

Supporting Information

Low-temperature processed additive-incorporated CsPbIBr₂-based inverted perovskite solar cell

Tuhin Ghosh and Debabrata Pradhan*

Materials Science Centre, Indian Institute of Technology, Kharagpur 721 302, India

*E-mail: deb@matsc.iitkgp.ac.in

Characterization. The crystallographic phase of the as-synthesized CsPbIBr₂ was analyzed by powder X-ray diffraction (PXRD) using Bruker D2 PHASER X-ray diffractometer (Cu K α radiation, $\lambda = 1.54 \text{ \AA}$) in the 2θ range of 10–50°. The 2D XRD was performed using BRUKER D8 DISCOVER. The surface morphology of the as-synthesized materials was examined using field emission scanning electron microscope (FESEM) (MERLIN, Zeiss) operated at 5 kV. The thickness of different layers of solar device was examined by FESEM (Gemini 500, Zeiss) operated at 5 kV. The formation of highly crystalline structure of both with and without MACl incorporated CsPbIBr₂ were examined by transmission electron microscopy (TEM, TECNAI G2, FEI). The chemical composition and electronic structure of the as-deposited perovskite films was studied by X-ray photoelectron spectroscopy (XPS) with a PHI 5000 VersaProbe-II (ULVAC, PHI, Inc.) spectrometer, equipped with a micro-focused (100 μm , 25 W, 15 kV) monochromatic Al K α X-ray source with an energy of 1486.6 eV. The topographical features of both without and with MACl-incorporated CsPbIBr₂ were examined by atomic force microscopy (AFM) (Agilent Technology, 5500). The optical absorbance and emission of the as-prepared samples was studied using UV-vis spectrophotometer (PerkinElmer Lambda 750) and photoluminescence spectrophotometer (PerkinElmer LS55), respectively. The time-resolved photoluminescence (TRPL) measurement was performed by time-correlated single

photon counting (TCSPC) spectrometer of IBH (U.K.). To characterize the PV performances and current–voltage (J – V) response of the fabricated devices, simulated AM 1.5 sunlight with a power intensity of 100 mW/cm² (Xenon Short Arc operated at 150 W using a Solar simulator, PHOTO EMISSION TECH., USA, SS50AAA) was used. The PV response, i.e., J – V curve of the solar device with an active device area and aperture area of of 0.1 cm², was recorded by a source meter (4200 SCS, Keithley) with a scan rate of 0.01 V/s. The external quantum efficiency (EQE) of the fabricated devices was measured using the Enlitech (QE-R) system. The impedance spectroscopic analysis was carried with a biopotentiostat (SP 150, BioLogic). The transient photocurrent (TPC) and transient photovoltage (TPV) were carried out in Metrohm AUTOLAB electrochemical analyzer. The glass rod used to form an active perovskite films has following dimension: length = 30 cm, radius = 0.25 cm, mass = 18.32 g, and weight = 0.18 N. It is to be noted that no additional force and/or power was used to prepare the perovskite thin film.

