Supplementary Information

Biochemical investigations using mass spectrometry to monitor JMJD6-catalysed hydroxylation of multi-lysine containing bromodomain-derived substrates

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1. Abbreviations

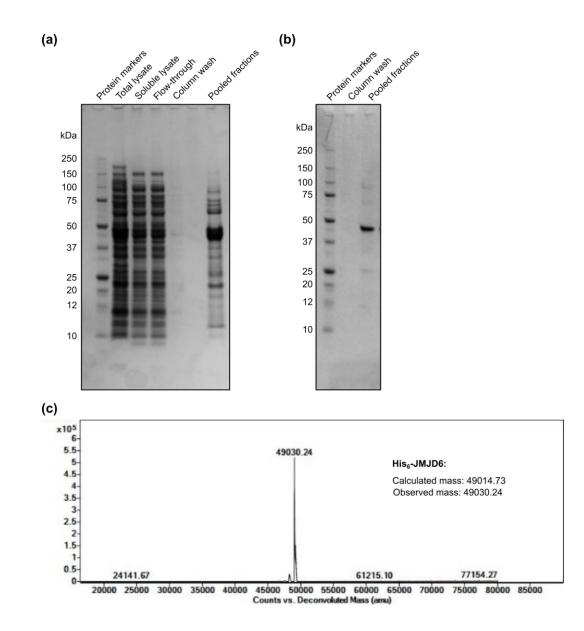
| 20G | 2-oxoglutarate |
|-------------------------|---|
| AR | androgen receptor |
| AR-V7 | androgen receptor slice variant 7 |
| AspH | aspartate/asparagine β-hydroxylase |
| BBOX | γ-butyobetaine hydroxylase |
| BID | basic residue enriched interaction domain |
| BRD | bromodomain-containing protein |
| CP4H | collagen prolyl 4-hydroxylase |
| CROP | cisplatin resistance-associated overexpressed protein |
| CTAD | C-terminal activation domain |
| DHX9 | DExH-box helicase 9 |
| DKC1 | dyskerin pseudouridine synthase 1 |
| DSBH | double stranded β-helix |
| EGFD | epidermal growth factor-like domain |
| ERα | estrogen receptor-α |
| FECH | ferrochelatase |
| FIH | factor inducible hypoxia inducible factor-a |
| FLT1 | vascular endothelial growth factor receptor 1 |
| GBB | γ-butyobetaine |
| H3/4 | histone 3/4 |
| HIF | hypoxia-inducible factor |
| HSP70 | heat shock protein 70 |
| JMJD6 | Jumonji-C domain-containing protein 6 |
| KDM | histone N^{ε} -methyl lysine demethylase |
| $K_{\rm m}^{ m app}$ | apparent Michaelis constant |
| $k_{\rm cat}^{\rm app}$ | turnover number |
| LAA | <i>L</i> -ascorbic acid |
| LC-MS | liquid chromatography-mass spectrometry |
| LUC7L2 | LUC7-like 2 |
| MALDI-MS | matrix-assisted laser desorption/ionization mass spectrometry |
| MINA53 | MYC-induced nuclear antigen |
| mRNA | messenger ribonucleic acid |

| NKAP | NF-KB-activating protein |
|-----------------------|---|
| NMR | nuclear magnetic resonance |
| NOG | <i>N</i> -oxalylglycine |
| NO66 | nucleolar protein 66 |
| PHYH | phytanoyl-CoA dioxygenase |
| RBM39 | RNA-binding protein 39 |
| RPL8 | ribosomal protein L8 |
| RPL27A | ribosomal protein L27A |
| RPS6 | ribosomal protein S6 |
| O ₂ | dioxygen |
| PDB | protein data bank |
| PHD | prolyl hydroxylase domain-containing protein |
| pVHL | von Hippel-Lindau protein |
| SRSF11 | serine and arginine rich splicing factor 11 |
| SPE-MS | solid-phase extraction coupled to mass spectrometry |
| TET | ten-eleven translocation |
| TRAF6 | tumor necrosis factor receptor associated factor 6 |
| U2AF65 | splicing factor U2 auxiliary factor 65 kDa subunit |
| U2SURP | U2 SnRNP associated SURP domain containing |
| USP42 | ubiquitin-specific peptidase 42 |
| v_{\max}^{app} | apparent maximum velocities |

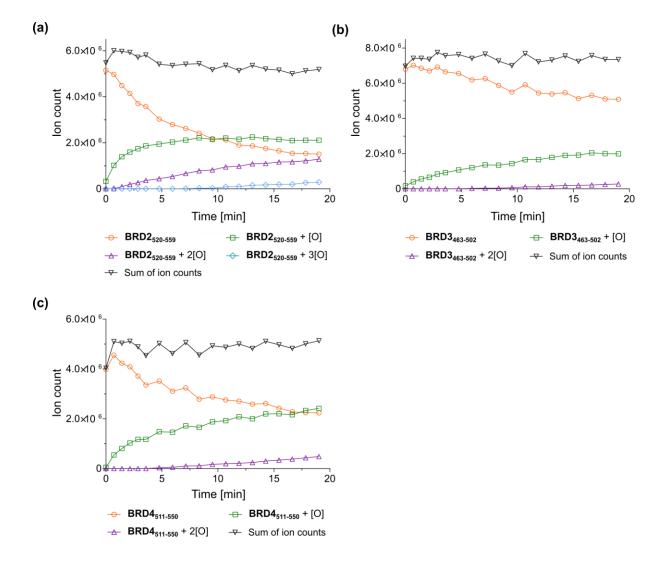
2. Supplementary figures

Supplementary Figure S1. Recombinant human JMJD6. Recombinant human JMJD6 (fulllength with an N-terminal His₆-tag) was >95% pure as analysed by sodium dodecyl sulphate polyacrylamide gel electrophoresis (SDS-PAGE) with Coomassie staining and electrospray ionization mass spectrometry (ESI-MS).

