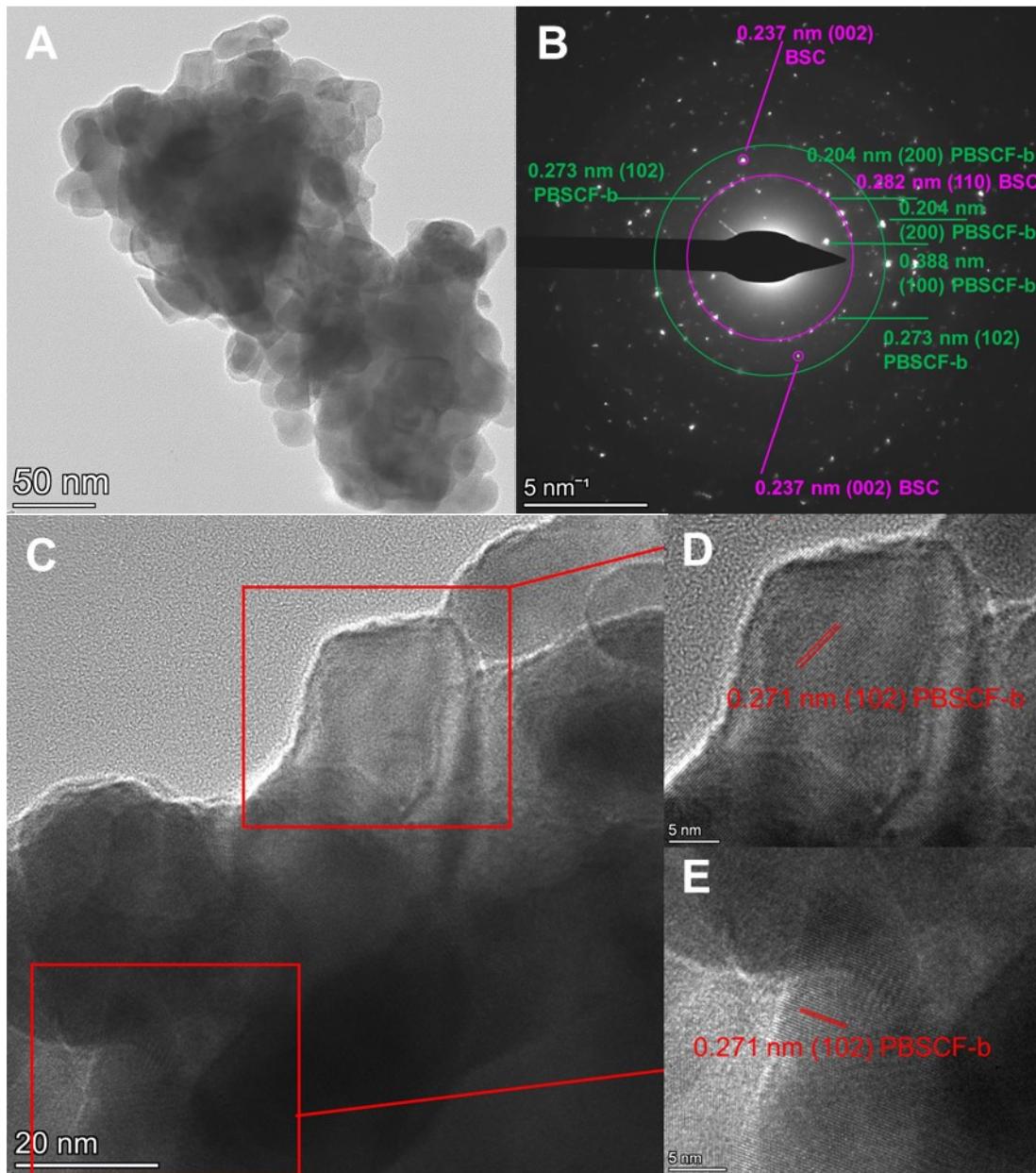
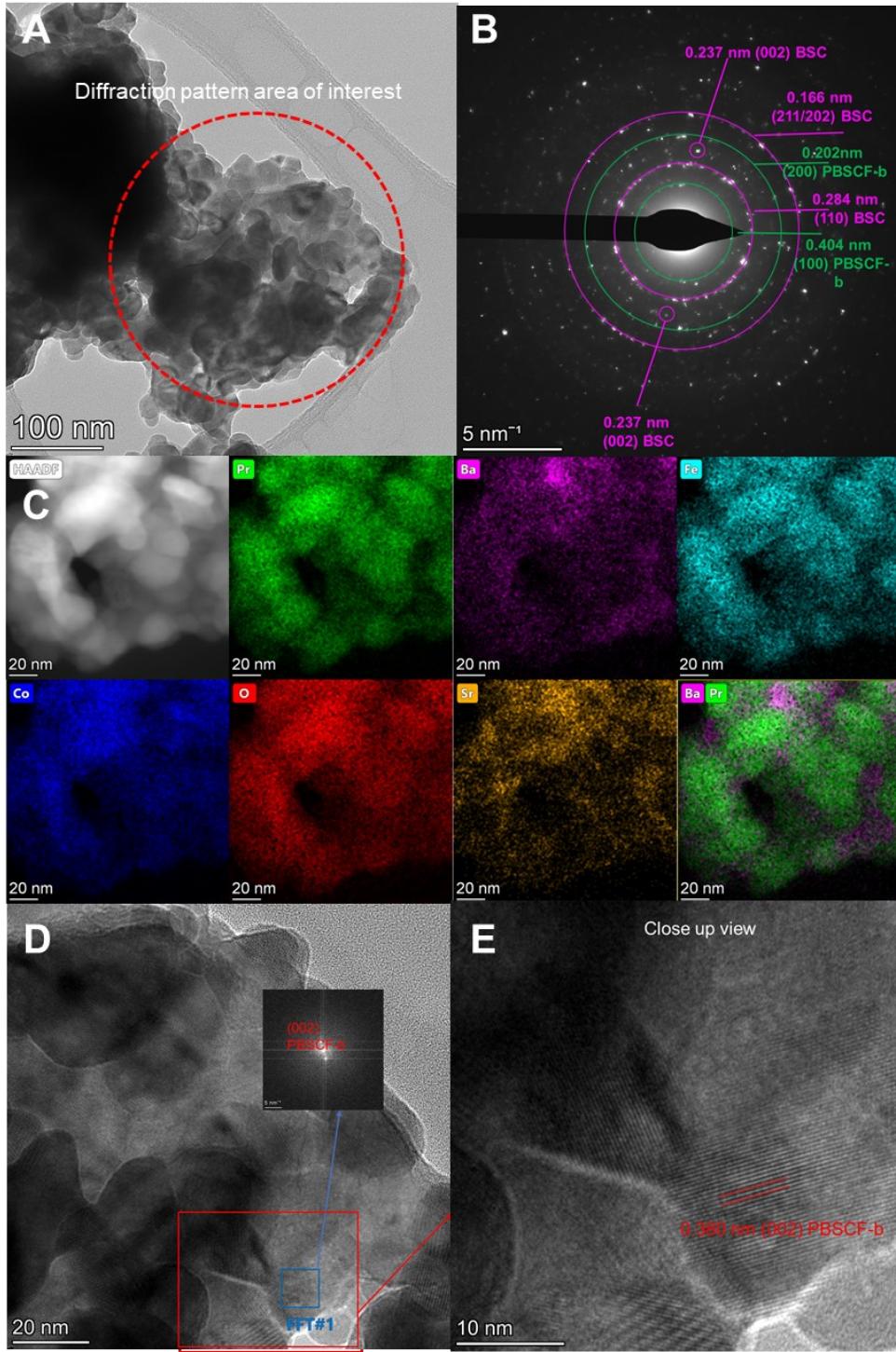


