

Supporting Information

3D Crinkled MXene/TiO₂ Heterostructure with Interfacial Coupling for Ultra-Fast and Reversible Potassium Storage

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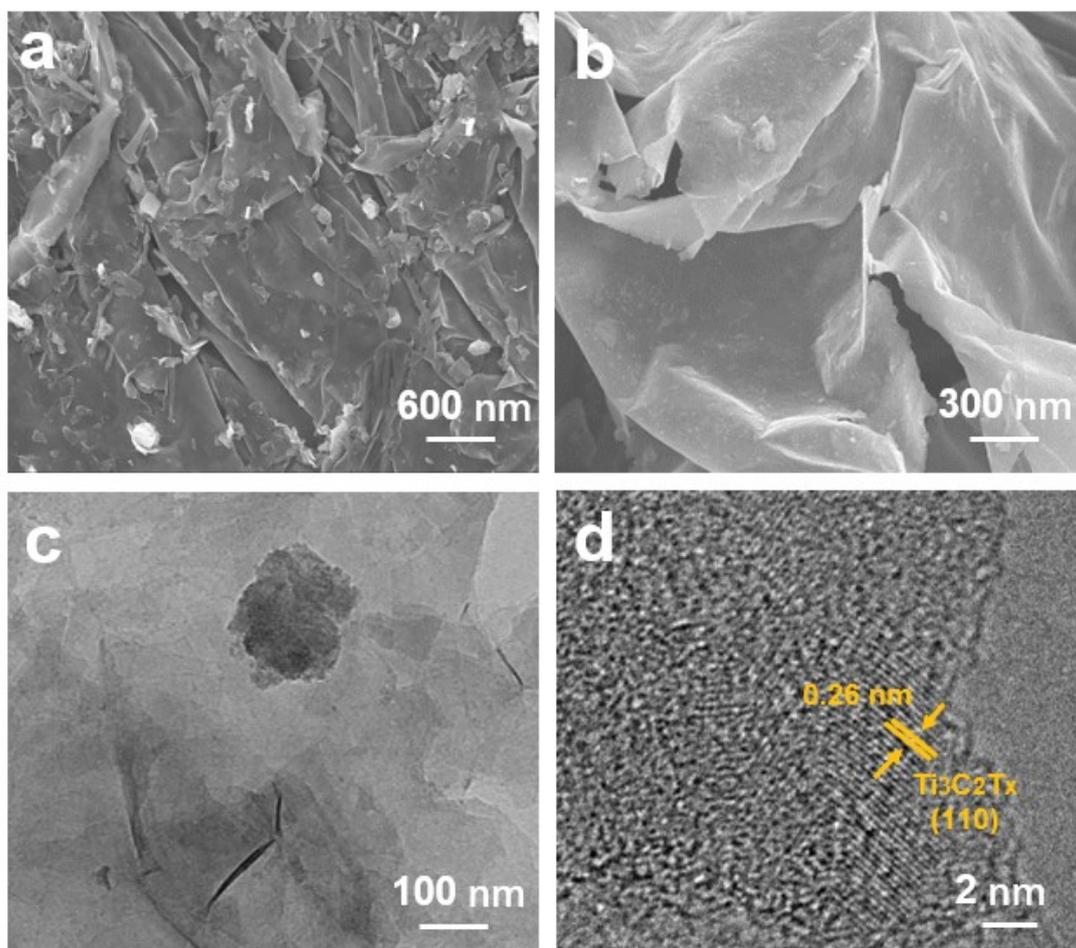


Figure S1 (a)-(b) SEM images, (c) TEM image and (d) HRTEM image of MXene.

