

*Supporting Information*

# Functionalized nanozyme with drug loading for enhanced tumor combination treatment of catalytic therapy and chemotherapy

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## Chemodynamic performance

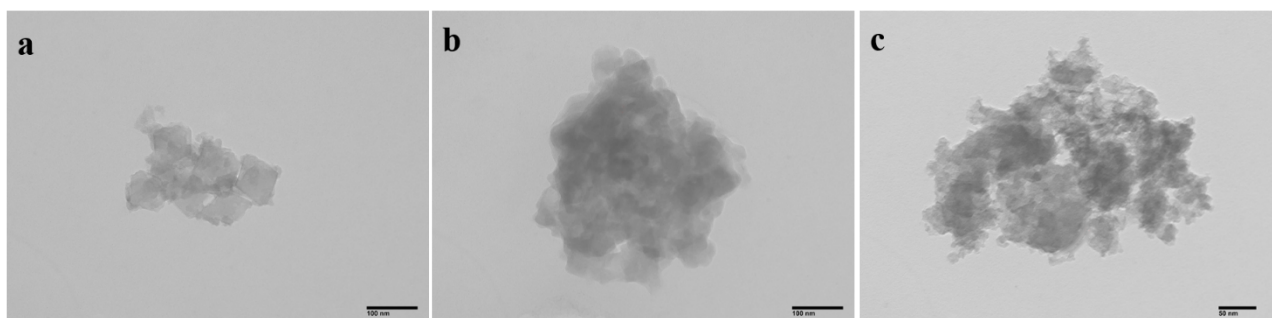
To clarify the catalytic activity of Zr/Ce-MOFs and Zr/Ce-MOFs/DOX/MnO<sub>2</sub>, MB was applied as an indicator to monitor ·OH production. MB can be degraded by ·OH produced from the disproportionation reaction between H<sub>2</sub>O<sub>2</sub> and Zr/Ce-MOFs and Zr/Ce-MOFs/DOX/MnO<sub>2</sub>, then assayed at 665 nm on a UV spectrometer. According to the time-dependent absorbance curve, all the corresponding average initial velocities of absorbance were calculated. Next, these average initial velocities were transformed to initial velocities (V<sub>0</sub>) of ·OH production via the Beer-Lambert law (eqn (1)), and then plotted as curves against the corresponding concentrations and fitted by the Michaelis-Menten equation (eqn (3)) to obtain the Michaelis-Menten constant (K<sub>M</sub>) and the maximum velocity (V<sub>max</sub>).

$$A=kb c \quad (1)$$

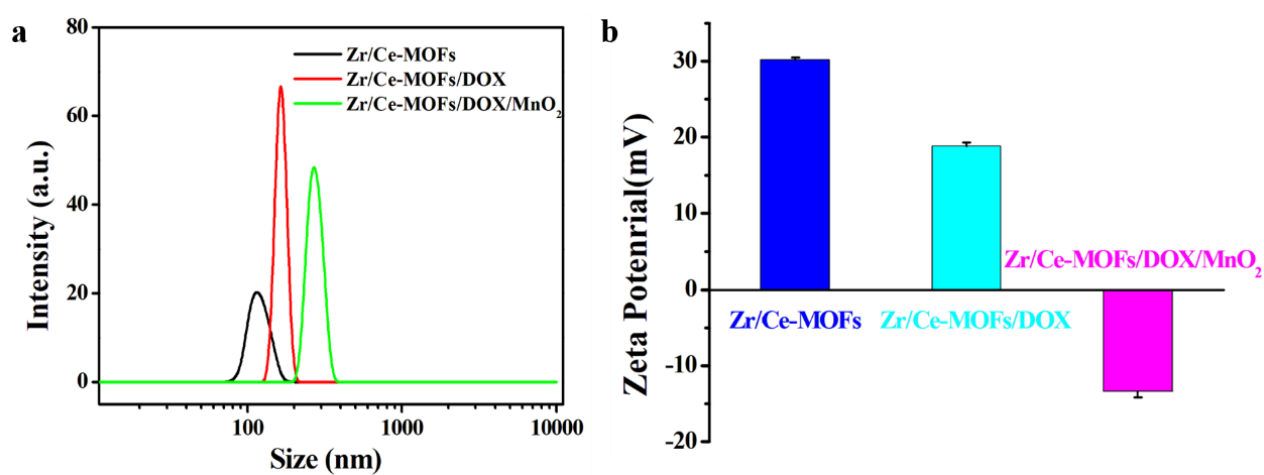
$$V_0=(V_{\max} \times [S]) / (K_M + [S]) \quad (2)$$

