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# **Supporting Information**

### Catechol exerts in situ surface oxygen vacancy effect on BiOI: an innovative

## signal transduction mode for cathodic photoelectrochemistry

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Fig. S1 (A) XRD pattern and (B) SEM image of BiOI.



Fig. S2 (A) The effect of test buffer medium pH, and (B) applied bias on photocurrents of BiOI modified electrode to CA (10.0  $\mu$ M). (C) Photocurrents of BiOI modified electrode when incubated with different concentrations of CA (from left to right, 0.0, 0.01, 0.1, 1.0, 5.0, 10.0, 50.0, and 100.0  $\mu$ M). (D) The corresponding calibration curve of the photocurrent change (I/I<sub>0</sub>) with CA concentrations on a logarithmic scale.



Fig. S3 The color changes of BiOI before and after reaction with CA.



**Fig. S4** (A) Survey XPS spectra of BiOI before and after reaction with CA, and corresponding high-resolution XPS spectra of (B) C 1s, (C) O 1s, (D) Bi 4f, and (E) I 3d.



Fig. S5 Photocurrent responses of BiOI modified electrode before (blue line) and after (red line) reaction with  $10.0 \mu$ M of CA in deaerated or aerated solutions.

Sample	<b>B</b> <sub>1</sub>	<b>B</b> <sub>2</sub>	$\tau_1$ (ns)	$\tau_2(ns)$	$\chi^2$	$ au_{avg}(ns)$
Before	5431.27	3029.61	1.51	4.06	1.15	3.04
After	5621.32	2719.36	2.03	5.24	1.01	3.81

Table S1. Excited state lifetimes of BiOI before and after reaction with CA

The fitting model is  $I(t) = B_1 e^{-\tau/\tau 1} + B_2 e^{-\tau/\tau 2}$ . The average fluorescence lifetime  $(\tau_{avg})$  is calculated according to the equation:  $\tau_{avg} = (B_1 \tau_1^2 + B_2 \tau_2^2)/(B_1 \tau_1 + B_2 \tau_2)$ , where B represent weighting parameter,  $\tau$  is decay time.



Fig. S6 The optimization of the (A) phenol concentration, (B) NADH concentration, (C) reaction temperature, and

(D) reaction time of the bioreaction for the PEC detection of TYR. The concentration of TYR was 0.1 U/mL.

As shown in Fig. S6, when the phenol concentration was 1.0 mM, the NADH concentration was 1.0 mM, the reaction temperature was 37 °C, and the bioreaction time was 20 min, the maximal photocurrent responses of the PEC detection system for TYR were obtained.



Fig. S7 (A) Photocurrents of the BiOI modified electrode to different concentrations (0.0, 0.0001, 0.0005, 0.001, 0.005, 0.01, 0.1, and 1.0 U/mL) of TYR. (B) The corresponding calibration curve of the photocurrent increment  $(I/I_0)$  with TYR concentrations on a logarithmic scale.

Method	Material	Linear range (U/mL)	LOD (U/mL)	Reference
SERS	<i>p</i> -TC/AuNPs	0.1 - 100.0	0.07	[1]
BL	TYR/LH <sub>2</sub>	0.1-200.0	0.1	[2]
FL	Dopa/CQDs	2.3×10 <sup>-2</sup> -0.8	7.0×10 <sup>-3</sup>	[3]
FL	Au NCs	6.0×10 <sup>-3</sup> –3.6	6.0×10 <sup>-3</sup>	[4]
PEC	CdS NPs	Not given	0.1	[5]
PEC	BiOI	$1.0 \times 10^{-4} - 1.0$	2.5×10 <sup>-5</sup>	This work

Table S2. Comparison with different methods for the detection of TYR

Note: FL: Fluorescence; BL: Bioluminescence; SERS: Surface-enhanced Raman scattering.



**Fig. S8** UV-Vis absorption spectra of various solutions: Au NPs (black line); TYR (red line); Magainin I-C (green line); Au NPs/TYR/Magainin I-C bioconjugates (blue line).

The characteristic absorption peaks for Au nanoparticles (NPs), TYR, and Magainin I-C could be seen in the Au NPs/TYR/Magainin I-C bioconjugates (Fig. S8), which confirmed that TYR and Magainin I-C were successfully fixed to Au NPs and subsequently used as catalytic label (for signify the detection) and recognition element for *E. coli* O157:H7.



**Fig. S9** The effect of the incubation time of (A) *E. coli* O157:H7 and Magainin I, (B) *E. coli* O157:H7 and Au NPs/TYR/Magainin I-C bioconjugates on the responses of the PEC immunoassay. The concentration of *E. coli* O157:H7 was 1.0×10<sup>5</sup> CFU/mL.

