Supplementary information

Dual host-guest interactions-mediated photoswitchable fluorescent supramolecular polymers for anti-counterfeiting and encryption

Hong Wang^{1,a,b}, Jia Tang^{1,a}, Haitao Deng^a, Yong Tian^a, Zhong Lin^a, Jiaxi Cui^c, and Jian Chen*,a,b

^a Key Laboratory of Theoretical Organic Chemistry and Functional Molecule of Ministry of Education, Hunan Provincial Key Laboratory of Controllable Preparation and Functional Application of Fine Polymers, Hunan Provincial Key Lab of Advanced Materials for New Energy Storage and Conversion, Hunan Province College Key Laboratory of QSAR/QSPR, School of Chemistry and Chemical Engineering, Hunan University of Science and Technology, Xiangtan, Hunan 411201, China.

^bSchool of Resource & Environment and Safety Engineering, Hunan University of Science and Technology, Xiangtan, Hunan 411201, China.

cInstitute of Fundamental and Frontier Sciences, University of Electronic Science and Technology

of China, Chengdu, Sichuan, 611731 China

* Corresponding authors: <u>cj0066@gmail.com (J. Chen)</u>

¹These authors contributed equally.

Experimental section

Materials

All starting chemicals and solvents were purchased from commercial sources and used without further treatment, unless indicated otherwise. 9-(diethylamino)-5-oxo-5H-benzo[a]phenoxazin-2-yl-methacrylate (NRME) are prepared by our previous work and 2-(3,3-dimethyl-6'-nitrospiro[indoline-2,2'-thiochromen]-1-yl)ethanol (STP-OH) are prepared by previous work ^[1-2]. The water used throughout this work is the double-distilled water which was further purified with a Milli-Q system.

Characterization

¹H NMR spectra was recorded on a 500 MHz NMR spectrometer (Bruker Avance). UV-Vis spectra were recorded on a Shimadzu UV-2501PC spectrophotometer at room temperature. Fluorescence spectra and fluorescence lifetime (τ) measurements were carried out with a time-correlated single photon counting (TCSPC) nanosecond fluorescence spectrometer (Edinburgh FLS920) at ambient temperature (298 K).

Spectral test

Taking **SPS-4** as an example: the as-prepared solution containing NC-DSA-CN (0.35 mg), P(MMA₉₁-*co*-BA₅₂-*co*-P5A₇-*co*-AD₇) (0.93 mg), CDSP (7.2 µg), and DMSO (3 mL) was transferred to a cuvette. Subsequently, all samples were exposed to visible light for 2 min. For photo-response experiment, the samples were exposed to UV for different times and then tested. The fluorescence and absorption spectra were obtained using a time-correlated single photon counting (TCSPC) nanosecond fluorescence spectrometer (Edinburgh FLS920), and a Shimadzu UV-2501PC spectrophotometer, respectively. The excitation wavelength is 420 nm ($\lambda_{ex} = 420$ nm). For other samples containing NC-DSA-CN or CDSP, detailed sample information can be found in Table 1.

Information encryption

SPS-5 without CDSP and SPS-4 with CDSP were selected as encryption unit to add into the 96-well plates based on the binary codes of the standard 8 bit ASCII characters. Subsequently, all samples were exposed to visible light for 5 min. Finally, the optical photographs of these samples were recorded before and after UV irradiation.

Association constants between P5A and NC-DSA-CN

¹H NMR was used to measure the association constants (K_a) between P5A and NC-DSA-CN. P5A and NC-DSA-CN were dissolved in DMSO. The concentration of NC-DSA-CN was kept at 100 mM. The concentrations of P5A were 0, 3.7, 7.5, 11.2, 15.0, 18.1, 22.5 and 30 mM. ¹H NMR spectra of P5A/NC-DSA-CN mixtures in DMSO were measured. Due to the host-guest interaction between P5A and NC-DSA-CN, the signals of protons will shift. A modified Benesi-Hildebrand equation (1) was used for the calculation of the association constants between P5A and NC-DSA-CN.

$$\frac{1}{\Delta\delta_{obs}} = \frac{1}{\Delta\delta} \cdot \frac{1}{K_a} \cdot \frac{1}{C_{P5A}} + \frac{1}{\Delta\delta}$$
(1)

where $\Delta \delta_{obs}$ is the observed shifts of the peaks; K_a is the association constant; $\Delta \delta$ is a constant correlated to the concentration of the NC-DSA-CN; and C_{P5A} is the concentration of P5A.



