

## Supporting Information for

### **Metallic Cobalt/Cobalt Sulfide Hetero-nanostructures Embedded within N-doped Graphitic Carbon Nanocage for Hydrogen Evolution Reaction**

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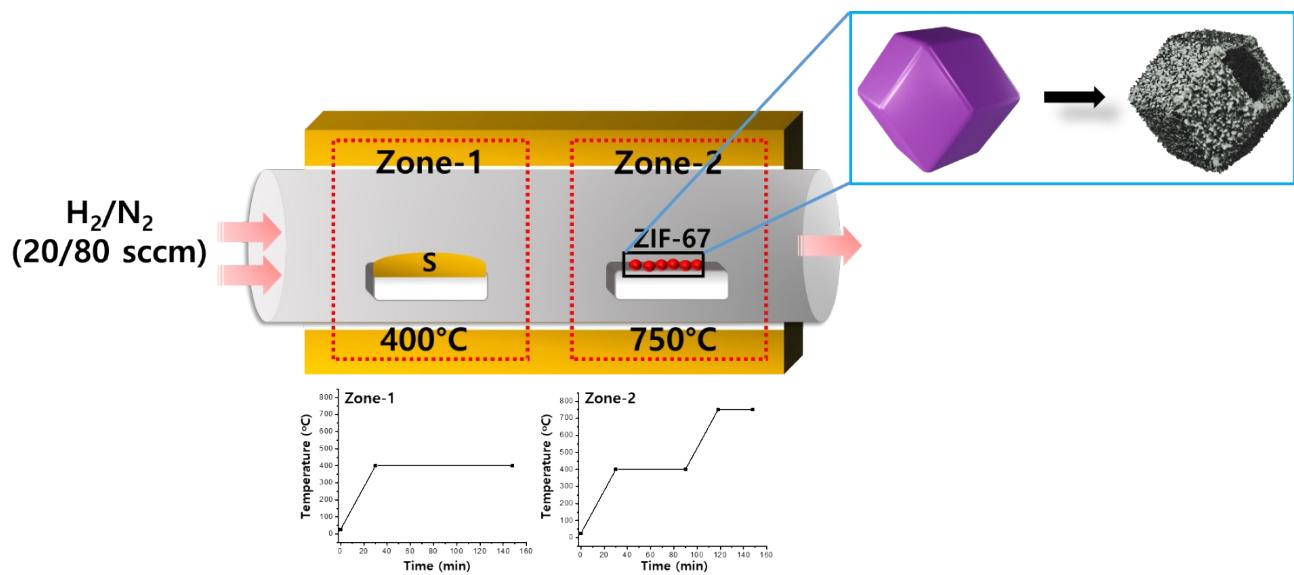
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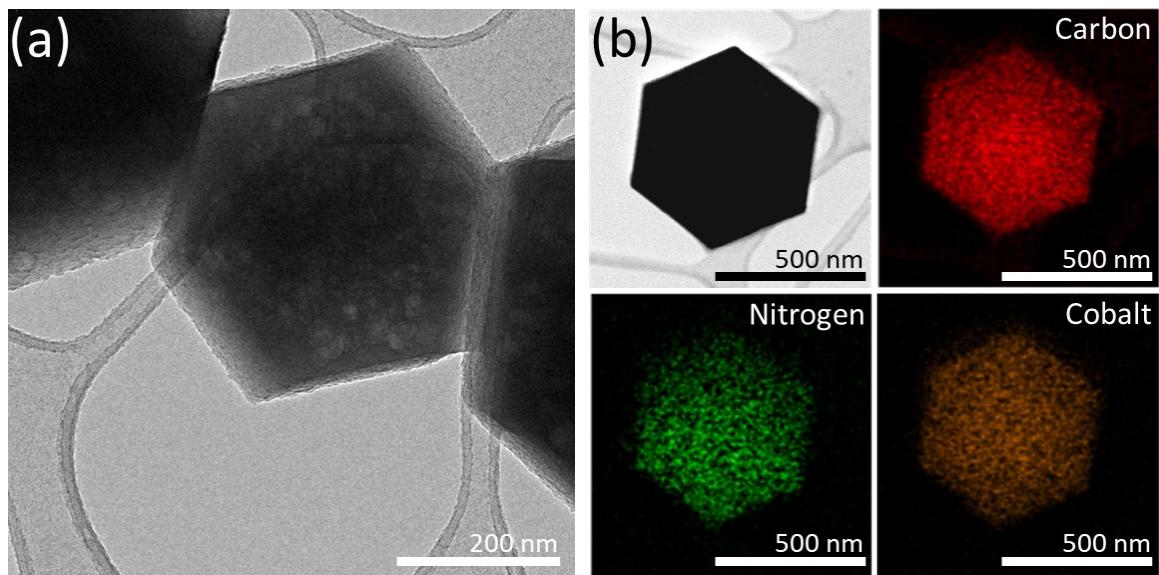
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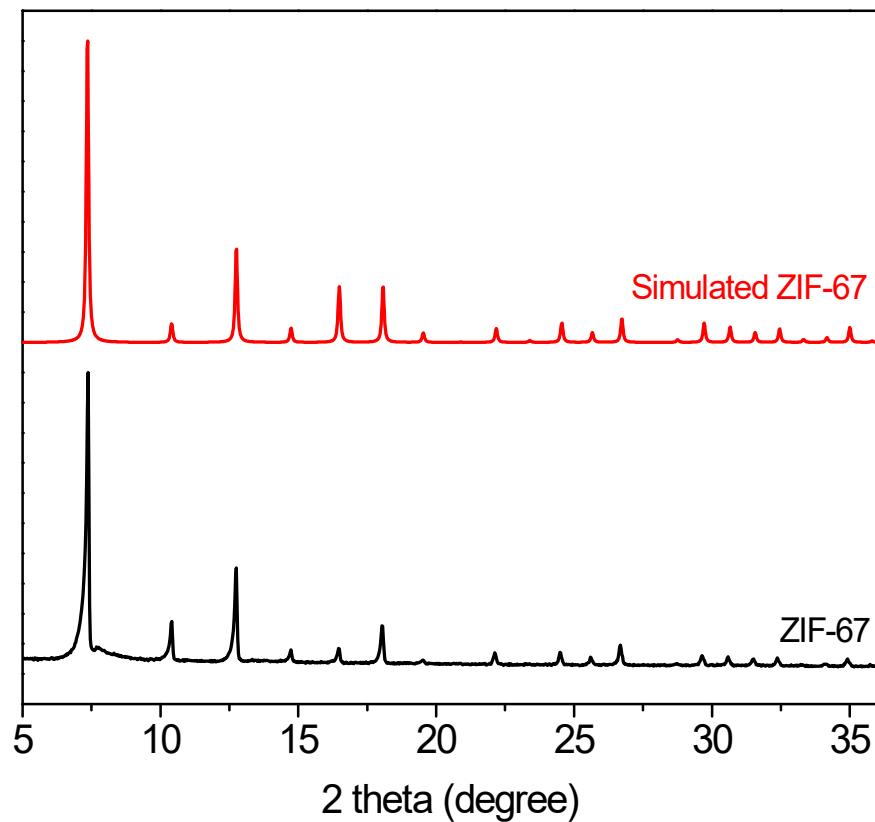
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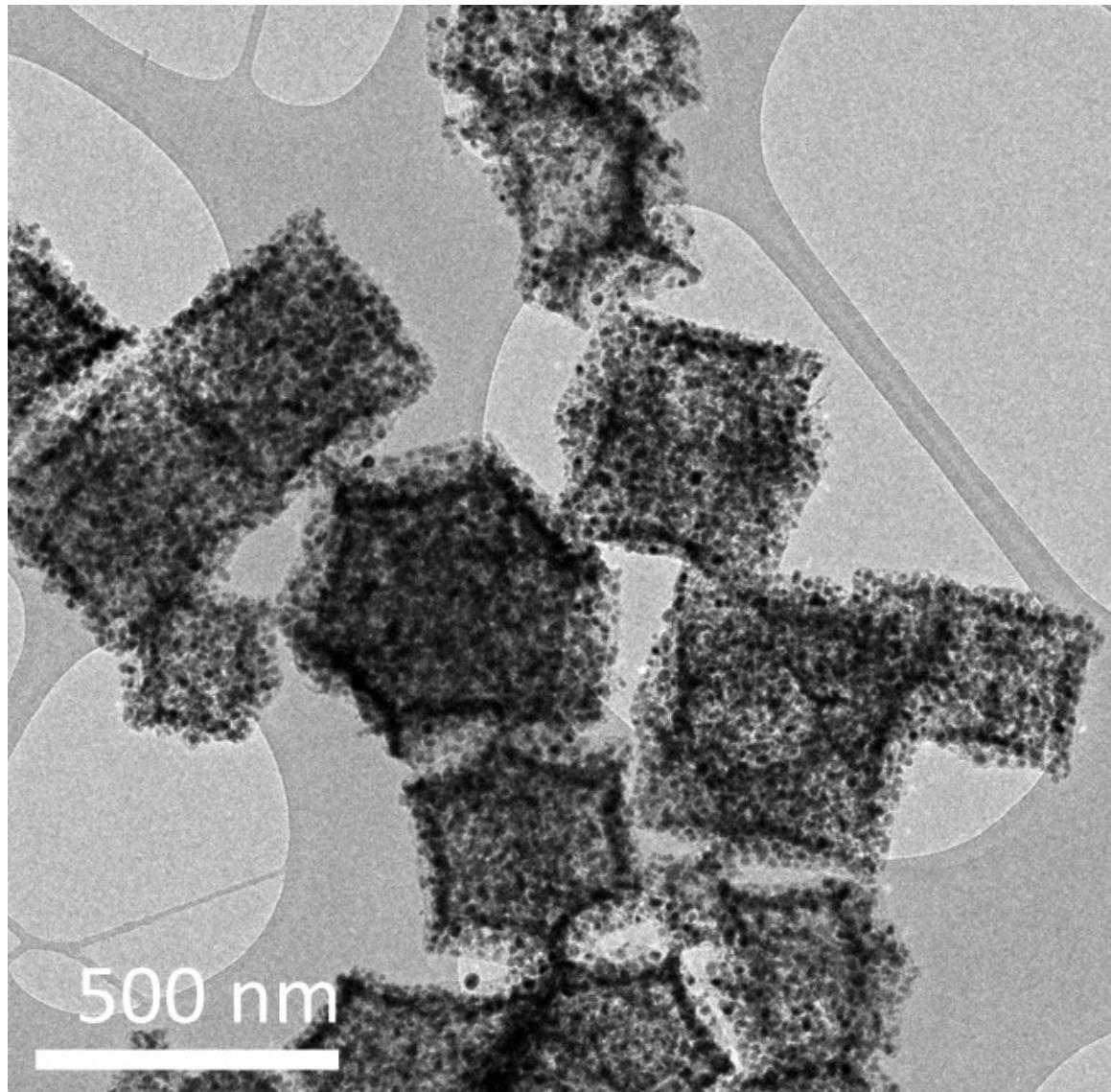
**Figure S1.** Schematic of fabrication of hollow nanocages composed of metallic cobalt/cobalt sulfide heterostructures embedded within N-doped graphitic carbon (Co/Co<sub>x</sub>S<sub>y</sub>@NC-750) through thermal annealing at 750 °C in the presence of H<sub>2</sub> and S.



