

Electronic Supplementary Information (ESI)

Electrochemically Engineered Domain: Nickel– Hydroxide/Nickel Nitride Composite for Alkaline HER Electrocatalysis

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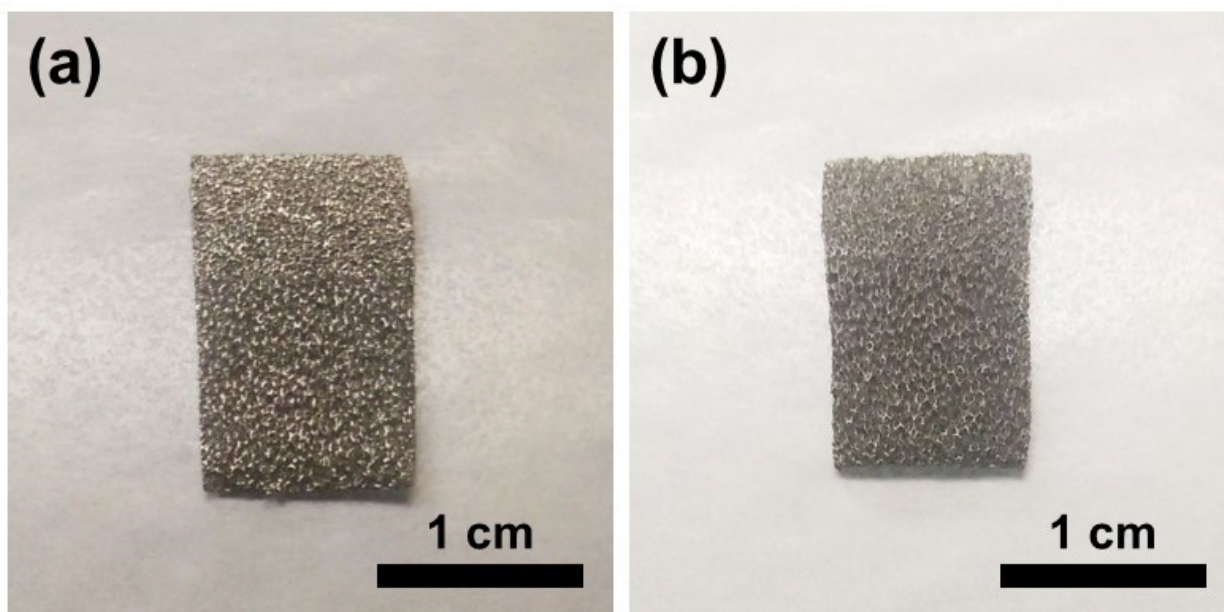


Fig. S1 Digital photographs of pristine (a) NF and (b) Ni₃N/NF.

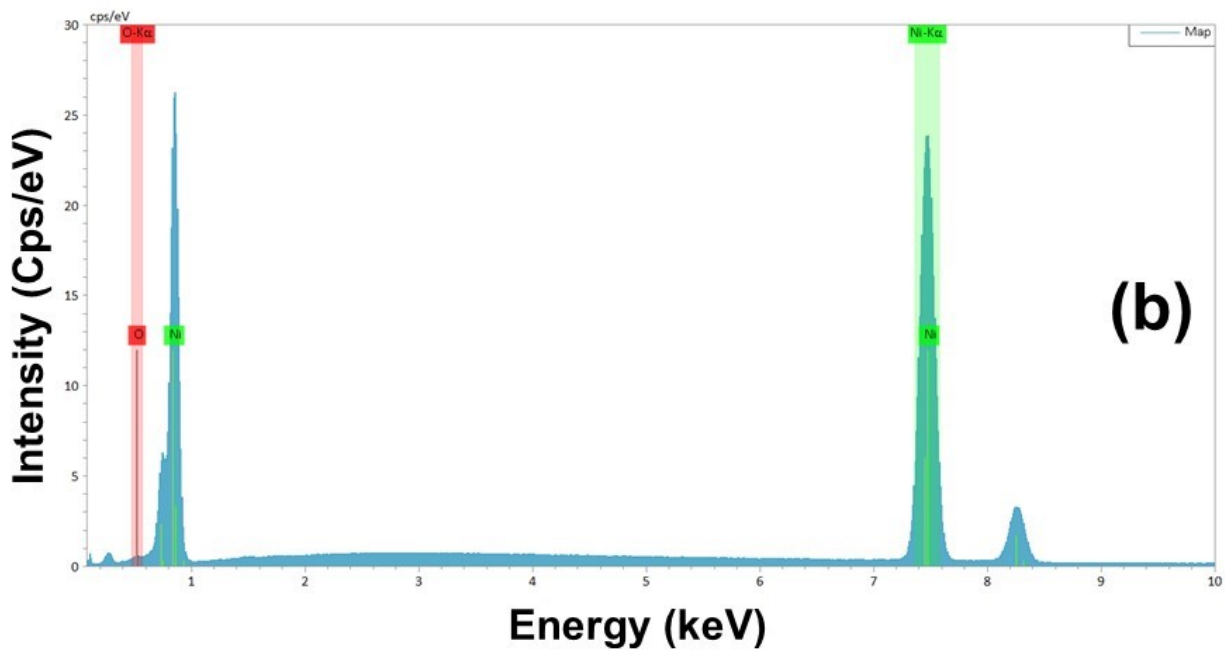
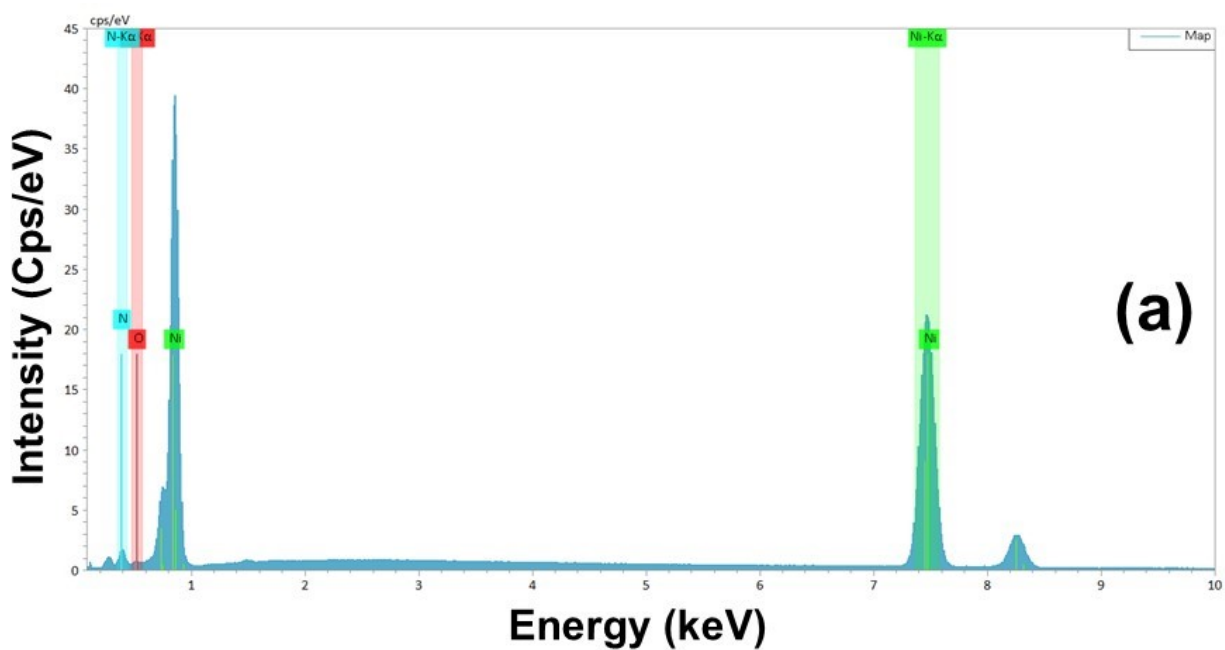


Fig. S2 EDX spectra of pristine (a) Ni₃N/NF and (b) NF.

