

Electronic Supplementary Material (ESI).
This journal is © The Royal Society of Chemistry

RSCPublishing

Electronic Supplementary Information

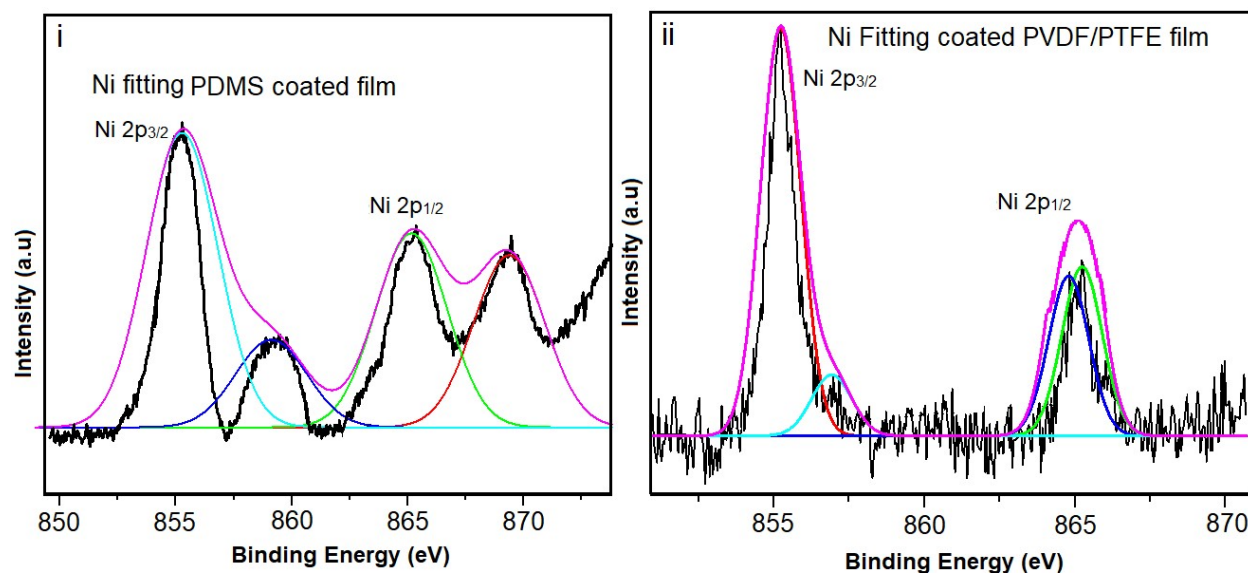
Extreme Low Temperature Environment Operatable Hybrid Dual-Functioning Energy Device Driven from Supercapacitor/Piezo-Tribo Electric Generator System

Samayanan Selvam^{a,†}, Young-Kwon Park^{b,†}, Jin-Heong Yim^{a,*}

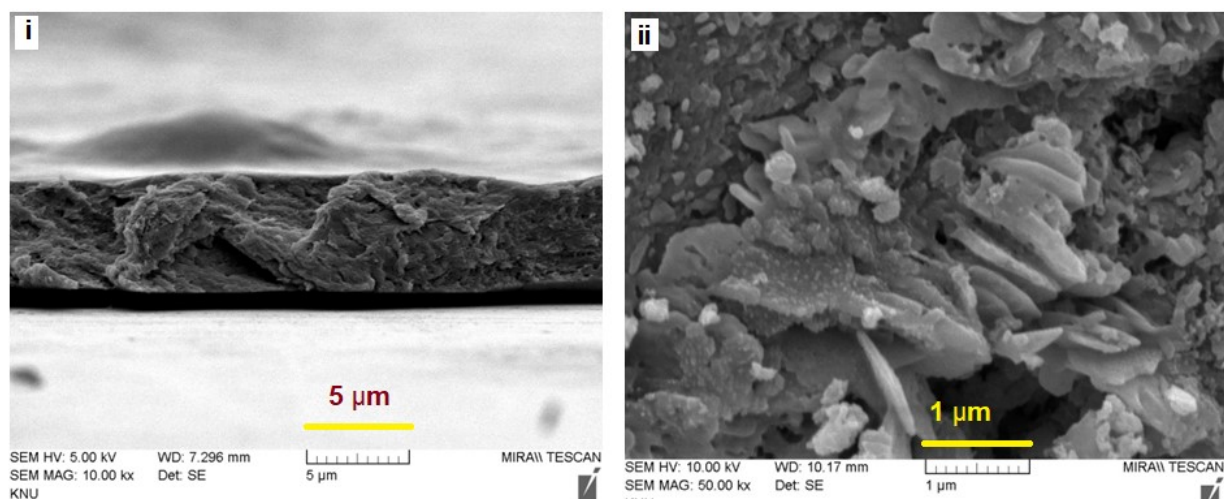
^aDivision of Advanced Materials Engineering, Kongju National University, Budaedong 275, Seobuk-gu, Cheonan-si, Chungnam 31080, South Korea