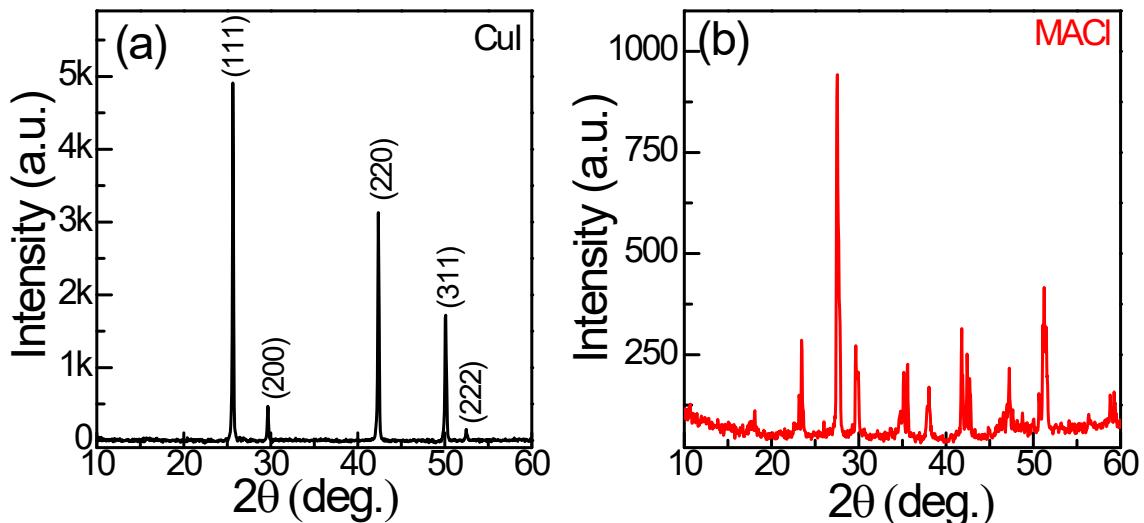


Fig. S1. XRD patterns of the as-synthesized (a) CuI and (b) MACl

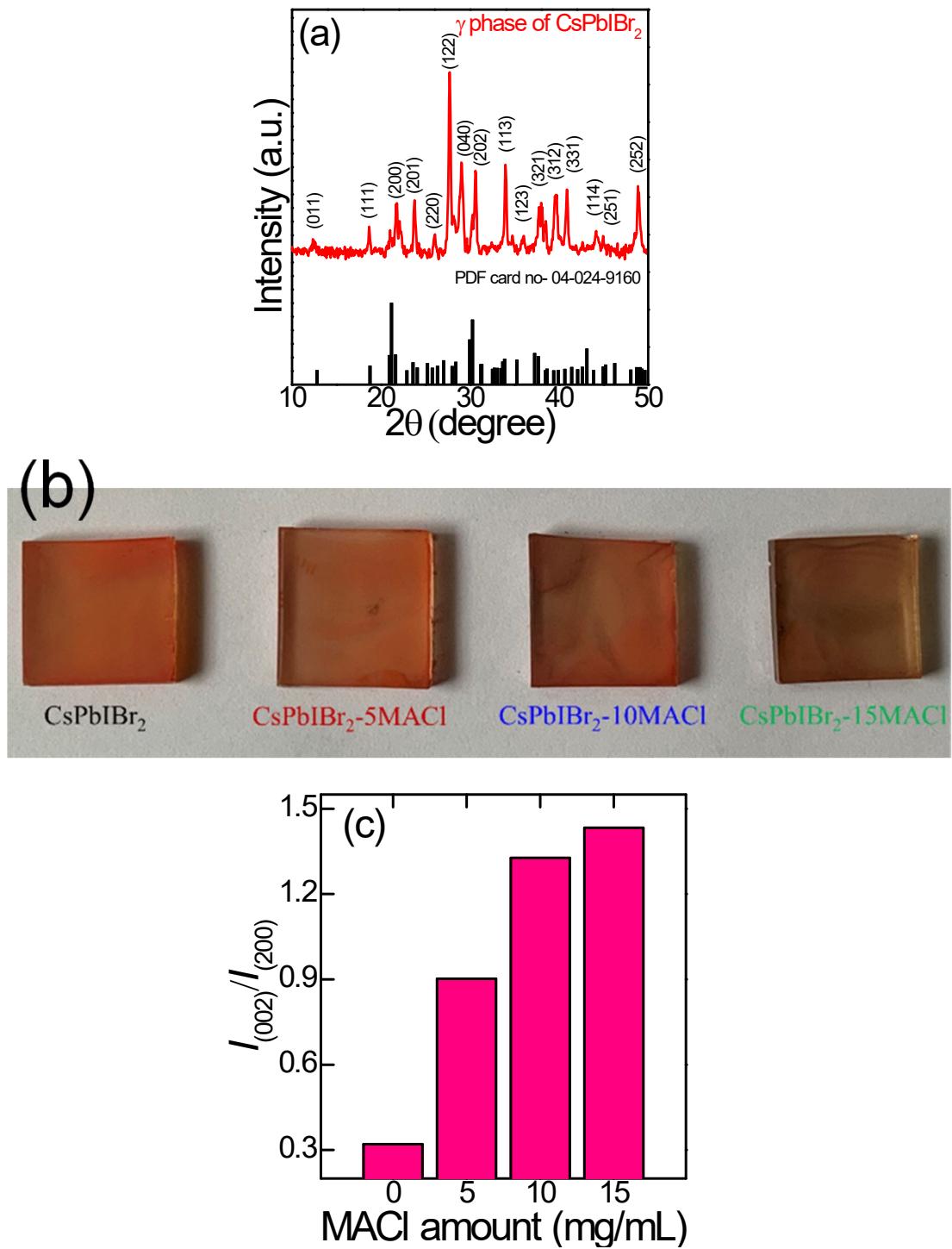


Fig. S2. (a) XRD pattern of the as-synthesized solid-state mixture of CsPbIBr_2 perovskite. The orthorhombic (γ) phase of the as-synthesised CsPbIBr_2 perovskite powder matches well with the JCPDS card No. 04-024-9160, (b) photographs and (c) $(002)/(200)$ direction intensity of CsPbIBr_2 and $\text{M}\text{A}\text{C}\text{l}$ -incorporated CsPbIBr_2 thin films.

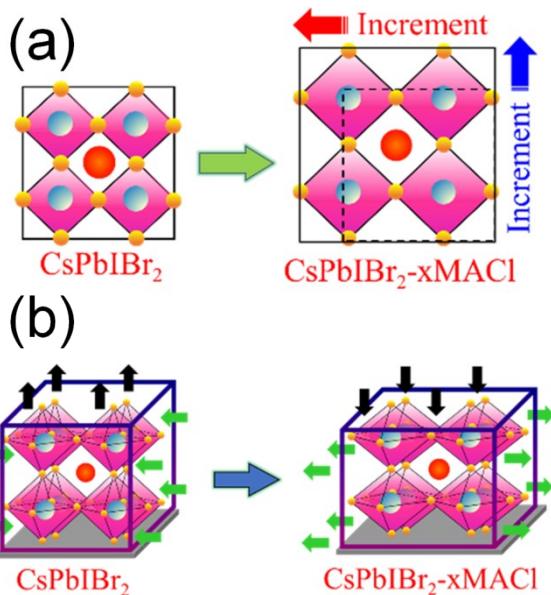


Fig. S3. (a) Schematic representation of the lattice expansion and (b) conversion of lattice strain from compressive to tensile due to MACl incorporation within CsPbIBr_2

