

## Electronic Supplementary Information

# Unraveling the Modification Effect at NiO<sub>x</sub>/perovskite interfaces for Efficient and Stable Inverted Perovskite Solar Cells

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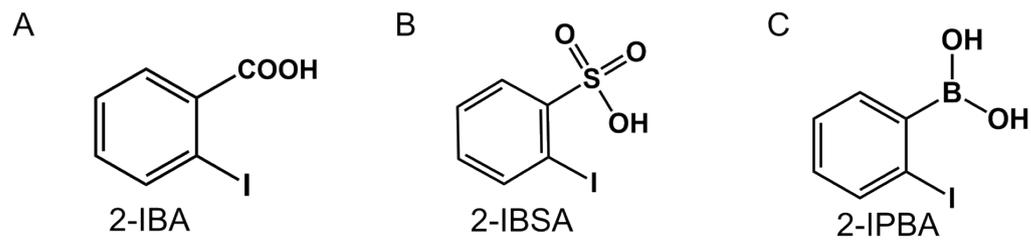


Figure S1. Chemical structures of (A) 2-IBA, (B) 2-IBSA and (C) 2-IPBA.

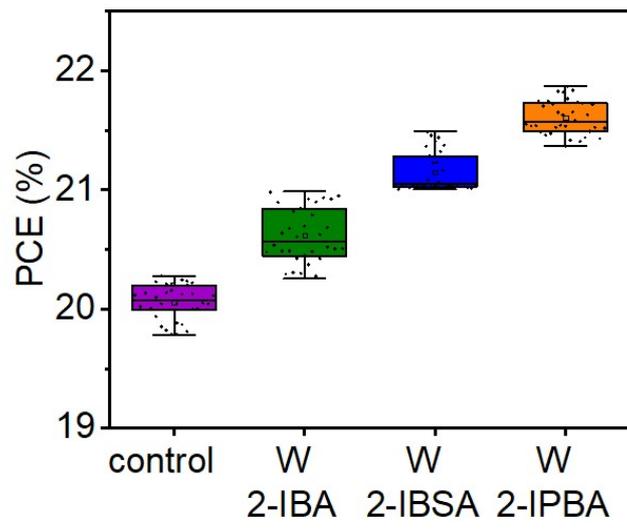


Figure S2. PCE statistics obtained from 35 individual control, 2-IBA, 2-IBSA and 2-IPBA modified PSCs.

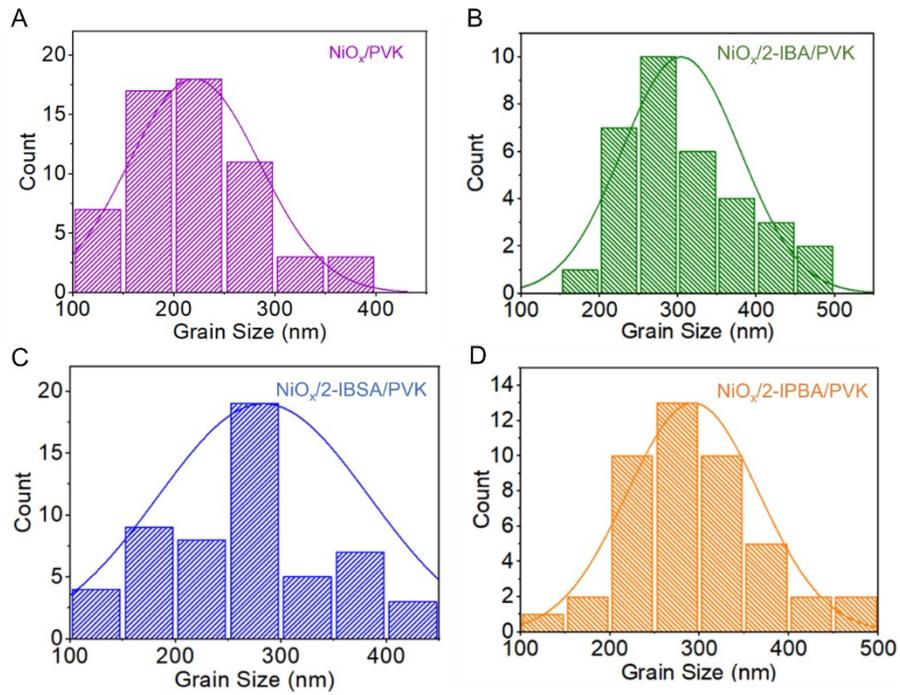


Figure S3. Size distributions of the perovskite grains deposited on (A)  $\text{NiO}_x$ , (B)  $\text{NiO}_x/2\text{-IBA}$ , (C)  $\text{NiO}_x/2\text{-IBSA}$  and (D)  $\text{NiO}_x/2\text{-IPBA}$ , respectively.