^bSchool of Environmental Engineering, University of Seoul, Seoul, 02504 Korea

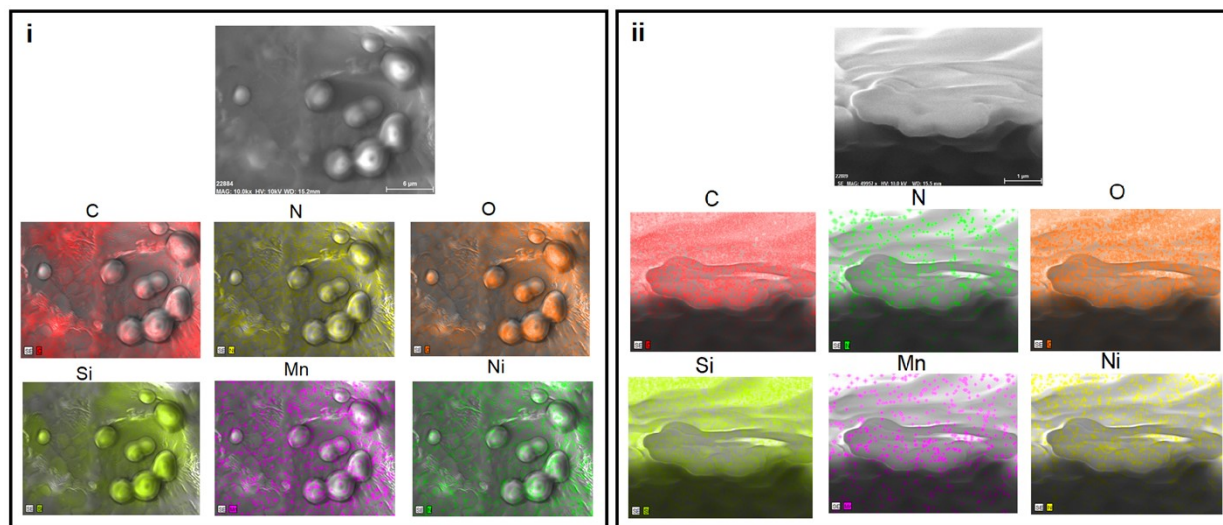
E-mail: jhyim@kongju.ac.kr;



