

Supplementary Material

Enabling high loading of well-dispersed Ni₂CoP₂ catalysts on 3D-printed electrode for efficient electrocatalysis

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Supplementary Figures

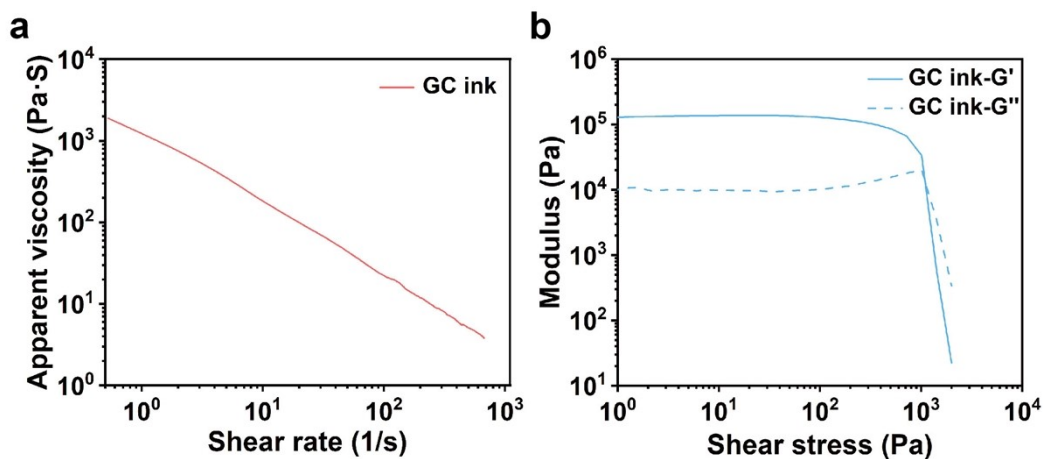


Fig. S1. (a) Apparent viscosity as a function of shear rate for GC inks. (b) Storage modulus (G') and loss modulus (G'') as a function of shear stress for GC ink.

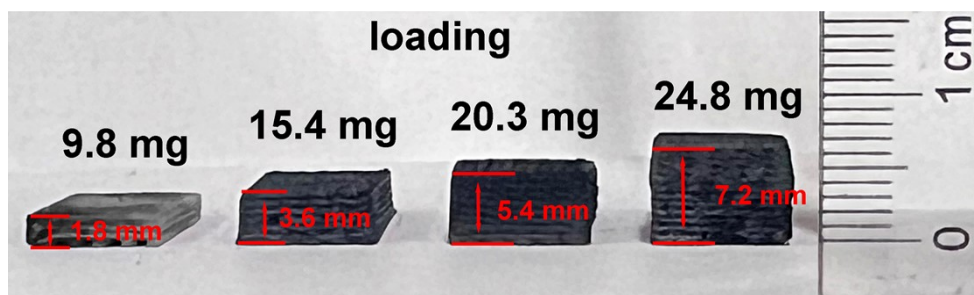


Fig. S2. The optical photo of different 3DP GC electrodes with 6, 12, 18, and 24 layers.

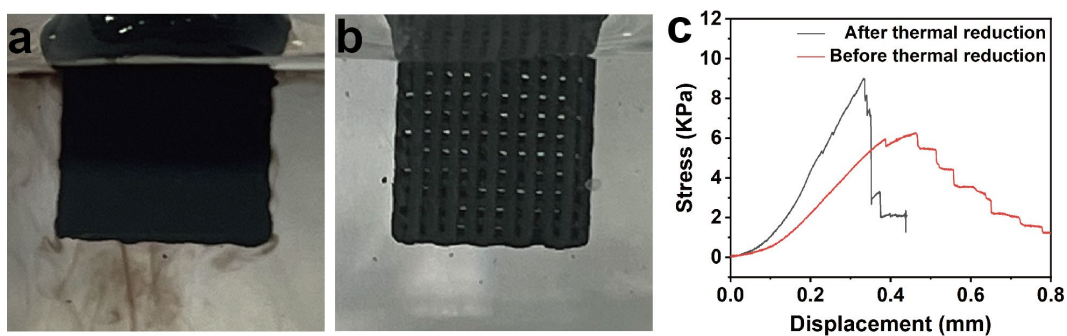


Fig. S3. (a) The 3DP GC electrode before thermal reduction, (b) The 3DP GC electrode after thermal reduction, (c) Flexural stress.