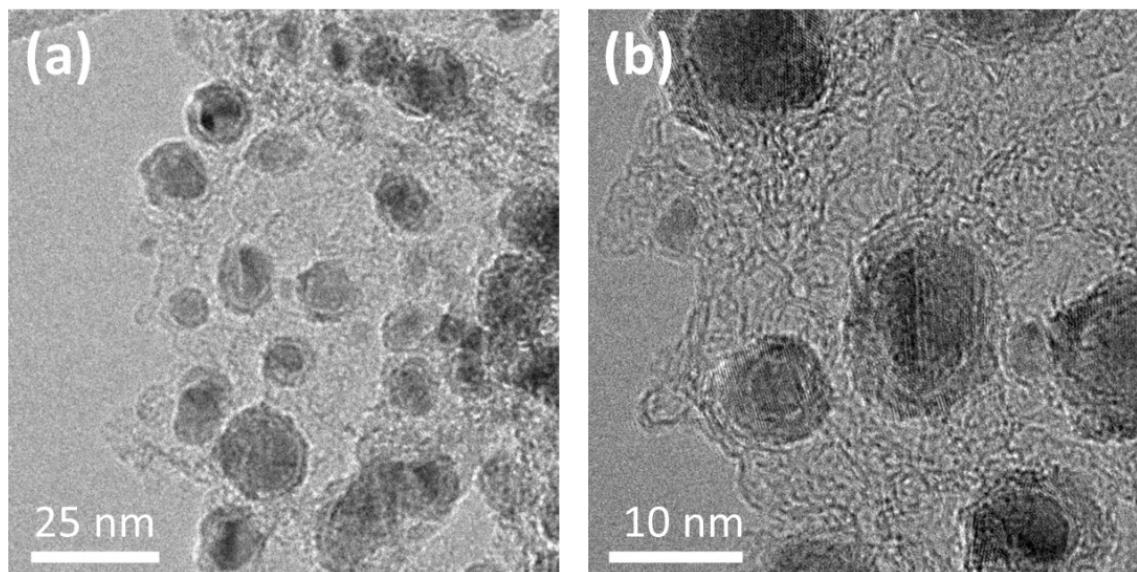
**Figure S2.** (a) TEM and (b) EDX mapping of ZIF-67 with C (red), N (green), and Co (orange).