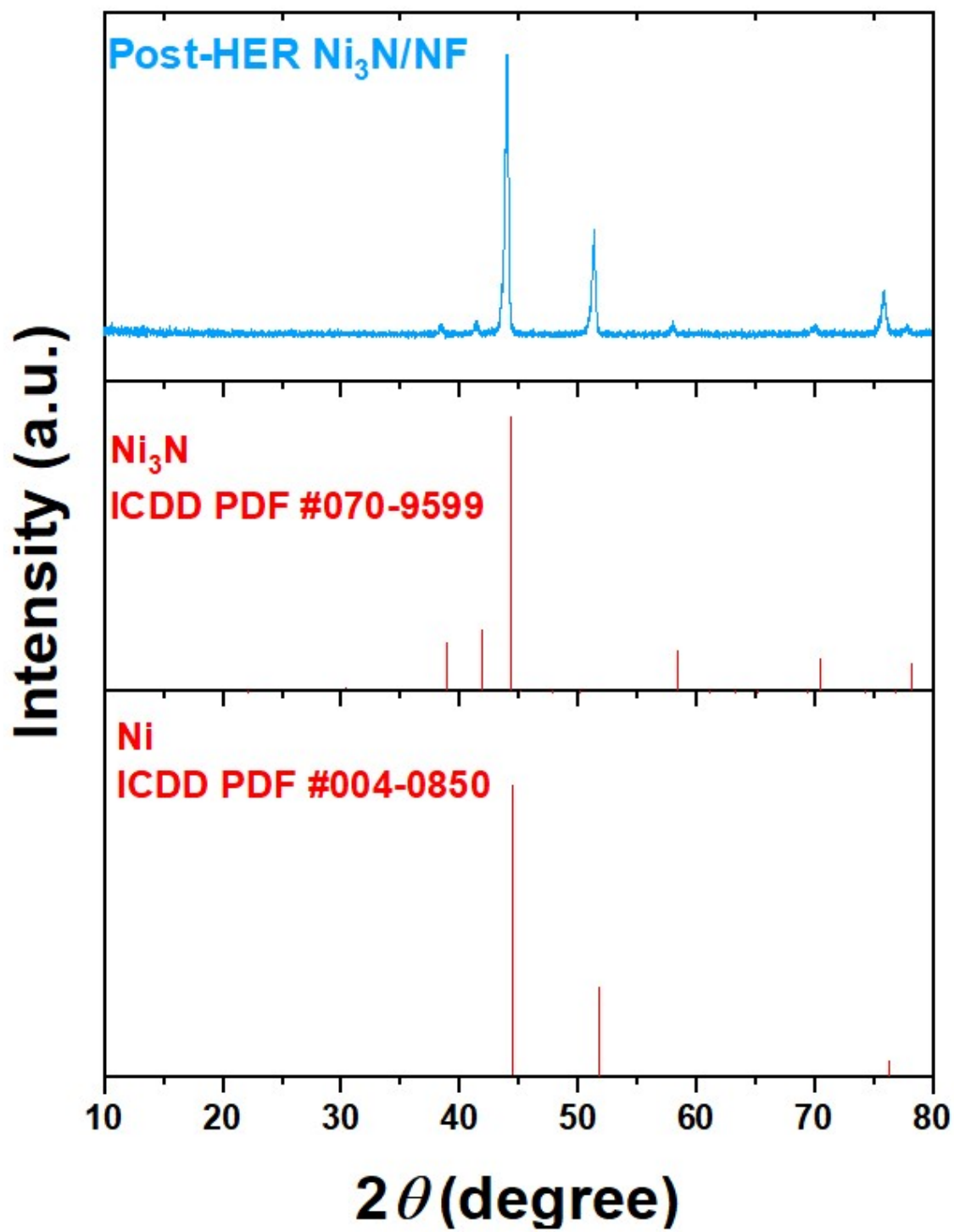


Fig. S3 XRD pattern of Ni₃N/NF after long-term HER testing.

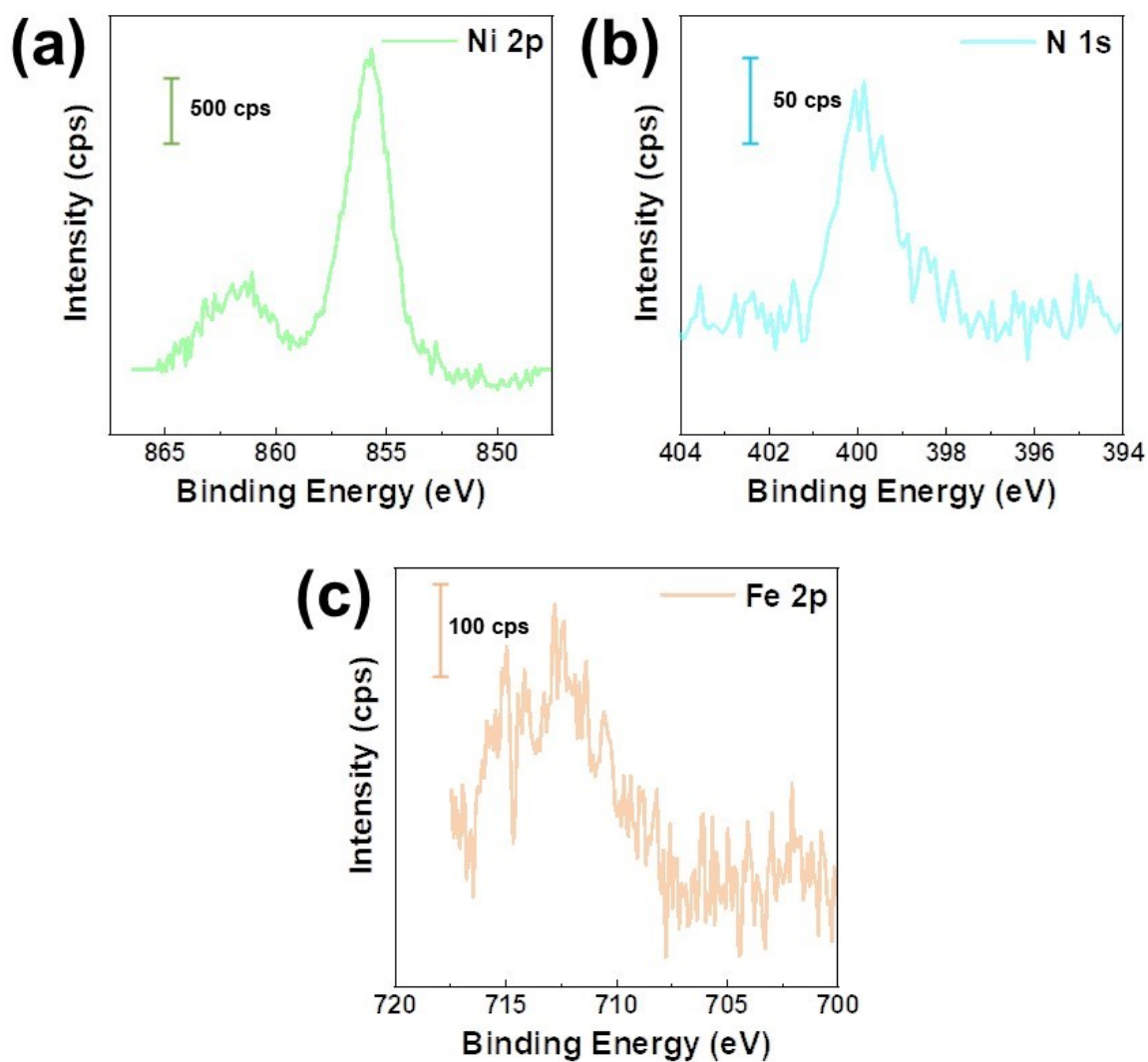


Fig. S4 (a) Ni 2p, (b) N 1s, and (c) Fe 2p XPS spectra of $\text{Ni}_3\text{N}/\text{NF}$ after long-term HER testing.

