

Supporting Information

Tracing the origin of elevated springtime atmospheric sulfate on the Southern Himalayan-Tibetan Plateau

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Supplementary Notes

Note S1. Chemical and S multi-Isotope measurements for PM_{2.5} aerosol samples collected at QOMS

Water-soluble ion analysis: A filter punch was soaked in 10 mL of Milli-Q water added by ultrasonification for 30 min. The water extracts were filtered using a 0.45 μ m Millipore filters. An ion chromatograph instrument operating at 1.0 ml min⁻¹ was used for this purpose [Metrohm IC]. Calibration of the instrument was done using commercial standards (~10 to 100 parts per million). Standard stock solutions from analytical grade reagents and salts were used for quality assurance of the analytical data (within 5%). Blank filters were also analysed with the samples and the concentrations in the aerosol samples were suitably corrected for contribution from the blanks.

Metal analysis: A filter punch was digested with ultra-pure mixing of nitric acid, hydrogen peroxide, and hydrofluoric acid (6 mL HNO₃ + 2 mL H₂O₂ + 1 mL HF) at 180°C for 6 hours adapted from the protocol detailed elsewhere¹. After filtration through a 0.45 μ m pore filter, metal concentrations were measured by inductively coupled plasma-mass spectrometry (Agilent 8900 Triple Quadrupole ICP-MS). Quality control was performed by repeated analysis of multi-element calibration standards (SCP science) along with blank correction to report the aerosol concentration.

S multi-isotope analysis: Column chemistry is used to extract sulfate from the filters as discussed in our previous publication². After purification, the extracts were analyzed for S multi-isotopes at CRPG on a Neptune Plus MC-ICP-MS (Thermo Fisher Scientific Neptune Plus) coupled with a desolvation membrane (Aridus II, Cetac) to reduce the interference of oxides and hydrides on the masses ³²S, ³³S and ³⁴S and to increase the sensitivity to the signal^{3,4}. The Paris et al., protocol³ is used for analysis by standard-sample bracketing using an in-house Na₂SO₄ standard to correct for instrumental isotope fractionation. Each sample was run a minimum of four independent times, usually five, with each run consisting of fifty acquisition of 4.2 seconds. The sample requirement for this method is ~40 nmol and as such this could become a preferable method for high-resolution S multi-isotope analysis.

The reported values (see Supplementary Table S1) are given as 1σ (standard deviation) of at least 5 independent measurements on the Neptune for a given sample. Assessment of quality control, accuracy, and precision of this method for S multi-isotope analysis in aerosols has been discussed in detail in our recent publication². The triple sulfur isotopes (³²S, ³³S, ³⁴S) are reported relative to the Vienna-Canyon Diablo Troilite (V-CDT) standard. Quality control using various standards and three separate seawater samples was conducted as detailed elsewhere², Procedural blanks were also run with each set of samples and represent less than 1% of the total sulfur of a sample.