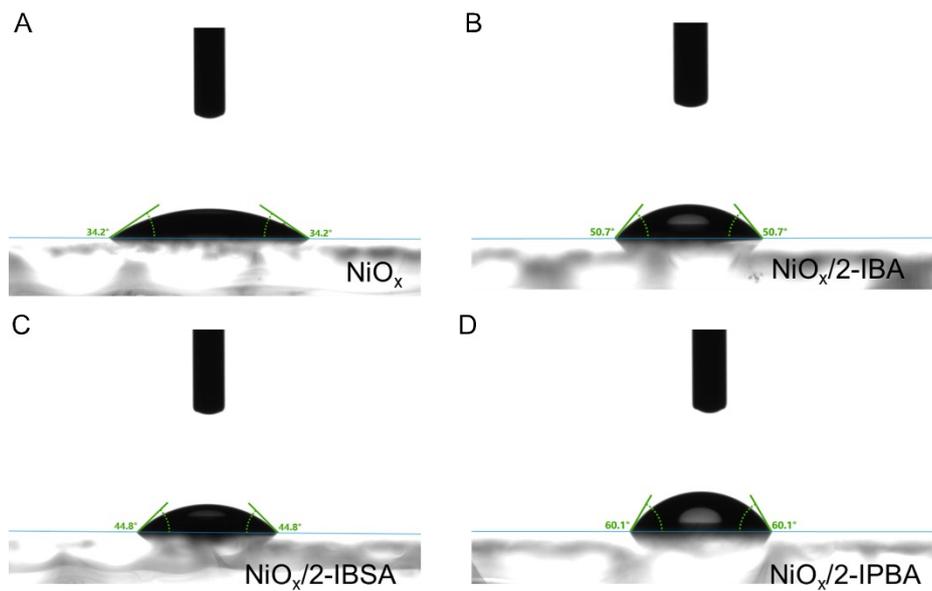


Figure S4. Contact angle testing for (A) the  $\text{NiO}_x$ , (B)  $\text{NiO}_x/2\text{-IBA}$ , (C)  $\text{NiO}_x/2\text{-IBSA}$  and (D)  $\text{NiO}_x/2\text{-IPBA}$  substrates.

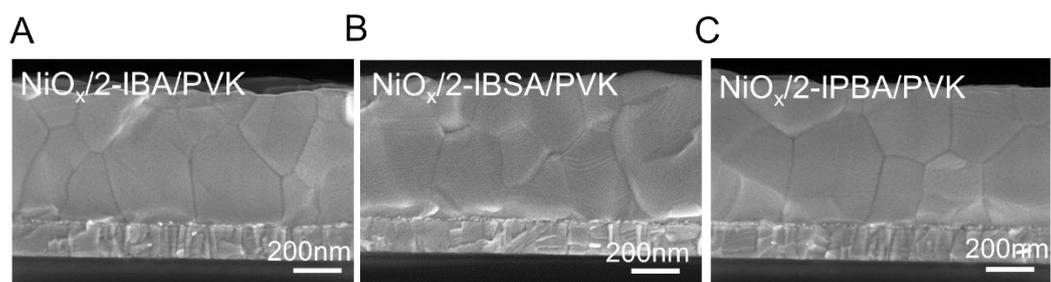


Figure S5. Cross-sectional SEM images of the perovskite films deposited on the (A) NiO<sub>x</sub>/2-IBA, (B) NiO<sub>x</sub>/2-IBSA and (C) NiO<sub>x</sub>/2-IPBA substrate.

