

Electronic Supplementary Information for

## Excellent Intrinsic Chern insulators: Monolayer PdTaX<sub>2</sub> (X=Se, Te)

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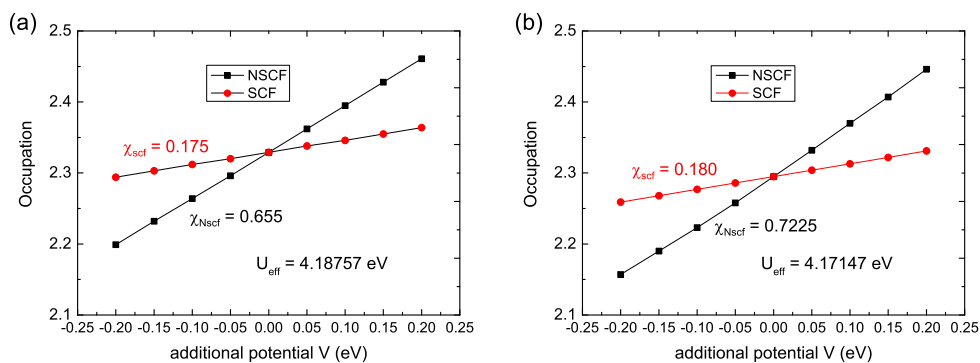


Figure S1: The  $U_{eff}$  for Ta in PdTaSe<sub>2</sub> and PdTaTe<sub>2</sub> monolayers.

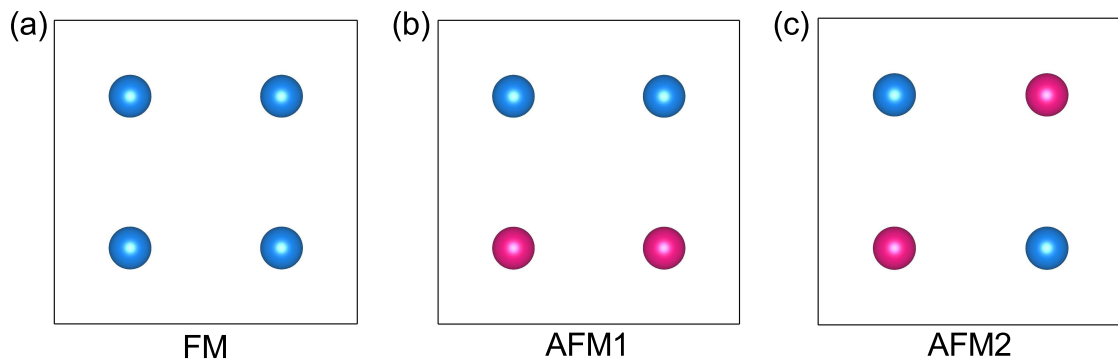


Figure S2: The different magnet configurations we employed to test the ground state, in which the  $2 \times 2 \times 1$  supercell is used. The blue and purple represent the spin up or spin down Ta atoms, respectively.

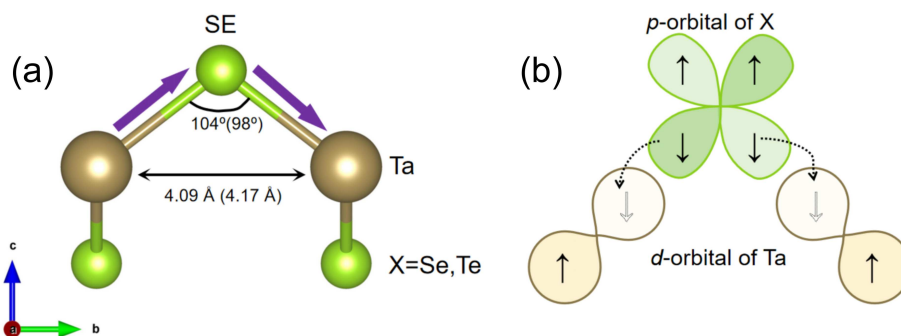


Figure S3: (a) The schematic diagram of the super-exchange (SE, Purple arrow), in which X atom represent Se or Te atoms. The angle and distance of Ta-Se-Ta(Ta-Te-Ta) is  $104^\circ(98^\circ)$  and  $4.09(4.17)$  Å. (b) Schematic diagram of the electrons hopping between X and Ta atoms.