(a) 10-12% SDS-PAGE analysis of purified His₆-JMJD6 after Ni(II)-affinity chromatography (HisTrap column); (b) 10-12% SDS-PAGE analysis of His₆-JMJD6 after subsequent size-exclusion chromatography (Superdex 75); (c) Deconvoluted mass spectrum of purified His₆-JMJD6 shows potential evidence for JMJD6 self-hydroxylation, as reported;¹ calculated mass of His₆-JMJD6 = 49014.73 Da, observed mass: 49030.24 Da (*i.e.*, mass difference: ~ +16 Da).

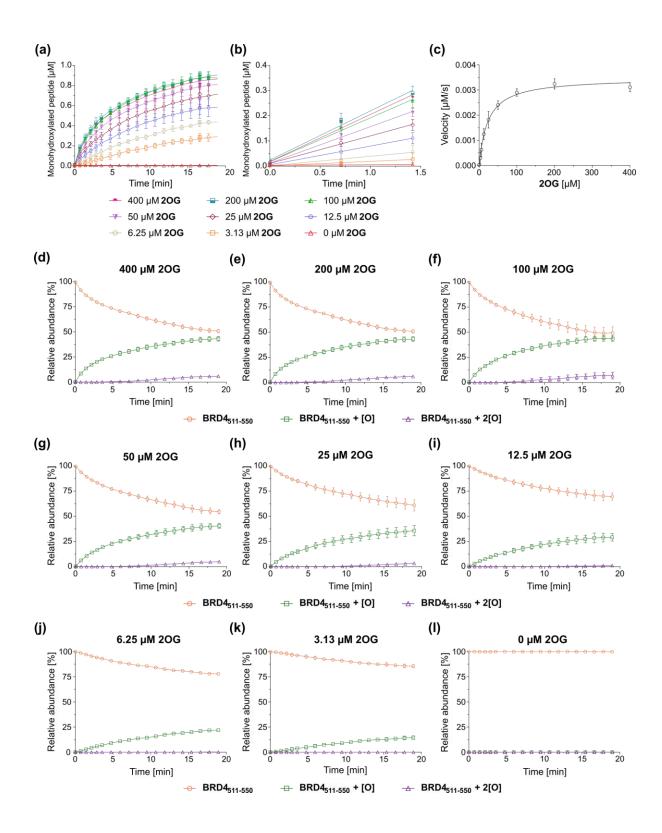


Supplementary Figure S2. Ion counts of BRD-derived peptides measured during JMJD6catalysed hydroxylation. The sum of the MS ion counts of the substrate peptide and all hydroxylated product peptides (black line) remains approximately constant throughout the JMJD6-catalyzed reactions of: (a) BRD2₅₂₀₋₅₅₉, (b) BRD3₄₆₃₋₅₀₂ and (c) BRD4₅₁₁₋₅₅₀. The results indicate that the SPE-MS assays are sufficiently robust to enable quantification of the extent of the JMJD6-catalysed BRD2₅₂₀₋₅₅₉, BRD3₄₆₃₋₅₀₂ and BRD4₅₁₁₋₅₅₀ hydroxylation. Hydroxylation reactions were performed as described in the Experimental Procedures section using His₆-JMJD6 (0.05 μ M), 2OG (200 μ M), (NH₄)₂Fe(SO₄)₂·6H₂O (FAS; 2 μ M), BRD substrate (2 μ M) and *L*-ascorbic acid (LAA; 100 μ M) in Tris buffer (50 mM, pH 7.5).



