

Metal-phenolic coordination frameworks nanozyme exhibits dual enzyme mimic activity and its application of effective colorimetric detection of biomolecules

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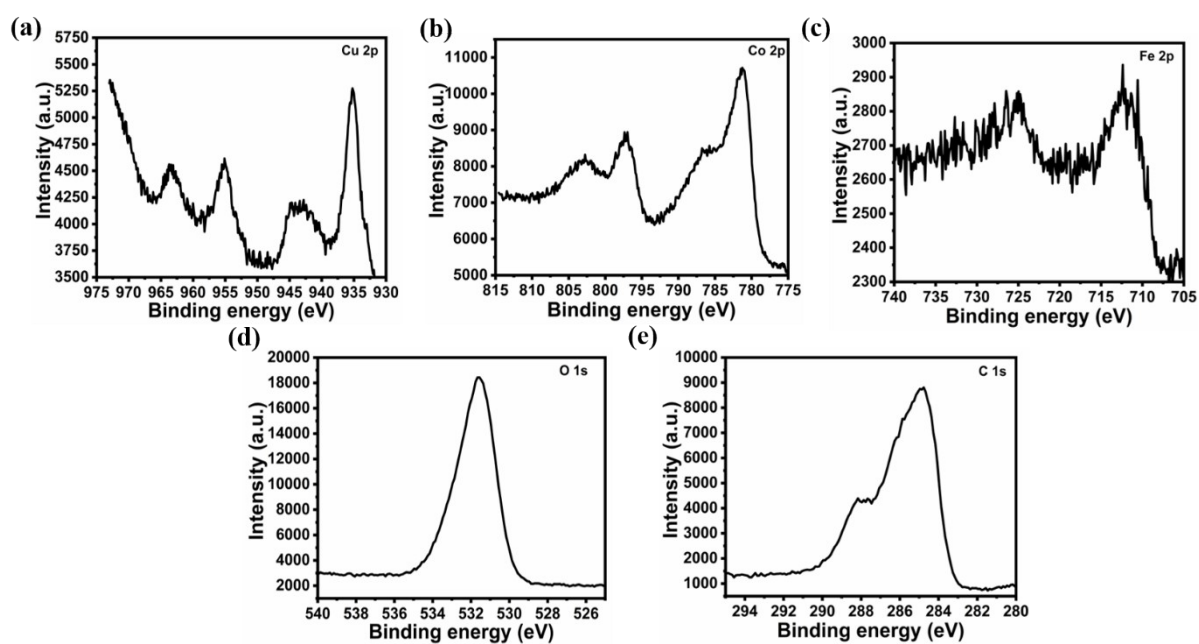


Figure S1 High-resolution XPS spectra of nanozymes showing Cu 2p, Co 2p, Fe 2p, O 1s and C 1s core energy levels.

