

Supporting Information

Formamidinium Iodide for Instantaneous and Fluorescent Detection of Pb^{2+} in Water

Md Ashiqur Rahman Laskar^{†, a}, Md Tawabur Rahman^{†, a, b}, Khan Mamun Reza^a, Abdullah Al Maruf^c, Nabin Ghimire^a, Brian Logue^d, Qiquan Qiao^{a, e*}

^aDepartment of Electrical Engineering and Computer Science, South Dakota State University, Brookings, SD 57007, USA

^bDepartment of Electrical and Electronic Engineering, Khulna University of Engineering & Technology, Khulna 9203, Bangladesh

^cDepartment of Physics, South Dakota State University, Brookings, SD 57007, USA

^dDepartment of Chemistry and Biochemistry, South Dakota State University, Brookings, SD 57007, USA
^eDepartment of Mechanical and Aerospace Engineering, Syracuse University, Syracuse, NY 13244, USA

[†]These authors contributed equally

***Email:** quqiao@syr.edu, tawabur@eee.kuet.ac.bd

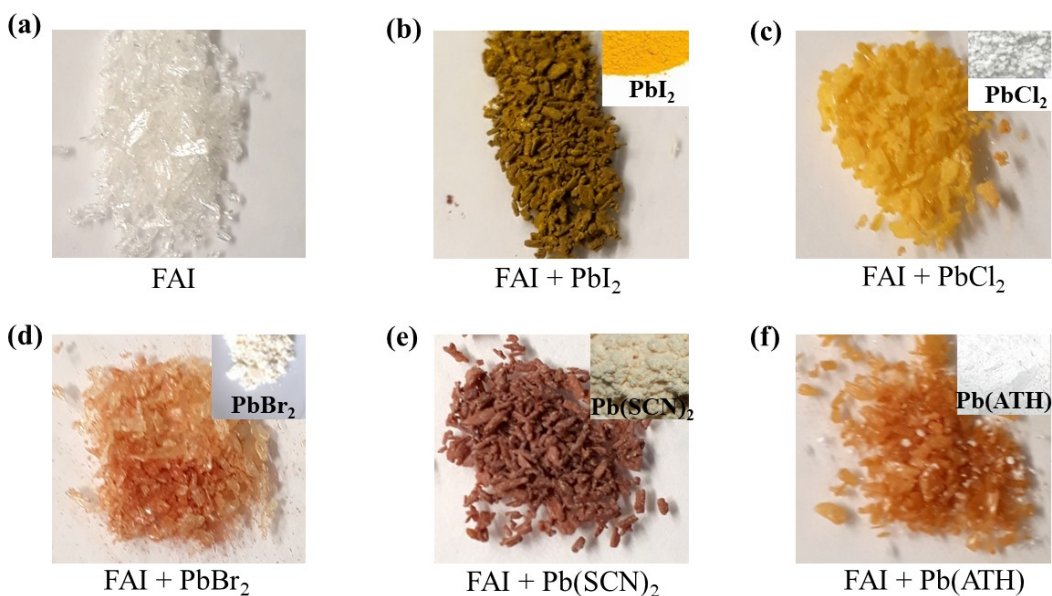


Figure S1. a) FAI powder. b-f) Photographs of FAI after mixing (5:1) with different Pb^{2+} species like PbI_2 , PbCl_2 , PbBr_2 , $\text{Pb}(\text{SCN})_2$, $\text{Pb}(\text{CH}_3\text{COO})_2 \cdot 3\text{H}_2\text{O}$. Here $\text{Pb}(\text{ATH})$ stands for Lead (II) Acetate Trihydrate or $\text{Pb}(\text{CH}_3\text{COO})_2 \cdot 3\text{H}_2\text{O}$.

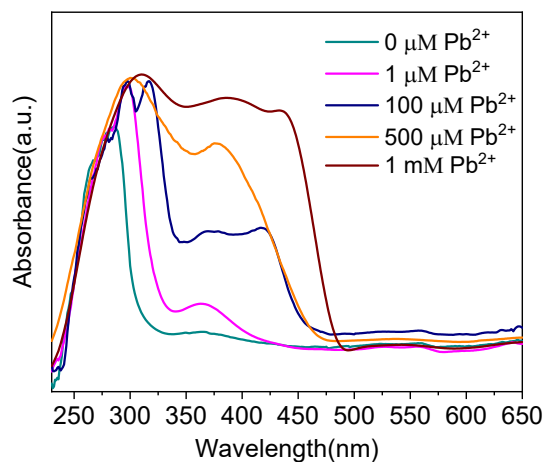


Figure S2. UV-Vis absorbance spectra of sensing solutions with different concentrations of Pb^{2+} .

