

Supporting Information

Core-shell BaPP@BDO-Rh benzimidazole oligomer photocatalyst can effectively promote NADH regeneration and CO₂ conversion to fumarate

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1. Supplementary figures

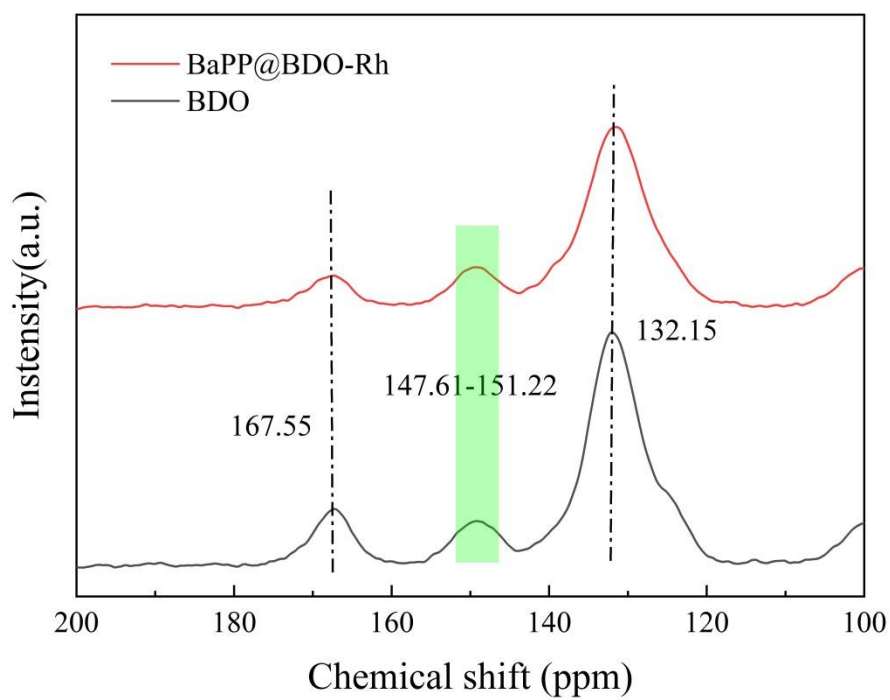
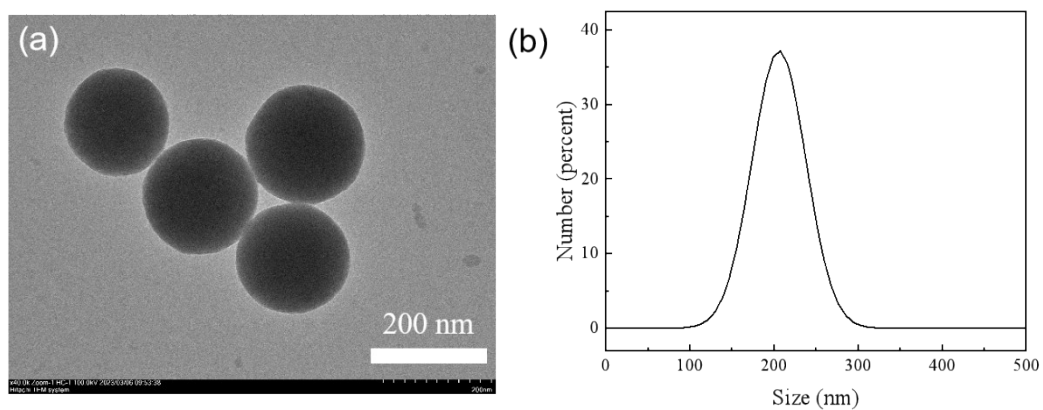


Fig. S1. The solid state ^{13}C NMR spectra of BDO and BaPP@BDO-Rh.



37

38 **Fig. S2.** (a) TEM images of nano-BaSO₄, (b) Particle size distribution of
 39 nano-BaSO₄. Particle size is obtained by testing the light scattering
 40 particle size instrument (Marvin, ZS-90).

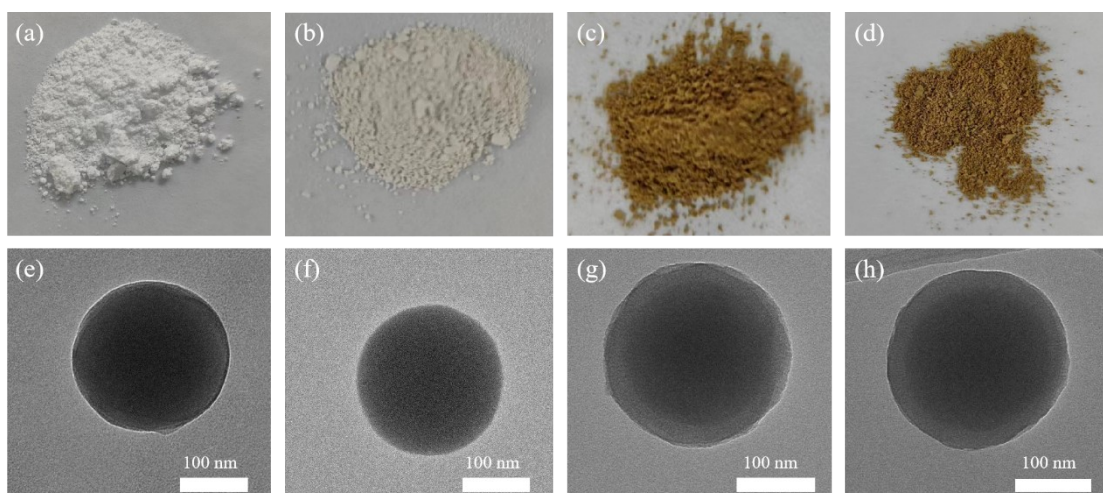
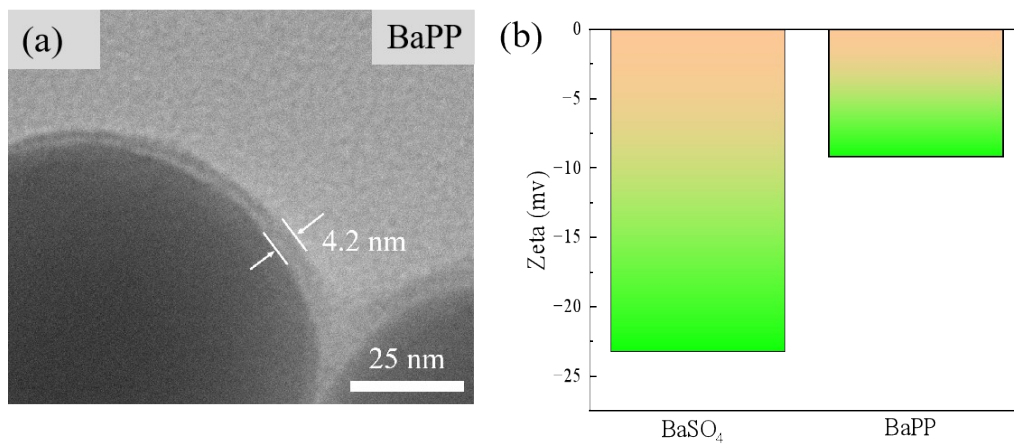