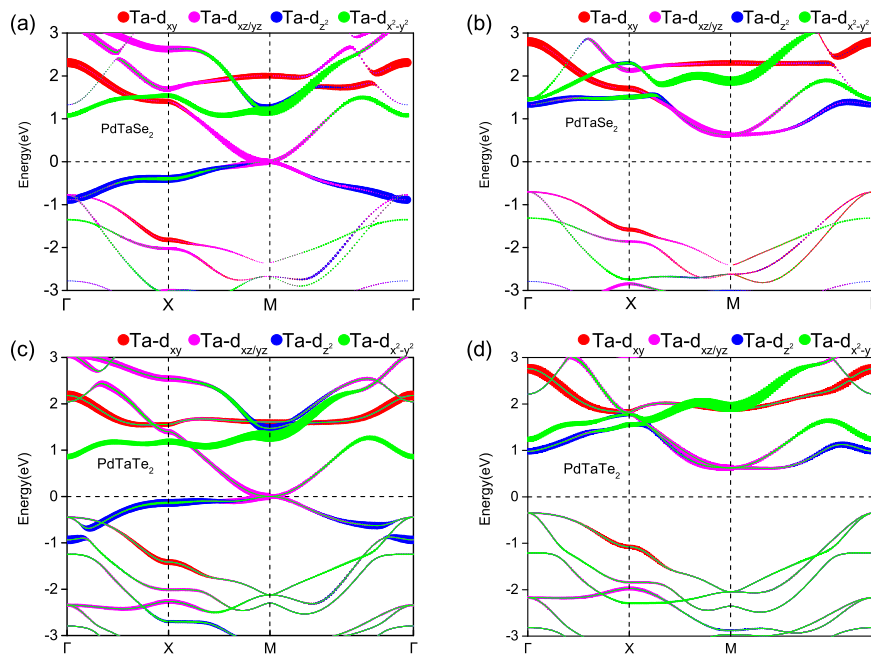
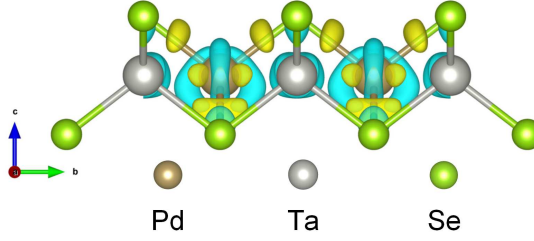
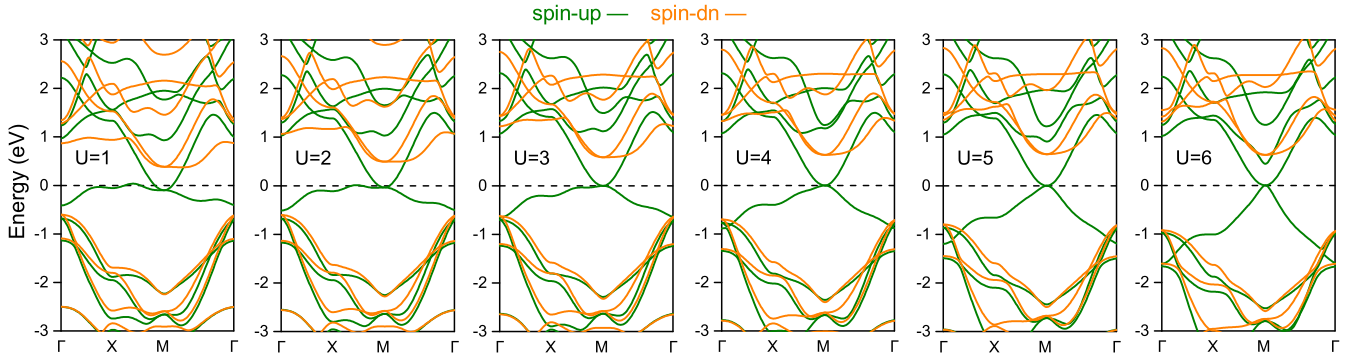
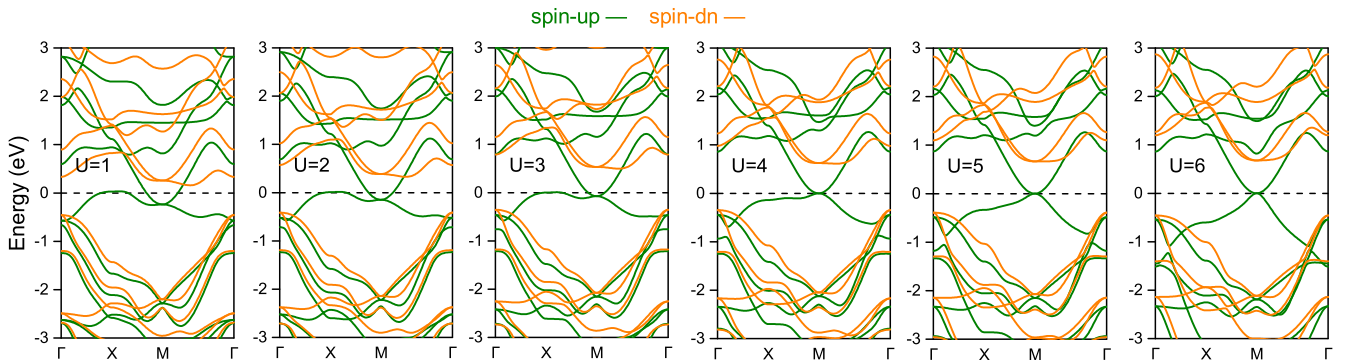


