

## Electronic Supporting Information

### Mechanoresponsive luminescence triggered by a supercooled transition of a copper(I) complex

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#### Synthesis.

Copper(I) iodide, tetrabutylphosphonium iodide and the solvent were purchased from Aldrich and used as received.

**(PBU<sub>4</sub>)<sub>2</sub>[Cu<sub>2</sub>L<sub>4</sub>] (1As).** CuI (132 mg, 0.7 mmol) and PBU<sub>4</sub>I (267 mg, 0.7 mmol) were dissolved in 20 mL of acetonitrile. After complete dissolution, the solvent is evaporated leading to a viscous product forming rapidly a white powder with yield of 95 % (M = 1154 g.mol<sup>-1</sup>).

Anal. Calcd (% wt.) for **1As** for C<sub>32</sub>H<sub>72</sub>Cu<sub>2</sub>L<sub>4</sub>P<sub>2</sub>: C, 33.32; H, 6.29. Found: C, 33.14, H, 5.98.

Anal. Calcd (% wt.) for **1RE** for C<sub>32</sub>H<sub>72</sub>Cu<sub>2</sub>L<sub>4</sub>P<sub>2</sub>: C, 33.32; H, 6.29. Found: C, 33.03; H, 6.01.

#### Characterizations.

Elemental analyses (C, H, N) were performed by the Service de microanalyses de l'ICSN - CNRS Gif-sur-Yvette.

The TG (Thermogravimetric Analysis) analyses were performed with a NETZCH STA449 F3 equipment. The DSC (Differential Scanning Calorimetry) measurements were performed with a NETZSCH DSC 200 F3 equipment at 1 °C/min.

Luminescence spectra were recorded with a Fluorolog 3 spectrofluorimeter (Horiba Jobin Yvon). Low temperature measurements were realized in an OptistatDN-V Oxford cryostat. The absolute photoluminescence quantum yields (PLQY) were measured by using the Jobin Yvon integrating sphere. Emission lifetimes ( $\tau$ ) were recorded with a time-resolved photoluminescence setup consisting of a frequency-tripled regenerative amplified femtosecond Ti:sapphire laser system from Spectra Physics to obtain  $\lambda_{\text{ex}} = 267$  and 400 nm and a C7700 streak camera from Hamamatsu coupled to an SP2300 imaging Acton spectrograph for the luminescence detection. Data were analyzed by exponential curve fitting using *Origin* software.

Powder X-ray diffraction (PXRD) diagrams were recorded on a Bruker D8 Advance instrument in the Bragg-Brentano geometry, equipped with a front germanium monochromator, a copper anode (CuK-L3 radiation  $\lambda = 1.540598 \text{ \AA}$ ) and a LynxEye PSD detector. The calculated patterns were obtained from the single crystal data (SCXRD) using the *Mercury* software.

Single crystal X-Ray diffraction (SCXRD) analyses were performed on a crystal of **1** (colorless prism, dimensions = 0.450 x 0.330 x 0.250 mm) mounted on the goniometer head of a D8 Venture (Bruker-AXS) diffractometer equipped with a CMOS-PHOTON70 detector, using Mo- $K\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ , multilayer monochromator) at  $T = 297(2) \text{ K}$ . Crystal structure was solved by dual-space algorithm using *SHELXT* program,<sup>1</sup> and then refined with full-matrix least-squares methods based on  $F^2$  (*SHELXL*).<sup>2</sup> All non-Hydrogen atoms were refined with anisotropic atomic displacement parameters. H atoms were finally included in their calculated positions and treated as riding on their parent atom with constrained thermal parameters. The  $(\text{PBU}_4)^+$  cations present structural disorder with two positions for each, having 75/25 and 50/50 occupancy ratio. Details of the crystal data and structure refinements are summarized in Table S1.

Raman spectra from 73 to 1500  $\text{cm}^{-1}$  and from 2500 to 3500  $\text{cm}^{-1}$  were measured with a Raman T64000 Jobin Yvon triple spectrometer. All spectra were recorded using an excitation wavelength of 633 nm emitted by a Helium Neon laser. The Rayleigh line was eliminated with a premonochromator.

