Electronic Supplementary Information

A calix[3]carbazole-based cavitand: synthesis, structure and its complexation with fullerenes

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Contents

1. General information	S2
2. Synthesis and characterization	S2
3. NMR and mass spectra of 2	S3
4. X-ray crystal structures and crystallographic data	S6
5. Photophysical properties	S25
6. Electrochemical measurements	S27
7. UV-vis absorption spectra studies of complexations	S28
8. MALDI-TOF mass spectra studies of complexations	S29
9. Color changes of complexations	S30
10. Determination of the association constants	S30
11. Theoretical calculations	
12. Cartesian coordinates of the optimized structures	
13. References	S50

1. General information

All the reagents and solvents were commercially available and used without further purification. Unless otherwise stated, the reactions were carried out under inert and anhydrous conditions. Thin layer chromatography (TLC) was performed by silica gel GF₂₅₄. The melting point was determined by WRR melting point apparatus without correction. ¹H, ¹³C NMR spectra were recorded on Bruker[®] AVIII 300 MHz NMR, Bruker[®] AVIII 500 MHz NMR and Bruker 700 MHz NMR spectrometers and reported in parts per million (ppm) from internal standard TMS. Matrix-Assisted Laser Desorption/Ionization Time-of-Flight (MALDI-TOF) mass spectrometry were recorded on the Autoflex III. Single crystal data were collected on a Bruker Smart APEXII CCD diffractometer using graphite monochromated CuKα radiation. Ultraviolet-visible spectra were recorded on PerkinElmer[®] UV/Vis/NIR spectrometer (Lambda 950), and fluorescence spectra and transient photoluminance decay characteristics were measured using the Edinburgh Instruments FLS1000 spectrometer. Based on density functional theory (DFT), the structures were optimized by using the B3LYP functional and 6-31G(d) basis set carried out by Gaussian 16 program.^{\$\$1}

2. Synthesis and characterization

The synthetic procedures of calix[3]carbazole 1 were described in the literatures.^{S2, S3}



Calix[3]carbazole-based deep cavitand (2): To a solution of **1** (172 mg, 0.2 mmol) and Cs_2CO_3 (1.303 g, 4 mmol) in anhydrous dimethyl sulfoxide (20 mL) was added 4,5-difluorophthalonitrile (148 mg, 0.9 mmol) under argon. The mixture was stirred and heated at 80 °C for 24 hours. Then the reaction was quenched by the addition of water (50 mL), and extracted with dichloromethane (3 × 100 mL). The combined organic layer was dried with anhydrous MgSO₄, and concentrated under a reduced pressure

to give a crude solid. The solid was purified by column chromatography with CH₂Cl₂ as the eluent to give **2** as light yellow solid (167 mg, 68 %). M.p.: > 280 °C. ¹H NMR (300 MHz, CDCl₃, 298K): δ 7.99 (s, 6H), 7.62 (s, 6H), 7.57 (t, *J* = 7.2 Hz, 6H), 7.47 (t, *J* = 7.4 Hz, 3H), 7.39 (d, *J* = 8.2 Hz, 6H), 7.09 (s, 6H), 4.67 (d, *J* = 13.1 Hz, 3H), 3.89 (d, *J* = 13.1 Hz, 3H). ¹³C NMR (126 MHz, CDCl₃, 298 K): δ 153.7, 141.6, 136.8, 130.5, 128.6, 126.6, 122.3, 120.9, 114.7, 112.4, 102.4, 29.9. MALDI-TOF MS: *m/z* calcd for [M] C₈₁H₃₉N₉O₆: 1233.3023; found: 1233.3021.

3. NMR and mass spectra of 2



Fig. S1 ¹H NMR spectrum (300 MHz, CDCl₃, 298K) of 2.



Fig. S2 ¹³C NMR spectrum (126 MHz, CDCl₃, 298K) of 2.



Fig. S3 MALDI-TOF mass spectrum of 2.



Fig. S4 ¹H-¹H COSY spectrum (700 MHz, CDCl₃, 298 K) of 2.



Fig. S5 ¹H-¹³C HSQC spectrum (700 MHz, CDCl₃, 298 K) of 2.