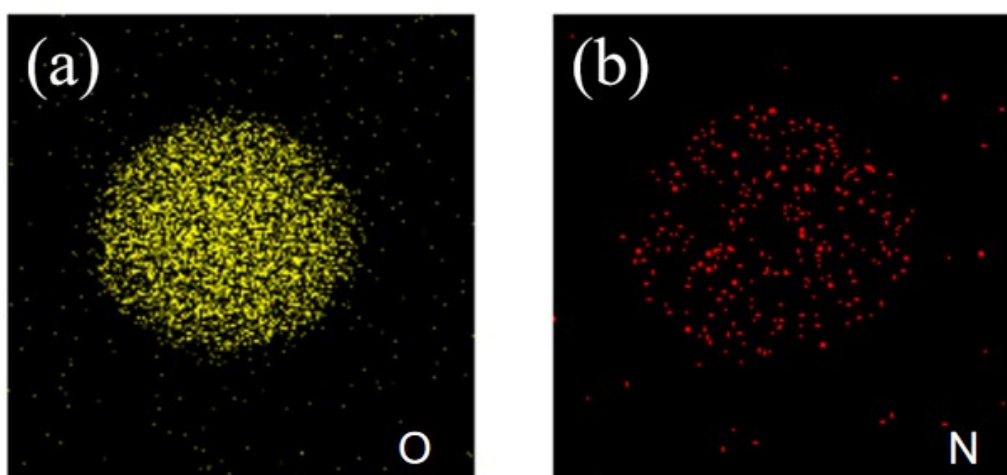
Fig. S3. (a-d) Optical images of (a) BaSO_4 , (b) BaPP, (c) BaPP@BDO and (d) BaPP@BDO-Rh. (e-h) TEM images of (e) BaSO_4 , (f) BaPP, (g) BaPP@BDO and (h) BaPP@BDO-Rh. BaPP@BDO and BaPP@BDO-Rh appear as bright yellow hydrogels in aqueous solution and as earthy-yellow powders when dried.



49

50 **Fig. S4.** (a) High-resolution TEM image of BaPP. (b) Zeta potential of

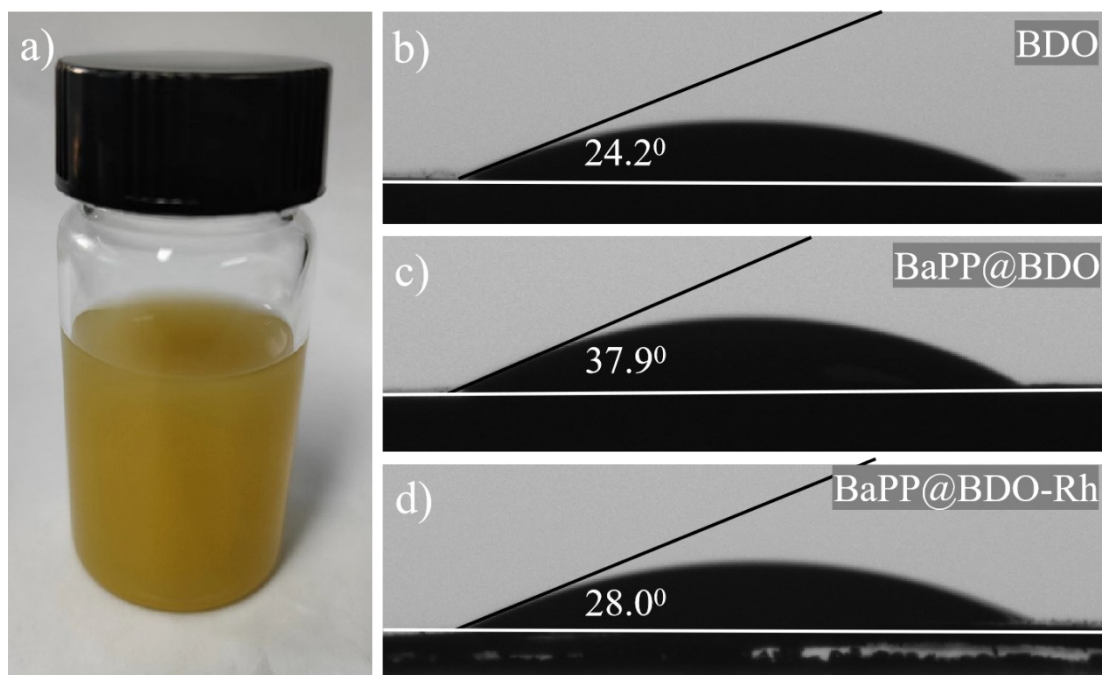
51 BaSO₄ and BaPP.