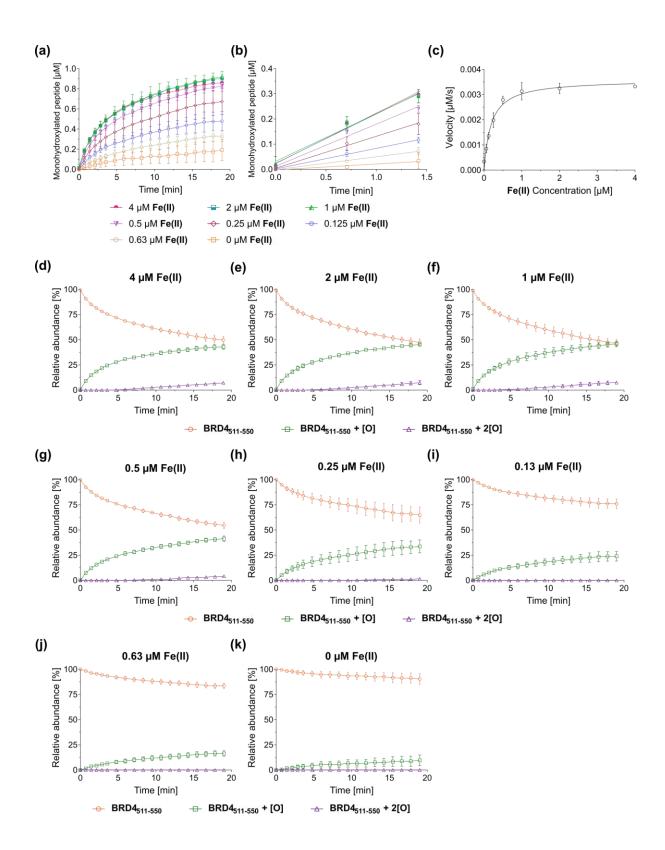
Supplementary Figure S3. Determination of the JMJD6 kinetic parameters for 2OG (continues on the following page). SPE-MS was used to determine the extent of JMJD6catalysed mono- and di-hydroxylation of the BRD4₅₁₁₋₅₅₀ fragment peptide for the specified 2OG concentrations. Initial rates of the JMJD6-catalysed mono-hydroxylation of BRD4₅₁₁₋₅₅₀ were used to determine the maximum velocity (v_{max}^{app}) and Michaelis constant (K_m^{app}) of JMJD6 for 2OG. Measurement times were normalized to the first sample injection analysed after the addition of the JMJD6 enzyme mixture to the substrate mixture (containing 2OG, FAS, LAA, and BRD4₅₁₁₋₅₅₀), by which time low levels of BRD4₅₁₁₋₅₅₀ mono-hydroxylation were manifest. Data are the mean of three independent runs (n = 3; mean ± standard deviation, SD). Conditions: His₆-JMJD6 (50 nM), 2OG (as specified), FAS (2 μ M), BRD4₅₁₁₋₅₅₀ (2 μ M) and LAA (100 μ M) in Tris buffer (50 mM, pH 7.5).

(a) Abundance of the mono-hydroxylated BRD4₅₁₁₋₅₅₀ peptide, following the addition of 50 nM JMJD6 to the substrate mixture (t = 0 min), for the specified 2OG concentrations; (b) initial reaction velocities used for the kinetic analysis of the JMJD6-catalysed mono-hydroxylation of BRD4₅₁₁₋₅₅₀; (c) Michaelis-Menten curve used to determine the kinetic parameters of JMJD6 for 2OG. The JMJD6 v_{max}^{app} and K_m^{app} values of JMJD6 for 2OG are $3.5 \pm 0.1 \cdot 10^{-3} \,\mu\text{M} \cdot \text{s}^{-1}$ and $23.3 \pm 2.5 \,\mu\text{M}$, respectively, as determined by non-linear regression; (d-l) time course of the JMJD6-catalysed hydroxylation of BRD4₅₁₁₋₅₅₀ showing the relative abundance of the BRD4₅₁₁₋₅₅₀ substrate (BRD4₅₁₁₋₅₅₀; orange circles), mono-hydroxylated BRD4₅₁₁₋₅₅₀ (BRD4₅₁₁₋₅₅₀ (BRD4₅₁₁₋₅₅₀ (BRD4₅₁₁₋₅₅₀ (BRD4₅₁₁₋₅₅₀); orange circles), mono-hydroxylated BRD4₅₁₁₋₅₅₀ + 2[O]; purple triangles), in the presence of varied 2OG concentrations: (d) 400 μ M; (e) 200 μ M; (f) 100 μ M; (g) 50 μ M; (h) 25 μ M; (j) 12.5 μ M; (j) 6.25 μ M; (k) 3.13 μ M; (l) 0 μ M. No evidence for triple (or higher order) BRD4₅₁₁₋₅₅₀ hydroxylation was observed under the employed reaction conditions.



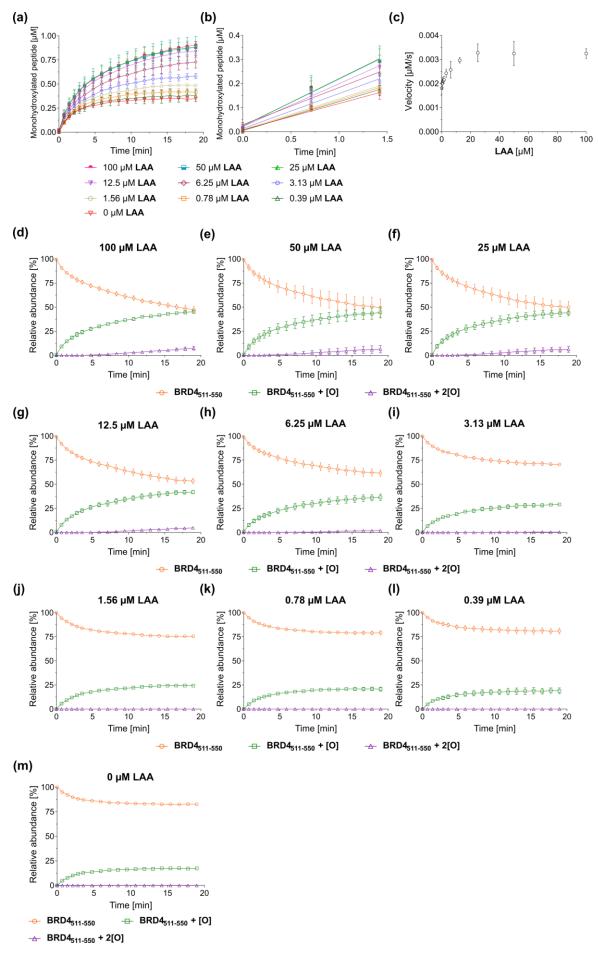
Supplementary Figure S4. Determination of the JMJD6 kinetic parameters for Fe(II) (continues on the following page). SPE-MS was used to determine the extent of JMJD6-catalysed mono- and di-hydroxylation of the BRD4₅₁₁₋₅₅₀ fragment peptide for the specified Fe(II) concentrations. Initial rates of the JMJD6-catalysed mono-hydroxylation of BRD4₅₁₁₋₅₅₀ were used to determine the maximum velocity (v_{max}^{app}) and apparent Michaelis constant (K_m^{app}) of JMJD6 for Fe(II). Measurement times were normalized to the first sample injection analysed after the addition of the JMJD6 enzyme mixture into the substrate mixture, by which time low levels of BRD4₅₁₁₋₅₅₀ mono-hydroxylation were manifest. Data are the mean of three independent runs (n = 3; mean ± SD). Conditions: His₆-JMJD6 (50 nM), 2OG (200 μ M), FAS (as specified), BRD4₅₁₁₋₅₅₀ (2 μ M) and LAA (100 μ M) in Tris buffer (50 mM, pH 7.5).

