Supporting Information

Highly Stable Lanthanide Cluster-based Luminescent Materials Constructed from β -diketone to 1,10phenanthroline Exhibiting Ultrahigh Photoluminescence and Efficient Pesticide Detection

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Fig. S1 The coordination modes of (a) $btfa^-$ and (b) mpa^- lignads in compounds Eu_{2b} and Eu_{2p} . Green Ln; yellow O, grey C, white, H.



Fig. S2 The snub diphenoid geometry of the octa-coordinated Eu^{3+} in Eu_{2b} .



Fig. S3 The vertical distance between the two planes formed by the benzene rings on the two adjacent $bfta^{-}$ ligands in Eu_{2b} .



Fig. S4 The tricapped trigonal prism geometry of nona-coordinated Eu^{3+} in Eu_{2p} .



Fig. S5 (a, b, c) In compound Eu_{2p} , the dihedral angles between the three planes (plane1, plane2, plane3) in phen1 and the plane of the benzene ring in mpa⁻, and the vertical distances from the centers of the plane1, plane2 and plane3 in phen1 to the plane of the benzene ring in mpa⁻.



Fig. S6 The vertical distance between the adjacent planes of two mpa⁻ ligands (a), and that of two phen2 ligands (b).



Fig. S7 The TGA of (a) compounds Eu_{2b} , Gd_{2b} and (b) compounds Eu_{2p} , Gd_{2p} under nitrogen atmosphere.



Fig. S8 The decay curves of (a) compounds Eu_{2b} and (b) Eu_{2p} .



Fig. S9 (a) The solid-state UV-vis spectra of Gd_{2b} and Gd_{2p} at room temperature; (b) the solid-state phosphorescence emission spectra of Gd_{2b} and Gd_{2p} at 77 K.



Fig. S10 (a) The temperature-dependent emission spectra of Eu_{2p} @PMMA; (b) the emission intensity (anchored at 614 nm) of Eu_{2p} @PMMA from 293 to 363 K.



Fig. S11 Time-dependent PXRD patterns of Eu_{2p}@PMMA from 25 to 100 °C.



Fig. S12 The emission sepctra of WEu_{2b} , WEu_{2p} and the mixed commercial phosphors.



Fig. S13 The emission spectra (a) and the emission intensity (b, anchored at 614 nm) of Eu_{2p} toward different pesticides in EtOH solution.



Fig. S14 The emission spectra of the selectivity experiment of Eu_{2p} toward DCN in the presence of other different pesticides.



Fig. S15 The PXRD patterns of Eu_{2p} befor and after soaking in EtOH solution of DCN.



Fig. S16 The PXRD patterns of Eu_{2b} and Gd_{2b} .



Fig. S17 The absorption spectra of different pesticides in EtOH solution and the

excitation spectrum of $\mathbf{Eu}_{2\mathbf{p}}$ at $\lambda_{em} = 614$ nm.



Fig. S18 IR spectra in $500 - 4000 \text{ cm}^{-1}$ for (a) Eu_{2b} , Gd_{2b} and (b) Eu_{2p} , Gd_{2p} .

compound	Eu _{2b}	Gd _{2b}
Formula	$C_{58}H_{46}Eu_2F_{12}O_{16}$	$C_{58}H_{46}Gd_2F_{12}O_{16}$
FW	1530.87	1541.45
T/K	120	120
Cry. System	triclinic	triclinic
Space group	<i>P</i> -1	<i>P</i> -1
<i>a</i> /Å	10.1101(15)	10.1164(17)
b /Å	12.1486(17)	12.138(2)
c /Å	13.7123(18)	13.680(2)
α / °	112.931(10)	112.871(11)
β/°	93.700(11)	93.806(13)
γ / °	97.746(12)	97.821(14)
$V/\text{\AA}^3$	1524.3(4)	1520.2(4)
Ζ	1	1
Dc /g cm ^{-3}	1.668	1.684
μ / mm ⁻¹	15.489	14.872
Data/parameters	5434/399	5415/399
2 <i>θ</i> / °	7.058 - 139.04	8.03 - 139.19
Obs. reflections	13374	20545
F (000)	756.0	758.0
GOOF	1.101	1.011
$R_1[I > 2\sigma(I)]^a$	0.0971	0.0409
wR ₂ (All data) ^b	0.1198	0.0501

Table S1. Crystallographic Data for Compounds Eu_{2b} , Gd_{2b} .

