Supporting material to Surface Chemical Mechanisms of Trimethyl Aluminum in Atomic Layer Deposition of AlN

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Effect of spin polarization

The energy difference between non-spin polarized and spin polarized calculations and magnetization of structures A_0 , B_0 , C_1 , C_2 , BC_1 , and BC_2 are given in Table SI. For the minima structures the magnetization when considering spin polarization was negligible and the difference in energy between spin polarized calculation and non-spin polarized were less than 1.5 kJ mol⁻¹. This agrees with the expected closed shell configuration of these structures as the system can form full octets. For the transition state BC_1 the magnetization was found to be 1.88 indicating a triplet state with the energy being around 26 kJ mol⁻¹ less for the spin polarized triplet compared to the non-polarized singlet state. The transition state BC_2 has a magnetization of 0.94 with one unpaired electron and an energy difference of -5 kJ mol⁻¹.

	Electronic energy	Electron energy	Energy	Magnetization
	non-spin polarized	spin polarized (eV)	difference	(spin polarized)
	(eV)		(kJ mol ⁻¹)	
A ₀	-1564.550	-1564.550	0.0	0.00
B ₀	-1565.507	-1565.507	0.0	0.00
C ₁	-1566.506	-1566.518	-1.1	0.05
C ₂	-1570.761	-1570.761	0.0	0.00
BC ₁	-1564.993	-1565.269	-26.6	1.88
BC ₂	-1565.191	-1565.242	-5.0	0.94

Adsorption and Decomposition structures

The adsorption and decomposition structures of a first TMA molecule on amino terminated AlN are shown in Figs. S1 - S8. Surface oxidation and last methyl protonation are shown in Fig. S9 - S12. Adsorption and decomposition of a second TMA molecule are shown in Figs. S13 - S21.



Figure S1. Amino terminated (0001)-AlN surface A_0 .



Figure S2. Adduct between surface amino group and TMA molecule (B_0) .



Figure S3. Methyl protonation transition state (BC_1) from TMA adduct.



Figure S4. DMA fragment (C_1) bridging adduct amino group and deprotonated amino group.



Figure S5. Reductive elimination transition state (CD_1) from DMA fragment (C_1) . Methyl groups are pushed together.



Figure S6. Aluminum adatom bridging two adduct amino group and one deprotonating amino group (D₁).



Figure S7. Reductive elimination transition state (CD_2) from TMA adduct (B_0) .



Figure S8. MMA fragment bridging three adduct amino groups (C_2).



Figure S9. Methyl protonation transition state (CD_2) from MMA fragment.



Figure S10. MMA fragment bridging one adduct amino groups (D_3) and two deprotonated amino groups after oxidation of the surface.



Figure S11. Methyl protonation transition state (DE₃) from MMA fragment at oxidized surface.



Figure S12. Aluminum adatom bridging three deprotonated amino groups (E_3) .



Figure S13. Adduct between surface amino group and TMA molecule (D_5) with a neighboring MMA fragment.



Figure S14. Methyl protonation transition state (DE₅) from a TMA adduct with neighboring MMA fragment.



Figure S15. DMA fragment bridging two adduct amino groups and a deprotonated amino group (E_5) , shared with an MMA fragment.



Figure S16. Methyl protonation transition state (\mathbf{EF}_5) from DMA fragment with neighboring MMA fragment.



Figure S17. Bridging MMA fragment between two deprotonated amino groups and one adduct amino group (F_5) with neighboring MMA fragment.



Figure S18. Reductive elimination transition state (DE_6) from a TMA adduct with neighboring MMA fragment.



Figure S19. Bridging MMA fragment three adduct amino groups (E_6) with neighboring MMA fragment.



Figure S20. Reductive elimination transition state (**EF**₇) from a DMA fragment bridging three amino groups with neighboring MMA fragment.



Figure S21. Bridging aluminum adatom between two adduct amino groups and one deprotonated amino group (**F**₇) with neighboring MMA fragment.