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# **Supporting information**

# Electrochemical valorization of HCl for the production of chlorine *via* proton-filter functional covalent organic framework

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### **Chemicals required**

4, 4',4''-(1,3,5-triazine-2,4,6-triyl)trianiline (Tta, TCI Chemicals), 2,6-diformylpyridine (Dfp, TCI Chemicals) 1,3,5-tris(4-aminophenyl)benzene (Tab, TCI Chemicals), 1, 4-dioxane (Sigma-Aldrich), mesitylene (Sigma-Aldrich), acetic acid (Merck), N, N-dimethylacetamide (DMA) (Sigma-Aldrich), acetone (Merck), methanol (Fisher Scientific), were used as received. All the reactions were carried out in oven-dried 35 ml heavy walled (HW) pressure glass vessel capped with Teflon screw cap with rubber internal thread under an air atmosphere unless otherwise mentioned.  $Rh_xS_y/C$  (30%) was obtained from De Nora;  $RuO_2$  was obtained from Sigma-Aldrich

### **Physical characterization**

For the preliminary analysis of the synthesized COFs, powder X-ray diffraction (P-XRD) measurements were performed on Rigaku Smart Lab II with Cu K $\alpha$  ( $\lambda$  = 1.5405 Å) radiation source operating at 40 kV and 40 mA. The patterns were recorded with divergent slit of 1/16° over the  $2\theta$  range of 2–50° with step size = 0.02°. Fourier transform infrared (FT-IR) spectra were taken on a Bruker Optics ALPHA-E spectrometer with a universal Zn-Se ATR (attenuated total reflection) accessory in the 600- 4000 cm<sup>-1</sup> region or using a Diamond ATR (Golden Gate) with 24 scan rate and 4 cm<sup>-1</sup> resolution. Solid-state <sup>13</sup>C cross polarization-magic angle spinning (CP/MAS) NMR spectra of the COFs were recorded on a Bruker Avance 500 Wide Bore (500MHz) spectrometer at ambient temperature with a magic angle spinning rate of 18.0 kHz and the data were processed using Top Spin software. The BET and porosity analyses were performed on an Anton Paar Autosorb iQ combined physisorption and chemisorption instrument. For each measurement, 20 – 30 mg of COF samples were used and were activated at 80 °C for 16 hours in N<sub>2</sub> gas and utilised liquid N<sub>2</sub> bath (77 K) to collect full isotherms. Further, the surface areas were calculated using the multipoint Brunauer-Emmett-Teller (BET) model, and pore size distributions were evaluated/calculated using the non-local density functional theory (NLDFT). TGA analysis was performed using a PerkinElmer Simultaneous Thermal analyzer STA 6000 under N<sub>2</sub> environment at a heating rate of 15 °C min<sup>-1</sup> and a temperature range of 30-900 °C.<sup>1</sup>

The morphological and elemental analyses was done by scanning electron microscopy (SEM, JEOL JSM-7610F FEG-SEM. The samples for SEM analyses were prepared by drop-casting 10  $\mu$ L of material slurry on a silicon substrate (slurry prepared by dispersing it in isopropyl alcohol) and drying in air followed by Pt coating (nano-sized film) using the JEOL JEC-300FC Auto Fine instrument. Further, transmission electron microscopy (TEM) and high-resolution TEM (HR-TEM) were employed for the in-depth morphological analysis by FEI Tecnai TEM 20 kV. The TEM samples were prepared by drop casting the material slurry over carbon grids (TED PELLA, INC. 200 mesh) and then drying overnight in desiccators. X-ray photoelectron spectroscopy XPS measurements were performed using a Supra+ instrument (Kratos, Manchester, UK) equipped with an Al K $\alpha$  excitation source and a monochromator. The charge neutralizer was on during the measurements. The take-off angle was 90°. XPS measurements and data processing were performed using ESCApe 1.5 software (Kratos). The powder

samples were placed on a carbon tape attached to the silicon wafer. The area analysed was 300 by 700 microns. The measurements were performed at a pass energy of 20 eV. The base pressure in the main analysis chamber was 8 x  $10^{-8}$  mbar. The binding energy scale was corrected based on the C-C/C-H peak at 284.8 eV in the C 1s spectrum.

# **Electrochemical investigation**

The synthesized catalyst was meticulously homogenized in a mortar and pestle and 3.00 mg of it was added to 50  $\mu$ I IPA followed by the addition of 945  $\mu$ I of DI water and 5  $\mu$ I of 0.25% Nafion solution. The mixture was ultrasonicated for an hour to obtain a uniform catalyst slurry. The carbon paper was cleansed in 5 M HNO3 to get rid of surface oxides before analysis, then it was ultrasonically processed for 15 minutes in water, followed by another 15 minutes in isopropyl alcohol (IPA). Following that, the 40  $\mu$ I of catalytic ink was drop-casted on carbon paper and left to dry at room temperature. The electrochemical measurements were performed using Biologic VSP-300 potentiostat/galvanostat connected with EC-Lab software. The measurements were carried out using three-electrode assembly, which comprised of Ag/AgCI/3M KCI as reference electrode and graphite electrode was employed as counter electrode. The working electrode was prepared by drop-casting 20  $\mu$ L of catalyst slurry followed by drying at room temperature.

The electrocatalytic activity w.r.t oxygen reduction reactions were analysed by linear sweep voltammetric (LSV) experiments were performed in  $O_2$  and Ar saturated 0.4 M HCl at a scan rate of 10 mV s<sup>-1</sup> across a potential range of 0.5 to -0.1 V vs. RHE. The hydrodynamic experiments viz, rotating disk electrode (RDE) using a glassy carbon disk electrode from Pine and experiments were performed using a speed controlling unit (AFMSRCE, Pine Research Instrument Inc., USA) at various rotation rate in the range of 100 to 1300 rpm. All the potential are converted to the RHE scale for uniformity unless detailed otherwise.

 $E_{RHE} = E_{Ag/AgCl} + E_{Ag/AgCl} + 0.059 \, pH$ 

The current density is calculated using the equation:

$$j = {i \choose A}$$

where i is the current in mA and A is the size of the electrode in cm<sup>2</sup>.

Where  $E_{RHE}$  is the converted potential vs. RHE,  $E_{Ag/AgCl}$  is the experimentally measured potential vs. Ag/AgCl (saturated KCl), and  $\hat{E}_{Ag/AgCl}$  is 0.1976 V at 25 °C.

# Rotating disk electrode (RDE) analysis

The RDE measurements were performed for each variants by sweeping the potential from 0.5 V to -0.1 V in the oxygen saturated 0.4 M HCl solution at different rotation rates ranging from 0 to 1300 rpm. The measurements were recorded in Autolab 302 N modular potentiostat/galvanostat connected with Nova 1.11 software.

The mentioned K-L plots were extracted from RDE polarization curves using the following equation:

$$1/j = 1/j_k + 1/j_d$$

Where, 'j' stands for the measured current density, ' $j_k$ ' and ' $j_d$ ' represent kinetic and diffusion current densities, respectively. Furthermore, the slope of the K-L plot was computed by linearly fitting the graph between the inverse of the current density measured (y-axis) and inverse of square root of rotation rate (x-axis) to obtain the number of electrons transferred during the reaction. For calculating the kinetic parameters, the following equation was utilized;

$$j_d = 0.62 n F D^{2/3} \upsilon^{-1/6} D_{02}$$

Here, n depicts the number of electrons transferred during the ORR, F is the Faraday constant (96500 C mol<sup>-1</sup>), D is the diffusion coefficient of dissolved oxygen ( $1.43*10^{-5}$  cm<sup>2</sup> s<sup>-1</sup>),  $\upsilon$  is the kinematic viscosity ( $1.13*10^{-2}$  cm<sup>2</sup> s<sup>-1</sup>) and D<sub>02</sub> represents the concentration of dissolved oxygen (1.61 mol cm<sup>-3</sup>) in 0.4 M HCl.