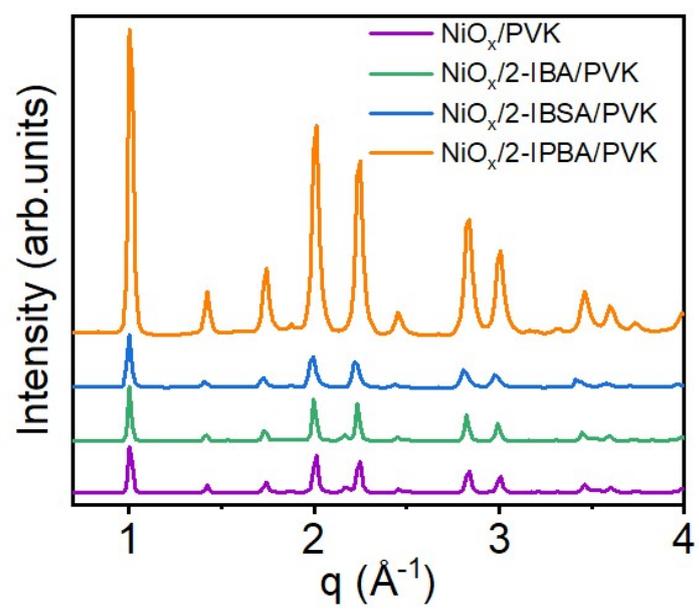


Figure S6. The sector-averaged integrals obtained from the 2D GIWAXS data for the perovskite films deposited on the NiO<sub>x</sub>, NiO<sub>x</sub>/2-IBA, NiO<sub>x</sub>/2-IBSA and NiO<sub>x</sub>/2-IPBA substrates.

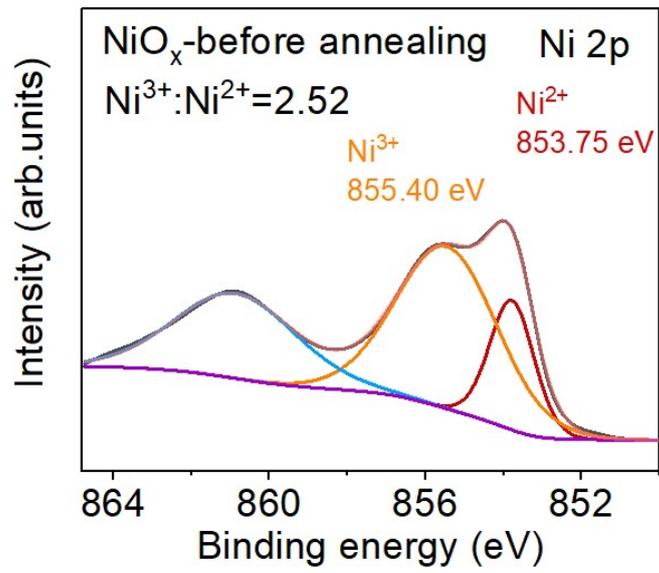


Figure S7. XPS spectra of Ni 2p for the NiO<sub>x</sub> film before thermal annealing.

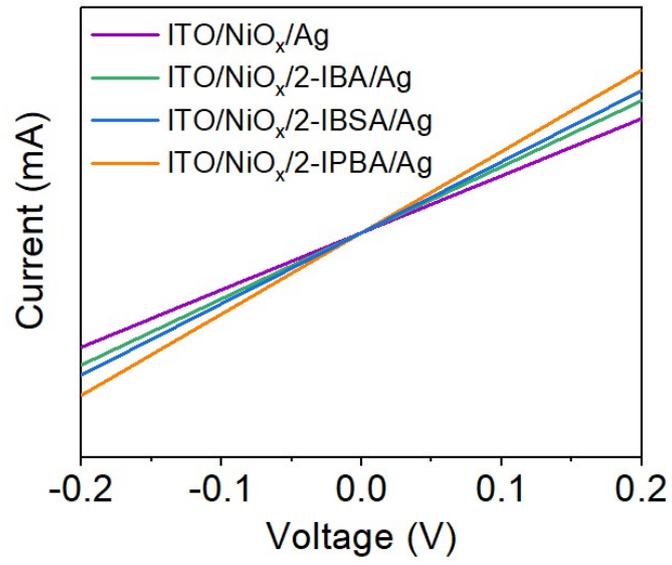


Figure S8. I-V curves of the NiO<sub>x</sub>, NiO<sub>x</sub>/2-IBA, NiO<sub>x</sub>/2-IBSA and NiO<sub>x</sub>/2-IPBA films, respectively.