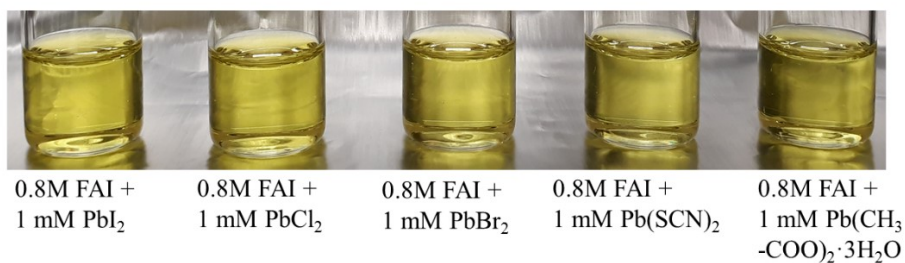


Figure S3. Images of sensing solutions with 1 mM Pb^{2+} from different source compounds.

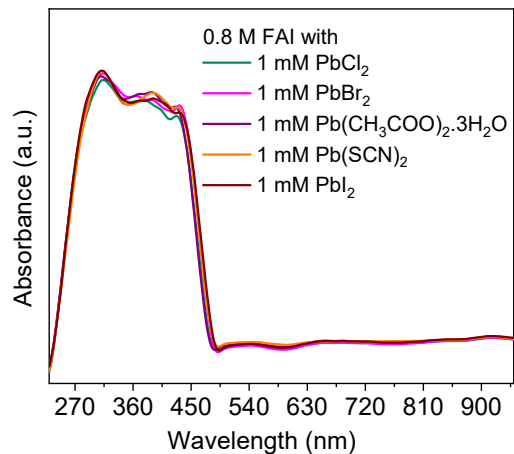


Figure S4. UV-Vis absorbance spectra of sensing solutions with 1 mM Pb^{2+} from different source compounds.

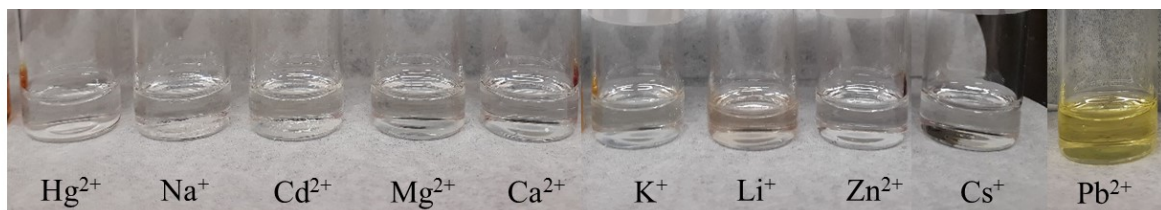


Figure S5. Images of sensing solutions with different metal ions at 500 μM .

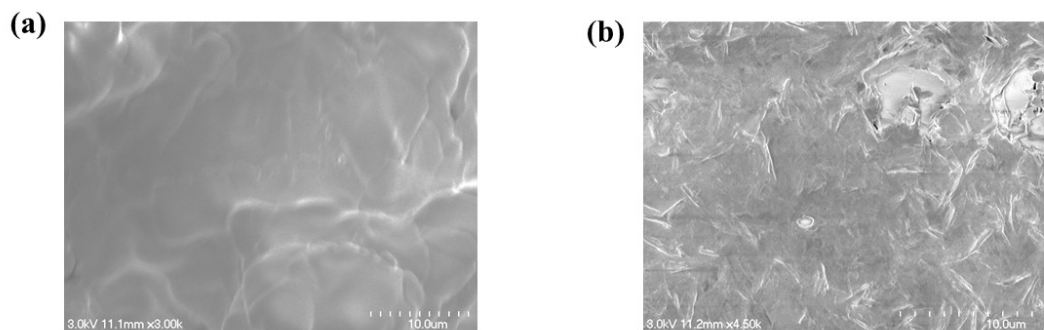


Figure S6. SEM images of a) FAI b) PbI_2 .

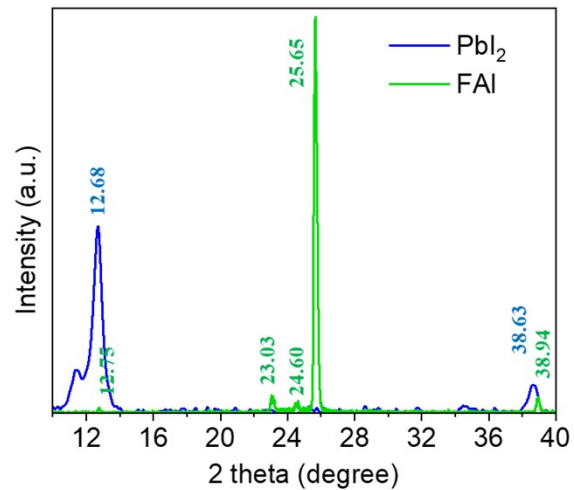


Figure S7. XRD of FAI and PbI_2 .