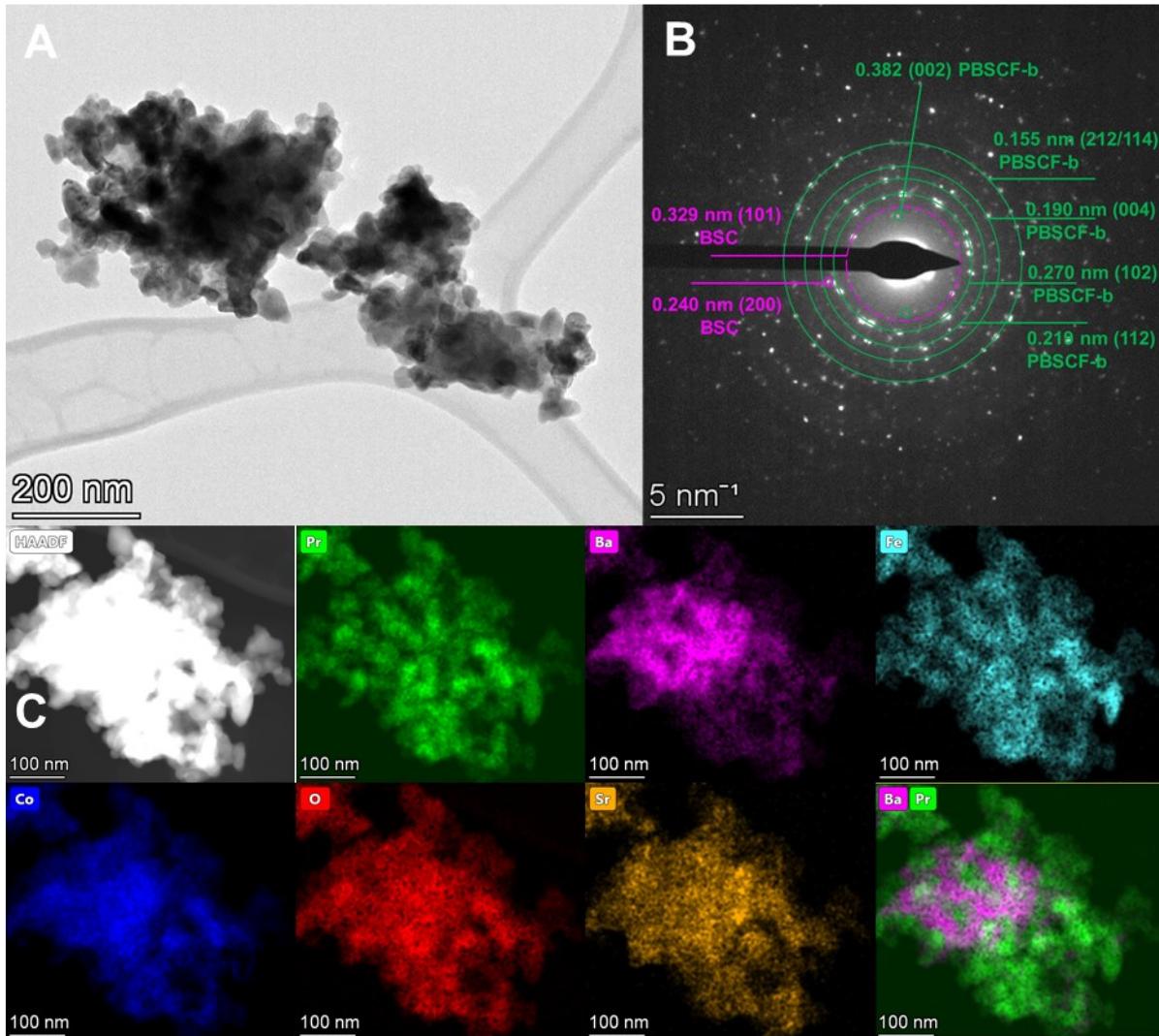
**Figure S1. XRD patterns of the previous PBSCF-a phase and the hybrid electrode.**  
The previous PBSCF-a is a single-phase double perovskite, which was synthesized by a wet chemistry method and sintered at 900 °C for 5 hours. The hybrid electrode consists of two phases, PBSCF-b and BSC, which was prepared by the same wet chemistry method and sintered at 750-770 °C.