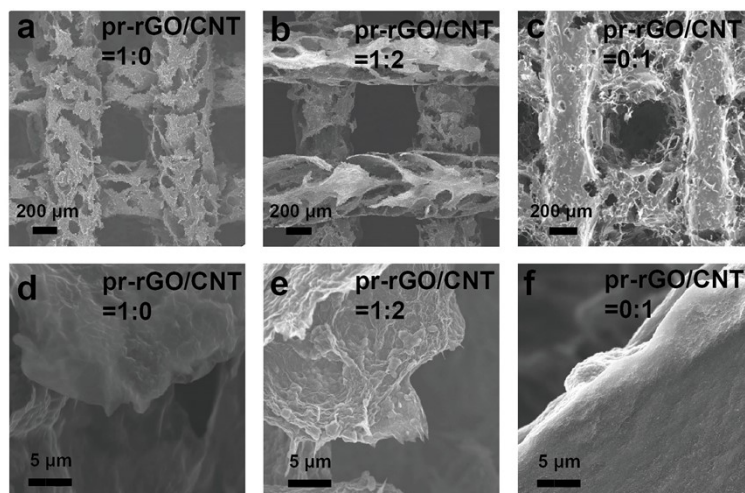


Fig. S4. SEM characterization of three electrodes with different GC proportions (a, d) pr-rGO/CNT=1:0, (b, e) pr-rGO/CNT=1:2, (c, f) pr-rGO/CNT=0:1.

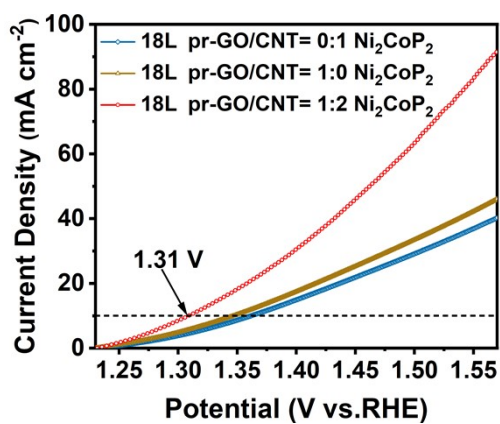


Fig. S5. The polarization curves of 18L-3DP electrodes with different ratios of CNT and pr-GO.

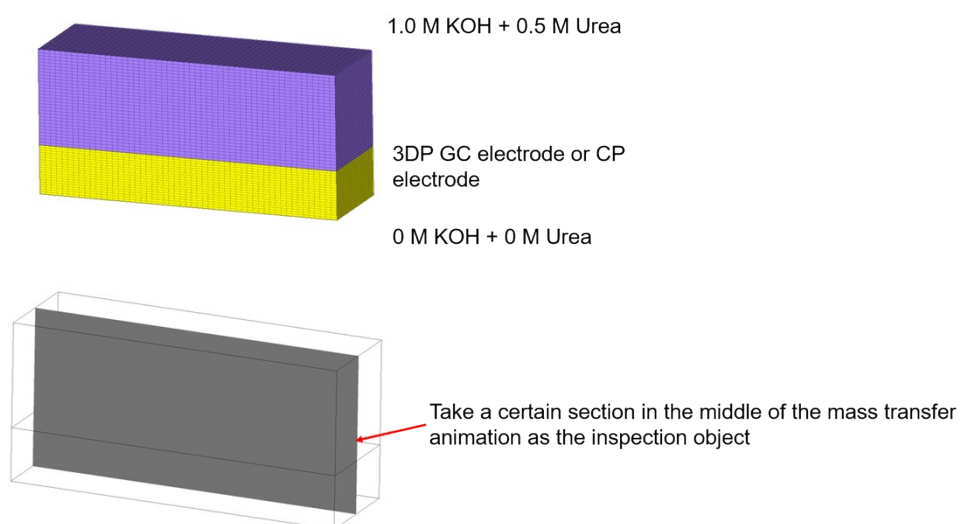


Fig. S6. The mass transport model.