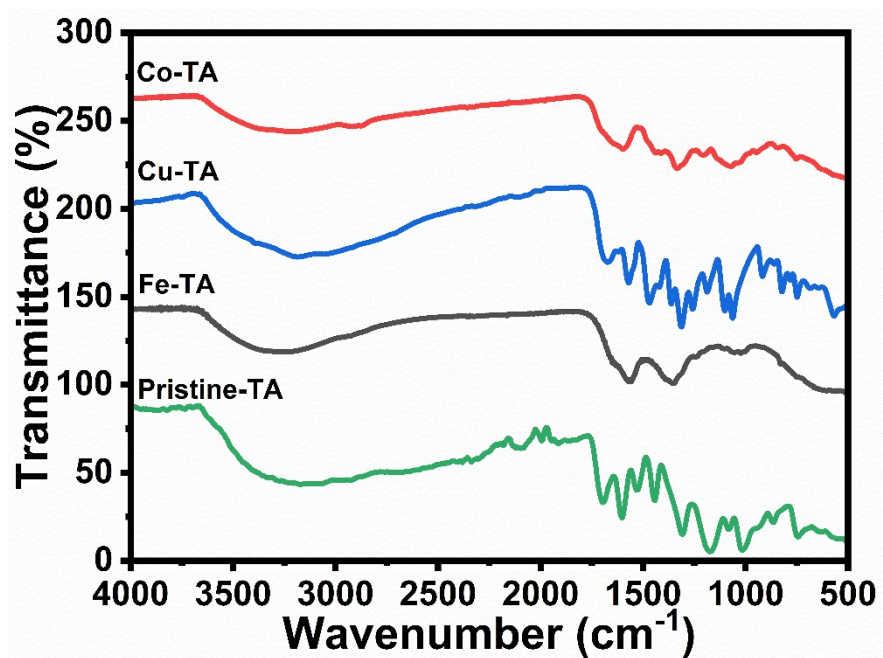


Figure S2 FT-IR spectra of Cu-TA, Co-TA, Fe-TA and Pristine-TA.

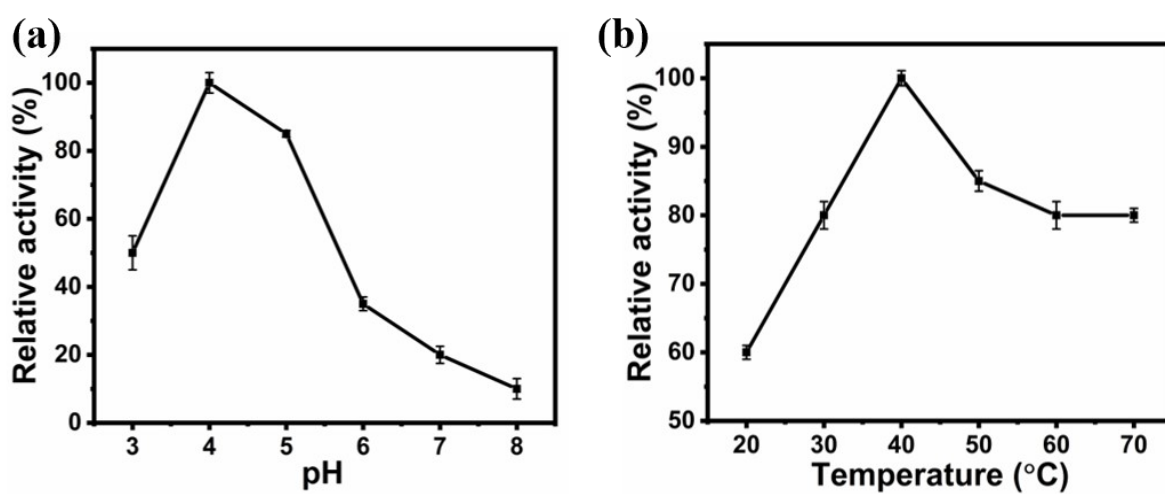


Figure S3 The effect of (a) pH and (b) temperature of the peroxidase and catalase activity of Fe-TA nanozyme.