# Electrochemical impedance spectroscopy (EIS)

EIS measurements were performed at 1.6 V vs. RHE under illumination over a frequency range from 7 MHz to 10 Hz. Further, to obtain charge transfer resistance ( $R_{ct}$ ) the EIS curves were z fit using an appropriate equivalent circuit. Furthermore, the rate constant,  $k^0$  was calculated from  $R_{ct}$  by using the formulae:

$$k^0 = \frac{RT}{n^2 F^2 A C R_{ct}}$$

Where R is gas constant, T is temperature in Kelvin, n is the no. of electrons involved in the reaction, F is the faraday constant, A is the area of the electrode, C is the concentration of the reacting species and  $R_{ct}$  is the charge transfer resistance.

# Electrochemical surface area (ECSA)

The ECSA was estimated by electrochemical double-layer capacitance ( $C_{dl}$ ) via CV plots obtained at different scan rates ranging from 20-300 mV s<sup>-1</sup>, with a sweep potential ranging from 0.63 to 0.75 V vs. RHE. The  $C_{dl}$  can be calculated from the slope (k) of the current density against scan rate at 0.7 V vs. RHE.

Further, the ECSA was estimated according to the equation:

$$ECSA = \frac{C_{dl}}{C_s}$$

where  $C_s$  is a specific capacitance of 40  $\mu$ F cm<sup>-2</sup> Additionally, specific surface area (SSA) has been calculated using the equation:

 $SSA = \frac{ECSA}{loading \ of \ the \ catalyst}$ 

#### Scanning electrochemical microscopy (SECM) studies

The scanning electrochemical microscopic (SECM) approach was used to gain insight into the local electrocatalytic activity and distribution of active sites over the catalyst spot. Sensolytics Base SECM (Sensolytics, Bochum, Germany) with stepper-motor-driven x-y-z stages and an additional three-axis piezo-positioning system working in conjunction with the bipotentiostat (Autolab 204, Metrohm) was utilized to record SECM analysis. To perform the SECM measurements, four electrodes assembly was used including Pt- micro-electrode (WE2) having diameter of 25  $\mu$ m, Ag/AgCl/3M KCl as reference, Pt-coil as counter and glassy carbon plate containing catalyst spot as a working electrode (WE1) respectively in oxygen saturated 0.4 M HCl solution. The electrocatalytic ability of each spot vs. Pt-microelectrode towards oxygen reduction in 0.4 M HCl solution was measured using the redox competitive mode of SECM. The GC (WE1) electrode was polarized at different potentials (0.5, 0.2, 0.0 and -0.1 V) whereas, Pt- micro-electrode (WE2) was constantly polarized at 0.2 V vs. RHE. The array scan was performed in x-y using tilt-control mode and the data was analyzed using Gwyddion and Origin 8.5 software. Moreover, two distinct pulse potential profile was also employed in order to see the stability and durability of the catalysts.





Fig. S1 Synthetic scheme of (a) Tab-Dfp and (b) Tta-Dfp.<sup>1</sup>



**Fig. S2** Comparison of experimental P-XRD pattern of **Tta-Dfp** (-); with corresponding simulated, and Pawley refined difference; Background; inset shows Rp, Rwp, and Rwp (w/o bck) (lattice parameters: R, weighted profile R values: Rwp, Un-weighted profile R values Rp).

