

## Supplementary Information

### A Highly Sensitive Ion-Selective Chemiresistive Sensor for Online Monitoring of Lead Ions in Water

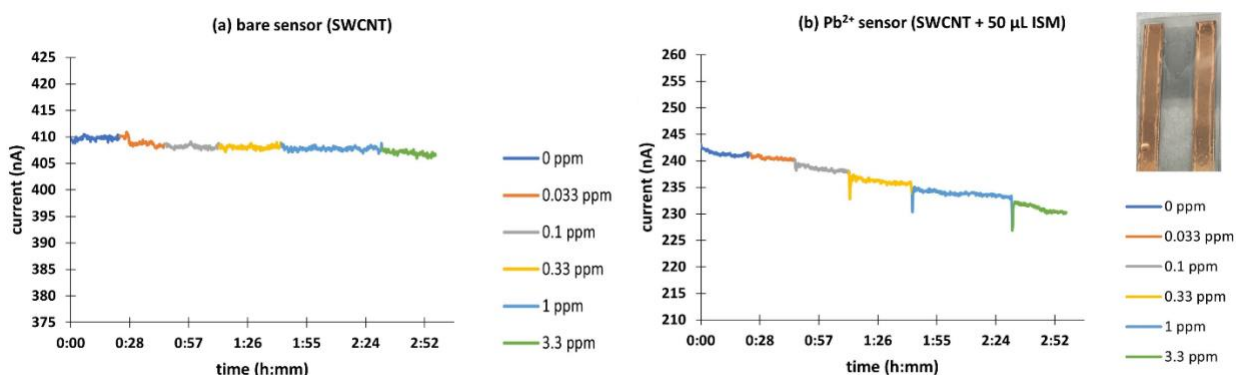
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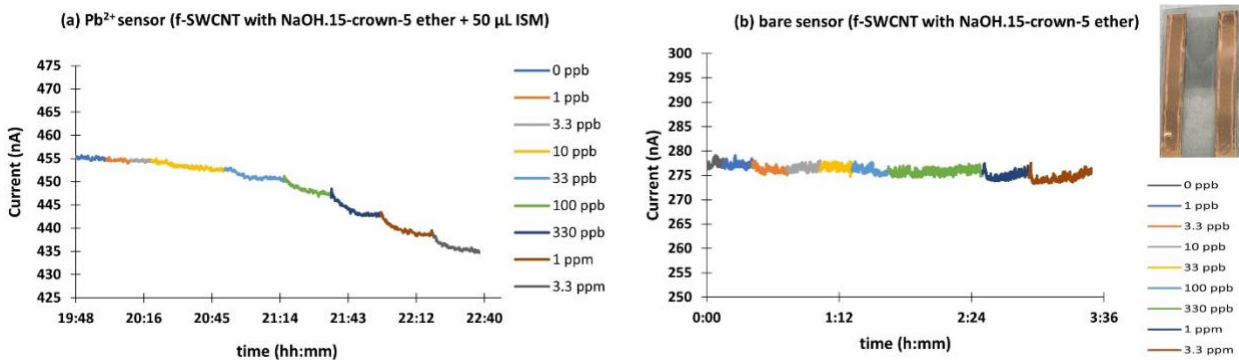
<sup>b</sup>Department of Biomedical Engineering, McMaster University, 1280 Main Street West, Hamilton, Ontario L8S 4L7, Canada

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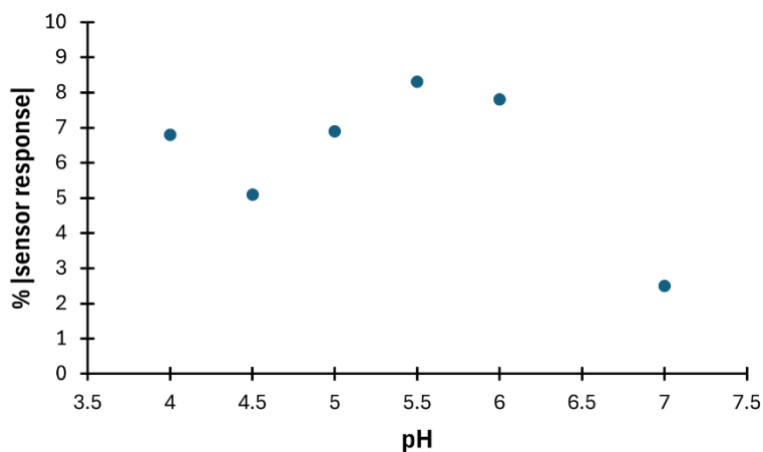
\*Corresponding Author: [pkruse@mcmaster.ca](mailto:pkruse@mcmaster.ca)