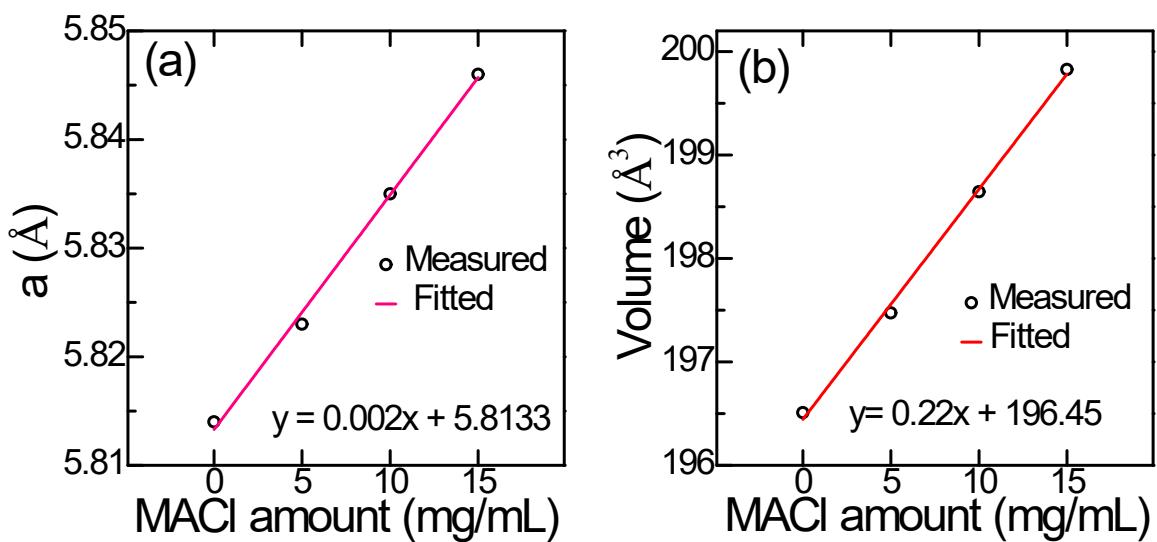


Fig. S4. Fitting plot of (a) lattice parameter and (b) crystal volume of without and with MACl-incorporated CsPbIBr_2 films

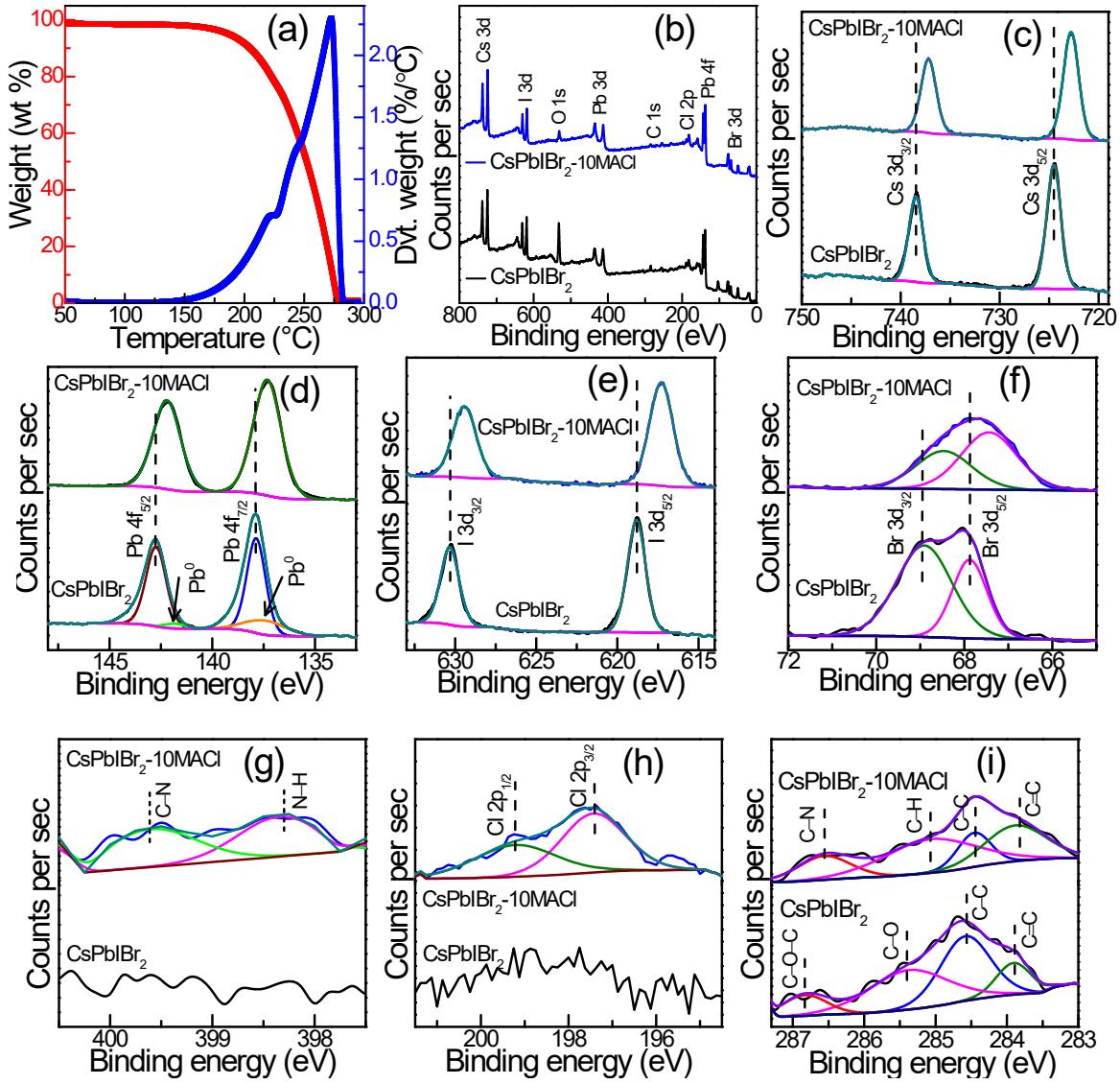


Fig. S5. (a) TGA plot of MACl, and (b) XPS survey, (c) Cs 3d, (d) Pb 4f, (e) I 3d, (f) Br 3d, (g) N 1s, (h) Cl 2p and (i) C 1s spectra of CsPbIBr₂-10MACl and CsPbIBr₂

Table S1. The percentage of C–N and C=N bonding using N 1s region spectra (Fig. S6g)

Active layer	C–N (%)	N–H (%)
CsPbIBr ₂	Absent	Absent
CsPbIBr ₂ -10MACl	48.46	51.54

Table S2. The carbon atomic percentages with different bonding using XPS analysis (Fig. S5i)

Active layer	C=C	C–C	C–O–C	C–O	C–H	C–N
CsPbIBr ₂	13.16	38.75	10.85	37.25	Absent	Absent
CsPbIBr ₂ -10MACl	42.96	6.37	Absent	Absent	33.39	17.28