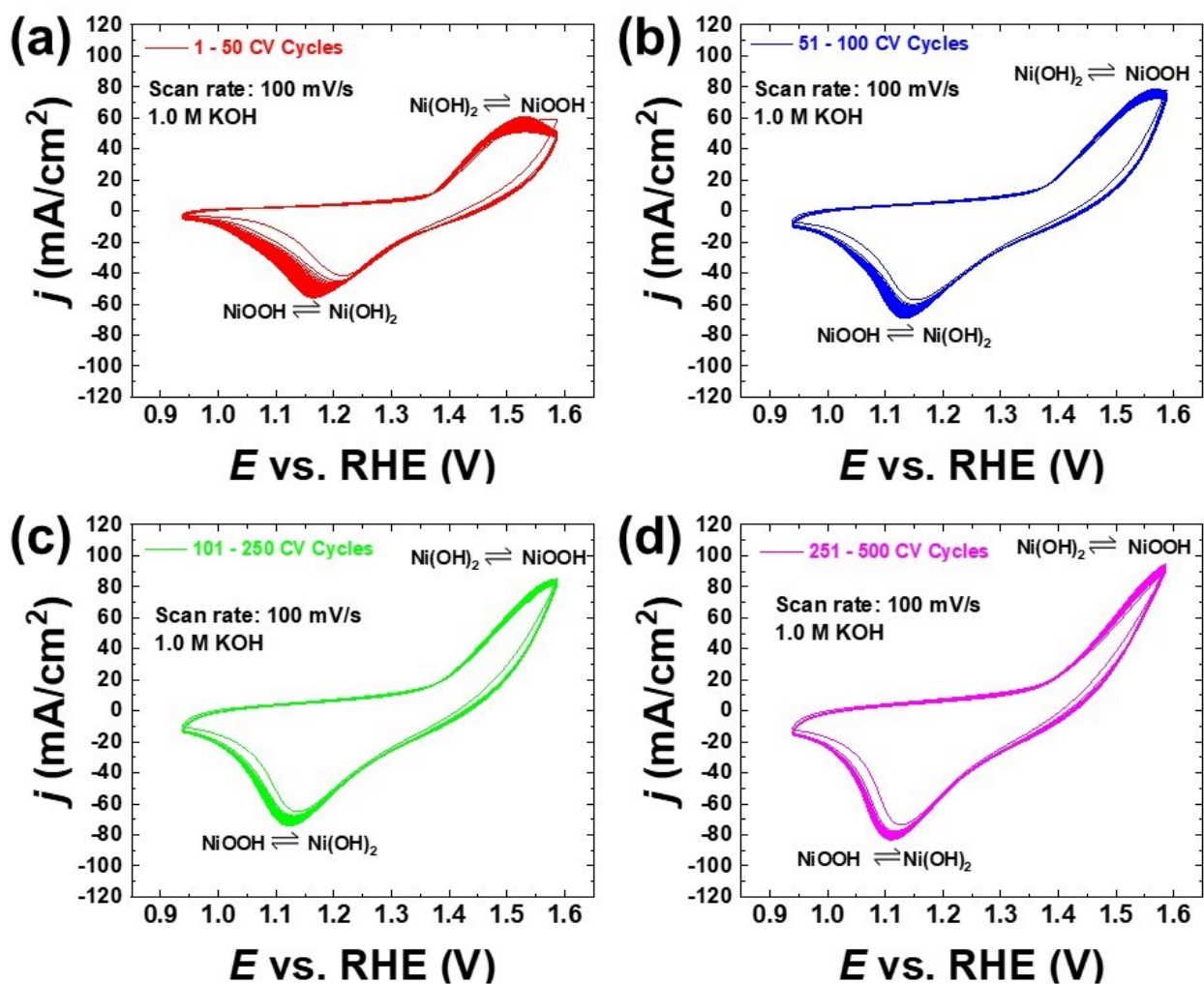


Fig. S5 CV activation plots for $\text{Ni}_3\text{N}/\text{NF}$: (a) 1-50 CV cycles, (b) 51-100 CV cycles, (c) 101-250 CV cycles, and (d) 251-500 CV cycles.

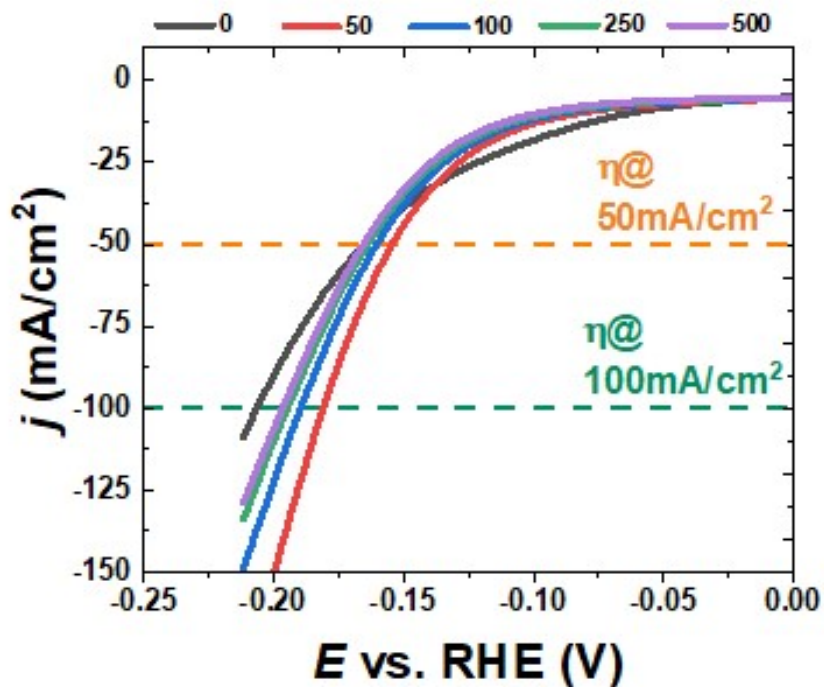


Fig. S6 LSV curves of Ni₃N/NF after CV activation with different cycle numbers: 0-500 cycles.

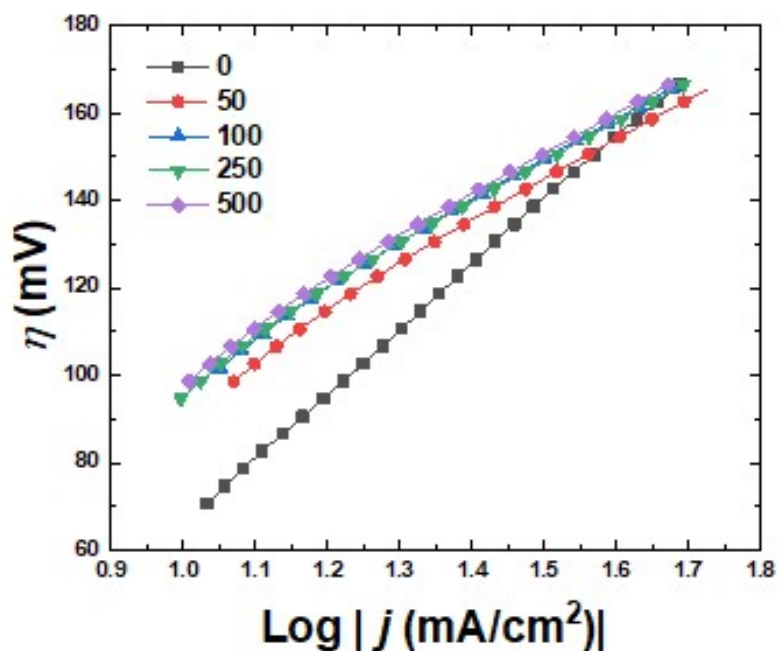


Fig. S7 Tafel slopes of Ni₃N/NF after CV activation with different cycle numbers: 0-500 cycles.