 $aR_1 = \sum ||FO| - |F_C|| / \sum |F_O|; bwR_2 = \{\sum [w(F_O^2 - F_C^2)^2] / \sum [w(F_O^2)^2] \}^{1/2}$

compound	Eu _{2p}	Gd _{2p}
Formula	$C_{84}H_{72}Cl_{2}Eu_{2}N_{8}O_{22}$	$C_{84}H_{72}Cl_2Gd_2N_8O_{22}$
FW	1920.31	1930.89
T / K	120	150
Cry. System	triclinic	triclinic
Space group	<i>P</i> -1	<i>P</i> -1
a /Å	11.970	11.9791(14)
b /Å	12.237	12.2597(14)
c /Å	13.604	13.6056(18)
α / °	75.15	74.888(10)
β/°	75.36	75.411(10)
γ / °	79.39	79.339(9)
V / Å ³	1848.3	1851.5(4)
Ζ	1	1
Dc /g cm ⁻³	1.725	1.732
μ / mm $^{-1}$	13.426	12.862
Data/parameters	6209/525	6723/525
2 <i>θ</i> / °	6.886 - 129.998	6.89 - 139.376
Obs. reflections	18605	27933
F (000)	988.0	1000.0
GOOF	1.027	1.010
$R_1[I > 2\sigma(I)]^a$	0.0374	0.0328
wR ₂ (All data) ^b	0.0917	0.0745

Table S2. Crystallographic Data for Compounds Eu_{2p} , Gd_{2p} .

 $aR_1 = \sum ||FO| - |F_C|| / \sum |F_O|; bwR_2 = \{\sum [w(F_O^2 - F_C^2)^2] / \sum [w(F_O^2)^2] \}^{1/2}$

Eu1-Eu1 ¹	4.0620(15)	O2-C12	1.188(11)	
Eu1-O2	2.354(8)	O4-C4	1.275(12)	
Eu1-O3	2.458(8)	O5-C15	1.231(11)	
Eu1-O4 ¹	2.472(9)	F2-C25	1.319(16)	
Eu1-O5	2.311(9)	F3-C28	1.331(14)	
Eu1-O6	2.378(9)	O6-C9	1.238(11)	
Eu1-O7	2.360(9)	O7-C10	1.200(11)	
Eu1-O8 ¹	2.489(8)	O8-C4	1.276(12)	
Eu1-O8	2.409(8)	F4-C25	1.320(12)	
Eu1-C4 ¹	2.875(6)	F5-C28	1.356(19)	
O1-C1	1.350(16)	F6-C25	1.341(17)	
O1-C26	1.419(17)	C1-C8	1.39(2)	
F1-C28	1.322(13)	C1-C21	1.396(18)	
O2-Eu1-Eu1 ¹	147.8(2)	O3-Eu1-O8 ¹	78.2(3)	_
O2-Eu1-O3	75.3(3)	O3-Eu1-C4 ¹	76.4(3)	
O2-Eu1-O4 ¹	87.2(3)	O4 ¹ -Eu1-Eu1 ¹	84.7(2)	
O2 -Eu1-O6	136.0(3)	O4 ¹ -Eu1-O8 ¹	52.4(3)	
O2-Eu1-O7	71.9(3)	O4 ¹ -Eu1-C4 ¹	26.2(3)	
O2 -Eu1-O8	134.4(3)	O5-Eu1-Eu1 ¹	115.9(2)	
O2 -Eu1-O8 ¹	134.8(3)	O5-Eu1-O2	72.6(3)	
O2-Eu1-C4 ¹	110.7(3)	O5-Eu1-O3	100.2(3)	
O3-Eu1-Eu1 ¹	72.6(2)	O5-Eu1-O4 ¹	158.8(3)	
O3-Eu1-O4 ¹	80.4(3)	O5-Eu1-O6	83.9(3)	

Table S3. Selected bond distances (Å) and band angles (°) of Eu_{2b} .

