

Supplementary Information

Amplifying the Electrochemical Footprint of < 1000 Molecules in a Dissolving Microdroplet

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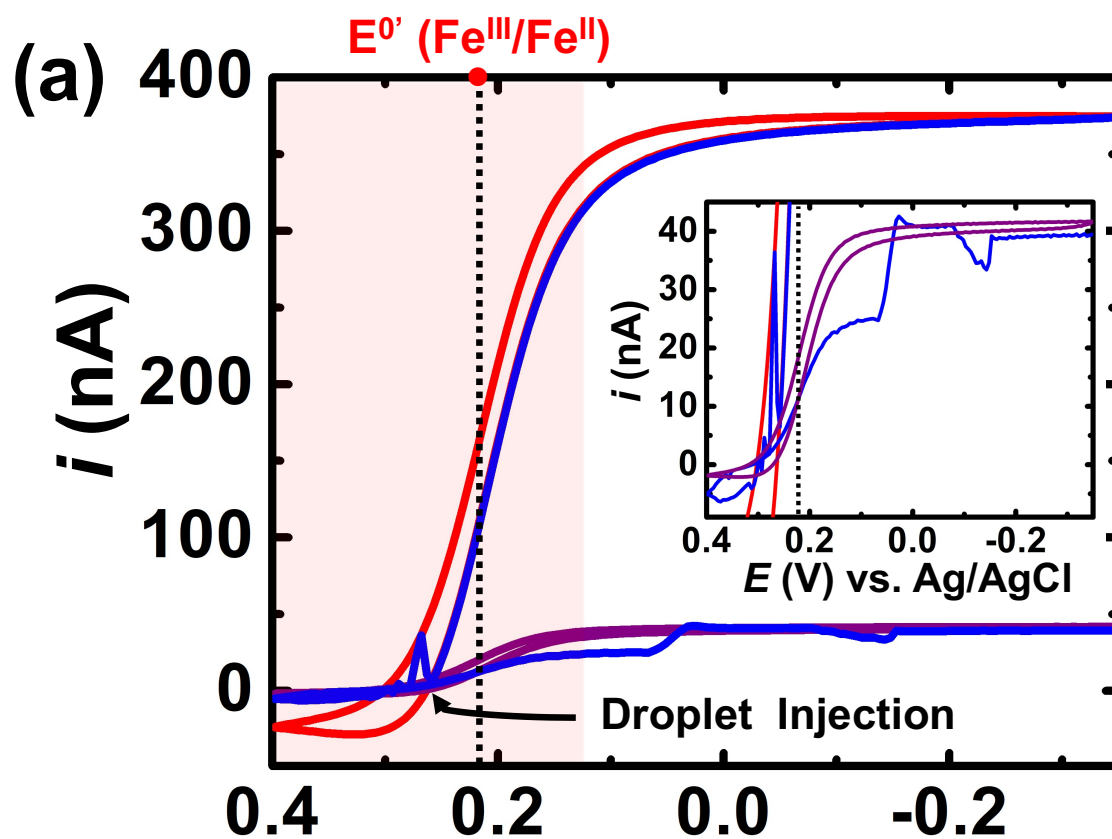


Figure S1. (a) Cyclic voltammograms recorded during the dissolution of a DCE droplet containing 0.5 mM $(\text{Cp}^*)_2\text{Fe}^{\text{III}}$ in an aqueous bulk phase of 200 mM $\text{K}_3[\text{Fe}(\text{CN})_6]$ in 10 mM NaClO_4 . The dashed lines represent the standard apparent potential for the redox couple $\text{Fe}(\text{CN})_6^{3-} / \text{Fe}(\text{CN})_6^{4-}$. Inset (i) shows a close up of the purple voltammogram showing suppressed redox activity of $\text{Fe}(\text{CN})_6^{3-} / \text{Fe}(\text{CN})_6^{4-}$ and absence of any signal from $\text{Cp}_2^*(\text{Fe})^{\text{II}}$.