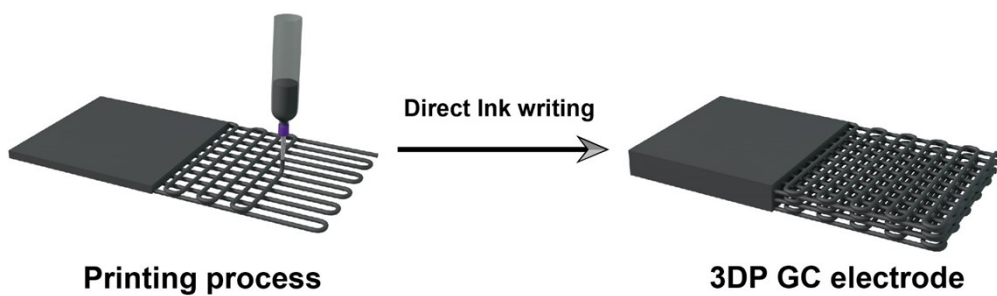


Fig. S7. The printing process of 3DP GC

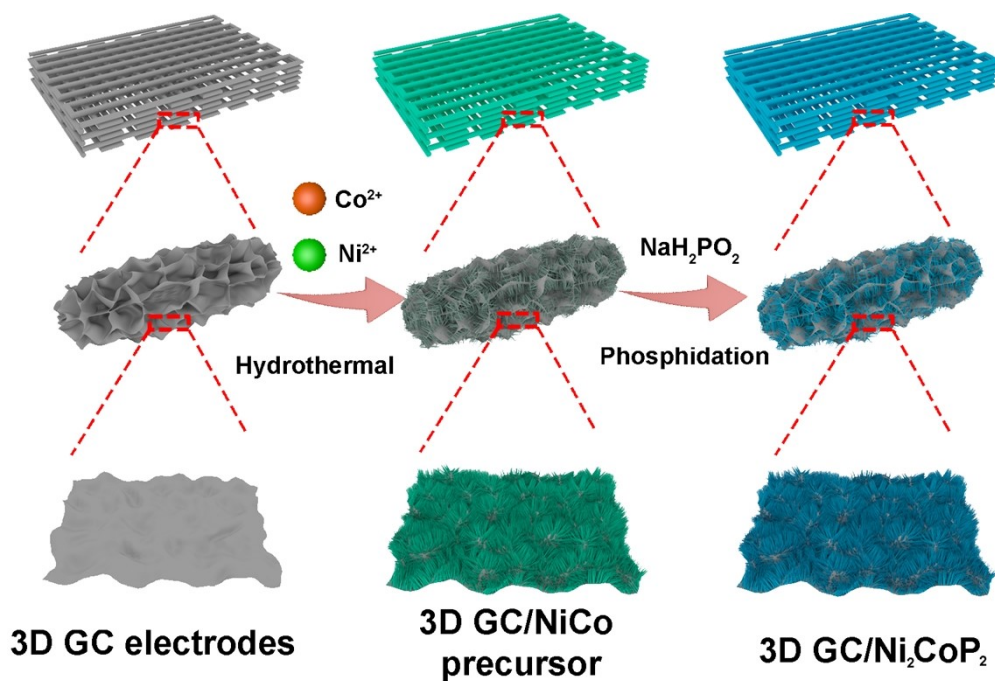


Fig. S8. The preparation of 3DP GC/Ni₂CoP₂.

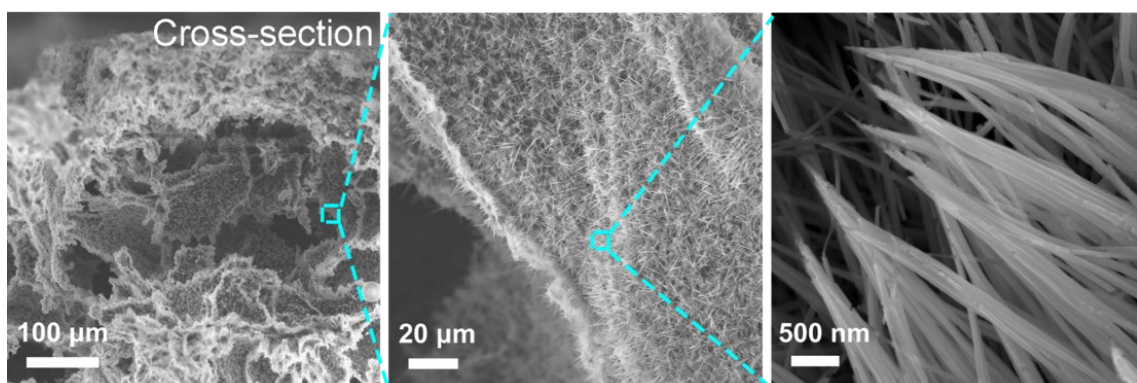


Fig. S9. The SEM images of 3DP GC/Ni₂CoP₂ (cross-section).