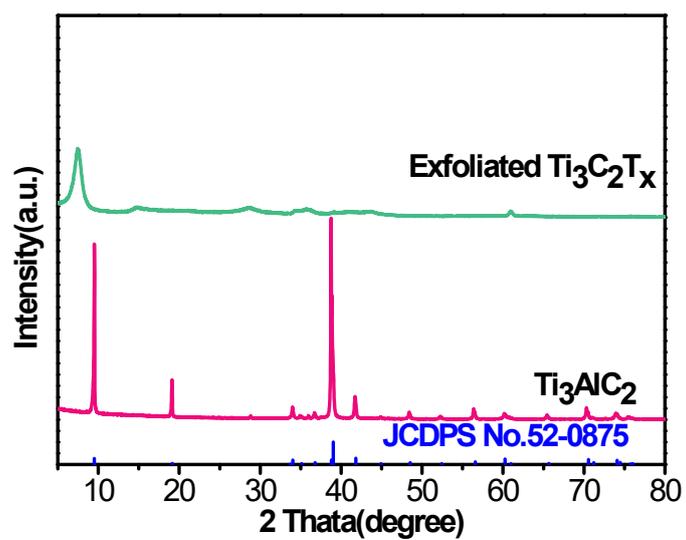


Figure S2 XRD patterns of Ti₃AlC₂ and Ti₃C₂T_x nanosheets.

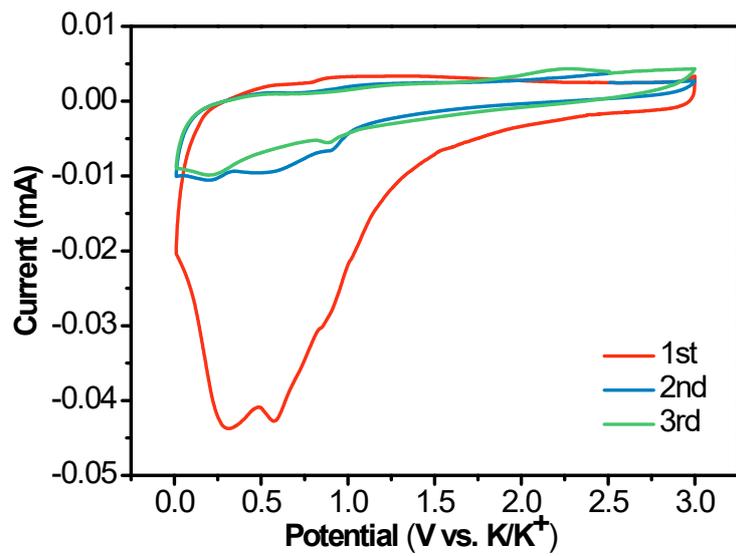


Figure S3 CV curves of CM at a scan rate of 0.1 mV s⁻¹.

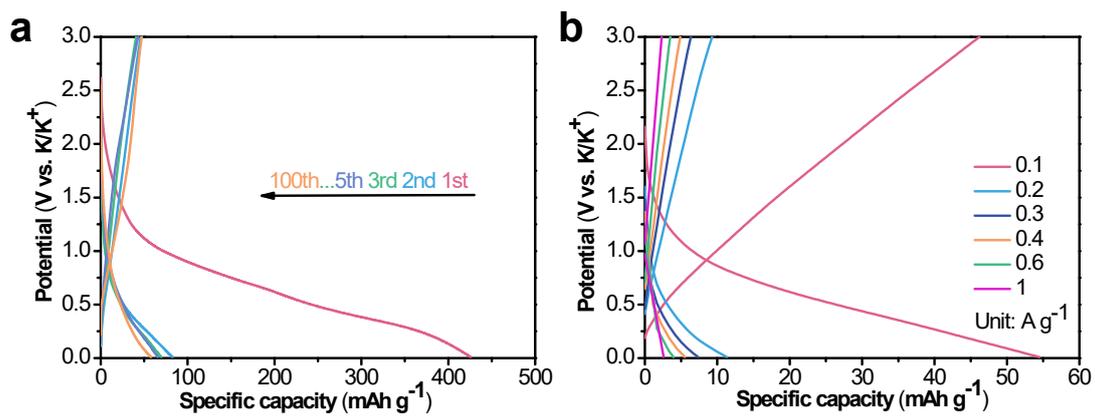


Figure S4 (a) Galvanostatic charge/discharge curves of CM at 0.05 A g^{-1} . (b) The charge/discharge profiles of CM at different current densities.