Tta-Dfp (Space group P3)											
	Unit cell parameters										
		a =	: 35.92 Å,	b = 35.9	92 Å, c = 9	9.75 Å; α =	= 90°, β =	90°, γ =	120°		
Atoms	X	У	Z	Atoms	X	У	Z	Atoms	X	у	Z
H1	-0.87009	-1.07606	-0.03212	C15	-2.86932	-1.95742	-0.03478	C45	-0.61776	-1.95232	0.63011
H2	-0.94418	-1.07672	-0.031/2	C16	-2.82751	-1.95827	-0.03486	N46	-0.60947	-1.98885	0.63271
H3	-0.86773	-0.92391	-0.03489		-2.82686	-2.00033	-0.03397	C47	-0.64499	-2.03419	0.63486
H4	0 7006	-0.92540	-0.03504	U18 N40	-2.11232	2.03094	-0.03072	C48	0 70909	-2.07203	0.03705
	-0.7990	-1.00711	-0.02731	C20	-2.71311	-2.05555	-0.03125	C49	-0.70000	-2.24203	0.03033
H7	-0.0210	-1 28719	-0.04197	C20	-2.68642	-1 97343	-0.03202	C51	-0.66575	-2.20043	0.03303
H8	-0 75791	-1 21712	-0.0330	C22	-2 64043	-1.96401	-0.03613	C52	-0.62349	-2 15838	0.000000
H9	-0.60815	-1.1316	-0.00588	C23	-2.63112	-2.00065	-0.03583	C53	-0.62377	-2.20093	0.64821
H10	-0.60075	-1.20048	-0.00748	C24	-2.66821	-2.04682	-0.03313	C54	-0.6661	-2.24371	0.64221
C11	-0.95693	-0.99844	-0.03369	C25	-2.65777	-2.08524	-0.03348	C55	-0.66671	-2.28897	0.64316
C12	-0.86933	-1.04304	-0.0331	C26	-2.71915	-2.25334	-0.03246	N56	-0.37786	-1.67016	0.62749
C13	-0.91188	-1.0436	-0.03306	C27	-2.72353	-2.21334	-0.0329	N57	-0.66833	-2.11576	0.6372
C14	-0.91237	-0.99902	-0.034	C28	-2.68407	-2.16871	-0.02488	N58	-0.62611	-2.28996	0.64318
C15	-0.86932	-0.95742	-0.03478	C29	-2.63974	-2.1649	-0.01479	H59	-0.53273	-1.80838	0.66862
C16	-0.82751	-0.95827	-0.03486	C30	-2.63535	-2.20481	-0.01481	H60	-0.46416	-1.73173	0.66373
C17	-0.82686	-1.00033	-0.03397	C31	-2.67493	-2.24954	-0.02371	H61	-0.37789	-1.79335	0.59316
C18	-0.77232	-1.03094	-0.03072	C32	-2.66994	-2.29181	-0.02389	H62	-0.44769	-1.87115	0.5951
N19	-0.71311	-1.05533	-0.03125	N33	-2.99841	-2.03912	-0.03364	H63	-0.58842	-1.87658	0.61819
C20	-0.72317	-1.01999	-0.03202	N34	-2.78191	-1.99728	-0.0339	C74	-0.37331	-1.70883	0.62752
C21	-0.68642	-0.97343	-0.03434	N35	-2.69067	-2.1287	-0.02607	C75	-0.49961	-1.80619	0.64973
022	-0.64043	-0.96401	-0.03613	N36	-2.62685	-2.28/55	-0.02395	C75	-0.4599	-1./01/0	0.62914
C23	-0.03112	1 04692	-0.03583	H109	-2.09308	1 02020	-0.03469	C79	-0.41539	1 7062	0.02014
C24	-0.00021	-1.04002	-0.03313		-2.01190	-1.92020	-0.0376	C70	-0.4110	-1.84081	0.00930
C25	-0.03777	-1.00524	-0.03340	H1	-2.39324	-1.99301	-0.0370	C80	-0.43143	-1.84637	0.01021
C27	-0.72353	-1 21334	-0.0329	H2	-1.92328	-1 86746	-0.03172	C81	-0.57921	-1 90353	0.62681
C28	-0.68407	-1.16871	-0.02488	H3	-2.07609	-1.94382	-0.03489	N104	-0.53573	-1.89305	0.63087
C29	-0.63974	-1.1649	-0.01479	H4	-2.07454	-1.86938	-0.03504	H112	-0.72051	-2.07917	0.637
C30	-0.63535	-1.20481	-0.01481	H5	-1.93289	-1.73249	-0.02731	H113	-0.73632	-2.01406	0.63222
C31	-0.67493	-1.24954	-0.02371	H6	-1.92408	-1.54568	-0.04197	H114	-0.6713	-1.93278	0.62726
C32	-0.66994	-1.29181	-0.02389	H7	-1.71281	-1.46325	-0.03984	N56	-0.37786	-2.67016	0.62749
N33	-0.99841	-1.03912	-0.03364	H8	-1.78288	-1.54079	-0.04041	N56	0.62214	-1.67016	0.62749
N34	-0.78191	-0.99728	-0.0339	H9	-1.8684	-1.47655	-0.00588	N56	0.67016	-0.7077	0.62749
N35	-0.69067	-1.1287	-0.02607	H10	-1.79952	-1.40027	-0.00748	N56	-1.2923	-2.62214	0.62749
N36	-0.62685	-1.28755	-0.02395	C11	-2.00156	-1.95849	-0.03369	N56	-0.2923	-1.62214	0.62749
H109	-0.69308	-0.94454	-0.03469	C12	-1.95696	-1.82629	-0.0331	C74	-1.3/331	-2.70883	0.62752
H110	-0.61196	-0.92828	-0.038	C13	-1.9564	-1.86828	-0.03306	C74	-0.29117	-2.66448	0.62752
	0.02204	1 70402	-0.0370	C14 C15	2.00090	1 0110	-0.034	C74	0.22552	1 62660	0.02752
H2	-0.92394	-1.867/6	-0.03212	C15	-2.04230	-1.9119	-0.03478	C74	0.66448	-0.62669	0.02752
H3	-1 07609	-1 94382	-0.03489	C17	-1 99967	-1 82653	-0.03397	H64	-1 79169	-0.02003	0.02702
H4	-1.07454	-1.86938	-0.03504	C18	-1.96906	-1.74138	-0.03072	H65	-1.86551	-0.9213	0.2993
H5	-0.93289	-1.73249	-0.02731	N19	-1.94467	-1.65778	-0.03125	H66	-1.94028	-1.07214	0.30235
H6	-0.92408	-1.54568	-0.04197	C20	-1.98001	-1.70318	-0.03202	H67	-1.86607	-1.0714	0.30203
H7	-0.71281	-1.46325	-0.03984	C21	-2.02657	-1.71299	-0.03434	H68	-1.72941	-0.92907	0.29529
H8	-0.78288	-1.54079	-0.04041	C22	-2.03599	-1.67642	-0.03613	H69	-1.54183	-0.91917	0.30756
H9	-0.8684	-1.47655	-0.00588	C23	-1.99935	-1.63047	-0.03583	H70	-1.45865	-0.70855	0.33337
H10	-0.79952	-1.40027	-0.00748	C24	-1.95318	-1.62139	-0.03313	H71	-1.53643	-0.77838	0.33228
C11	-1.00156	-1.95849	-0.03369	C25	-1.91476	-1.57253	-0.03348	H72	-1.47361	-0.86339	0.2587
C12	-0.95696	-1.82629	-0.0331	C26	-1.74666	-1.46581	-0.03246	H73	-1.39701	-0.79478	0.2627
C13	-0.9564	-1.86828	-0.03306	C27	-1.78666	-1.51019	-0.0329	C82	-1.95428	-0.99733	0.30123
C14	-1.00098	-1.91335	-0.034	C28	-1.83129	-1.51536	-0.02488	C83	-1.82369	-0.95413	0.3002
C15	-1.04258	-1.9119	-0.03478	C29	-1.8351	-1.47484	-0.01479	C84	-1.86622	-0.95437	0.30043
C16	-1.04173	-1.86924	-0.03486	C30	-1./9519	-1.43054	-0.01481	C85	-1.90902	-0.99676	0.30142
017	-0.99967	-1.82653	-0.03397	031	1.75046	-1.42539	-0.023/1	086	1.90817	1.03883	0.30215
N10	-0.90900	-1.74130	-0.03072	N22	-1.70019	-1.3/013	-0.02389	C89	-1.000/9	-1.03052	0.30198
1117	0.07701	1.00110	0.00120	1400	1.1.00000	1.00020	0.00004	500	1.02024	0.00040	0.00030

 Table S1. Fractional atomic coordinates of unit cell and lattice parameters of Tta-Dfp.