Fig. S6 2D NOESY spectrum (700 MHz, CDCl₃, 298 K) of 2.

4. X-ray crystal structures and crystallographic data



Fig. S7 ORTEP drawing of **2-c1** from (a) top view and (b) side view (the thermal ellipsoids are displayed at a 30 % probability).

CCDC	2305712
Empirical formula	C ₈₁ H ₃₉ N ₉ O ₆
Formula weight	1233.30
Temperature/K	169.99(10)
Crystal system	monoclinic
Space group	$P2_{l}/m$
a/Å	12.7611(3)
b/Å	18.5038(3)
c/Å	14.1018(3)
$\alpha/^{\circ}$	90
β/°	100.643(2)
$\gamma^{/\circ}$	90
Volume/Å ³	3272.56(12)
Z	4
pcalcg/cm ³	1.394
μ/mm^{-1}	0.959
F(000)	1415.0
Crystal size/mm ³	$0.33 \times 0.26 \times 0.14$
Radiation	$CuK\alpha$ ($\lambda = 1.54184$)
2Θ range for data collection/°	6.378 to 151.102
Index ranges	$-16 \le h \le 15, -19 \le k \le 23, -17 \le l \le 17$
Reflections collected	25286
Independent reflections	$6729 [R_{int} = 0.0449, R_{sigma} = 0.0362]$
Data/restraints/parameters	6729/0/445
Goodness-of-fit on F ²	1.060
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0457, wR_2 = 0.1254$
Final R indexes [all data]	$R_1 = 0.0602, wR_2 = 0.1341$
Largest diff. peak/hole / e Å ⁻³	0.26/-0.27

 Table S1. X-ray crystallographic and the refinement data of 2-c1.



Fig. S8 The multiple intramolecular C-H $\cdots \pi$ interactions of 2-c1.



Fig. S9 The multiple intermolecular $\pi \cdots \pi$ interactions of two 2-c1 molecules.



Fig. S10 The packing mode of 2-c1 viewed along *a* axis.



Fig. S11 The packing mode of 2-c1 viewed along *b* axis.



Fig. S12 The packing mode of 2-c1 viewed along *c* axis.



Fig. S13 ORTEP drawing of 2-c2 from (a) top view and (b) side view (the thermal ellipsoids are displayed at a 30 % probability).

CCDC	2305713
Empirical formula	$C_{82}H_{41}Cl_2N_9O_6$
Formula weight	1317.26
Temperature/K	169.98(10)
Crystal system	monoclinic
Space group	<i>12/a</i>
a/Å	30.8326(16)
b/Å	13.5754(7)
c/Å	31.1005(15)
$lpha/^{\circ}$	90
β/°	106.960(6)
$\gamma/^{\circ}$	90
Volume/Å ³	12451.4(12)
Ζ	8
pcalcg/cm ³	1.359
µ/mm ⁻¹	1.059
F(000)	5243.0
Crystal size/mm ³	0.3 imes 0.2 imes 0.1
Radiation	$CuK\alpha \ (\lambda = 1.54184)$
2Θ range for data collection/°	5.942 to 152.224
Index ranges	$-38 \le h \le 38, -16 \le k \le 16, -39 \le l \le 38$
Reflections collected	80263
Independent reflections	12769 [$R_{int} = 0.0935$, $R_{sigma} = 0.0458$]
Data/restraints/parameters	12769/282/975
Goodness-of-fit on F ²	1.532
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.1210, wR_2 = 0.2740$
Final R indexes [all data]	$R_1 = 0.1954, wR_2 = 0.3023$
Largest diff. peak/hole / e Å ⁻³	0.54/-0.46

 Table S2. X-ray crystallographic and the refinement data of 2-c2.



Fig. S14 The multiple intramolecular C-H $\cdots \pi$ interactions of 2-c2.



Fig. S15 The multiple intermolecular $\pi \cdots \pi$ and C-H $\cdots \pi$ interactions of two 2-c2 molecules.



Fig. S16 The packing mode of 2-c2 viewed along *b* axis.