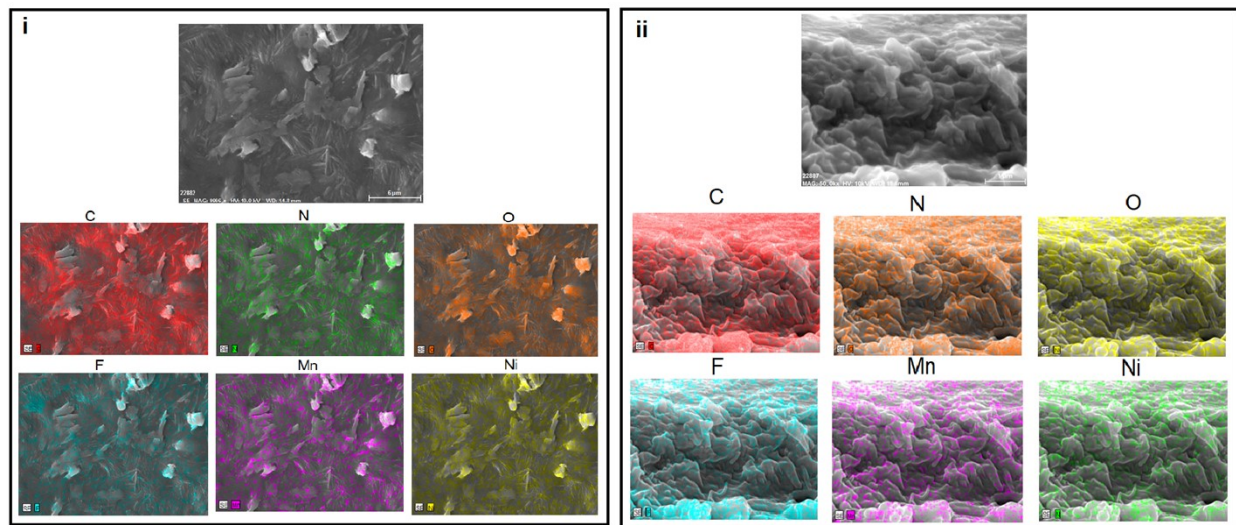
ESI. Fig. 1. (i and ii) Ni XPS fitting details of PDMS and PVDF-PTFE coated composite



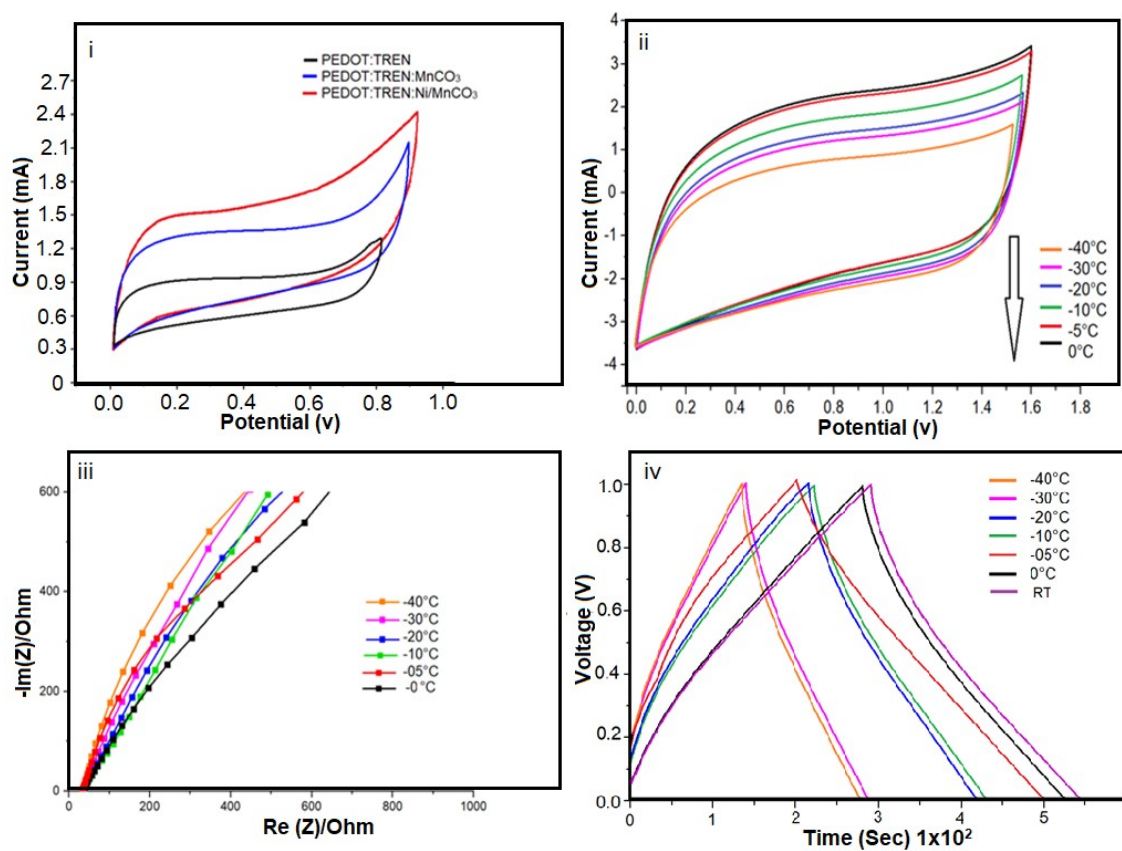
ESI. Fig. 2. (i and ii) SEM images of Cross section images of PEDOT:TREN:Ni@MnCO₃ composite