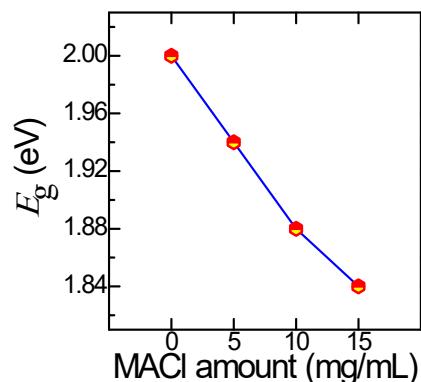


Fig. S6. Band gap variation of CsPbIBr₂ and CsPbIBr₂-xMACl perovskite layer

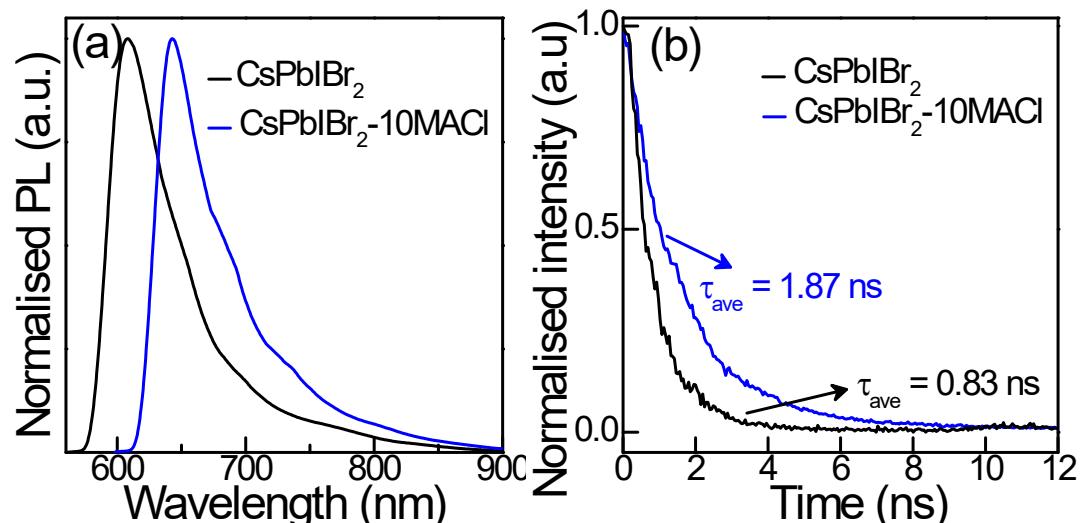


Fig. S7. (a) PL and (b) TRPL variation of CsPbIBr₂ and CsPbIBr₂-10MACl perovskite layer

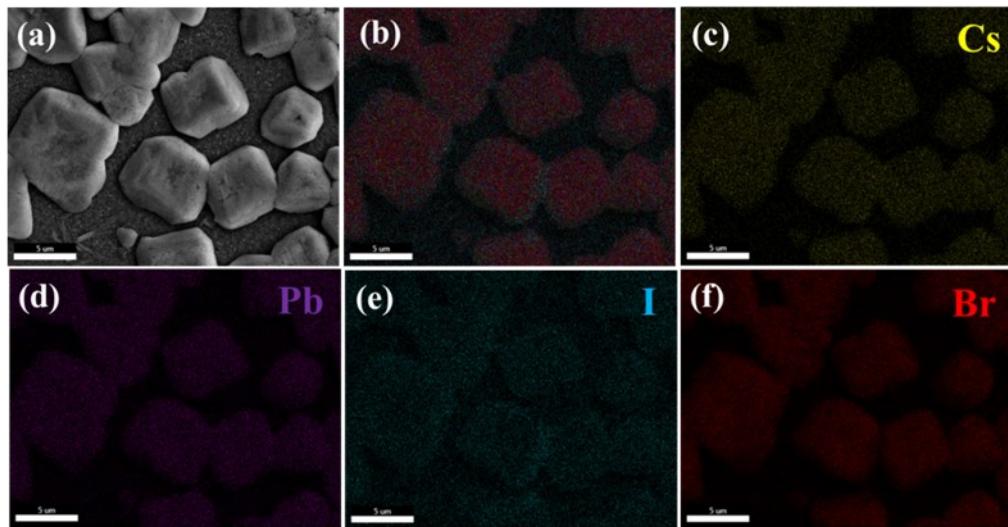


Fig. S8. (a) SEM image of CsPbIBr_2 film and corresponding EDX elemental mapping, (b) overlapped EDX signal of all elements shown in (c) Cs, (d) Pb, (e) I, and (f) Br in the CsPbIBr_2 film.