### Computational details.

Density Functional Theory (DFT) calculations were conducted using the Gaussian16 package<sup>3</sup> based on previous computations on  $(\text{PPh}_4)_2[\text{Cu}_2\text{I}_4]$ .<sup>4</sup> The PBE0 functional, in conjunction with the triple  $\zeta$  basis set DEF2TZVP for Cu and I atoms,<sup>5,6</sup> was employed to examine the  $[\text{Cu}_2\text{I}_4]^{2-}$  moiety within the ideal  $D_{2h}$  symmetry point group. Inner electrons (relativistic effects) were not explicitly treated, as they are accounted for in the pseudo-potentials. Spin-orbit coupling was not considered in this study. The vibrational analysis utilized the anharmonic GVPT2 model, as described in previous publications.<sup>7</sup> Within this model, it becomes feasible to simulate the Raman signature, including energies and intensities, while considering various anharmonic effects such as overtones, combination bands, Fermi, and Darling-Dennisson resonances. Given the small size of the structure under investigation, our analysis extended to computing second overtones (MaxQuanta=3). In this context, we identified a strong band at 105  $\text{cm}^{-1}$  corresponding to Cu-I stretching which must correspond to the experimental 120  $\text{cm}^{-1}$  band. Two vibrational overtones were also calculated at 138 and 147  $\text{cm}^{-1}$  (corresponding fundamentals peaking at 69 ( $b_{2g}$ , Cu-I twisting) and 73 ( $b_{3u}$ , Cu-I wagging)  $\text{cm}^{-1}$ , respectively). These vibrational overtones can correspond to the experimental 146  $\text{cm}^{-1}$  band.

## Additional Data.

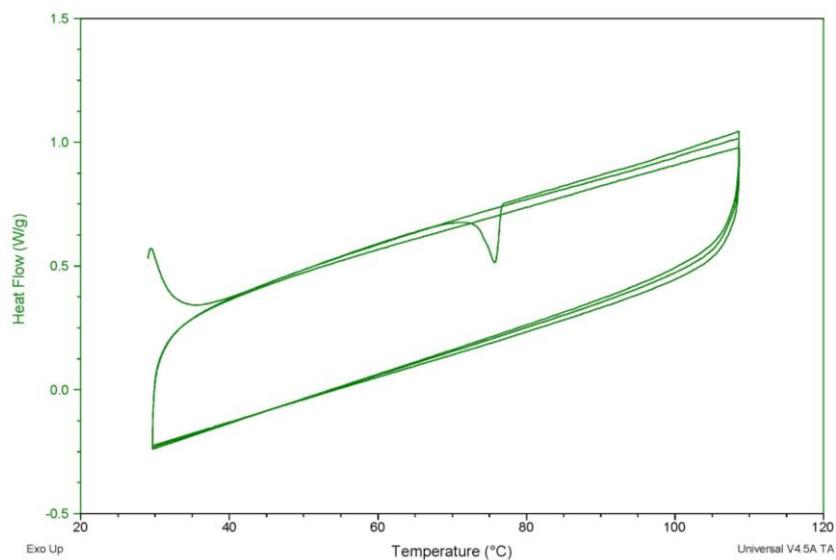


Figure S1. DSC (differential scanning calorimetry) curves of **1** at 1 °C/min.

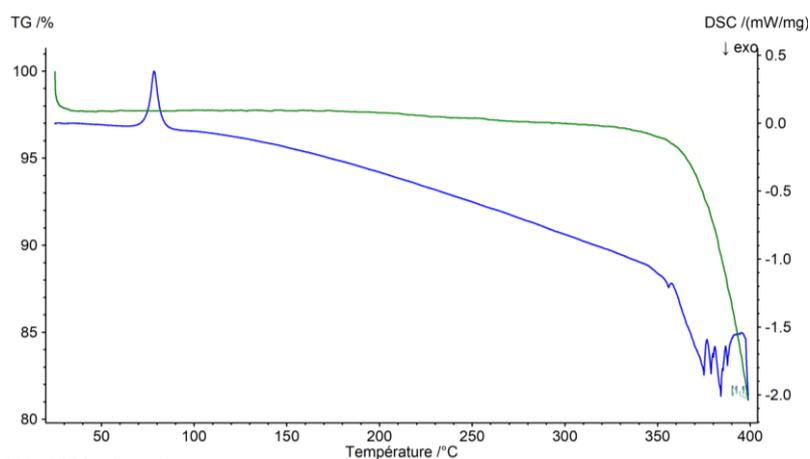


Figure S2. ATG-DTA curves of **1**.

Table S1. SCXRD and structure refinement data of  $(\text{Bu}_4\text{P})_2[\text{Cu}_2\text{I}_4]$  (**1**) at 293 K.

