

Polyarylether-based COFs coordinated by Tb³⁺ for fluorescent detection of anthrax-biomarker dipicolinic acid

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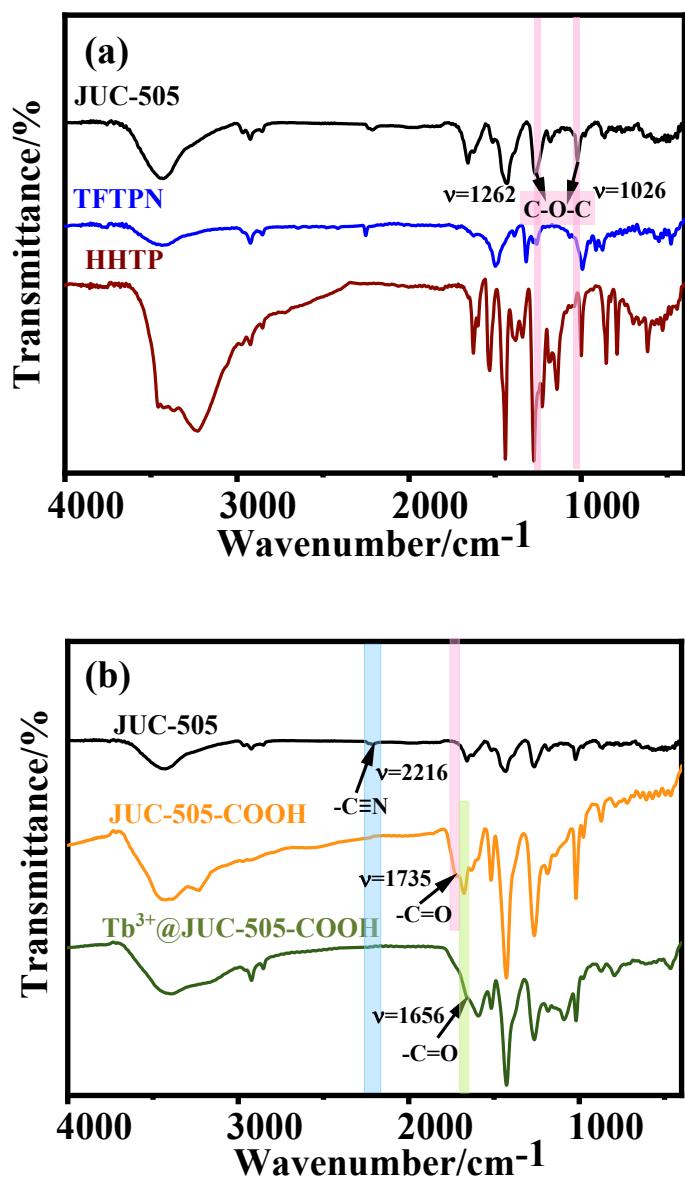


Figure S1. (a)FT-IR spectra of JUC-505, (b) FT-IR spectra of JUC-505, JUC-505-COOH, and Tb^{3+} @JUC-505-COOH.