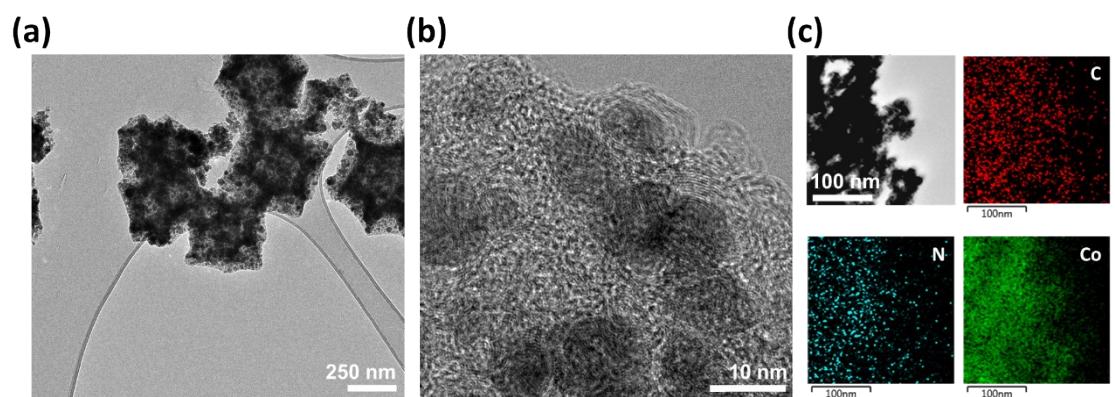
**Figure S3.** PXRD of ZIF-67 and its simulated pattern. The simulated PXRD data using Mercury 3.3 program.



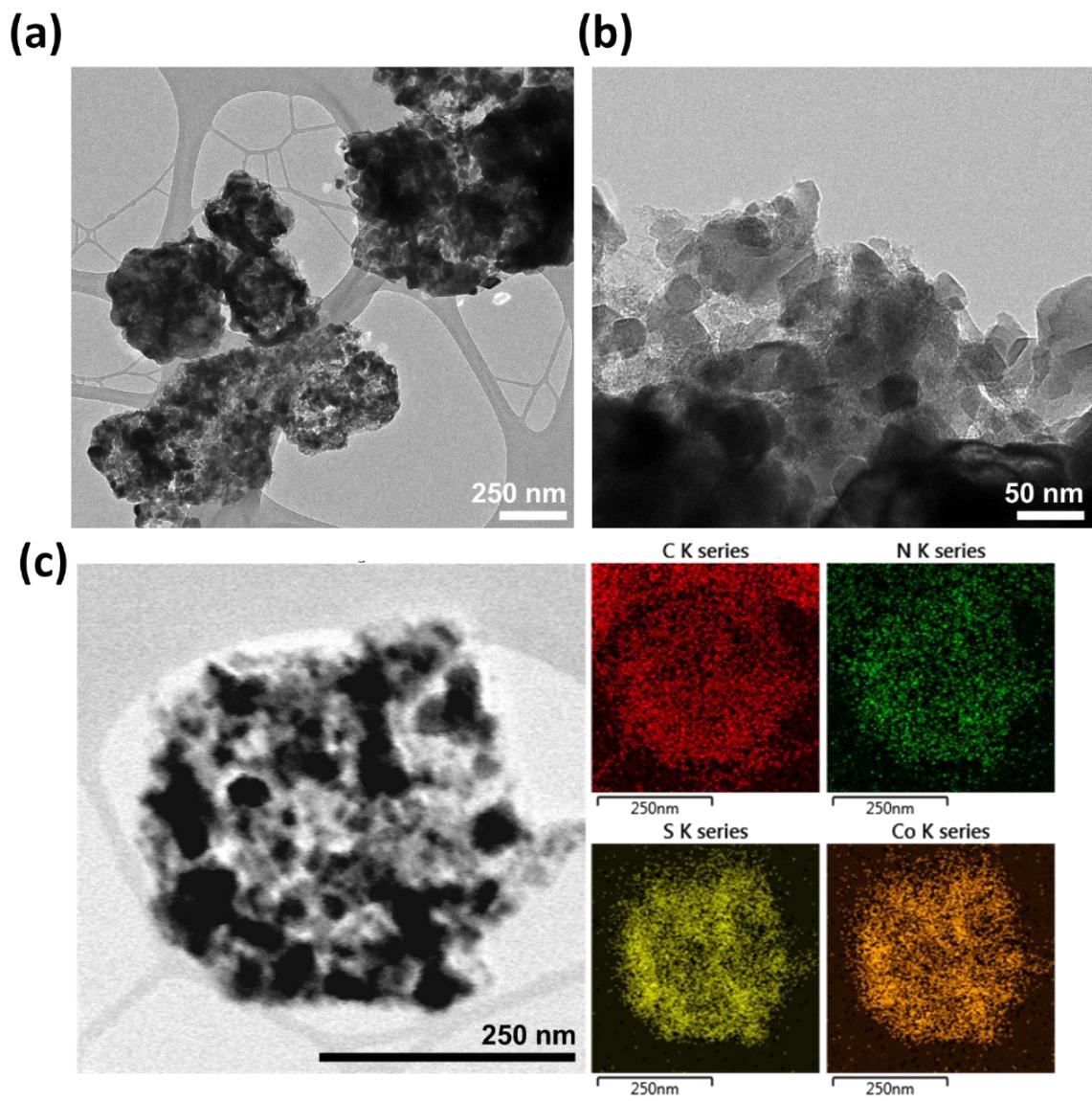
**Figure S4.** TEM of Co/Co<sub>x</sub>S<sub>y</sub>@NC-750.