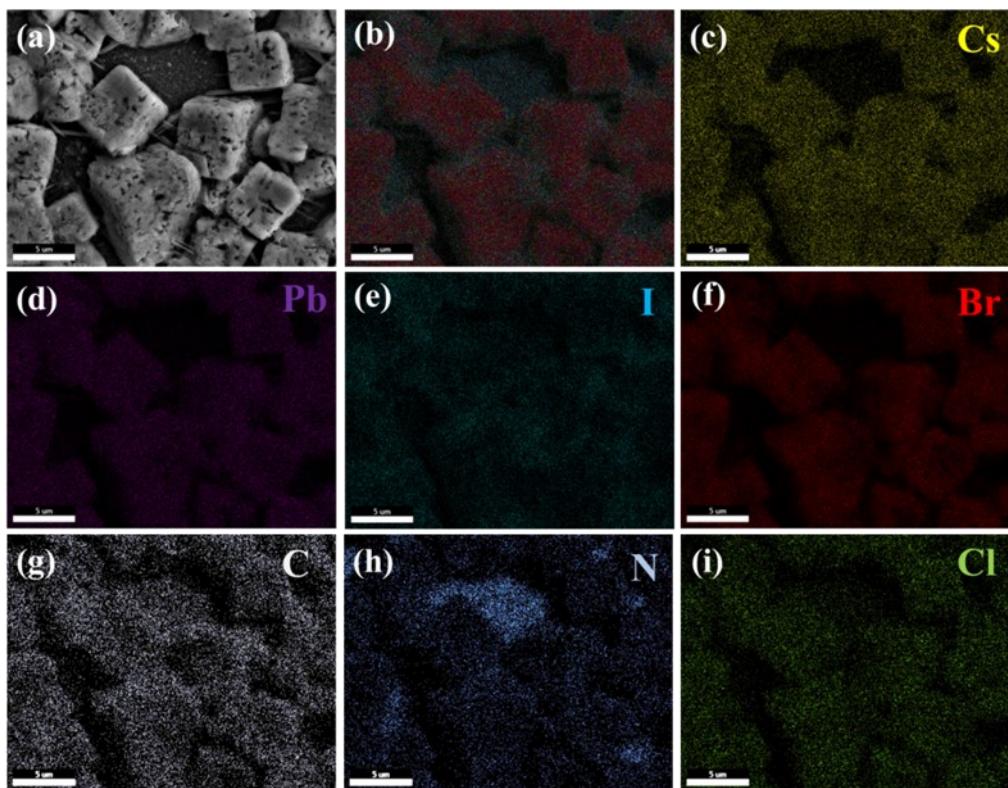


Fig. S9. (a) SEM image of $\text{CsPbIBr}_2\text{-}10\text{MACl}$ film and corresponding EDX elemental mapping, (b) overlapped EDX signals of all elements shown in (c) Cs, (d) Pb, (e) I, (f) Br, (g) C, (h) N, and (i) Cl in the $\text{CsPbIBr}_2\text{-}10\text{MACl}$ film.

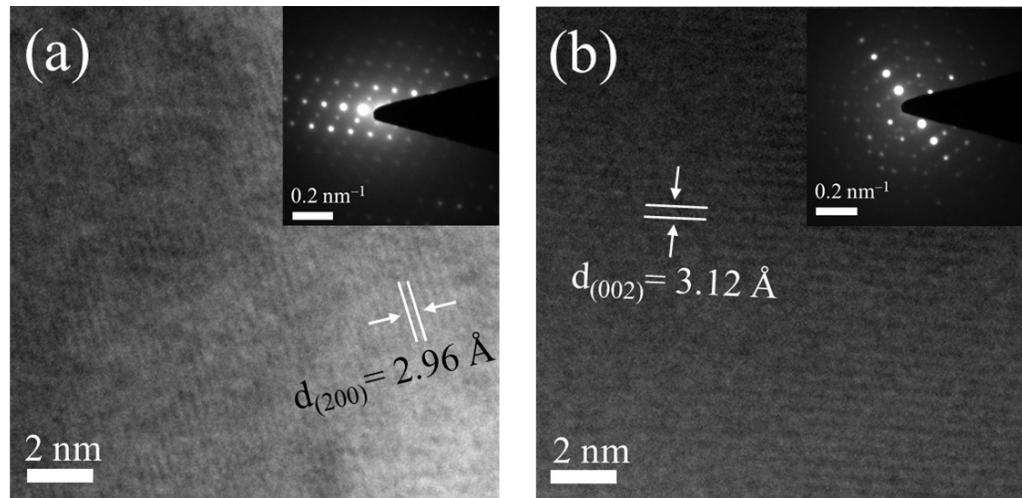


Fig. S10. TEM analysis of (a) CsPbIBr_2 and (b) $\text{CsPbIBr}_2\text{-}10\text{MACl}$ with measured inter planar spacing.

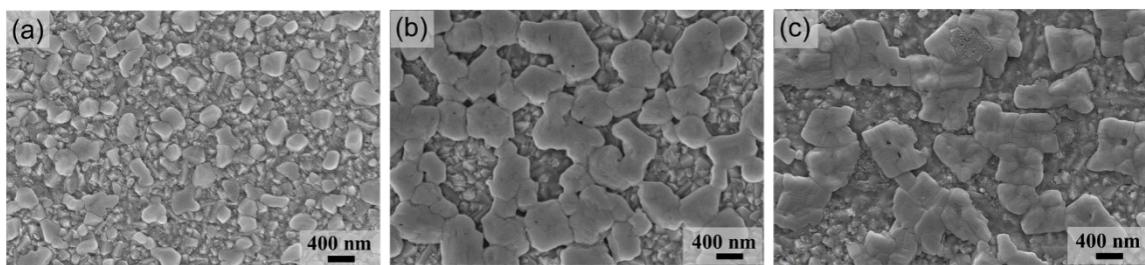


Fig. S11. FESEM images of the perovskite thin films obtained by varying the annealing time (a) 15 min, (b) 30 min, and (c) 45 min.

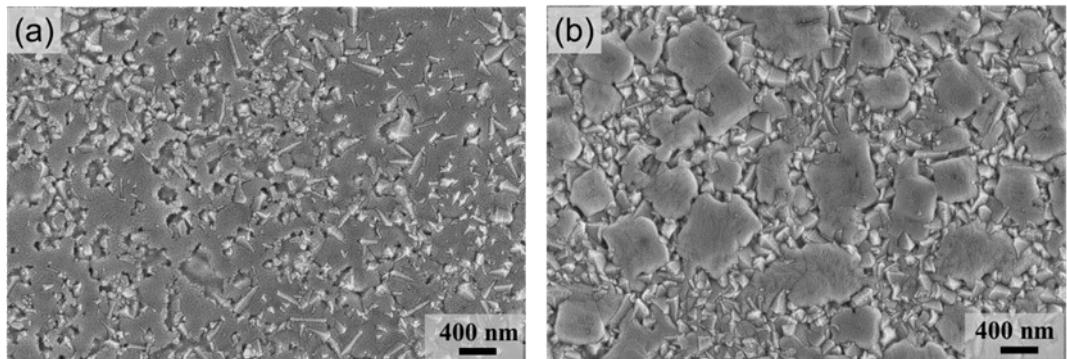


Figure S12. FESEM images of (a) CsPbIBr_2 and (b) $\text{CsPbIBr}_2\text{-}10\text{MACl}$ perovskite thin films obtained by spin coating method.