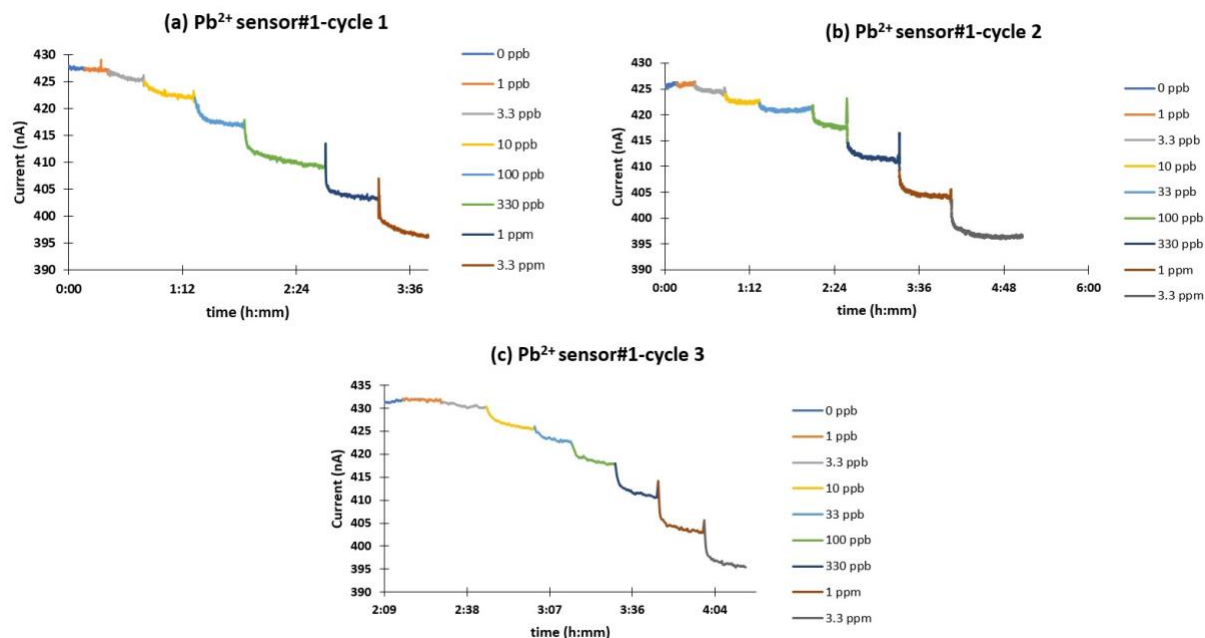
**Fig. S1** Sensitivity comparison of a bare device and a Pb<sup>2+</sup> sensor to lead ions in different concentrations (from 33 ppb to 3.3 ppm). (a) Resonance of a bare sensor made of SWCNT as a resistive film and (b) a Pb<sup>2+</sup> sensor made of SWCNT as a resistive film and coated with ISM (Drop-casted 50 µL of ISM solution on the SWCNT at the circular opening (Image of the actual sensor on the top right of the figure)).



**Fig. S2** Response comparison of a Pb<sup>2+</sup> sensor and a bare sensor to lead ions in different concentrations (from 1 ppb to 3.3 ppm). (a) Resonse of a Pb<sup>2+</sup> sensor made of f-SWCNT (with NaOH.15-crown-5 ether) as a resistive film coated with ISM (Drop-casted 50 μL of ISM solution on the SWCNT film at the circular opening) and (b) a bare sensor made of f-SWCNT (with NaOH.15-crown-5 ether) as a resistive film (Image of the actual sensor on the top right of the figure).