Supplementary Figures

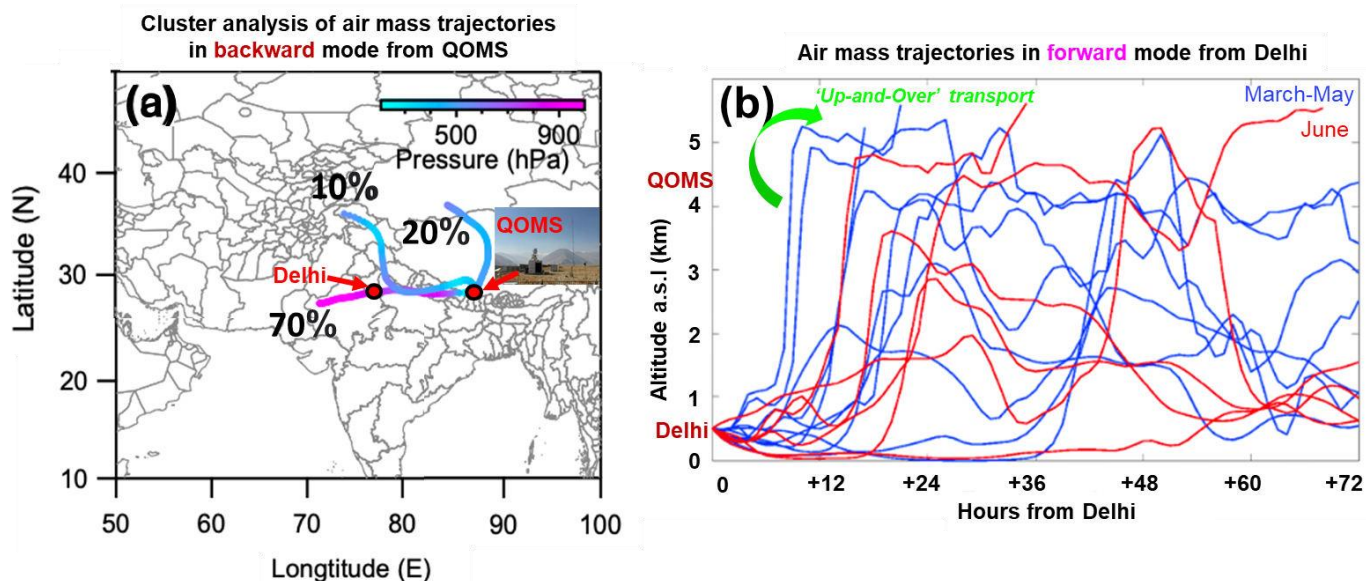


Figure S1. Air mass transport at QOMS during spring of 2021. (a) Cluster analysis of 7-day air mass back-trajectories computed hourly using NOAA Hybrid Single-Particle Lagrangian Integrated Trajectory model (HYSPLIT) version 4^{14,15}. Over the course of the campaign the air masses were predominantly from the S Asian region. As such, synoptic sampling was possible as air masses were linking the sites Delhi and QOMS, (b) Forward trajectories were computed from Delhi for which the height of air masses is shown (for 72 h). the distinct springtime feature 'up-and-over' transport can be seen wherein air masses are rapidly transported to higher altitudes from the Indo-Gangetic Plain to the HTP.