Table S3. Photophysical properties of PSC devices with active layer, i.e., CsPbIBr₂ and CsPbIBr₂-10MACl.

<i>CsPbIBr₂-based Cells</i>	<i>V_{OC}</i> (V)	<i>J_{SC}</i> (mA/cm ²)	<i>FF</i>	<i>PCE</i> (%)	<i>CsPbIBr₂-10MACl-based Cells</i>	<i>V_{OC}</i> (V)	<i>J_{SC}</i> (mA/cm ²)	<i>FF</i>	<i>PCE</i> (%)
1	0.9	11.40	0.511	5.24	1	1.16	12.06	0.586	8.20
2	0.89	10.96	0.499	4.87	2	1.14	11.70	0.580	7.73
3	0.79	10.53	0.508	4.22	3	1.15	11.98	0.584	8.04
4	0.85	11.01	0.501	4.69	4	1.14	12.08	0.579	7.97
5	0.91	11.35	0.498	5.14	5	1.13	12.06	0.585	7.97
6	0.84	11.28	0.502	4.75	6	1.15	12.00	0.585	8.07
7	0.89	11.3	0.489	4.92	7	1.13	11.76	0.581	7.72
8	0.87	11.35	0.495	4.89	8	1.1	12.48	0.582	7.99
9	0.84	10.88	0.496	4.53	9	1.14	11.68	0.563	7.49
10	0.78	11.18	0.502	4.38	10	1.17	11.93	0.565	7.89
11	0.86	11.38	0.511	5.00	11	1.18	11.71	0.584	8.07
12	0.89	11.10	0.49	4.84	12	1.11	11.71	0.584	7.59
13	0.84	11.03	0.505	4.68	13	1.15	11.98	0.570	7.86
14	0.86	10.97	0.46	4.34	14	1.09	12.01	0.585	7.66
15	0.86	11.29	0.504	4.90	15	1.11	11.87	0.580	7.65
16	0.88	11.25	0.516	5.11	16	1.11	11.70	0.576	7.48
17	0.83	10.98	0.483	4.40	17	1.09	11.82	0.579	7.47
18	0.85	11.39	0.493	4.77	18	1.15	10.96	0.585	7.37
19	0.89	11.38	0.497	5.03	19	1.13	11.90	0.581	7.82
20	0.82	11.29	0.461	4.28	20	1.14	11.98	0.585	7.99

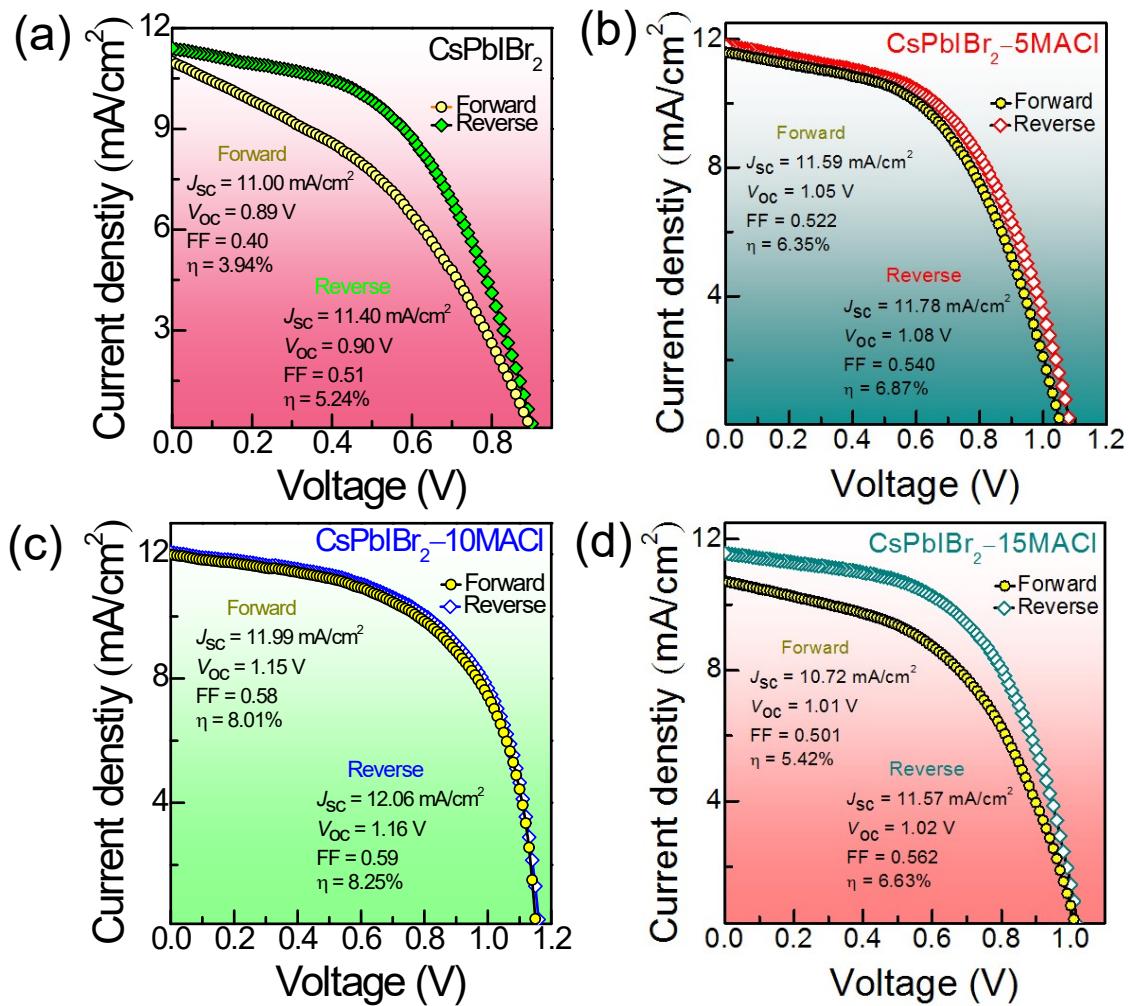


Fig. S13. $J-V$ curves of the (a) CsPbIBr_2 , (b) $\text{CsPbIBr}_2 - 5\text{MACl}$, (c) $\text{CsPbIBr}_2 - 10\text{MACl}$, and (d) $\text{CsPbIBr}_2 - 15\text{MACl}$ -based PSC with both scan directions

Table S4. PV performances of the present fabricated PSC with the other CsPbIBr₂ based PSCs as reported in literature.

