

Crystal-Field Regulation Enables Broadband-to-Line Emission Switching in Cr³⁺-Activated Pyroxenes

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Figure S1. PL spectra of (a, b) LiScSi₂O₆:0.04Cr³⁺ and (d, e) LiScGe₂O₆:0.04Cr³⁺ under compression and decompression. The intensity of ²E→⁴A₂ transition under HP, ⁴T₂→⁴A₂ transition at 0.3 GPa and after release of (c) LiScSi₂O₆:0.04Cr³⁺ and (f) LiScGe₂O₆:0.04Cr³⁺. The pressure-dependent ²E→⁴A₂ transition of (g) LiScSi₂O₆:0.04Cr³⁺, (h) LiScGe₂O₆:0.04Cr³⁺ (i) ruby under HP. (j) The comparative analysis of PL spectra of LiScSi₂O₆:0.04Cr³, LiScGe₂O₆:0.04Cr³⁺ and ruby at 16.0 GPa and 22.0 GPa.

Figure S2. Diagram of the bond lengths and bond angles of ScO₆ in (a) C₂/c and (b) P₂₁/c phases. The calculated Sc-O distances and O-Sc-O angles as a function of pressure in (c, d) LiScSi₂O₆ and (e, f) LiScGe₂O₆.

Figure S3. Partial enlargements of (a) *in situ* HP XRD pattern and (b) Lattice volume of LiScSi₂O₆. β angle of (c) LiScSi₂O₆ and (d) LiScGe₂O₆ as a function of pressure.

Figure S4. LeBail fitting results of (a) LiScSi₂O₆ at 0.4 GPa, 3.0 GPa, 24.1 GPa and (b) LiScGe₂O₆ at 1.1 GPa, 20.3 GPa.

Figure S5. PL spectra of (a) LiScSi₂O₆:0.02Eu³⁺ and (b) LiScGe₂O₆:0.02Eu³⁺ measured at 80 K and 300 K.

Figure S6. Crystal structures of (a) LiScSi₂O₆ and (b) LiScGe₂O₆. The twist angles of (c) SiO₄ tetrahedra chains in LiScSi₂O₆ and (d) GeO₄ tetrahedra chains in LiScGe₂O₆.

Table S1. Energy level transition and crystal field strength (D_q/B) of Some Cr³⁺ activated Phosphors at ambient conditions.