Fig. S17 ORTEP drawing of $C_{60}@2$ from (a) top view and (b) side view (the thermal ellipsoids are displayed at a 30 % probability).

CCDC	2305715	
Empirical formula	C153H47Cl4N9O6	
Formula weight	2248.79	
Temperature/K	169.98(10)	
Crystal system	monoclinic	
Space group	$P2_1/n$	
a/Å	13.0791(4)	
b/Å	32.0488(9)	
c/Å	27.0224(9)	
$\alpha/^{\circ}$	90	
β/°	94.916(3)	
$\gamma/^{\circ}$	90	
Volume/Å ³	11285.3(6)	
Ζ	4	
pcalcg/cm ³	1.324	
μ/mm^{-1}	1.493	
F(000)	4576.0	
Crystal size/mm ³	$0.45 \times 0.35 \times 0.31$	
Radiation	$CuK\alpha (\lambda = 1.54184)$	
2Θ range for data collection/°	4.286 to 152.546	
Index ranges	$-16 \le h \le 16, -39 \le k \le 23, -33 \le l \le 33$	
Reflections collected	80169	
Independent reflections	22551 [$R_{int} = 0.0963$, $R_{sigma} = 0.0906$]	
Data/restraints/parameters	22551/633/1550	
Goodness-of-fit on F ²	1.512	
Final R indexes [I>=2σ (I)]	$R_1 = 0.1425, wR_2 = 0.4056$	
Final R indexes [all data]	$R_1 = 0.1961, wR_2 = 0.4366$	
Largest diff. peak/hole / e Å ⁻³	2.07/-1.08	

Table S3. X-ray crystallographic and the refinement data of $C_{60}@2$



Fig. S18 Crystal structure of complex C_{60} (*a*) 2 with two *o*-DCBs around from top view.



Fig. S19 Crystal structure of complex C_{60} (*a*) 2 with two *o*-DCBs around from side view.



Fig. S20 Space-filling model of the 1:1 complexation between 2 and C_{60} from top view.



Fig. S21 Space-filling model of the 1:1 complexation between 2 and C₆₀ from side view.



Fig. S22 Side view of 2 showing the cavity depth.



Fig. S23 The multiple intermolecular interactions between 2 and encapsulated C_{60} molecule.



Fig. S24 Intermolecular interactions between the adjacent complexes C_{60} (*a*) 2 and four *o*-DCBs.



Fig. S25 The multiple intermolecular $\pi \cdots \pi$ interactions of two adjacent complexes C₆₀@2.



Fig. S26 The ordered packing mode of complexes C_{60} (*i*)2.



Fig. S27 The packing mode of complexes $C_{60}(a)$ viewed along *a* axis.



Fig. S28 The tubular packing structure of complexes C_{60} (a)2.



Fig. S29 ORTEP drawing of $C_{70}@2$ from (a) top view and (b) side view (the thermal ellipsoids are displayed at a 30 % probability).

CCDC	2305716
Empirical formula	$C_{163}H_{47}Cl_4N_9O_6$
Formula weight	2368.89
Temperature/K	169.99(10)
Crystal system	monoclinic
Space group	$P2_1/n$
a/Å	26.5926(7)
b/Å	32.9430(7)
c/Å	30.1206(12)
$\alpha/^{\circ}$	90
β/°	113.007(4)
$\gamma/^{\circ}$	90
Volume/Å ³	24288.0(14)
Z	8
pcalcg/cm ³	1.296
μ/mm^{-1}	1.417
F(000)	9632.0
Crystal size/mm ³	$0.42 \times 0.39 \times 0.37$
Radiation	$CuK\alpha$ ($\lambda = 1.54184$)
2Θ range for data collection/°	4.498 to 152.092
Index ranges	$-32 \le h \le 32, -30 \le k \le 41, -37 \le l \le 37$
Reflections collected	175708
Independent reflections	$48649 \ [R_{int} = 0.1184, R_{sigma} = 0.1032]$
Data/restraints/parameters	48649/1187/3277
Goodness-of-fit on F ²	1.395
Final R indexes [I>= 2σ (I)]	$R_1 = 0.1108, wR_2 = 0.2615$
Final R indexes [all data]	$R_1 = 0.2002, wR_2 = 0.2962$
Largest diff. peak/hole / e Å ⁻³	0.96/-0.63

Table S4. X-ray crystallographic and the refinement data of $C_{70}@2$



Fig. S30 Crystal structure of complexes C_{70} (*a*) 2 with four *o*-DCBs around from top view.



