

# **Investigating the role of interstitial water molecules in copper hexacyanoferrates for sodium-ion battery cathodes**

*Donghyeon Kim<sup>†</sup>, Ahreum Choi<sup>†,\*</sup>, Changhyun Park, Min-Ho Kim, Hyun-Wook Lee<sup>\*</sup>*

School of Energy and Chemical Engineering, Ulsan National Institute of Science and  
Technology (UNIST), Ulsan 44919, Republic of Korea

<sup>†</sup>These authors contributed equally to this work.

\*Corresponding author: [archoi0210@unist.ac.kr](mailto:archoi0210@unist.ac.kr) (A.C.), [hyunwooklee@unist.ac.kr](mailto:hyunwooklee@unist.ac.kr) (H.-W.L.)

**Table S1 | The Results of ICP Analysis of CuHCF<sub>2</sub>-1.4H<sub>2</sub>O, CuHCF<sub>2</sub>-1.6H<sub>2</sub>O, and CuHCF<sub>2</sub>-1.8H<sub>2</sub>O.**

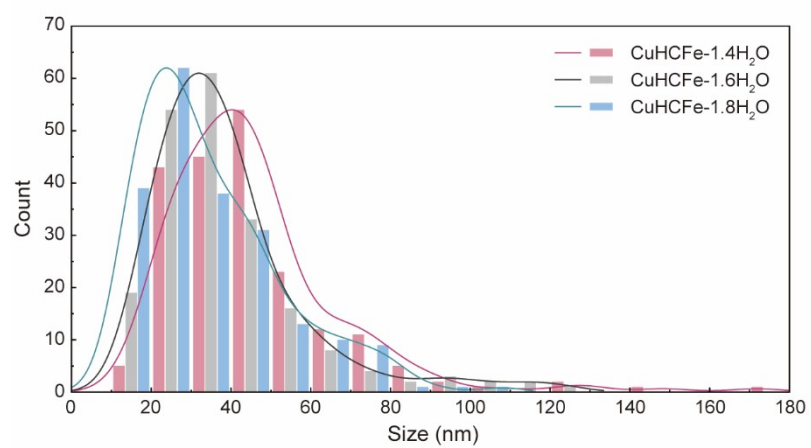
| <b>mg/kg</b>                            | <b>Cu</b> | <b>Fe</b> |
|-----------------------------------------|-----------|-----------|
| CuHCF <sub>2</sub> -1.4H <sub>2</sub> O | 245660    | 141986    |
| CuHCF <sub>2</sub> -1.6H <sub>2</sub> O | 231223    | 134597    |
| CuHCF <sub>2</sub> -1.8H <sub>2</sub> O | 232902    | 137121    |

**Table S2 | The Results of EA of CuHCF<sub>e</sub>-1.4H<sub>2</sub>O, CuHCF<sub>e</sub>-1.6H<sub>2</sub>O, and CuHCF<sub>e</sub>-1.8H<sub>2</sub>O for C and N.**

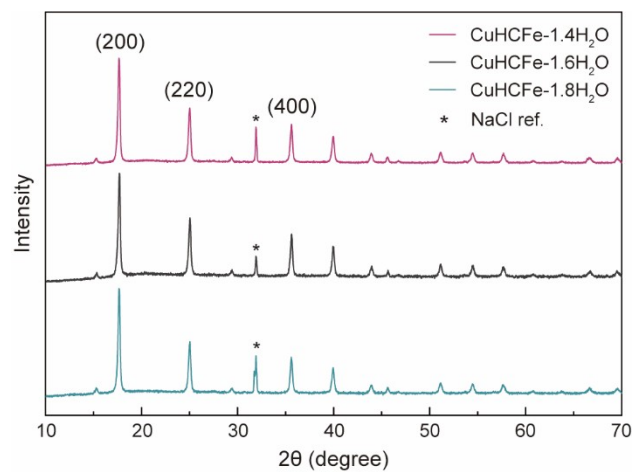
| wt%                                     | C     | N     |
|-----------------------------------------|-------|-------|
| CuHCF <sub>e</sub> -1.4H <sub>2</sub> O | 22.68 | 23.42 |
| CuHCF <sub>e</sub> -1.6H <sub>2</sub> O | 23.33 | 24.73 |
| CuHCF <sub>e</sub> -1.8H <sub>2</sub> O | 23.55 | 24.71 |