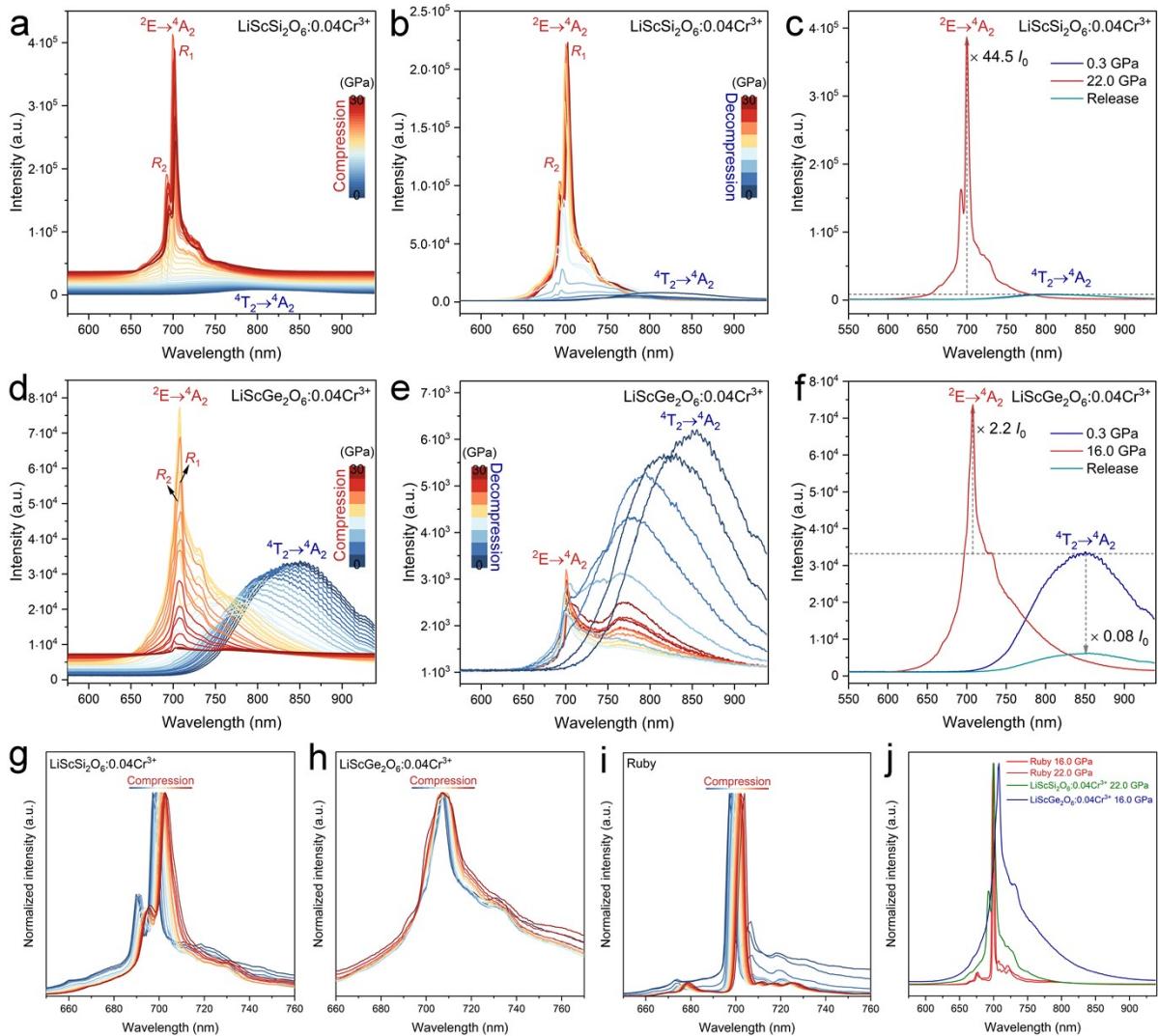


Figure S1. PL spectra of (a, b) $\text{LiScSi}_2\text{O}_6:0.04\text{Cr}^{3+}$ and (d, e) $\text{LiScGe}_2\text{O}_6:0.04\text{Cr}^{3+}$ under compression and decompression. The intensity of $^2\text{E}\rightarrow^4\text{A}_2$ transition under HP, $^4\text{T}_2\rightarrow^4\text{A}_2$ transition at 0.3 GPa and after release of (c) $\text{LiScSi}_2\text{O}_6:0.04\text{Cr}^{3+}$ and (f) $\text{LiScGe}_2\text{O}_6:0.04\text{Cr}^{3+}$. The pressure-dependent $^2\text{E}\rightarrow^4\text{A}_2$ transition of (g) $\text{LiScSi}_2\text{O}_6:0.04\text{Cr}^{3+}$, (h) $\text{LiScGe}_2\text{O}_6:0.04\text{Cr}^{3+}$ (i) ruby under HP. (j) The comparative analysis of PL spectra of $\text{LiScSi}_2\text{O}_6:0.04\text{Cr}^{3+}$, $\text{LiScGe}_2\text{O}_6:0.04\text{Cr}^{3+}$ and ruby at 16.0 GPa and 22.0 GPa.