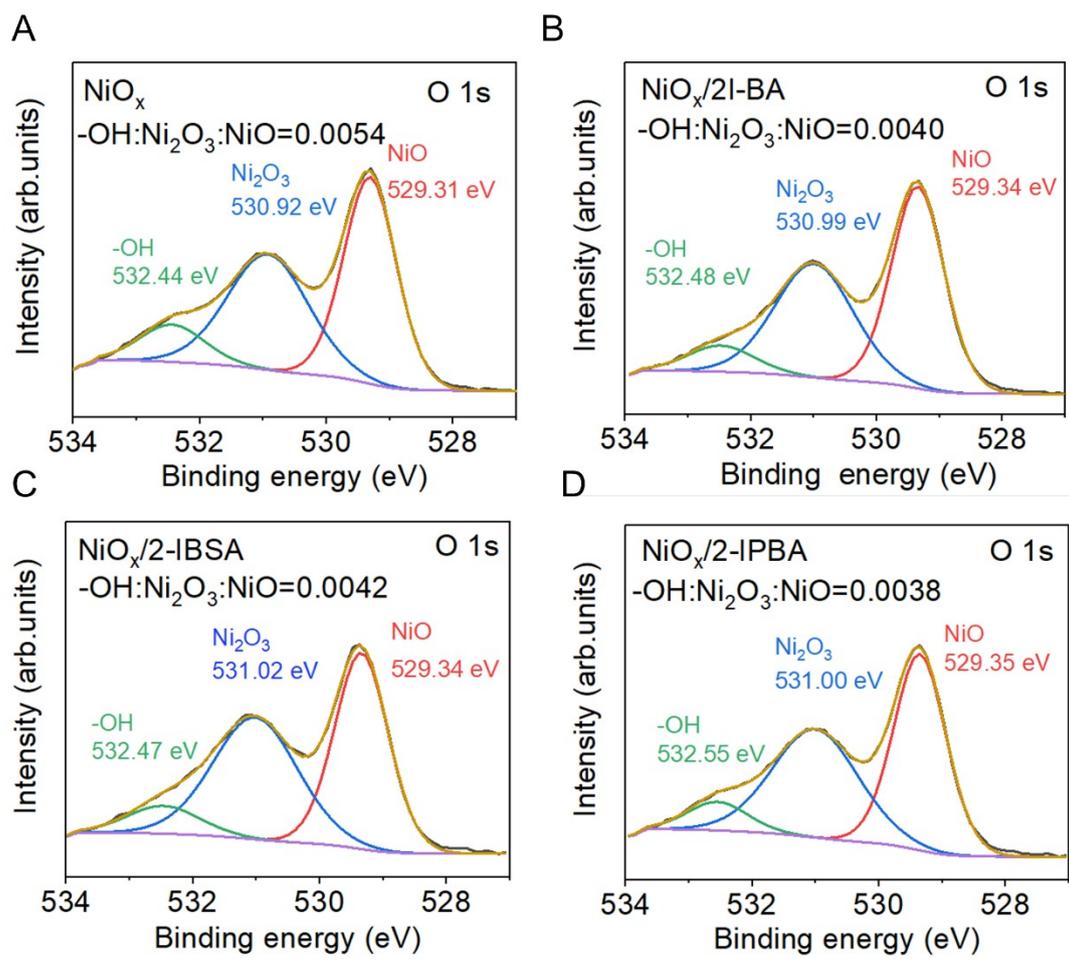


Figure S9. XPS spectra of O 1s for (A) the NiO<sub>x</sub>, (B) NiO<sub>x</sub>/2-IBA, (C) NiO<sub>x</sub>/2-IBSA and (D) NiO<sub>x</sub>/2-IPBA films.

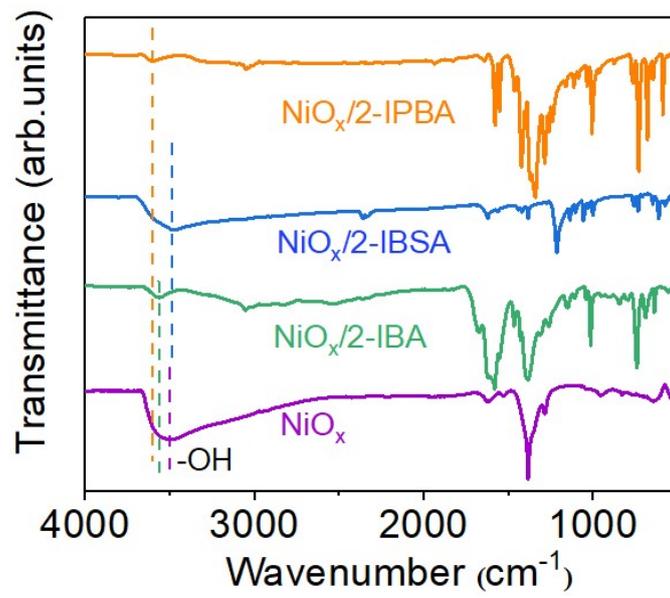


Figure S10. FT-IR spectra of the NiO<sub>x</sub>, NiO<sub>x</sub>/2-IBA, NiO<sub>x</sub>/2-IBSA and NiO<sub>x</sub>/2-IPBA films.