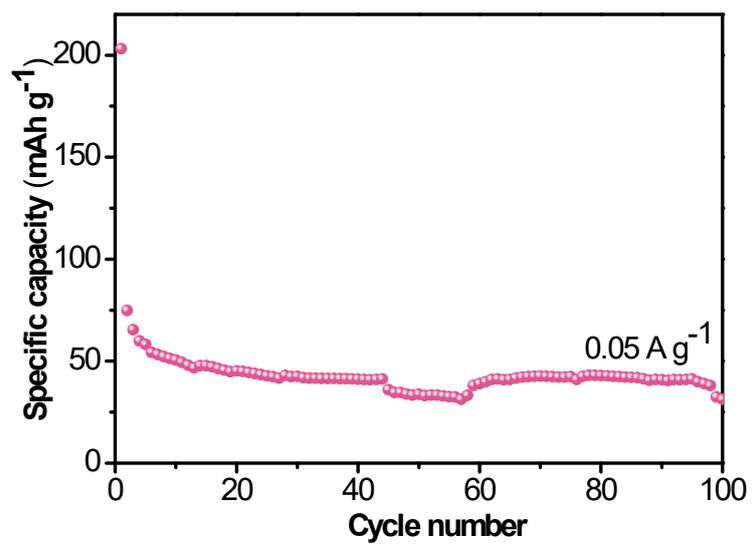


Figure S5 Cycle performance of MXene at 0.05 A g^{-1} over 100 cycles.

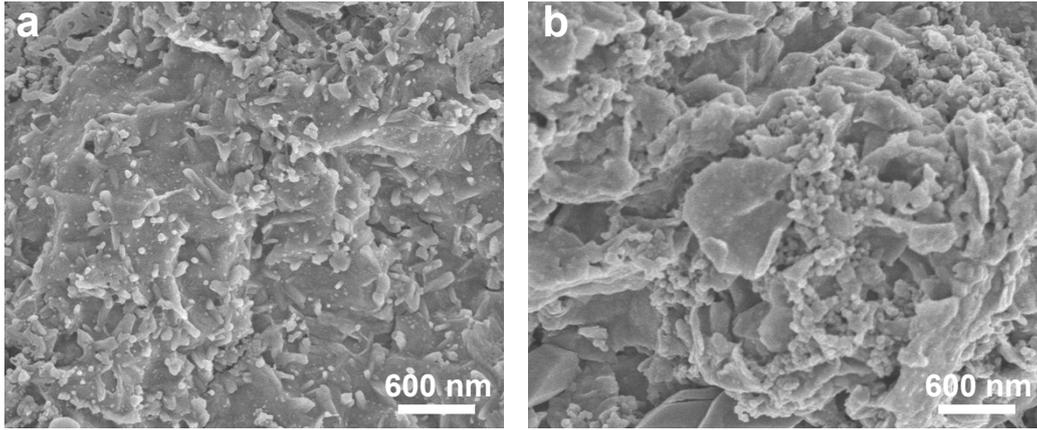


Figure S6 (a) SEM images of CM/TiO₂ after cycle at 0.05 A g⁻¹. (b) SEM images of CM after cycle at 0.05 A g⁻¹.

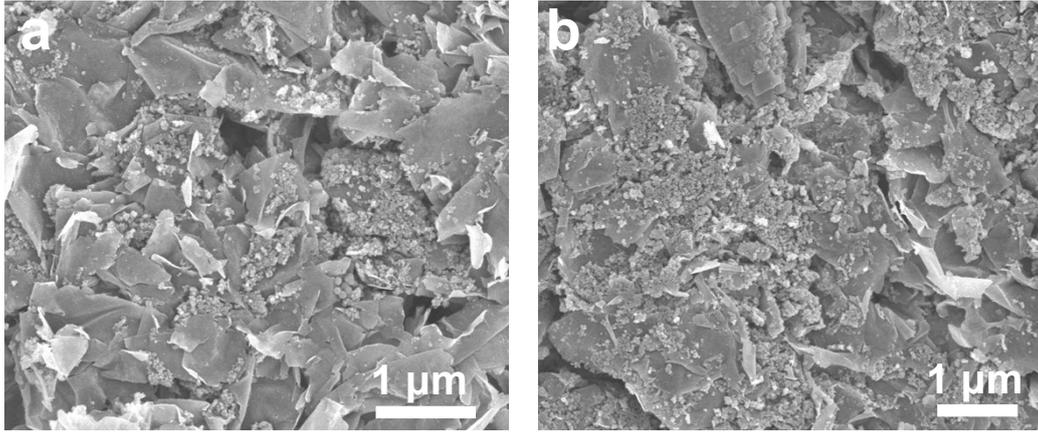


Figure S7 (a) SEM images of CM/TiO₂-3h. (b) SEM images of CM/TiO₂-4h.