Figure S4: The orbital resolved band structures of PdTaSe<sub>2</sub> and PdTaTe<sub>2</sub> monolayers, respectively. (a) and (b) are the spin-up and spin-down channel for the d orbitals of Ta for PdTaSe<sub>2</sub>, while (c) and (d) depict the corresponding channels for PdTaTe<sub>2</sub>.

Table S1: The Bader charge transfer of the two monolayers.

Elements	Pd	Ta	Se
PdTaSe <sub>2</sub>	+0.479 e	-1.681 e	+0.601 e
PdTaTe <sub>2</sub>	+0.667 e	-1.473 e	+0.403 e

Figure S5: Differential charge density of PdTaSe<sub>2</sub>, where yellow(blue) represents gained(lost) electrons.Figure S6: The band structures with the  $U_{eff}$  from 1 to 6 eV for PdTaSe<sub>2</sub>. All of the  $E_f$  are set to zero.Figure S7: The band structures with the  $U_{eff}$  from 1 to 6 eV for PdTaTe<sub>2</sub>. All of the  $E_f$  are set to zero.

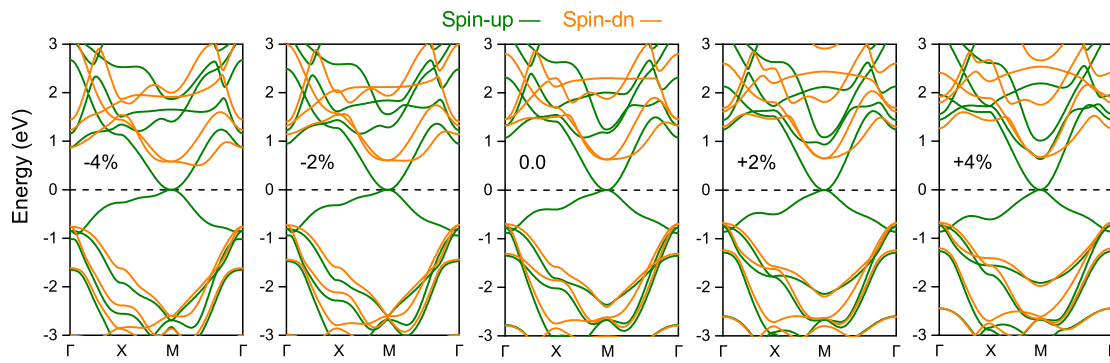


Figure S8: The band structures of PdTaSe<sub>2</sub> monolayer under biaxial strains ranging from  $-4\%$  to  $+4\%$ . All of the  $E_f$  are set to zero.

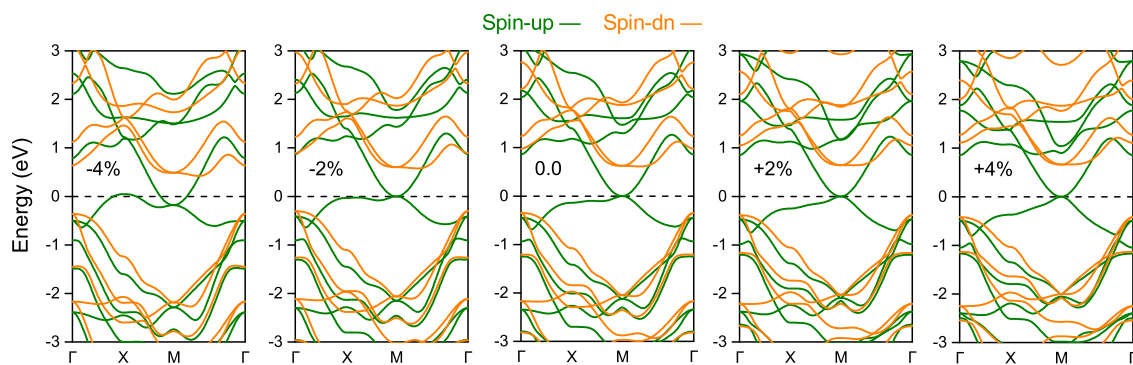


Figure S9: The figures present the band structures of PdTaTe<sub>2</sub> monolayer under biaxial strains ranging from  $-4\%$  to  $+4\%$ . All of the  $E_f$  are set to zero.

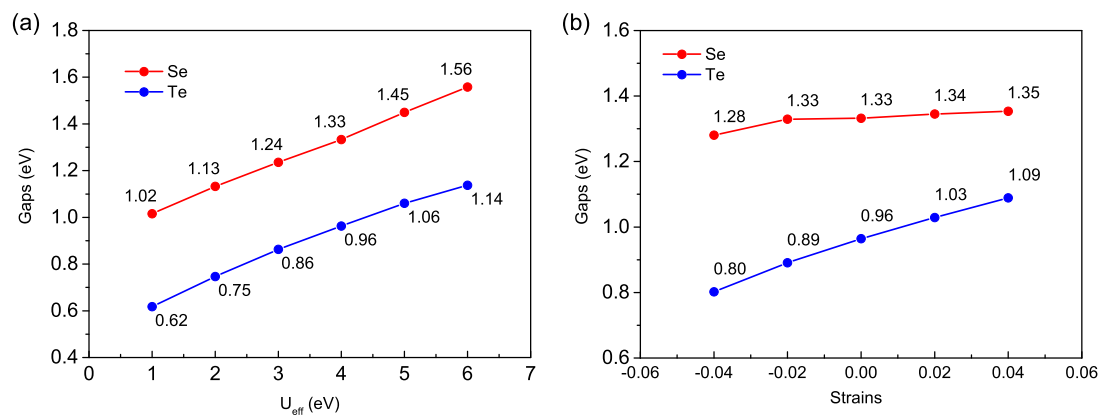


Figure S10: The spin gaps of the spin down channel for PdTaSe<sub>2</sub> (in red) and PdTaTe<sub>2</sub> (in blue) with different values of  $U_{eff}$  or a range of biaxial strains.