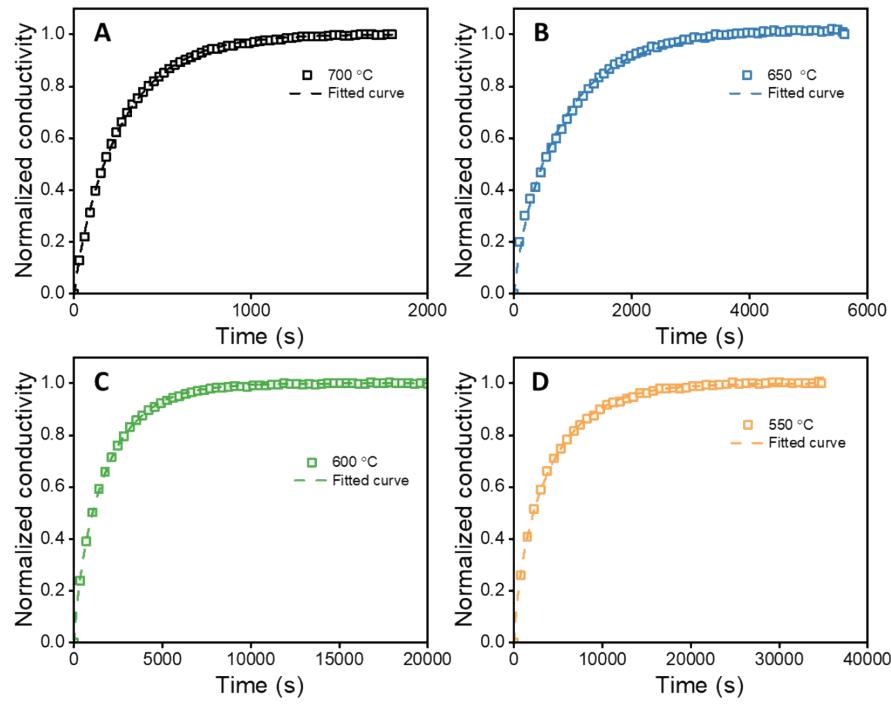
**Figure S2.** (A) TEM images of the hybrid electrode. (B) SAED patterns indicate the hybrid electrode is composed of the PBSCF-b phase and BSC phase. (C) HR-TEM image of the hybrid electrode grain and the zoomed-in areas: (D) and (E). The lattice fringes with the d-spacing of 0.271 nm, correspond to the (102) planes of the new PBSCF-b phase.



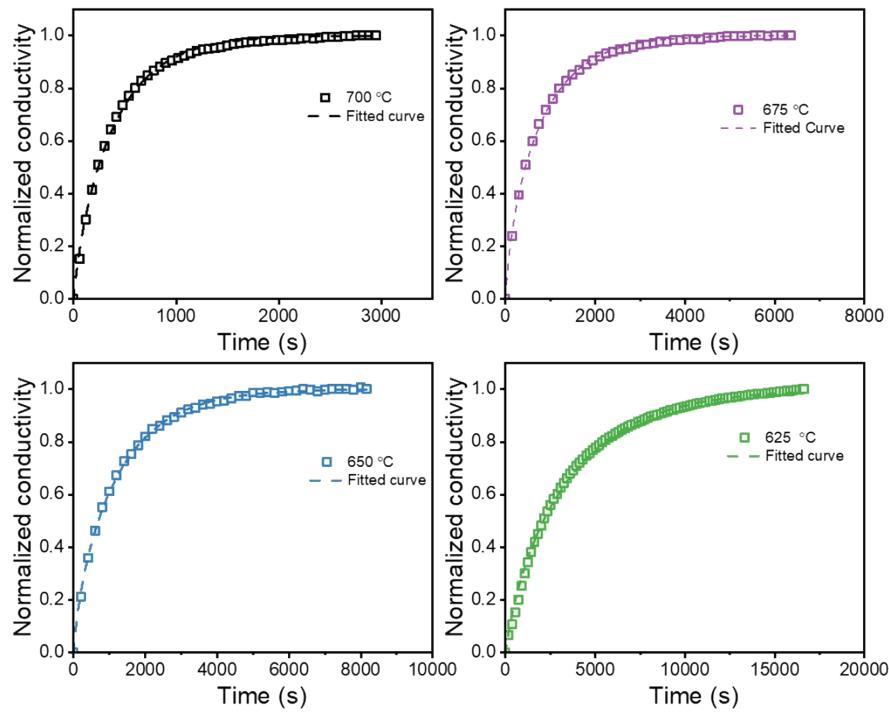
**Figure S3.** (A) Additional TEM images of the hybrid electrode. (B) Additional SAED patterns show the hybrid electrode is composed of a new PBSCF-b phase and the BSC phase. (C) EDS mapping images of the hybrid electrode show the BSC phase mainly contains Ba, Sr, Co, and O elements. (D) HR-TEM image of the hybrid electrode grain and the zoomed-in areas. (E) HR-TEM image shows lattice planes with d-spacing of 0.380 nm, corresponding to the (002) planes of the new PBSCF-b phase.



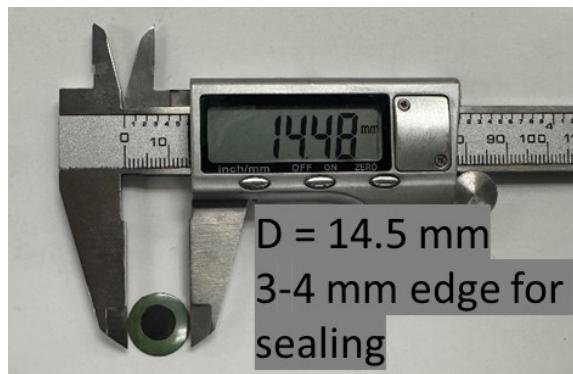
**Figure S4.** (A) TEM image of the hybrid electrode. (B) SAED patterns show the lattices with the d-spacings of 0.382 nm, 0.155 nm, 0.190 nm, 0.270 nm, and 0.219 nm, corresponding to the (002), (212/114), (004), (102/110), and (112) planes of PBSCF-b, respectively. And those with the d-spacing of 0.329 nm and 0.240 nm correspond to the (101) and (200) planes of BSC, respectively. (C) EDS mapping images of the hybrid electrode confirm that it is a composite of PBSCF-b and BSC phases.