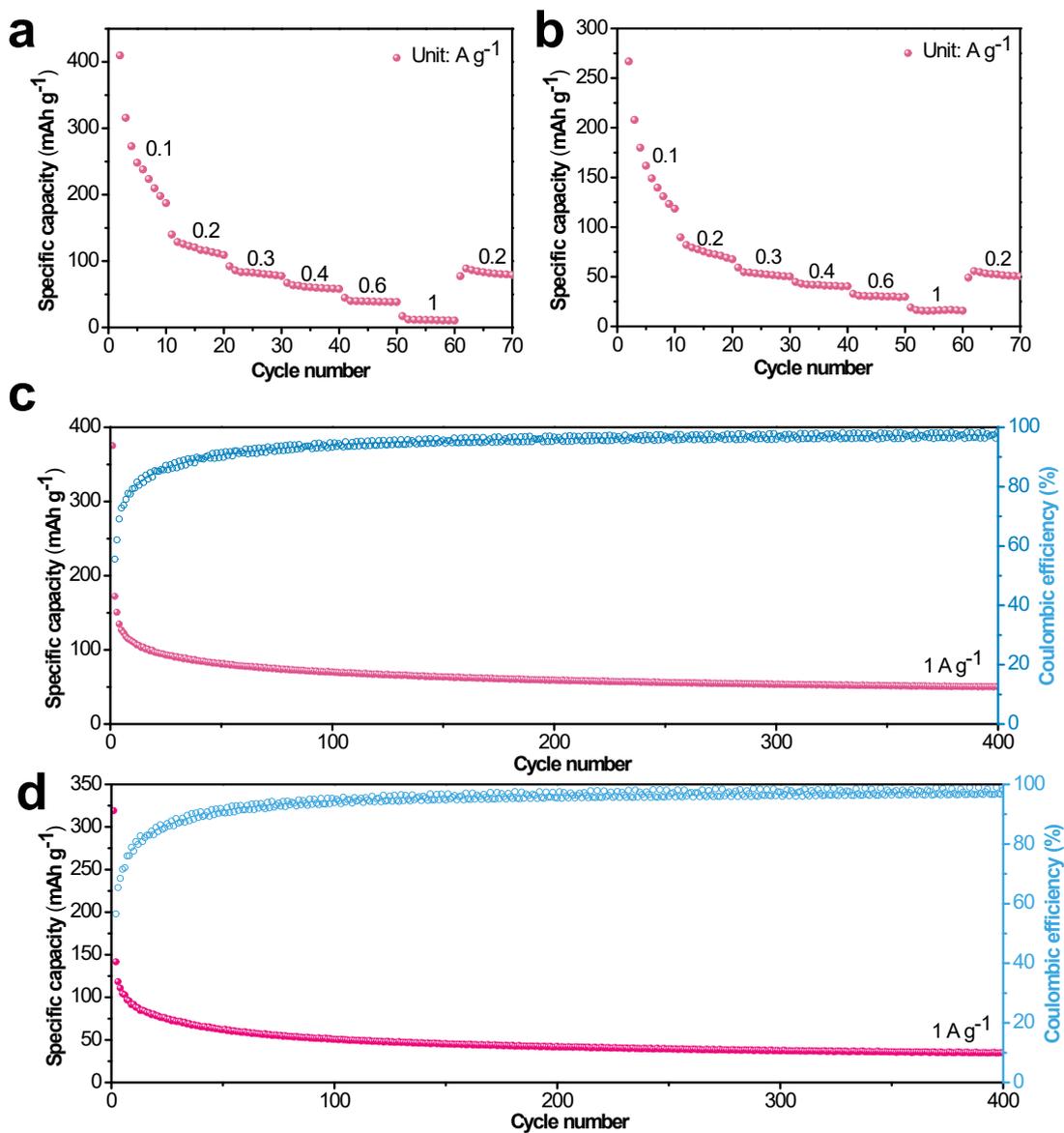


Figure S8 (a) Rate capability of CM/TiO₂-3h at various current densities. (b) Rate capability of CM/TiO₂-4h at various current densities. (c) Long cycle performance of CM/TiO₂-3h at 1 A g⁻¹. (d) Long cycle performance of CM/TiO₂-4h at 1 A g⁻¹.