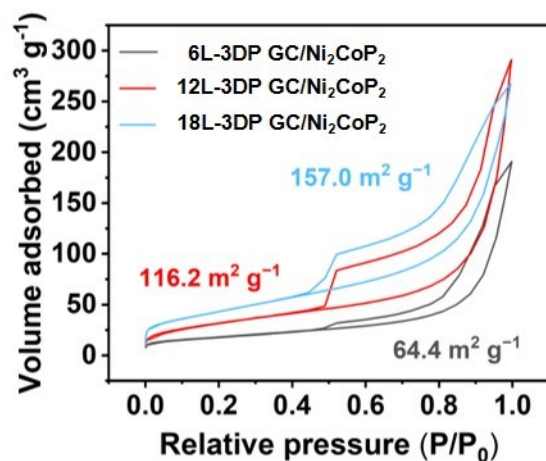


Fig. S10. N₂ sorption isotherm of 6L, 12L, 18L-3DP GC/Ni₂CoP₂.

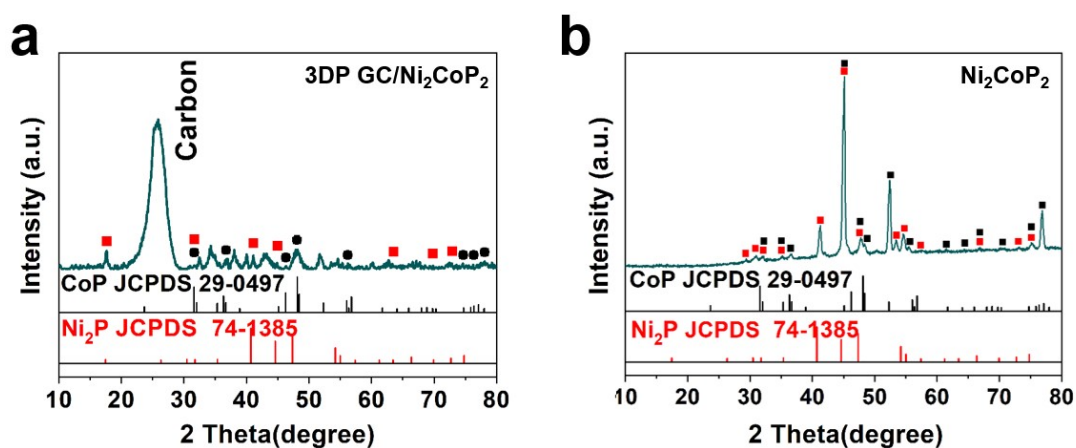


Fig. S11. XRD spectrum of a) 3DP GC/Ni₂CoP₂, b) Ni₂CoP₂. (red square for Ni₂P, black square for CoP)

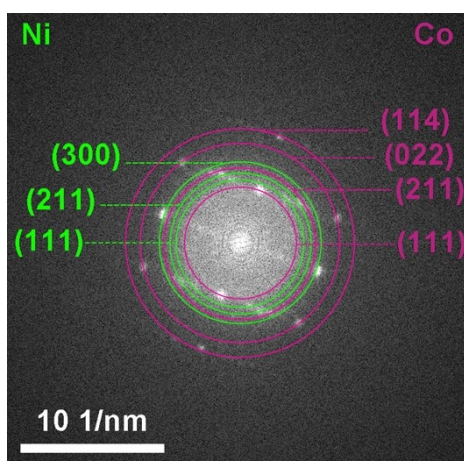


Fig. S12. The SAED pattern of Ni₂CoP₂.