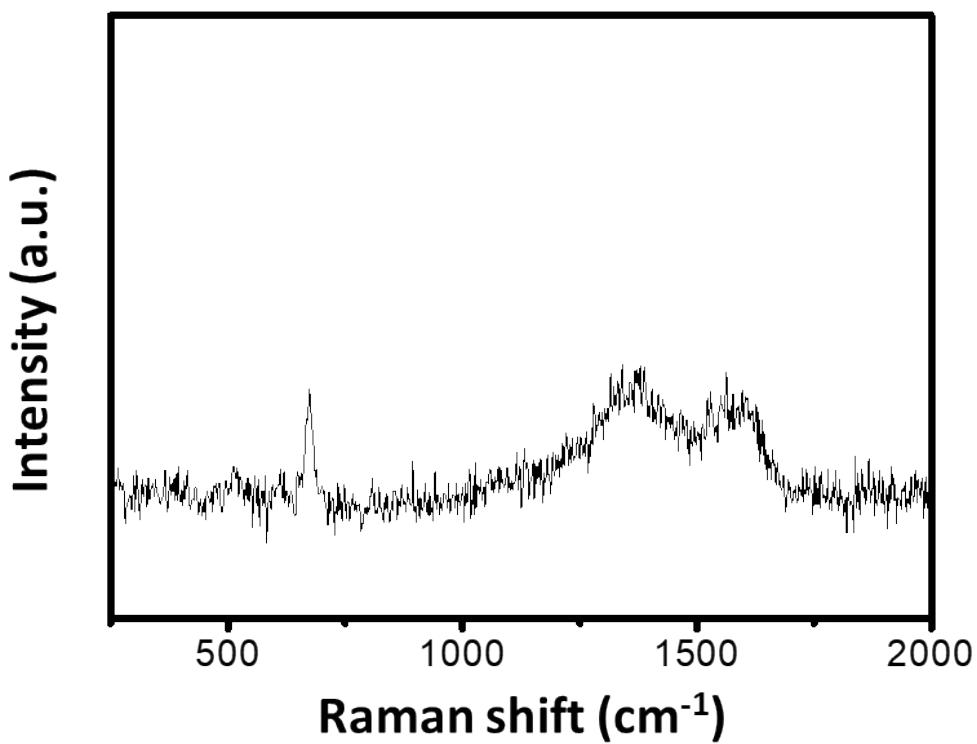
**Figure S5.** Higher-magnification TEM images of Co/Co<sub>x</sub>S<sub>y</sub>@NC-750.



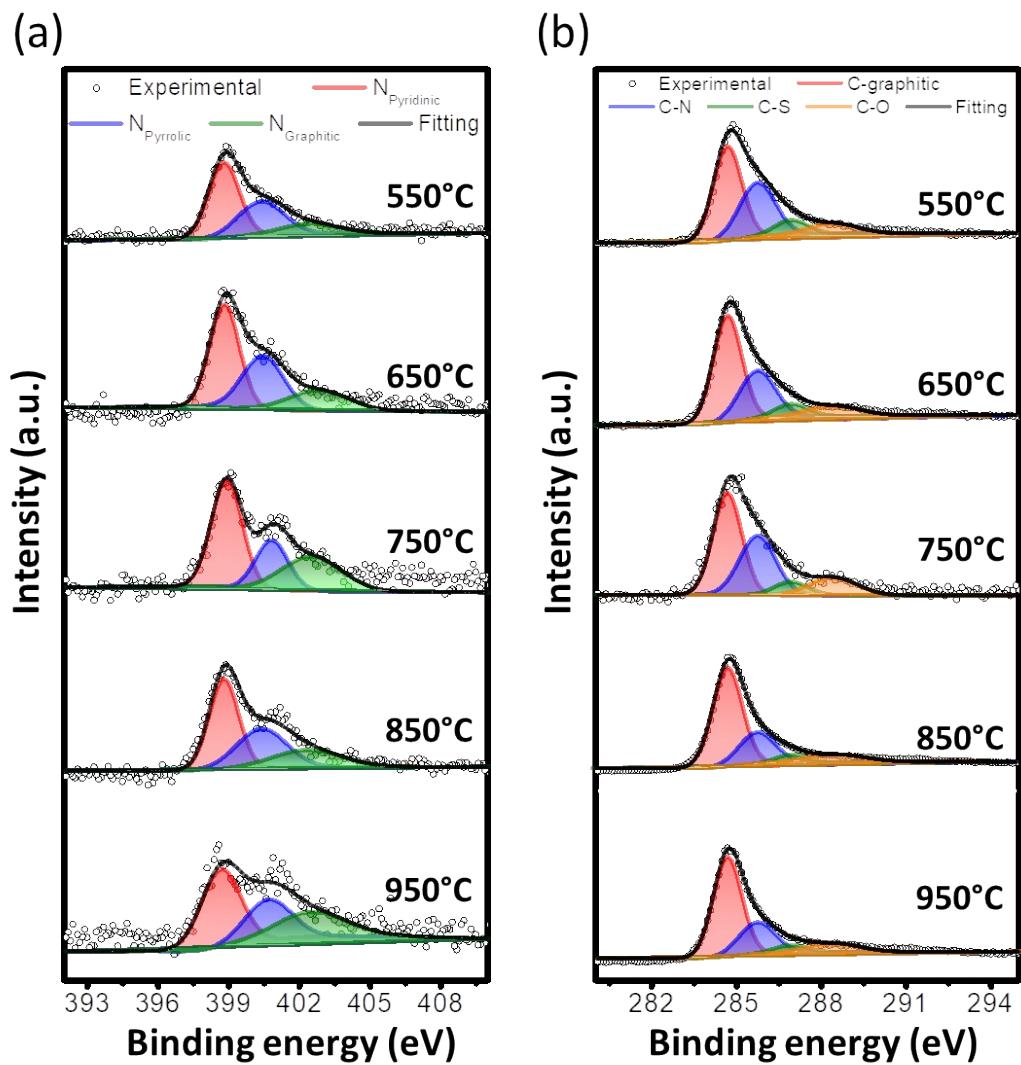
**Figure S6.** (a, b) TEM images and (c) EDX mapping of Co@NC. C (red), N (blue), and Co (green).



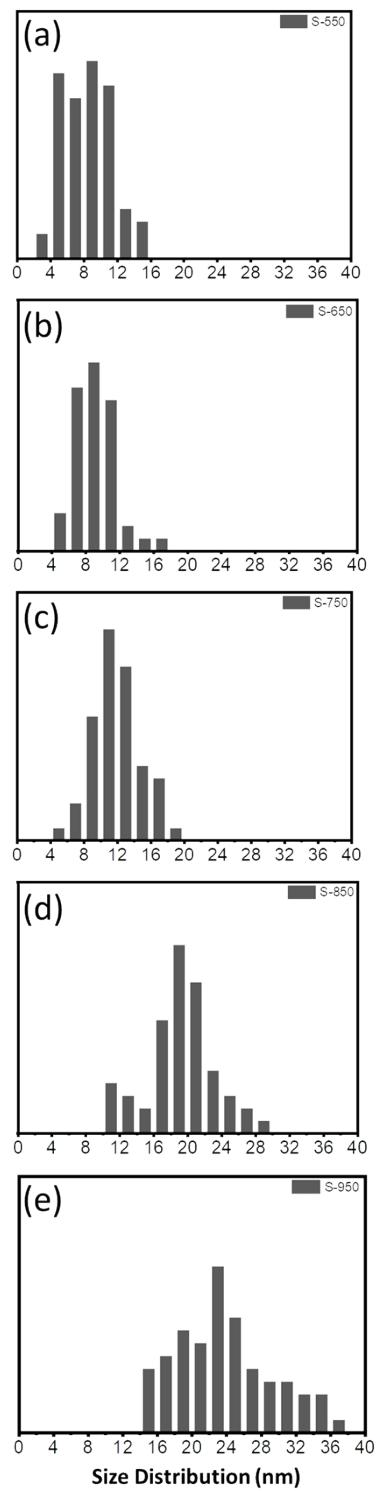
**Figure S7.** (a, b) TEM images and (c) EDX mapping of  $\text{Co}_x\text{S}_y@\text{NC}$ . C (red), N (green), S (yellow), and Co (orange).