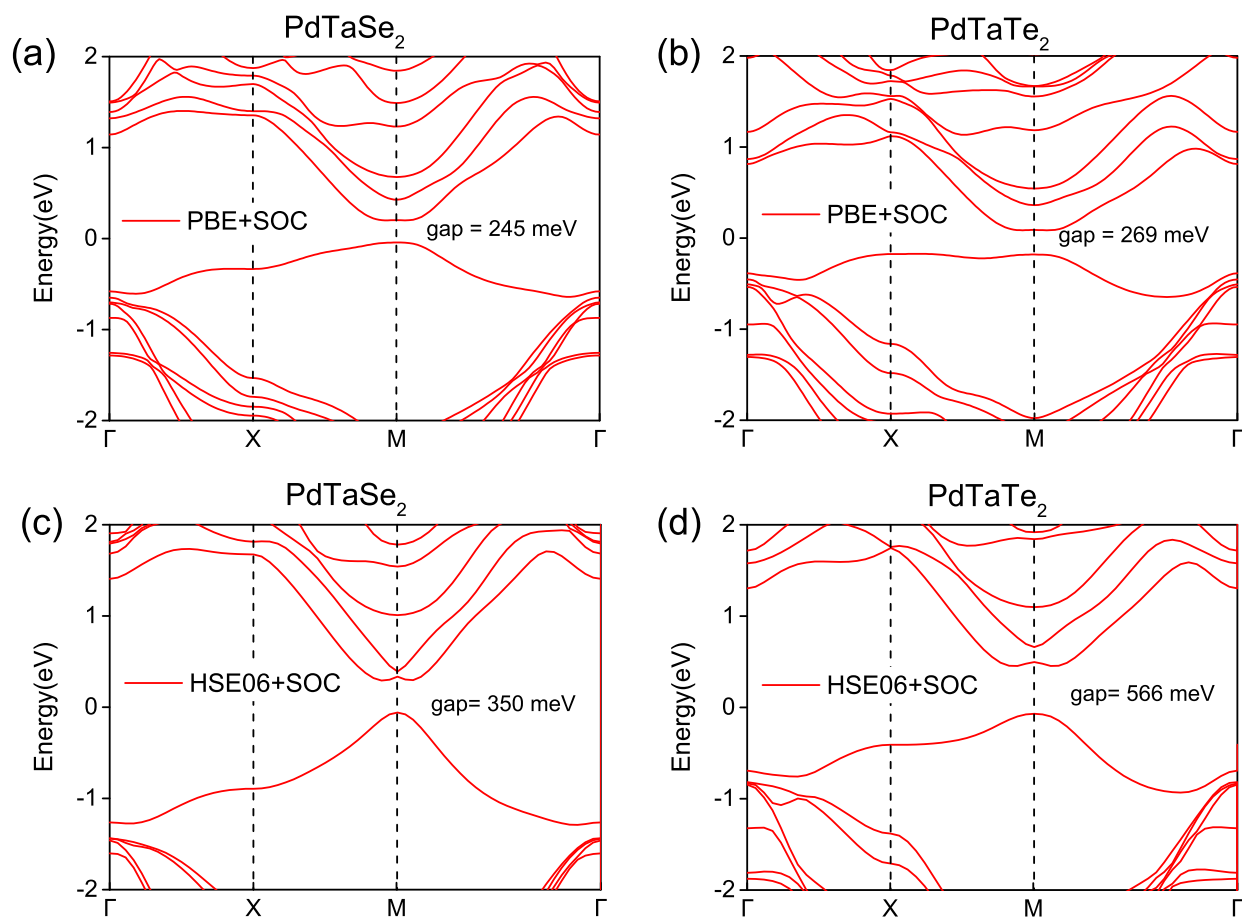


Figure S11: The orbital resolved band structures of PdTaSe<sub>2</sub> and PdTaTe<sub>2</sub> monolayers, respectively. (a) and (b) are the spin-up and spin-down channels for the d orbitals of Ta for PdTaSe<sub>2</sub>, while (c) and (d) depict the corresponding channels for PdTaTe<sub>2</sub>.