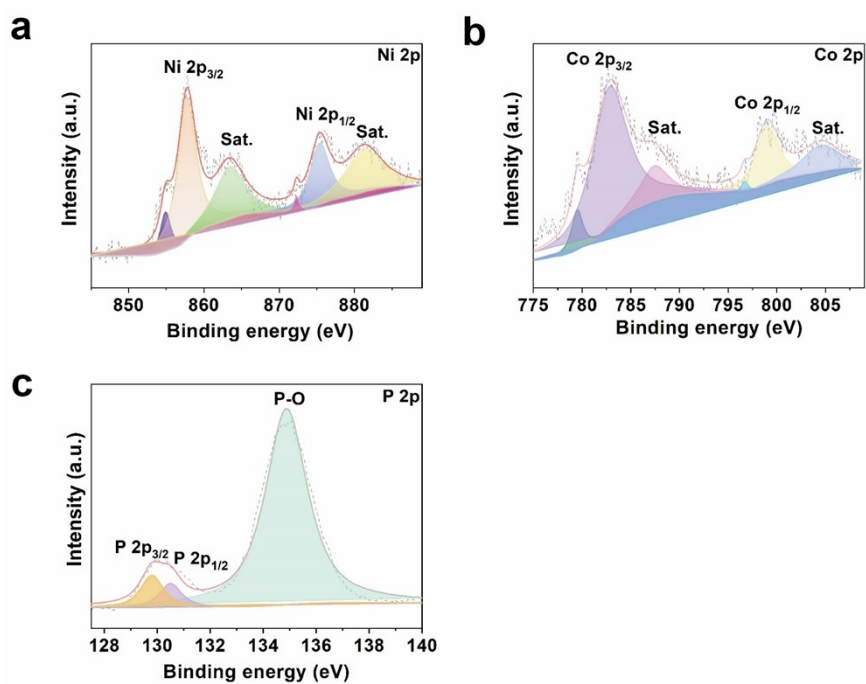


Fig. S13. High-resolution XPS spectra of 3DP GC/Ni₂CoP₂, a) Ni 2p, b) Co 2p, c) P 2p.

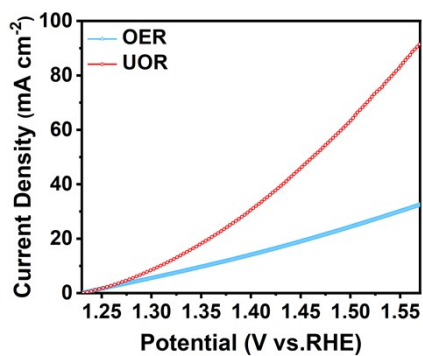


Fig. S14. UOR and OER LSV curves of 18L-3DP GC/Ni₂CoP₂.

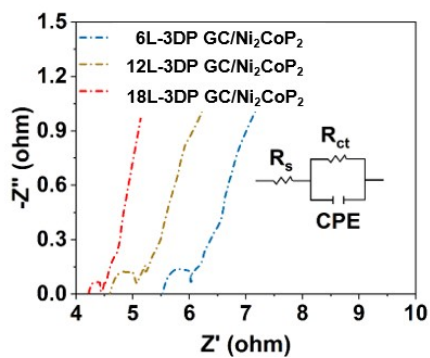


Fig. S15. Nyquist plots of different electrocatalysts for UOR.

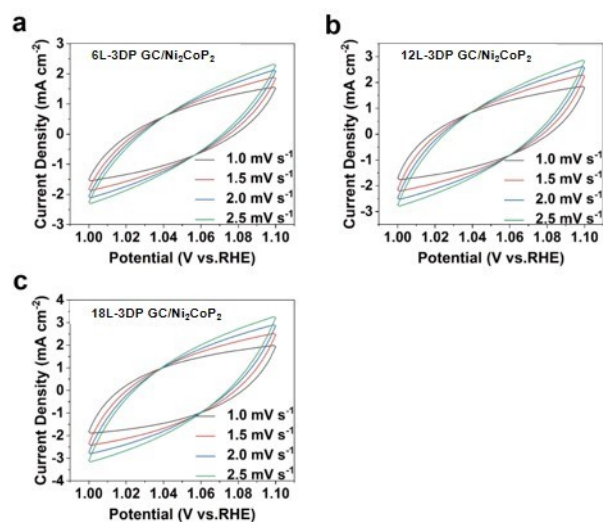


Fig. S16. CV curves of 3DP GC/Ni₂CoP₂ with different layers with different rate from 1 to 2.5 mV s⁻¹ for UOR.

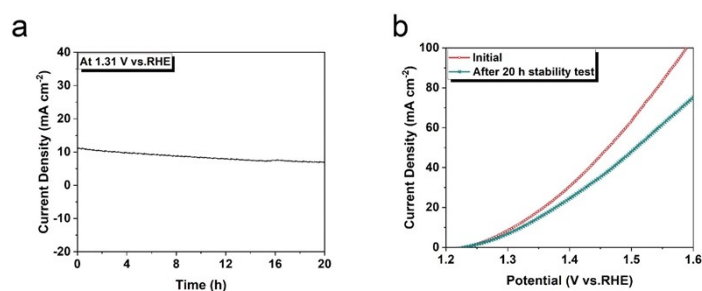


Fig. S17. a) Chronopotentiometry plots of 18L-3DP GC/Ni₂CoP₂ obtained at a constant potential of 1.31 V for UOR. b) LSV curves of 18L-3DP GC/Ni₂CoP₂ before and after 20 h stability for UOR.