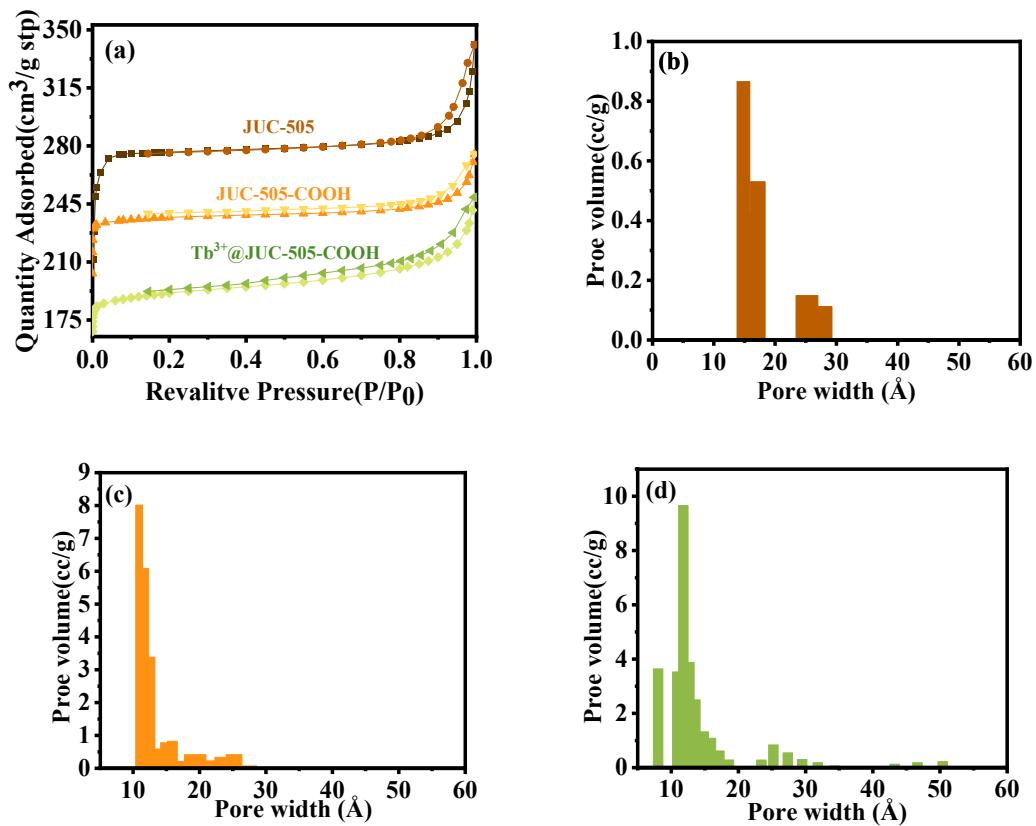


Figure S2. (a) N_2 adsorption and desorption isotherms for JUC-505 (brown curve), JUC-505-COOH (orange curve) and $\text{Tb}^{3+}@JUC-505\text{-COOH}$ (green curve) measured at 77 K. Pore size distributions of JUC-505 (b), JUC-505-COOH (c) and $\text{Tb}^{3+}@JUC-505\text{-COOH}$ (d)

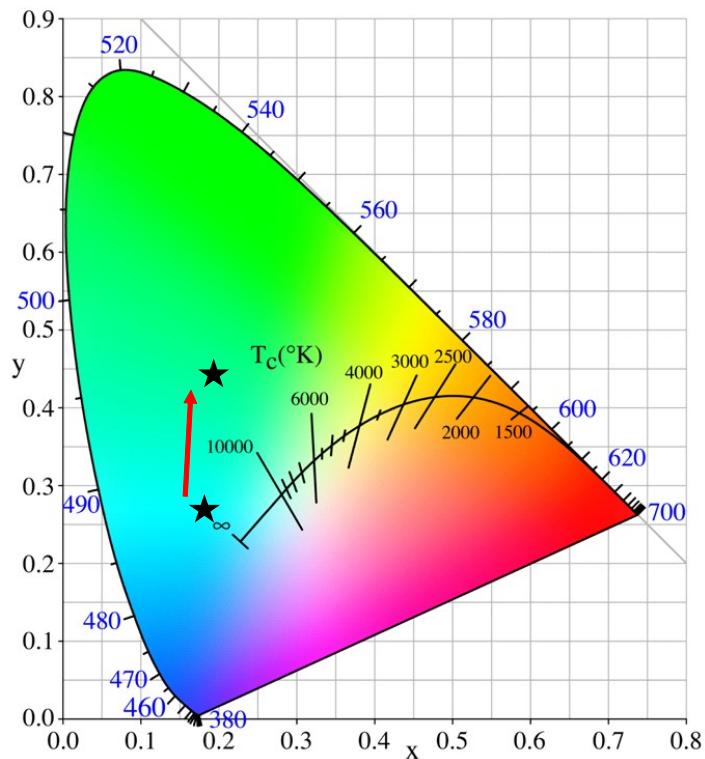


Figure S3. CIE chromaticity diagram of Tb^{3+} @JUC-505-COOH.