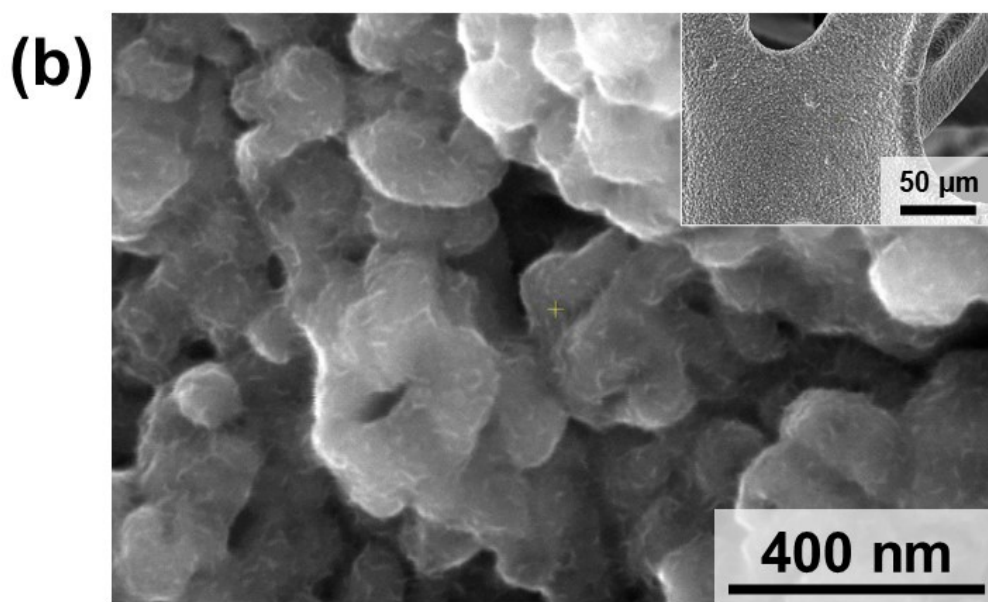
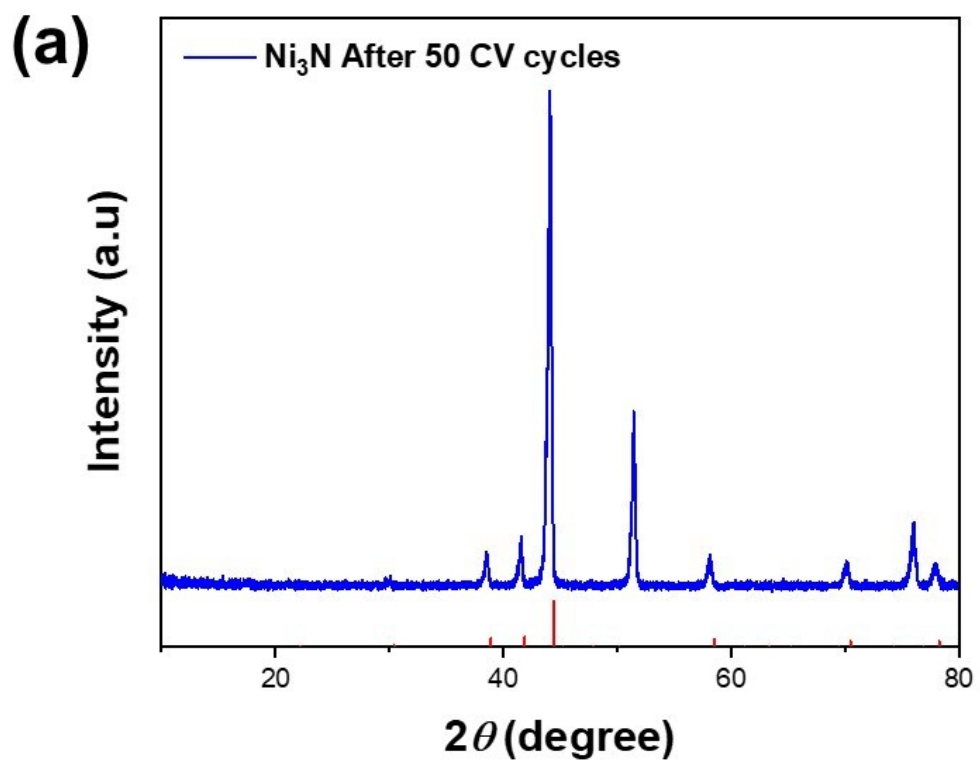


Fig. S8 (a) XRD pattern and (b) SEM images of 50 CV-activated $\text{Ni}_3\text{N}/\text{NF}$.

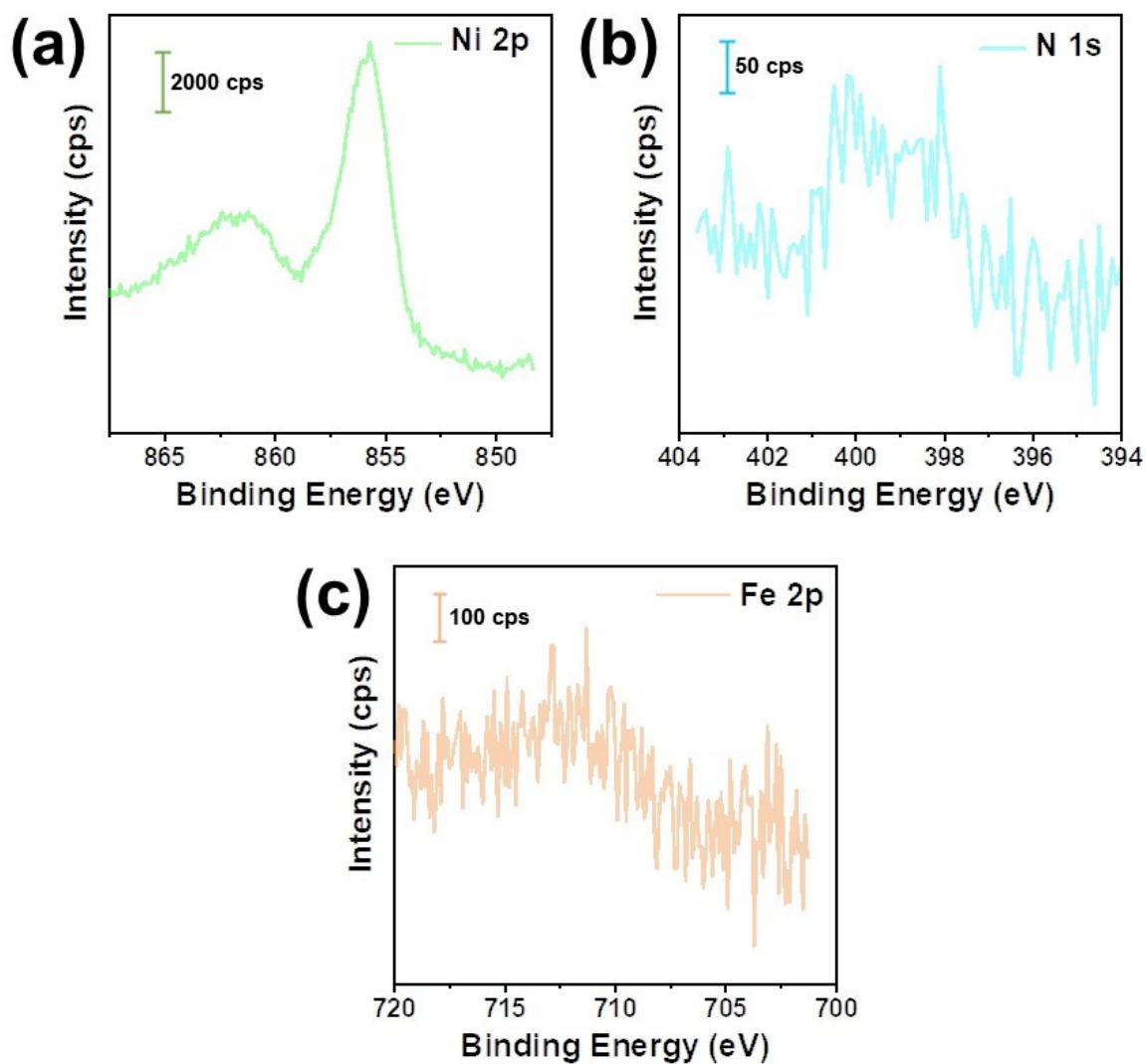


Fig. S9 (a) Ni 2p, (b) N 1s, and (c) Fe 2p XPS spectra of 50 CV-activated Ni₃N/NF.