ESI Fig. 3. EDS mapping of PEDOT:TREN:Ni@MnCO₃/PDMS Composite Surface (i) and Cross section (ii)



ESI Fig. 4 EDS mapping of PEDOT:TREN:Ni@MnCO₃/PVDF-PTFE Surface (i) and Cross section (ii)



ESI Fig. 5 CV profile of PEDOT:TREN, PEDOT:TREN:MnCO₃ and PEDOT:TREN, PEDOT:TREN:Ni/MnCO₃ composite without GO (i), CV details at various current rate (ii) and EIS studies (iii) and (iv) Charge-discharge curves.

Calculations

The Gravimetric specific capacitance was calculated from the galvanostatic discharge curves, using the following equation 1.

$$C = \frac{I\Delta t}{m\Delta V} \text{1}$$

Also, the aerial capacitance was calculated from equation 2

$$C = \frac{I\Delta t}{A\Delta V} \text{2}$$

Where (I) is charge or discharge current, Δt (s) is the time for a full charge or discharge, m (g) designates the mass of the active material, A is the area of the active materials and ΔV signifies the voltage change after a full charge or discharge.

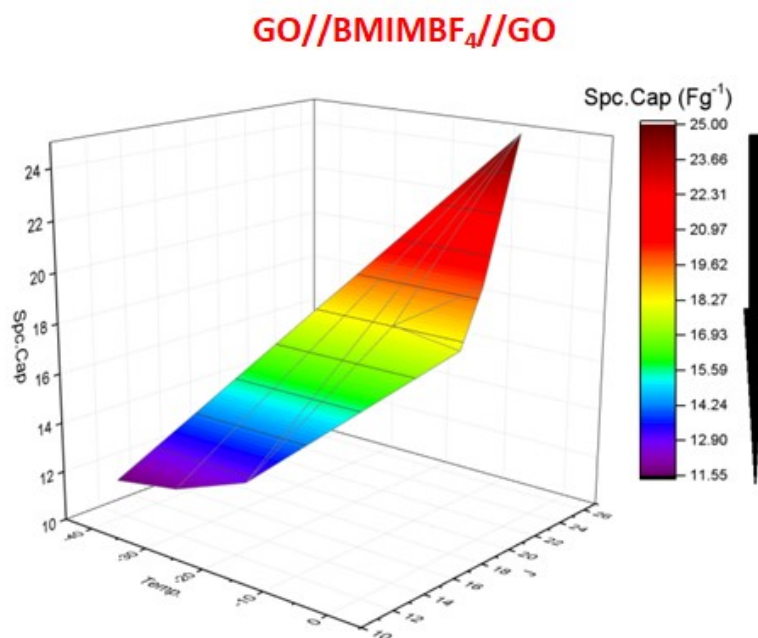
The energy density (E) considered by equation 3 .

$$E = \frac{C(\Delta V)^2}{2} \text{ WhKg}^{-1} \text{3}$$

Where C is the specific capacitance of the active materials, and ΔV is the potential window of discharge.¹⁻⁵