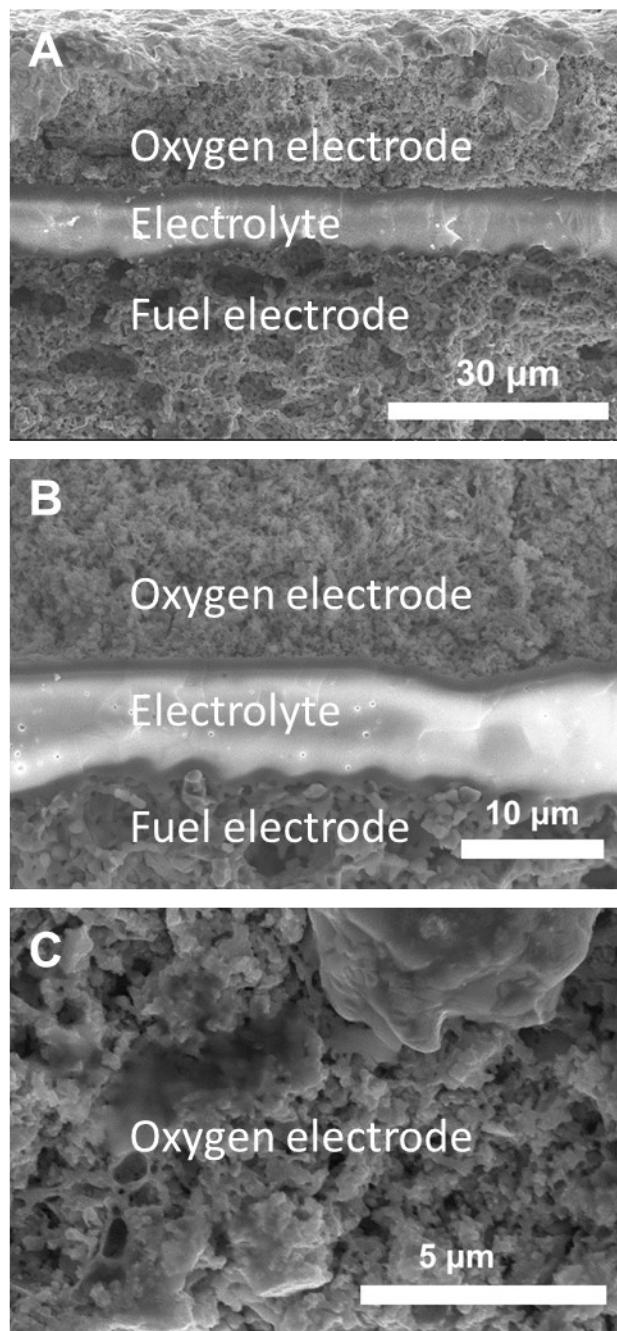
**Figure S5.** ECR experimental results and the fitted results for PBSCF-b coated with BSC at 550 °C, 600 °C, 650 °C, and 700 °C after changing the oxygen partial pressure from 2 vol.% (490 sccm Ar + 10 sccm O<sub>2</sub>) to 20 vol.% (400 sccm Ar + 100 sccm O<sub>2</sub>).



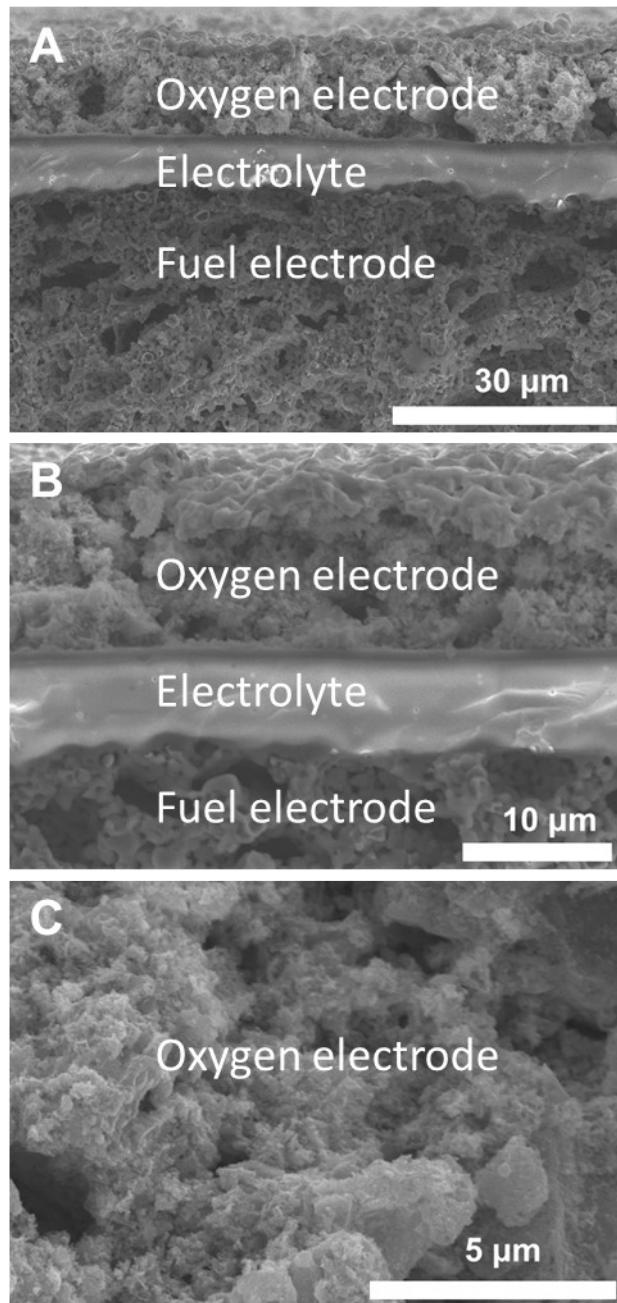
**Figure S6.** ECR experimental results and the fitted results for the previous PBSCF-a at 625 °C, 650 °C, 675 °C, and 700 °C after changing the oxygen partial pressure from 2 vol.% (490 sccm Ar + 10 sccm O<sub>2</sub>) to 20 vol.% (400 sccm Ar + 100 sccm O<sub>2</sub>).



**Figure S7.** Photos of a representative fuel electrode-supported SOEC fabricated in this work to evaluate the oxygen electrodes. The oxygen electrode effective area is 0.5 cm<sup>2</sup>.



**Figure S8.** Representative SEM images of SOECs studied in this work, which are highly reproducible, highly uniform, and defect-free, and have dense electrolytes (thickness of electrolyte = ~8  $\mu\text{m}$ ).



**Figure S9.** SEM images of the SOEC with PBSCF-b+BSC hybrid electrode after performance evaluation in steam electrolysis mode. No cracking or electrolyte-electrode delamination were observed.