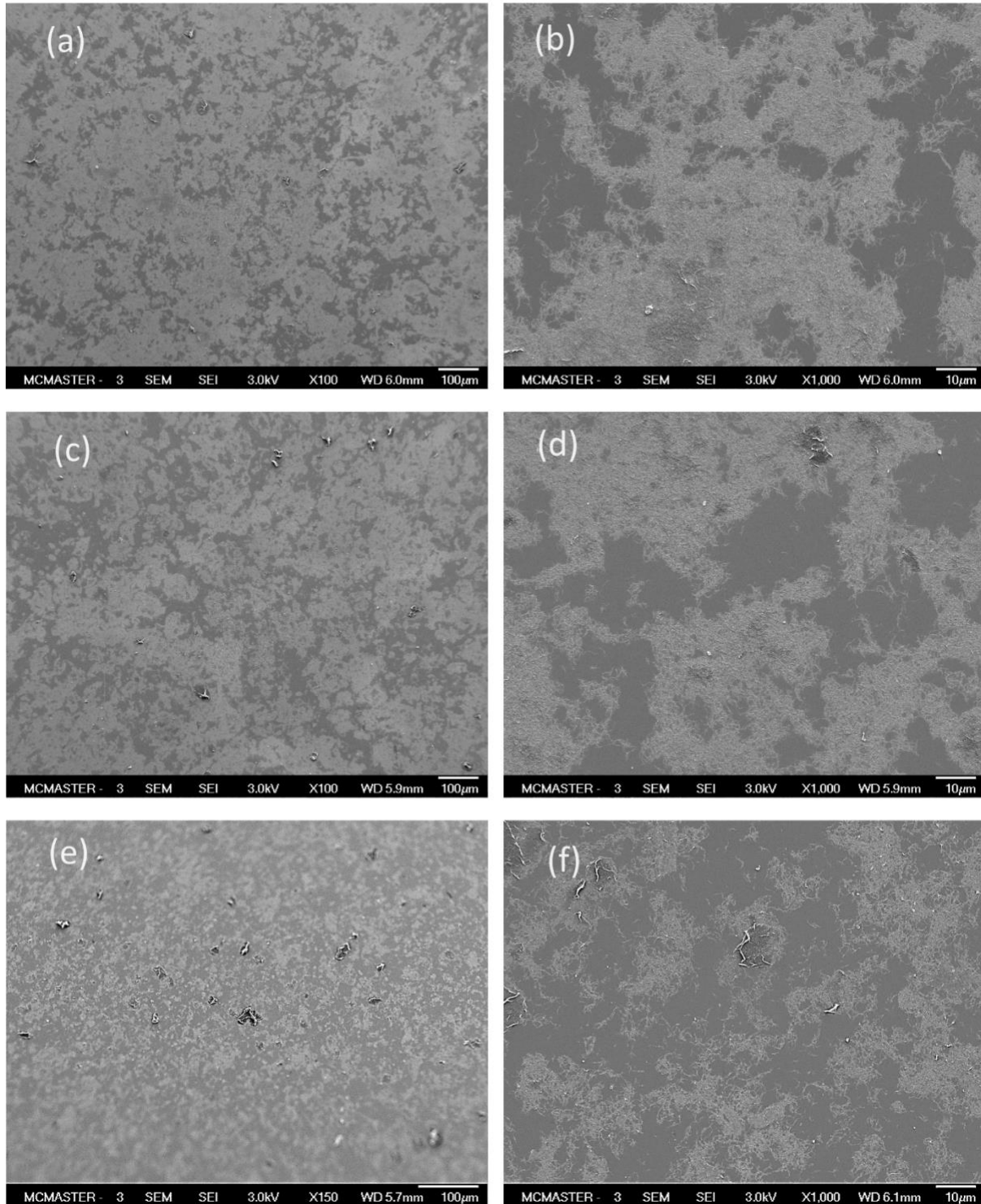
**Fig. S3** Pb<sup>2+</sup> sensor responses in a range of pH 4-7 in 200 mg/L tris-acetate buffer solution.