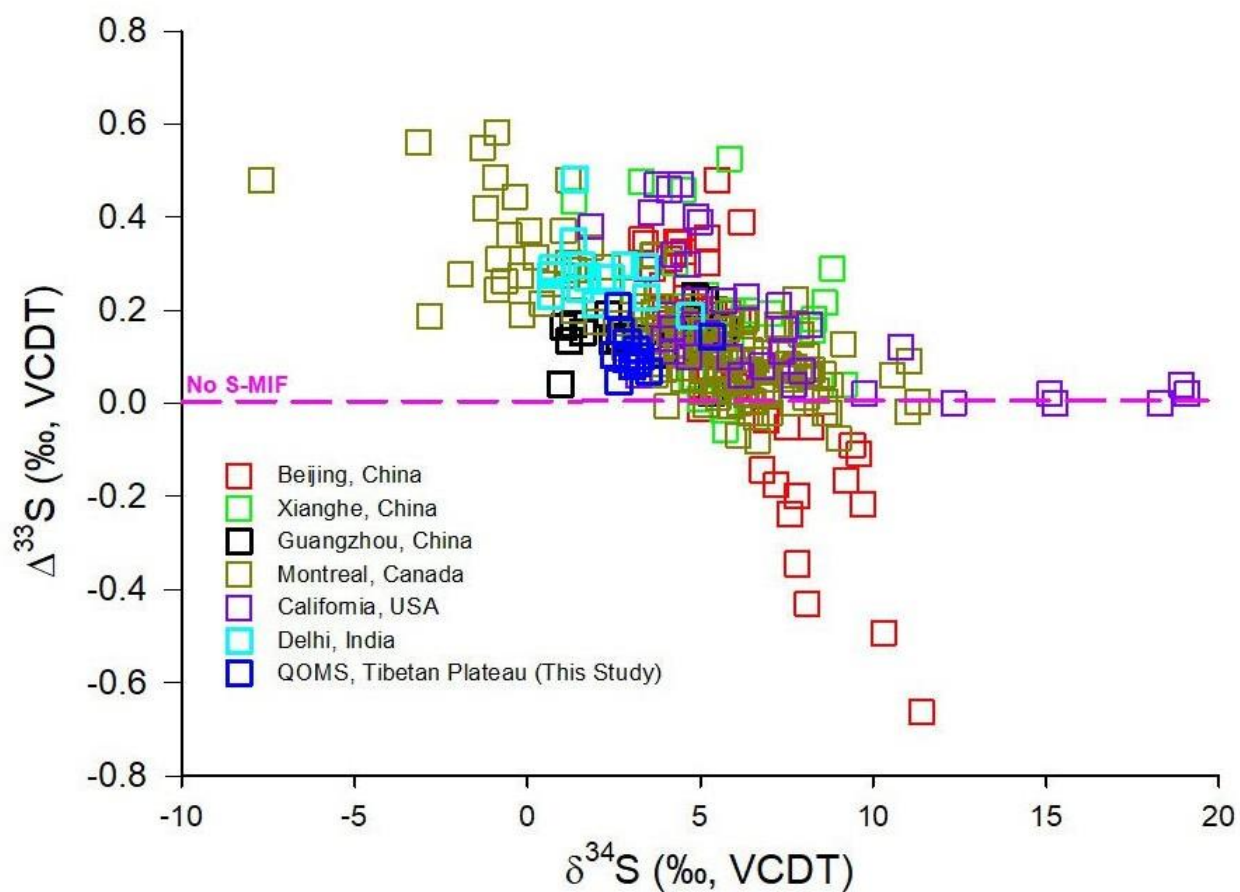


Figure S2. A comparison of aerosol sulfur isotope anomalies from published records^{2,6-11}.