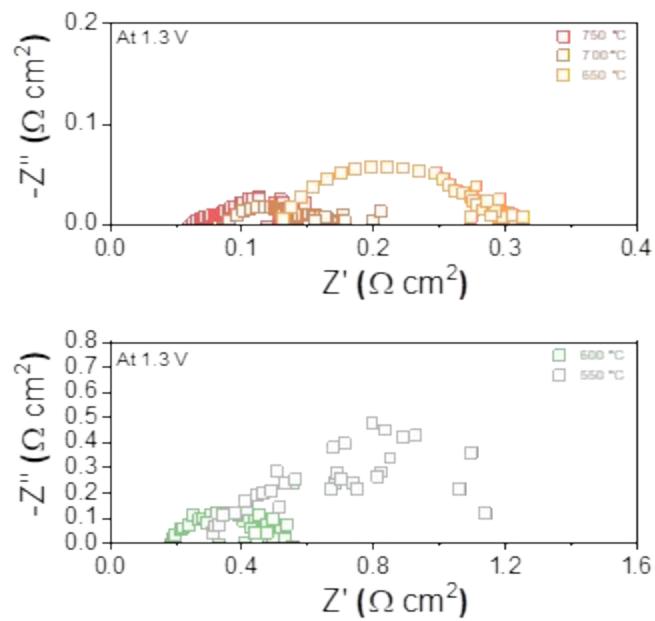


Figure S10. EIS spectra collected at an applied voltage of 1.3 V and an operating temperature ranging from 550 to 750 °C for SOEC with the hybrid electrode.

**Table S1.** Comparisons YSZ electrolyte-based SOEC performance.

Half-cell configuration	Oxygen electrode materials	Temperature (°C)	Current densities at 1.3 V (A cm <sup>-2</sup> )	Literature
Ni-YSZ YSZ	Hybrid electrode	750	4.4	This work
		700	2.3	
		650	1.4	
		600	0.9	
		550	0.46	
Ni-YSZ YSZ GDC	SSC-SDC	800	4.08	1
		750	3.13	
		700	1.9	
		650	1.1	
Ni-YSZ YSZ GDC	STFC	800	3	2
		750	2.3	
		700	1.5	
		650	1	
		600	0.6	
LSCFN55 YSZ	SSNC	800	0.7	3
		750	0.45	
430L YSZ SSZ	SSZ-Nd <sub>2</sub> O <sub>3</sub> /NNO	800	2.05	4
		750	1.62	
		700	1.06	
		650	0.6	
Ni-YSZ Ni-SSZ SSZ SNDC	PBSCF-GDC	750	2.1	5
		700	1.45	
		650	0.94	
		600	0.51	
Ni-YSZ YSZ	LSM-DYSB	700	1.32	6
		650	0.9	
		600	0.47	
Ni-YSZ YSZ SNDC	LSCF-SNDC	750	1.37	7
		700	0.97	
		650	0.58	
Ni-YSZ YSZ GDC	LSCF-GDC	800	2.75	8
		750	1.7	
		700	1.25	
		650	0.6	
Ni-YSZ YSZ	LSM-YSB	800	1.52	9
		750	0.97	
		700	0.7	
		650	0.46	
Ni-YSZ YSZ GDC	STFC-PrOx	800	4.25	10
		750	3.6	

		700	2.9	
		650	2.25	
		600	1.25	
		550	0.5	
Ni-YSZ YSZ GDC	PBFZr-GDC	700	2.14	11
Ni-YSZ YSZ GDC	LSCF-GDC	750	2.1	12
Ni-YSZ YSZ GDC	PBNF0.1	750	0.65	13
Ni-YSZ YSZ	PBSM3	750	2.75	14
		700	1.85	
Ni-YSZ YSZ	LSM-YSZ	700	0.40	15
Ni-YSZ YSZ GDC	LSCF	700	0.82	16
Ni-YSZ YSZ	LSCF-YSZ	750	0.91	17
Ni-YSZ YSZ SDC	BSCF-SDC	750	0.23	18
Ni-YSZ YSZ	PNO	700	0.28	19
Ni-YSZ YSZ GDC	LSCF-GDC	750	0.34	20
Ni-YSZ YSZ GDC	LSCF-GDC	750	0.77	21
Ni-YSZ ScSZ	Mn1.3Co1.3Cu0.4O4-ScSZ	750	1.4	22
Ni-YSZ YSZ GDC	LSFN-GDC	750	0.53	23
Ni-YSZ YSZ GDC	BCO-PBCC	750	1.36	24

**Table S2.** Performance comparison of this work and other state-of-the-art YSZ anode-supported hydrogen-fueled SOFCs reported recently.

Half-cell configuration	Oxygen electrode materials	Peak power density at 750 °C (W cm <sup>-2</sup> )	Literature
Ni-YSZ YSZ	Hybrid electrode	2.4	This work
Ni-YSZ YSZ GDC	STFC	1.75	2
Ni-YSZ YSZ GDC	STFC-PrO <sub>x</sub>	3	10
Ni-YSZ YSZ GDC	LSCF-GDC-PrO <sub>x</sub>	2	25
Ni-YSZ YSZ	LSCF-GDC	1.6	26
Ni-YSZ YSZ GDC	LSF-GDC-Pr <sub>6</sub> O <sub>11</sub>	1.57	27
Ni-YSZ YSZ GDC	PBFZr-GDC	1.9	11
Ni-YSZ YSZ GDC	PBCC-GDC	1.74	28
Ni-YSZ YSZ	SCP-GDC	1.4	29
Ni-YSZ YSZ	PBSCF-GDC	1.37	29
Ni-YSZ YSZ	MCO-GDC	0.7	30
Ni-YSZ YSZ	CMO-SDC	0.64	31
Ni-YSZ YSZ SDC	BCFN/BCO-LSCF	1.4	25
Ni-YSZ YSZ GDC	BSNM-SDC-Ag	1.3	32
Ni-YSZ YSZ GDC	SSC-SDC	2.44	1
Ni-YSZ YSZ ESB	LSM-ESB	1.75	33
Ni-YSZ YSZ GDC	NBCCF@GDC	1	34

**Table S3.** XRD refinement results of the hybrid electrode material.

Sample	Phase	Space group	a, b, c (Å)	α, β, γ (°)	Wt. frac. (%)	Chi	wR <sub>p</sub> (%)	R <sub>p</sub> (%)	d-spacing
New PBSCF-b+BSC	New PBSCF-b	Tetragonal; P4/mmm	3.88, 3.88, 7.74	90, 90, 90	89.5	1.978	13.68	11.04	0.195 nm, (200); 0.387 nm, (002); 0.273 nm, (102); 0.158 nm, (212);
	BSC	Hexagonal; P63/mmc	5.58, 5.58, 4.63	90, 90, 120	10.5				0.279 nm, (110); 0.167 nm, (202); 0.169 nm, (211); 0.230 nm, (002);

**Table S4.** Comparison in peak power densities at various temperature using different single cell configuration fueled by propane.