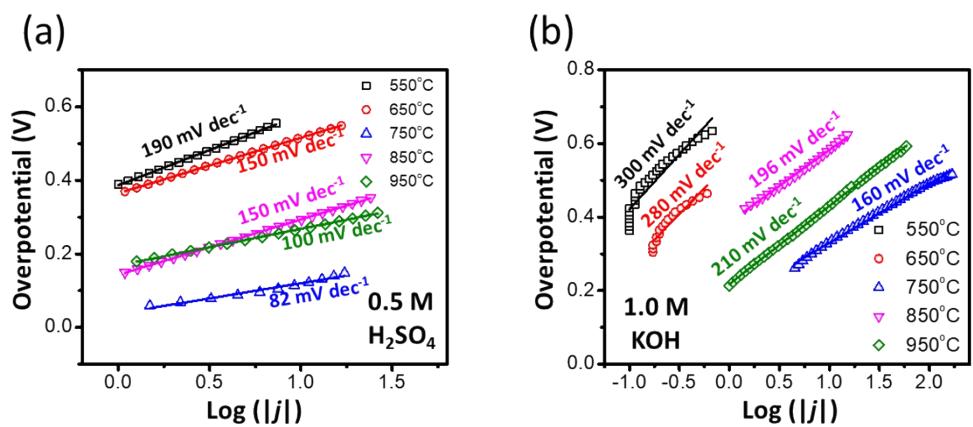
**Figure S8.** Raman spectrum of  $\text{Co}_3\text{O}_4$  prepared by calcinating ZIF-67 in the air.



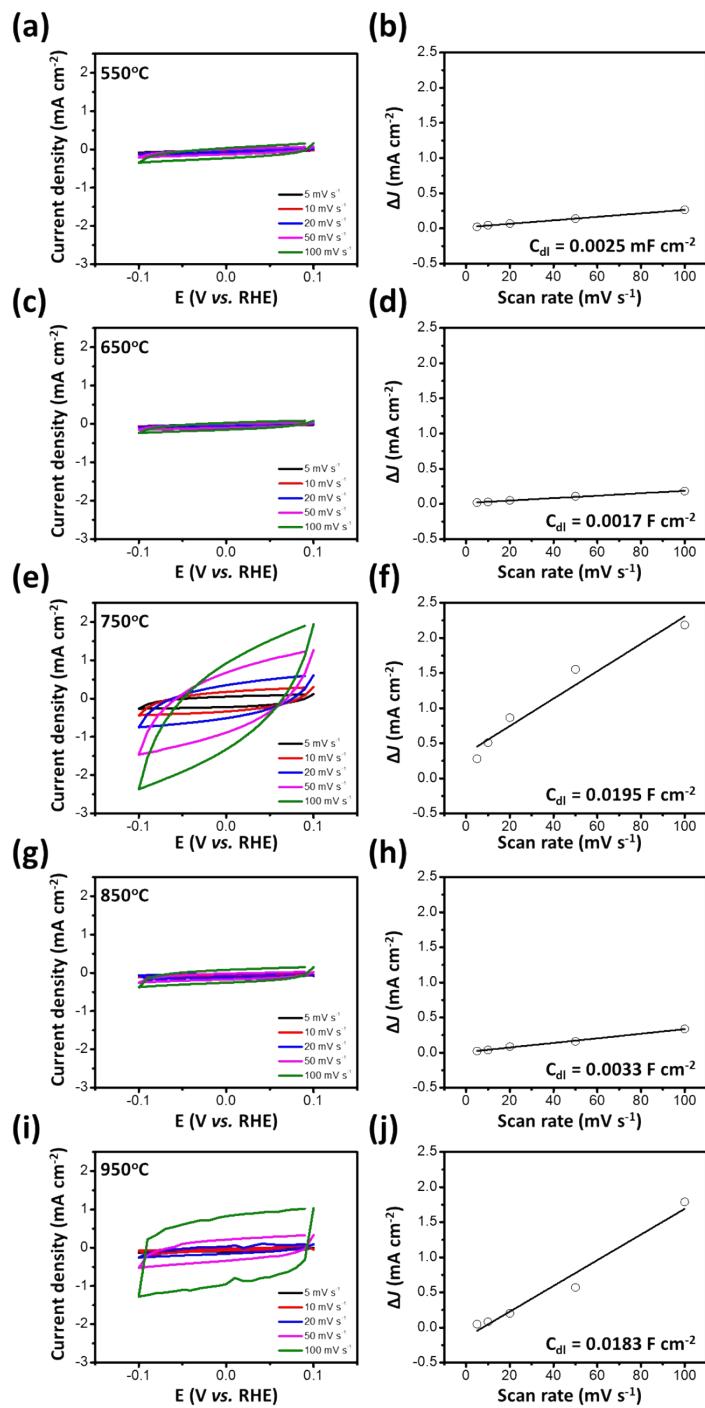
**Figure S9.** XPS spectra of (a) N 1s and (b) C 1s of S-T samples.