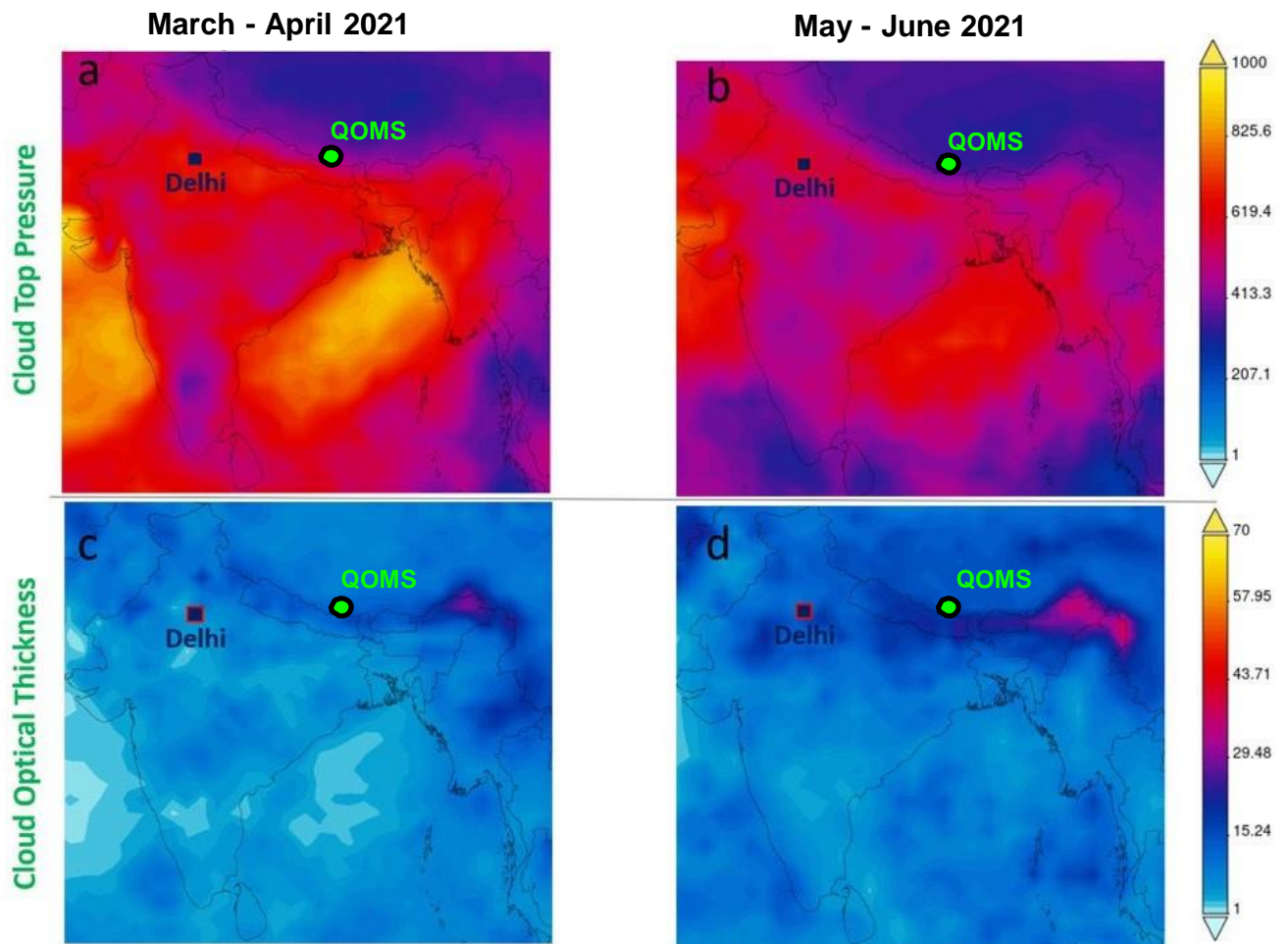


Figure S3. Time-averaged map of cloud parameters. MODIS AQUA, TERRA-based [product MYD08_D3 v6.1, MOD08_D3 v6.1]¹⁶ monthly mean of (a, b) cloud top pressure [in mb] and (c,d) cloud optical properties shown for the months of March-April and May-June 2021 over the S Asian region [<https://giovanni.gsfc.nasa.gov/giovanni/>].