C20	-0.98001	-1.70318	-0.03202	N34	-2.00272	-1.78463	-0.0339	C89	-1.73797	-0.96515	0.29759
C21	-1.02657	-1.71299	-0.03434	N35	-1.8713	-1.56197	-0.02607	N90	-1.65416	-0.94021	0.29802
C22	-1.03599	-1.67642	-0.03613	N36	-1.71245	-1.3393	-0.02395	C91	-1.69951	-0.97575	0.29803
C23	-0.99935	-1.63047	-0.03583	H109	-2.05546	-1.74854	-0.03469	C92	-1.70903	-1.02222	0.29887
C24	-0.95318	-1.62139	-0.03313	H110	-2.07172	-1.68368	-0.038	C93	-1.67229	-1.03138	0.29994
C25	-0.91476	-1.57253	-0.03348	H111	-2.00699	-1.60223	-0.0376	C94	-1.62644	-0.99458	0.30033
C26	-0.74666	-1.46581	-0.03246	C11	-1.95693	-1.99844	-0.03369	C95	-1.61761	-0.94848	0.29916
C27	-0.78666	-1.51019	-0.0329	C11	-0.95693	-1.99844	-0.03369	C96	-1.56878	-0.90993	0.30061
C28	-0.83129	-1.51536	-0.02488	C11	-1.00156	-0.95849	-0.03369	C97	-1.46159	-0.74225	0.31726
C29	-0.8351	-1.47484	-0.01479	C11	-3.04151	-2.04307	-0.03369	C98	-1.50608	-0.78211	0.31688
C30	-0.79519	-1.43054	-0.01481	C11	-2.04151	-3.04307	-0.03369	C99	-1.51164	-0.82655	0.2977
C31	-0.75046	-1.42539	-0.02371	C32	-1.66994	-2.29181	-0.02389	C100	-1.47146	-0.83028	0.2773
C32	-0.70819	-1.37813	-0.02389	C32	-1.66994	-1.29181	-0.02389	C101	-1.42704	-0.79054	0.27846
N33	-0.96088	-1.95929	-0.03364	C32	-2.70819	-2.37813	-0.02389	C102	-1.42147	-0.74601	0.29826
N34	-1.00272	-1.78463	-0.0339	C32	-1.62187	-3.33006	-0.02389	C103	-1.3/412	-0.70392	0.29867
N35	-0.8713	-1.56197	-0.02607	032	-0.62187	-1.33006	-0.02389	N105	-1.95525	-0.95672	0.30118
N36	-0.71245	-1.3393	-0.02395	N33	-1.99841	-2.03912	-0.03364	N106	-1.78109	-0.99908	0.30031
H109	-1.05540	-1.74804	-0.03469	N33	-0.99841	-2.03912	-0.03364	N107	-1.00032	-0.80044	0.29734
	1.00600	-1.00300	-0.030	NOO	-2.90000	-1.95929	-0.03364	NIUO	-1.3009	-0.00040	0.29071
	1 20507	1 1 2 0 0 1	-0.0370	N22	-0.90000	2 00150	-0.03304		1 67026	1 06706	0.29002
	1 12254	1 05592	-0.03212	N26	1 62695	2 29755	0.03304		1 50906	1 00202	0.30007
	1 05619	1 12227	0.03172	N26	1 62695	1 29755	-0.02395		2 07012	0 97092	0.30133
НА	-1.13062	-1.13227	-0.03409	N36	-1.02005	-2 3303	-0.02395	H65	-2.07913	-0.07002	0.29904
H5	-1.15002	-1 20010	-0.03304	N36	-2.7 1245	-3 3393	-0.02395	H66	-2.0707	-0.86814	0.2995
H6	-1 45432	-1.3784	-0.02701	N36	-0.6607	-1.37315	-0.02395	H67	-1 9286	-0.79467	0.30203
H7	-1 53675	-1 24956	-0.03984	H37	0 4017	-1 06409	0.63798	H68	-2 07093	-0.80034	0.29529
H8	-1.45921	-1.24209	-0.04041	H38	0.25863	-1.27491	0.63124	H69	-2.08083	-0.62266	0.30756
H9	-1.52345	-1.39185	-0.00588	H39	0.25941	-1.20069	0.62875	H70	-2.29145	-0.7501	0.33337
H10	-1.59973	-1.39925	-0.00748	H40	0.40971	-1.12641	0.65232	H71	-2.22162	-0.75805	0.33228
C11	-1.04151	-1.04307	-0.03369	H41	0.40924	-1.20026	0.65454	H72	-2.13661	-0.61022	0.2587
C12	-1.17371	-1.13067	-0.0331	C42	0.30852	-1.04372	0.63493	H73	-2.20522	-0.60223	0.2627
C13	-1.13172	-1.08812	-0.03306	C43	0.29935	-1.00698	0.63232	C82	-2.00267	-0.95695	0.30123
C14	-1.08665	-1.08763	-0.034	C44	0.33616	-0.96115	0.62961	C83	-2.04587	-0.86956	0.3002
C15	-1.0881	-1.13068	-0.03478	C45	0.38224	-0.95232	0.63011	C84	-2.04563	-0.91185	0.30043
C16	-1.13076	-1.17249	-0.03486	N46	0.39053	-0.98885	0.63271	C85	-2.00324	-0.91226	0.30142
C17	-1.17347	-1.17314	-0.03397	C47	0.35501	-1.03419	0.63486	C86	-1.96117	-0.86934	0.30215
C18	-1.25862	-1.22768	-0.03072	C48	0.36562	-1.07263	0.63705	C87	-1.96148	-0.82727	0.30198
N19	-1.34222	-1.28689	-0.03125	C49	0.29192	-1.24283	0.63533	C88	-2.00354	-0.82678	0.30093
C20	-1.29682	-1.27683	-0.03202	C50	0.29225	-1.20043	0.63383	C89	-2.03485	-0.77282	0.29759
C21	-1.28701	-1.31358	-0.03434	C51	0.33425	-1.1579	0.63909	N90	-2.05979	-0.71395	0.29802
C22	-1.32358	-1.35957	-0.03613	C52	0.37651	-1.15838	0.6468	C91	-2.02425	-0.72376	0.29803
C23	-1.36953	-1.36888	-0.03583	C53	0.37623	-1.20093	0.64821	C92	-1.9///8	-0.68681	0.29887
C24	-1.37861	-1.33179	-0.03313	054	0.3339	-1.243/1	0.64221	C93	-1.96862	-0.64091	0.29994
025	-1.42/4/	1.34223	-0.03348	U55	0.33329	-1.28897	0.64316	C94	-2.00542	-0.63186	0.30033
C26	1 40001	1 27647	-0.03240	N50	0.02214	-0.07010	0.6272	C95	-2.05152	-0.00913	0.29910
C22	-1.40901	-1.2/04/	-0.0329	N59	0.33107	-1.113/0	0.0372	C90	-2.09007	-0.00000	0.30001
C20	-1.40404	-1.31393	-0.02400	H50	0.37309	-0.80838	0.04310	C97	-2.23773	-0.7 1934	0.31720
C30	-1 56946	-1.36465	-0.01473	H60	0.40727	-0.000000	0.00002	C99	-2.21709	-0.68509	0.31000
C31	-1.57461	-1 32507	-0.02371	H61	0.0000	-0 79335	0.59316	C100	-2 16972	-0.64118	0.2377
C32	-1 62187	-1 33006	-0.02389	H62	0.55231	-0.87115	0.5951	C100	-2 20946	-0.6365	0.27846
N33	-1.04071	-1.00159	-0.03364	H63	0.41158	-0.87658	0.61819	C102	-2.25399	-0.67546	0.29826
N34	-1.21537	-1.21809	-0.0339	C74	0.62669	-0.70883	0.62752	C103	-2.29608	-0.6702	0.29867
N35	-1.43803	-1.30933	-0.02607	C75	0.50039	-0.80619	0.64973	N105	-2.04328	-0.99853	0.30118
N36	-1.6607	-1.37315	-0.02395	C76	0.5401	-0.76176	0.6481	N106	-2.00092	-0.78201	0.30031
H109	-1.25146	-1.30692	-0.03469	C77	0.58461	-0.75619	0.62814	N107	-2.13356	-0.69188	0.29734
H110	-1.31632	-1.38804	-0.038	C78	0.5884	-0.7963	0.60936	N108	-2.33954	-0.70844	0.29871
H111	-1.39777	-1.40476	-0.0376	C79	0.54857	-0.84081	0.61021	H115	-1.94874	-0.69323	0.29862
H1	-1.92394	-2.79403	-0.03212	C80	0.50414	-0.84637	0.6296	H116	-1.93294	-0.6123	0.30067
H2	-1.92328	-2.86746	-0.03172	C81	0.42079	-0.90353	0.62681	H117	-1.99797	-0.59603	0.30155
H3	-2.07609	-2.94382	-0.03489	N104	0.46427	-0.89305	0.63087	H64	-2.12918	-1.20831	0.29904
H4	-2.07454	-2.86938	-0.03504	H112	0.27949	-1.07917	0.637	H65	-2.05579	-1.13449	0.2993
H5	-1.93289	-2.73249	-0.02731	H113	0.26368	-1.01406	0.63222	H66	-2.13186	-1.05972	0.30235