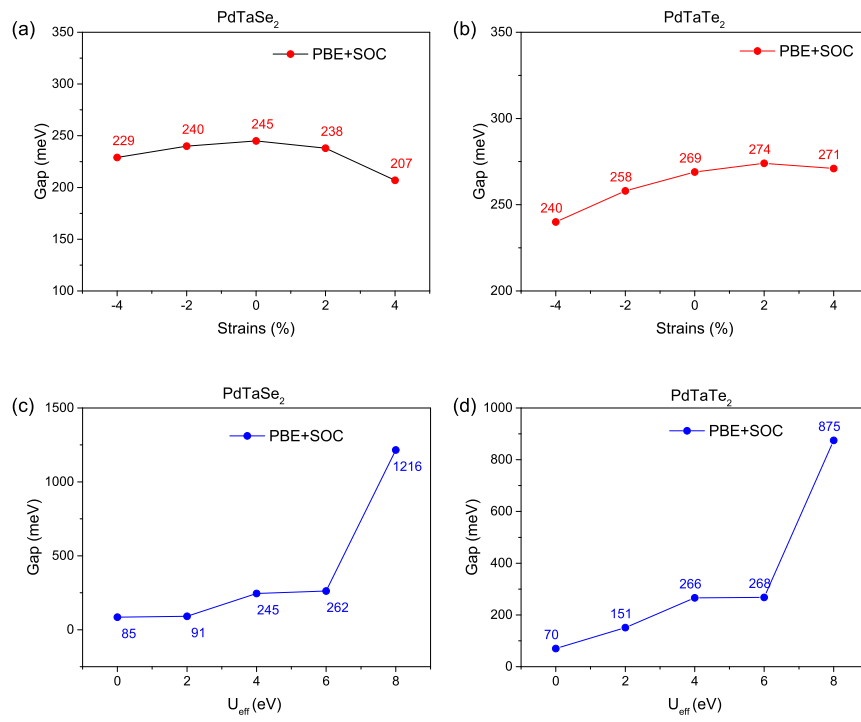


Figure S12: (a) and (b) depict the band gaps for PdTaSe<sub>2</sub> and PdTaTe<sub>2</sub> under strains ranging from -4% to 4% with spin-orbit coupling (SOC). In (c) and (d), the band gaps for these materials are presented under various  $U_{eff}$  values.