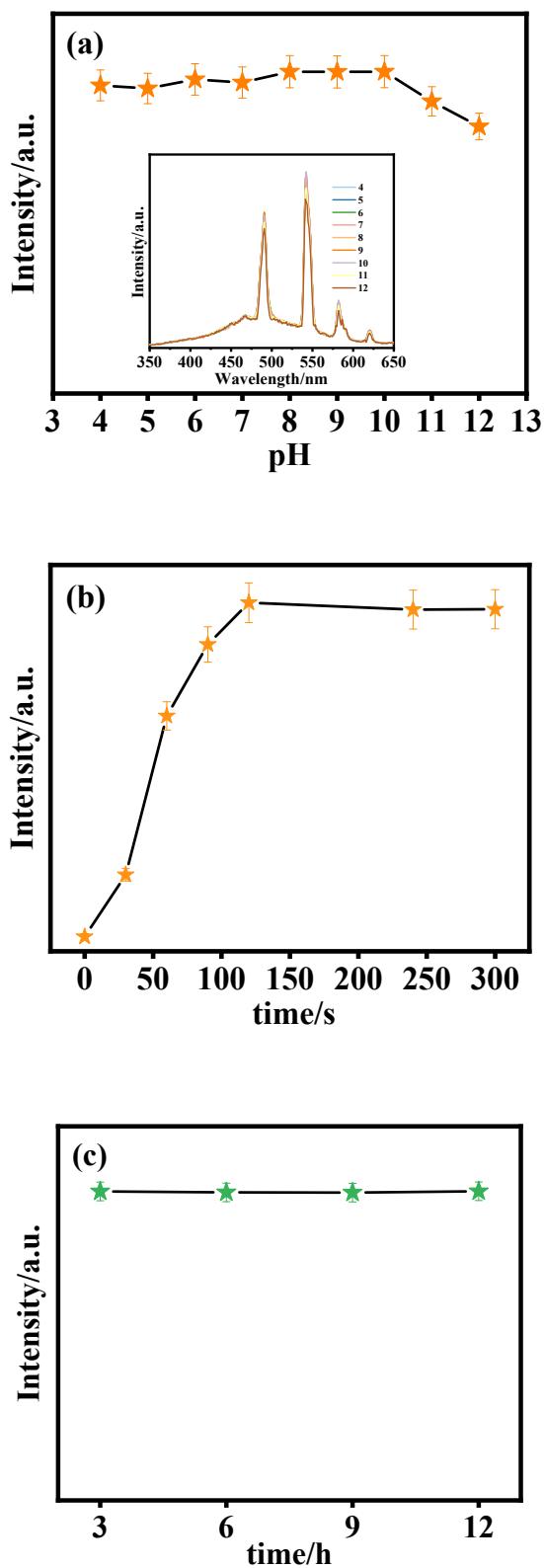


Figure S4. **(a)** Influence of pH on Tb^{3+} @JUC-505-COOH/DPA system. **(b)** Time-dependent fluorescence intensity of Tb^{3+} @JUC-505-COOH suspensions in DPA, **(b)**

time-light stability of the Tb^{3+} @JUC-505-COOH in the presence of DPA.