	<b><math>(\text{Bu}_4\text{P})_2[\text{Cu}_2\text{I}_4]</math></b>
CCDC n°	2314646
Chemical formula	$\text{C}_{32} \text{H}_{72} \text{P}_2 \text{Cu}_2 \text{I}_4$
fw	1153.51
Crystal system	Triclinic
Space group	$P - 1$
a, Å	12.4248(14)
b, Å	12.9746(16)
c, Å	16.828(2)

$\alpha$ , deg	86.253(5)
$\beta$ , deg	75.221(4)
$\gamma$ , deg	64.491(4)
$V$ , Å <sup>3</sup>	2364.3(5)
$Z$	2
$\rho_{\text{calc}}$ , g/cm <sup>3</sup>	1.620
$\mu$ , mm <sup>-1</sup>	3.596
Reflections collected	31523
Independent reflections	8951
$R_{\text{int}}$	0.0457
Reflections $I > 2\sigma(I)$	6254
Parameters	489
GOF on $F^2$	1.035
$R_1^a/wR_2^b$ ( $I > 2\sigma(I)$ )	0.0561/0.1429
$R_1^a/wR_2^b$ (all)	0.0807/0.1673

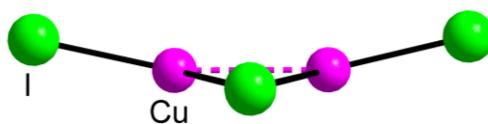
<sup>a</sup>  $R_1 = [\sum \text{abs}(\text{abs}(F_o) - \text{abs}(F_c))]/[\sum \text{abs}(F_o)]$ . <sup>b</sup>  $wR_2 = [\sum(w(F_o^2 - F_c^2)^2)/\sum(w(F_o^2)^2)]^{0.5}$ .

**Table S2.** Selected intramolecular bonds lengths and angles of (Bu<sub>4</sub>P)<sub>2</sub>[Cu<sub>2</sub>I<sub>4</sub>] (**1**) from SCXRD at 293 K.

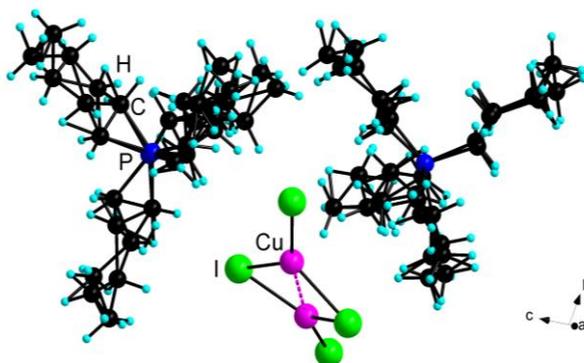
	<b>1</b> CRYST
<b>Cu<sup>...</sup>Cu</b>	2.6576(14)
<b>Cu-I<math>\mu</math><sub>2</sub></b>	2.5450(12) 2.5805(12) 2.5855(12) 2.5924(13)
<b>Cu-I<math>\mu</math><sub>1</sub></b>	2.4936(11) 2.4907(12)
<b>I<math>\mu</math><sub>1</sub>-Cu-I<math>\mu</math><sub>2</sub></b>	125.02(4) 118.34(4) 122.03(5) 122.85(5)
<b>I<math>\mu</math><sub>2</sub>-Cu-I<math>\mu</math><sub>2</sub></b>	116.62(4) 115.12(4)
<b>I-Cu-I-Cu</b>	165.52(7) 164.75(7) 165.68(7) 164.59(7)

**Table S3.** Photophysical data of **1**.

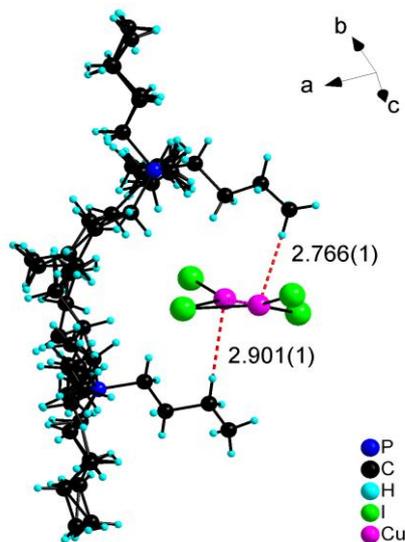
	$\lambda_{\max}$ (nm)	PLQY	$\tau$ ( $\mu\text{s}$ )	$k_r$ ( $\mu\text{s}^{-1}$ )	$k_{nr}$ ( $\mu\text{s}^{-1}$ )
<b>1<sub>AS</sub></b>	520	0.90	4.6	0.20	0.02
<b>1<sub>sc</sub></b>	650	0.04	0.47	0.09	2.04
<b>1<sub>RE</sub></b>	520	0.87	3.0	0.29	0.04