**Table S3 | The Chemical Formula of CuHCFE-1.8H<sub>2</sub>O and CuHCFE-1.4H<sub>2</sub>O at Pristine State, SOC 0 and SOC 100.**

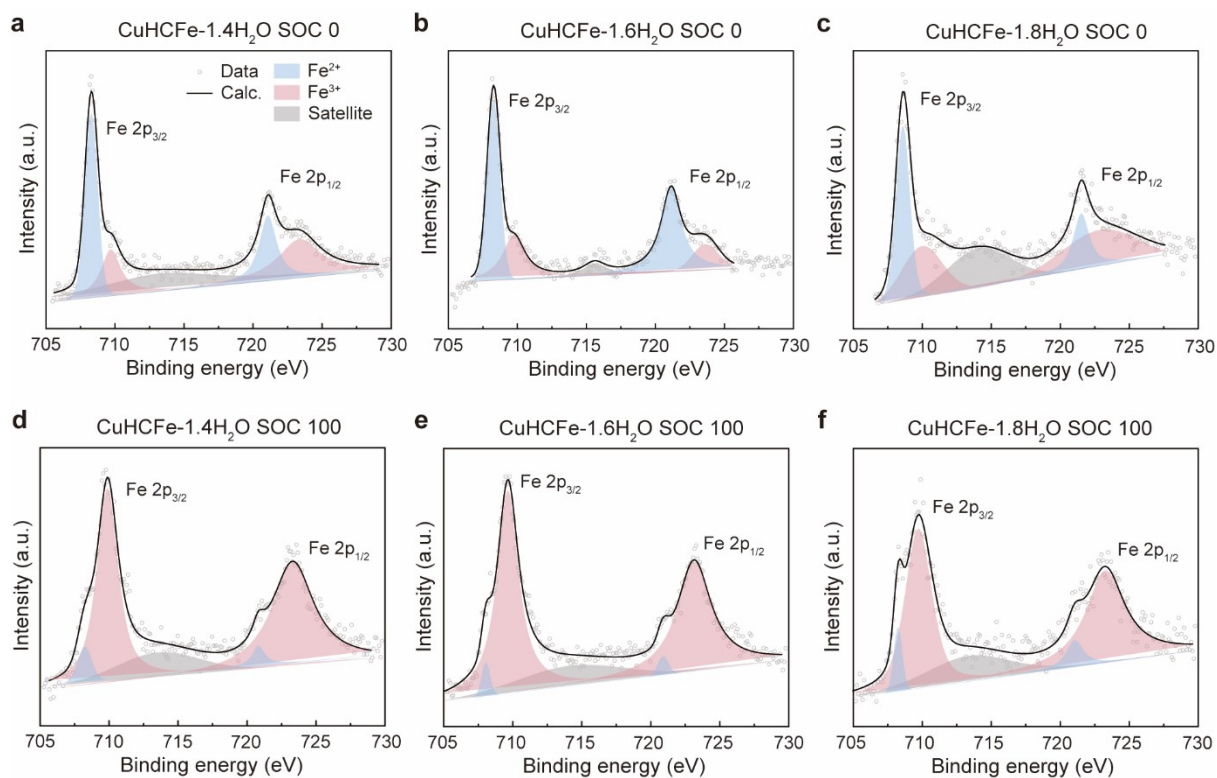
|                            | <b>Pristine</b>                                                  | <b>SOC 0</b>                                                                         | <b>SOC 100</b>                                                                       |
|----------------------------|------------------------------------------------------------------|--------------------------------------------------------------------------------------|--------------------------------------------------------------------------------------|
| CuHCFE-1.8H <sub>2</sub> O | Cu[Fe(CN) <sub>6</sub> ] <sub>0.658</sub> · 1.75H <sub>2</sub> O | Na <sub>0.488</sub> Cu[Fe(CN) <sub>6</sub> ] <sub>0.657</sub> · 2.88H <sub>2</sub> O | Na <sub>0.279</sub> Cu[Fe(CN) <sub>6</sub> ] <sub>0.673</sub> · 2.12H <sub>2</sub> O |
| CuHCFE-1.4H <sub>2</sub> O | Cu[Fe(CN) <sub>6</sub> ] <sub>0.670</sub> · 1.41H <sub>2</sub> O | Na <sub>0.611</sub> Cu[Fe(CN) <sub>6</sub> ] <sub>0.667</sub> · 3.87H <sub>2</sub> O | Na <sub>0.291</sub> Cu[Fe(CN) <sub>6</sub> ] <sub>0.674</sub> · 2.07H <sub>2</sub> O |