EIS. Table 1. Specific capacitance calculation details of PEDOT:TREN:Ni@MnCO₃

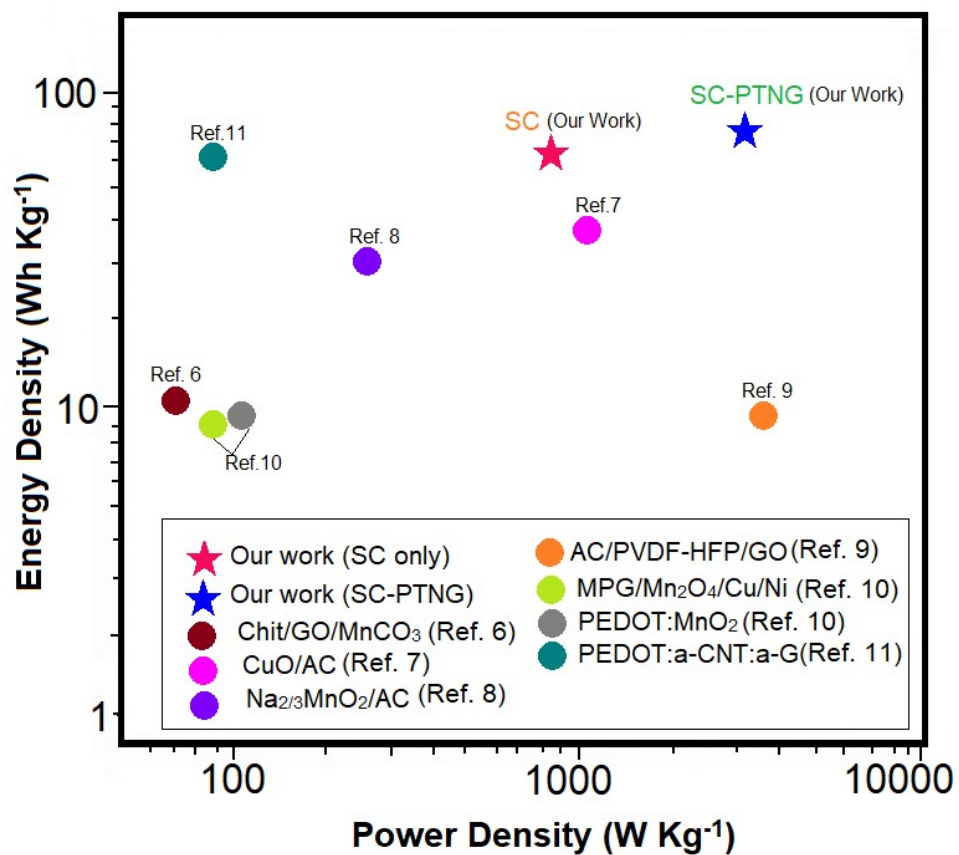
Temp.	Gravimetric Cap.
	Fg ⁻¹
-40°C	275
-30°C	283
-20°C	350
-10°C	426
-05°C	504
0°C	491
RT	475



