Supporting Information

Selectively aggregating natural ligands into efficient AIEgens by human telomeric duplex-G-quadruplex junction

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Table S1.	Sequences	used in	1 this	work
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Entry	Sequence (from 5'to 3')
Q	GGGTTAGGGTTAGGGG
AQ	AGGGTTAGGGTTAGGGG
AQTT	AGGGTTAGGGTTAGGGTTAGGGTT
TAQ	TAGGGTTAGGGTTAGGGG
TAQT	TAGGGTTAGGGTTAGGGT
TAQTT	TAGGGTTAGGGTTAGGGTTAGGGTT
TTQA	TTGGGTTAGGGTTAGGGA
TTAQ	TTAGGGTTAGGGTTAGGG
TTAQTT	TTAGGGTTAGGGTTAGGGTTAGGGTT
TTAQTTA	TTAGGGTTAGGGTTAGGGTTA
TAQ-3iG	5' -TAGGGTTAGGGTTAGGGTTAGG-3'-3'-G-5'
TAQ-5iG	5'-TA-3'-3'-G-5'-GGTTAGGGTTAGGGTTAGGG-3'
TAQ-5iTAG	3'-TAG-5'-5'-GGTTAGGGTTAGGGTTAGGG-3'
mPu22	TGAGGGTGGGGGGGGGGGGAA
ckit2	CGGGCGGGCGCGAGGGAGGGT
PS2.M	GTGGGTAGGGCGGGTTGG
1XAV	TGAGGGTGGGTAGGGTGGGTAA
PW17	GGGTAGGGCGGGTTGGG
T3TT	GGGTTTGGGTGGGGGG
T2T2T2	GGGGTTGGGGTTGGGGG
NHEIII	TGGGGAGGGTGGGGAGGGTGGGGAAGG
TBA	GGTTGGTGTGGTTGG
D	CCCTAACCCTTAGGGTTAGGG
D-TTAQTT	CCCTAACCCTTAGGGTTAGGGTTAGGGTTAGGGTTAGGGTTAGGGTT
D- <u>T</u> TAQTT	ACCCTAACCCTTAGGGTTAGGGTTAGGGTTAGGGTTAGGGTTAGGGTT
D- <u>TT</u> AQTT	AACCCTAACCCTTAGGGTTAGGGTTAGGGTTAGGGTTAGGGTTAGGGTT
D- <u>TTA</u> QTT	TAACCCTAACCCTTAGGGTTAGGGTTAGGGTTAGGGTTAGGGTTAGGGTT
(TTAQ) ₂ TT	TTAGGGTTAGGGTTAGGGTTAGGGTTAGGGTTAGGGTTAGGGTTAGGGTT



Fig. S1. Fluorescence excitation and emission spectra of THP in formamide.



Fig. S2. DLS measurements of THP aggregating at variant concentrations in water.



Fig. S3. Absorption spectra of THP in (A) DMF, (B) acetonitrile, (C) methanol, (D) ethanol, (E) glycol, and (F) isopropanol, respectively.



Fig. S4. Emission spectra (A, excitation at 355 nm) and (B) particle size measurements of THP (50 μ M) in variant solvents.



Fig. S5. (A) Fluorescence excitation and emission spectra of THP in bindng with TTAQTT in PBS by systematically changing the [THP]/([THP]+[TTAQTT]) ratio, but keeping their total concentration at 2 μ M. (B) Fluorescence excitation and emission spectra of THP (10 μ M) in water by increasing the CB7 concentration. (C) Fluorescence excitation and emission spectra of THP in bindng with CB7 in water by systematically changing the [THP]/([THP]+[CB7]) ratio, but keeping their total concentration at 5 μ M. (D) The corresponding fluorescence responses at 495 nm in (C) as a function of [THP]/([THP]+[CB7]. (E) The CB7 structure and the predicted binding mode of CB7 with THP.



Fig. S6. Normalized absorption spectra with gradual addition of CB7 to THP (35 $\mu M)$ in water.



Fig. S7. Fluorescence responses of THP at variant concentration in the presence of DNA (1 μ M) in 20 mM PBS (pH 7.0) containing 100 mM (A) Na⁺ and (B) Li⁺, respectively.



Fig. S8. Job's plot analysis of binding stoichiometry of THP with (A) D-TTAQTT, (B) D-<u>T</u>TAQTT, (C) D-<u>TTAQTT</u>, and (D) D-<u>TTAQTT</u> in 20 mM PBS (pH 7.0) containing 100 mM K^+ , respectively.



Fig. S9. Normalized absorption spectra of THP (30 μ M) with increasing concentration of (A) D-TTAQTT, (B) D-<u>T</u>TAQTT, (C) D-<u>TT</u>AQTT, and (D) D-<u>TTA</u>QTT, respectively.



Fig. S10. Difference absorption spectra according to Fig. S9.



Fig. S11. DLS analysis of THP (10 μ M) in interacting with (A) D-TTAQTT, (B) D-<u>T</u>TAQTT, (C) D-<u>TT</u>AQTT, and (D) D-<u>TTA</u>QTT, respectively.



Fig. S12. Effect of monovalent cation ions (K^+ , Na^+ and Li^+) on the fluorescence response of THP (50 μ M) in the absence and presence of D-TTAQTT, TTAQTT, and (TTAQ)₂TT (1 μ M in G4 unit).