(a) Abundance of the mono-hydroxylated BRD4₅₁₁₋₅₅₀ peptide, following the addition of 50 nM JMJD6 to the substrate mixture (t = 0 min), for the specified Fe(II) concentrations; (b) initial reaction velocities used for the kinetic analysis of the JMJD6-catalysed mono-hydroxylation of BRD4₅₁₁₋₅₅₀; (c) Michaelis-Menten curve used to determine the kinetic parameters of JMJD6 for Fe(II). The JMJD6 v_{max}^{app} and K_m^{app} values of JMJD6 for Fe(II) are $3.6 \pm 0.1 \cdot 10^{-3} \,\mu\text{M} \cdot \text{s}^{-1}$ and $0.19 \pm 0.02 \,\mu\text{M}$, respectively, as determined by non-linear regression; (d-k) time course of the JMJD6-catalysed hydroxylation of BRD4₅₁₁₋₅₅₀ showing the relative abundance of the BRD4₅₁₁₋₅₅₀ substrate (BRD4₅₁₁₋₅₅₀; orange circles), mono-hydroxylated BRD4₅₁₁₋₅₅₀ (BRD4₅₁₁₋₅₅₀ (BRD4₅₁₁₋₅₅₀ (BRD4₅₁₁₋₅₅₀); purple triangles), in the presence of varied Fe(II) concentrations: (d) 4 μ M; (e) 2 μ M; (f) 1 μ M; (g) 0.5 μ M; (h) 0.25 μ M; (i) 0.125 μ M; (j) 0.063 μ M; (k) 0 μ M. No evidence for triple (or higher order) BRD4₅₁₁₋₅₅₀ hydroxylation was observed under the employed reaction conditions.



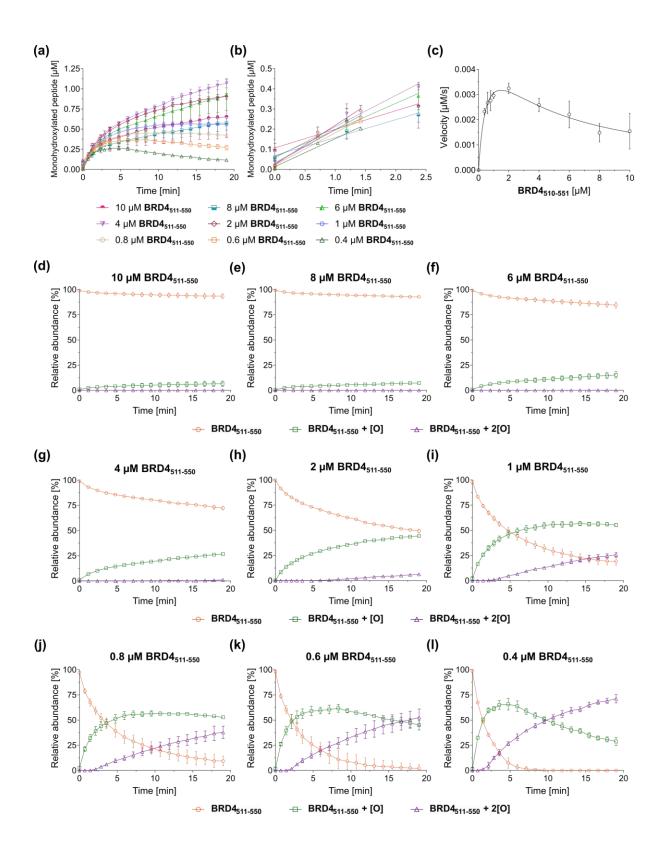
Supplementary Figure S5. Effect of LAA on the JMJD6-catalysed hydroxylation of BRD4₅₁₁₋₅₅₀ (continues on the following page). SPE-MS was used to determine the extent of JMJD6-catalysed mono- and di-hydroxylation of the BRD4₅₁₁₋₅₅₀ fragment peptide for the specified concentrations of L-ascorbic acid (LAA). Measurement times were normalized to the first sample injection analysed after the addition of the JMJD6 enzyme mixture to the substrate mixture, by which time low levels of BRD4₅₁₁₋₅₅₀ mono-hydroxylation were manifest. Data are the mean of three independent runs (n = 3; mean \pm SD). Conditions: His₆-JMJD6 (50 nM), 2OG (200 μ M), FAS (2 μ M), BRD4₅₁₁₋₅₅₀ (2 μ M) and LAA (as specified) in Tris buffer (50 mM, pH 7.5).