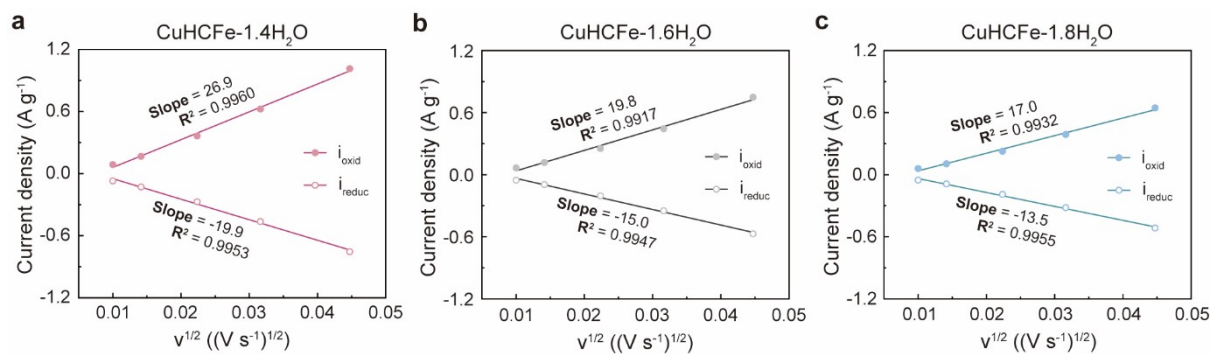
**Figure S1 | The Particle Size Distribution of CuHCF Samples.**



**Figure S2 | The XRD Spectra of CuHCFE Samples.** The structures of CuHCFE-1.4H<sub>2</sub>O, CuHCFE-1.6H<sub>2</sub>O, and CuHCFE-1.8H<sub>2</sub>O are cubic.