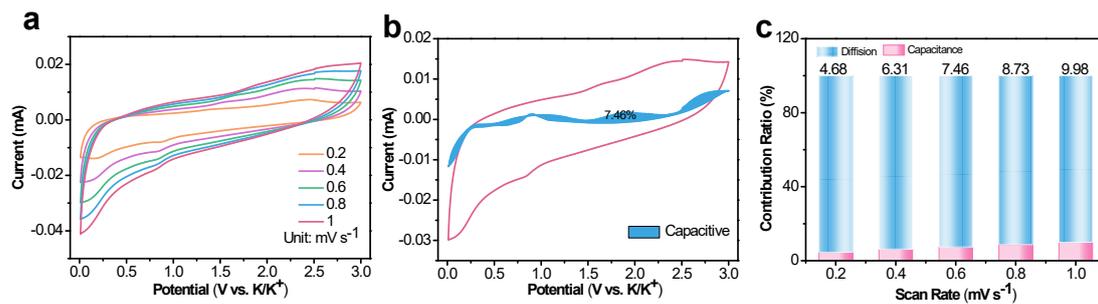


Figure S9 (a) CV profiles of CM at different scan rates from 0.2 to 1.0 mV s⁻¹. (b) Capacitive contribution of CM at 0.6 mV s⁻¹. (c) Normalized contribution ratio of capacitive capacities in CM at various scan rates.

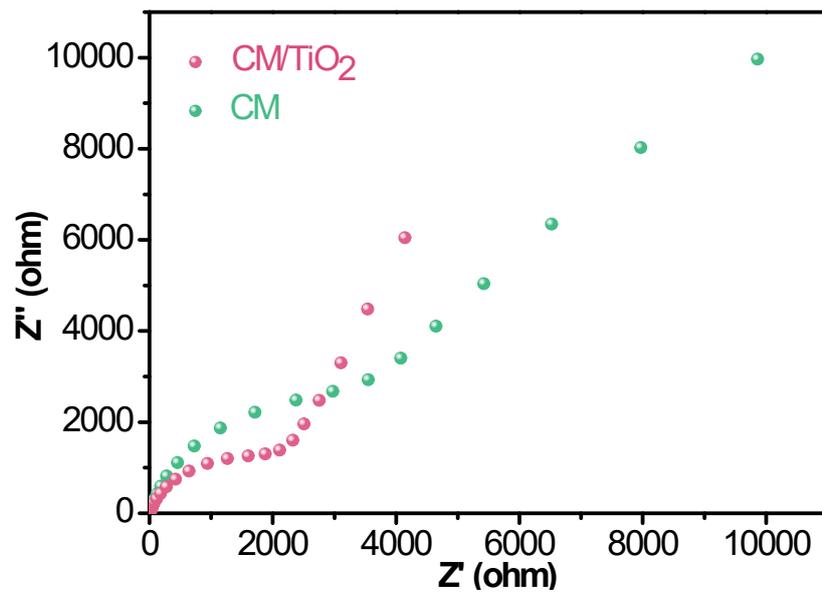


Figure S10 Nyquist plot of CM/TiO₂ and CM after 100 cycles.