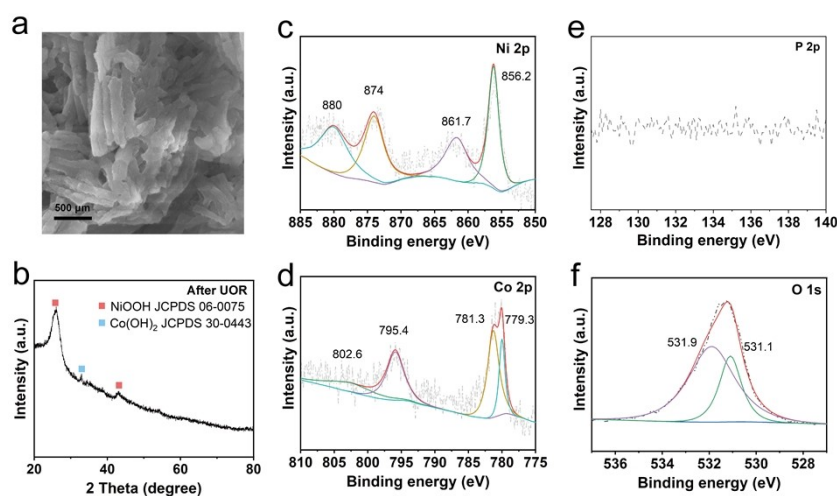


Fig. S18. a) The SEM of Ni₂CoP₂ after UOR, b) The XRD of Ni₂CoP₂ after UOR, c-f) The XPS of Ni₂CoP₂ after UOR.

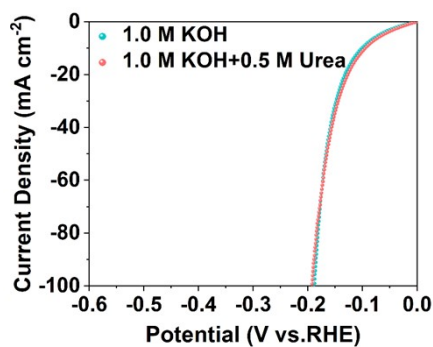


Fig. S19. HER LSV curves of 18L-3DP GC/Ni₂CoP₂ in 1.0 M KOH and 1.0 M KOH with 0.5 M urea.

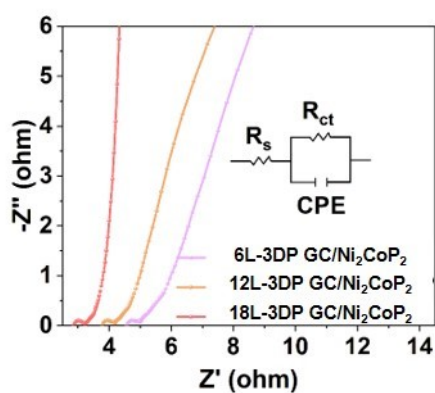


Fig. S20. Nyquist plots of different electrocatalysts for the HER.

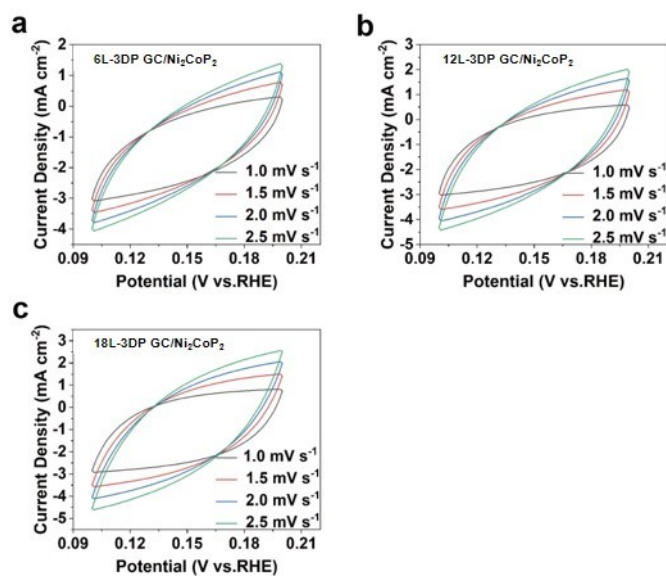


Fig. S21. CV curves of 3DP GC/Ni₂CoP₂ with different layers with different rate from 1 to 2.5 mV s⁻¹ for HER.

