## **Supporting information**

A Fluorescent Quaternary Phosphonium Main-Chain Type Polymer: An Opportunity to Fabricate Functional Materials with Excellent Antibacterial Activity and Bacterial Imaging Capability

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### 1. <sup>1</sup>H and <sup>13</sup>C NMR Spectra of Compounds.



**Figure S1.** <sup>1</sup>H NMR spectrum of TPE-2CH<sub>3</sub> measured in CDCl<sub>3</sub> at room temperature.



**Figure S2.** <sup>13</sup>C NMR spectrum of TPE-2CH<sub>3</sub> measured in CDCl<sub>3</sub> at room temperature.



9.5 9.0 8.5 8.0 7.5 7.0 6.5 6.0 5.5 5.0 4.5 4.0 3.5 3.0 2.5 2.0 1.5 1.0 0.5 0.0 Chemical shift (ppm)

**Figure S3.** <sup>1</sup>H NMR spectrum of TPE-CH<sub>2</sub>-Br measured in CDCl<sub>3</sub> at room temperature.

**S5** 



**Figure S4.** <sup>13</sup>C NMR spectrum of TPE-CH<sub>2</sub>-Br measured in CDCl<sub>3</sub> at room temperature.



**Figure S5.** <sup>1</sup>H NMR spectrum of TPE-2CH<sub>2</sub>Br measured in CDCl<sub>3</sub> at room temperature.



**Figure S6.** <sup>13</sup>C NMR spectrum of TPE-2CH<sub>2</sub>Br measured in CDCl<sub>3</sub> at room temperature.



**Figure S7.** <sup>1</sup>H NMR spectrum of TPE-2N<sup>+</sup> measured in CDCl<sub>3</sub> at room temperature.



**Figure S8.** <sup>13</sup>C NMR spectrum of TPE-2N<sup>+</sup> measured in CDCl<sub>3</sub> at room temperature.



**Figure S9.** <sup>1</sup>H NMR spectrum of TPE-2P<sup>+</sup> measured in CDCl<sub>3</sub> at room temperature.



**Figure S10.** <sup>13</sup>C NMR spectrum of TPE-2P<sup>+</sup> measured in CDCl<sub>3</sub> at room temperature.



**Figure S11.** <sup>1</sup>H NMR spectrum of TPE-Ammonium polymer measured in DMSO- $d_6$  at room temperature.



**Figure S12.** <sup>1</sup>H NMR spectrum of TPE-Phosphonium polymer measured in CDCl<sub>3</sub> at room temperature.

## 2. <sup>31</sup>P NMR Spectra of Compounds



Figure S13. <sup>31</sup>P NMR spectrum of DPPH measured in CDCl<sub>3</sub> at room

temperature.



Figure S14. <sup>31</sup>P NMR spectrum of TPE-2P<sup>+</sup> measured in CDCl<sub>3</sub> at room

temperature.



Figure S15. <sup>31</sup>P NMR spectrum of TPE-Phosphonium polymer measured

in CDCl<sub>3</sub> at room temperature.

#### 3. UV-vis and PL Spectra of Compounds



**Figure S16**. Normalized UV-*vis* absorption spectra of TPE-2N<sup>+</sup> in DMF, DMSO, EtOH and MeOH.



**Figure S17**. Normalized UV-*vis* absorption spectra of TPE-2P<sup>+</sup> in DMF, DMSO, EtOH and MeOH.



**Figure S18**. Photoluminescence spectra of TPE-2N<sup>+</sup> and TPE-2P<sup>+</sup> in solid

state. ( $\lambda_{ex} = 316$  nm).



**Figure S19**. Normalized UV-*vis* absorption spectra of TPE-Ammonium polymer in DMF, DMSO, EtOH and MeOH.



**Figure S20**. Normalized UV-*vis* absorption spectra of TPE-Phosphonium polymer in DMF, DMSO, EtOH, MeOH and DCM.



**Figure S21.** Photoluminescence spectra of TPE-Ammonium polymer and TPE-Phosphonium polymer in solid state. ( $\lambda_{ex} = 316$  nm).

#### 4. FT-IR spectrum of Compounds



**Figure S22.** FT-IR spectra of TPE- $2N^+$ , TPE- $2P^+$  and polymers (KBr tablet).

#### 5. AIE test of two polymers



**Figure S23**. Photoluminescence spectra of (A) TPE-Ammonium polymer and (B) TPE-Phosphonium polymer at different ratios of water/ethanol. Inset: photographs of TPE-Ammonium polymer in ethanol and in water/ethanol mixtures when the addition of water was 95%, respectively, taken under illumination at 365 nm; (C) Plots of the maximum emission intensities of TPE-Phosphonium and TPE-Ammonium polymer with  $V_{water}$ :  $V_{EtOH}$  from 0 to 0.95. (Conc.: 1 mg/mL).

#### 6. Wettability test



**Figure S24**. Contact angle of (A) TPE-2N<sup>+</sup>, (B) TPE-2P<sup>+</sup>, (C) TPE-Ammonium polymer and (D) TPE-Phosphonium polymer.

### 7. SEM images



Figure S25. SEM images of the xerogel of TPE-Phosphonium polymer with CNCs. Scale bar:  $2 \mu m$ .