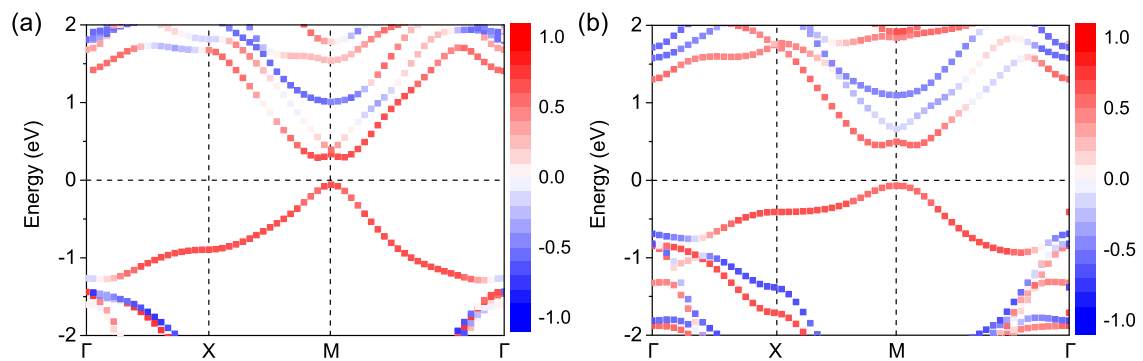


Figure S13: (a) and (b) are the normalized spin-resolved band structures of the PdTaSe<sub>2</sub> and PdTaTe<sub>2</sub> monolayers obtained by HSE06 functional in the presence of SOC, with the easy magnetization direction of the  $z$  axis.