Single cell configuration (fuel electrode  electrolyte  oxygen electrode)	Temperature (°C)	Peak power density (W cm <sup>-2</sup> )	Literature
Ni-YSZ YSZ Hybrid electrode	750	2.6	This work
	700	1.7	
	650	0.95	
	600	0.54	
	550	0.26	
RP-SFM-SDC LDC LSGM LSCF-SDC	800	0.7	35
	750	0.35	
LSCrFeCo-GDC LDC LSGM LSCF-GDC	800	0.5	36
	750	0.38	
	700	0.18	
L-PBMCO LDC LSGM NBSCF-GDC	800	0.33	37
RP-PSFN-CFA LDC LSGM BCFN	800	0.59	38
PBMO LDC LSGM NBSCF-GDC	800	0.75	39
	750	0.3	
	700	0.17	
NTO-Ni-YSZ YSZ LSM-YSZ	700	0.15	40
BaO/Ni-YSZ YSZ SDC-LSCF	750	0.9	41
Ru-CeO <sub>2</sub>  PSZ Ni-YSZ YSZ LSCF-GDC	750	0.48	42
	700	0.385	
PSCFN-Co-Fe LSGM BCFN	800	0.6	43
	750	0.32	
RP-PSFR-FRA-GDC LSGM LSCF-GDC	800	0.5	44
	750	0.25	
	700	0.12	

**Table S5.** Crystal structure results obtained from TEM for the PBSCF-b phase.

Mul ti.	(h,k,l)	d* / nm <sup>-1</sup>	Θ <sub>Scatt</sub> / Deg.	Inten s.	Θ <sub>Bragg</sub> / mRad	V <sub>r</sub> [V  nm  e]	V <sub>i</sub> [V  nm  e]	Ampli. [V  nm  e]	d / nm	LP factor	s
1	(0,0,0)	0.000	0.000		0.000	12.339	0.935	12.375			
2	(0,0,1)	1.310	0.189	1	1.642	-0.154	0.247	0.291	0.764	0.7636	1
4	(1,0,0)	2.558	0.368	301	3.208	4.917	0.479	4.940	0.391	0.3909	1
2	(0,0,2)	2.619	0.377	561	3.284	9.623	0.699	9.649	0.382	0.3818	4
8	(1,0,1)	2.874	0.413	489	3.604	4.695	0.471	4.719	0.348	0.348	2
4	(1,1,0)	3.617	0.520	1	4.536	0.249	0.256	0.357	0.276	0.2764	2
8	(1,0,2)	3.661	0.527	300	4.591	4.146	0.453	4.170	0.273	0.2731	5
8	(1,1,1)	3.847	0.553	1000	4.824	7.779	0.642	7.806	0.260	0.2599	3
2	(0,0,3)	3.929	0.565	1	4.927	0.266	0.255	0.368	0.255	0.2545	9
8	(1,1,2)	4.466	0.642	2	5.600	0.274	0.252	0.372	0.224	0.2239	6
8	(1,0,3)	4.688	0.674	167	5.879	3.497	0.430	3.523	0.213	0.2133	10
4	(2,0,0)	5.116	0.736	244	6.415	6.259	0.593	6.287	0.195	0.1955	4
2	(0,0,4)	5.238	0.753	114	6.569	6.135	0.589	6.163	0.191	0.1909	16
8	(2,0,1)	5.281	0.759	1	6.622	0.252	0.247	0.353	0.189	0.1894	5
8	(1,1,3)	5.341	0.768	434	6.697	6.034	0.585	6.062	0.187	0.1872	11
8	(2,1,0)	5.720	0.822	98	7.172	2.955	0.408	2.983	0.175	0.1748	5
8	(2,0,2)	5.747	0.826	355	7.207	5.655	0.571	5.684	0.174	0.174	8
8	(1,0,4)	5.830	0.838	93	7.310	2.904	0.405	2.932	0.172	0.1715	17
16	(2,1,1)	5.868	0.844	183	7.358	2.886	0.404	2.915	0.170	0.1704	6
16	(2,1,2)	6.291	0.904	150	7.889	2.700	0.395	2.729	0.159	0.159	9
8	(1,1,4)	6.366	0.915	1	7.983	0.197	0.242	0.312	0.157	0.1571	18
8	(2,0,3)	6.450	0.927	1	8.089	0.192	0.241	0.309	0.155	0.155	13
2	(0,0,5)	6.548	0.941	0	8.211	0.187	0.241	0.305	0.153	0.1527	25
16	(2,1,3)	6.939	0.998	111	8.702	2.444	0.387	2.474	0.144	0.1441	14
8	(1,0,5)	7.030	1.011	54	8.816	2.410	0.385	2.441	0.142	0.1422	26