<i>Sl No.</i>	<i>Device architecture</i>	J_{SC} (mA/cm^2)	V_{OC} (V)	PCE (%)	Temp. (°C)	<i>Ref.</i>
1.	FTO/CuI/CsPbIBr ₂ /Al	11.40	0.90	5.24	110	This work
2.	FTO/CuI/CsPbIBr ₂ -10MACl/Al	12.06	1.16	8.25	110	This work
3.	FTO/TiO ₂ /CsPbIBr ₂ -10MACl/PEDOT:PSS/Au	12.89	1.26	10.18	110	This Work
4.	FTO/c-TiO ₂ /CsPbIBr ₂ /Carbon	10.66	1.24	9.16	280	1
5.	FTO/NiO _x /CsPbIBr ₂ /MoO _x /Au	10.56	0.85	5.52	160	2
6.	FTO/c-TiO ₂ /CsPbIBr ₂ /Spiro-OMeTAD /Au	9.69	1.22	8.02	320	3
7.	FTO/c-TiO ₂ /m-TiO ₂ /CsPbIBr ₂ /Spiro-OMeTAD /Au	7.8	1.12	6.30	300	4
8.	FTO/c-TiO ₂ /CsPbIBr ₂ /Au	8.7	0.921	4.70	250	5
9.	FTO/SnO ₂ /C60/CsPbIBr ₂ /Spiro-OMeTAD/Au	8.32	1.18	7.34	150	6
10.	ITO/ZnO/CsPbIBr ₂ /Spiro-OMeTAD/Ag	11.52	1.27	10.16	160	7
11.	ITO/SnO ₂ /CsPbIBr ₂ /Carbon	8.50	1.23	7.00	180	8
12.	FTO/In ₂ S ₃ /CsPbIBr ₂ /Spiro-OMeTAD /Ag	7.76	1.09	5.59	160	9
13.	FTO/c-TiO ₂ /CsPbIBr ₂ (GuaSCN 3%)/Spiro-OMeTAD/Au	10.90	1.23	12.05	280	10
14.	FTO/c-TiO ₂ /CsPbIBr ₂ (BAI0.1%)/Spiro-OMeTAD/Au	10.78	1.25	11.63	160	11
15.	FTO/TiO ₂ /PEG-CsPbIBr ₂ /Spiro-OMeTAD/Ag	11.10	1.21	12.25	200	12

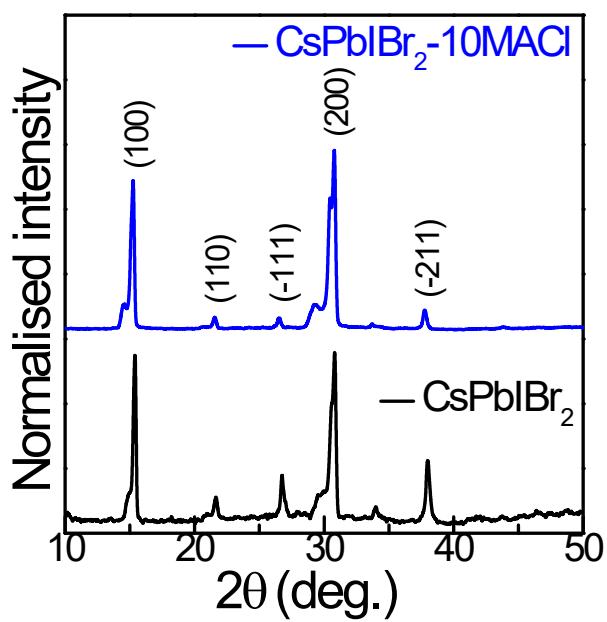


Fig. S14. XRD patterns of CsPbIBr_2 and $\text{CsPbIBr}_2\text{-}10\text{M}\text{ACl}$ film after 3 months of preparation.

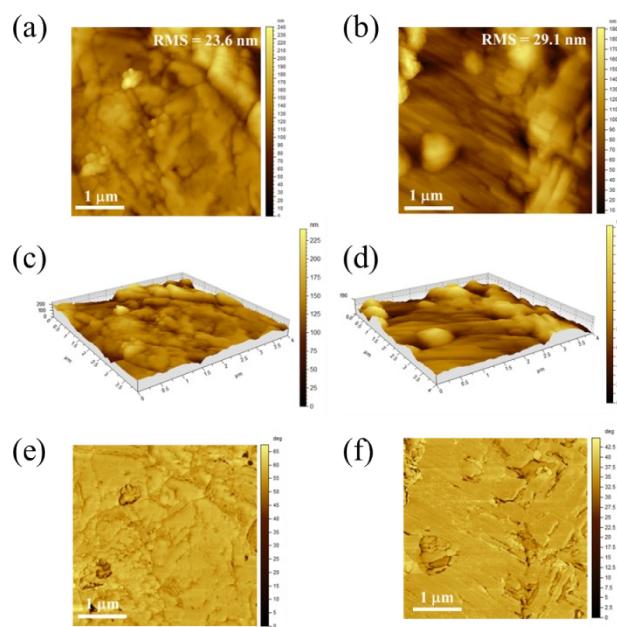


Fig. S15. AFM analysis of (a, c and e) $\text{CsPbIBr}_2\text{-}10\text{M}\text{ACl}$ and (b, d and f) CsPbIBr_2 .