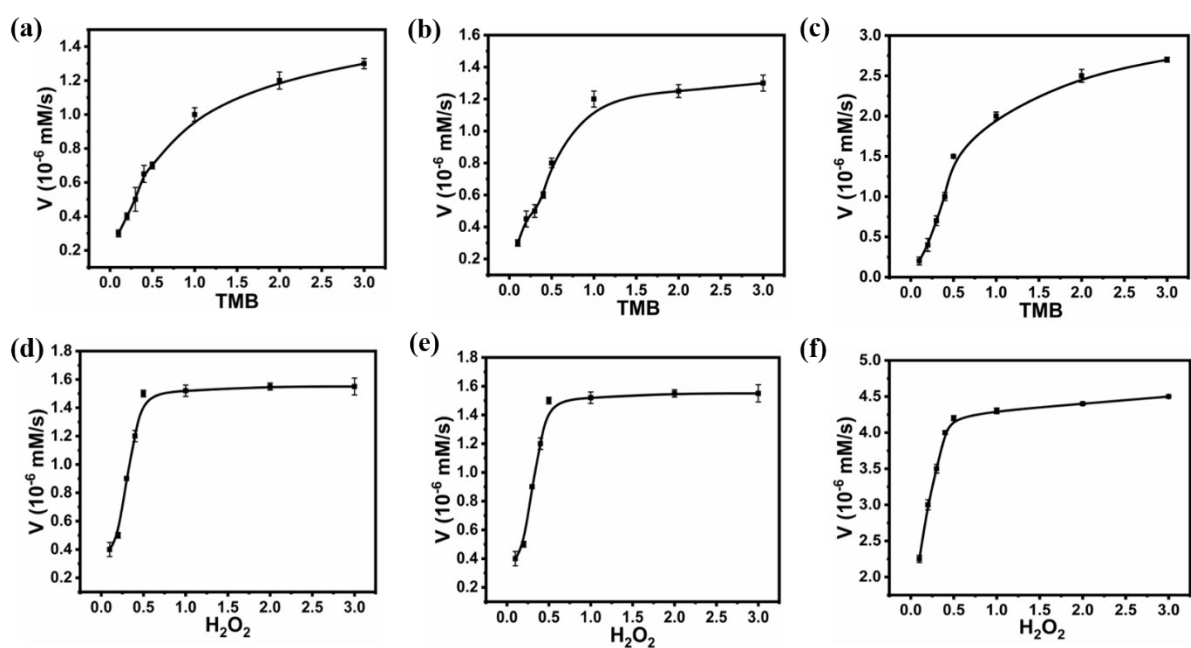


Figure S4 The Michaelis-Menten curves for different peroxidase-like nanozymes using H_2O_2 and TMB as substrates

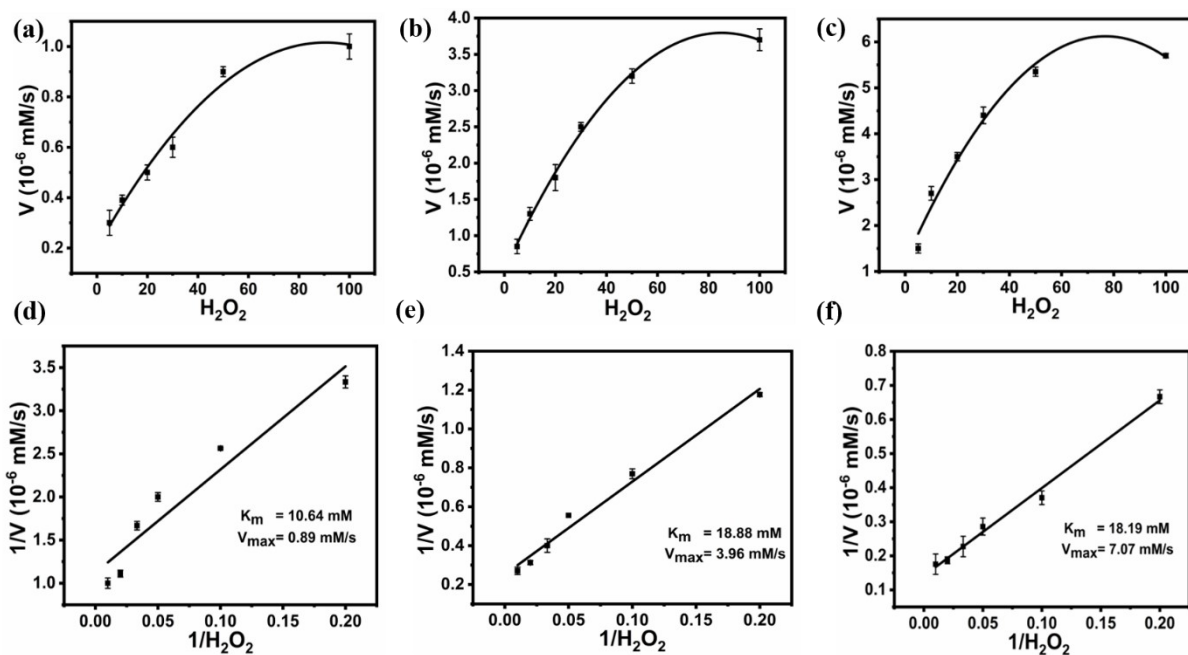


Figure S5 The Michaelis-Menten curves and double reciprocal plots for various catalase-like activities of nanozymes.

