Supplementary Information

Enhanced efficiency and stability of blue perovskite light-emitting

diodes through dual defect passivation

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Experimental section

Materials and Preparation. The precursor solutions of perovskites were prepared by dissolving PEGDA, TPPB, CsBr, PbBr₂ and PbCl₂ with a molar ratio of x/0.075/1.5/0.37/0.63 at a concentration of 10 wt.%. The post-treatment solutions were prepared by dissolving various concentration PEGDA in toluene.

Film and Device Fabrication. The perovskite films were spin-coated on PEDOT:PSS (Clevios P VP 4083) layer with a spin-coating speed of 6000 rpm, and then annealed at 130 °C for 15 min. The post-treatments were conducted through spin-coating PEGDA in toluene solutions on annealed perovskite films with a speed of 3000 rpm. Finally, the TPBi electron-transport layer, and the LiF/Al electrode were thermally evaporated, respectively.

Device Characterization. All devices were characterized through a system combining a fibre integration sphere (FOIS-1) coupled with a Keithley 2400 source meter and a QE-6500 spectrometer. The stability of devices was measured in a nitrogen-filled glovebox.

Film Characterization. The PL spectra were obtained by using a QE65 Pro spectrometer, with perovskite films excited by a 375 nm CW laser. The excitation-intensity-dependent PLQEs were obtained by a PLQE-LD CP001 system (Nanjing Ouyi Optoelectronics Technology). The in-situ PL spectra were measured by an ISPL-HI001 system (Nanjing Ouyi Optoelectronics Technology). The perovskite films were excited by a 375 nm CW laser and the PL spectra were collected by using a QE Pro spectrometer. Time-resolved photoluminescence spectra were measured by using an Edinburgh FLS980. The films were excited by a 375 nm pulsed laser with an intensity of 4 nJ cm⁻². XPS spectra were measured by a PHI5000 VersaProbe. FTIR spectra were recorded by using a Thermo Scientific Nicolet iS50. The samples were spin-coated on substrates and measured with a reflection accessory. The XRD data were obtained by a RIGAKU Smartlab 3kW X-ray diffractometer. The SEM measurements were performed by using a JEOL JSM-7610F plus scanning electron microscope.



Fig. S1. Characteristics of perovskite films and devices with PEGDA additive. (a) EL spectra of PEGDA_{0.5}–CsPb(Br_{0.65}Cl_{0.35})₃ perovskite LED under various biases. (b) XRD data of perovskites without and with 0.5 mg mL⁻¹ PEGDA. (c-d) SEM images of perovskites without (c) and with 0.5 mg mL⁻¹ PEGDA (d). Scale bar, 1 μ m.



Fig. S2. Characteristics of perovskite films with PEGDA post-treatment. (a) Contour plot of in-situ PL spectra of perovskites with 4 mg mL⁻¹ PEGDA. (b) Excitation-intensity-dependent PLQEs of PEGDA_{0.5}–CsPb(Br_{0.65}Cl_{0.35})₃ films without and with 4 mg mL⁻¹ PEGDA. (c) Time-resolved PL decay transients of PEGDA_{0.5}–CsPb(Br_{0.65}Cl_{0.35})₃ films without and with 4 mg mL⁻¹ PEGDA. (d) SEM image of perovskite with 4 mg mL⁻¹ PEGDA. Scale bar, 1 μ m. (e) XRD data of perovskites without and with 4 mg mL⁻¹ PEGDA.



Fig. S3. Characteristics of the control $CsPb(Br_{0.65}Cl_{0.35})_3$ perovskite and LEDs with 4 mg mL⁻¹ PEGDA. (a) Excitation-intensity-dependent PLQEs. (b) Current density and luminance versus driving voltage. (c) Dependence of EQE versus current density. (d) EL spectra.



Fig. S4. Characteristics of devices based on PEGDA_{0.5}–CsPb(Br_{0.65}Cl_{0.35})₃ perovskite with 4 mg mL⁻¹ PEGDA post-treatment. (a) Histogram of peak EQEs from 50 devices. (b) EL spectra of device under continuous measurement.