52

53 **Fig. S5.** (a) O and (b) N element mappings of BaPP@BDO-Rh.

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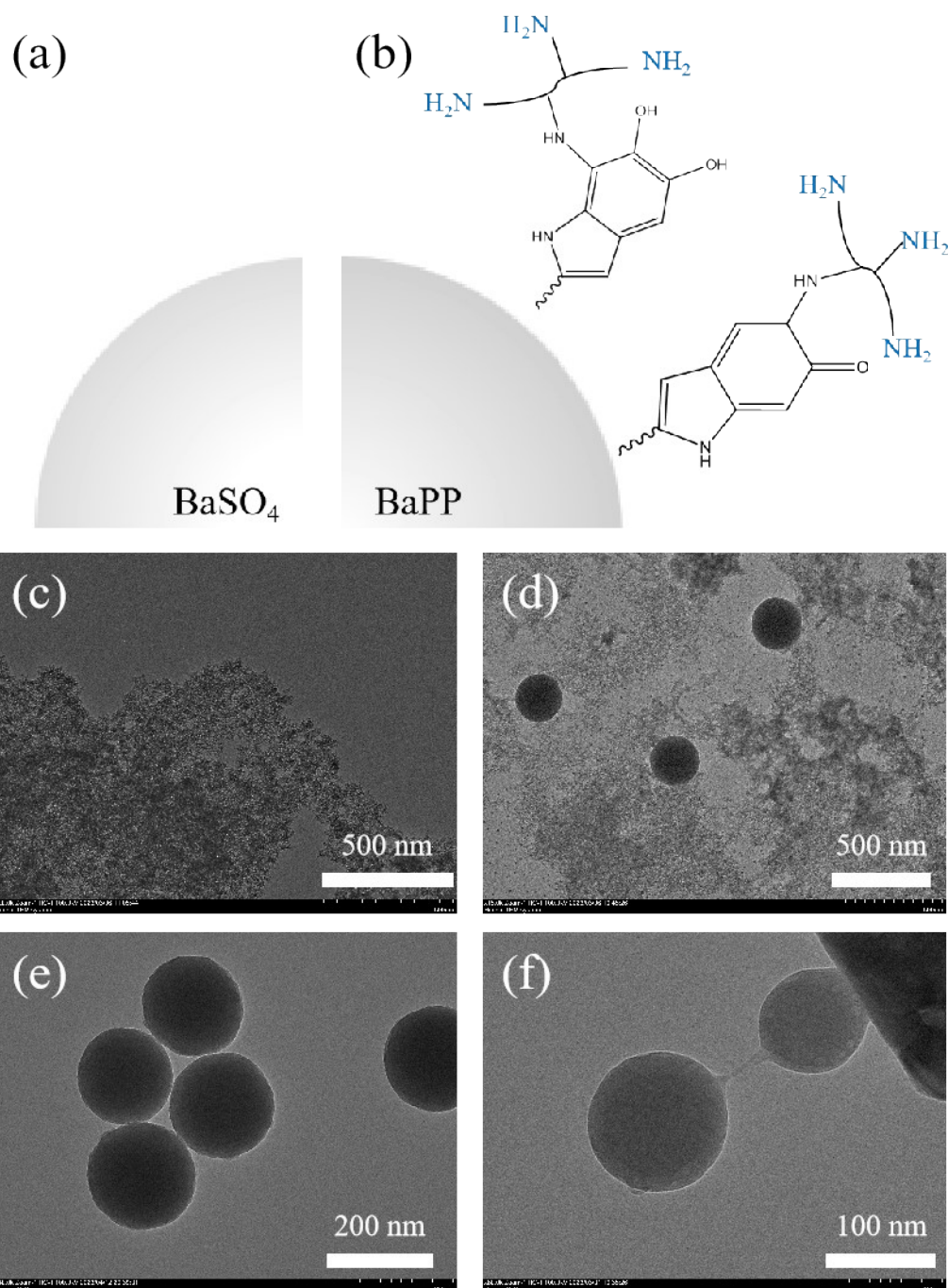
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56 **Fig. S6.** a) Macroscopic image of BaPP@BDO-Rh in aqueous solution.

57 Water contact angle test of b) BDO, c) BaPP@BDO and d) BaPP@BDO-

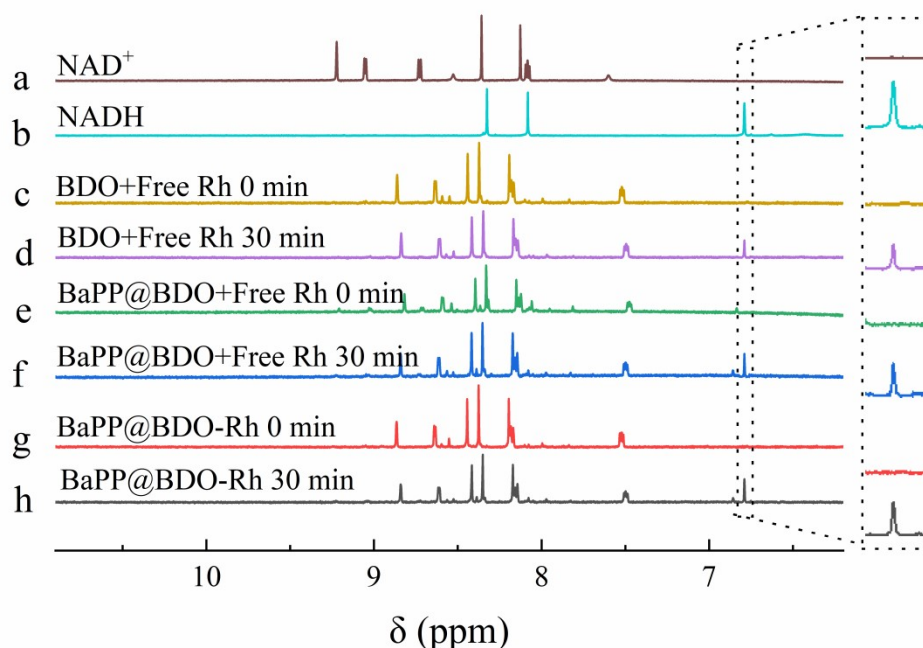
58 Rh.