Similarly, as shown in Fig. S9, when both Magainin I and *E. coli* O157:H7, *E. coli* O157:H7 and Au NPs/TYR/Magainin I-C bioconjugates were incubated for 30 min, the best PEC detection results for E. coli O157:H7 were obtained.

Method	Material	Linear range (CFU/mL)	LOD (CFU/mL)	Reference
Colorimetry	CNTs	$1.0 \times 10^2 - 1.0 \times 10^5$	$1.0 \times 10^{2}$	[6]
CL	Luminol	$4.3 \times 10^{3} - 4.3 \times 10^{5}$	$1.2 \times 10^{3}$	[7]
ECL	Ru complex	$5.0 \times 10^2 - 5.0 \times 10^5$	$1.2 \times 10^{2}$	[8]
EC	Fc	$1.0 \times 10^{3} - 1.0 \times 10^{7}$	1.0×10 <sup>3</sup>	[9]
EC		$1.0 \times 10^{3} - 1.0 \times 10^{8}$	1.0×10 <sup>3</sup>	[10]
EC	AuNPs	$1.0 \times 10^2 - 1.0 \times 10^5$	$1.0 \times 10^{2}$	[11]
PEC	BiOI	$5.0 - 1.0 \times 10^{6}$	3.0	This work

Table S3. Comparison with different methods for the detection of E. coli O157:H7

CL: Chemiluminescence; ECL: Electrochemiluminescence; EC: Electrochemistry.

### Table S4. Spiked recovery tests for E. coli O157:H7 in skim milk

Spiked samples	Spiked <i>E. coli</i> O157:H7 (CFU/mL)	Found (CFU/mL)	Recovery (%)
1	0.0	Not found	
2	50.0	48.50	97.0
3	5.0×10 <sup>3</sup>	5.17×10 <sup>3</sup>	103.4
4	5.0×10 <sup>5</sup>	4.76×10 <sup>5</sup>	95.2

Table S5	Comparison	of results for	skim milk se	amples using t	the proposed	PEC sensor	and FLISA kit
Table 55.	Comparison	of results for	SKIIII IIIIK Se	impies using	ine proposed		

Spiked samples	Method; concentration:	t <sub>exp</sub>	
	mean±SD (RSD) (CFU/µL,		
	ELISA	PEC	
1	4.06±0.23	3.97±0.27	1.01
2	4.52±1.15	4.61±1.19	1.21
3	5.49±0.63	5.37±0.59	0.97
4	5.73±2.01	5.79±1.94	1.14
5	6.11±1.41	6.02±1.48	0.89
6	6.54±0.91	6.61±0.84	1.03
7	7.02±0.33	6.94±0.28	1.64

#### References

[1] Lu. Wang, Z. F. Gan, D. Guo, H. L. Xia, F. T. Patrice, M. E. Hafez and D. W. Li, *Anal. Chem.*, 2019, 91, 6507–6513.

- [2] S. Li, R. Hu, S. Q. Wang, X. D. Guo, Y. Zeng, Y. Li and G. Q. Yang, Anal. Chem., 2018, 90, 9296–9300.
- [3] L. J. Chai, J. Zhou, H. Feng, C. Tang, Y. Y. Huang and Z. S. Qian, *ACS Appl. Mater. Interfaces*, 2015, 7, 23564–23574.
- [4] Y. Teng, X. F. Jia, J. Li and E. K. Wang, Anal. Chem., 2015, 87, 4897-4902.
- [5] H. B. Yildiz, R. Freeman, R. Gill and I. Willner, Anal. Chem., 2008, 80, 2811–2816.
- [6] H. Zhang, Y. P. Shi, F. Lan, Y. Pan, Y. K. Lin, J. Z. Lv, Z. H. Zhu, Q. Jiang and C. Q. Yi, *Chem. Commun.*, 2014, **50**, 1848–1850.
- [7] Y. Zhang, C. Tan, R. H. Fei, X. X. Liu, Y. Zhou, J. Chen, H. C. Chen, R. Zhou and Y. G. Hu, *Anal. Chem.*, 2014, **86**, 1115–1122.
- [8] Z. J. Li, H. Y. Yang, L. J. Sun, H. L. Qi, Q. Gao and C. X. Zhang, Sens. Actuat. B: Chem., 2015, 210, 468–474.
- [9] Y. X. Li, R. Afrasiabi, F. Fathi, N. Wang, C. L. Xiang, R. Love, Z. She and H. B. Kraatz, *Biosens. Bioelectron.*, 2014, **58**, 193–199.
- [10] M. R. Akanda, V. Tamilavan, S. Park, K. Jo, M. H. Hyun and H. Yang, *Anal. Chem.*, 2013, 85, 1631–1636.
- [11] A. R. H. Abdel-Azzem Hassan, A. D. L. Escosura-Muñiz and A. Merkoçi, *Biosens*. *Bioelectron.*, 2015, **67**, 511–515.