H6	-1.92408	-2.54568	-0.04197	H114	0.3287	-0.93278	0.62726	H67	-2.20533 -1.13393	0.30203
H7	-1.71281	-2.46325	-0.03984	H37	0.06409	-1.53421	0.63798	H68	-2.19966 -1.27059	0.29529
H8	-1.78288	-2.54079	-0.04041	H38	0.27491	-1.46646	0.63124	H69	-2.37734 -1.45817	0.30756
H9	-1.8684	-2.47655	-0.00588	H39	0.20069	-1.5399	0.62875	H70	-2.2499 -1.54135	0.33337
H10	-1.79952	-2.40027	-0.00748	H40	0.12641	-1.46388	0.65232	H71	-2.24195 -1.46357	0.33228
C11	-2.00156	-2.95849	-0.03369	H41	0.20026	-1.3905	0.65454	H72	-2.38978 -1.52639	0.2587
C12	-1.95696	-2.82629	-0.0331	C42	0.04372	-1.64776	0.63493	H73	-2.39777 -1.60299	0.2627
C13	-1.9564	-2.86828	-0.03306	C43	0.00698	-1.69367	0.63232	C82	-2.04305 -1.04572	0.30123
C14	-2.00098	-2.91335	-0.034	C44	-0.03885	-1.70269	0.62961	C83	-2.13044 -1.17631	0.3002
C15	-2.04258	-2.9119	-0.03478	C45	-0.04768	-1.66544	0.63011	C84	-2.08815 -1.13378	0.30043
C16	-2.04173	-2.86924	-0.03486	N46	-0.01115	-1.62062	0.63271	C85	-2.08774 -1.09098	0.30142
C17	-1.99967	-2.82653	-0.03397	C47	0.03419	-1.6108	0.63486	C86	-2.13066 -1.09183	0.30215
C18	-1.96906	-2.74138	-0.03072	C48	0.07263	-1.56175	0.63705	C87	-2.17273 -1.13421	0.30198
N19	-1.94467	-2.65//8	-0.03125	C49	0.24283	-1.46525	0.63533	C88		0.30093
C20	-1.98001	-2.70318	-0.03202	050	0.20043	-1.50732	0.63383	C89	-2.22/18 -1.26203	0.29759
021	-2.02037	-2.71299	-0.03434	051	0.1579	-1.50785	0.63909	090	-2.28005 -1.34584	0.29802
022	-2.03599	-2.0/042	-0.03013	052	0.15838	-1.40511	0.0408	C91	-2.2/024 -1.30049	0.29803
C23	1 05210	-2.03047	-0.03003	C53	0.20093	1 42204	0.04021	C92	2.31319 -1.29097	0.29007
C24	1 01/76	2.02139	-0.03313	C54	0.24371	1 27774	0.04221	C93	2 26914 1 27256	0.29994
C25	1 74666	2.07200	-0.03340	055 NE6	0.20097	1 7077	0.04310	C94	2 2 2 2 0 0 14 - 1.37 3 3 0	0.30033
C20	1 79666	2.40001	0.03240	N50	0.32904	1 55257	0.02749	C95	2 24115 1 42122	0.29910
C28	-1.83120	-2.51019	-0.0329	N58	0.11370	-1.33615	0.0372	C90	-2.34113 -1.43122	0.30001
C20	-1.8351	-2.31330	-0.02400	H59	-0 10162	-1 72/35	0.66862	C98	-2 27603 -1 40302	0.31688
C30	-1 79519	-2.43054	-0.01481	H60	-0.26827	-1 73243	0.66373	C99	-2 31491 -1 48836	0.01000
C31	-1 75046	-2 42539	-0.02371	H61	-0.20027	-1 58454	0.59316	C100	-2 35882 -1 52854	0.2773
C32	-1.70819	-2.37813	-0.02389	H62	-0.12885	-1.57654	0.5951	C101	-2.3635 -1.57296	0.27846
N33	-1.96088	-2.95929	-0.03364	H63	-0.12342	-1.71184	0.61819	C102	-2.32454 -1.57853	0.29826
N34	-2.00272	-2.78463	-0.0339	C74	-0.29117	-1.66448	0.62752	C103	-2.3298 -1.62588	0.29867
N35	-1.8713	-2.56197	-0.02607	C75	-0.19381	-1.69342	0.64973	N105	-2.00147 -1.04475	0.30118
N36	-1.71245	-2.3393	-0.02395	C76	-0.23824	-1.69814	0.6481	N106	-2.21799 -1.21891	0.30031
H109	-2.05546	-2.74854	-0.03469	C77	-0.24381	-1.6592	0.62814	N107	-2.30812 -1.44168	0.29734
H110	-2.07172	-2.68368	-0.038	C78	-0.2037	-1.6153	0.60936	N108	-2.29156 -1.6311	0.29871
H111	-2.00699	-2.60223	-0.0376	C79	-0.15919	-1.61062	0.61021	H115	-2.30677 -1.25551	0.29862
H1	-1.20597	-2.12991	-0.03212	C80	-0.15363	-1.64949	0.6296	H116	-2.3877 -1.32064	0.30067
H2	-1.13254	-2.05582	-0.03172	C81	-0.09647	-1.67568	0.62681	H117	-2.40397 -1.40194	0.30155
H3	-1.05618	-2.13227	-0.03489	N104	-0.10695	-1.64268	0.63087	H64	-3.07913 -1.87082	0.29904
H4	-1.13062	-2.20516	-0.03504	H112	0.07917	-1.64134	0.637	H65	-3.0787 -1.94421	0.2993
H5	-1.26751	-2.2004	-0.02731	H113	0.01406	-1.72226	0.63222	H66	-2.92786 -1.86814	0.30235
H6	-1.45432	-2.3784	-0.04197	H114	-0.06722	-1.73852	0.62726	H67	-2.9286 -1.79467	0.30203
H7	-1.53675	-2.24956	-0.03984	H37	0.53421	-1.4017	0.63798	H68	-3.07093 -1.80034	0.29529
H8	-1.45921	-2.24209	-0.04041	H38	0.46646	-1.25863	0.63124	H69	-3.08083 -1.62266	0.30756
H9	-1.52345	-2.39185	-0.00588	H39	0.5399	-1.25941	0.62875	H70		0.33337
H10	-1.59973	-2.39925	-0.00748	H40	0.46388	-1.40971	0.65232	H/1	-3.22162 -1.75805	0.33228
011	1 1 1 2 2 7 4	-2.04307	-0.03369	H41	0.3905	1 20952	0.05454	H/2	-3.13001 -1.01022	0.2587
C12	-1.1/3/1	-2.1300/	-0.0331	C42	0.04770	-1.30032	0.03493	C22	-3.20322 -1.00223	0.2021
C14	-1.13172	-2.00012	-0.03300	C43	0.09307	-1.29900	0.03232	C82	-3.00207 -1.95095	0.30123
C15	-1.00000	-2 13068	-0.03478	C45	0.66544	-1 38224	0.02301	C84	-3.04563 -1.91185	0.0002
C16	-1 13076	-2.10000	-0.03486	N46	0.00044	-1 39053	0.63271	C85	-3 00324 -1 91226	0.30142
C17	-1 17347	-2 17314	-0.03397	C47	0.6108	-1 35501	0.63486	C86	-2 96117 -1 86934	0.30215
C18	-1.25862	-2.22768	-0.03072	C48	0.56175	-1.36562	0.63705	C87	-2.96148 -1.82727	0.30198
N19	-1.34222	-2.28689	-0.03125	C49	0.46525	-1.29192	0.63533	C88	-3.00354 -1.82678	0.30093
C20	-1.29682	-2.27683	-0.03202	C50	0.50732	-1.29225	0.63383	C89	-3.03485 -1.77282	0.29759
C21	-1.28701	-2.31358	-0.03434	C51	0.50785	-1.33425	0.63909	N90	-3.05979 -1.71395	0.29802
C22	-1.32358	-2.35957	-0.03613	C52	0.46511	-1.37651	0.6468	C91	-3.02425 -1.72376	0.29803
C23	-1.36953	-2.36888	-0.03583	C53	0.42284	-1.37623	0.64821	C92	-2.97778 -1.68681	0.29887
C24	-1.37861	-2.33179	-0.03313	C54	0.42239	-1.3339	0.64221	C93	-2.96862 -1.64091	0.29994
C25	-1.42747	-2.34223	-0.03348	C55	0.37774	-1.33329	0.64316	C94	-3.00542 -1.63186	0.30033
C26	-1.53419	-2.28085	-0.03246	N56	0.7077	-1.62214	0.62749	C95	-3.05152 -1.66913	0.29916
C27	-1.48981	-2.27647	-0.0329	N57	0.55257	-1.33167	0.6372	C96	-3.09007 -1.65885	0.30061
C28	-1.48464	-2.31593	-0.02488	N58	0.33615	-1.37389	0.64318	C97	-3.25775 -1.71934	0.31726
C29	-1.52516	-2.36026	-0.01479	H59	0.72435	-1.46727	0.66862	C98	-3.21789 -1.72397	0.31688
C30	-1.56946	-2.36465	-0.01481	H60	0.73243	-1.53584	0.66373	C99	-3.17345 -1.68509	0.2977

