

ARTICLE

## Exploring the Impact of Lithium Halide-based Redox Mediators in Suppressing CO<sub>2</sub> Evolution in Li-O<sub>2</sub> Cells

Received 00th January 20xx,  
Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

Sri Harsha Akella <sup>a</sup>, Muniyandi Bagavathi <sup>a</sup>, Rosy <sup>b</sup>, Daneil Sharon <sup>c</sup>, Capraz Ozgur <sup>d</sup>, Malachi Noked <sup>a\*</sup>

<sup>a</sup>. Department of Chemistry, Bar-Ilan University, Ramat Gan, 529002, Israel.

Bar-Ilan Institute of Nanotechnology and Advanced Materials, Ramat Gan, 529002, Israel. Email: Malachi.Noked@biu.ac.il \*Corresponding Author

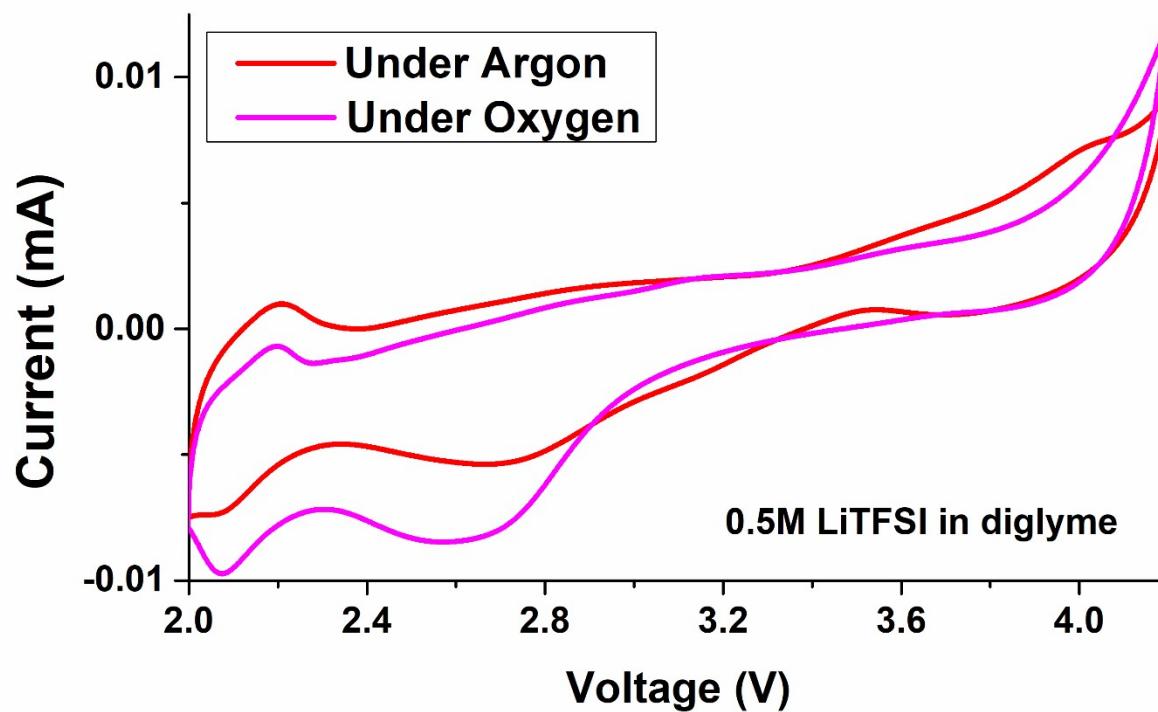
<sup>b</sup>. Department of Chemistry, Indian Institute of Technology (Banaras Hindu University), Uttar Pradesh, Varanasi, 221005, India. Email: rosy.chy@iitbhu.ac.in

<sup>c</sup>. The Institute of Chemistry, The Hebrew University of Jerusalem, Jerusalem, 9190401 Israel. Email: daniel.sharon@mail.huji.ac.il