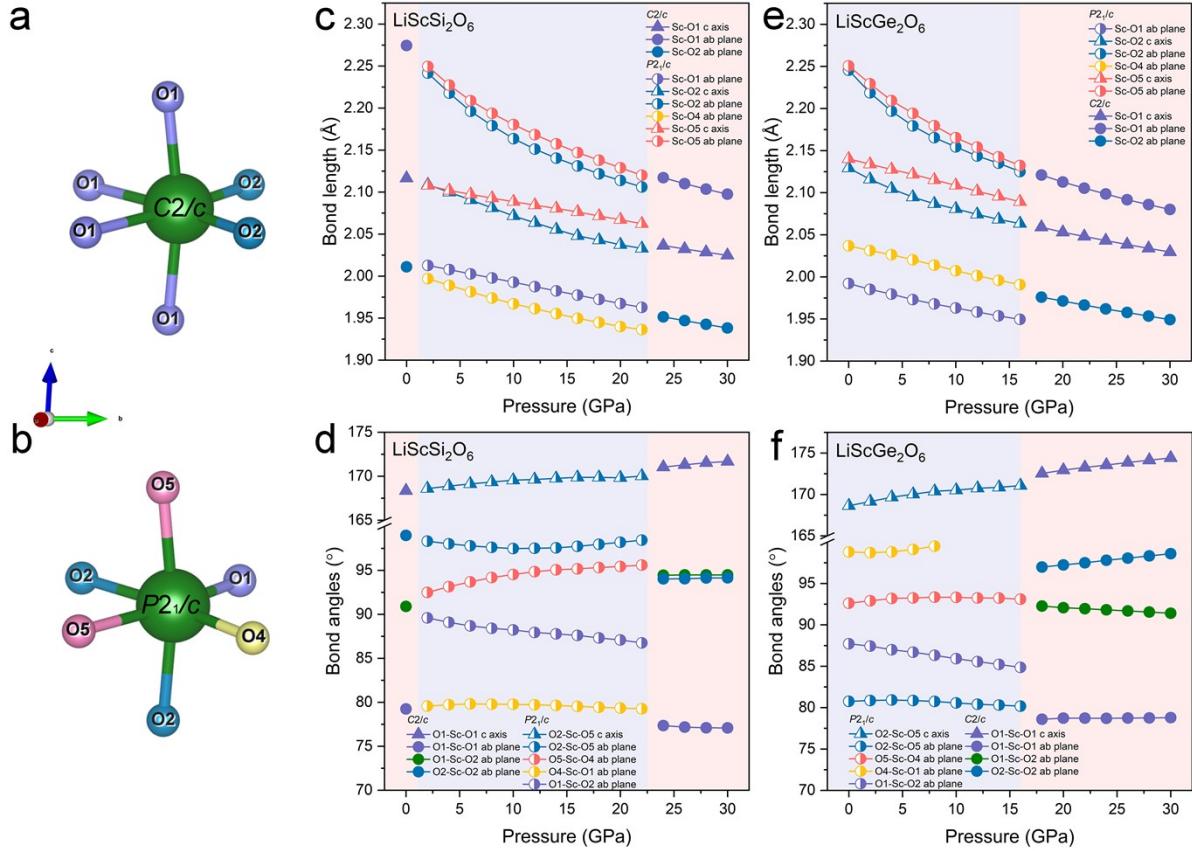


Figure S2. Diagram of the bond lengths and bond angles of ScO_6 in (a) $C2/c$ and (b) $P2_1/c$ phases. The calculated Sc-O distances and O-Sc-O angles as a function of pressure in (c, d) $\text{LiScSi}_2\text{O}_6$ and (e, f) $\text{LiScGe}_2\text{O}_6$.