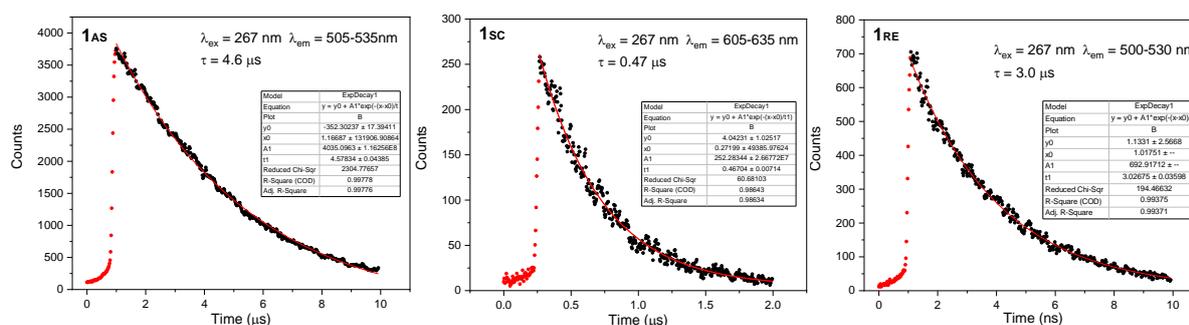
**Figure S3.** Molecular structure of  $[\text{Cu}_2\text{I}_4]^{2-}$  anion showing its folded geometry with  $\text{I}\mu_1\text{-Cu-I}\mu_2\text{-Cu}$  mean torsional angles of  $165.14(7)^\circ$  at 293 K.



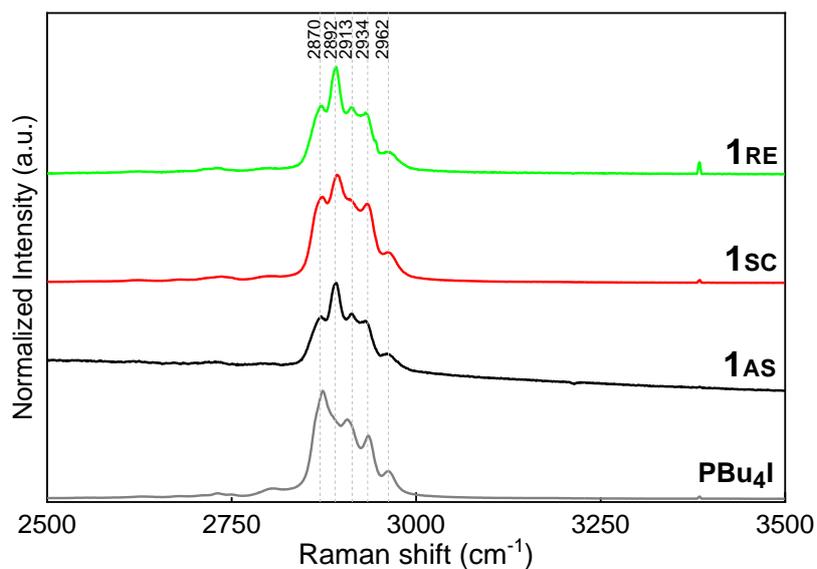
**Figure S4.** Molecular structure of  $(\text{Bu}_4\text{P})_2[\text{Cu}_2\text{I}_4]$  (**1**) showing the structural disorder of the two  $(\text{Bu}_4\text{P})^+$  cations at 293 K.



**Figure S5.** Intermolecular Cu...H contacts in **1** structure at 293 K.



**Figure S6.** Fitted emission decay curves at 293 K. The red dots are not considered for the fits.



**Figure S7.** Raman spectra of **1**, as synthesized (**1AS**), in the supercooled state (**1sc**) and in the mechanically recrystallized state (**1RE**), from 2500 to 3500  $\text{cm}^{-1}$  at 293 K. The vibrations are from the butyl chains of the  $(\text{Bu}_4\text{P})^+$  cation.

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