$$1/V_0=(K_M/V_{\max}) \times (1/[S]) + (1/V_{\max}) \quad (3)$$

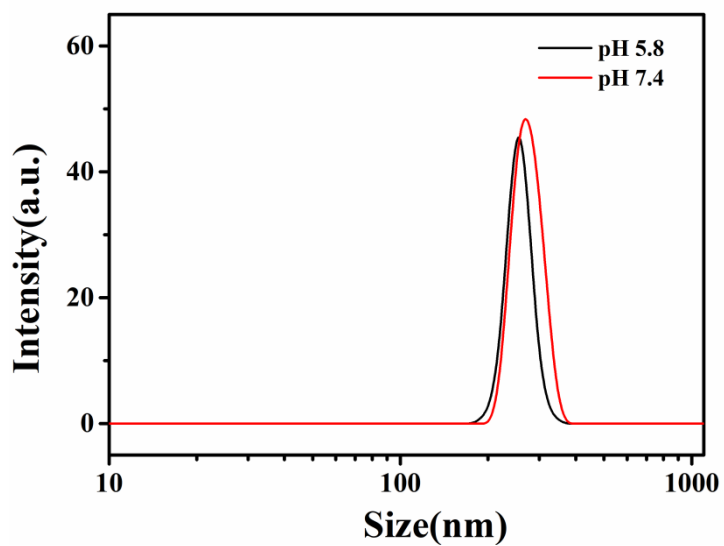
[S]-the concentration of H<sub>2</sub>O<sub>2</sub>



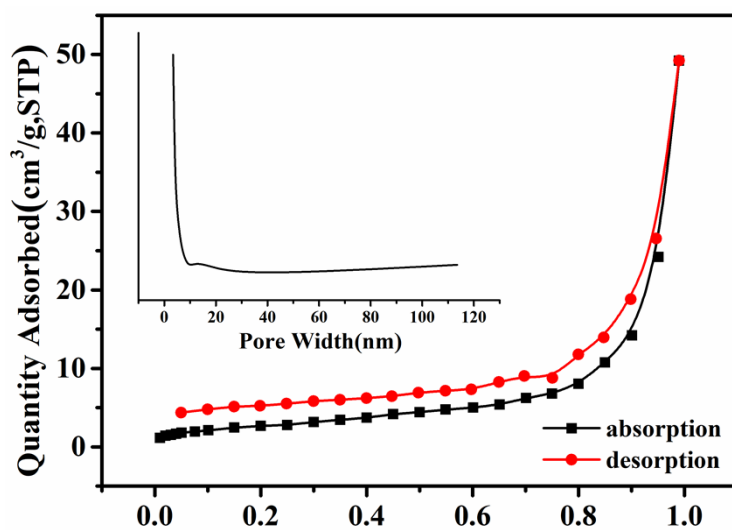
**Fig. S1.** TEM images of (a) Zr/Ce-MOFs, (b) Zr/Ce-MOFs/DOX and (c) Zr/Ce-MOFs/DOX/MnO<sub>2</sub>.



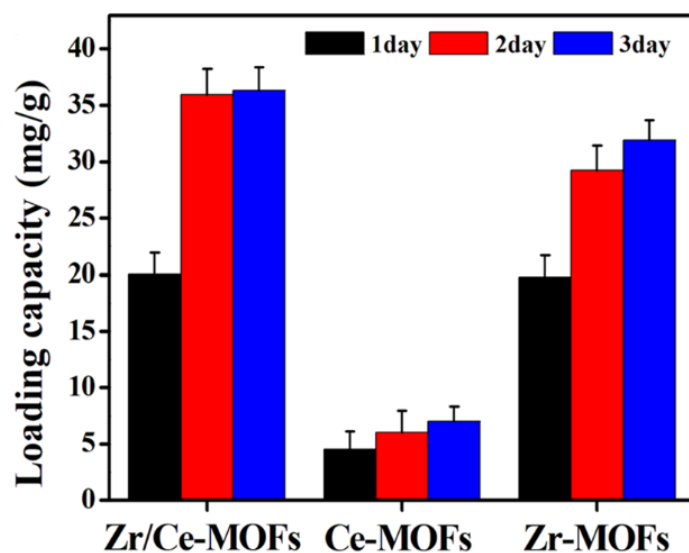
**Fig. S2.** (a) DLS and (b) Zeta potential of Zr/Ce-MOFs, Zr/Ce-MOFs/DOX and Zr/Ce-MOFs/DOX/MnO<sub>2</sub>. The error bar is the standard deviation of experimental data for three times.