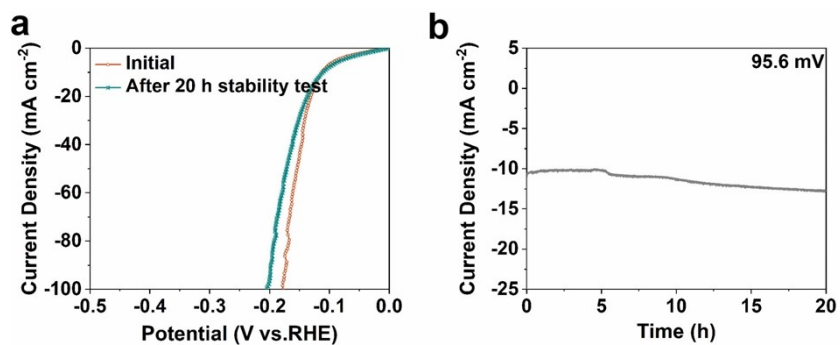


Fig. S22. a) LSV curves of 18L-3DP GC/Ni₂CoP₂ before and after 20 h stability test for the HER. b) Chronopotentiometry plots of 18L-3DP GC/Ni₂CoP₂ obtained at a constant potential of 0.0956 V for HER.

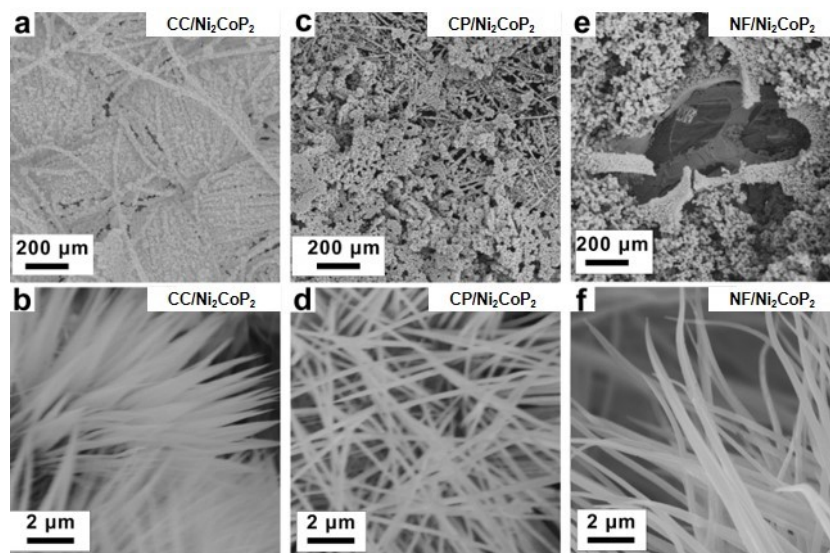


Fig. S23. The SEM images of Ni₂CoP₂ on (a,b) CC, (c,d) CP, (e,f) NF.

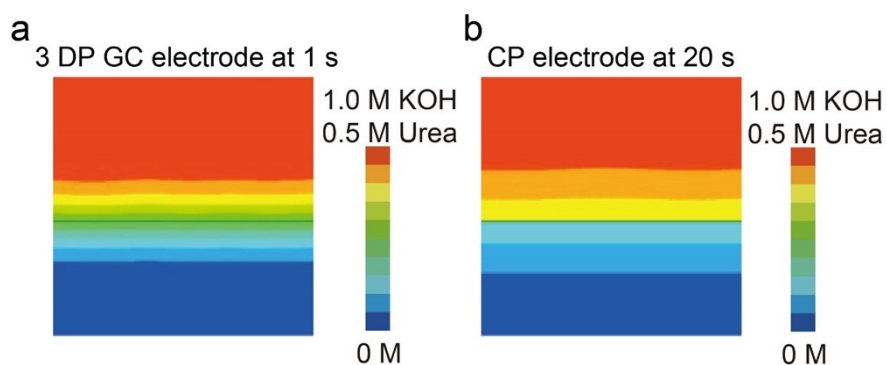


Fig. S24. Simulation of 1.0 M KOH + 0.5 M urea mass transfer under different 3D structured electrodes. a) 3DP GC electrode. b) CP electrode.

Table S1. The corresponding EIS parameters of the 6L-3DP GC/Ni₂CoP₂, 12L-3DP GC/Ni₂CoP₂ and 18L-3DP GC/Ni₂CoP₂ for UOR.