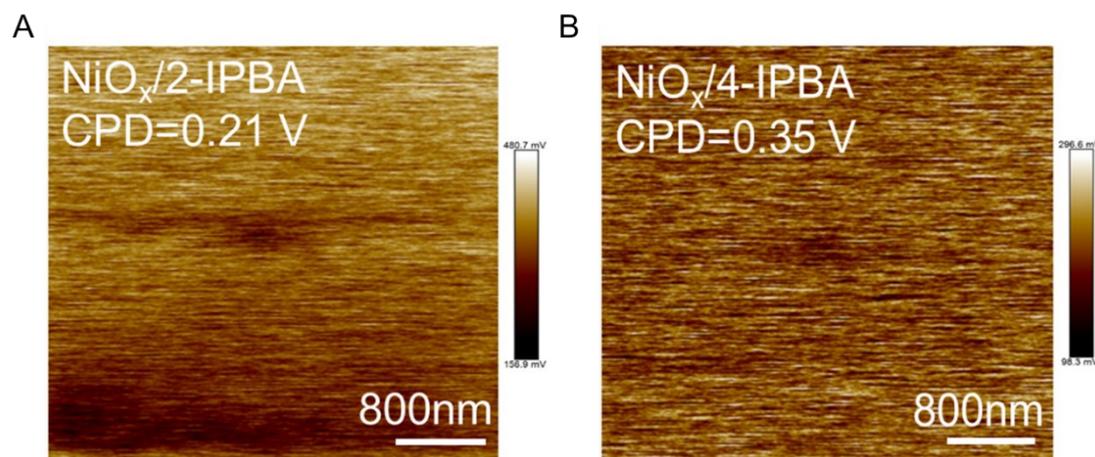
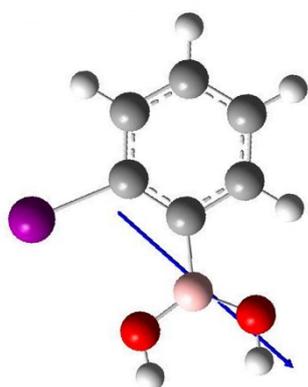


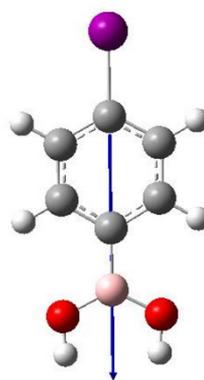
Figure S11. KPFM images of (A) the  $\text{NiO}_x/2\text{-IPBA}$  and (B)  $\text{NiO}_x/4\text{-IPBA}$  films.

A



2-IPBA  
2.85D

B



4-IPBA  
4.55D

Figure S12. Molecular structures and dipole moments of (A) 2-IPBA and (B) 4-IPBA molecules.

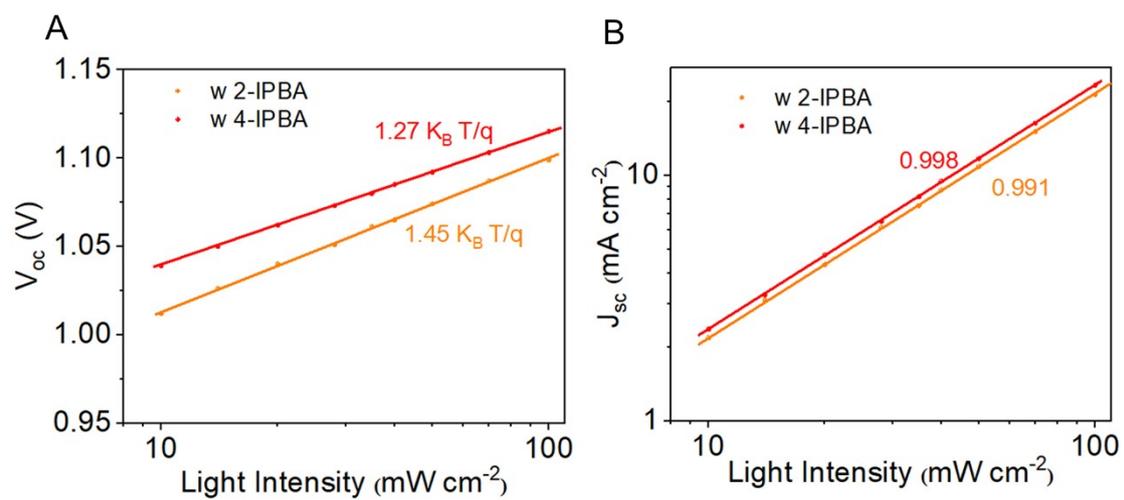


Figure S13. (A)  $V_{oc}$  and (B)  $J_{sc}$  versus light intensity of PSCs with the 2-IPBA and 4-IPBA modification.