59



60
 61 **Fig. S7.** The possible surface molecular structures of (a) BaSO_4 and (b)
 62 BaPP . (c-e) TEM images of (c) BDO, (d) BaSO_4 / BDO and (e,f)
 63 BaPP @BDO. By comparison, it is found that BaSO_4 without PEI/PDA
 64 modification is separated from BDO. Only PEI/PDA modified BaSO_4 and
 65 BDO can form core-shell structure.

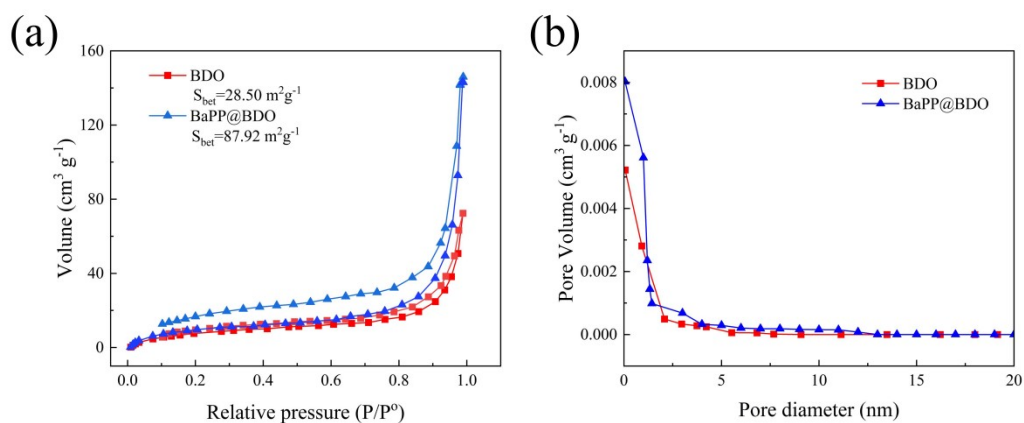
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68 **Fig. S8.** ^1H NMR spectra of NAD^+ (a) in D_2O , NADH (b) in D_2O , NADH
 69 regeneration system containing BDO+Free Rh (c,d), BaPP@BDO+Free
 70 Rh (e,f) and BaPP@BDO-Rh (g,h) before and after irradiated at xenon
 71 lamp. These peaks belonged to the hydrogen atoms on the pyridine ring
 72 after NAD^+ was added to the reaction system. This phenomenon is
 73 thought to be caused by solvation effect.¹ Interestingly, no shift was
 74 observed in the characteristic peaks of NADH. By comparing the lines c
 75 and d, e and f, and g and h, the characteristic peak at 6.79 ppm validated
 76 the production of 1,4-NADH after illumination. Due to the limitation of
 77 detection signal, no signal of side-product was detected.

78

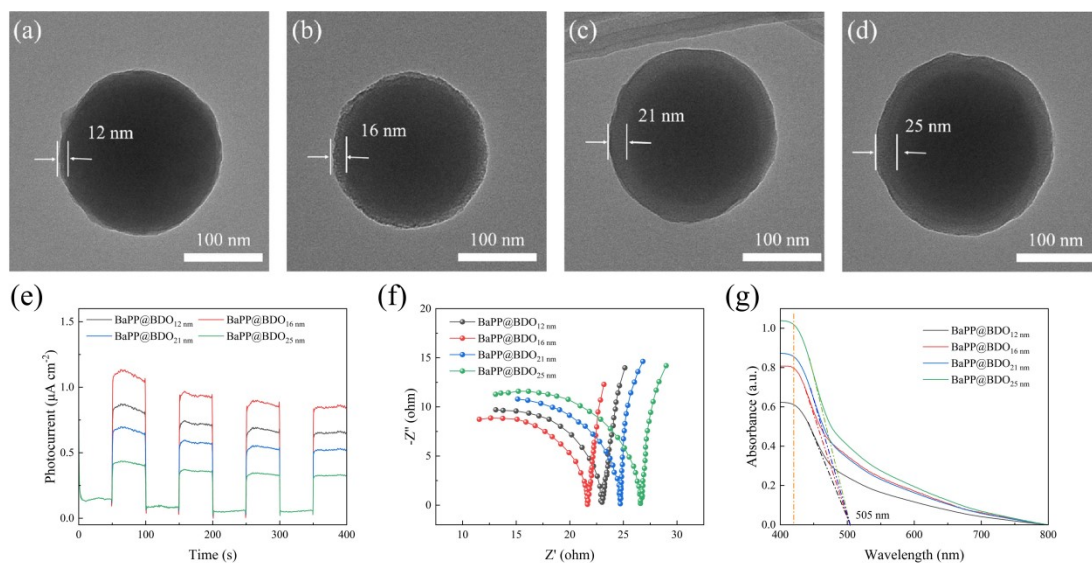


79

80 **Fig. S9.** (a) N₂ adsorption-desorption isotherms and (b) pore size

81 distribution curves of BDO and BaPP@BDO.