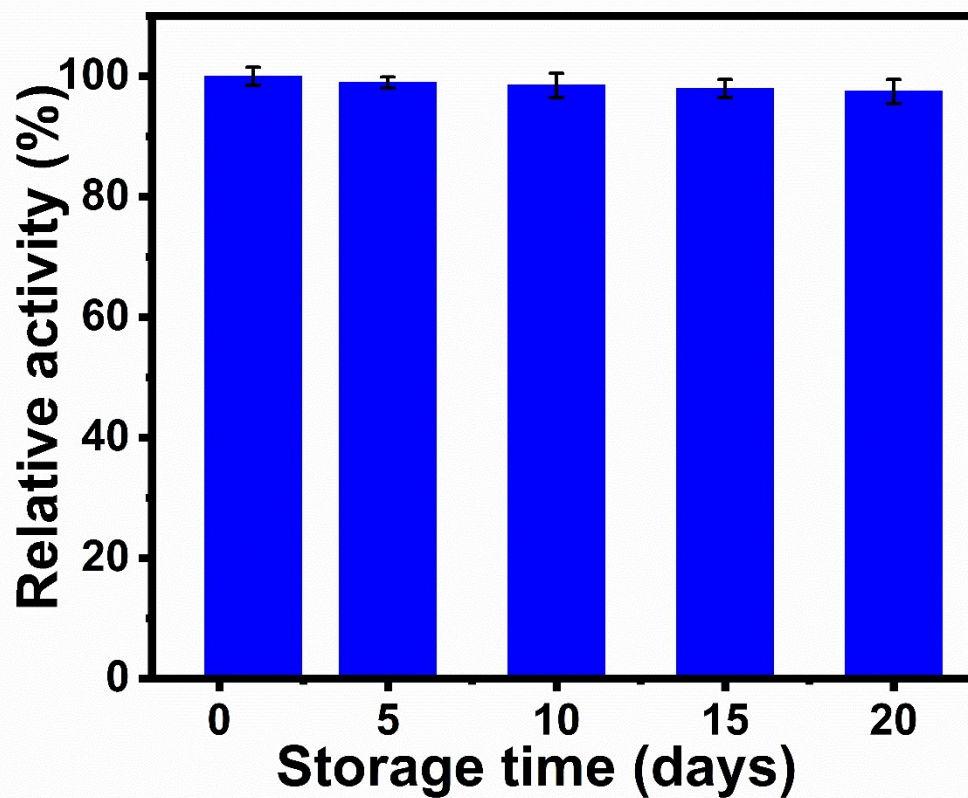


Figure S6 The remarkable stability of Fe-TA nanozyme after 21 days of storage.