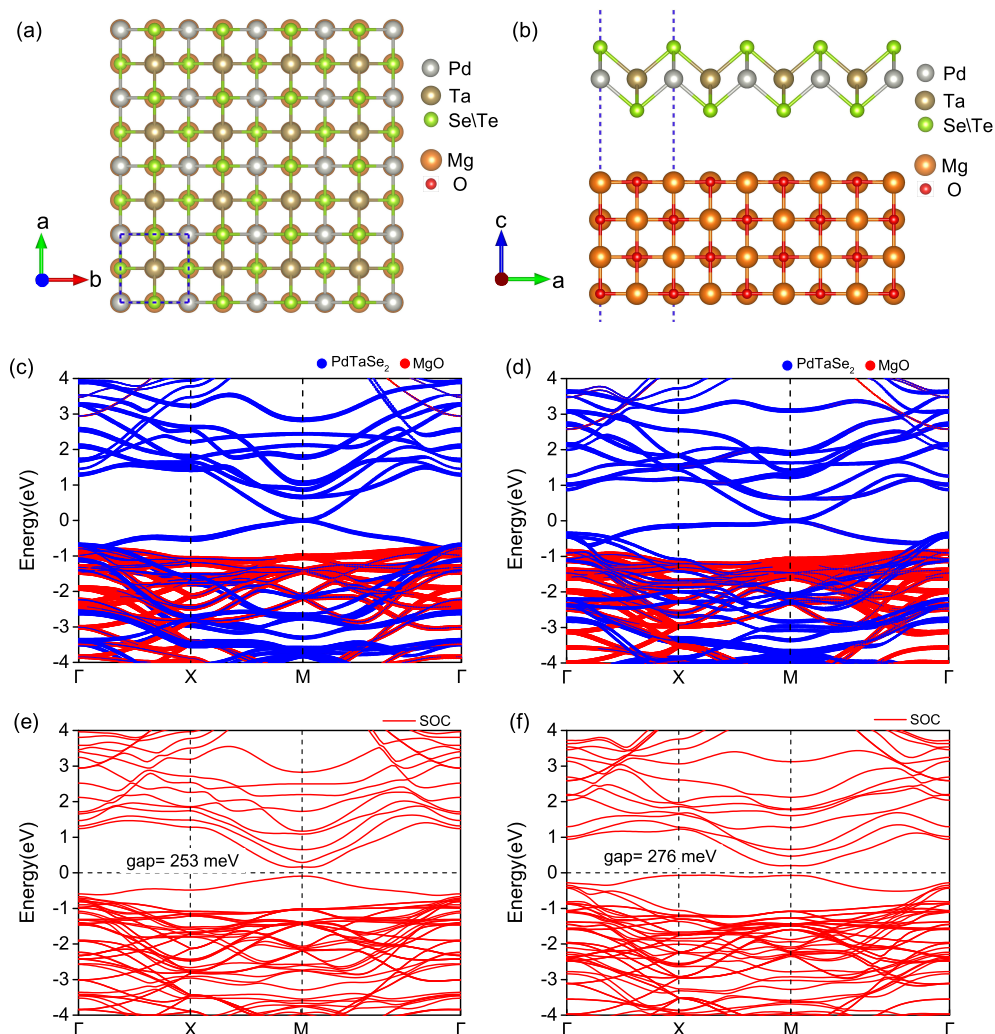


Figure S14: (a) and (b) are the top and side views of the equilibrium adsorption structure for the two monolayers adsorbed on a reconstructed MgO-(001) surface. The interlayer spaces for PdTaSe<sub>2</sub>/MgO and PdTaTe<sub>2</sub>/MgO are measured to be 3.63 Å and 4.08 Å, respectively. Band structures without spin-orbit coupling (SOC) are shown in (c) and (d) for PdTaSe<sub>2</sub>/MgO and PdTaTe<sub>2</sub>/MgO, while (e) and (f) display the band structures with SOC considered.

Table S2: The comparison of some previously proposed 2D FM materials with ours.  $T_c$  is Curie temperature. TG or T denotes whether the gap type is topological and global or topological but not global. The calculation or fabrication methods are also listed for each materials. (ME: Mechanical exfoliation, MBE: molecular beam epitaxy, SF: self-flux method)

Materials	$T_c$	gap	TG or T	QAHE	methods	ref.
CrI <sub>3</sub>	45	/	/	N	ME	Comput. Mater. Sci. 197, 110638 (2021)
CrGeTe <sub>3</sub>	30	/	/	N	ME	Nature 546, 265 (2017)
Fe <sub>5</sub> GeTe <sub>2</sub>	280	/	/	N	ME	ACS nano 13, 4436 (2019).
VSe <sub>2</sub> /MoS <sub>2</sub>	300	/	/	N	MBE	Nanotechnol. 13, 289 (2018).