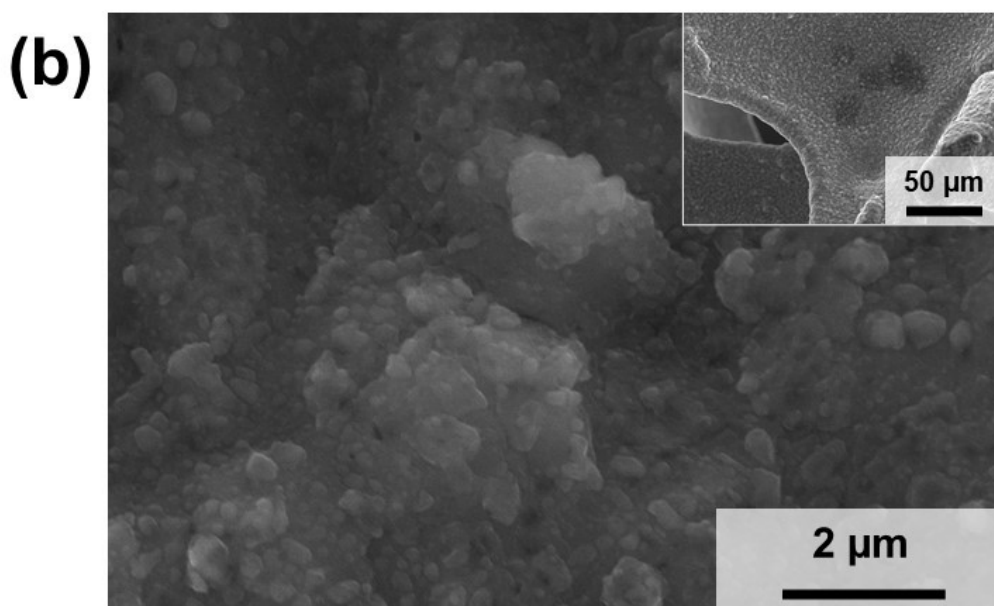
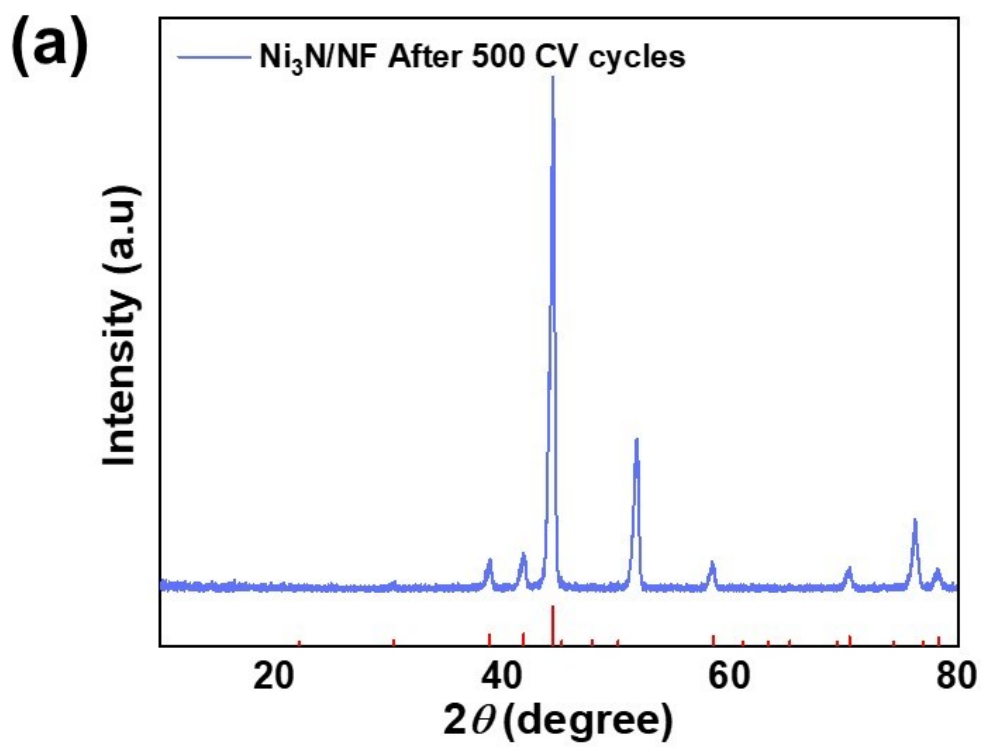


Fig. S10 (a) XRD pattern and (b) SEM images of 500 CV-activated $\text{Ni}_3\text{N/NF}$.

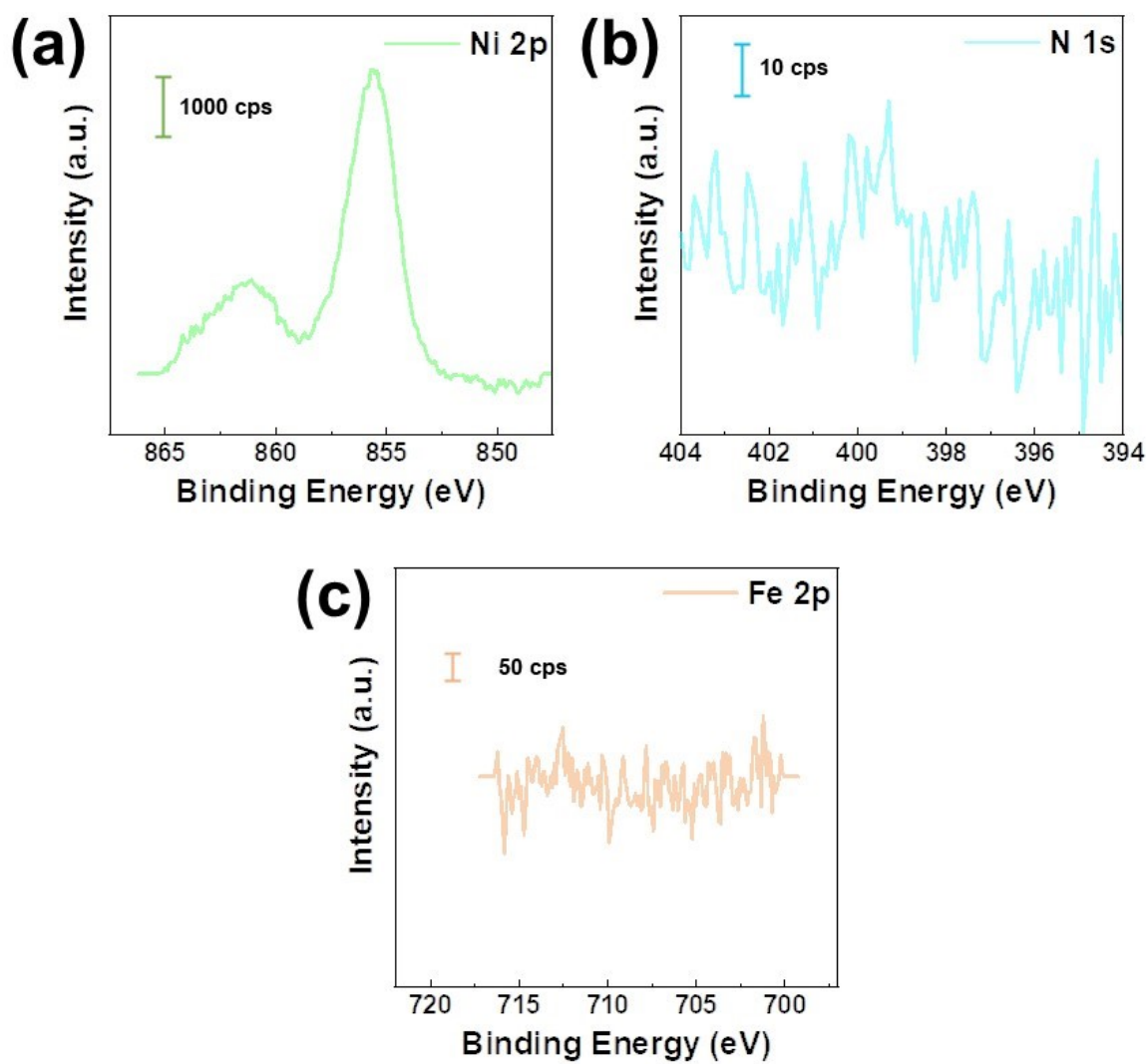


Fig. S11 (a) Ni 2p, (b) N 1s, and (c) Fe 2p XPS spectra of 500 CV-activated Ni₃N/NF.