(a) Abundance of the mono-hydroxylated BRD4₅₁₁₋₅₅₀ peptide, following the addition of 50 nM JMJD6 to the substrate mixture (t = 0 min), for the specified LAA concentrations; (b) initial reaction velocities used for the kinetic analysis of the JMJD6-catalysed mono-hydroxylation of BRD4₅₁₁₋₅₅₀; (c) effect of LAA concentration on the initial reaction velocity of the JMJD6-catalysed mono-hydroxylation of BRD4₅₁₁₋₅₅₀; (d-k) time course of the JMJD6-catalysed hydroxylation of BRD4₅₁₁₋₅₅₀ showing the relative abundance of the BRD4₅₁₁₋₅₅₀ substrate (BRD4₅₁₁₋₅₅₀; orange circles), mono-hydroxylated BRD4₅₁₁₋₅₅₀ (BRD4₅₁₁₋₅₅₀ (BRD4₅₁₁₋₅₅₀ + [O]; green squares) and di-hydroxylated BRD4₅₁₁₋₅₅₀ (BRD4₅₁₁₋₅₅₀ + 2[O]; purple triangles), in the presence of varied LAA concentrations: (d) 100 μ M; (e) 50 μ M; (f) 25 μ M; (g) 12.5 μ M; (h) 6.25 μ M; (i) 3.13 μ M; (j) 1.56 μ M; (k) 0.78 μ M; (l) 0.39 μ M; (m) 0 μ M. No evidence for triple (or higher order) BRD4₅₁₁₋₅₅₀ hydroxylation was observed under the employed reaction conditions.



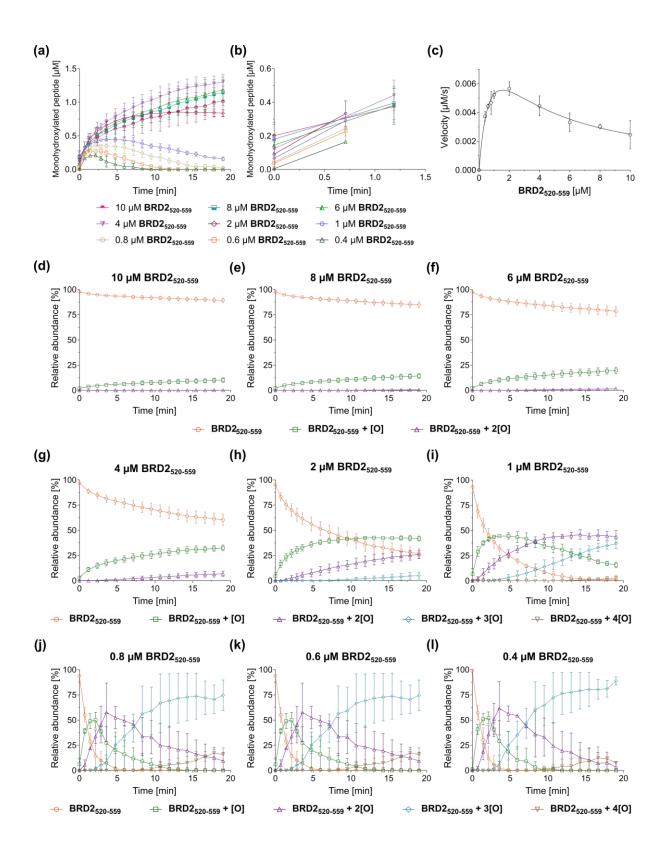
Supplementary Figure S6. Determination of the JMJD6 kinetic parameters for BRD4₅₁₁-550 (continues on the following page). SPE-MS was used to determine the extent of JMJD6catalysed mono- and di-hydroxylation of the BRD4₅₁₁₋₅₅₀ fragment peptide for the specified BRD4₅₁₁₋₅₅₀ concentrations. Initial rates of the JMJD6-catalysed mono-hydroxylation of BRD4₅₁₁₋₅₅₀ were used to determine the maximum velocity (v_{max}^{app}) and Michaelis constant (K_m^{app}) of JMJD6 for BRD4₅₁₁₋₅₅₀. Measurement times were normalized to the first sample injection analysed after the addition of the JMJD6 enzyme mixture to the substrate mixture, by which time low levels of BRD4₅₁₁₋₅₅₀ mono-hydroxylation were manifest. Data are the mean of three independent runs (n = 3; mean ± SD). Conditions: His₆-JMJD6 (50 nM), 2OG (200 μ M), FAS (2 μ M), BRD4₅₁₁₋₅₅₀ (as specified) and LAA (100 μ M) in Tris buffer (50 mM, pH 7.5).