82



83

84 **Fig. S10.** (a-d) TEM images of (a) BaPP@BDO-Rh_{12 nm}, (b)

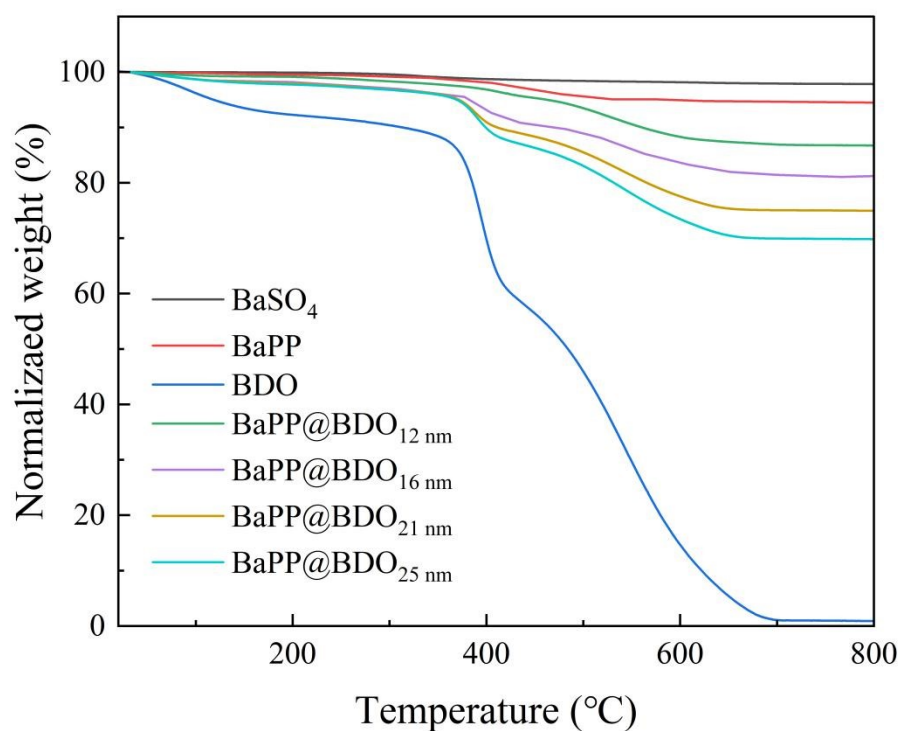
85 BaPP@BDO-Rh_{16 nm}, (c) BaPP@ BDO-Rh_{21 nm} and (d) BaPP@BDO-Rh₂₅

86 nm. (e) Photocurrent responses , (f) EIS Nyquist plots and UV-vis diffuse

87 reflectance spectra of BaPP@BDO-Rh_{12 nm}, BaPP@BDO-Rh_{16 nm},

88 BaPP@BDO-Rh_{21 nm} and BaPP@BDO-Rh_{25 nm}.

89



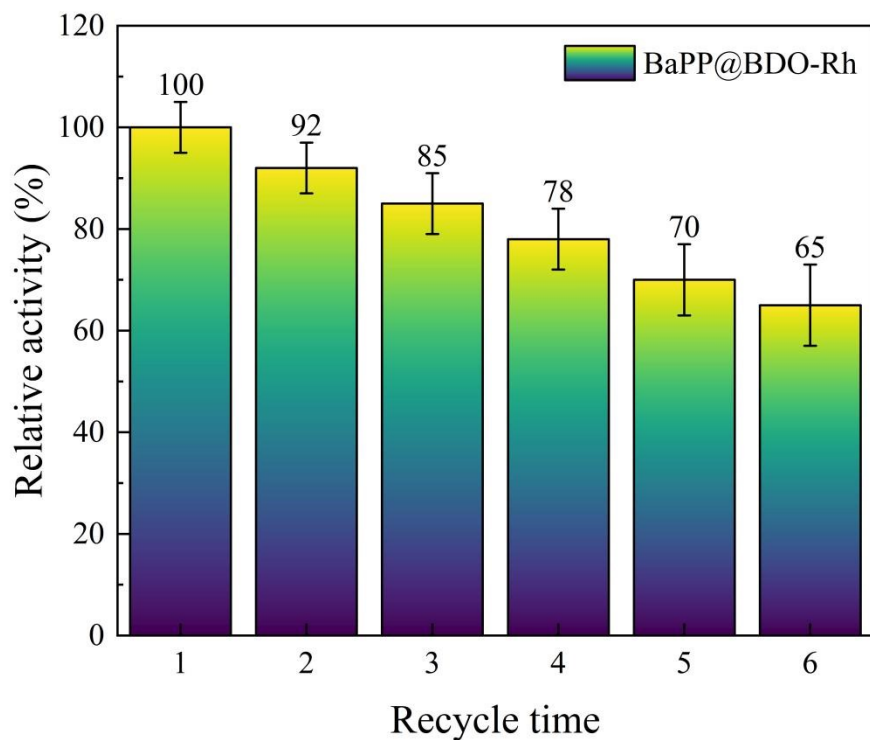
90

91 **Fig. S11.** Thermogravimetric analysis (TGA) profile of BaSO_4 , BaPP ,

92 BDO , $\text{BaPP@BDO}_{12 \text{ nm}}$, $\text{BaPP@BDO}_{16 \text{ nm}}$, $\text{BaPP@BDO}_{21 \text{ nm}}$ and

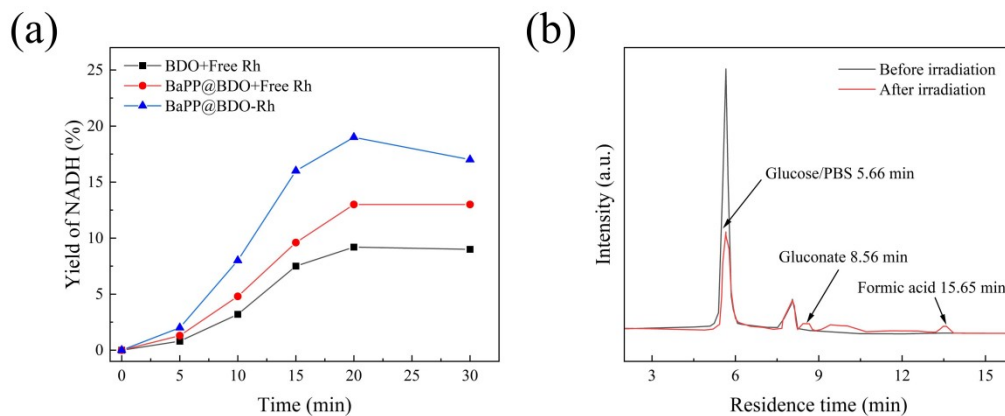
93 $\text{BaPP@BDO}_{25 \text{ nm}}$.