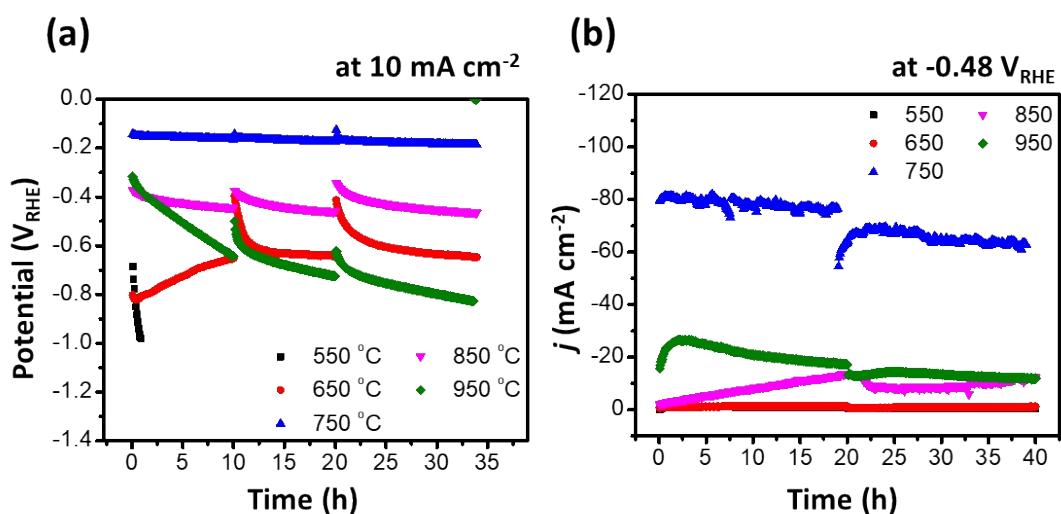
**Figure S10.** Size distribution of S-T sample synthesized at (a) 550 °C, (b) 650 °C, (c) 750 °C; (d) 850 °C, and (e) 950 °C.



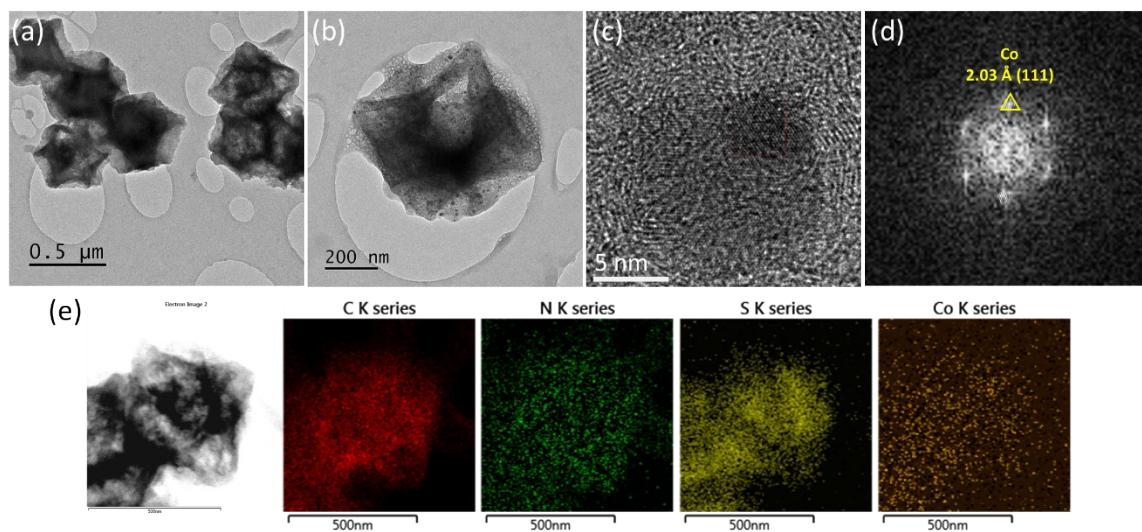
**Figure S11.** Tafel slope of the electrocatalysts in 0.5 M  $\text{H}_2\text{SO}_4$  and 1.0 M KOH.



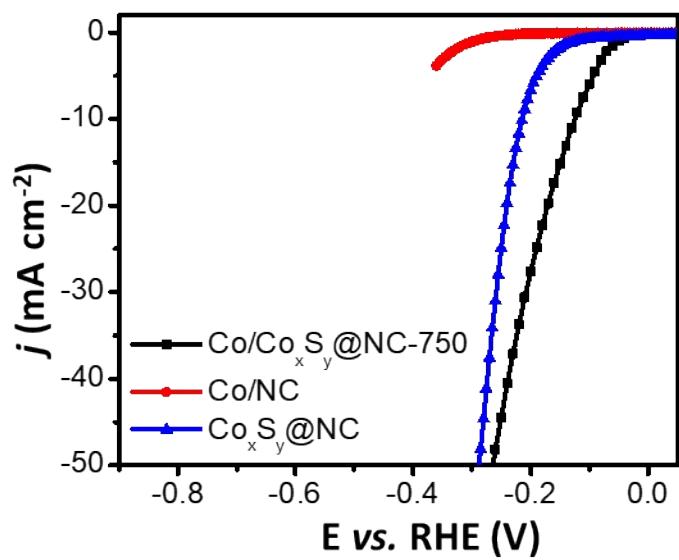
**Figure S12.** (a, c, e, g, i) Cyclic voltammetry and (b, d, f, h, j) Fitting plots of  $\Delta J$  versus scan rates for S-T samples obtained at different temperatures.



**Figure S13.** Stability of the sample obtained at different temperatures: 550 to 950 °C in (a) 0.5 M  $\text{H}_2\text{SO}_4$  and (b) 1.0 M KOH. Noted that: S-750 and Co/ $\text{Co}_x\text{S}_y@\text{NC}$ -750 refer to the similar sample.



**Figure S14.** (a, b, c) TEM, (d) FFT, and (e) EDX mapping of Co/Co<sub>x</sub>S<sub>y</sub>@NC-750 after long-term stability test in 0.5 M H<sub>2</sub>SO<sub>4</sub>.



**Figure S15.** Comparison of HER performance of  $\text{Co/Co}_x\text{S}_y@\text{NC-750}$  versus  $\text{Co/NC}$  and  $\text{Co}_x\text{S}_y/\text{NC}$  in  $0.5 \text{ M H}_2\text{SO}_4$ .

**Table S1.** Composition of Co<sup>0</sup>, Co<sup>II</sup>, and Co<sup>III</sup> species in *S-T* samples determined from XPS.

<b>Samples</b>	<b>Co<sup>0</sup></b>	<b>Co<sup>II</sup></b>	<b>Co<sup>III</sup></b>
S-550	12.9	26.4	60.7
S-650	26.7	20.9	52.4
S-750 (Co/Co <sub>x</sub> S <sub>y</sub> @NC-750)	28.3	26.1	45.7
S-850	23.8	19.0	57.2
S-950	0	16.5	83.5