**Table S6.** Crystal structure results obtained from TEM for the BSC phase.

Multi .	(h,k,l)	d* / nm <sup>-1</sup>	Θ <sub>Scatt</sub> / Deg.	Intens .	Θ <sub>Bragg</sub> / mRad	V <sub>r</sub> [V  nm  e ]	V <sub>i</sub> [V  nm  e ]	Ampli. [V  nm  e ]	d / nm	LP factor	s
1	(0,0,0 )	0.000	0.00 0		0.000	28.782	2.090	28.858			
6	(1,0,0 )	2.084	0.30 0	168	2.613	4.877	0.693	4.926	0.48 0	0.480	1
6	(1,1,0 )	3.609	0.51 9	69	4.525	4.112	0.659	4.165	0.27 7	0.277	2
6	(2,0,0 )	4.167	0.59 9	1000	5.225	16.942	1.342	16.995	0.24 0	0.240	4
2	(0,0,2 )	4.232	0.60 9	321	5.307	16.749	1.337	16.805	0.23 6	0.236	4
12	(1,0,2 )	4.717	0.67 8	77	5.915	3.482	0.631	3.539	0.21 2	0.212	5
12	(2,1,0 )	5.513	0.79 3	51	6.913	3.060	0.610	3.121	0.18 1	0.181	5
12	(1,1,2 )	5.562	0.80 0	50	6.974	3.036	0.609	3.096	0.18 0	0.179	6
12	(2,0,2 )	5.939	0.85 4	783	7.448	12.636	1.197	12.692	0.16 8	0.168	8
6	(3,0,0 )	6.251	0.89 9	18	7.838	2.711	0.589	2.775	0.16 0	0.160	9
24	(2,1,2 )	6.950	0.99 9	51	8.715	2.419	0.579	2.488	0.14 4	0.143	9
6	(2,2,0 )	7.218	1.03 8	221	9.051	10.446	1.116	10.506	0.13 9	0.138	8
12	(3,1,0 )	7.512	1.08 0	20	9.420	2.210	0.564	2.281	0.13 3	0.133	1
12	(3,0,2 )	7.548	1.08 5	20	9.466	2.197	0.563	2.268	0.13 2	0.132	1
6	(4,0,0 )	8.334	1.19 8	139	10.45 1	8.905	1.043	8.966	0.12 0	0.120	1
12	(2,2,2 )	8.367	1.20 3	275	10.49 2	8.864	1.041	8.925	0.12 0	0.119	1
2	(0,0,4 )	8.464	1.21 7	44	10.61 4	8.743	1.034	8.804	0.11 8	0.118	1
24	(3,1,2 )	8.622	1.24 0	25	10.81 2	1.855	0.533	1.930	0.11 6	0.116	1
12	(1,0,4 )	8.716	1.25 3	12	10.93 0	1.828	0.531	1.904	0.11 5	0.114	1
12	(3,2,0 )	9.082	1.30 6	10	11.38 9	1.728	0.521	1.805	0.11 0	0.110	1
12	(1,1,4 )	9.201	1.32 3	10	11.53 8	1.697	0.517	1.774	0.10 9	0.108	1
12	(4,0,2 )	9.347	1.34 4	187	11.72 1	7.716	0.978	7.778	0.10 7	0.107	2
12	(2,0,4 )	9.434	1.35 6	181	11.83 0	7.622	0.973	7.684	0.10 6	0.106	2

12	(4,1,0 )	9.548	1.37 3	9	11.97 3	1.610	0.507	1.688	0.10 5	0.104 7	1 7
24	(3,2,2 )	10.02 0	1.44 0	14	12.56 5	1.500	0.494	1.580	0.10 0	0.099 8	1 7

**Table S7.** Summary of the stability testing results of our SOECs with the hybrid electrode

Operation	Fuel humidity	Current density	Temperature	Degradation rate
Discharging	3% humidified H <sub>2</sub>	0.8 A cm <sup>-2</sup>	650 °C	2 mV/h
	3% humidified H <sub>2</sub>	0.2 A cm <sup>-2</sup>	550 °C	1 mV/h
	3% humidified C <sub>3</sub> H <sub>8</sub>	0.4 A cm <sup>-2</sup>	600 °C	8 mV/h
Charging	40% humidified H <sub>2</sub>	0.8 A cm <sup>-2</sup>	650 °C	3 mV/h
	40% humidified H <sub>2</sub>	3.0 A cm <sup>-2</sup>	650 °C	8 mV/h
	40% humidified H <sub>2</sub>	5.0 A cm <sup>-2</sup>	750 °C	4 mV/min

## References

- 1 H. Shimada, T. Yamaguchi, H. Kishimoto, H. Sumi, Y. Yamaguchi, K. Nomura and Y. Fujishiro, *Nat Commun*, 2019, **10**, 5432.
- 2 S.-L. Zhang, H. Wang, M. Y. Lu, A.-P. Zhang, L. V. Mogni, Q. Liu, C.-X. Li, C.-J. Li and S. A. Barnett, *Energy Environ. Sci.*, 2018, **11**, 1870–1879.
- 3 Z. Teng, Z. Xiao, G. Yang, L. Guo, X. Yang, R. Ran, W. Wang, W. Zhou and Z. Shao, *Materials Today Energy*, 2020, **17**, 100458.
- 4 T. Chen, Y. Zhou, M. Liu, C. Yuan, X. Ye, Z. Zhan and S. Wang, *Electrochemistry Communications*, 2015, **54**, 23–27.
- 5 H. Yu, H. Im and K. T. Lee, *Adv Funct Materials*, 2022, 2207725.
- 6 B.-H. Yun, K. J. Kim, D. W. Joh, M. S. Chae, J. J. Lee, D. Kim, S. Kang, D. Choi, S.-T. Hong and K. T. Lee, *J. Mater. Chem. A*, 2019, **7**, 20558–20566.
- 7 J. H. Park, C. H. Jung, K. J. Kim, D. Kim, H. R. Shin, J.-E. Hong and K. T. Lee, *ACS Appl. Mater. Interfaces*, 2021, **13**, 2496–2506.
- 8 J. Kim, S. Im, S. H. Oh, J. Y. Lee, K. J. Yoon, J.-W. Son, S. Yang, B.-K. Kim, J.-H. Lee, H.-W. Lee, J.-H. Lee and H.-I. Ji, *Sci. Adv.*, 2021, **7**, eabj8590.
- 9 J. Yan, Z. Zhao, L. Shang, D. Ou and M. Cheng, *Journal of Power Sources*, 2016, **319**, 124–130.
- 10 B.-K. Park, R. Scipioni, Q. Zhang, D. Cox, P. W. Voorhees and S. A. Barnett, *J. Mater. Chem. A*, 2020, **8**, 11687–11694.
- 11 G. Li, Y. Gou, X. Cheng, Z. Bai, R. Ren, C. Xu, J. Qiao, W. Sun, Z. Wang and K. Sun, *ACS Appl. Mater. Interfaces*, 2021, **13**, 34282–34291.
- 12 K. Joong Yoon, M. Biswas, H.-J. Kim, M. Park, J. Hong, H. Kim, J.-W. Son, J.-H. Lee, B.-K. Kim and H.-W. Lee, *Nano Energy*, 2017, **36**, 9–20.
- 13 Z. Li, B. Yang, B. Qian, S. Wang, Y. Zheng, L. Ge and H. Chen, *Separation and Purification Technology*, 2023, **308**, 123002.
- 14 K. T. Bae, I. Jeong, D. Kim, H. Yu, H.-N. Im, A. Akromjon, C.-W. Lee and K. T. Lee, *Chemical Engineering Journal*, 2023, **461**, 142051.
- 15 T. Liu, Y. Wang, Y. Zhang, S. Fang, L. Lei, C. Ren and F. Chen, *Electrochemistry Communications*, 2015, **61**, 106–109.
- 16 D. Kim, J. W. Park, M. S. Chae, I. Jeong, J. H. Park, K. J. Kim, J. J. Lee, C. Jung, C.-W. Lee, S.-T. Hong and K. T. Lee, *J. Mater. Chem. A*, 2021, **9**, 5507–5521.
- 17 Y. Tan, A. Wang, L. Jia, D. Yan, B. Chi, J. Pu and J. Li, *International Journal of Hydrogen Energy*, 2017, **42**, 4456–4464.
- 18 D. Heidari, S. Javadpour and S. H. Chan, *Energy Conversion and Management*, 2017, **136**, 78–84.
- 19 M. A. Laguna-Bercero, H. Monzón, A. Larrea and V. M. Orera, *J. Mater. Chem. A*, 2016, **4**, 1446–1453.
- 20 M.-B. Choi, B. Singh, E. D. Wachsman and S.-J. Song, *Journal of Power Sources*, 2013, **239**, 361–373.
- 21 H.-J. Choi, Y.-H. Na, M. Kwak, T. W. Kim, D.-W. Seo, S.-K. Woo and S.-D. Kim, *Ceramics International*, 2017, **43**, 13653–13660.
- 22 K. J. Kim, I. Thaheem, I. Jeong, H. Yu, J. H. Park and K. T. Lee, *Journal of Power Sources*, 2022, **539**, 231611.
- 23 Y. Tian, W. Wang, Y. Liu, L. Zhang, L. Jia, J. Yang, B. Chi, J. Pu and J. Li, *ACS Appl. Energy Mater.*, 2019, **2**, 3297–3305.