94



95

96 **Fig. S12.** Recycling stability of BaPP@BDO-Rh. The recycle stability of
97 the catalytic system was evaluated by recovering the photocatalysts from
98 the catalytic system by centrifugation (8000 rpm), and then, after washing
99 three times, redispersing them into a new reaction solution for the next
100 cycle.



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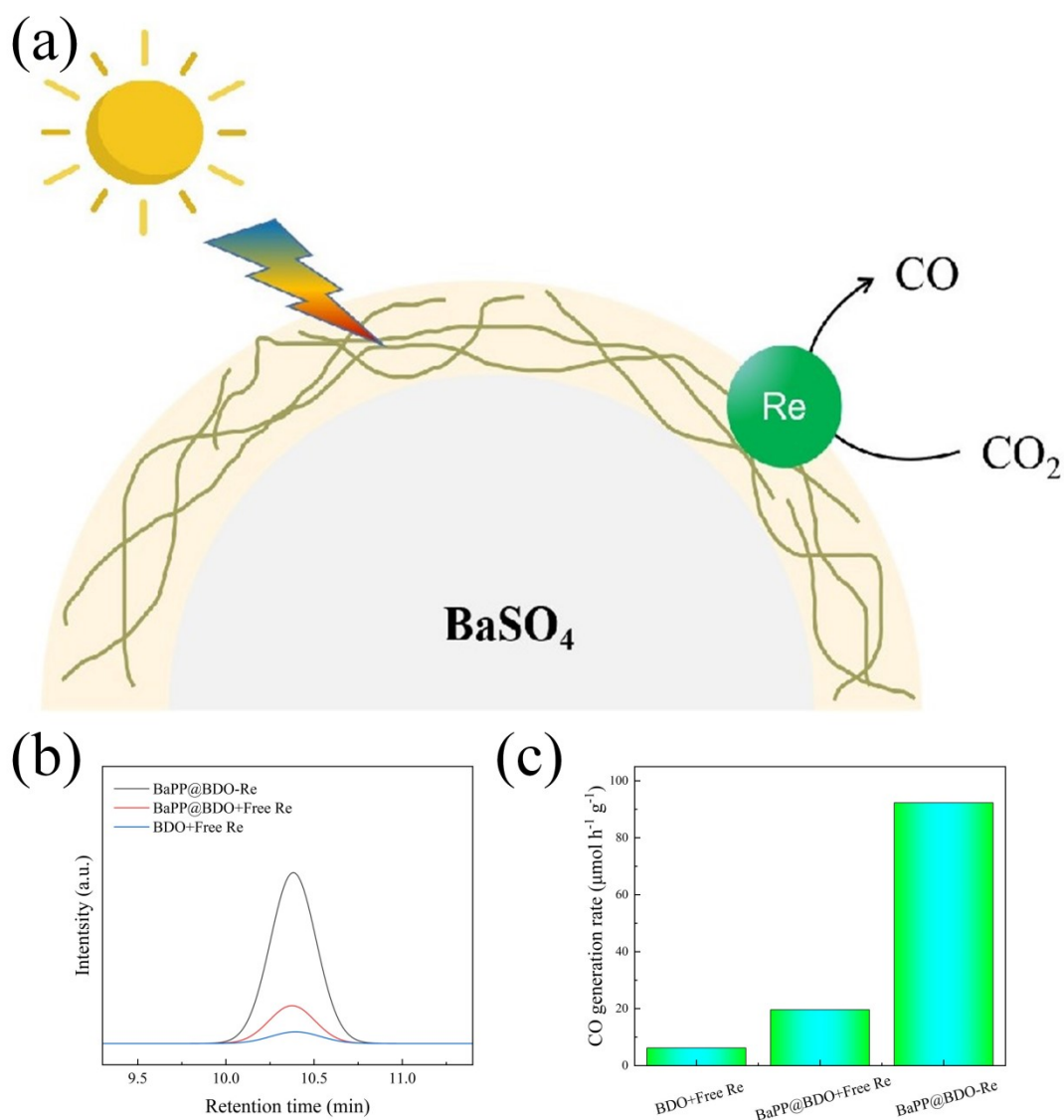
102 **Fig. S13.** (a) Yield of NADH as a function of reaction time, (b) HPLC

103 analysis of the reaction solution before/after the reaction. Reaction

104 conditions: [photocatalyst] = 1 g L⁻¹; [Glucose/PBS] = 1000/100 mM ,

105 pH = 8.5; [NAD⁺] = 1 mM; [free Rh] = 0.0188 mM; volume = 3 mL.