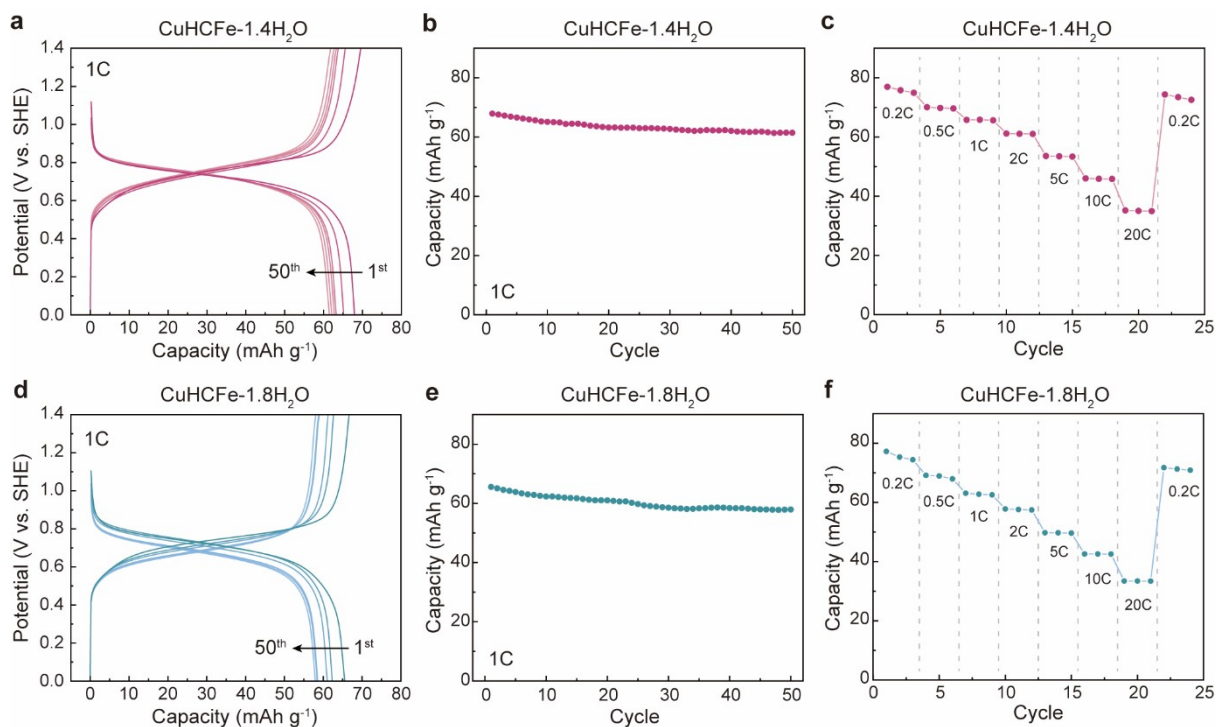


**Figure S3 | The X-ray Photoelectron Spectroscopy (XPS) Spectra of Fe 2p. CuHCF Electrode in Aqueous Electrolyte.** a-c) The XPS Spectra of a) CuHCF-1.4H<sub>2</sub>O, b) CuHCF-1.6H<sub>2</sub>O, and c) CuHCF-1.8H<sub>2</sub>O at SOC 0. d-f) The XPS Spectra of d) CuHCF-1.4H<sub>2</sub>O, e) CuHCF-1.6H<sub>2</sub>O, and f) CuHCF-1.8H<sub>2</sub>O at SOC 100.