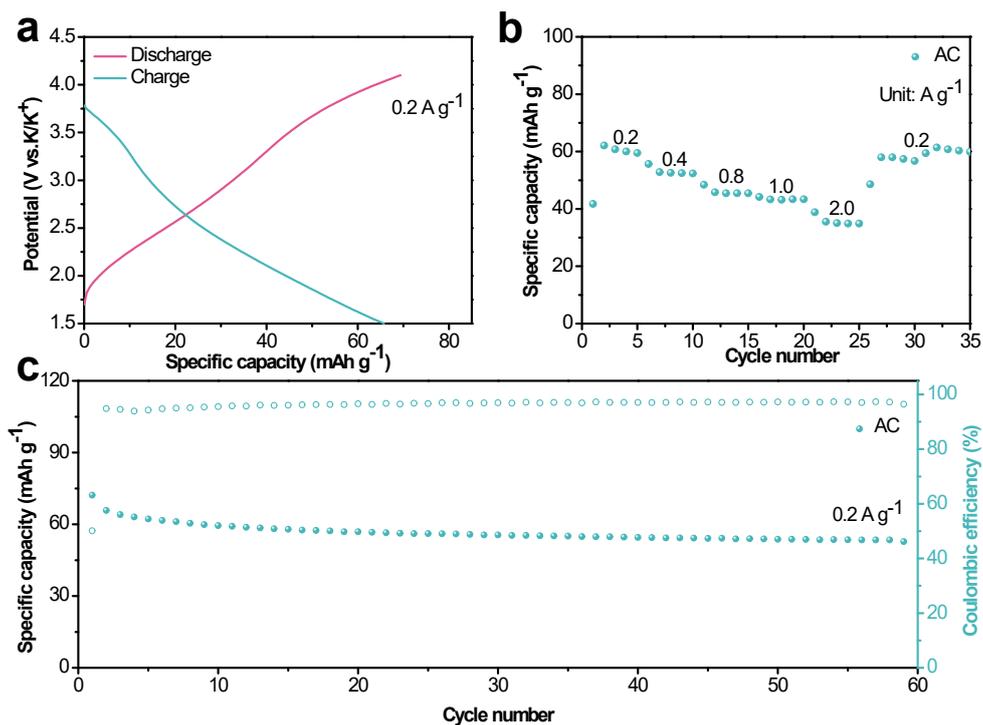


Figure S11 (a) Galvanostatic charge/discharge curves of AC electrode at 0.2 A g⁻¹. (b) Rate capability of AC electrode at various current densities. (c) Cycle performance of AC electrode at 0.2 A g⁻¹.

Table S1. The comparisons of the reported anode for PIBs.

Electrodes	Low rate capacity	High rate capacity	Cycle number
This work	301 mA h g ⁻¹ (0.1 A g ⁻¹)	94 mA h g ⁻¹ (0.4 A g ⁻¹)	150 mA h g ⁻¹ (0.05 A g ⁻¹ , 100 cycles)
Ti ₃ CN[1]	181 mA h g ⁻¹ (0.02 A g ⁻¹)	80 mA h g ⁻¹ (0.5 A g ⁻¹)	60 mA h g ⁻¹ (0.1 A g ⁻¹ , 100 cycles)
f-MXene[2]	119 mA h g ⁻¹ (0.08 A g ⁻¹)	80 mA h g ⁻¹ (0.5 A g ⁻¹)	120 mA h g ⁻¹ (0.1 A g ⁻¹ , 200 cycles)
a-Ti ₃ C ₂ T _x [3]	167 mA h g ⁻¹ (0.08 A g ⁻¹)	90 mA h g ⁻¹ (0.2 A g ⁻¹)	50 mA h g ⁻¹ (0.1 A g ⁻¹ , 120 cycles)
Ti ₃ C ₂ [4]	119 mA h g ⁻¹ (0.01 A g ⁻¹)	57 mA h g ⁻¹ (0.5 A g ⁻¹)	30 mA h g ⁻¹ (0.2 A g ⁻¹ , 500 cycles)
h-MXene[2]	69 mA h g ⁻¹ (0.08 A g ⁻¹)	39 mA h g ⁻¹ (0.5 A g ⁻¹)	100 mA h g ⁻¹ (0.1 A g ⁻¹ , 200 cycles)
MXene Ti ₃ C ₂ [5]	105 mA h g ⁻¹ (0.06 A g ⁻¹)	20 mA h g ⁻¹ (0.5 A g ⁻¹)	80 mA h g ⁻¹ (0.05 A g ⁻¹ , 100 cycles)
N-UT-Ti ₃ C ₂ T _x [6]	166 mA h g ⁻¹ (0.09 A g ⁻¹)	141 mA h g ⁻¹ (0.3 A g ⁻¹)	100 mA h g ⁻¹ (0.1 A g ⁻¹ , 100 cycles)

Reference

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- [5] Z. Guo, G. Dong, M. Zhang, M. Gao, L. Shao, M. Chen, H. Liu, M. Ni, D. Cao, K. Zhu, Sulfur-Decorated $Ti_3C_2T_x$ MXene for High-Performance Sodium/Potassium-Ion Batteries, *Chemistry – An Asian Journal* 18(18) (2023).
- [6] Y. Zhao, G. Dong, M. Zhang, D. Wang, Y. Chen, D. Cao, K. Zhu, G. Chen, Surface-engineered $Ti_3C_2T_x$ MXene enabling rapid sodium/potassium ion storage, *2D Materials* 10(1) (2022).