C31	-1.57461	-2.32507	-0.02371	H61	0.58454	-1.62211	0.59316	C100	-3.16972	-1.64118	0.2773
C32	-1.62187	-2.33006	-0.02389	H62	0.57654	-1.55231	0.5951	C101	-3.20946	-1.6365	0.27846
N33	-1.04071	-2.00159	-0.03364	H63	0.71184	-1.41158	0.61819	C102	-3.25399	-1.67546	0.29826
N34	-1.21537	-2.21809	-0.0339	C74	0.66448	-1.62669	0.62752	C103	-3.29608	-1.6702	0.29867
N35	-1.43803	-2.30933	-0.02607	C75	0.69342	-1.50039	0.64973	N105	-3.04328	-1.99853	0.30118
N36	-1.6607	-2.37315	-0.02395	C76	0.69814	-1.5401	0.6481	N106	-3.00092	-1.78201	0.30031
H109	-1.25146	-2.30692	-0.03469	C77	0.6592	-1.58461	0.62814	N107	-3.13356	-1.69188	0.29734
H110	-1.31632	-2.38804	-0.038	C78	0.6153	-1.5884	0.60936	N108	-3.33954	-1.70844	0.29871
H111	-1.39777	-2.40476	-0.0376	C79	0.61062	-1.54857	0.61021	H115	-2.94874	-1.69323	0.29862
H1	-1.87009	-3.07606	-0.03212	C80	0.64949	-1.50414	0.6296	H116	-2.93294	-1.6123	0.30067
H2	-1.94418	-3.07672	-0.03172	C81	0.67568	-1.42079	0.62681	H117	-2.99797	-1.59603	0.30155
H3	-1.86773	-2.92391	-0.03489	N104	0.64268	-1.46427	0.63087	H64	-3.12918	-2.20831	0.29904
H4	-1.79484	-2.92546	-0.03504	H112	0.64134	-1.27949	0.637	H65	-3.05579	-2.13449	0.2993
H5	-1.7996	-3.06711	-0.02731	H113	0.72226	-1.26368	0.63222	H66	-3.13186	-2.05972	0.30235
H6	-1.6216	-3.07592	-0.04197	H114	0.73852	-1.3287	0.62726	H67	-3.20533	-2.13393	0.30203
H7	-1.75044	-3.28719	-0.03984	H37	-0.93591	-2.53421	0.63798	H68	-3.19966	-2.27059	0.29529
H8	-1.75791	-3.21712	-0.04041	H38	-0.72509	-2.46646	0.63124	H69	-3.37734	-2.45817	0.30756
H9	-1.60815	-3.1316	-0.00588	H39	-0.79931	-2.5399	0.62875	H70	-3.2499	-2.54135	0.33337
H10	-1.60075	-3.20048	-0.00748	H40	-0.87359	-2.46388	0.65232	H71	-3.24195	-2.46357	0.33228
C11	-1.95693	-2.99844	-0.03369	H41	-0.79974	-2.3905	0.65454	H72	-3.38978	-2.52639	0.2587
C12	-1.86933	-3.04304	-0.0331	C42	-0.95628	-2.64776	0.63493	H73	-3.39777	-2.60299	0.2627
C13	-1.91188	-3.0436	-0.03306	C43	-0.99302	-2.69367	0.63232	C82	-3.04305	-2.04572	0.30123
C14	-1.91237	-2.99902	-0.034	C44	-1.03885	-2.70269	0.62961	C83	-3.13044	-2.17631	0.3002
C15	-1.86932	-2.95742	-0.03478	C45	-1.04768	-2.66544	0.63011	C84	-3.08815	-2.13378	0.30043
C16	-1.82751	-2.95827	-0.03486	N46	-1.01115	-2.62062	0.63271	C85	-3.08774	-2.09098	0.30142
C17	-1.82686	-3.00033	-0.03397	C47	-0.96581	-2.6108	0.63486	C86	-3.13066	-2.09183	0.30215
C18	-1.77232	-3.03094	-0.03072	C48	-0.92737	-2.56175	0.63705	C87	-3.1/2/3	-2.13421	0.30198
N19	-1./1311	-3.05533	-0.03125	C49	-0.75717	-2.46525	0.63533	000	-3.17322	-2.1/6/6	0.30093
020	-1.72317	-3.01999	-0.03202	050	-0.79957	-2.50732	0.63383	C89	-3.22718	-2.26203	0.29759
021	-1.68642	-2.97343	-0.03434	051	-0.8421	-2.50785	0.63909	N90	-3.28605	-2.34584	0.29802
022	-1.04043	-2.96401	-0.03013	052	-0.84162	-2.40511	0.0408	091	-3.2/024	-2.30049	0.29803
023	-1.03112	-3.00005	-0.03583	053	-0.79907	-2.42284	0.04821	092	-3.31319	-2.29097	0.29887
024	-1.66821	-3.04682	-0.03313	054	-0.75629	-2.42239	0.64221	C93	-3.35909	-2.32771	0.29994
C25	-1.00///	-3.08524	-0.03348	U00	-0.71103	-2.3/1/4	0.04310	C94	-3.30814	-2.3/300	0.30033
C20	1 72252	2 21224	-0.03240	N50	-1.32904	-2.1011	0.6272	C95	-3.33007	-2.30239	0.29910
C20	1 60407	2 16071	-0.0329	NEO	-0.00424	-2.00207	0.0372	C90	2 20066	2.43122	0.30001
C20	1 62074	2 16/0	0.02400	1150	1 10162	2.33013	0.04310	C97	-3.20000	2.00041	0.31720
C29	1 62525	2 20/121	0.01479	H59 H60	1 26927	2.72433	0.00002	C90	-3.27003	2.49392	0.31000
C31	-1.67/03	-3.20401	-0.01401	H61	-1.20027	-2.73243	0.00070	C100	-3 35882	-2.40030	0.2311
C32	-1 66001	-3 20181	-0.02371	H62	-1 12885	-2.50-54	0.53510	C100	-3 3635	-2.52004	0.278/6
N33	-1.00334	-3.03012	-0.02003	H63	-1 123/2	-2.37034	0.0001	C102	-3.30000	-2.57250	0.20826
N34	-1 78191	-2 99728	-0.0000-	C74	-1 20117	-2.66448	0.62752	C102	-3 3298	-2 62588	0.20020
N35	-1 69067	-3 1287	-0.02607	C75	-1 19381	-2 69342	0.64973	N105	-3.00147	-2 04475	0.30118
N36	-1 62685	-3 28755	-0.02395	C76	-1 23824	-2 69814	0.6481	N106	-3 21799	-2 21891	0.30031
H109	-1.69308	-2.94454	-0.03469	C77	-1.24381	-2,6592	0.62814	N107	-3.30812	-2.44168	0.29734
H110	-1.61196	-2.92828	-0.038	C78	-1.2037	-2.6153	0.60936	N108	-3.29156	-2.6311	0.29871
H111	-1.59524	-2.99301	-0.0376	C79	-1.15919	-2.61062	0.61021	H115	-3.30677	-2.25551	0.29862
H1	-2.20597	-2.12991	-0.03212	C80	-1.15363	-2.64949	0.6296	H116	-3.3877	-2.32064	0.30067
H2	-2.13254	-2.05582	-0.03172	C81	-1.09647	-2.67568	0.62681	H117	-3.40397	-2.40194	0.30155
H3	-2.05618	-2.13227	-0.03489	N104	-1.10695	-2.64268	0.63087	H64	-2.79169	-1.92087	0.29904
H4	-2.13062	-2.20516	-0.03504	H112	-0.92083	-2.64134	0.637	H65	-2.86551	-1.9213	0.2993
H5	-2.26751	-2.2004	-0.02731	H113	-0.98594	-2.72226	0.63222	H66	-2.94028	-2.07214	0.30235
H6	-2.45432	-2.3784	-0.04197	H114	-1.06722	-2.73852	0.62726	H67	-2.86607	-2.0714	0.30203
H7	-2.53675	-2.24956	-0.03984	H37	-0.46579	-2.4017	0.63798	H68	-2.72941	-1.92907	0.29529
H8	-2.45921	-2.24209	-0.04041	H38	-0.53354	-2.25863	0.63124	H69	-2.54183	-1.91917	0.30756
H9	-2.52345	-2.39185	-0.00588	H39	-0.4601	-2.25941	0.62875	H70	-2.45865	-1.70855	0.33337
H10	-2.59973	-2.39925	-0.00748	H40	-0.53612	-2.40971	0.65232	H71	-2.53643	-1.77838	0.33228
C11	-2.04151	-2.04307	-0.03369	H41	-0.6095	-2.40924	0.65454	H72	-2.47361	-1.86339	0.2587
C12	-2.17371	-2.13067	-0.0331	C42	-0.35224	-2.30852	0.63493	H73	-2.39701	-1.79478	0.2627
C13	-2.13172	-2.08812	-0.03306	C43	-0.30633	-2.29935	0.63232	C82	-2.95428	-1.99733	0.30123
C14	-2.08665	-2.08763	-0.034	C44	-0.29731	-2.33616	0.62961	C83	-2.82369	-1.95413	0.3002
C15	-2.0881	-2.13068	-0.03478	C45	-0.33456	-2.38224	0.63011	C84	-2.86622	-1.95437	0.30043
C16	-2.13076	-2.17249	-0.03486	N46	-0.37938	-2.39053	0.63271	C85	-2.90902	-1.99676	0.30142