	18L-3DP GC/Ni ₂ CoP ₂	12L-3DP GC/Ni ₂ CoP ₂	6L-3DP GC/Ni ₂ CoP ₂
R _s (Ω cm ²)	4.22	4.60	5.52
R _{ct} (Ω cm ²)	0.24	0.52	0.59

Table S2. The corresponding EIS parameters of the 6L-3DP GC/Ni₂CoP₂, 12L-3DP GC/Ni₂CoP₂ and 18L-3DP GC/Ni₂CoP₂ for HER.

	18L-3DP GC/Ni ₂ CoP ₂	12L-3DP GC/Ni ₂ CoP ₂	6L-3DP GC/Ni ₂ CoP ₂
R _s (Ω cm ²)	2.82	3.68	4.65
R _{ct} (Ω cm ²)	0.42	0.57	0.69

Table S3. Comparison of UOR performance of 3DP GC/Ni₂CoP₂ with Ni, Co-based electrocatalysts. (*j*: current density; E: potential)

Catalysts	<i>j</i> (mA cm ⁻²)	E (V)	Reference
a-MoS ₂ /CoS/Co _{0.85} Se HNTs	50	1.38	<i>Nanoscale</i> , 2020, 12 , 991
Ni ₂ P/Fe ₂ P/NF	10	1.36	<i>J. Colloid Interf. Sci.</i> , 2019, 541 , 279-286
Co-Ni-S@NF	10	1.31	<i>J. Mater. Chem. A</i> , 2022, 10 , 24137
Co-V-NiS ₂	10	1.35	<i>ACS Catal.</i> , 2022, 12 , 569-579
Co-NiMoO ₄ -Ar	10	1.324	<i>J. Mater. Chem. A</i> , 2022, 10 , 16825
Ni(OH)S/NF	10	1.34	<i>Appl. Catal. B Environ.</i> , 2022, 312 , 121389
NC-FNCP	10	1.37	<i>Nano Res.</i> , 2022, 15 , 1916-1925
CoNi LDH	10	1.32	<i>ACS Appl. Mater. Interfaces</i> , 2022, 14 , 16222-16232
Ni ₃ S ₂ /NF	10	1.34	<i>ACS Appl. Mater. Interfaces</i> , 2021, 13 , 35709-35718
3DP GC/Ni₂CoP₂	10	1.31	This work

Table S4 Comparison of HER performance of 3DP GC/Ni₂CoP₂ with Ni, Co-based electrocatalysts. (*j*: current density; η : overpotential)

Catalysts	<i>j</i> (mA cm ⁻²)	η (mV)	Reference
a-MoS ₂ /CoS/Co _{0.85} Se HNTs	10	127	<i>Nanoscale</i> , 2020, 12 , 991
Ni ₂ P/Fe ₂ P/NF	10	115	<i>J. Colloid Interf. Sci.</i> , 2019, 541 , 279-286
Ni-Co-P HNBS	10	107	<i>Energy Environ. Sci.</i> , 2018, 11 , 872
Fe-CoP HTPAs	10	98	<i>Small</i> , 2018, 14 , 1704233
NiCoP/NF	10	95	<i>J Mater. Chem. A</i> , 2023, 11 , 1256
NiCoP-90	10	93	<i>Adv. Energy Mater.</i> , 2023, 12 , 2300499
Ni _{0.2} S/Cu ₅ FeS ₄	10	140	<i>J. Colloid Interf. Sci.</i> , 2020, 578 , 668-676
r-NiS/h-NiS	10	101	<i>J. Colloid Interf. Sci.</i> , 2023, 633 , 640-648
Ni ND _{s45}	10	204	<i>Mol. Catal.</i> , 2021, 516 , 112006
Co _{0.9} Ni _{0.1} Se with vacancies	10	185.7	<i>Mol. Catal.</i> , 2022, 525 , 112339
Ni-SA/NC	10	102	<i>Angew. Chem. Int. Ed.</i> , 2020, 59 , 22743-22748
NiCo ₂ -P	10	183	<i>Adv. Mater.</i> , 2020, 33 , 2003846
HC NiCo/C	10	99	<i>ACS Appl. Mater. Interfaces</i> , 2021, 13 , 9932-9941
NiCo DASs/N-C	10	189	<i>Adv. Funct. Mater.</i> , 2023, 33 , 2210867
NiCoO-NiCo/C	10	123	<i>Appl. Catal. B Environ.</i> , 2021, 292 , 120170
3DP GC/Ni ₂ CoP ₂	10	95.6	This work

Video S1. Electrolyte transmission behaviour of 3DP GC electrode.

Video S2. Electrolyte transmission behaviour of CP electrode.