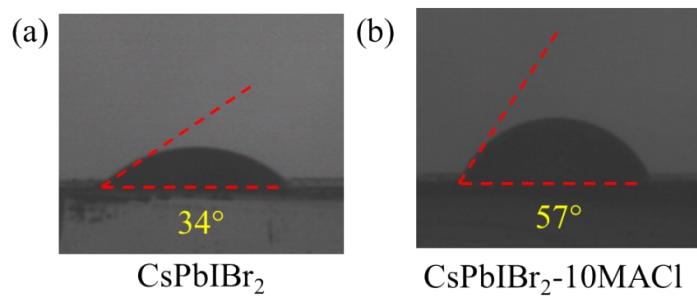


Fig. S16. Contact angle measurements of (a) CsPbIBr_2 and $\text{CsPbIBr}_2\text{-}10\text{MACl}$.

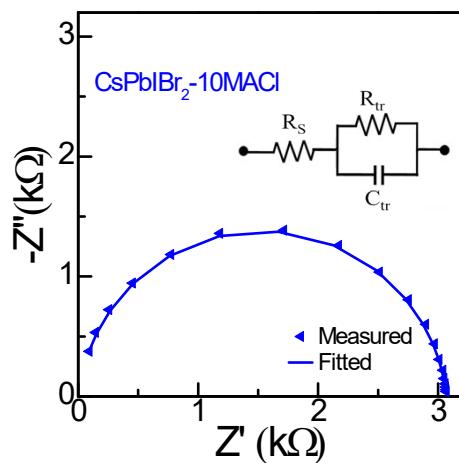


Fig. S17. Fitted Nyquist plot of $\text{CsPbIBr}_2\text{-}x\text{MACl}$ and inset shows the fitted circuit diagram.

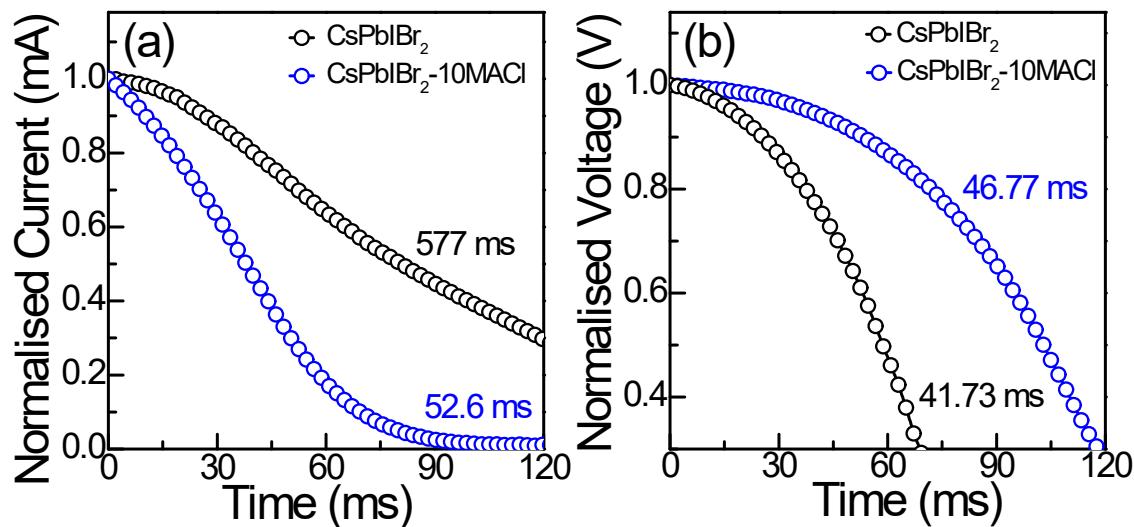


Fig. S18. (a) TPC and (b) TPV measurements of $\text{CsPbIBr}_2\text{-}10\text{MACl}$ and CsPbIBr_2 .

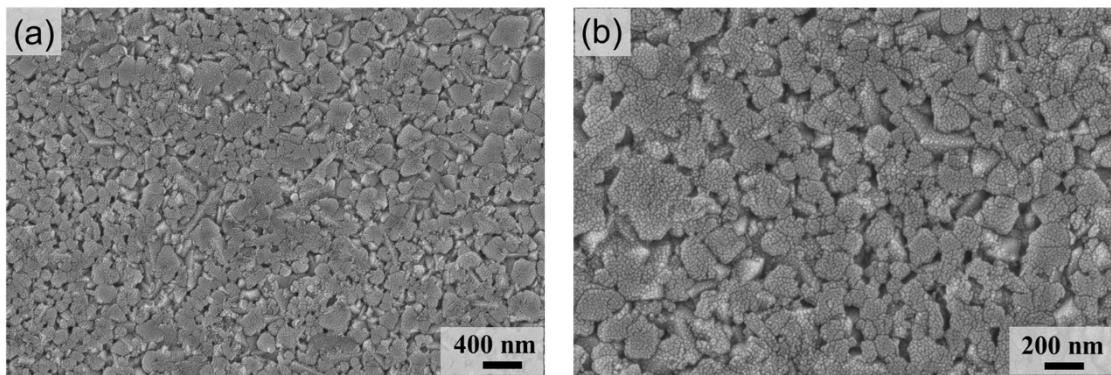


Fig. S19. (a,b) FESEM images of CsPbIBr_2 thin film obtained using salt precursors.

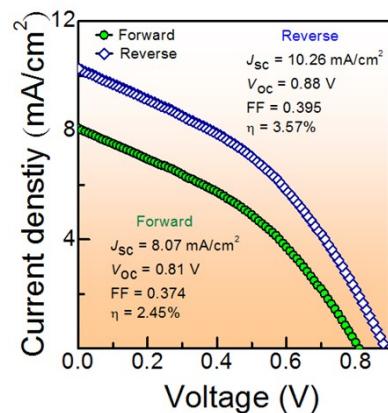


Fig. S20. J - V curves of the device fabricated with CsPbIBr_2 thin film prepared using salt precursors.

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