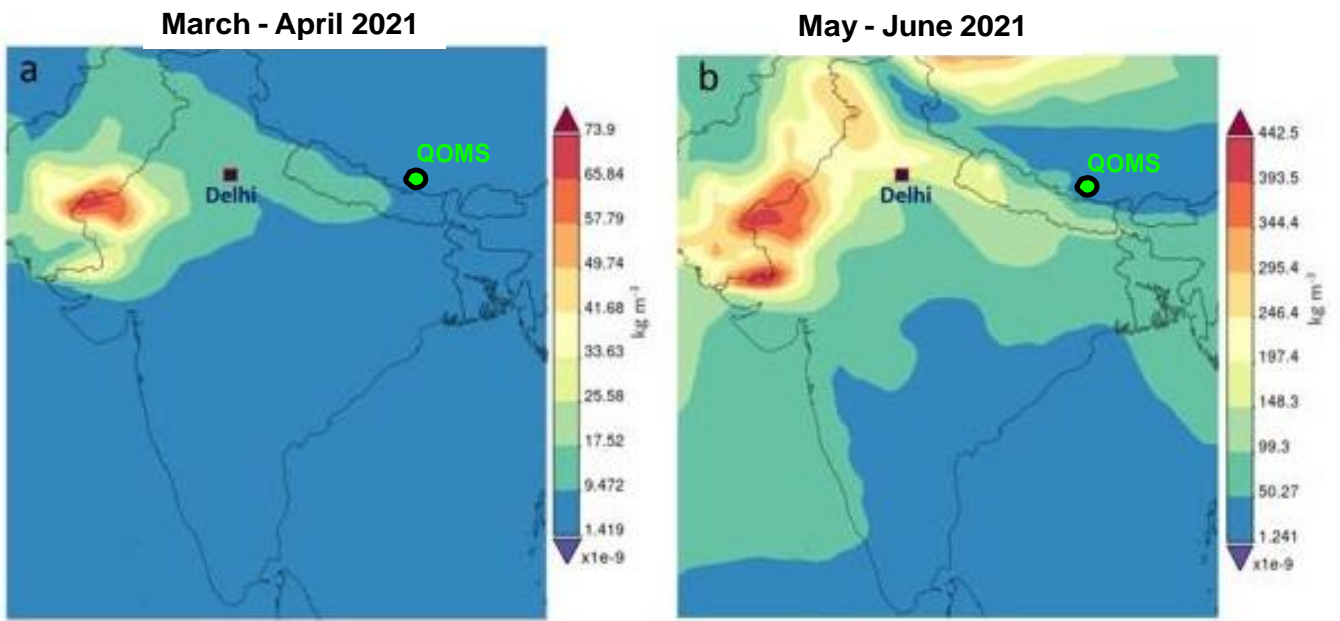


Figure S4. Time-averaged map of dust surface mass concentration. Computed monthly $0.5^\circ \times 0.625^\circ$ grid using the MERRA-2 model M2T1NXAER v5. 12.4 [<https://giovanni.gsfc.nasa.gov/giovanni/>]¹⁷.