**Table S2.** Performance of earth-abundant MOF-derived HER catalysts.

Sample	Overpotential	Tafel slope	electrolyte	Ref.
<b>Co/Co<sub>x</sub>S<sub>y</sub>@NC-750</b>	<b><math>\eta_{10} = 130</math> mV</b>	<b>82 mV/dec</b>	<b>0.5M H<sub>2</sub>SO<sub>4</sub></b>	<b>This work</b>
<b>Co/Co<sub>x</sub>S<sub>y</sub>@NC-750</b>	<b><math>\eta_{10} = 330</math> mV</b>	<b>160 mV/dec</b>	<b>1.0M KOH</b>	<b>This work</b>
Ir-COF@ZIF <sub>800</sub>	$\eta_{10} = 48$ mV	44 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[1]
CoS <sub>2</sub> @NSC	$\eta_{10} = 95$ mV	69 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[2]
CPCS8	$\eta_{10} = 156$ mV	74 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[3]
ZnCo-11-NC	$\eta_{10} = 213$ mV	77 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[4]
Fe <sub>0.39</sub> Co <sub>0.61</sub> S <sub>2</sub> /rGO	$\eta_{10} = 198$ mV	94 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[5]
Sn-CoS <sub>2</sub> /CC	$\eta_{10} = 161$ mV	94 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[6]
Co@NC/B-NCNTS	$\eta_{10} = 182$ mV	105 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[7]
Co-NC@Mo <sub>2</sub> C	$\eta_{10} = 143$ mV	60 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[8]
CoP@NC/GR	$\eta_{10} = 105$ mV	47.5 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[9]
Ni <sub>2</sub> P/Ni@C	$\eta_{10} = 149$ mV	61.2 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[10]
Co <sub>3</sub> O <sub>4</sub> /C	$\eta_{10} = 194.8$ mV	71.6 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[11]
Pd-CoS <sub>2</sub> MoS <sub>2</sub> /C-600	$\eta_{10} = 144$ mV	59.9 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[12]
MoS <sub>2</sub> /CoS <sub>2</sub> NTs	$\eta_{10} = 90$ mV	30 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[13]
NiRu <sub>2</sub> @NC-600	$\eta_{10} = 85$ mV	45.54 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[14]
rePCN-222(Hf)-700	$\eta_{10} = 31.6$ mV	22 mV/dec	0.5M H <sub>2</sub> SO <sub>4</sub>	[15]

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CoSP/NC	$\eta_{10} = 183$ mV	64.25 mV/dec	1.0M KOH	[16]
Co-M-Fe/Ni	$\eta_{10} = 149$ mV	65 mV/dec	1.0M KOH	[17]
Co@N-SC/N-HCP@CC	$\eta_{10} = 66$ mV	65 mV/dec	1.0M KOH	[18]
CoS <sub>2</sub> QDs@rGO	$\eta_{10} = 287$ mV	78 mV/dec	1.0M KOH	[19]
CoP/CNTHPs	$\eta_{10} = 147$ mV	78.1 mV/dec	1.0M KOH	[20]
Co-S-P/CC	$\eta_{10} = 167$ mV	86mV/dec	1.0M KOH	[21]
CoNiP/CoNi/N-rGO	$\eta_{10} = 150$ mV	47.51 mV/dec	1.0M KOH	[22]
HNi/NiO/C	$\eta_{10} = 87$ mV	91.7 mV/dec	1.0M KOH	[23]
Ru-MoO <sub>2</sub> @PC/rGO	$\eta_{10} = 126$ mV	43.5 mV/dec	1.0M KOH	[24]
Ni@NC-600	$\eta_{10} = 180$ mV	120 mV/dec	1.0M KOH	[25]
Fe-Ni@NC-CNTs	$\eta_{10} = 202$ mV	113.7 mV/dec	1.0M KOH	[26]
ZnSP/NC	$\eta_{10} = 171$ mV	54.78 mV/dec	1.0M KOH	[27]
Co/Ni@GC/NCNTs/CN	$\eta_{10} = 150$ mV	86 mV/dec	1.0M KOH	[28]
Co/Ni-MOFs@P	$\eta_{10} = 194$ mV	58.3 mV/dec	1.0M KOH	[29]
Co/Ni-MOFs@Se	$\eta_{10} = 247$ mV	88.9 mV/dec	1.0M KOH	[29]

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## References

- [1] Z. Duan, M. Liu, X. Li, S. Yang, S. Liu, Q. Xu, G. Zeng, Integrating multifunctional catalytic sites in COF@ZIF-67 derived carbon for the HER and ORR, *Chemical Communications*, 58 (2022) 13214-13217.
- [2] S. Feng, X. Li, J. Huo, Q. Li, C. Xie, T. Liu, Z. Liu, Z. Wu, S. Wang, Controllable Synthesis of CoS<sub>2</sub>@N/S-Codoped Porous Carbon Derived from ZIF-67 for as a Highly Efficient Catalyst for the Hydrogen Evolution Reaction, *ChemCatChem*, 10 (2018) 796-803.
- [3] H.B.M. Sidek, X. Jin, M.S. Islam, S.-J. Hwang, 1D Composite Nanorods of Cobalt Phosphide-Cobalt Sulfide with Improved Electrocatalyst Performance, *ChemCatChem*, 11 (2019) 6099-6104.
- [4] R. Wang, Q. Yuan, P. Sun, R. Nie, X. Wang, Tuning the active sites in the cobalt-based nitrogen-doped carbon by zinc for enhancing hydrogen evolution reaction, *Journal of Alloys and Compounds*, 789 (2019) 100-107.
- [5] M. Wang, W. Tang, S. Liu, X. Liu, X. Chen, X. Hu, L. Qiao, Y. Sui, Design of earth-abundant ternary Fe<sub>1-x</sub>CoxS<sub>2</sub> on RGO as efficient electrocatalysts for hydrogen evolution reaction, *Journal of Alloys and Compounds*, 862 (2021) 158610.
- [6] F. Liu, W. He, Y. Li, F. Wang, J. Zhang, X. Xu, Y. Xue, C. Tang, H. Liu, J. Zhang, Activating sulfur sites of CoS<sub>2</sub> electrocatalysts through tin doping for hydrogen evolution reaction, *Applied Surface Science*, 546 (2021) 149101.
- [7] Z.-J. Jiang, G. Xie, L. Guo, J. Huang, Z. Jiang, Co nanoparticles coupling induced high catalytic activity of nitrogen doped carbon towards hydrogen evolution reaction in acidic/alkaline solutions, *Electrochimica Acta*, 342 (2020) 136076.
- [8] Q. Liang, H. Jin, Z. Wang, Y. Xiong, S. Yuan, X. Zeng, D. He, S. Mu, Metal-organic frameworks derived reverse-encapsulation Co-NC@Mo<sub>2</sub>C complex for efficient overall water splitting, *Nano Energy*, 57 (2019) 746-752.