Cr <sub>3</sub> Te <sub>4</sub>	344	/	/	N	MBE	Adv. Mater. 33, 2103360 (2021)
Fe <sub>3</sub> GaTe <sub>2</sub>	380	/	/	N	SF	Nat. Commun. 13, 5067 (2022)
GaFeSe	1280	/	/	N	DFT	Phys. Rev. B 106, 125404 (2022)
InFeSe	1028	403	TG	Y	DFT	Phys. Rev. B 106, 125404 (2022)
TlFeSe	980	/	/	N	DFT	Phys. Rev. B 106, 125404 (2022)
MnBi <sub>2</sub> Se <sub>4</sub> /Bi <sub>2</sub> Se <sub>3</sub>	>300	100	TG	Y	MBE	Nano Lett. 17, 3493 (2017).
(Bi <sub>0.8</sub> Mn <sub>0.2</sub> ) <sub>2</sub> Se <sub>3</sub>	12	16	TG	Y	DFT	Phys. Rev. Lett. 109, 076801 (2012).
Co <sub>3</sub> Pb <sub>3</sub> Se <sub>2</sub>	42	77	TG	Y	DFT	Phys. Rev. B 103, 014410 (2021).
Pb <sub>3</sub> Pt <sub>2</sub> Cl <sub>9</sub>	248	100	TG	Y	DFT	Phys. Rev. B 104, 205401 (2021)
TiTe	650	309	TG	Y	DFT	Nano Lett. 22, 5379 (2022).
LiFeSe	1500	35	TG	Y	DFT	Phys. Rev. Lett. 125, 086401 (2020)
Fe <sub>2</sub> I <sub>2</sub> Br	403	260	TG	Y	DFT	Nanoscale 13, 12956 (2021)
Fe <sub>2</sub> I <sub>2</sub> Cl	429	223	TG	Y	DFT	Nanoscale 13, 12956 (2021)
FeBr	417	166	TG	Y	DFT	J. Mater. Chem. C 10, 8381 (2022).
FeS	/	628	T	Y	DFT	Nat. Commun. 13, 919 (2022).
FeSe	/	798	T	Y	DFT	Nat. Commun. 13, 919 (2022).
PdTaSe <sub>2</sub>	1049	334	TG	Y	DFT	/
PdTaTe <sub>2</sub>	1003	692	TG	Y	DFT	/