(a) Abundance of the mono-hydroxylated BRD4₅₁₁₋₅₅₀ peptide, following the addition of 50 nM JMJD6 to the substrate mixture (t = 0 min), for the specified BRD4₅₁₁₋₅₅₀ concentrations; (b) initial reaction velocities used for the kinetic analysis of the JMJD6-catalysed mono-hydroxylation of BRD4₅₁₁₋₅₅₀; (c) Michaelis-Menten curve used to determine the kinetic parameters of JMJD6 for BRD4₅₁₁₋₅₅₀. The JMJD6 v_{max}^{app} and K_m^{app} values of JMJD6 for BRD4₅₁₁₋₅₅₀ are $5.8 \pm 1.2 \cdot 10^{-3} \,\mu$ M·s⁻¹ and $0.62 \pm 0.26 \,\mu$ M, respectively, as determined by non-linear regression; (d-l) time course of the JMJD6-catalysed hydroxylation of BRD4₅₁₁₋₅₅₀ showing the relative abundance of the BRD4₅₁₁₋₅₅₀ substrate (BRD4₅₁₁₋₅₅₀; orange circles), mono-hydroxylated BRD4₅₁₁₋₅₅₀ (BRD4₅₁₁₋₅₅₀ (BRD4₅₁₁₋₅₅₀ + [O]; green squares) and di-hydroxylated BRD4₅₁₁₋₅₅₀ concentrations: (d) 10 μ M; (e) 8 μ M; (f) 6 μ M; (g) 4 μ M; (h) 2 μ M; (i) 1 μ M; (j) 0.8 μ M; (k) 0.6 μ M; (l) 0.4 μ M. No evidence for triple (or higher order) BRD4₅₁₁₋₅₅₀ hydroxylation was observed under the employed reaction conditions.



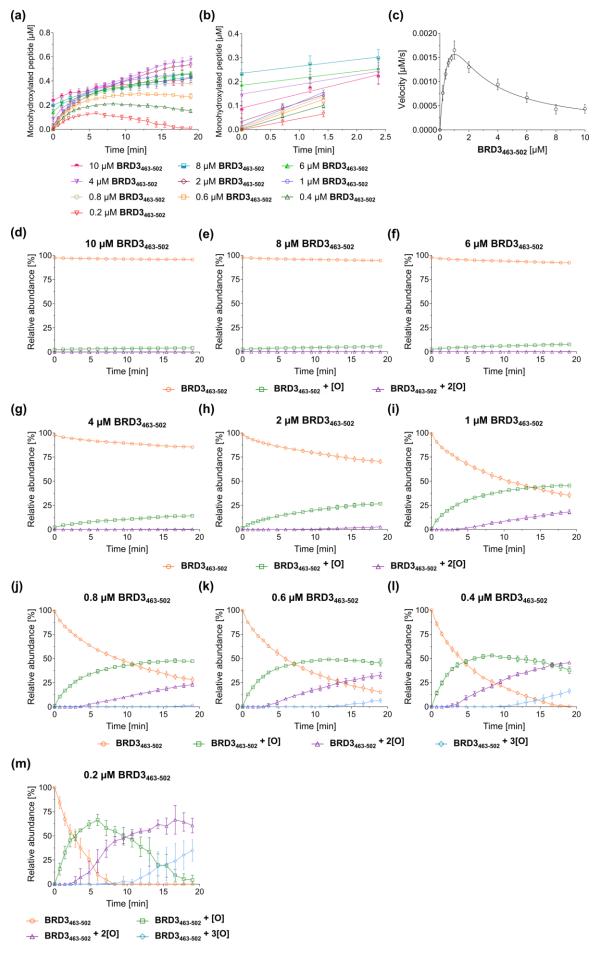
Supplementary Figure S7. Determination of the JMJD6 kinetic parameters for BRD2₅₂₀-559 (continues on the following page). SPE-MS was used to determine the extent of JMJD6catalysed mono- and di-hydroxylation of the BRD2₅₂₀₋₅₅₉ fragment peptide for the specified BRD2₅₂₀₋₅₅₉ concentrations. Initial rates of the JMJD6-catalysed mono-hydroxylation of BRD2₅₂₀₋₅₅₉ were used to determine the maximum velocity (v_{max}^{app}) and Michaelis constant (K_m^{app}) of JMJD6 for BRD2₅₂₀₋₅₅₉. Measurement times were normalized to the first sample injection analysed after addition of the JMJD6 enzyme mixture to the substrate mixture, by which time low levels of BRD2₅₂₀₋₅₅₉ mono-hydroxylation were manifest. Data are the mean of three independent runs (n = 3; mean ± SD). Conditions: His₆-JMJD6 (50 nM), 2OG (200 μ M), FAS (2 μ M), BRD2₅₂₀₋₅₅₉ (as specified) and LAA (100 μ M) in Tris buffer (50 mM, pH 7.5).

(a) Abundance of the mono-hydroxylated BRD2₅₂₀₋₅₅₉ peptide, following the addition of 50 nM JMJD6 to the substrate mixture (t = 0 min), for the specified BRD2₅₂₀₋₅₅₉ concentrations; (b) initial reaction velocities used for the kinetic analysis of the JMJD6-catalysed mono-hydroxylation of BRD2₅₂₀₋₅₅₉; (c) Michaelis-Menten curve used to determine the kinetic parameters of JMJD6 for BRD2₅₂₀₋₅₅₉. The JMJD6 v_{max}^{app} and K_m^{app} values of JMJD6 for BRD2₅₂₀₋₅₅₉ are $1.2 \pm 0.2 \cdot 10^{-2} \mu M \cdot s^{-1}$ and $0.89 \pm 0.32 \mu M$, respectively, as determined by non-linear regression; (d-m) time course of the JMJD6-catalysed hydroxylation of BRD2₅₂₀₋₅₅₉ showing the relative abundance of the BRD2₅₂₀₋₅₅₉ substrate (BRD2₅₂₀₋₅₅₉; orange circles), mono-hydroxylated BRD2₅₂₀₋₅₅₉ (BRD2₅₂₀₋₅₅₉ + 2[O]; purple triangles), trihydroxylated BRD2₅₂₀₋₅₅₉ (BRD2₅₂₀₋₅₅₉ + 4[O]; brown inverse triangles) in the presence of varied BRD2₅₂₀₋₅₅₉ concentrations: (d) 10 μ M; (e) 8 μ M; (f) 6 μ M; (g) 4 μ M; (h) 2 μ M; (i) 1 μ M; (j) 0.8 μ M; (k) 0.6 μ M; (l) 0.4 μ M.