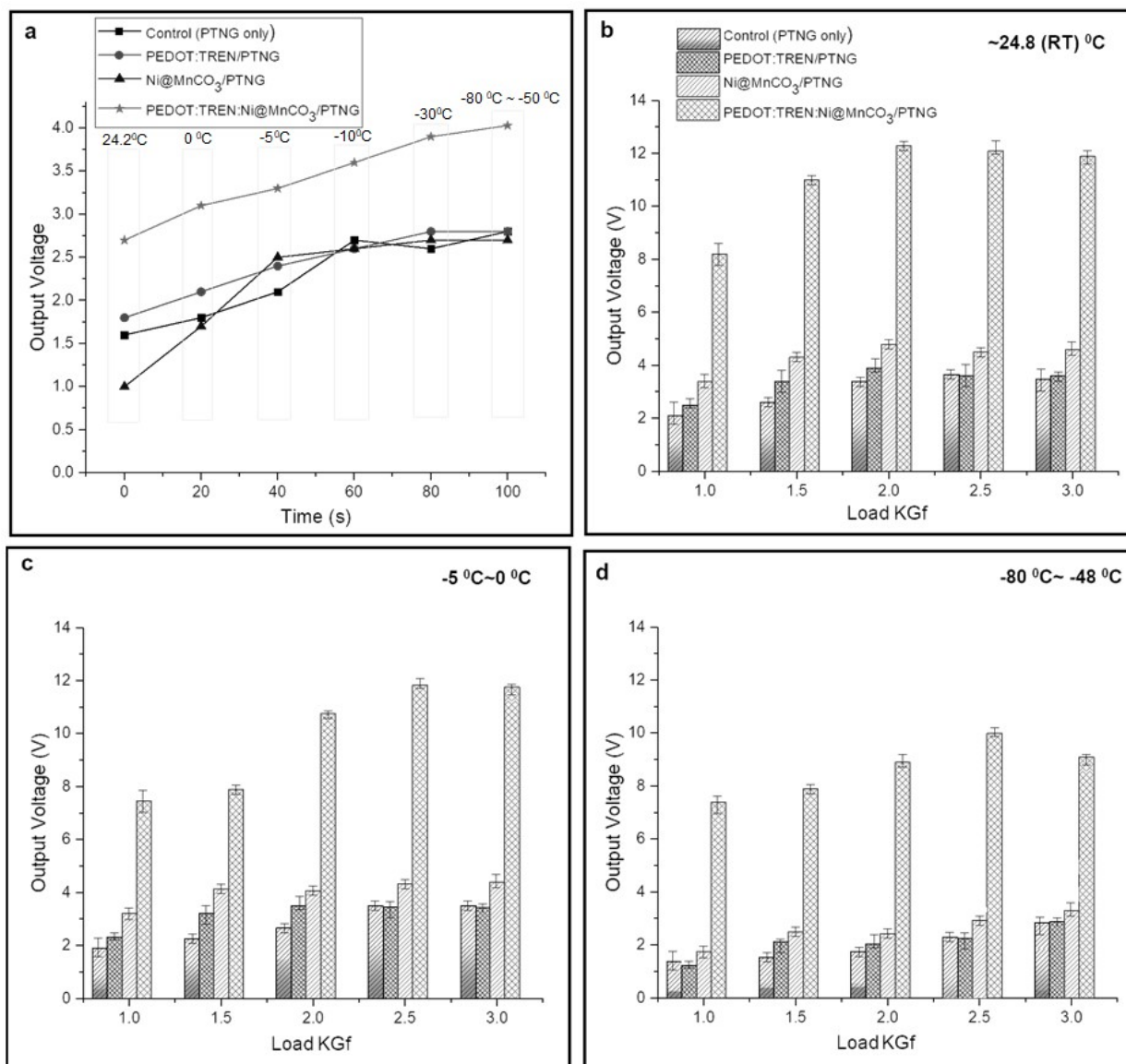
ESI. Fig 6 Electrolyte performance at low temperature conditions in terms of Specific capacitance and calculation details

EIS. Table 2. Specific capacitance calculation details of Electrolyte performance

Temp.	Gravimetric Cap.
	Fg ⁻¹
-40°C	11.6
-30°C	11.9
-20°C	12.7
-10°C	18.3
-05°C	25
0°C	20
RT	18



ESI. Fig 7. Energy density and Power density comparison from Ragone plot from reports on MnCO₃, Ni, PEDOT related Asymmetric supercapacitors (ASC)



ESI. Fig 8. Control experiment PTNG performance test under bending and twist conditions at after 1000 cycles workout; (a); output voltage error bar diagram of PTNG performance test under various load condition and temperature ranges (b-d) respectively.

ESI. Table 3. Comparative analysis of similar SC-PTNG from the literature.