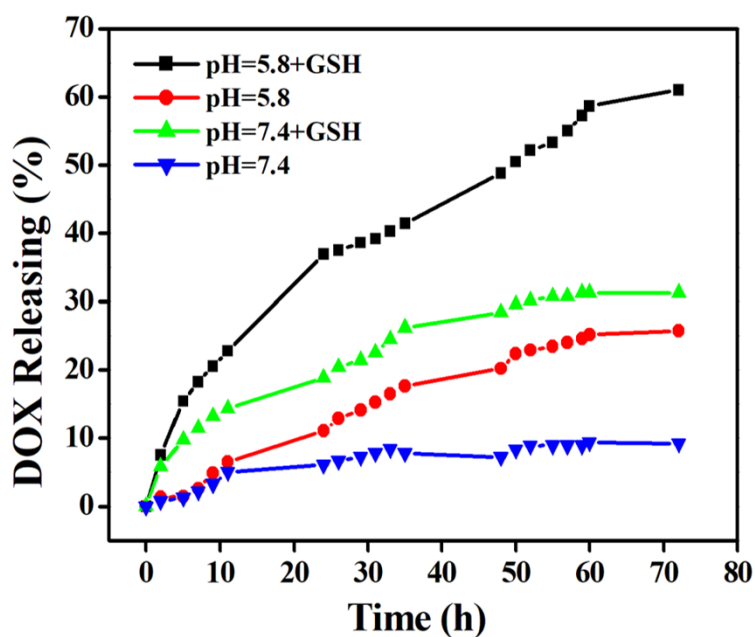
**Figure S3.** DLS size distribution of Zr/Ce-MOFs/DOX/MnO<sub>2</sub> in PBS of different pH.



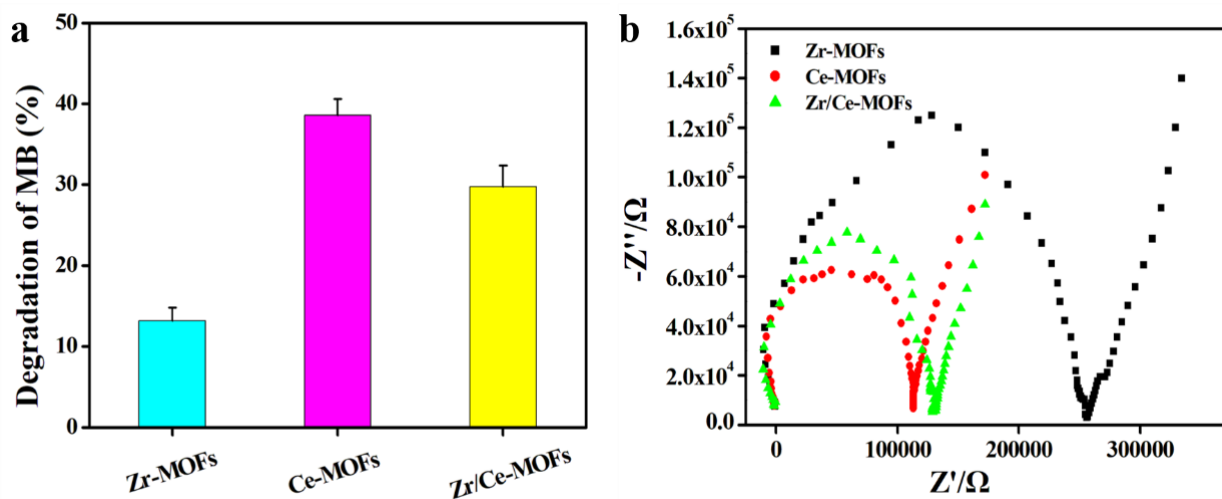
**Figure S4.** Nitrogen adsorption curve of Zr/Ce-MOFs.