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Symmetry code: 1-X,1-Y,-Z

Gd1-Gd1 ¹	4.0474(10)	O4-C13	1.275(7)
Gd1-O1 ¹	2.446(4)	O5-C2	1.377(6)
Gd1-O2	2.342(3)	O5-C27	1.431(6)
Gd1-O3	2.370(3)	F1-C20	1.328(7)
Gd1-O4	2.303(4)	O6-C12	1.246(6)
Gd1-O6	2.346(3)	F2-C20	1.330(7)
Gd1-07	2.395(3)	F3-C29	1.321(8)
Gd1-O8 ¹	2.487(3)	F4-C20	1.323(7)
Gd1-O8	2.392(3)	F5-C29	1.327(7)
Gd1-C1 ¹	2.840(5)	O8-C1	1.283(6)
O1-C1	1.250(6)	F6-C29	1.346(8)
O2-C28	1.255(6)	C1-C8	1.509(7)
O3-C26	1.288(6)	C2-C7	1.393(7)
O1 ¹ -Gd1-Gd1 ¹	84.88(8)	O2-Gd1-C1 ¹	110.12(13)
O1 ¹ -Gd1-O8 ¹	52.75(11)	O3-Gd1-Gd1 ¹	75.73(8)
O1 ¹ -Gd1-C1 ¹	26.02(13)	O3-Gd1-O1 ¹	107.54(13)
O2-Gd1-Gd1 ¹	147.82(8)	O3-Gd1-O7	146.61(12)
O2-Gd1-O1 ¹	86.85(12)	O3-Gd1-O8	74.82(11)
O2-Gd1-O3	136.26(12)	O3-Gd1-O8 ¹	81.42(11)
O2-Gd1-O6	71.68(12)	O3-Gd1-C11	96.65(14)
O2-Gd1-O7	75.49(12)	O4-Gd1-Gd1 ¹	115.76(9)
O2-Gd1-O8 ¹	134.75(11)	O4-Gd1-O1 ¹	158.75(12)

Table S4. Selected bond distances (Å) and band angles (°) of Gd_{2b} .

Symmetry code: 1-X,1-Y,-Z

O2-Gd1-O8

O4-Gd1-O2

72.97(12)

134.44(11)

 Eu1-Eu1 ¹	3.9414(5)	01-C1	1.269(5)
Eu1-O1 ¹	2.380(3)	O2-C2	1.281(5)
Eu1-O2	2.344(3)	O3-C1	1.249(5)
Eu1-O2 ¹	2.641(3)	O4-C2	1.254(5)
Eu1-O3	2.383(3)	O5-C9	1.424(6)
Eu1-O4 ¹	2.430(3)	O5-C13	1.375(5)
Eu1-N1	2.579(3)	N1-C8	1.363(5)
Eu1-N2	2.635(3)	N1-C34	1.336(5)
Eu1-N3	2.634(4)	N2-C3	1.360(5)
Eu1-N4	2.636(4)	N2-C33	1.320(5)
Eu1-C2 ¹	2.912(4)	N3-C23	1.361(6)
C11-O7	1.419(4)	N3-C26	1.332(6)
C11-O8	1.411(4)	O6-C30	1.371(6)
O1 ¹ -Eu1-Eu1 ¹	68.56(7)	O2 ¹ -Eu1-Eu1 ¹	35.19(6)
O1 ¹ -Eu1-O2 ¹	71.67(9)	O2-Eu1-O11	74.90(10)
O1 ¹ -Eu1-O3	137.88(9)	O2-Eu1-O2 ¹	75.67(10)
O1 ¹ -Eu1-O4 ¹	75.83(10)	O2-Eu1-O3	75.62(10)
O1 ¹ -Eu1-N1	81.16(10)	O2-Eu1-O4 ¹	124.88(10)
O1 ¹ -Eu1-N2	132.55(10)	O2-Eu1-N1	142.97(10)
O1 ¹ -Eu1-N3	137.70(10)	O2-Eu1-N2	150.75(10)
O1 ¹ -Eu1-N4	75.80(10)	O2-Eu1-N3	99.64(10)
O1 ¹ -Eu1-C2 ¹	74.05(10)	O2-Eu1-N4	76.11(10)
O2-Eu1-Eu1	40.48(7)	O2 ¹ -Eu1-C2 ¹	26.08(10)

Table S5. Selected bond distances (Å) and band angles (°) of Eu_{2p} .