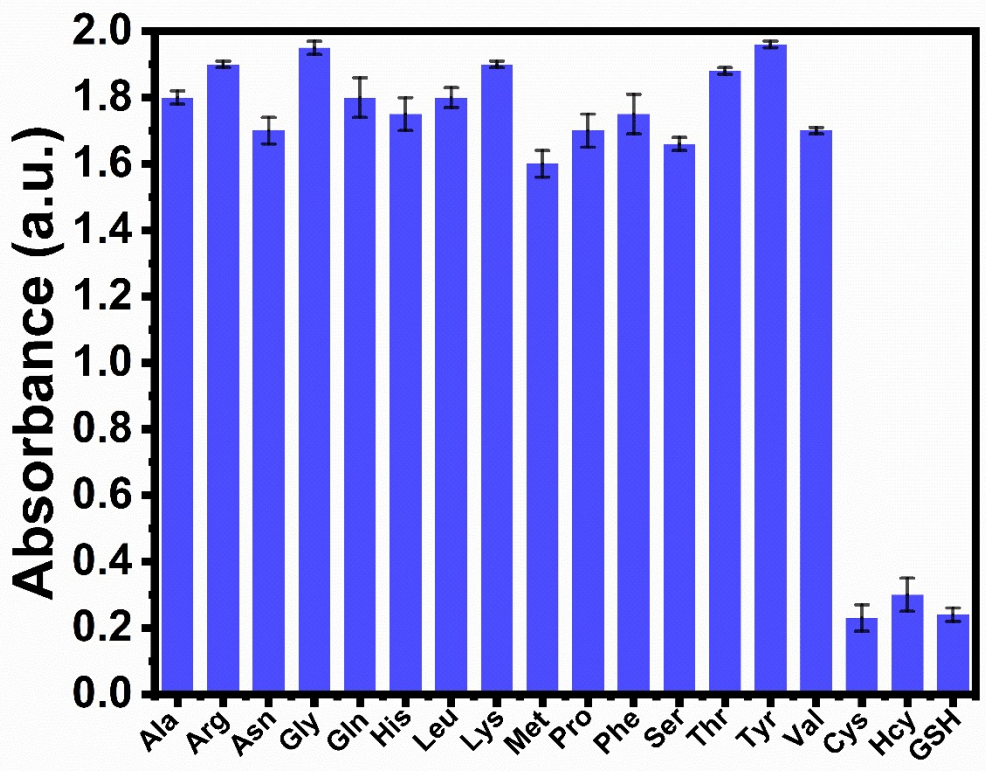


Figure S7 Selectivity performance of Fe-TA nanozyme in detecting Cys, Hcy and GSH in the presence of other possible interfering biomolecules.

Table S1 Comparison of the POD-like and CAT-like kinetic parameters of Cu-TA, Co-TA and Fe-TA nanozymes.

Enzyme mimic	MPNs	Substrate	K_m (mM)	V_{max} (Ms⁻¹)
POD	Cu-TA	H ₂ O ₂	0.261	2.04
		TMB	0.306	1.126
	Co-TA	H ₂ O ₂	0.497	2.39
		TMB	0.248	1.251
	Fe-TA	H ₂ O ₂	0.11	4.76
		TMB	0.348	1.651
CAT	Cu-TA	H ₂ O ₂	10.64	0.89
	Co-TA	H ₂ O ₂	18.88	3.96
	Fe-TA	H ₂ O ₂	18.19	7.07

Table S2 Comparison of the linear ranges and LOD of Cys, GSH, and Hcy with previous studies.

	Materials	Method	Linear ranges (μM)	LODs (μM)	References
Cys	CeO ₂ /CoO	Colorimetry	5-10	3.71	1
	CuMnO ₂	Colorimetry	20–300	11.26	2
	Pd–Fe ₃ O ₄ DBNPs	Colorimetry	0-250	3.1	3
	Pt@WO ₃ NSs	Colorimetry	0.01–15	1.2	4
	PdPt ₃ -LNT NDs	Colorimetry	0–200	3.099	5
	Fe-TA	Colorimetry	1-2	0.382	This study
Hcy	AgNPRs	Colorimetry	0-5	0.041	6
	CB-CQDs	Colorimetry	0.5–20	0.6	7
	MnO ₂ NFs-RhB	Colorimetry	0–30	0.087	8
	Fe ³⁺ - TMB	Colorimetry	2–24	2.09	9
	Fe ₃ O ₄ @CuO-GO	Colorimetry	5–200	1.8	10
	Fe-TA	Colorimetry	2-3	0.776	This study
GSH	Hemin/GQD	Colorimetry	1-50	200	11
	Pd150-PCRPs NPs	Colorimetry	2–300	1.65	12
	Fe ₃ O ₄ /CNDs	Colorimetry	0–20	0.058	13
	PCN-224-Mn	Colorimetry	0.5-60	0.233	14
	Mn ₃ O ₄ microspheres	Colorimetry	5–60	0.889	15
	Fe-TA	Colorimetry	2-9	0.750	This study

Table S3 The performance of Fe-TA colorimetric sensor in spiked real water samples. (n=3)

Targets	Samples	Spiked (μM)	Found (μM)	Recovery \pm RSD (%)
Cys	Tap water	5	5.12	102.4 \pm 2.24
	River water	10	10.13	101.3 \pm 3.84
	Lake water	15	14.9	99.3 \pm 4.02
Hcy	Tap water	5	4.97	99.4 \pm 1.27
	River water	10	9.9	99 \pm 5.12
	Lake water	15	14.95	99.7 \pm 6.1
GSH	Tap water	5	5.02	100.4 \pm 1.98
	River water	10	9.93	99.3 \pm 3.14
	Lake water	15	15.2	101 \pm 3.55

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