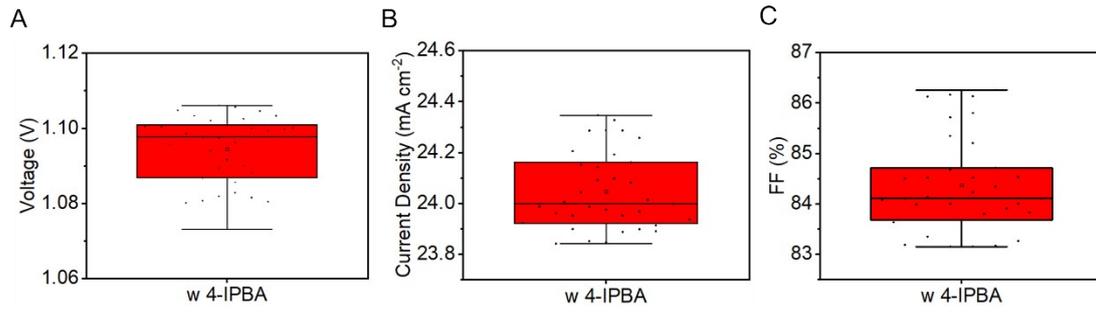


Figure S14. (A)  $V_{oc}$ , (B)  $J_{sc}$  and (C) FF statistics obtained from 35 PSCs with the 4-IPBA modification.

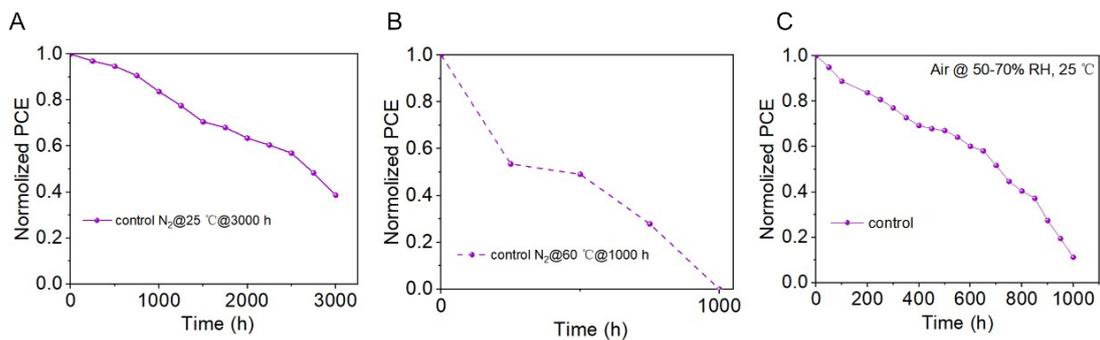


Figure S15. Normalized PCEs of the control device stored in (A)  $N_2$  atmosphere at  $25\text{ }^\circ\text{C}$  for 3000 h, (B)  $60\text{ }^\circ\text{C}$  for 1000 h and (C) air (RH of 50-70%,  $25\text{ }^\circ\text{C}$ ) for 1000 h.

Table S1. The average data collected out of 35 cells

	<b>Voc (V)</b>	<b>Jsc (mA cm<sup>-2</sup>)</b>	<b>FF (%)</b>	<b>PCE (%)</b>
<b>control</b>	1.03±0.008	24.23±0.277	79.89±1.324	20.06±0.148
<b>W 2-IBA</b>	1.06±0.010	24.19±0.160	80.93±0.144	20.61±0.236
<b>W 2-IBSA</b>	1.10±0.002	23.95±0.196	80.32±0.206	21.20±0.166
<b>W 2-IPBA</b>	1.10±0.001	23.57±0.336	82.08±0.709	21.70±0.140

Table S2. FWHM and DoO parameters from Figure 2E-H

<b>Samples</b>		$\chi \approx 0$	$\chi \approx 40$	$\chi \approx 65$
<b>NiO<sub>x</sub>/PVK</b>	FWHM	52.18	48.24	17.68
	DoO	2.34%	28.06%	11.20%
<b>NiO<sub>x</sub>/2I-BA/PVK</b>	FWHM	34.41	31.81	41.76
	DoO	5.58%	30.38%	15.62%
<b>NiO<sub>x</sub>/2I-BSA/PVK</b>	FWHM	20.57	41.14	36.91
	DoO	6.21%	22.48%	10.64%
<b>NiO<sub>x</sub>/2I-PBA/PVK</b>	FWHM	8.37	43.66	42.76
	DoO	8.02%	18.58%	8.28%