Self-powered supercapacitor (SC-PTNG)	System	Electrolyte	Temp. (°C)	Specific capacitance	Charging Volt/ Output Volt)	Ref
PEDOT:TREN:PDMS:Ni@MnCO ₃ / PEDOT:TREN:PVDF-PTFE:Ni@MnCO ₃	SC- Piezo-Tribo hybrid	[BMIM][BF ₄]	RT	542 Fg ⁻¹	22 V	Our work
PEDOT:TREN:PDMS:Ni@MnCO ₃ / PEDOT:TREN:PVDF-PTFE:Ni@MnCO ₃	SC- Piezo-Tribo hybrid	[BMIM][BF ₄]	-80	317 Fg ⁻¹	11.9 V	Our work
AgNWS/NiOH/ P(VDF-TrFE)	SC- Piezo-Tribo hybrid	(PVA/KOH	RT	3.47 mFcm ⁻²	150 V	ESI. Ref. 12
Siloxene–PVDF piezofiber	SC- Piezo- separately	TEABF ₄	RT	27.58 mFcm ⁻²	207 mV	ESI. Ref. 13
Co-Fe ₂ O ₃ @ACC	SC- Piezo- separately	PVA-KCl- BaTiO ₃	RT	2.8 mFcm ⁻²	120 mV	ESI. Ref. 14
3D AG/PTFE/PDMS	SC-Tribo separately	NA	RT	550 Fg ⁻¹	3.2V	ESI. Ref. 15
Silicone-CF fiber	SC-Tribo separately	H ₃ PO ₄ / PVA gel	RT	31.25Fg ⁻¹	42.9 V	ESI. Ref. 16

ESI. References

1. N. F. M. Yusof, N. H. Idris, M. F. Din, S. R. Majid, N. Harun, M. M. Rahman, (2020) 10:9207 | <https://doi.org/10.1038/s41598-020-66148-w>
2. M. Manoj, D. Mangalaraj, N. Ponpandian and C. Viswanathan, RSC Adv., 2015, 5, 48705-48711
3. Z. Yang, Y. Jiang, L. X. Yu, B. Wen, F. Li, S. Sun and T. Hou, J. Mater. Chem., 2005, 15, 1807–1811
4. S. Selvam, B. Balamuralitharan, S.N. Karthick, K.V.Hemalatha, K. Prabakar and Hee-Je Kim, Anal. Methods, 2016, 8, 7937-7943.
5. H. Chen, Z. Yan, X.Y. Liu, X. L. Guo, Y. X. Zhang, Z. H. Liu, Rational design of microsphere and microcube MnCO₃@MnO₂ heterostructures for supercapacitor electrodes, Journal of Power Sources 353 (2017) 202-209
6. S. Selvam, J.-H. Yim, J. Energy Storage, 2021, 43, 103300
7. A. K. Mishra, A. K. Nayak, A. K. das, D. Pradhan, J. Phys. Chem. C 2018, 122, 11249
8. A. A. Nechikott, P. K. Nayak, RSC Adv., 2023, 13, 14139
9. X. Yang, F. Zhang, L. Zhang, T. Zhang, Y. Huang and Y. Chen, Adv. Funct. Mater., 2013, 23, 3353–3360.
10. W. Li, X. Xu, C. Liu, M. C. Tekell, J. Ning, J. Guo, J. Zhang, D. Gan, Adv. Funct. Mater. 2017, 27, 1702738
11. Y. Zhou, N. Lachman, M. Ghaffari, H. Xu, D. Bhattacharya, P. Fattahi, M. R. Abidian, S. Wu, K. K. Gleason, B. L. Ward;e. Q. M. Zhang, J. Mater. Chem. A, 2014, 2, 9964
12. S. Qin, Q. Zhabg, X. Yang, M. Liu, Q. Sun, Z. L. Wang, Adv. Energy Mater. 2018, 8, 1800069

13. K. Krishnamoorthy, P. Pazhamalai, V. K. Mariappan, S. S. Nardekar, S. Sahoo, S.-J. Kim, . Nat Commun 11, 2351 (2020).
<https://doi.org/10.1038/s41467-020-15808-6>
14. D. Zhou, F. Wang, J. Yang, L, -Z, Fan, Chemical Engineering Journal, 2021, 406, 126825
15. J. Dong, S. Huang, J. Luo, J. Zhao, F. R. Fan, Z. -Q. Tian, Jung, Nano Energy, 2022, 95, 106971
16. Y. Yang, L. Xie, Z. Wen, C. Chen, X. Chen, A. Wei, P. Cheng, X. Xie, X. Sun, ACS Appl. Mater. Interfaces 2018, 10, 42356