Symmetry code:¹1-X,-Y,1-Z

Gd1-Gd1 ¹	3.9340(7)	C11-O9	1.411(4)
Gd1-O1	2.366(2)	C11-O10	1.417(5)
Gd1-O2 ¹	2.422(3)	O1-C8	1.255(4)
Gd1-O3 ¹	2.368(2)	O2-C2	1.239(4)
Gd1-O5 ¹	2.639(2)	O3-C8	1.261(4)
Gd1-O5	2.338(2)	O4-C3	1.375(5)
Gd1-N1	2.621(3)	O4-C26	1.430(5)
Gd1-N2	2.623(3)	O5-C2	1.282(4)
Gd1-N3	2.565(3)	N1-C4	1.367(5)
Gd1-C2 ¹	2.903(4)	N1-C25	1.331(5)
Gd1-N4	2.614(3)	N2-C7	1.327(5)
Cl1 -07	1.429(3)	N2-C15	1.372(5)
C11-O8	1.410(4)	N3-C1	1.371(5)
O1-Gd1-Gd1 ¹	69.21(6)	O2 ¹ -Gd1-N1	65.48(9)
O1-Gd1-O2 ¹	97.46(9)	O2 ¹ -Gd1-N2	137.70(9)
O1-Gd1-O3 ¹	137.86(8)	O2 ¹ -Gd1-N3	74.22(9)
O1-Gd1-O5 ¹	72.14(8)	O2 ¹ -Gd1-C2 ¹	24.88(9)
O1-Gd1-N1	75.87(9)	O2 ¹ -Gd1-N4	132.95(9)
O1-Gd1-N2	124.57(9)	O3 ¹ -Gd1-Gd1 ¹	68.80(6)
O1-Gd1-N3	138.35(9)	O3 ¹ -Gd1-O2 ¹	75.71(9)
O1-Gd1-C21	82.88(9)	O3 ¹ -Gd1-O5 ¹	71.76(8)
O1-Gd1-N4	76.90(9)	O3 ¹ -Gd1-N1	132.36(9)
O2 ¹ -Gd1-Gd1 ¹	85.19(6)	O3 ¹ -Gd1-N2	75.58(9)
O2 ¹ -Gd1-O5 ¹	50.90(8)	O31-Gd1-N3	80.67(9)

Table S6. Selected bond distances (Å) and band angles (°) of Gd_{2p} .

Symmetry code: ¹1-X,-Y,1-Z

Com. ^b	Eu _{2b}	Gd _{2b}
Refcode ^a	Eu1	Gd1
OP-8	24.391	24.579
HPY-8	16.733	16.875
HBPY-8	13.713	13.703
CU-8	13.432	13.431
SAPR-8	7.619	7.572
TDD-8	6.271	6.356
JGBF-8	11.602	11.527
JETBPY-8	20.865	20.846
JBTPR-8	5.395	5.375
BTPR-8	6.432	6.44
JSD-8	4.332	4.326
TT-8	14.056	14.064
ETBPY-8	21.262	21.396

Table S7. The CShM (Continuous Shape Measurements) values of Ln^{3+} ions with octa-coordinate mode in compounds Eu_{2b} , Gd_{2b} .

^{*a*}OP-8, Octagon; HPY-8, Heptagonal pyramid; HBPY-8, Heptagonal bipyramid; CU-8, Cube; SAPR-8, Square antiprism; TDD-8, Triangular dodecahedron; JGBF-8, Johnson gyrobifastigium; JETBPY-8, Johnson elongated triangular bipyramid; JBTPR-8, Biaugmented trigonal prism; BTPR-8, Biaugmented trigonal prism; JSD-8, Snub diphenoid; TT-8, Triakis tetrahedron; ETBPY-8, Elongated trigonal bipyramid. ^{*b*}Lanthanide ions in each asymmetric unit of compounds **Eu_{2b}**, **Gd_{2b}**.