106



107

108 **Fig. S14.** Scheme of (a) BaPP@BDO-Re for photocatalytic CO₂

109 reduction. (b) Photocatalytic CO₂ reduction by BDO with free Re,

110 BaPP@BDO with free Re and BaPP@BDO-Re under 420 nm excitation

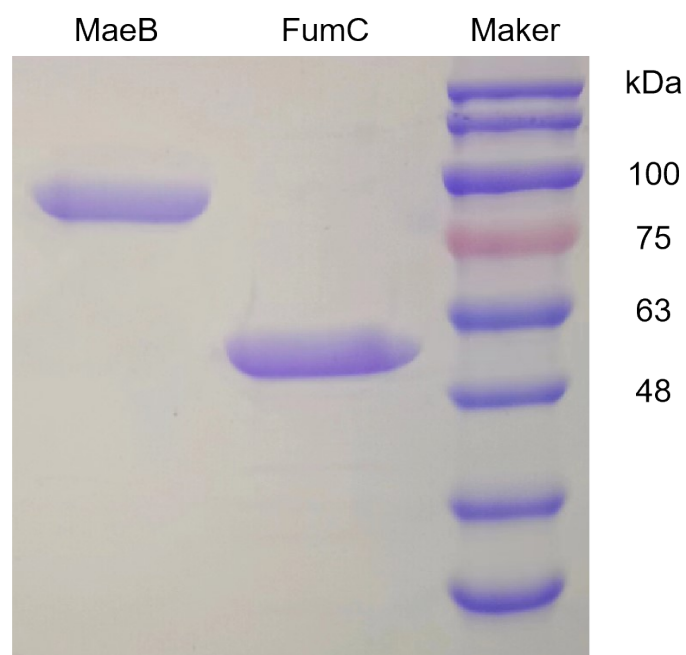
111 detected by gas chromatography and (c) calculated CO evolution rate.

112 Reaction conditions: photocatalyst (0.6 mg) was dispersed in 6 mL of

113 CH₃CN, and 0.4 mL of TEOA in 20 mL septum-sealed glass vials.

114

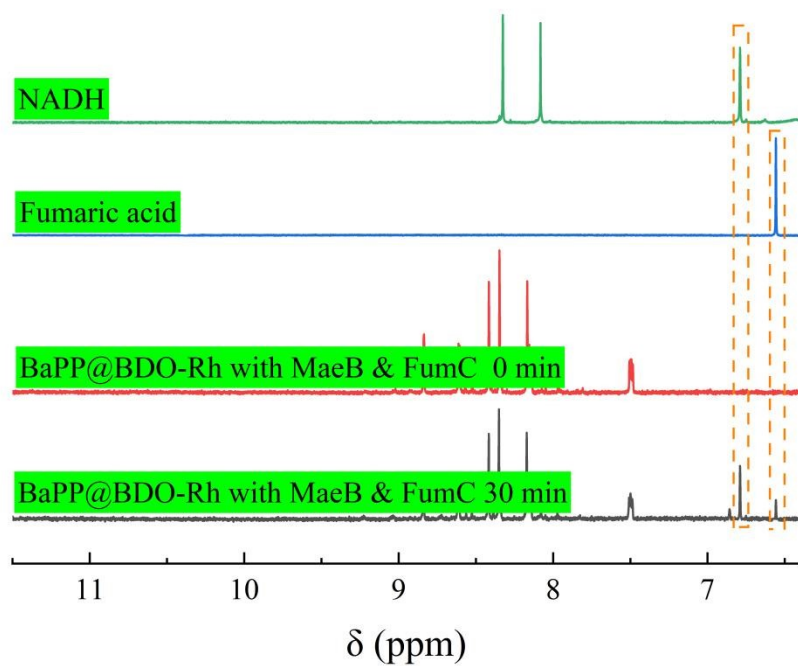
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117 **Fig. S15.** SDS-PAGE analysis of the recombinant enzymes. Malate
118 dehydrogenase (MaeB);Fumarate hydratase (FumC).

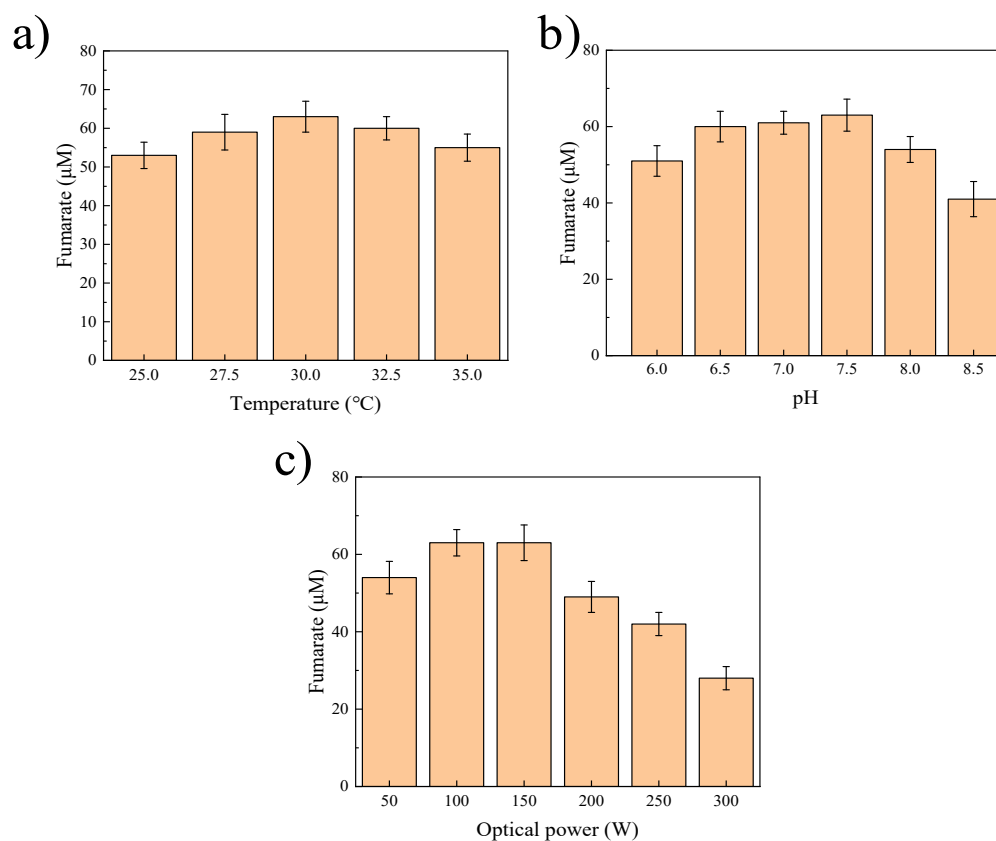
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121 **Fig. S16.** ^1H NMR spectra of NADH and fumarate in a photoenzyme

