Electronic Supporting Information (ESI) for

The role of terminal coordinated amides in a series of Ca-tatb frameworks: pore size regulation and fluorescence sensing tunability

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Figure S1 The crystal image of the as-made compound 1.



Figure S2 The crystal image of the as-made compound 2.



Figure S3 The crystal image of the as-made compound 3.



Figure S4 PXRD patterns of the as-made compounds 1, 2 and 3. Simulated PXRD pattern of compound 2 is included for comparison.



Figure S5 (a) The 1D chain-like SBU in compound **2** with the coordination environment of Ca^{2+} . (b) The 3D structure of compound **2** containing DMA solvents viewed along the *a* axis. (c) The channel environment in zoom in mode. The structure figures are depicted according to the single crystal data reported in *Inorg. Chim. Acta*, 2018, **478**, 8.



Figure S6 The channel structure of compound 2 that depicted by Material Studio.



Figure S7 TG curve for the as-made compounds 1, 2 and 3.



Figure S8 TG curve for compound 2 before and after solvent exchange.



Figure S9 The excitation and emission spectra of H₄tatb ligand at solid state measured at room temperature ($\lambda_{ex} = 410 \text{ nm}$ and $\lambda_{em} = 490 \text{ nm}$).



Figure S10 The excitation and emission spectra of compound 1 at solid state measured at room temperature ($\lambda_{ex} = 370$ nm and $\lambda_{em} = 480$ nm).



Figure S11 The excitation and emission spectra of compound 2 at solid state measured at room temperature ($\lambda_{ex} = 370 \text{ nm}$ and $\lambda_{em} = 480 \text{ nm}$).



Figure S12 The excitation and emission spectra of compound 3 at solid state measured at room temperature ($\lambda_{ex} = 370 \text{ nm}$ and $\lambda_{em} = 480 \text{ nm}$).



Figure S13 The FL spectra of compound 1 dispersed in solvents and nitrobenzene.



Figure S14 The FL spectra of compound 2 dispersed in solvents and nitrobenzene.



Figure S15 The FL spectra of compound 3 dispersed in solvents and nitrobenzene.



Figure S16 PXRD for the as-made compounds before and after being immersed in 10⁻³ M TNP.



Figure S17 The IR for the as-made compounds before and after being immersed in 10⁻³ M TNP.

Compounds	$K_{\rm sv}$ / ${ m M}^{-1}$	Reference
Tb(L)(OH)	7.73 × 10 ⁻²	S1
Eu(BTB)H ₂ O	6.76 × 10 ⁻²	- S2
Tb(BTB)H ₂ O	3.25 × 10 ⁻²	
Cd(NDC) _{0.5} (PCA)	3.5×10^{4}	S3
[(CH ₃)2NH ₂] ₃ [Zn ₄ Na(BPTC) ₃]·4CH ₃ OH·2DMF	3.2×10^{4}	S4
$[Zn(NDC)(H_2O)]_n$	6×10^4	- \$5
$[Cd(NDC)(H_2O)]_n$	$2.385 imes 10^4$	
${[Tb(L)_{1.5}(H_2O)]\cdot 3H_2O}_n$	7.47×10^{4}	S6
$Zr_6O_4(OH)_6(L)_6$	$2.9 imes 10^4$	S7
$Eu_3(L)_3(HCOO)(\mu_3-OH)_2(H_2O)$	2.1×10^{4}	S8
$Zn_8(ad)_4(BPDC)_6O\cdot 2Me_2NH_2$	$4.6 imes 10^4$	S9
$Zr_6O_4(OH)_6(L)_6$	$5.8 imes 10^4$	S10
[Cd(NDC)L] ₂ ·H ₂ O	3.7×10^{4}	S11
$Zn_4(DMF)(Ur)_2(NDC)_4$	10.83×10^4	S12
$[{Zn(BINDI)_{0.5}(bpa)_{0.5}(H_2O)}.4H_2O]_n(MOF1)$	$4.9 imes 10^4$	- S13
$[\{Zn(BINDI)_{0.5}(bpe)\}\cdot 3H_2O]_n (MOF2)$	1.29×10^{4}	
Cu-CIP	$1.07 imes 10^4$	S14
$\{Mn(Tipp)(A)_2\}_n \cdot 2H_2O$	11.8×10^{4}	S15
{ $(Me_2NH_2)_4[Eu_4(DDAC)_3(HCO_2)(OH_2)_2] \cdot 8DMF \cdot 9H_2O$ } _n	$8.6 imes 10^4$	S16
$[Zn_3(TIAB)_2(IMDC)_2] \cdot (NO_3)_2 \cdot (DMF)_2 \cdot (H_2O)_2$	5.68×10^{4}	S17
[Zn(bipa)(suc)] _n	6.48×10^{4}	S18

Table S1 Selected MOF FL sensors towards TNP.

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Figure S18 The SEM photographs for the as-made compound 2 before (up) and after (down) immersing treatment in 10^{-3} M TNP.



Figure S19 The time-dependent absorption spectra for compounds 1 (a) and 3 (b) that dispersed in 0.5×10^{-4} M TNP.