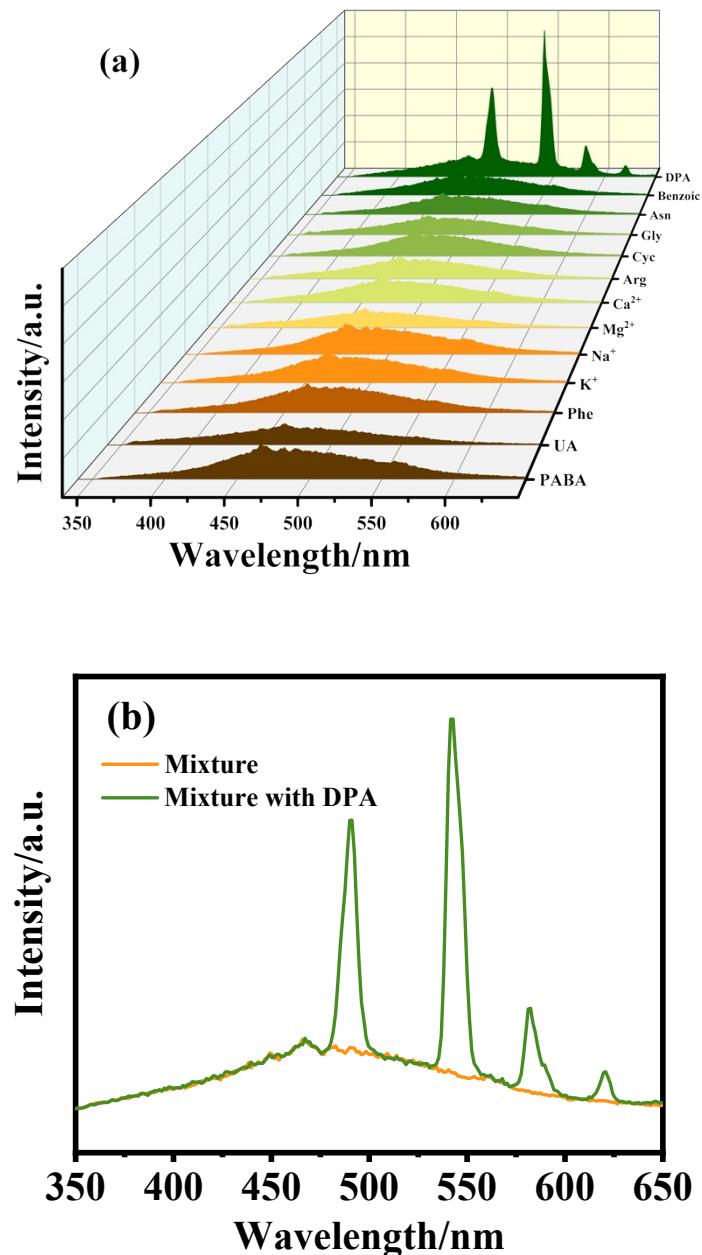


Figure S5. Emission spectra of Tb^{3+} @JUC-505-COOH introduced into different body fluids components **(a)** and their mixture **(b)**.

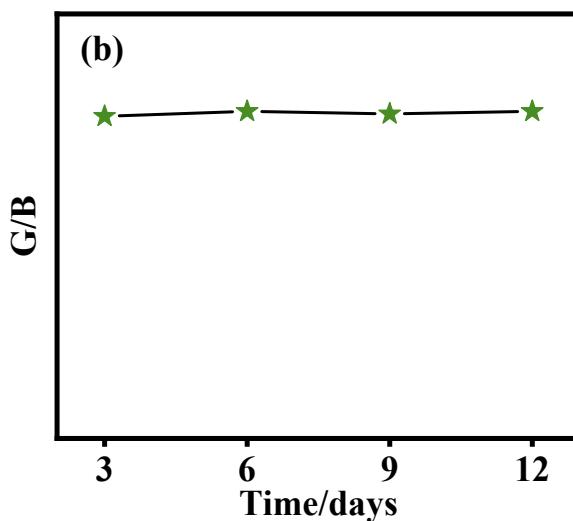
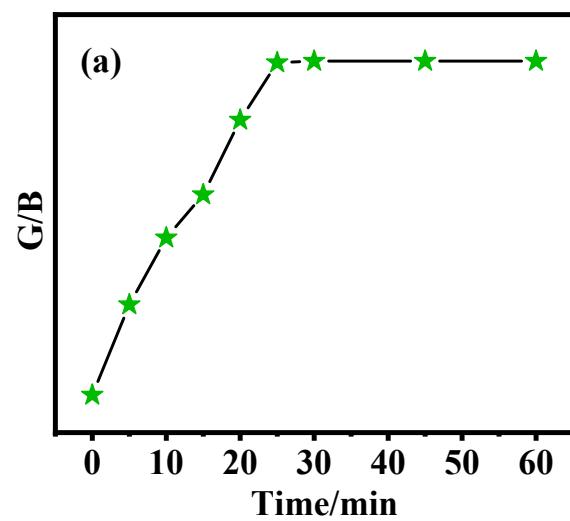


Figure S6. (a) Time-dependent G/B intensity of paper-based probe in DPA, (b) time-light stability of the paper-based probe in the presence of DPA.

Table S1 Comparison of the DPA detection properties of Tb^{3+} @JUC-505-COOH with other sensor reported in previous literatures.

Materials	Linear range (μM)	LOD (μM)	Reference s
R6H@Eu(BTC)	0-120	4.5	[S1]
His@ZIF-8/ Tb^{3+}	0.08-10	0.02	[S2]
Eu/CdTe	20-100	1.72	[S3]
Eu-PB MOG 1	0-50	0.298	[S4]
EBT-Eu	2-10	2	[S5]
Tb^{3+} @JUC-505-COOH	0-120	0.615	This work

References

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