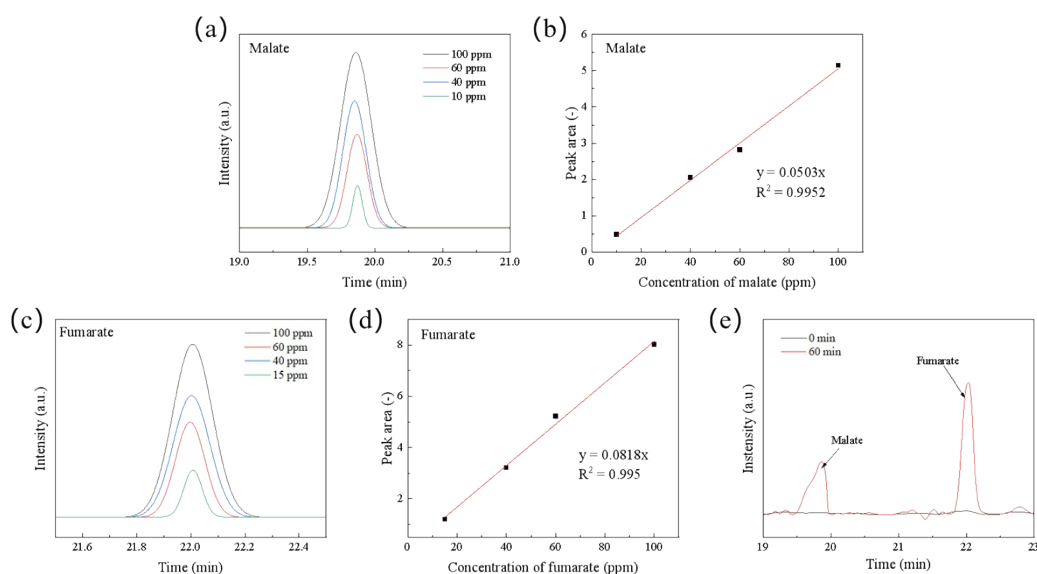
122 catalyzed system.



123

124 **Fig. S17.** Effect of temperature (a), pH (b) and optical power (c) on

125 photoenzyme-catalyzed synthesis of fumarate.



126

127 **Fig. S18.** (a) Ion chromatogram of standard malate at different
 128 concentrations. (b) Relationship between the malate concentration and the
 129 detection peak area. (c) Ion chromatogram of standard fumarate at
 130 different concentrations. (d) Relationship between the fumarate
 131 concentration and the detection peak area. (e) Ion chromatograms of
 132 photoenzyme reaction for 0 min and 60 min.

133

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135

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137 **2. Supplementary tables**

138 **Table S1.** Parameters of time-resolved transient PL of BaPP@BDO and
139 BaPP@BDO-Rh. After calculating² the parameters in Table S1, the
140 fluorescence lifetimes of BaPP@BDO and BaPP@BDO-Rh
141 can be calculated as 1.54 ns and 0.86 ns.

Photocatalyst	τ_1	τ_2	B_1	B_2
BaPP@BDO-Rh	0.4244	1.9088	0.054	0.005
BaPP@BDO	0.4265	1.9008	0.0057	0.004

142 **Table S2** The photocatalytic regeneration performance of NADH by
 143 different photocatalysts.

Photocatalyst	Mediator	Reaction	Yield (%)	TOF (h ⁻¹)
		equilibrium time (min)		
BaPP@BDO-Rh (Our study)	Rh	8	43.5	131.16
PCN@TA/PEI-Rh ³	Rh	20	37.8	70.82
SiPP@CPNL-Rh ²	Rh	28	39.6	44.8
GCN@M/TiO ₂ ⁴	Rh	20	58	42.67
Co1/C ₃ N ₄ ⁵	Rh	10	98	33.01
Rh-NU-1006 ⁶	Rh	120	28	20.69
TM@HOF-101 ⁷	Rh ^[a]	40	74.5	6.36
DBTS-CMP ₁ ⁸	Rh	45	84	3.75
ACN ⁹	Rh	60	62.3	3.36

ATCN-DSCN 10	Rh	15	74	2.95
TCPP/SiO ₂ /Rh HNPs ¹¹	Rh	180	75	1.67
CTF ¹²	Rh	120	75.9	0.76
AM/M/BP HNSs ¹³	Rh	180	89	0.5

144 [a] Rh is [Cp*Rh(bpy)H₂O]²

Table S3. Reactions catalyzed by enzymes and BaPP@BDO-Rh in this system.

Enzyme	Reaction
Malate dehydrogenase (MaeB)	$\text{NADH} + \text{Pyruvate} + \text{CO}_2 \rightarrow \text{Malate} + \text{NAD}^+ + \text{H}^+ + 2\text{e}^-$
Fumarate hydratase (FumC)	$\text{Malate} \rightarrow \text{Fumarate} + \text{H}_2\text{O}$
BaPP@BDO-Rh	$\text{NAD}^+ + \text{H}^+ + 2\text{e}^- \rightarrow \text{NADH}$
Total reaction	$\text{Pyruvate} + \text{CO}_2 \rightarrow \text{Malate} + \text{H}_2\text{O}$

Table S4. Information about enzymes used in this study.

Enzyme	EC number	Source	Sp. Act (U mg ⁻¹)
Malate dehydrogenase (MaeB)	1.1.1.40	<i>E.coli</i>	1.76
Fumarate hydratase (FumC)	4.2.1.2	<i>R. capsulata</i>	6.29

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