (2) Chemical composition analysis of K_2CO_3 activated amine functionalized Zr-MOF-808. (3) EDX spectrum of K_2CO_3 activated amine functionalized Zr-MOF-808.



Figure S106. (Left) Comparison of the reaction profiles of DMNP hydrolysis with Zr-MOF-808-NH-TFA (red) and Hf-MOF-808-NH-TFA (black) after K_2CO_3 activation at 12 mol% MOF loading. (Right) Zoom in on the initial reaction profile of DMNP hydrolysis with Zr-MOF-808-NH-TFA (red) and Hf-MOF-808-NH-TFA (black) after K_2CO_3 activation at 12 mol% MOF loading.



Figure S107. ¹H NMR spectrum of digested K₂CO₃ activated Zr-MOF-808-PA(5X) in d₆-DMSO.



Figure S108. (Left) Reaction profile of DMNP hydrolysis with Zr-MOF-808-PA(5X) after K₂CO₃ activation (red). (Right) Corresponding ³¹P NMR spectra of DMNP hydrolysis with Zr-MOF-808-PA(5X) after K₂CO₃ activation.



Figure S109. (Left) Transmittance ATF-IR spectra of Zr-MOF-808-NH-TFA (black), Zr-MOF-808-NH-TFA K2CO3-activated (red), and Zr-MOF-808-NH₂ after acetone treatment (blue). (Right) ¹H NMR spectrum of digested Zr-MOF-808-NH-TFA (black), Zr-MOF-808-NH-TFA K2CO3-activated (red), and Zr-MOF-808-NH₂ after acetone treatment (blue).



Figure S110. (Left) Initial DMNP hydrolysis conversion with Zr-MOF-808-NH-Morph (red solid squares), spent Zr-MOF-808-NH-Morph (red unfilled squares), Zr-MOF-808-SALI-BA-Morph (black solid squares), and spent Zr-MOF-808-SALI-BA-Morph (black unfilled squares) at 12 mol% MOF loading.



Figure S111. % Transmittance ATF-IR spectra of Zr-MOF-808-NH-TFA (black), Zr-MOF-808-NH-TFA HClactivated (blue), Zr-MOF-808-NH-TFA K₂CO₃-activated (red), and Zr-MOF-808-NH-Morph (green).

Computational Modeling

Given that there is no reported synthesis of Zr-MOF-808-NH₂, computational modeling was utlized to construct a simulated single X-ray structure. The structure was then utilized to generate a simulated PXRD pattern as seen in Figure 2. Ab-initio DFT calculations were performed using the PBE functional and plane-wave basis sets as implemented in the Quantum Espresso 6.4 software package (P. Giannozzi et. al., J.Phys.: Condens.Matter 29, 465901 (2017)). Energy cutoffs were set at 50 Ry for the wavefunction and 400 Ry for density. The calculations used the primitive cell of MOF-808, with hydroxyl termination of the 6 open sites at each SBU. Following is the .cif file of the geometry optimised structure (Total energy is -8308.3829395732 Ry):

```
data
loop
symmetry_equiv_pos_as_xyz
х,у, z
cell length a
                24.732090
cell length b
                24.732082
cell length c
                24.732098
_cell_angle_alpha
                      60.000035
cell angle beta
                      60.000031
 cell angle gamma
                      60.000010
loop
atom site label
_atom_site_fract x
atom site fract y
atom site fract z
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0	0.190109	0.479561	0.055222
0	0.279118	0.059391	0.472930
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0	0.273449	0.477389	0.186361
0	0 067020	0 182615	0 470159
0	0 466804	0 276243	0 076588
0	0.100712	0.270243	0.070500
0	0.102/13	0.000141	0.201374
0	0.184645	0.2/456/	0.469494
0	0.4//499	0.061663	0.188491
0	0.059899	0.479299	0.271066
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0	0.066394	0.779245	0.967189
0	0.971719	0.183977	0.065394
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0	0.971510	0.782155	0.176446
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0	0.774801	0.978954	0.063102
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Zr	0.242135	0.101666	0.547910
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Zr	0.239387	0.553251	0.103383
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Zr	0.106453	0.101148	0.550091
0	0.062767	0.056779	0.658270

Zr	0.550317	0.239253	0.110187
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Zr	0.100762	0.105715	0.243952
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ZT O	0.103241	0.553188	0.235232
0	0.056903	0.659820	0.21/10/
Zr	0.236077	0.101869	0.109542
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Zr	0.700518	0.145664	0.148608
0	0.593433	0.186208	0.195142
Zr	0.147652	0.701435	0.005471
0	0.193230	0.593920	0.026083
Zr	0.006688	0.147027	0.146919
0	0.027287	0.191047	0.190599
7r	0 147297	0.006324	0.698586
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U V	0.029740	0.142204	0.102130
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Zr	0.699268	0.013279	0.145/41
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Zr	0.148969	0.141532	0.695652
0	0.193632	0.185822	0.588357
Zr	0.140148	0.010979	0.150130
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Zr	0.696392	0.148940	0.014291
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Zr	0.143467	0.700009	0.142658
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Zr	0.014159	0.141869	0.697819
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0	0.915333	0.197100	0.192881
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H	0.195826	0.730481	0.871382
~ *			

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Н	0.212208	0.729605	0.177098
0	0 921545	0 189457	0 694411
ч	0.921010	0 185788	0 727457
\cap	0 919291	0.105700	0 194238
U U	0.919291	0.000000	0.211503
0	0.106122	0.125575	0.211303
U	0.100132	0.190700	0.922760
П	0.100251	0.212909	0.002002
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Н	0.181534	0.211400	0.725640
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0	0.055260	0.330132	0.060323
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Н	0.375559	0.061497	0.517401
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