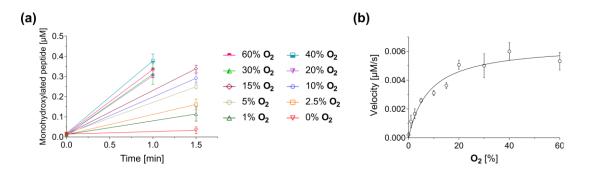
Supplementary Figure S8. Determination of the JMJD6 kinetic parameters for BRD3₄₆₃₋₅₀₂ (continues on the following page). SPE-MS was used to determine the extent of JMJD6catalysed mono- and di-hydroxylation of the BRD3₄₆₃₋₅₀₂ fragment peptide for the specified BRD3₄₆₃₋₅₀₂ concentrations. Initial rates of the JMJD6-catalysed mono-hydroxylation of BRD3₄₆₃₋₅₀₂ were used to determine the maximum velocity (v_{max}^{app}) and Michaelis constant (K_m^{app}) of JMJD6 for BRD3₄₆₃₋₅₀₂. Measurement times were normalized to the first sample injection analysed after the addition of the JMJD6 enzyme mixture to the substrate mixture, by which time low levels of BRD3₄₆₃₋₅₀₂ mono-hydroxylation were manifest. Data are the mean of three independent runs (n = 3; mean ± SD). Conditions: His₆-JMJD6 (50 nM), 2OG (200 μ M), FAS (2 μ M), BRD3₄₆₃₋₅₀₂ (as specified) and LAA (100 μ M) in Tris buffer (50 mM, pH 7.5).

(a) Abundance of the mono-hydroxylated BRD3₄₆₃₋₅₀₂ peptide, following the addition of 50 nM JMJD6 to the substrate mixture (t = 0 min), for the specified BRD3₄₆₃₋₅₀₂ concentrations; (b) initial reaction velocities used for the kinetic analysis of the JMJD6-catalysed mono-hydroxylation of BRD3₄₆₃₋₅₀₂; (c) Michaelis-Menten curve used to determine the kinetic parameters of JMJD6 for BRD3₄₆₃₋₅₀₂. The JMJD6 v_{max}^{app} and K_m^{app} values of JMJD6 for BRD3₄₆₃₋₅₀₂ are $5.2 \pm 1.3 \cdot 10^{-3}$ µM·s⁻¹ and 1.2 ± 0.4 µM, respectively, as determined by non-linear regression; (d-m) time course of the JMJD6-catalysed hydroxylation of BRD3₄₆₃₋₅₀₂ (BRD3₄₆₃₋₅₀₂ (BRD3₄₆₃₋₅₀₂ (BRD3₄₆₃₋₅₀₂ + [O]; green squares), di-hydroxylated BRD3₄₆₃₋₅₀₂ + 2[O]; purple triangles) and tri-hydroxylated BRD3₄₆₃₋₅₀₂ (BRD3₄₆₃₋₅₀₂ + 3[O]; light blue diamonds) in the presence of varied BRD3₄₆₃₋₅₀₂ concentrations: (d) 10 µM; (e) 8 µM; (f) 6 µM; (g) 4 µM; (h) 2 µM; (i) 1 µM; (j) 0.8 µM; (k) 0.6 µM; (l) 0.4 µM; (m) 0.2 µM.

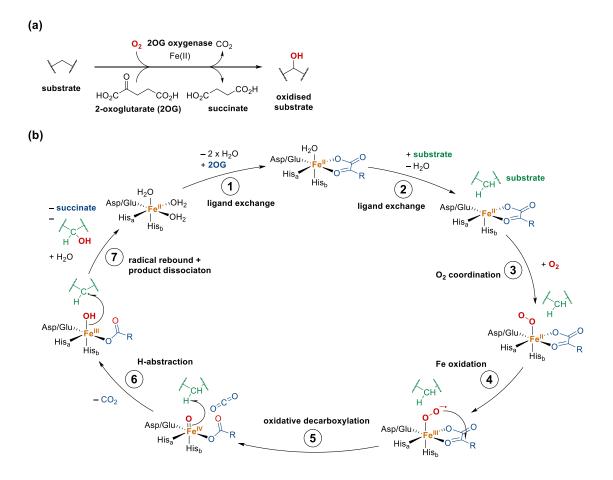


Supplementary Figure S9. Determination of the JMJD6 kinetic parameters for O₂. SPE-MS was used to determine the extent of JMJD6-catalysed mono-hydroxylation of the BRD4₅₁₁₋₅₅₀ fragment peptide for the specified O₂ concentrations. The concentration of O₂ [%] in the reaction buffer was converted into O₂ [μ M] by standard calibration ($y_{[\mu M]} = 9.026 \cdot x_{[\%]}$).² Hydroxylation reactions used to determine the JMJD6 kinetic parameters for O₂ were performed at 37 °C and quenched after 60 or 90 s, as indicated. Data are the mean of three independent runs (n = 3; mean ± SD). Note, no evidence for di- (or higher order) hydroxylation of BRD4₅₁₁₋₅₅₀ was observed under the reaction conditions used. Conditions: His₆-JMJD6 (50 nM), 2OG (200 μ M), FAS (2 μ M), BRD4₅₁₁₋₅₅₀ (2 μ M), LAA (100 μ M) and O₂ (as specified) in Tris buffer (50 mM, pH 7.5) at 37 °C.