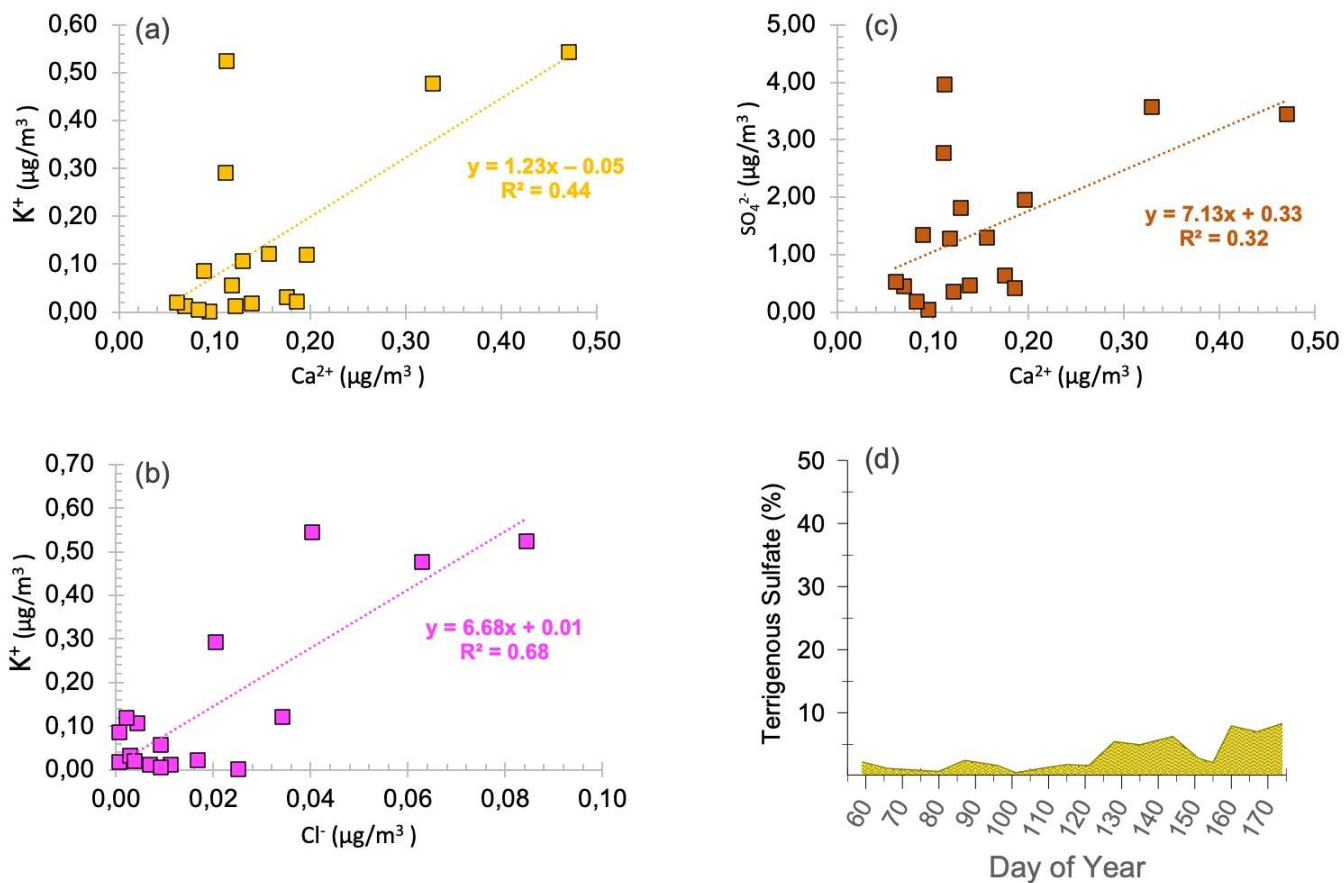


Figure S5. Establishing the role of mineral dust and biomass burning through correlations of water-soluble ions at QOMS during spring of 2021. The fraction of terrigenous sulfate in total sulfate is shown for the QOMS site. This fraction is calculated based on the ratio of SO_4^{2-} -to- Ca^{2+} concentrations in regional soil to the same in sampled aerosol⁵.