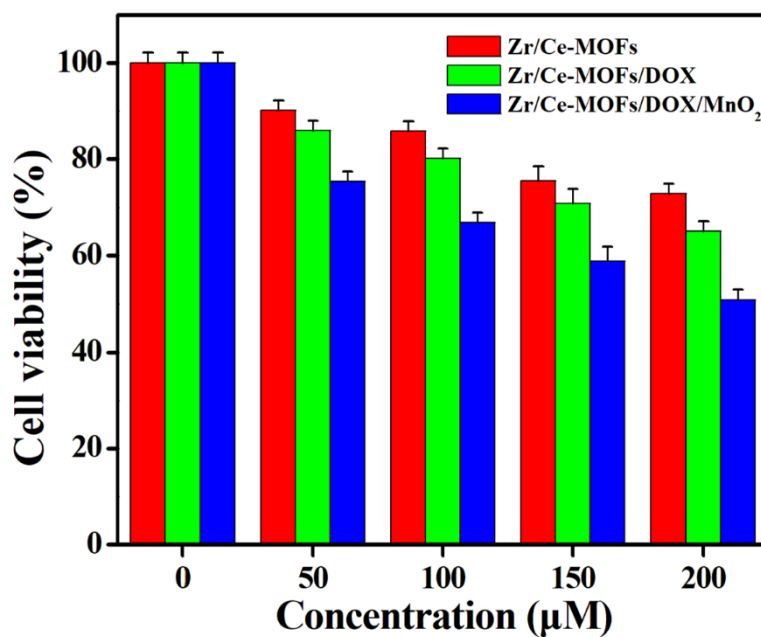
**Fig. S5.** The drug loading of different particle. The error bar is the standard deviation of experimental data for three times.



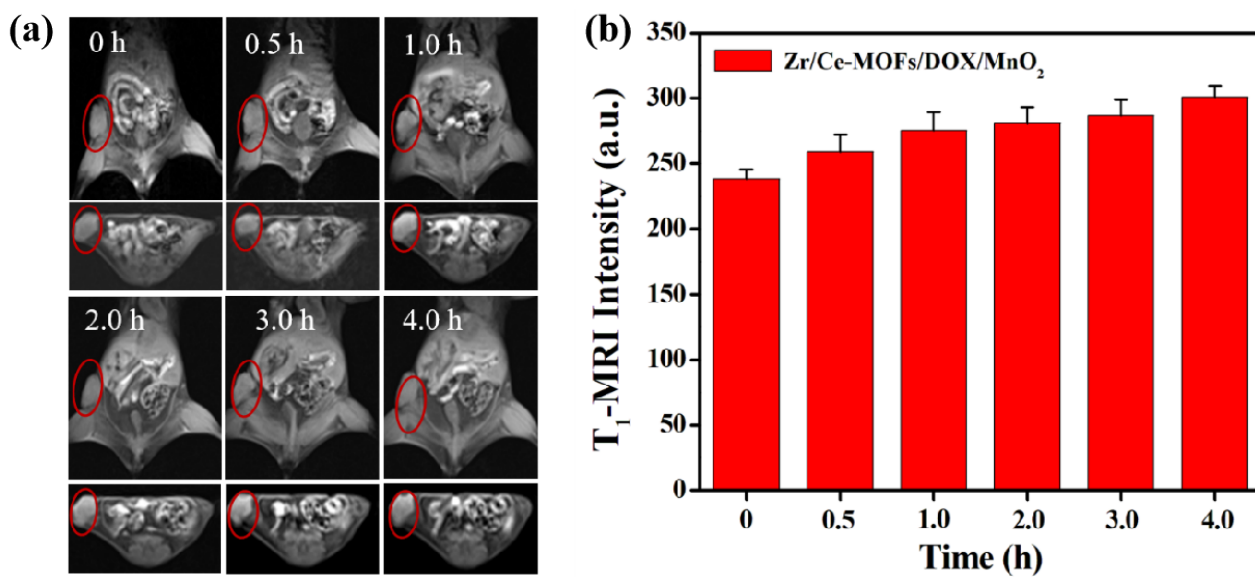
**Fig. S6.** The drug release of Zr/Ce-MOFs/DOX/MnO<sub>2</sub> at different pH values (pH = 7.4 and 5.8) and GSH concentrations (0 and 10 mM).



**Fig. S7.** (a) The degradation rate of MB of different materials; (b) EIS Nyquist plots of different materials. The error bar is the standard deviation of experimental data for three times.



**Fig. S8.** Cell viability of Zr/Ce-MOFs, Zr/Ce-MOFs/DOX and Zr/Ce-MOFs/DOX/MnO<sub>2</sub>. The error bar is the standard deviation of experimental data for three times.



**Fig. S9.** (a) T<sub>1</sub>-weighted images of tumor after injection with Zr/Ce-MOFs/DOX/MnO<sub>2</sub> at different time points; (b) The MRI signal at the injection site. The error bar is the standard deviation of experimental data for three times.