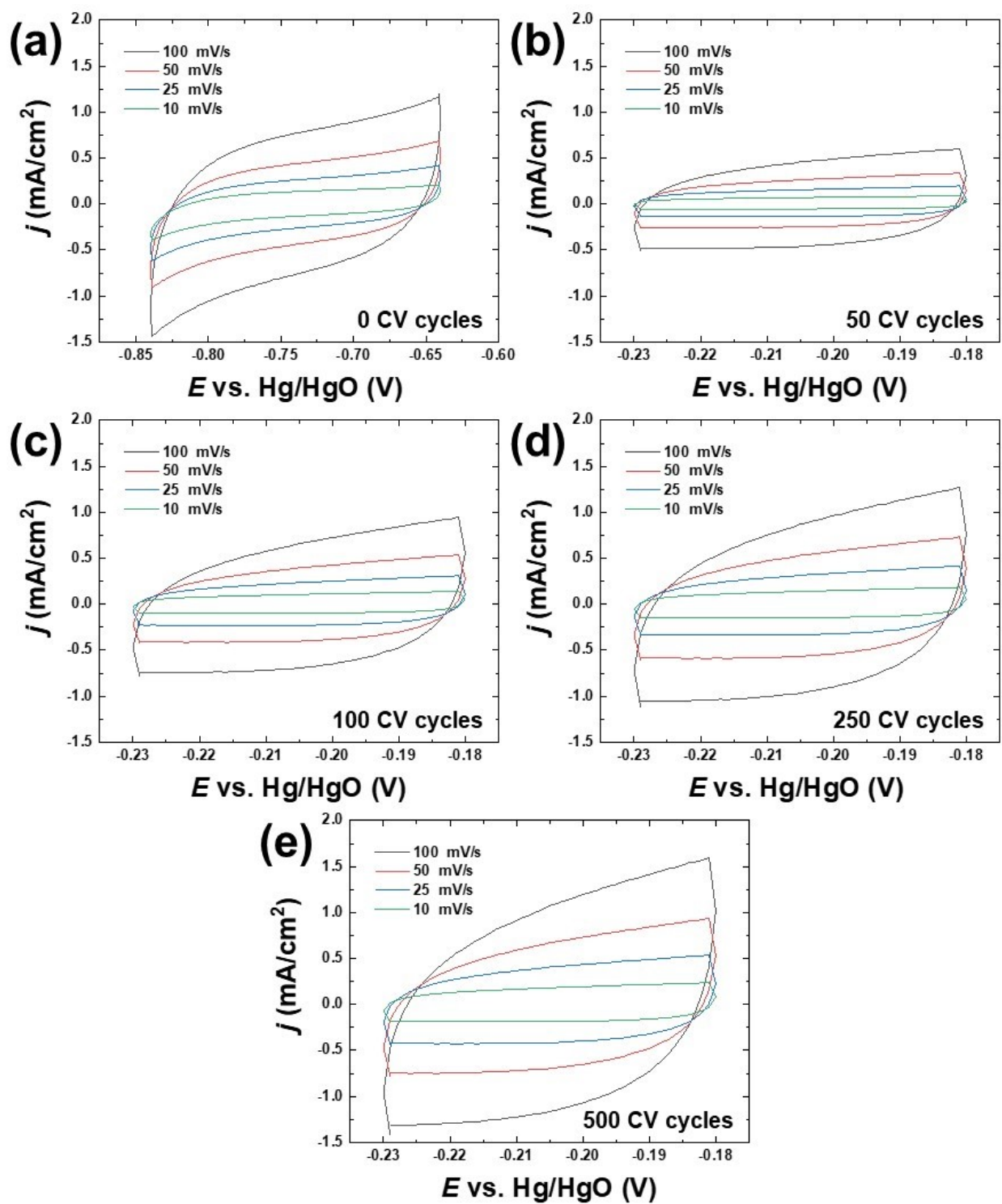


Fig. S12 Non-faradaic CV curves of (a) 0, (b) 50, (c) 100, (d) 250, and (e) 500 CV-activated Ni₃N/NF samples.

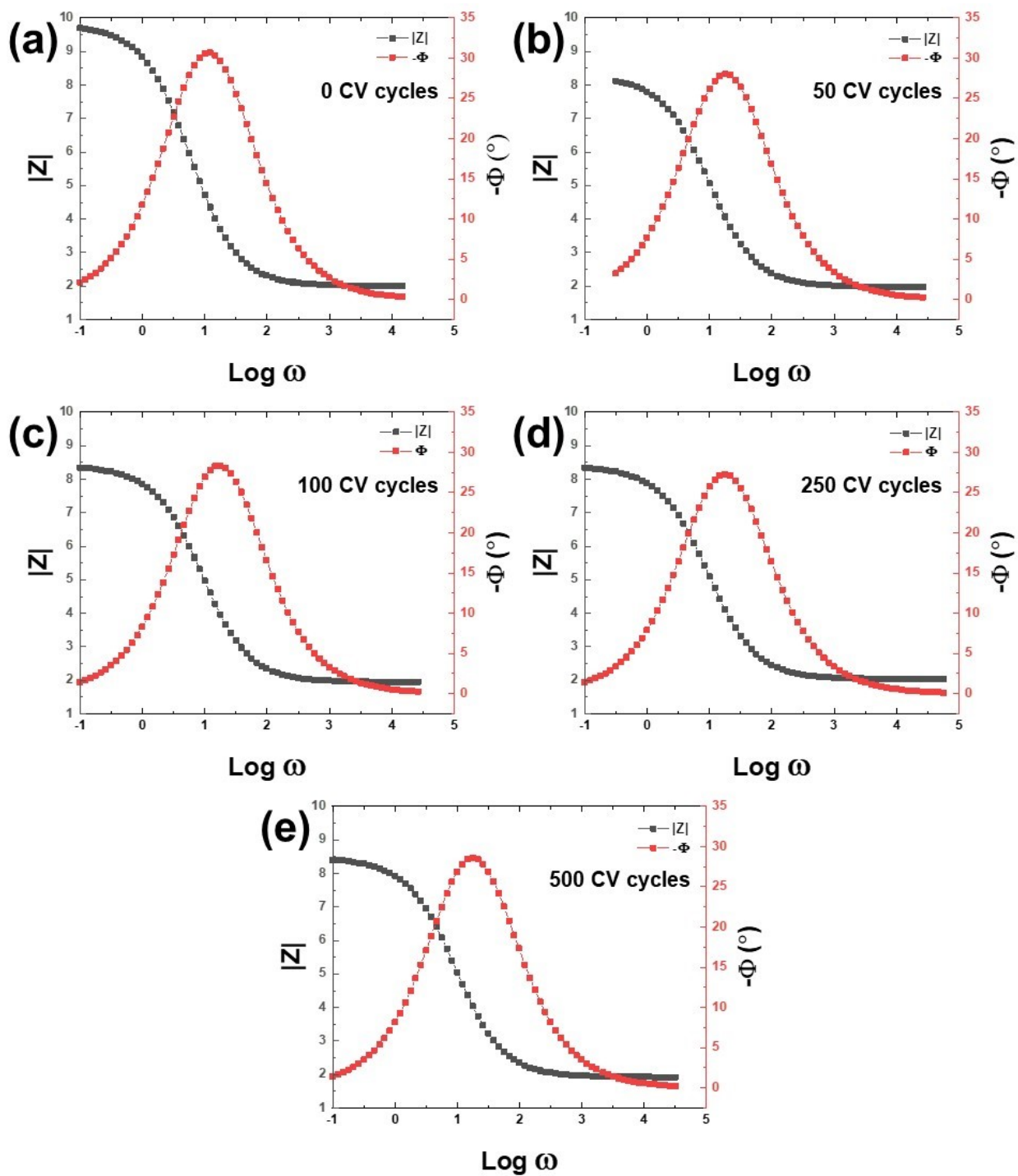


Fig. S13 Bode plots of (a) 0, (b) 50, (c) 100, (d) 250, and (e) 500 CV-activated $\text{Ni}_3\text{N}/\text{NF}$ samples.