- 24 Z. Yue, L. Jiang, Z. Chen, N. Ai, Y. Zou, S. P. Jiang, C. Guan, X. Wang, Y. Shao, H. Fang, Y. Luo and K. Chen, *ACS Appl. Mater. Interfaces*, 2023, **15**, 8138–8148.
- 25 Y. Niu, Y. Zhou, W. Lv, Y. Chen, Y. Zhang, W. Zhang, Z. Luo, N. Kane, Y. Ding, L. Soule, Y. Liu, W. He and M. Liu, *Adv. Funct. Mater.*, 2021, **31**, 2100034.
- 26 B.-K. Park and S. A. Barnett, *J. Mater. Chem. A*, 2020, **8**, 11626–11631.
- 27 X. Tong, Y. Xu, Đ. Tripković, P. V. Hendriksen, W.-R. Kiebach and M. Chen, *J. Mater. Chem. A*, 2020, **8**, 9039–9048.
- 28 Z. Chen, L. Jiang, S. He, C. Guan, Y. Zou, Z. Yue, N. Ai, S. P. Jiang, Y. Shao and K. Chen, *Applied Catalysis B: Environmental*, 2022, **305**, 121056.
- 29 K. Chen, N. Li, N. Ai, M. Li, Y. Cheng, W. D. A. Rickard, J. Li and S. P. Jiang, *J. Mater. Chem. A*, 2016, **4**, 17678–17685.
- 30 H. Liu, X. Zhu, M. Cheng, Y. Cong and W. Yang, *Chem. Commun.*, 2011, **47**, 2378–2380.
- 31 L. Zhang, D. Li and S. Zhang, *Ceramics International*, 2017, **43**, 2859–2863.
- 32 Y. Zhu, W. Zhou, Y. Chen and Z. Shao, *Angew. Chem. Int. Ed.*, 2016, **55**, 8988–8993.
- 33 Y. Liu, Y. Tian, W. Wang, Y. Li, S. Chattopadhyay, B. Chi and J. Pu, *ACS Appl. Mater. Interfaces*, 2020, **12**, 57941–57949.
- 34 Y. Tian, W. Wang, Y. Liu, A. Naden, M. Xu, S. Wu, B. Chi, J. Pu and J. T. S. Irvine, *ACS Catal.*, 2021, **11**, 3704–3714.
- 35 X. Xi, Z.-S. Cao, X.-Q. Shen, Y. Lu, J. Li, J.-L. Luo and X.-Z. Fu, *Journal of Power Sources*, 2020, **459**, 228071.
- 36 K.-Y. Lai and A. Manthiram, *Chem. Mater.*, 2018, **30**, 2515–2525.
- 37 O. Kwon, S. Sengodan, K. Kim, G. Kim, H. Y. Jeong, J. Shin, Y.-W. Ju, J. W. Han and G. Kim, *Nat Commun*, 2017, **8**, 15967.
- 38 C. Yang, J. Li, Y. Lin, J. Liu, F. Chen and M. Liu, *Nano Energy*, 2015, **11**, 704–710.
- 39 S. Sengodan, S. Choi, A. Jun, T. H. Shin, Y.-W. Ju, H. Y. Jeong, J. Shin, J. T. S. Irvine and G. Kim, *Nature Mater*, 2015, **14**, 205–209.
- 40 Z. Wang, Z. Wang, W. Yang, R. Peng and Y. Lu, *Journal of Power Sources*, 2014, **255**, 404–409.
- 41 L. Yang, Y. Choi, W. Qin, H. Chen, K. Blinn, M. Liu, P. Liu, J. Bai, T. A. Tyson and M. Liu, *Nat Commun*, 2011, **2**, 357.
- 42 Z. Zhan and S. Barnett, *Solid State Ionics*, 2005, **176**, 871–879.
- 43 L. Zhang, C. Yang, A. I. Frenkel, S. Wang, G. Xiao, K. Brinkman and F. Chen, *Journal of Power Sources*, 2014, **262**, 421–428.
- 44 M. Qin, T. Tan, K. Li, Z. Wang, H. Yang, Z. Liu, M. Zhou, T. Liu, C. Yang and M. Liu, *International Journal of Hydrogen Energy*, 2020, **45**, 21464–21472.