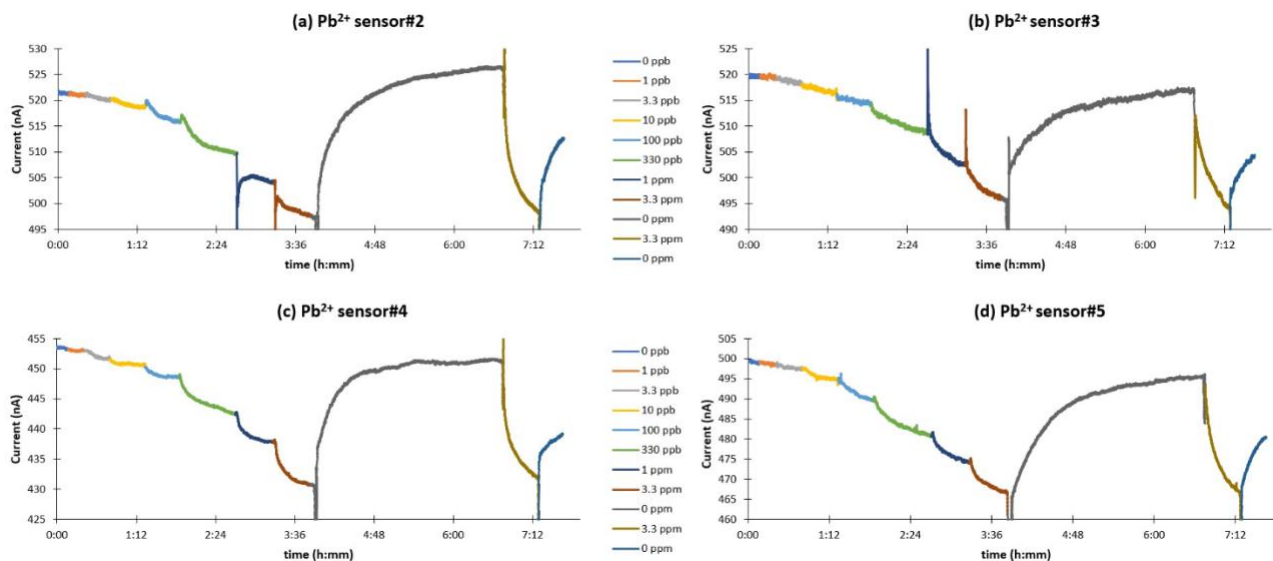
**Fig. S4** Sensor response (current changes) of a fabricated Pb<sup>2+</sup> sensor to various concentrations between 1 ppb to 3.3 ppm vs. time over three cycles. (a) sensor response in first cycle, (b) sensor response in second cycle, (c) sensor response in third cycle.

**Table S1** Performance of tested Pb<sup>2+</sup> sensors in three batches with different resistance ranges of CNT films

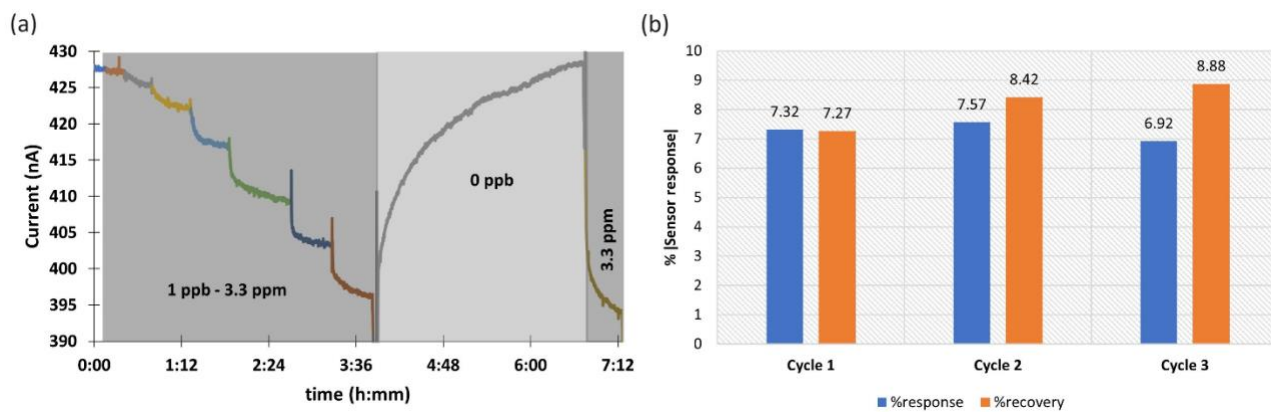
Sensor	Resistance of CNT film (kΩ)	% Sensor response	Lowest detectable concentration (ppb)	Response time at first detectable concentration (s)
<b>Batch No. 1</b>				
Sensor#1	9.2	7.6	3.3	8.6
Sensor#2	9.6	7.5	3.3	8.1
Sensor#3	8.9	7.8	3.3	7.9
<b>Batch No. 2</b>				
Sensor#9	5.5	6.8	10	10.2
Sensor#10	6.1	7.2	10	11.5
Sensor#11	7.8	7.7	3.3	10.9
<b>Batch No. 3</b>				
Sensor#13	12.3	4.3	10	7.5
Sensor#14	12.8	3.8	33	7.2
Sensor#15	13.2	2.9	33	7.8