047	0 47047	0 4 7 0 4 4	0 0 0 0 0 7	047	0.0000	0.05504	0.00400	0.00	0.00047	0,00000	0.00045
C17	2.1/34/	2.17314	-0.03397	C47	-0.3092	2.30001	0.03400		2.90017	2.03003	0.30213
N10	2.20002	2.22700	-0.03072	C40	0.43025	2.30302	0.03705		2.00019	1 00646	0.30190
C20	2.34222	2.20009	-0.03125	C49	-0.55475	2.29192	0.03033		2.02324	1 06515	0.30093
C20	2.29002	2 21259	-0.03202	C50	0.49200	2 22/25	0.00000	NOO	2.13191	1 04021	0.29139
C22	2 2 2 2 2 2 5 9	2 25057	0.03434	C51	0.5349213	2.33423	0.03909	C01	2 60051	1 07575	0.29002
C22	2 26052	2.33937	0.03013	C52	0.57716	2.37031	0.0400	C91	2 70002	2 02222	0.29003
C23	-2.30933	-2.30000	-0.03303	C53	-0.57761	-2.37023	0.04021	C92	-2.70903	-2.02222	0.29007
C24	-2.37001	-2.33173	-0.03313	C55	-0.62226	-2.3333	0.04221	C94	-2.07223	-1 00/58	0.29994
C26	-2.72/7/	-2.3+223	-0.03246	N56	-0.02220	-2.00020	0.0-010	C95	-2.02044	-1.00-00	0.00000
C27	-2.00+10	-2.20003	-0.03240	N57	-0.2323	-2.02214	0.02743	C96	-2.56878	-1 00003	0.23310
C28	-2.40301	-2.21041	-0.0023	N58	-0.66385	-2.33107	0.0072	C97	-2.00070	-1 7/225	0.30001
C29	-2.52516	-2.31000	-0.02+00	H59	-0.27565	-2.07003	0.66862	C98	-2.50608	-1.79211	0.31688
C30	-2.52010	-2.00020	-0.01481	H60	-0.26757	-2.53584	0.66373	C99	-2.500000	-1 82655	0.01000
C31	-2.50340	-2.30+03	-0.01401	H61	-0.41546	-2.00004	0.50316	C100	-2.31104	-1.83028	0.2377
C32	-2.62187	-2.02007	-0.02071	H62	-0.47346	-2.52211	0.5951	C101	-2.47140	-1 79054	0.27846
N33	-2 04071	-2 00159	-0.03364	H63	-0 28816	-2 41158	0.61819	C102	-2 42147	-1 74601	0.29826
N34	-2 21537	-2 21809	-0.0339	C74	-0.33552	-2 62669	0.62752	C103	-2.37412	-1 70392	0.20020
N35	-2.43803	-2.30933	-0.02607	C75	-0.30658	-2.50039	0.64973	N105	-2.95525	-1.95672	0.30118
N36	-2.6607	-2.37315	-0.02395	C76	-0.30186	-2.5401	0.6481	N106	-2.78109	-1.99908	0.30031
H109	-2.25146	-2.30692	-0.03469	C77	-0.3408	-2.58461	0.62814	N107	-2.55832	-1.86644	0.29734
H110	-2.31632	-2.38804	-0.038	C78	-0.3847	-2.5884	0.60936	N108	-2.3689	-1.66046	0.29871
H111	-2.39777	-2.40476	-0.0376	C79	-0.38938	-2.54857	0.61021	H115	-2.74449	-2.05126	0.29862
H1	-2.87009	-2.07606	-0.03212	C80	-0.35051	-2.50414	0.6296	H116	-2.67936	-2.06706	0.30067
H2	-2.94418	-2.07672	-0.03172	C81	-0.32432	-2.42079	0.62681	H117	-2.59806	-2.00203	0.30155
H3	-2.86773	-1.92391	-0.03489	N104	-0.35732	-2.46427	0.63087	C103	-3.37412	-1.70392	0.29867
H4	-2.79484	-1.92546	-0.03504	H112	-0.35866	-2.27949	0.637	C103	-2.37412	-0.70392	0.29867
H5	-2.7996	-2.06711	-0.02731	H113	-0.27774	-2.26368	0.63222	C103	-3.29608	-2.6702	0.29867
H6	-2.6216	-2.07592	-0.04197	H114	-0.26148	-2.3287	0.62726	C103	-2.29608	-1.6702	0.29867
H7	-2.75044	-2.28719	-0.03984	H37	-0.5983	-2.06409	0.63798	C103	-1.3298	-0.62588	0.29867
H8	-2.75791	-2.21712	-0.04041	H38	-0.74137	-2.27491	0.63124	N108	-3.3689	-2.66046	0.29871
H9	-2.60815	-2.1316	-0.00588	H39	-0.74059	-2.20069	0.62875	N108	-2.33954	-1.70844	0.29871
H10	-2.60075	-2.20048	-0.00748	H40	-0.59029	-2.12641	0.65232	N108	-1.33954	-0.70844	0.29871
C11	-2.95693	-1.99844	-0.03369	H41	-0.59076	-2.20026	0.65454	N108	-3.29156	-1.6311	0.29871
C12	-2.86933	-2.04304	-0.0331	C42	-0.69148	-2.04372	0.63493	N108	-2.29156	-0.6311	0.29871
C13	-2.91188	-2.0436	-0.03306	C43	-0.70065	-2.00698	0.63232				
C14	-2.91237	-1.99902	-0.034	C44	-0.66384	-1.96115	0.62961				