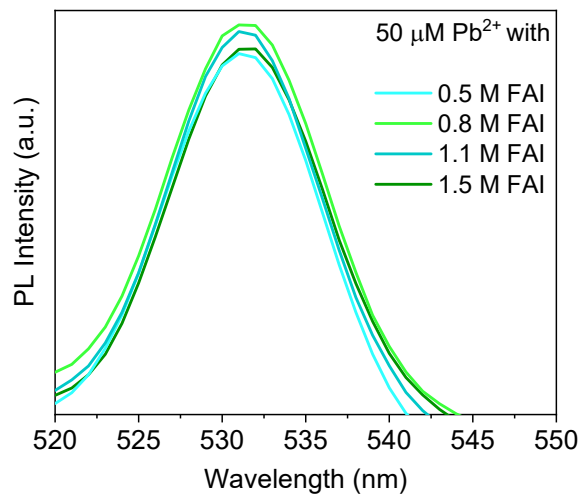


Figure S8. PL spectra of FAPbI_3 solutions at different FAI concentrations.

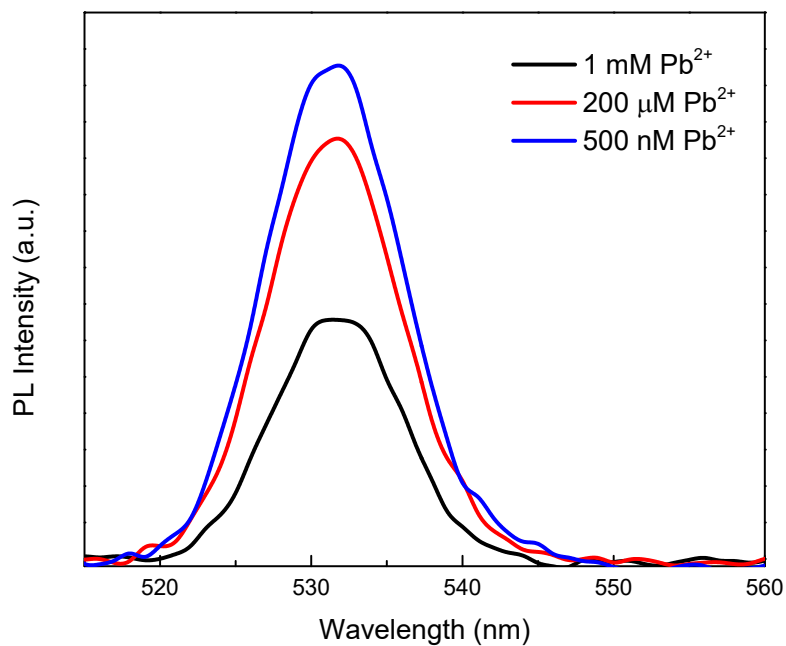


Figure S9. PL spectra of FAPbI₃ solutions with tap water samples

Table S1: Electrostatic potential (ESP) of FA⁺ under vacuum, DMF, and water

Material	Atom	Vacuum (kJ/mol)	DMF (kJ/mol)	Water (kJ/mol)
FA ⁺	N-C-N ^a	735.258	740.369	740.814

^aElectrons are delocalized on the resonance structure of N-C-N

Table S2. Comparison of sensing performance among different Pb²⁺ sensors.

Modifications	LOD (nM)	LDR (nM)	Methods	References
FAPbI ₃ perovskite	100 nM	100 nM - 1 mM	Colorimetric and Fluorometric	This work
CH ₃ NH ₃ PbBr ₃ perovskite	1.6 mM	1.6 - 200 mM	Fluorometric	[1]
AlGa _x N _{1-x} /Ga _x N _{1-x}	0.0576 nM	0.5 - 20 nM	HEMT	[2]
1T and 2H MoS ₂	0.031 nM	-	Electrochemical	[3]

Metal-organic framework (MOF)	7.7 pM	-	Fluorometric	[4]
Delonix regia leaf-derived CQDs	3.3 nM	10 - 180 μ M	Electrochemical	[5]
Ga ₂ O ₃ NPs	84 nM	0.3 - 80 μ M	Electrochemical	[6]
TiO ₂ @Gum Arabic-Carbon Paste Electrode	101.2 nM	5 - 50 nM	Electrochemical	[7]

HEMT: High electron mobility transistor; MoS₂: Molybdenum disulfide; NPs: Nanoparticles; CQDs: Carbon quantum dots.

References

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