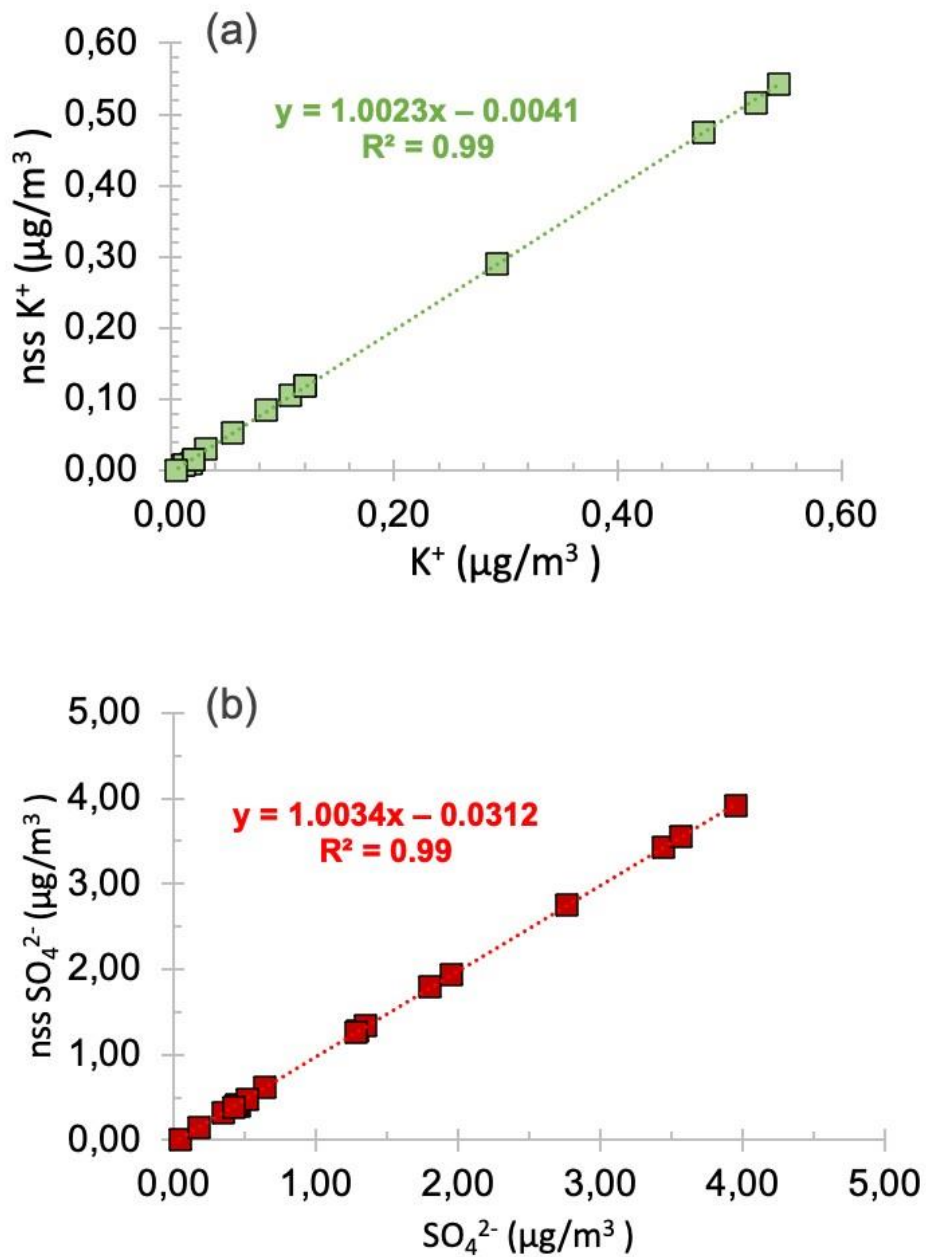


Figure S6. Marine influence on sulfate and potassium concentrations at QOMS. The non-sea salt potassium and sulfate concentrations are shown along with the total potassium and sulfate concentrations.