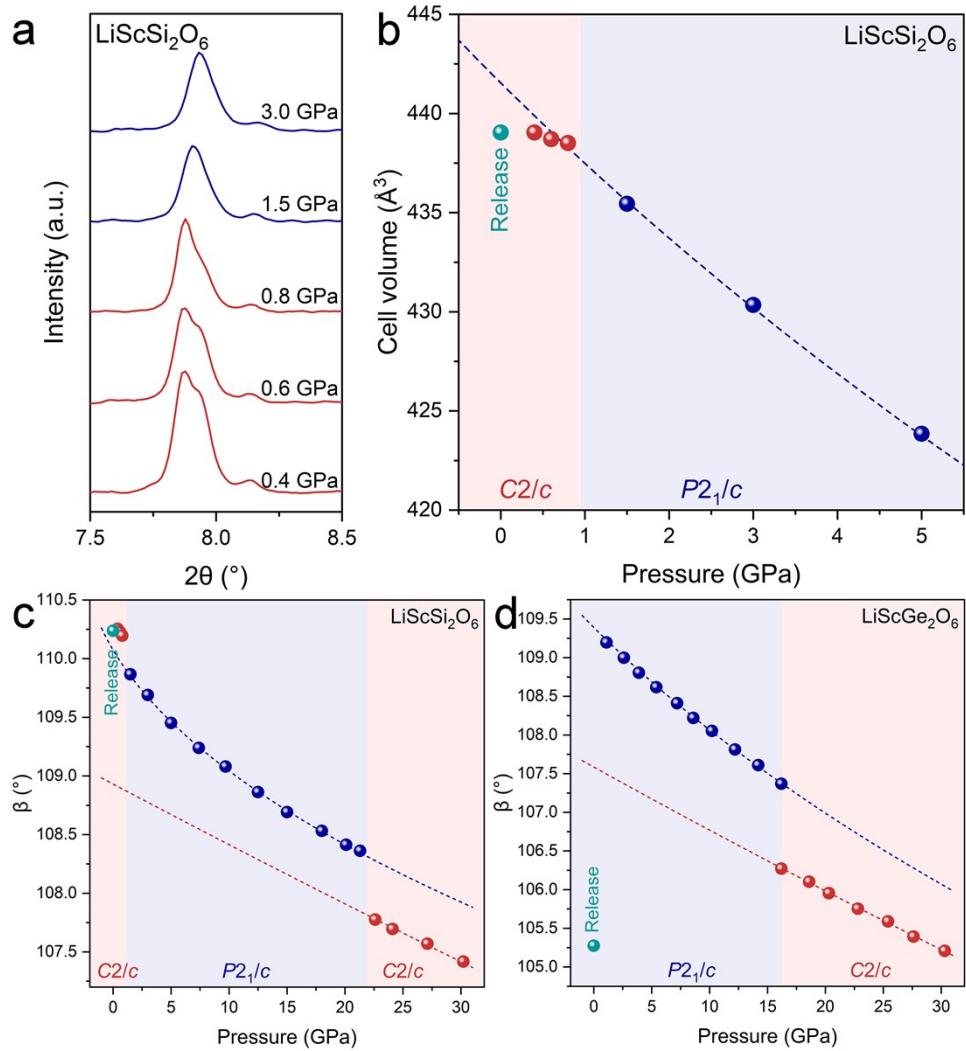


Figure S3. Partial enlargements of (a) *in situ* HP XRD pattern and (b) Lattice volume of $\text{LiScSi}_2\text{O}_6$. β angle of (c) $\text{LiScSi}_2\text{O}_6$ and (d) $\text{LiScGe}_2\text{O}_6$ as a function of pressure.