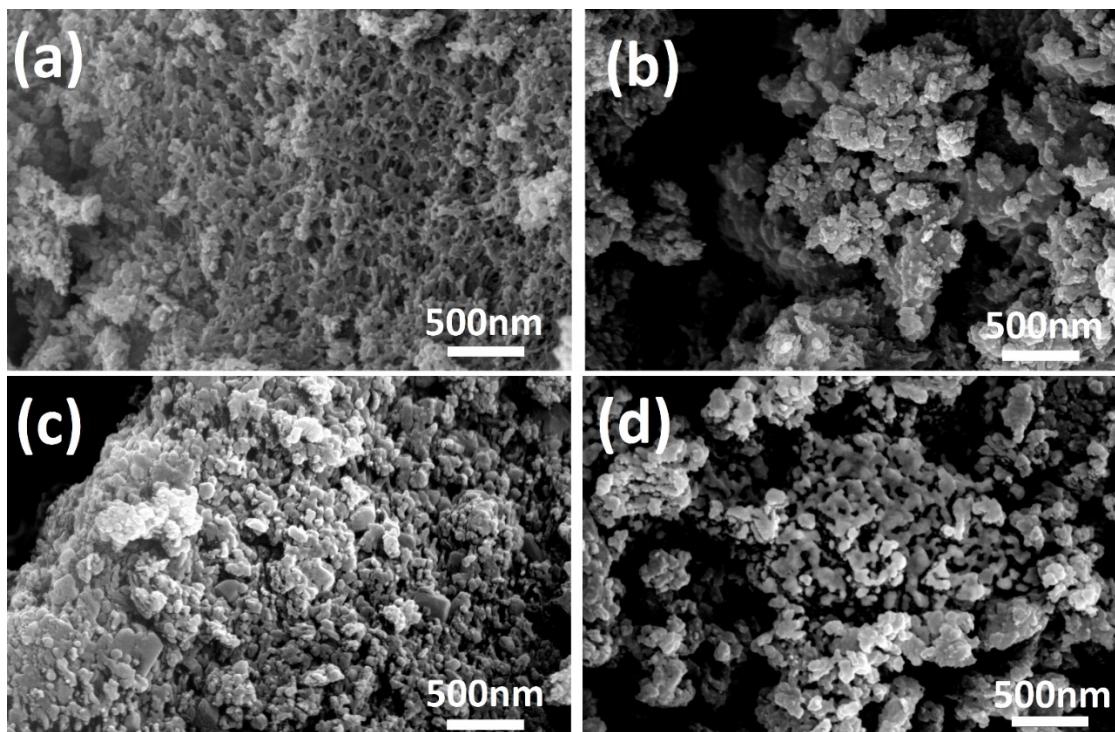
<sup>d</sup>. The School of Chemical Engineering, Oklahoma State University, Stillwater, Oklahoma 74078, United States of America. Email: ocapraz@okstate.edu

<sup>†</sup> Footnotes relating to the title and/or authors should appear here.

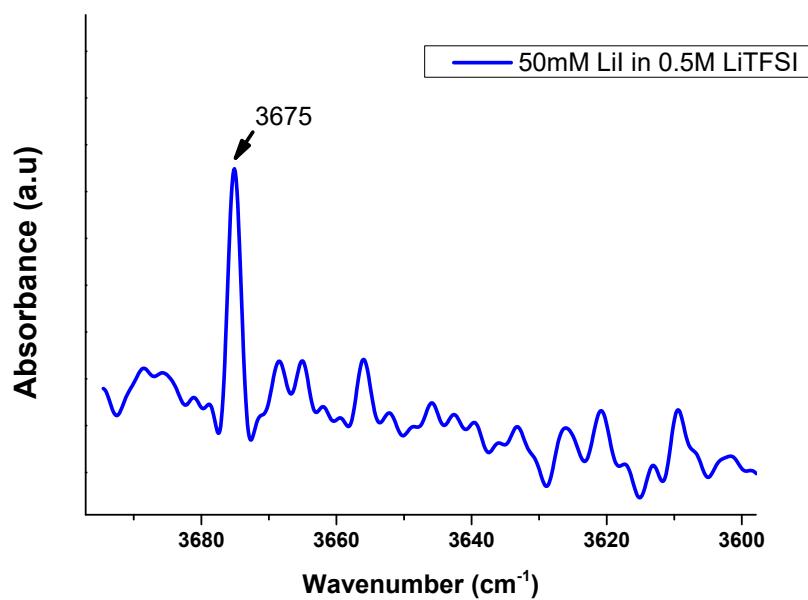
Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/x0xx00000x