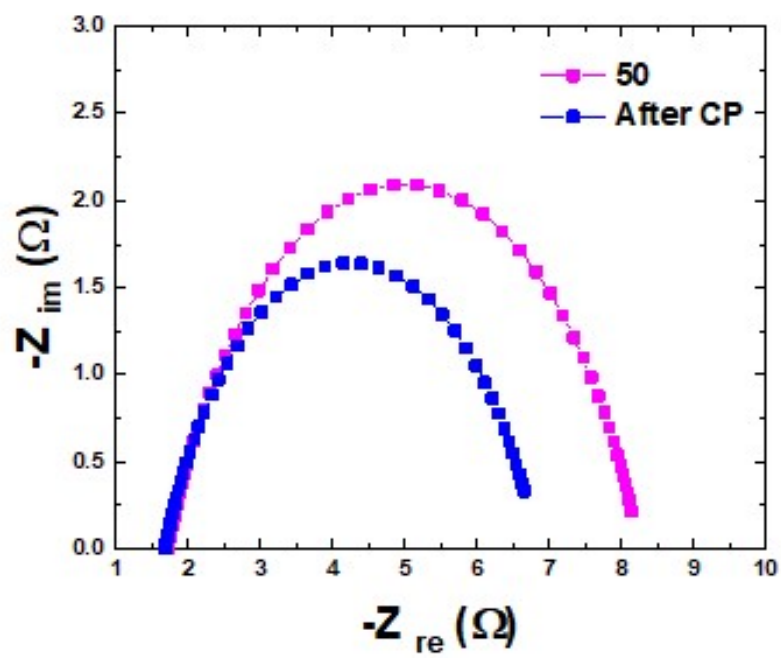


Fig. S14 Nyquist plots for 50 CV-activated Ni_3N/NF before and after long-term HER testing.

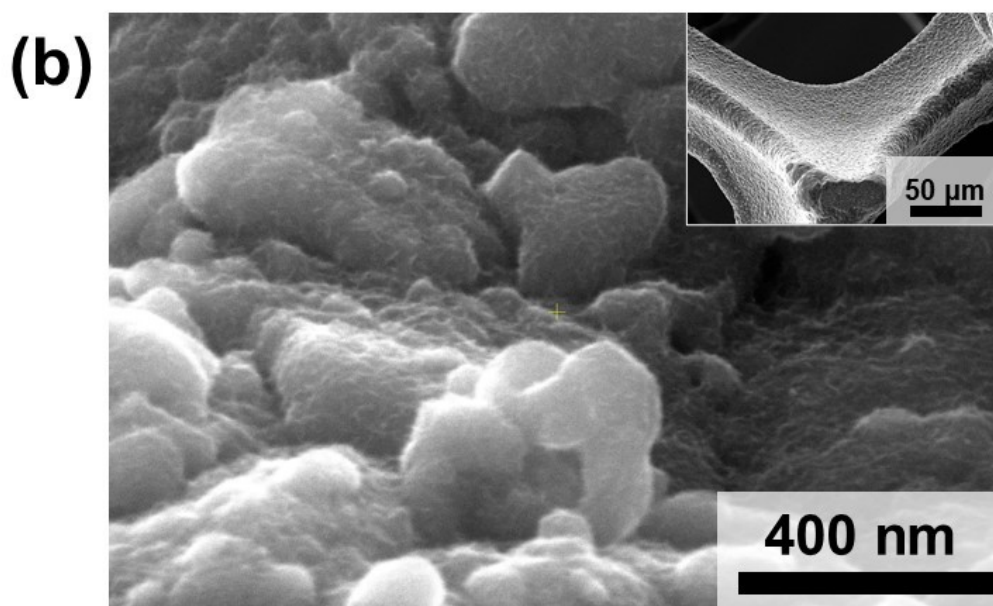
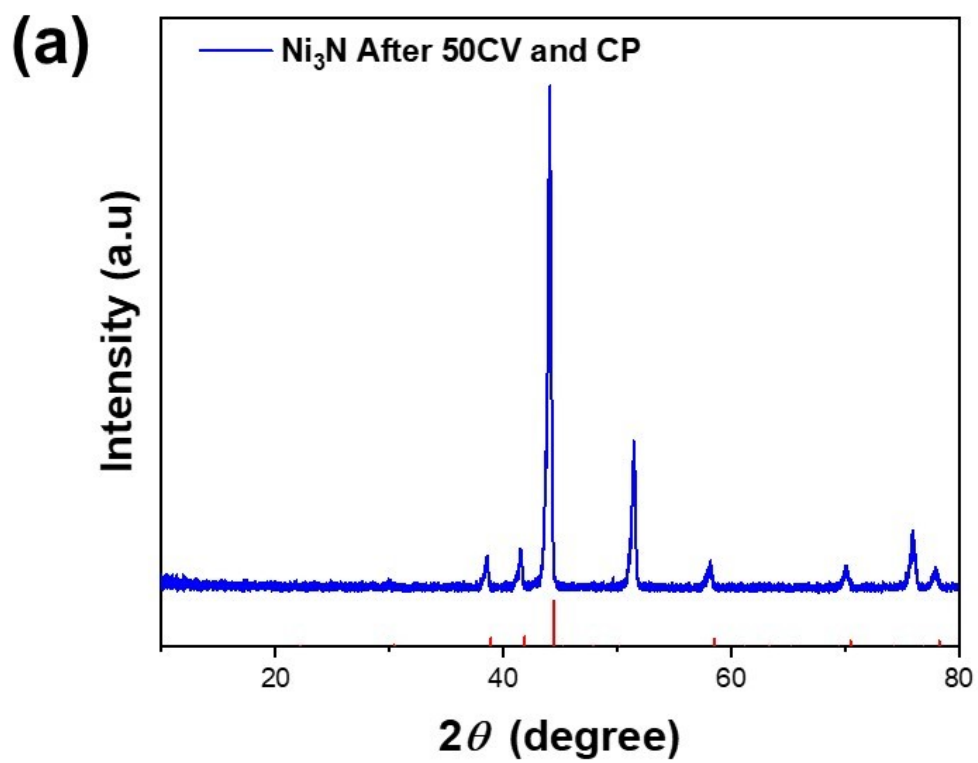


Fig. S15 (a) XRD pattern and (b) SEM images of 50 CV-activated $\text{Ni}_3\text{N}/\text{NF}$ after long-term HER testing.