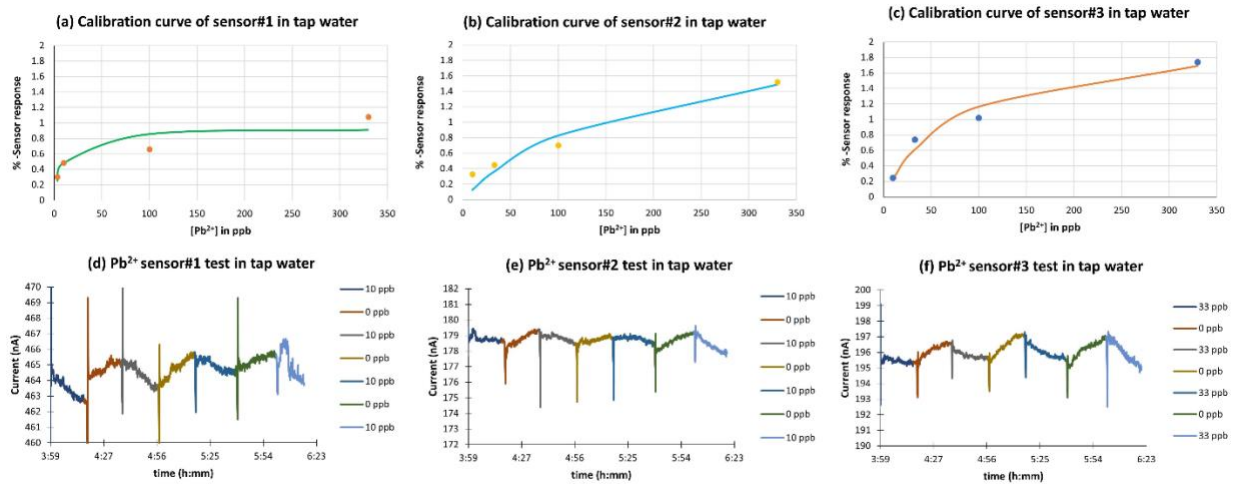
**Fig. S5** SEM images of air-brushed CNT films of three different batches in two bar scales of 100  $\mu\text{m}$  and 10  $\mu\text{m}$ . (a & b) SEM images of a representative CNT film of batch No. 1 (resistance 9.2  $\text{k}\Omega$ ), (c & d) batch No. 2 (resistance 5.5  $\text{k}\Omega$ ), (e & f) batch No. 3 (resistance 12.3  $\text{k}\Omega$ ).



**Fig. S6** Responses of several fabricated  $Pb^{2+}$  sensors and their recovery over time in a  $Pb^{2+}$  concentration range of 1 ppb to 3.3 ppm. (a) performance of  $Pb^{2+}$  sensor#2, (b)  $Pb^{2+}$  sensor#3, (c)  $Pb^{2+}$  sensor#4, and (d)  $Pb^{2+}$  sensor#5.