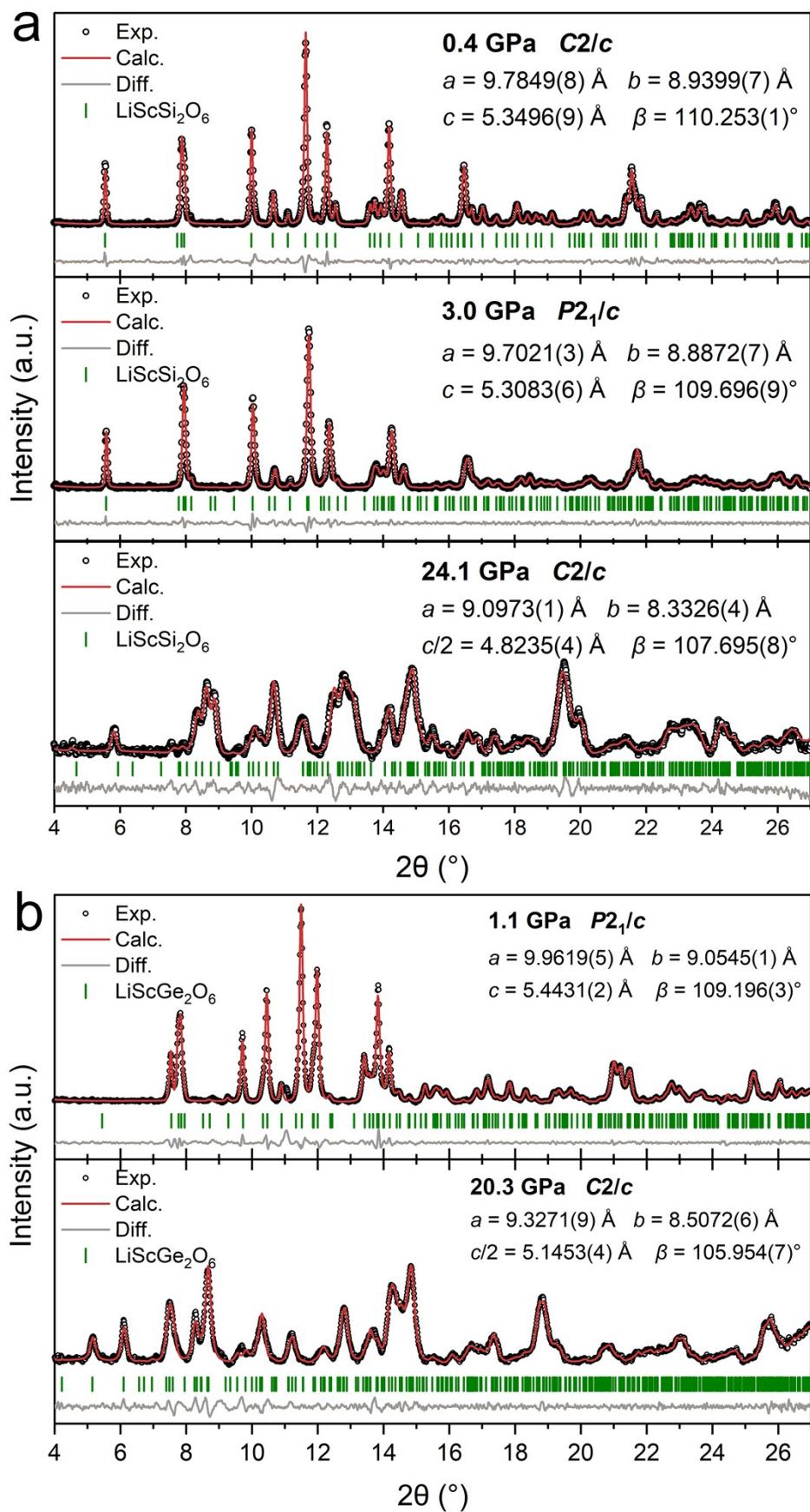


Figure S4. LeBail fitting results of (a) $\text{LiScSi}_2\text{O}_6$ at 0.4 GPa, 3.0 GPa, 24.1 GPa and (b) $\text{LiScGe}_2\text{O}_6$ at 1.1 GPa, 20.3 GPa.