Fig. S3 FT-IR spectra of Tta, Dfp and Tta-Dfp.<sup>1</sup>



**Fig. S4** a) <sup>13</sup>C CPMAS NMR spectrum of **Tab-Dfp** b) Structural building unit of **Tab-Dfp**; asterisks denote spinning sidebands or trapped trace of washing solvent (DMAc) due to the microporous network.<sup>1</sup>



**Fig. S5** a) <sup>13</sup>C CPMAS NMR spectrum of **Tta-Dfp** b) Structural building unit of **Tta-Dfp**; asterisks denote spinning sidebands or trapped trace of washing solvent (DMAc) due to the microporous network.<sup>1</sup>



Fig. S6 UV-Vis spectra of Tta- Dfp.<sup>1</sup>



Fig. S7 TGA of Tta- Dfp.<sup>1</sup>



Fig. S8 BET N<sub>2</sub> adsorption/desorption of Tta-Dfp.<sup>1</sup>



Fig. S9 Pore-size distribution profile of Tta-Dfp.<sup>1</sup>



Fig. S10 BET N<sub>2</sub> adsorption/desorption of Tab-Dfp.<sup>1</sup>



Fig. S11 Pore-size distribution profile of Tab-Dfp.<sup>1</sup>



Fig. S12 FE-SEM images for (a) Tta-Dfp and (b) Tab-Dfp.



<b>Table S2.</b> Elemental composition (weight percentage) analysis of all the elements.										
Element s	EDS		CHN-O							
	Tta-Dfp	Tab-Dfp	Tta-Dfp	Tab-Dfp						
С	81	90.9	82.5	89.5						
N	19	9.1	17.5	10.5						

Fig. S13 Elemental dot mapping images for (a) C, (b) N in Tta-Dfp and (c) C, (d) N in Tab-Dfp.



Fig. S14A LSV curves for the comparison of Tta-Dfp and Tab-Dfp with the benchmark CER electrocatalysts ( $RuO_2$  and  $Rh_xS_v/C$ )



Fig. S14B Tafel slopes extracted from Fig. 2a for Tta-Dfp and Tab-Dfp.



Fig. S15 Nyquists plot for Tta-Dfp and Tab-Dfp.



Fig. S16 Equivalent circuit obtained after fitting Nyquist plots of Tta-Dfp.

Table S3. EIS analysis (R <sub>ct</sub> and rate constant) of Tta-Dfp and Tab-Dfp during CER in 0.4 M HCI.									
Catalyst	<i>R</i> <sub>s</sub> (Ω, Solution resistance)	$R_p(\Omega, \text{Polarization} $ resistance)	<i>R<sub>ct</sub></i> (Ω, Charge- transfer resistance)	Rate constant, k (x 10 <sup>-5</sup> cm s <sup>-1</sup> )					
Tta-Dfp	13.70	203.9	190.20	1.40					
Tab-Dfp	0.91	235.91	235.00	1.13					



**Fig. S17** (a), (b) Cyclic voltammograms in the non-faradaic region for CER at different scan rates (20 mV s<sup>-1</sup> to 300 mV s<sup>-1</sup>) and (c), (d) corresponding current density *vs.* scan rate curves for the measurement of  $C_{dl}$ , ECSA and SSA.

Table S4. C <sub>dl,</sub> ECSA and SSA of Tta-Dfp and Tab-Dfp in 0.4 M HCl.									
Catalyst C <sub>dl</sub> * (μF) at 0.7 V vs. RHE ECSA (mm <sup>2</sup> ) SSA (mm <sup>2</sup> /mg)									
Tta-Dfp	1.08	2.70	22.5						
Tab-Dfp	0.989	2.47	20.5						

#### Quantification of chlorine gas: lodometric titration

The amount of chlorine (Cl<sub>2</sub>) gas evolved was estimated by performing the chronoamperometry for 60 s at 1.5 V vs. RHE in 0.4 M HCl. Further, the obtained Cl<sub>2</sub> was trapped by the immediate addition of an excess amount of potassium iodide (KI), and the solution was kept to attain equilibrium. The obtained amount was determined by performing lodometric titration versus 0.1 M Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and the following reactions will take place.

 $Cl_2 + K \rightarrow l_3 + Cl (Equation 1)$ 

 $I_3^{-}+2 S_2 O_3^{2-} \rightarrow 3I^{-}+S_4 O_6^{2-}$  (Equation 2)

### Calculation

Total volume of 0.1 M Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> consumed during titration = 2.1mL

Volume taken in the flask (0.4 M HCl and excess KI) = 30 mL

On applying the molarity equation:

$$M_1 * V_1 = M_2 * V_2$$

 $0.1 M * 2.1 mL = M_2 * 30 mL$ 

 $M_2 = 0.007 M (I_3)$ 

As can be seen in the stoichiometric equation two thio molecules are required for one  $(I_3^-)$ , therefore the concentration of thio would be equal to = 0.0035 M or mol/I.

Further, the amount of evolved chlorine was used to calculate the Faradaic efficiency as:

$$Q_{theo} = \frac{i * t}{nF}$$
$$Q_{theo} = \frac{22 \text{ mA.s}}{2 * 96500 \text{ C/mol}} = 1.1 * 10^{-7} \text{ mol}$$

Upon converting the obtained value in mol/l considering the 30 ml of volume taken to evolve the chlorine:  $0.0038* 10^{-3}$  mol/l

F.E. (%) = 
$$\frac{Q_{exp}}{Q_{theo}}$$
  
F.E. (%) =  $\frac{0.0035 \text{ mol/l}}{0.0038 \text{ mol } l} = 92\%$ 



**Fig. S18** (a) LSV curves recorded CER of **Tta-Dfp** for 200 cycles at a scan rate of 50 mV s<sup>-1</sup> in 0.4 HCl and (b) chronoamperometry curve in 0.4 HCl at 1.5 V *vs.* RHE for 50 h.



**Fig. S19A** RDE polarization curve (a) **Tab-Dfp**, (b) **Tta-Dfp** and K-L plots derived for (c) **Tab-Dfp**, (d) **Tta-Dfp** catalysts from respective RDE polarization curves at various rotation rates at different potentials (0.05 V, 0.00 V and -0.05 V *vs.* RHE).



Fig. S19B Chronoamperometry analysis performed for 10 minutes of operation @0 V and 2 minutes of shutdown (i.e. system off) for  $RuO_2$  and  $Rh_xS_y/C$ .



Fig. S20 Approach curve (I vs. tip substrate distance) for tip positioning.



**Fig. S21** 2D SECM images of **Tta-Dfp** and **Tab-Dfp** catalyst spot at a reduction potential of 0.5 *vs.* RHE



**Fig. S22** RC-SECM x-line scan of the catalyst's spots polarized at different potentials (-0.1, 0.0, 0.2 and 0.5 V vs. RHE).



**Fig. S23** (a) Sequential chronoamperometric curves at Pt tip in 0.4 M HCl in the potential range of 0.15 V to 0.50 V *vs.* Ag/AgCl/3 M KCl with an interval of 50 mV and (b) Absolute current difference *vs.* applied potential obtained from Figure S23a.



**Fig. S24** Optical microscopic images before SECM analysis (a) **Tab-Dfp**, (b) **Tta-Dfp**, (c) Pt/C (20%) and after SECM analysis (a) **Tab-Dfp**, (b) **Tta-Dfp**, (c) Pt/C (20%).



**Fig. S25** (a) LSV under full cell conditions (CER-ODC) with **Tta-Dfp** employed at both anode and cathode, (b) chronopotentiometry curves for CER-HER and CER-ODC cells at 10 mA cm<sup>-2</sup> for 12 h each.

#### References

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