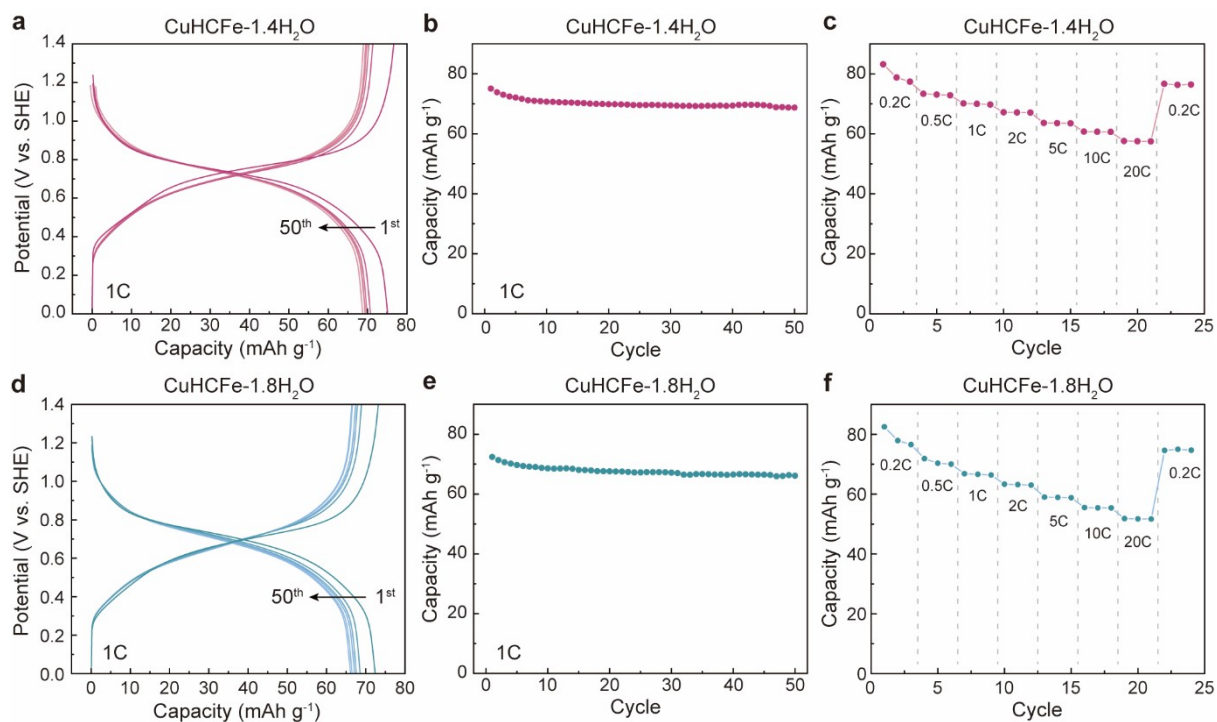


**Figure S4 | Scan Rate-current Density Graph of a) CuHCF-1.4H<sub>2</sub>O, b) CuHCF-1.6H<sub>2</sub>O, and c) CuHCF-1.8H<sub>2</sub>O.**

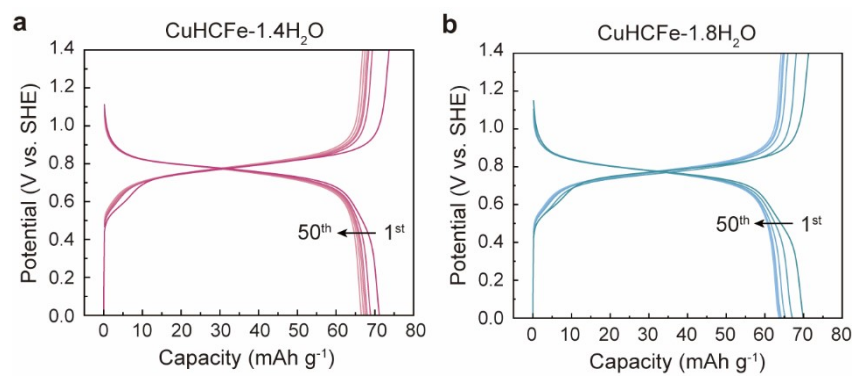




**Figure S5 | The Electrochemical Properties of CuHCF Samples in 1M NaClO<sub>4</sub> in DEGDME.** The charge-discharge curves of a) CuHCF-1.4H<sub>2</sub>O and d) CuHCF-1.8H<sub>2</sub>O. The specific capacity of b) CuHCF-1.4H<sub>2</sub>O and e) CuHCF-1.8H<sub>2</sub>O for 50 cycles at 1C. The rate capability of c) CuHCF-1.4H<sub>2</sub>O and f) CuHCF-1.8H<sub>2</sub>O from 0.2C (12 mA g<sup>-1</sup>) to 20C (1200 mA g<sup>-1</sup>).



**Figure S6 | The electrochemical properties of CuHCF samples in 1M NaClO<sub>4</sub> in acetonitrile.** The charge-discharge curves of a) CuHCF-1.4H<sub>2</sub>O and d) CuHCF-1.8H<sub>2</sub>O. The specific capacity of b) CuHCF-1.4H<sub>2</sub>O and e) CuHCF-1.8H<sub>2</sub>O for 50 cycles at 1C. The rate capability of c) CuHCF-1.4H<sub>2</sub>O and f) CuHCF-1.8H<sub>2</sub>O from 0.2C (12 mA g<sup>-1</sup>) to 20C (1200 mA g<sup>-1</sup>).



**Figure S7 | The Charge-discharge Curves of a) CuHCF-1.4H<sub>2</sub>O and b) CuHCF-1.8H<sub>2</sub>O in 1M NaClO<sub>4</sub> in EC:DEC (1:1) at 1C (60 mA g<sup>-1</sup>)**