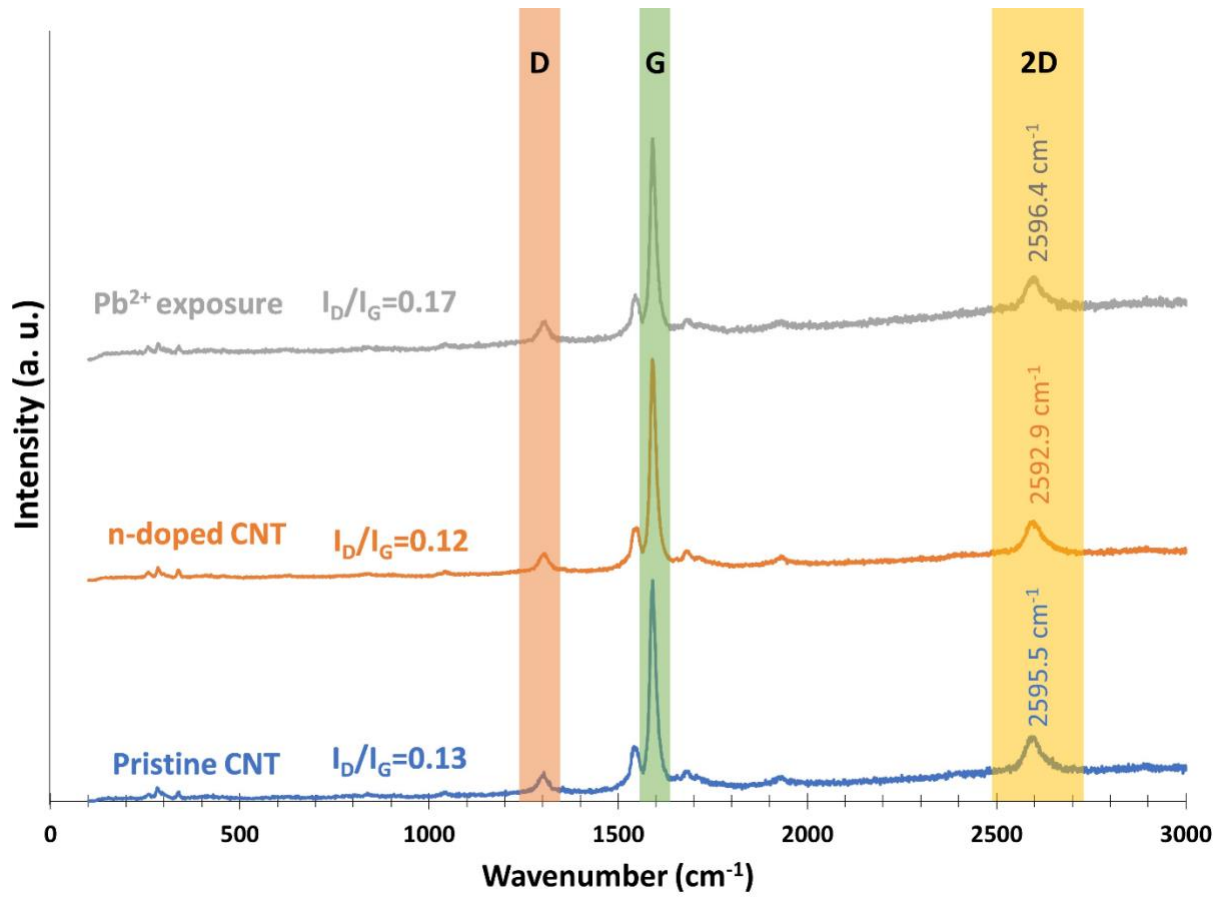
**Fig. S7** (a) Lead sensor's response and recovery between 0 ppb and 3300 ppb of  $Pb^{2+}$  ions, (b) absolute values of response and recovery percentages of a fabricated lead sensor between 0 ppb and 3300 ppb of  $Pb^{2+}$  ions over three cycles



**Fig. S8** Calibration curves of  $Pb^{2+}$  sensors prepared for real sample test and the response of the sensor to 10 ppb and 30 ppb of  $Pb^{2+}$  ion. (a) Calibration curve of sensor#1, (b) sensor#2 and (c) sensor#3 in tap water in a range of 10 ppb to 330 ppb of  $Pb^{2+}$  ions, (d) response of the sensor#1 to 10 ppb of  $Pb^{2+}$ , (e) response of the sensor#2 to 10 ppb of  $Pb^{2+}$ , and (f) response of the sensor#3 to 33 ppb of  $Pb^{2+}$  over three cycles ( $n=3$ ) in tap water.

**Table S2** Hall measurement data of the four-probe device on Si/SiO<sub>2</sub> substrate with sputter coated Au contacts after different fabrication steps, exposure to and removal of  $Pb^{2+}$  ions.

<i>Sample</i>	<i>Sheet resistivity (kΩ/sq)</i>	<i>Sheet Hall resistance coef. (m<sup>2</sup>/C)</i>	<i>Hall mobility (cm<sup>2</sup>/V.s)</i>	<i>Sheet concentration (Carrier density) (cm<sup>-2</sup>)</i>
Pristine SWCNT	8.304	2.46	70.7	$2.5 \times 10^{14}$
f-SWCNT	21.5	-43.2	20.1	$-1.4 \times 10^{13}$
f-SWCNT coated with ISM	24.04	579	241	$1.078 \times 10^{12}$
Exposed to 3 ppm of $Pb^{2+}$ for 24 hours	13.76	-14	10.2	$-4.461 \times 10^{13}$
Washing off $Pb^{2+}$ ions in DI water for 5 hours	9.806	16.9	211	$3.69 \times 10^{13}$



**Fig. S9** Raman spectra of pristine CNT, n-doped CNT and n-doped CNT exposed to 3 ppm of Pb<sup>2+</sup> ions for 5 hours.