- [9] M. Guo, F. Qiu, Y. Yuan, T. Yu, C. Yuan, Z.-H. Lu, Active Site Engineering in CoP@NC/Graphene Heterostructures Enabling Enhanced Hydrogen Evolution, Inorganic Chemistry, 60 (2021) 16761-16768.
- [10] X. Liu, W. Li, X. Zhao, Y. Liu, C.-W. Nan, L.-Z. Fan, Two Birds with One Stone: Metal–Organic Framework Derived Micro-/Nanostructured Ni<sub>2</sub>P/Ni Hybrids Embedded in Porous Carbon for Electrocatalysis and Energy Storage, Advanced Functional Materials, 29 (2019) 1901510.
- [11] K. Gothandapani, A.N. Grace, V. Venugopal, Mesoporous carbon-supported CO<sub>3</sub>O<sub>4</sub> derived from Zif-67 metal organic framework (MOF) for hydrogen evolution reaction in acidic and alkaline medium, International Journal of Energy Research, 46 (2022) 3384-3395.
- [12] Z.-X. Cai, J. Na, J. Lin, A.A. Alshehri, K.A. Alzahrani, Y.G. Alghamdi, H. Lim, J. Zheng, W. Xia, Z.-L. Wang, Y. Yamauchi, Hierarchical Tubular Architecture Constructed by Vertically Aligned CoS<sub>2</sub>-MoS<sub>2</sub> Nanosheets for Hydrogen Evolution Electrocatalysis, Chemistry – A European Journal, 26 (2020) 6195-6204.
- [13] B. Tang, Z.G. Yu, Y. Zhang, C. Tang, H.L. Seng, Z.W. Seh, Y.-W. Zhang, S.J. Pennycook, H. Gong, W. Yang, Metal–organic framework-derived hierarchical MoS<sub>2</sub>/CoS<sub>2</sub> nanotube arrays as pH-universal electrocatalysts for efficient hydrogen evolution, Journal of Materials Chemistry A, 7 (2019) 13339-13346.
- [14] S. Xu, Z. Li, K. Chu, G. Yao, Y. Xu, P. Niu, F. Zheng, NiRu nanoparticles encapsulated in a nitrogen-doped carbon matrix as a highly efficient electrocatalyst for the hydrogen evolution reaction, Dalton Transactions, 49 (2020) 13647-13654.
- [15] H.-M. Mei, S. Li, J.-R. Dong, L. Zhang, C.-Y. Su, Porphyrinic Metal-Organic Frameworks Derived Carbon-Based Nanomaterials for Hydrogen Evolution Reaction, ChemistrySelect, 5 (2020) 10988-10995.

- [16] L. Zhao, A. Yang, A. Wang, H. Yu, J. Dai, Y. Zheng, Metallic Co, CoS, and P co-doped N enriched carbon derived from ZIF-67 as an efficient catalyst for hydrogen evolution reaction, International Journal of Hydrogen Energy, 45 (2020) 30367-30374.
- [17] P. Chen, X. Duan, G. Li, X. Qiu, S. Wang, Y. Huang, A. Stavitskaya, H. Jiang, Construction of ZIF-67/MIL-88(Fe, Ni) catalysts as a novel platform for efficient overall water splitting, International Journal of Hydrogen Energy, 48 (2023) 7170-7180.
- [18] Z. Chen, Y. Ha, H. Jia, X. Yan, M. Chen, M. Liu, R. Wu, Oriented Transformation of Co-LDH into 2D/3D ZIF-67 to Achieve Co–N–C Hybrids for Efficient Overall Water Splitting, Advanced Energy Materials, 9 (2019) 1803918.
- [19] J. Jiang, R. Sun, X. Huang, H. Cong, J. Tang, W. Xu, M. Li, Y. Chen, Y. Wang, S. Han, H. Lin, CoS<sub>2</sub> quantum dots modified by ZIF-67 and anchored on reduced graphene oxide as an efficient catalyst for hydrogen evolution reaction, Chemical Engineering Journal, 430 (2022) 132634.
- [20] W. Liao, X. Tong, Y. Zhai, H. Dai, Y. Fu, M. Qian, G. Wu, T. Chen, Q. Yang, ZIF-67-derived nanoframes as efficient bifunctional catalysts for overall water splitting in alkaline medium, Dalton Transactions, 51 (2022) 7561-7570.
- [21] X. Liu, J. Dong, B. You, Y. Sun, Competent overall water-splitting electrocatalysts derived from ZIF-67 grown on carbon cloth, RSC Advances, 6 (2016) 73336-73342.
- [22] B. Chen, D. Kim, Z. Zhang, M. Lee, K. Yong, MOF-derived NiCoZnP nanoclusters anchored on hierarchical N-doped carbon nanosheets array as bifunctional electrocatalysts for overall water splitting, Chemical Engineering Journal, 422 (2021) 130533.
- [23] H.H. Do, M.A. Tekalgne, Q.V. Le, J.H. Cho, S.H. Ahn, S.Y. Kim, Hollow Ni/NiO/C composite derived from metal-organic frameworks as a high-efficiency electrocatalyst for the hydrogen evolution reaction, Nano Convergence, 10 (2023) 6.

- [24] J.-Y. Han, S.-H. Cai, J.-Y. Zhu, S. Yang, J.-S. Li, MOF-derived ruthenium-doped amorphous molybdenum dioxide hybrid for highly efficient hydrogen evolution reaction in alkaline media, *Chemical Communications*, 58 (2022) 100-103.
- [25] N. Cheng, N. Wang, L. Ren, G. Casillas-Garcia, N. Liu, Y. Liu, X. Xu, W. Hao, S.X. Dou, Y. Du, In-situ grafting of N-doped carbon nanotubes with Ni encapsulation onto MOF-derived hierarchical hybrids for efficient electrocatalytic hydrogen evolution, *Carbon*, 163 (2020) 178-185.
- [26] X. Zhao, P. Pachfule, S. Li, J.R.J. Simke, J. Schmidt, A. Thomas, Bifunctional Electrocatalysts for Overall Water Splitting from an Iron/Nickel-Based Bimetallic Metal–Organic Framework/Dicyandiamide Composite, *Angewandte Chemie International Edition*, 57 (2018) 8921-8926.
- [27] Y. Jing, H. Yin, Y. Zhang, B. Yu, MOF-derived Zn, S, and P co-doped nitrogen-enriched carbon as an efficient electrocatalyst for hydrogen evolution reaction, *International Journal of Hydrogen Energy*, 45 (2020) 19174-19180.
- [28] J. Li, J. Qian, X. Chen, X. Zeng, L. Li, B. Ouyang, E. Kan, W. Zhang, Three-dimensional hierarchical graphitic carbon encapsulated CoNi alloy/N-doped CNTs/carbon nanofibers as an efficient multifunctional electrocatalyst for high-performance microbial fuel cells, *Composites Part B: Engineering*, 231 (2022) 109573.
- [29] P. Luo, Z. Pang, Z. Qin, T. Wei, S. Li, Y. Hu, C. Wei, Strategies for improving Co/Ni-based bimetal-organic framework to water splitting, *International Journal of Hydrogen Energy*, 45 (2020) 28240-28251.