Com. ^b	Eu _{2p}	Gd _{2p}
Refcode ^a	Eu1	Gd1
EP-9	25.227	25.192
OPY-9	20.08	20.055
HBPY-9	16.932	16.953
JTC-9	14.295	14.233
JCCU-9	12.166	12.144
CCU-9	12.202	12.188
JCSAPR-9	4.596	4.603
CSAPR-9	4.511	4.525
JTCTPR-9	4.259	4.213
TCTPR-9	5.092	5.137
JTDIC-9	14.36	14.425
НН-9	10.87	10.876
MFF-9	4.394	4.416

Table S8. The CShM (Continuous Shape Measurements) values of Ln^{3+} ions withnona-coordinate mode in compounds Eu_{2p} , Gd_{2p} .

^{*a*}EP-9, Enneagon; OPY-9, Octagonal pyramid; HBPY-9, Heptagonal bipyramid; JTC-9, Johnson triangular cupola; JCCU-9, Capped cube; CCU-9, Spherical-relaxed capped cube; JCSAPR-9, Capped square antiprism; CSAPR-9, Spherical capped square antiprism; JTCTPR-9, Tricapped trigonal prism; TCTPR-9, Spherical tricapped trigonal prism; JTDIC-9, Tridiminished icosahedron; HH-9, Hula-hoop; MFF-9, Muffin. ^{*b*}Lanthanide ions in each asymmetric unit of compounds Eu_{2p}, Gd_{2p}.

Lanthanide compounds	QY	Ref.
Eu(hfa) ₃ (TPPO) ₂	65%	1
$[Eu(tftpa)_{1.5}(bpy)(H_2O)]_n$	74.50%	2
[Eu ₂ (m-BDC) ₃ (Phen) ₂] DMF	75%	3
$[Eu_2(2,3'-oba)_3(phen)_2]_n$	75.57%	4
[Eu(hfac) ₃ (DPEPO)]	80%	5
{ $[Eu(H_2O)(TPA)(ox)] \cdot 0.5H_2O$ }	89%	6
$[Eu(phen)_2(NO_3)_3]$	90%	7
${[EuKL_4(H_2O)_2] \cdot H_2O}_n$	92%	8
[Eu(pfbz) ₂ (phen)Cl]	97.70%	9
Eu _{2p}	96.79%	This work

Table S9. Some lanthanide compounds with high quantum yields.

Metal compounds	$K_{\rm sv}$ (M ⁻¹)	LOD	Ref.
Eu ₂ (dtztp)-MOF	6.25×10^{4}	5.28 ppm	10
Tb ₃ (HDDB)(DDB)(H ₂ O) ₆	6.42×10^{4}	$1.4 \times 10^{-7} \mathrm{M}$	11
$[Cd_{18}Sm_{6}L_{9}(OAc)_{36}]$	2.45×10^{4}	$6.7 \times 10^{-7} \mathrm{M}$	12
[Ag(CIP ⁻)]	5.2×10^{4}	$1.7 \times 10^{-7} \mathrm{M}$	13
[Ag(3-dpyb)(H ₃ odpa)]·H ₂ O	2.028×10^5	$1.154 \times 10^{-6} \mathrm{M}$	14
[Cd(tptc) _{0.5} (bpz)-CP]	4.71×10^4	112 ppb	15
[Cd3(CBCD)2(DMA)4(H2O)2]·10DMA	4.47×10^{4}	145 ppb	16
$[Mg_2(APDA)_2(H_2O)_3] \cdot 5DMA \cdot 5H_2O$	$7.5 imes 10^4$	150 ppb	17
Eum	3.21×10^{4}	1.21×10^{-7} M	This
2~2p			work

 Table S10. Some metal compounds for DCN detection.

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