Figure S3. ¹³C-NMR spectrum (in CDCl₃) of P5A-Aly.

170

150

130

110

90 80 70 60 50 40 30 20 10 0 f1 (ppm)



Figure S4. TOF MS spectrum of P5A-Aly.



Figure S6. ¹H-NMR spectrum (in CDCl₃) of P5A-Aly.



Figure S7. (a) The absorption spectra of NC-DSA-CN in DMSO. (b) Fluorescence emission spectra of NC-DSA-CN (10 μ M) in water/DMSO mixtures, f_w = 0-90 vol%. (c) Plot of peak intensity of NC-DSA-CN versus f_w in the mixtures. (d) The photographs of NC-DSA-CN (10 μ M) in water/DMSO mixtures, f_w = 0-90 vol%. The scale bars is 0.5 cm.



Figure S8. Synthetic route of P(MMA-co-BA-co-P5A-co-AD) (P0).



Figure S9. GPC trace of P(MMA-co-BA-co-P5A-co-AD).



Figure S10. ¹H-NMR spectrum (in CDCl₃) of P(MMA-co-BA-co-P5A-co-AD).



Figure S11. Fluorescence emission spectra of NC-DSA-CN in DMSO, the mixture of NC-DSA-CN and P5A, the mixture of NC-DSA-CN and P0 in DMSO.



Figure S12. The normalized emission spectrum of NC-DSA-CN and the absorption spectrum of CDSP (before and after 365 nm UV irradiation) in DMSO. (b) Fluorescence lifetime changes of SPS-4 under alternating 365 nm UV and 525 nm Visible light.



Figure S13. Fluorescence spectra of SPS-4 under alternating 365 nm UV and 525 nm Visible light (red trace: $\lambda_{ex} = 365$ nm, black trace: $\lambda_{ex} = 365$ nm, blue trace, $\lambda_{ex} = 560$ nm). The blue trace represents the fluorescence spectrum (λ_{ex} =560 nm) of CDSP in DMSO (1.8 µg/mL).

Antenna effect

The antenna effect under certain concentrations of donor and acceptor equals the ratio of the emission intensity at 655 nm of the acceptor upon excitation of the donor.¹

Antenna effect =
$$\frac{I_{SPS-3+UV(\lambda_{ex} = 365 nm)}^{655 nm} - I_{SPS-3(\lambda_{ex} = 365 nm)}^{655 nm}}{I_{SPS-5-UV(\lambda_{ex} = 560 nm)}}$$
where $I_{SPS-3+UV(\lambda_{ex} = 365 nm)}^{655 nm}$ and $I_{SPS-3(\lambda_{ex} = 365 nm)}^{655 nm}$ are the fluorescence intensities of SPS-4

under alternating 365 nm UV and 525 nm Visible light, $I_{SPS-5-UV(\lambda_{ex} = 560 \text{ nm})}^{655nm}$ the fluorescence

intensities of CDSP in DMSO (SPS-5) under the irradiation of UV, respectively. The antenna effect value was calculated as 7.99.



Figure S14. (a) CIE coordinates of supramolecular polymer solution SPS-4: A (initial state), B (after 365 nm UV irradiation). (b) Fluorescence transition of SPS-4. All scale bars are 0.5 cm.



Figure S15. Optical photographs of packaging box.

Reference

1. J.-J. Li, Y. Chen, J. Yu, N. Cheng and Y. Liu, *Adv. Mater.*, 2017, **29**, 1701905.