Table S3. Fitted TRPL parameters from Figure 3C

<b>Samples</b>	<b><math>\tau_1</math> (ns)</b>	<b><math>\tau_2</math> (ns)</b>	<b><math>A_1</math></b>	<b><math>A_2</math></b>	<b>ave. (ns)</b>
<b>NiO<sub>x</sub>/PVK</b>	20.75	149.46	39.84%	60.16%	98.18
<b>NiO<sub>x</sub>/2I-BA/PVK</b>	20.14	144.81	40.43%	59.57%	94.41
<b>NiO<sub>x</sub>/2I-BSA/PVK</b>	19.76	143.88	38.34%	61.66%	96.29
<b>NiO<sub>x</sub>/2I-PBA/PVK</b>	19.38	136.86	37.41%	62.59%	92.91

Table S4. Binding energy and atomic ratios of pristine NiO<sub>x</sub>, NiO<sub>x</sub> with 2-IBA treatment, NiO<sub>x</sub> with 2-IBSA treatment, and NiO<sub>x</sub> with 2-IPBA from Figure 3D-G and Figure S6

<b>Samples</b>	<b>Ni<sup>3+</sup> (eV) Ratio (%)</b>	<b>Ni<sup>2+</sup> (eV) Ratio (%)</b>	<b>Ni<sup>3+</sup>/ Ni<sup>2+</sup> (%)</b>
<b>NiO<sub>x</sub></b>	855.42	853.75	3.05
	51.32	16.82	
<b>NiO<sub>x</sub>/2I-BA</b>	855.46	853.76	2.59
	49.09	18.95	
<b>NiO<sub>x</sub>/2I-BSA</b>	855.48	853.76	2.56
	49.42	19.31	
<b>NiO<sub>x</sub>/2I-PBA</b>	855.52	853.76	2.63
	49.26	18.68	
<b>NiO<sub>x</sub>/before annealing</b>	855.40	853.75	2.52
	53.53	21.27	

Table S5. Fitted TRPL parameters from Figure 4E

<b>Samples</b>	<b><math>\tau_1</math> (ns)</b>	<b><math>\tau_2</math> (ns)</b>	<b><math>A_1</math></b>	<b><math>A_2</math></b>	<b>ave. (ns)</b>
<b>NiO<sub>x</sub>/4I-PBA/PVK</b>	16.41	137.80	42.58%	57.42%	86.11

Table S6. Comparing the performance of PSCs based on halogenated phenyl acids-modified NiO<sub>x</sub> HTLs

