Supporting Information

Constructing efficient and stable CsPbBr₃ nanocrystals via calcium and fluorine ions combined-treatment for light-emitting diodes

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Fig. S1 (a) Optical properties of PNCs with different amount of DDAF in step one.(b) The corresponding normalized absorption and PL spectra.



Fig. S2 The PL spectra of PNCs with different treated time in step two. (The inset represents the corresponding PL intensity changes.)



Fig. S3 (a) The PL spectra of PNCs treated with different amount of DDAF in step three. (The inset represents the corresponding PL intensity changes.) (b) The ambient light images of different DDAF treated PNCs with adding anti-solvent in step three.



Fig. S4 Photographs of photoluminescence efficiency of (a) CsPbBr₃, (b) CsPbBr₃@F, (c) CsPbBr₃@F·Ca and (d) CsPbBr₃@F·CaF PNCs.



Fig. S5 The photostability and thermal quenching curves of CsPbBr₃ PNCs



Fig. S6 (a) TEM (inset was the corresponding size distribution), (b) HRTEM, (c) HAADF-STEM and the corresponding elemental mapping of (d) Cs, (e) Pb and (f) Br of CsPbBr₃ PNCs.



Fig. S7 The elemental mapping of Cs, Pb and Br in a CsPbBr₃@F·CaF PNC sample.



Fig. S8 (a) XPS survey spectra and (b) high-resolution XPS spectra of Br 3*d* of CsPbBr₃, CsPbBr₃@F, CsPbBr₃@F·Ca and CsPbBr₃@F·CaF PNCs.



Fig. S9 (a) FTIR of CsPbBr₃, CsPbBr₃@F, CsPbBr₃@F·Ca, CsPbBr₃@F·CaF nanocrystals. (b) The enlarged images of area I and II of picture (a).



Fig. S10 ¹H NMR of CsPbBr₃, CsPbBr₃@F, CsPbBr₃@F·Ca, CsPbBr₃@F·CaF PNCs.



Fig. S11 SEM images of (a) CsPbBr₃@F and (b) CsPbBr₃@F·CaF films.



Fig. S12 AFM images of (a) CsPbBr₃@F and (b) CsPbBr₃@F·CaF films.



Fig. S13 UPS spectra of (a) CsPbBr₃@F and (b) CsPbBr₃@F·CaF PNCs samples.



Fig. S14 Current density as a function of voltage in (a) hole-only and (b) electrononly devices.

Samples	A ₁ (%)	τ ₁ (ns)	A2 (%)	τ ₂ (ns)	τ _{avg.} (ns)	A ₁ ' (%)	A2 ['] (%)	Ф (%)	Κ _r (μs ⁻¹)	K _{nr} (μs ⁻¹)
CsPbBr ₃	76.86	7.80	21.50	25.88	16.50	51.86	48.14	85.06	51.54	9.05
CsPbBr ₃ @F	85.55	6.96	16.22	24.54	14.00	59.95	40.05	91.44	65.29	6.11
CsPbBr ₃ @F·Ca	87.51	6.52	14.66	17.36	9.86	69.15	30.85	96.02	97.36	4.04
CsPbBr ₃ @F·CaF	96.81	6.61	6.91	19.50	8.85	82.60	17.40	98.95	111.77	1.19

Table S1 Summary of time-resolved PL bi-exponential fitting parameters forsolutions of the PNCs.

Note:

(1) A and τ are the amplitude and decay time constant respectively. τ_{avg} is calculated as

$$\tau_{avg} = \frac{A_1 \tau_1^2 + A_2 \tau_2^2}{A_1 \tau_1 + A_2 \tau_2}$$

(2) A_x ' represented the proportion of $\tau_x,$ recalculated as

$$A'_{x} = \frac{A_{x}\tau_{x}}{A_{1}\tau_{1} + A_{2}\tau_{2}}$$
 x = 1 or 2

(3)

$$\Phi = \frac{k_r}{k_r + k_{nr}} \ \tau_{avg.} = \frac{1}{k_r + k_{nr}}$$

kr represents radiative combination rate

 k_{nr} represents nonradiative combination rate