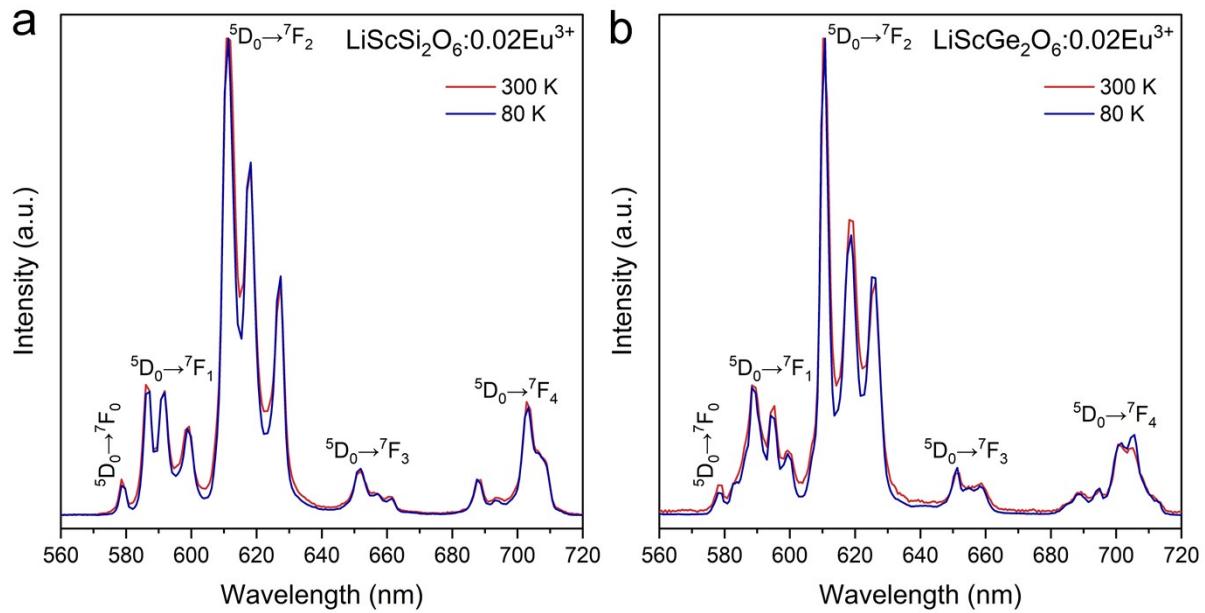


Figure S5. PL spectra of (a) $\text{LiScSi}_2\text{O}_6:0.02\text{Eu}^{3+}$ and (b) $\text{LiScGe}_2\text{O}_6:0.02\text{Eu}^{3+}$ measured at 80 K and 300 K.

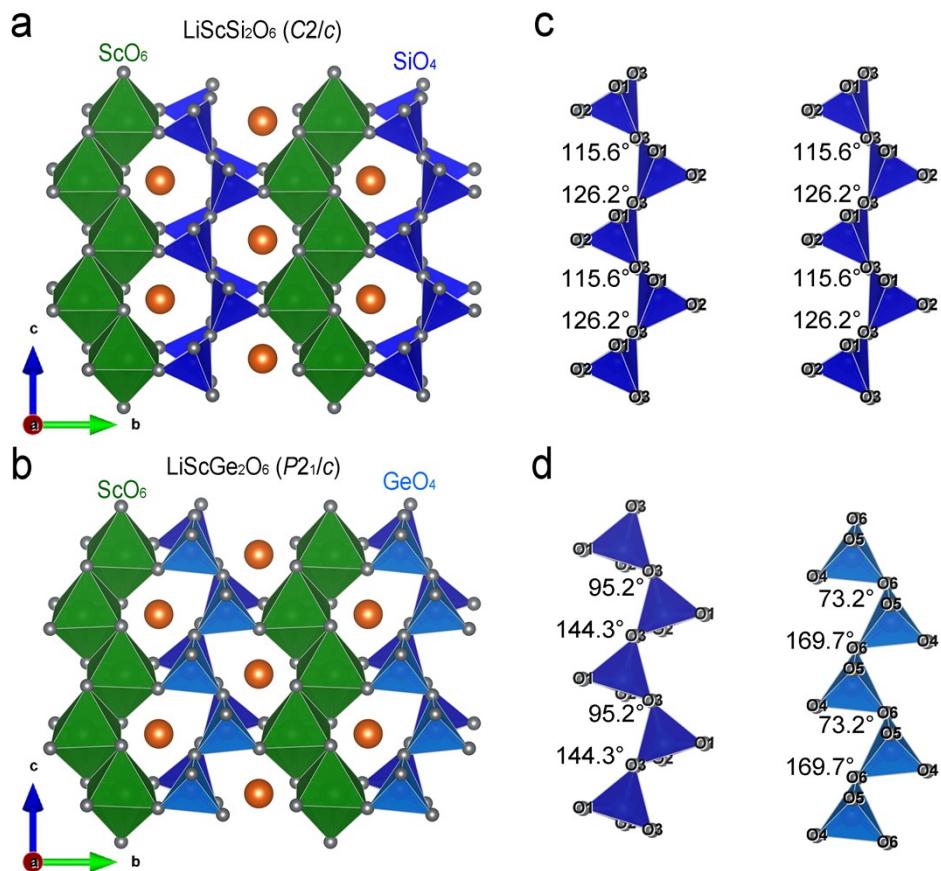


Figure S6. Crystal structures of (a) $\text{LiScSi}_2\text{O}_6$ and (b) $\text{LiScGe}_2\text{O}_6$. The twist angles of (c) SiO_4 tetrahedra chains in $\text{LiScSi}_2\text{O}_6$ and (d) GeO_4 tetrahedra chains in $\text{LiScGe}_2\text{O}_6$.