#### 8. EDS images



**Figure S26**. SEM images of the scanned area (A). Electron probe micro-analysis of xerogel tested by Energy Dispersive Spectrometer (EDS) (B). Scale bar: 5 μm.

#### 9. Antibacterial activity test



**Figure S27**. Antibacterial activities against *E. coli* (left) and *S. aureus* (right) of TPE-Phosphonium polymer (Conc.: 0.02 mg/mL).

#### **10. Bacterial imaging**



**Figure S28.** Fluorescence images of *E. coli* incubated with 0.2 mg/mL<sup>-1</sup> of TPE-Phosphonium polymer for 12 h, PI for 12h and merged. The bottom images were enlarged. Excitation wavelengths: TPE-Phosphonium polymer for 300 - 400 nm and PI for 440 - 550 nm. (scale bar =  $25 \mu$ m).



**Figure S29.** Fluorescence microscope images of *S. aureus* and *E. coli* incubated with TPE-Phosphonium polymer (0.2 mg/mL<sup>-1</sup>) for 12 h. Excitation wavelengths: 300 - 400 nm. (scale bar = 25 µm).

# 11. MIC data and the chemical structures of representative antibacterial agents

Samples	MIC (µg/mL)	
	S. aureus	E. coli
[Im][C] <sup>*</sup> (Polym. Chem-UK. <b>2018</b> , 9, 4611-4616)	460	440
[Qa][Cl]* (Polym. Chem-UK. 2018, 9, 4611-4616)	748	654
[DABCO][Cl][Br]*(Polym. Chem-UK. 2018, 9, 4611-4616)	2685	2350
S-P[Im][Cl]-L <sup>#</sup> (Polym. Chem-UK. 2018, 9, 4611-4616)	320	265
S-P[Im][Cl] <sup>#</sup> (Polym. Chem-UK. 2018, 9, 4611-4616)	275	235
S-P[Qa][Cl]-L <sup>#</sup> (Polym. Chem-UK. 2018, 9, 4611-4616)	365	285
S-P[Qa][Cl] <sup>#</sup> (Polym. Chem-UK. 2018, 9, 4611-4616)	257	220
S-P[DABCO][Cl][Br]-L <sup>#</sup> (Polym. Chem-UK. 2018, 9, 4611-4616)	887	787
S-P[DABCO][Cl][Br] <sup>#</sup> (Polym. Chem-UK. 2018, 9, 4611-4616)	550	497
Q-PGEDA-OP/TPE# (Chem. Mater. 2018, 30, 1782-1790)	15.5	125
Q-PGEDA-OA <sup>#</sup> (Chem. Mater. 2018, 30, 1782-1790)	31.5	500
P4VP-ManTPE <sup>#</sup> ( <i>Biomacromolecules</i> <b>2021</b> , 22, 2224-2232)	15.63	1000
P4VP-BPTPE <sup>#</sup> (Biomacromolecules <b>2021</b> , 22, 2224-2232)	7.81	500
M-P[Im][Br]& (Polym. Chem-UK. 2018, 9, 4611-4616)	52	26
S-P[Qa][Cl] <sup>#</sup> (Polym. Chem-UK. 2018, 9, 4611-4616)	120	58
M-P[Qa][Br] & (Polym. Chem-UK. 2018, 9, 4611-4616)	200	180
Diamine polymer 1 <sup>&amp;</sup> (Biomaterials 2017, 127, 36-48.)	3.9	7.8
Diamine polymer 2 <sup>&amp;</sup> (Biomaterials 2017, 127, 36-48.)	7.8	31.3
Diamine polymer 4 <sup>&amp;</sup> (Biomaterials 2017, 127, 36-48.)	3.9	3.9
Diamine polymer 1a <sup>&amp;</sup> (Biomaterials 2017, 127, 36-48.)	1.95	3.9
Diamine polymer 4a <sup>&amp;</sup> (Biomaterials 2017, 127, 36-48.)	1.95	3.9
Diamine polymer 5 <sup>&amp;</sup> (Biomaterials 2017, 127, 36-48.)	3.9	3.9
Diamine polymer 6 <sup>&amp;</sup> (Biomaterials 2017, 127, 36-48.)	3.9	7.8
TPA-Ammonium polymer <sup>&amp;</sup> (Our work)	1.95	3.90
TPE-Phosphonium polymer <sup>&amp;</sup> (Our work)	0.24	0.98

Table S1 MIC data from previous works and our work.

\*Small Cationic compound. #Side-chain polymer. &Main-chain polymer.



#### 12. Density functional theory (DFT) calculations



Figure **S30**. The chemical structure used for density functional theory (DFT) calculations (Upper left), and the optimized configuration (Lower left), The ESP diagram of TPE-Phosphonium polymer's molecular fragment (Right).

#### **13.** Cytotoxicity



**Figure S31.** HeLa cells were incubated with TPE-Phosphonium polymer for 24h at different concentration: 0, 5, 10, 20 and 40  $\mu$ g/mL.

#### 14. Cell imaging



**Figure S32.** Fluorescence images of HeLa cells incubated with 20  $\mu$ g/mL of TPE-Phosphonium polymer for 12 h (Excitation wavelength: 405 nm), the bright field images and the overlay images.