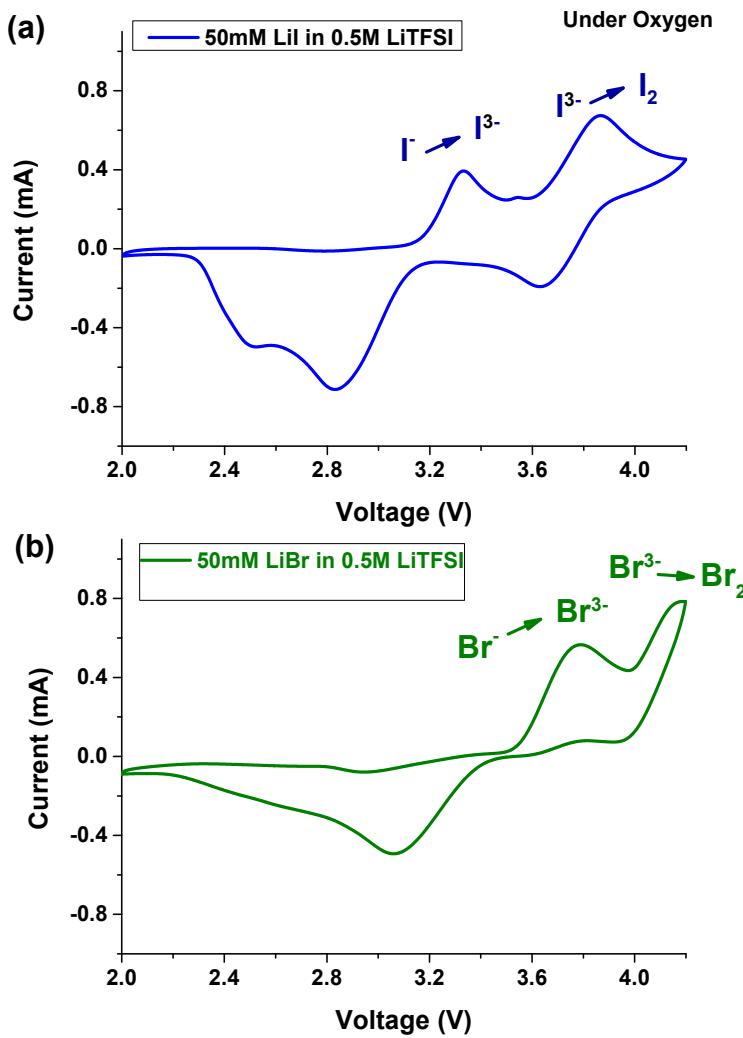
**Figure S1.** Cyclic voltammetry ( $5 \text{ mV s}^{-1}$ ) on Pt working electrode under argon atmosphere and oxygen atmosphere for 0.5M LiTFSI in diglyme.



**Figure S2.** HRSEM images of a) Pristine carbon paper, lithium oxide deposits on the carbon paper after ~30 hours discharge b) 0.5M LiTFSI c) 50mM LiBr in 0.5M LiTFSI, d) 50mM LiI in 0.5M LiTFSI in diglyme electrolytes.



**Figure S3.** ATR-FTIR spectra representing –OH vibrational band at 3675cm<sup>-1</sup> indicating the formation of LiOH in case of 50mM LiI in 0.5M LiTFSI in diglyme solution.



**Figure S4.** Cyclic voltammetry ( $5 \text{ mV s}^{-1}$ ) on Pt working electrode under oxygen atmosphere for (a) 50mM LiI in 0.5M LiTFSI and (b) 50mM LiBr in 0.5M LiTFSI in diglyme electrolyte.