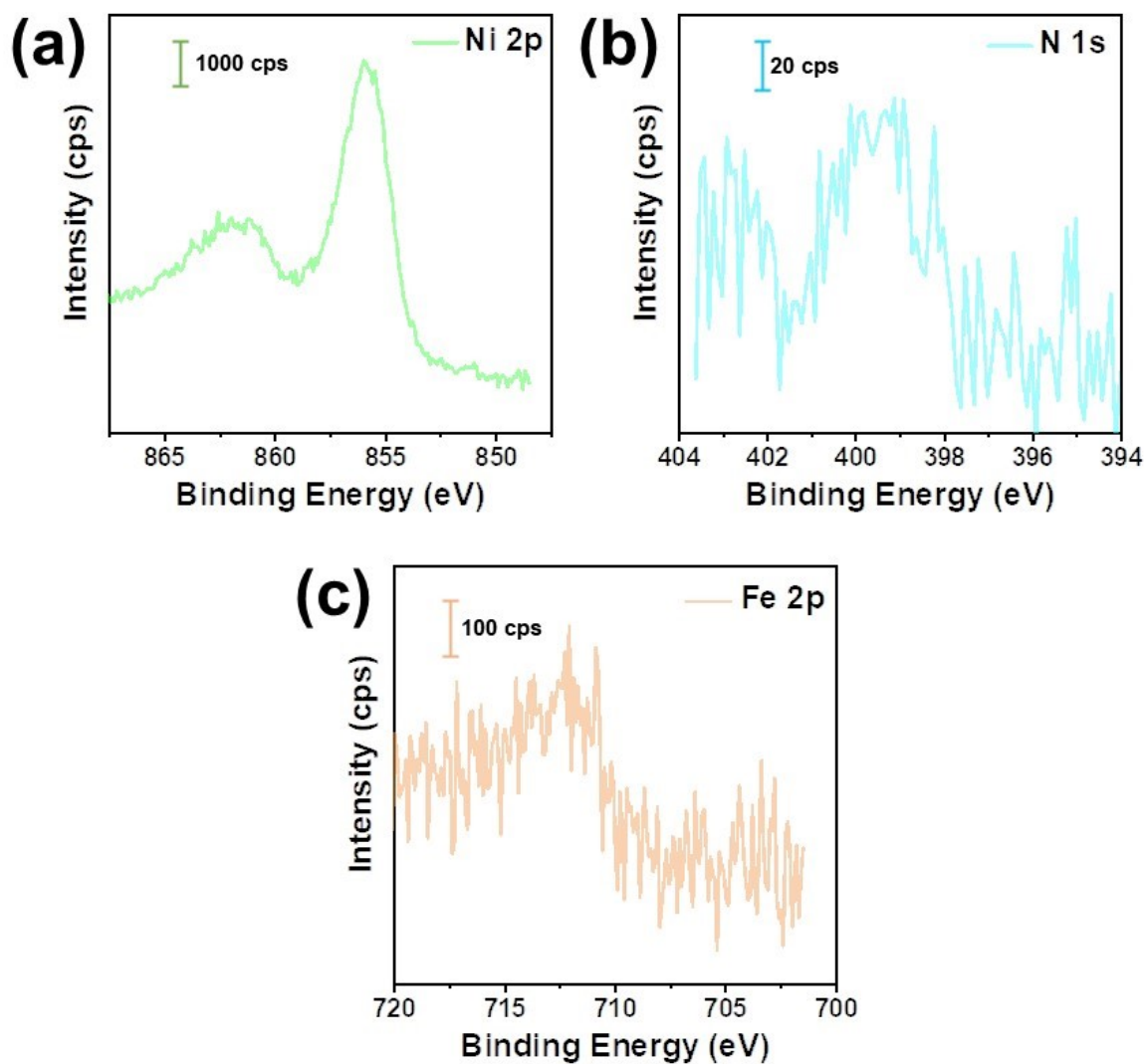


Fig. S16 (a) Ni 2p, (b) N 1s, and (c) Fe 2p XPS spectra of 50 CV-activated Ni₃N/NF after long-term HER testing.

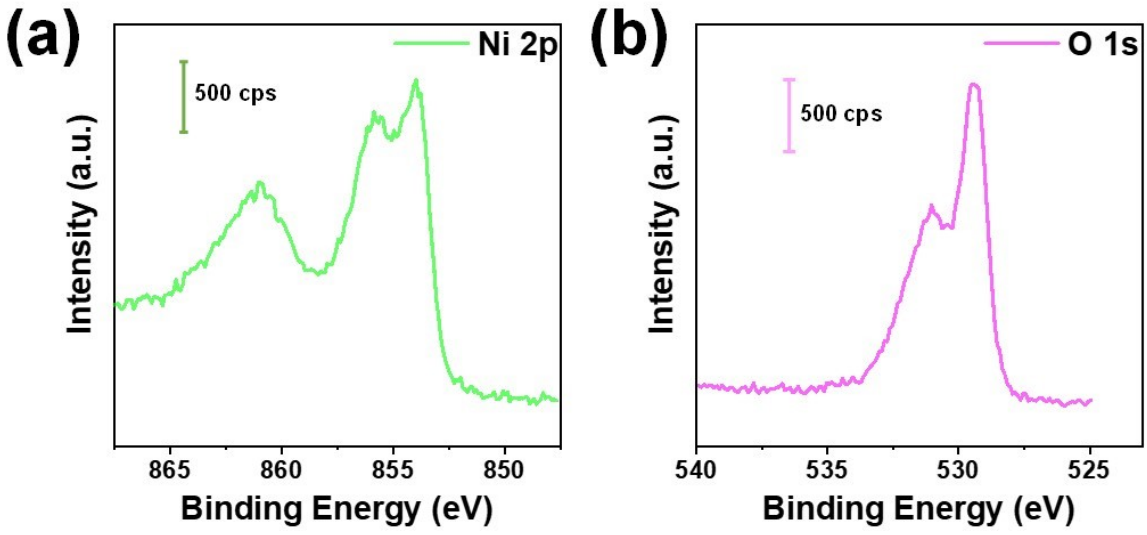


Fig. S17 (a) Ni 2p and (b) O 1s XPS spectra of NiO/NF.

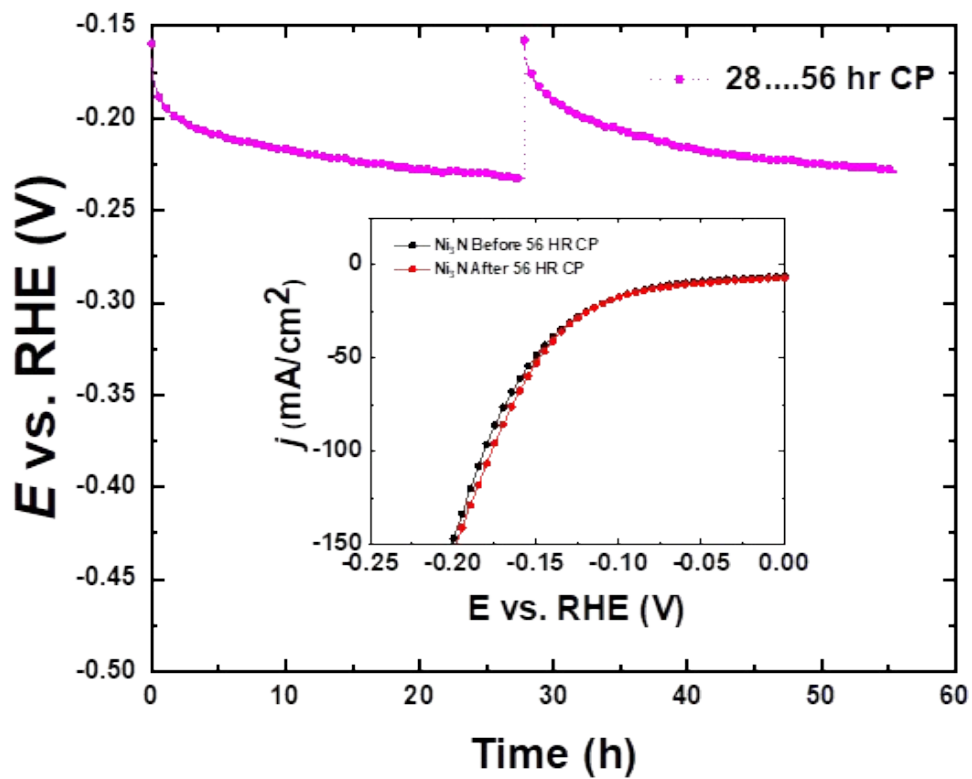


Fig. S18 ~56 h HER chronopotentiometry tests at -20 mA/cm^2 for $\text{Ni(OH)}_2\text{-Ni}_3\text{N/NF}$ and LSV curves before and after the chronopotentiometry (inset).

Table S1 An alkaline HER activity comparison (at a cathodic current density of 50 mA/cm²) of pristine Ni₃N/NF and electrochemically engineered Ni(OH)₂-Ni₃N/NF (after 50 CV cycles) in this work with previously reported similar electrocatalysts

This work pristine Ni ₃ N $\eta@50 \text{ mA/cm}^2 = 171.93 \pm 3.8 \text{ mV}$ After 50 CV cycles $\eta@50 \text{ mA/cm}^2 = 164 \pm 1.2 \text{ mV}$		
Sample Name	$\eta@50 \text{ mA/cm}^2 \text{ (mV)}$	Reference
Ni ₃ N/NF	~210	1
Ni ₃ N/NF	~220	2
Ni ₃ N _{1-x} /NF	~240	3
Ni ₃ N/NF	~200	4
Ni-S/NF	~350	5
Ni-S-OH/NF	~270	5
Ni ₃ N	~300	6
Ni ₃ N/NF	~210	7

Table S2 A alkaline HER activity comparison (at a cathodic current density of 100 mA/cm²) of pristine Ni₃N/NF and electrochemically engineered Ni(OH)₂-Ni₃N/NF (after 50 CV cycles) in this work with previously reported similar electrocatalysts

This work pristine Ni ₃ N $\eta@100 \text{ mA/cm}^2 = 207.27 \pm 3.06 \text{ mV}$ After 50 CV cycles $\eta@100 \text{ mA/cm}^2 = 192.27 \pm 1.53 \text{ mV}$		
Sample Name	$\eta@100 \text{ mA/cm}^2 \text{ (mV)}$	Reference
Ni ₃ N/NF	~300	1
Ni ₃ N/NF	~250	2
Ni ₃ N _{1-x} /NF	~350	3
Ni ₃ N/NF	~300	7
Ni-S/NF	~410	5
Ni-S-OH/NF	~310	5
Ni ₃ N	~350	6

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