(a) Initial reaction velocities used for the kinetic analysis of the JMJD6-catalysed mono-hydroxylation of BRD4₅₁₁₋₅₅₀ under different partial pressures of O₂; (b) Michaelis-Menten curve used to determine the kinetic parameters of JMJD6 for O₂. The JMJD6 v_{max}^{app} and K_m^{app} values of JMJD6 for O₂ are $6.5 \pm 0.4 \cdot 10^{-3} \,\mu\text{M} \cdot \text{s}^{-1}$ and $74.3 \pm 13.4 \,\mu\text{M}$, respectively, as determined by non-linear regression.



Supplementary Figure S10. Consensus mechanism for 2OG oxygenase catalysis. (a) 2OG oxygenases catalyse substrate oxidation (here: hydroxylation) in a manner coupled to the oxidative decarboxylation of 2OG to give succinate and carbon dioxide.³⁻⁵ Note that some 2OG oxygenases catalyse the substrate uncoupled turnover of 2OG to succinate and CO₂, at least in one case leading to a stable ternary enzyme:Fe(III):2OG complex.⁶ (b) The proposed consensus mechanism for hydroxylation reactions catalysed by 2OG oxygenases.³ Note, variations on this general catalytic cycle can occur, including with respect to the nature of the 2OG binding mode and the presence of Fe-binding waters, at different stages of catalysis.^{7, 8} Mechanistic studies using isotopically labelled O₂ have revealed high levels of O from O₂ is incorporated into the alcohol products of reactions catalysed by human 2OG oxygenases.⁹⁻¹¹ By contrast, (partial) incorporation of isotopically labelled O from water into alcohol products has been observed for reactions catalysed by some bacterial and fungal dioxygenases.¹²⁻¹⁷ Note, a consistently high level of a single isotopically labelled O from O₂ into the succinate coproduct has been reported.¹⁸



3. References

- M. Mantri, C. J. Webby, N. D. Loik, R. B. Hamed, M. L. Nielsen, M. A. McDonough, J. S. O. McCullagh, A. Böttger, C. J. Schofield and A. Wolf, *Med. Chem. Commun.*, 2012, 3, 80-85.
- 2. R. L. Hancock, N. Masson, K. Dunne, E. Flashman and A. Kawamura, *ACS Chem. Biol.*, 2017, **12**, 1011-1019.
- 3. S. Martinez and R. P. Hausinger, J. Biol. Chem., 2015, 290, 20702-20711.
- 4. K. S. Hewitson, N. Granatino, R. W. D. Welford, M. A. McDonough and C. J. Schofield, *Philos. Trans. R. Soc. A*, 2005, **363**, 807-828.
- 5. C. J. Schofield and R. Hausinger, *2-Oxoglutarate-Dependent Oxygenases*, The Royal Society of Chemistry, 2015.
- 6. G. Fiorini, S. A. Marshall, W. D. Figg, W. K. Myers, L. Brewitz and C. J. Schofield, *Sci. Rep.*, 2024, **14**, 26162.
- 7. M. A. McDonough, C. Loenarz, R. Chowdhury, I. J. Clifton and C. J. Schofield, *Curr. Opin. Struct. Biol.*, 2010, **20**, 659-672.
- 8. A. Brasnett, I. Pfeffer, L. Brewitz, R. Chowdhury, Y. Nakashima, A. Tumber, M. A. McDonough and C. J. Schofield, *Angew. Chem. Int. Ed.*, 2021, **60**, 14657-14663.
- 9. E. Holme, G. Lindstedt, S. Lindstedt and M. Tofft, *J. Biol. Chem.*, 1971, **246**, 3314-3319.
- 10. W. Min, T. P. Begley, J. Myllyharju and K. I. Kivirikko, *Bioorg. Chem.*, 2000, **28**, 261-265.
- L. A. McNeill, K. S. Hewitson, J. M. Gleadle, L. E. Horsfall, N. J. Oldham, P. H. Maxwell, C. W. Pugh, P. J. Ratcliffe and C. J. Schofield, *Bioorg. Med. Chem. Lett.*, 2002, **12**, 1547-1550.
- 12. J. E. Baldwin, R. M. Adlington, N. P. Crouch and I. A. C. Pereira, *Tetrahedron*, 1993, **49**, 7499-7518.
- 13. J. E. Baldwin, R. M. Adlington, N. P. Crouch, I. A. C. Pereira, R. T. Aplin and C. Robinson, J. Chem. Soc., Chem. Commun., 1993, 105-108.
- 14. B. Lindblad, G. Lindstedt and S. Lindstedt, J. Am. Chem. Soc., 1970, 92, 7446-7449.
- 15. P. J. Sabourin and L. L. Bieber, J. Biol. Chem., 1982, 257, 7468-7471.
- 16. Y. Kikuchi, Y. Suzuki and N. Tamiya, *Biochem. J.*, 1983, **213**, 507-512.
- 17. J. E. Baldwin, R. M. Adlington, N. P. Crouch and C. J. Schofield, *Tetrahedron*, 1988, 44, 643-650.

18. R. W. D. Welford, J. M. Kirkpatrick, L. A. McNeill, M. Puri, N. J. Oldham and C. J. Schofield, *FEBS Lett.*, 2005, **579**, 5170-5174.