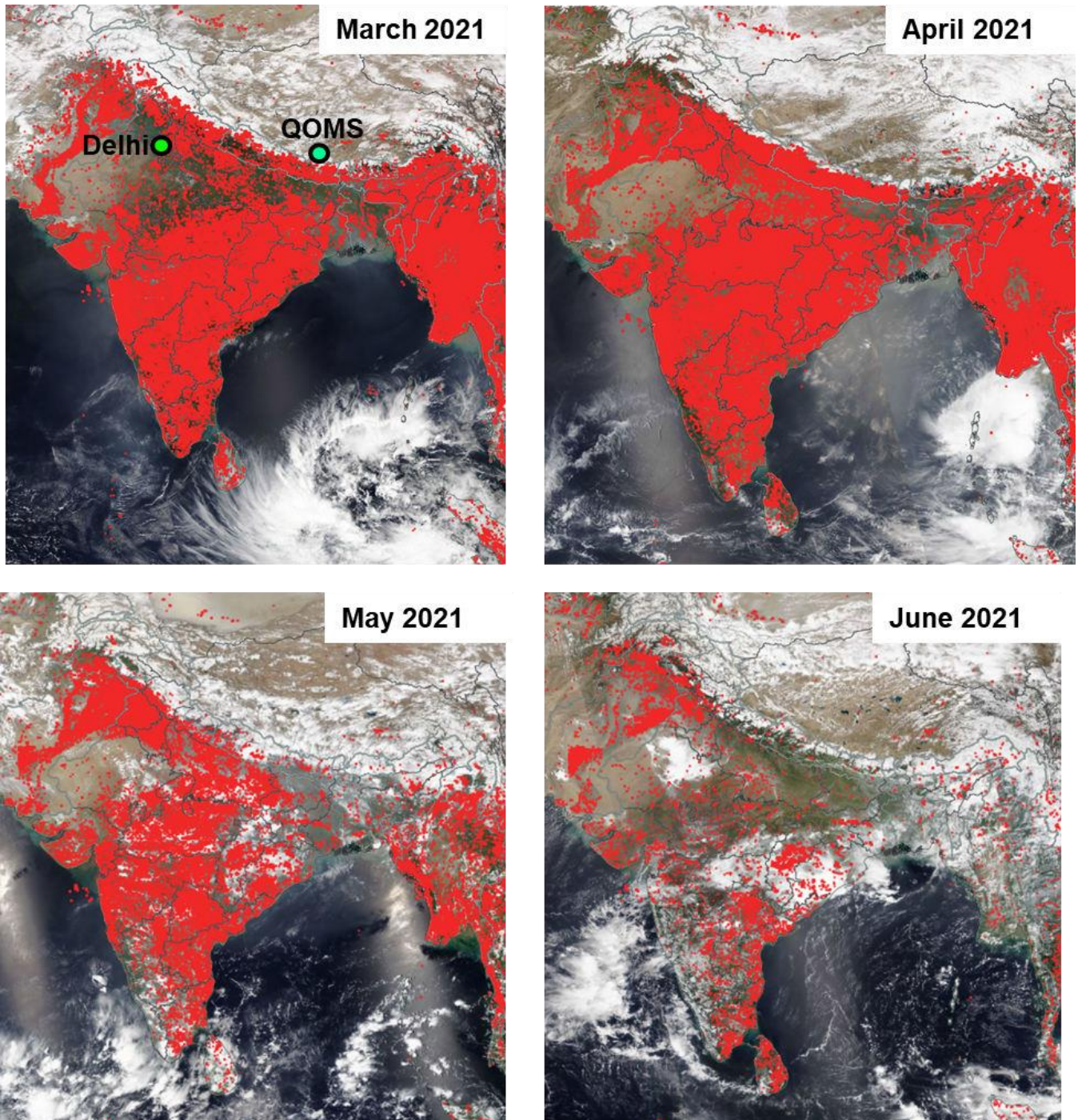


Figure S7. Open burning in S Asia during spring of 2021. The data is obtained from the Fire Information for Resource Management System (FIRMS)¹⁸ service of NASA <https://firms.modaps.eosdis.nasa.gov/>.

Table S1. Sampling details along with concentrations of water-soluble ions and S multi-isotope composition for samples collected with a frequency of 4-days at QOMS during spring 2021. Note that due to a lack of availability of sample material for isotopic analysis, the samples prior to April could not be measured for the same. However, based on a recent study¹², it is argued that the period from April till June is rather found to be devoid of influence from stratospheric intrusions (please refer to Figure 4 in Wang et al., 2021) and is precisely when a maximum influx of biomass burning aerosols is found at QOMS¹³. As such is the most suited scientifically to study the influence of biomass burning on the observed S-MIF signals at QOMS.

Sampling start date	Na ⁺ [μg/m ³]	K ⁺ [μg/m ³]	Ca ²⁺ [μg/m ³]	SO ₄ ²⁻ [μg/m ³]	δ ³⁴ [‰]	s.d.	Δ ³ [‰]	s.d.
28/02/2021	0.07	0.12	0.16	1.29	NA	NA	NA	NA
07/03/2021	0.05	0.09	0.09	1.35	NA	NA	NA	NA
21/03/2021	0.06	0.29	0.11	2.77	NA	NA	NA	NA
28/03/2021	0.04	0.54	0.47	3.44	NA	NA	NA	NA
06/04/2021	0.07	0.48	0.33	3.57	2.62	0.09	0.21	0.03
11/04/2021	0.19	0.52	0.11	3.96	2.71	0.05	0.15	0.03
19/04/2021	0.06	0.11	0.13	1.81	3.27	0.05	0.10	0.03
25/04/2021	0.06	0.12	0.20	1.96	2.87	0.03	0.10	0.02
01/05/2021	0.09	0.06	0.12	1.29	2.64	0.04	0.05	0.04
08/05/2021	0.24	0.02	0.14	0.46	2.50	0.05	0.10	0.04
15/05/2021	0.05	0.03	0.18	0.64	2.89	0.04	0.13	0.02
24/05/2021	0.11	0.01	0.12	0.35	3.13	0.04	0.11	0.04
31/05/2021	0.12	0.01	0.07	0.45	3.20	0.01	0.09	0.02
04/06/2021	0.15	0.02	0.06	0.52	3.04	0.06	0.08	0.05
09/06/2021	0.15	0.02	0.19	0.42	3.53	0.03	0.07	0.03

16/06/2021	0.14	0.00	0.09	0.05	5.34	0.04	0.14	0.02
23/06/2021	0.13	0.00	0.08	0.18	3.20	0.07	0.09	0.02

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