Table S1. Energy level transition and crystal field strength (D_q/B) of Some Cr³⁺ activated Phosphors at ambient conditions.

Materials	Cr sites	λ_{em} (nm)	Transition	D_q/B
LiInGe ₂ O ₆ :Cr	InO ₆	880	$^4T_2 \rightarrow ^4A_2$	1.66
LiInSi ₂ O ₆ :Cr	InO ₆	840	$^4T_2 \rightarrow ^4A_2$	1.75
InBO ₃ :Cr	InO ₆	820	$^4T_2 \rightarrow ^4A_2$	1.78
LiScP ₂ O ₇ :Cr	ScO ₆	880	$^4T_2 \rightarrow ^4A_2$	1.84
LiScSi ₂ O ₆ :Cr	ScO ₆	850	$^4T_2 \rightarrow ^4A_2$	1.87
NaInGe ₂ O ₆ :Cr	InO ₆	900	$^4T_2 \rightarrow ^4A_2$	1.89
LiScGe ₂ O ₆ :Cr	ScO ₆	844-886	$^4T_2 \rightarrow ^4A_2$	1.79-1.99
Ca ₂ LuZr ₂ Al ₃ O ₁₂ :Cr	CaO ₈ , LuO ₈ , ZrO ₆	650-850	$^4T_2 \rightarrow ^4A_2$	Cr1 1.92
				Cr2 2.38
Sr ₂ ScSbO ₆ :Cr	ScO ₆	890	$^4T_2 \rightarrow ^4A_2$	2.03
LiInP ₂ O ₇ :Cr	InO ₆	860	$^4T_2 \rightarrow ^4A_2$	2.09
CaMgGe ₂ O ₆ :Cr	MgO ₆	845	$^4T_2 \rightarrow ^4A_2$	2.12
ScBO ₃ :Cr	ScO ₆	800	$^4T_2 \rightarrow ^4A_2$	2.15
Sr ₉ Ga(PO ₄) ₇ :Cr	GaO ₆	850	$^4T_2 \rightarrow ^4A_2$	2.24
GaTaO ₄ :Cr	GaO ₆	840	$^4T_2 \rightarrow ^4A_2$	2.29
Bi ₂ Ga ₄ O ₉ :Cr	GaO ₆	800	$^4T_2, ^2E \rightarrow ^4A_2$	2.31
Ba ₃ Sc ₄ O ₉ :Cr	ScO ₆	835	$^4T_2 \rightarrow ^4A_2$	Cr1 2.42
				Cr2 2.40
[EA] ₂ NaCr _x Al _{1-x} (HCOO) ₆	AlO ₆	684.4, 686.4, 752.3	$^4T_2, ^2E \rightarrow ^4A_2$	2.76-2.33
La ₂ MgZrO ₆ :Cr	MgO ₆ , ZrO ₆	740, 760, 825	$^2E, ^4T_2 \rightarrow ^4A_2$	2.53
				2.49
Al ₂ O ₃ :Cr	AlO ₆	694.28	$^2E \rightarrow ^4A_2$	2.81
LaGaO ₃ :Cr	GaO ₆	739	$^2E \rightarrow ^4A_2$	2.89
Zn _{2-x} Al _{2x} Sn _{1-x} O ₄ :Cr	AlO ₆	800-722	$^4T_2 \rightarrow ^4A_2$	2.292-3.066
BaAl ₄ Sb ₂ O ₁₂ :Cr	AlO ₆ , SbO ₆	695, 710, 750	$^2E \rightarrow ^4A_2$	[SbO ₆] 2.6 [AlO ₆] 2.9
Y ₃ Al _{5-x} Ga _x O ₁₂ :Cr	AlO ₆ , GaO ₆	688-691	$^2E \rightarrow ^4A_2$	2.998-3.048