Modifiers	Architecture	Performance	Ref
4-iodo-2,3,5,6-tetrafluorobenzoic acid (I-TFBA)	ITO/NiO <sub>x</sub> /I-TFBA/ Cs <sub>0.05</sub> (FA <sub>0.92</sub> MA <sub>0.08</sub> ) <sub>0.95</sub> Pb(I <sub>0.92</sub> Br <sub>0.08</sub> ) <sub>3</sub> /PEAI/PCBM /BCP/Ag	V <sub>oc</sub> =1.17 V J <sub>sc</sub> =22.3 mA cm <sup>-2</sup> FF=82.2% PCE=22.02%	1
p-chlorobenzenesulfonic acid (CBSA)	ITO/NiO <sub>x</sub> /CBSA/ MAPbI <sub>3</sub> / PCBM/BCP/Ag	V <sub>oc</sub> =1.11 V J <sub>sc</sub> =23.23 mA cm <sup>-2</sup> FF=80.17% PCE=20.7%	2
Br-benzoic acid (Br-BA)	ITO/NiO <sub>x</sub> /Br-BA/MAPbI <sub>3</sub> / PCBM/C <sub>60</sub> /Ag	V <sub>oc</sub> =1.11 V J <sub>sc</sub> =21.7 mA cm <sup>-2</sup> FF=76.3% PCE=18.4%	3
4-bromobenzoic acid	ITO/NiO <sub>x</sub> /4Br-BA/ FAMAPbI <sub>3</sub> / PCBM/C <sub>60</sub> /Ag	V <sub>oc</sub> =1.01 V J <sub>sc</sub> =19.8 mA cm <sup>-2</sup> FF=64% PCE=12.6%	4
4-bromobenzylphosphonic acid	ITO/NiO <sub>x</sub> /4-Br-BPPA/ CsFAMAPbBr <sub>3</sub> I <sub>3</sub> / PCBM/Ag	V <sub>oc</sub> =1.09 V J <sub>sc</sub> =18.58 mA cm <sup>-2</sup> FF=67.18% PCE=12.73%	5
[2-(3,6-Dimethoxy-9H-carbazol-9-yl)ethyl]phosphonic Acid+4-Bromobenzoic acid	ITO/NiO <sub>x</sub> /MeO-2PACz+Br-BA/PVK/ PEA/ C <sub>60</sub> /SnO <sub>2</sub> /Ag	V <sub>oc</sub> =1.05 V J <sub>sc</sub> =23.40 mA cm <sup>-2</sup> FF=83% PCE=20.38%	6
<b>2-iodobenzoic acid (2-IBA)</b>	<b>ITO/NiO<sub>x</sub>/2-IBA/ Cs<sub>0.05</sub>(FA<sub>0.98</sub>MA<sub>0.02</sub>)<sub>0.95</sub> Pb(I<sub>0.98</sub>Br<sub>0.02</sub>)<sub>3</sub>/PCBM /BCP/Ag</b>	<b>V<sub>oc</sub>=1.06 V J<sub>sc</sub>=24.20 mA cm<sup>-2</sup> FF=80.81% PCE=20.77%</b>	<b>This Work</b>
<b>2-iodobenzenesulfonic acid (2-IBSA)</b>	<b>ITO/NiO<sub>x</sub>/2-IBSA/ Cs<sub>0.05</sub>(FA<sub>0.98</sub>MA<sub>0.02</sub>)<sub>0.95</sub> Pb(I<sub>0.98</sub>Br<sub>0.02</sub>)<sub>3</sub>/PCBM /BCP/Ag</b>	<b>V<sub>oc</sub>=1.10 V J<sub>sc</sub>=23.95 mA cm<sup>-2</sup> FF=80.97% PCE=21.37%</b>	<b>This Work</b>
<b>2-iodophenylboronic acid (2-IPBA)</b>	<b>ITO/NiO<sub>x</sub>/2-IPBA/ Cs<sub>0.05</sub>(FA<sub>0.98</sub>MA<sub>0.02</sub>)<sub>0.95</sub> Pb(I<sub>0.98</sub>Br<sub>0.02</sub>)<sub>3</sub>/PCBM /BCP/Ag</b>	<b>V<sub>oc</sub>=1.10 V J<sub>sc</sub>=23.76 mA cm<sup>-2</sup> FF=82.9% PCE=21.81%</b>	<b>This Work</b>
<b>4-iodophenylboronic acid (4-IPBA)</b>	<b>ITO/NiO<sub>x</sub>/4-IPBA/ Cs<sub>0.05</sub>(FA<sub>0.98</sub>MA<sub>0.02</sub>)<sub>0.95</sub> Pb(I<sub>0.98</sub>Br<sub>0.02</sub>)<sub>3</sub>/PCBM /BCP/Ag</b>	<b>V<sub>oc</sub>=1.08 V J<sub>sc</sub>=24.42 mA cm<sup>-2</sup> FF=86.18% PCE=22.91%</b>	<b>This Work</b>

## Notes and references

1. C. Zhang, X. Shen, M. Chen, Y. Zhao, X. Lin, Z. Qin, Y. Wang, L. Han, *Adv. Energy Mater.*, 2023, **13**, 2203250.
2. J. Zhang, J. Yang, R. Dai, W. Sheng, Y. Su, Y. Zhong, X. Li, L. Tan, Y. Chen, *Adv. Energy Mater.*, 2022, **12**, 2103674.
3. Q. Wang, C. C. Chueh, T. Zhao, J. Cheng, M. Eslamian, W. C. H. Choy and A. K. Y. Jen, *Chemsuschem*, 2017, **10**, 3794-3803.
4. H. Anizelli, T. W. David, P. Tyagi, E. Laureto, J. Kettle, *Sol. Energy*, 2020, **203**, 157.
5. J. Mangalam, T. Rath, S. Weber, B. Kunert, T. Dimopoulos, A. Fian, G. Trimmel, *J. Mater. Sci.: Mater. Electron.* 2019, **30**, 9602.
6. G. Xie, M. Ma, D. Yang, H. Yang, Y. Zeng, Y. Ma, S Wu, C, Liu, Y, Mai, *Chinese J. Lumin*, 2023, 44(6): 1023.