Fig. S31 Crystal structure of complexes C₇₀@2 with four *o*-DCBs around from side view.



Fig. S32 Space-filling model of the 1:1 complexation between 2 and C₇₀ from top view.



Fig. S33 Space-filling model of the 1:1 complexation between 2 and C_{70} from side view.



Fig. S34 Side view of 2 showing the cavity depth.



Fig. S35 The multiple intermolecular interactions between 2 and encapsulated C_{70} molecule and *o*-DCBs.



Fig. S36 The multiple intermolecular $\pi \cdots \pi$ interactions of two adjacent complexes C_{70} @2.



Fig. S37 The multiple intermolecular interactions between two adjacent complexes C_{70} @2 and *o*-DCB.



Fig. S38 The packing mode of complexes C_{70} (*a*)2 containing *o*-DCBs viewed along *c* axis.



Fig. S39 The tubular packing structure of complexes C₇₀@2 containing *o*-DCBs.



5. Photophysical properties

Fig. S40 UV-vis absorption spectrum of 2 (c = 1.0×10^{-5} M) in toluene at 298 K.



Fig. S41 Fluorescence spectrum of 2 (c = 1.0×10^{-5} M, excited at 311 nm) in toluene at 298 K.



Fig. S42 Transient decay spectrum of 2 ($c = 1.0 \times 10^{-5} \text{ mol/L}$) in toluene at 298 K.

6. Electrochemical measurements



Fig. S43 Cyclic voltammogram of 2 in CH_2Cl_2 containing 0.1 M *n*-Bu₄NPF₆ at room temperature at a scan rate of 0.1 V/s.



Fig. S44 Cyclic voltammogram of 2 in THF containing 0.1 M n-Bu₄NPF₆ at room temperature at a scan rate of 0.1 V/s.

7. UV-vis absorption spectra studies of complexations



Fig. S45 UV-vis absorption spectra of C_{60} and 2 (1.0×10⁻⁵ M) in the absence and presence of C_{60} (3 equiv.) in toluene.



Fig. S46 UV-vis absorption spectra of C_{70} and 2 (1.0×10⁻⁵ M) in the absence and presence of C_{70} (3 equiv.) in toluene.

8. MALDI-TOF mass spectra studies of complexations



Fig. S47 MALDI-TOF mass spectrum of complex C_{60} @2.



Fig. S48 MALDI-TOF mass spectrum of complex C₇₀@2.

9. Color changes of complexations







Fig. S50 Solution of C_{70} (left), C_{70} mixed with one equivalent of **2** (mid) and **2** (right), 1.0×10^{-3} M in toluene.

10. Determination of the association constants

Fluorescence titration experiments were performed at 298 K. A stock solution of the cavitand **2** (1.0×10^{-4} M in toluene) was mixed with a stock solution of fullerenes C₆₀ and C₇₀ (1.0×10^{-3} M in toluene) to give samples with a concentration of [H] = 1.0×10^{-5} M (for **2**) and [G] = (0^{-3}) ×10⁻⁵ M (for the corresponding fullerene). The emission spectra (excited at 311 nm) were measured for each sample, and the emission at 431 nm was monitored and fitted to a binding isotherm using Bindfit.^{S4-6} The

resulting fits were used to calculate K_a values in addition to an error margin for 95% confidence of fit.



Titration of fullerence C60 with cavitand 2

Fig. S51 Nonlinear fitting curve (left) and association constant (right) for host 2 and guest C₆₀.

Titration of fullerence C70 with cavitand 2



Fig. S52 Nonlinear fitting curve (left) and association constant (right) for host 2 and guest C₇₀.

11. Theoretical calculations



Fig. S53 Optimized structure of **2-c1** from (a) top view and (b) side view at the B3LYP/6-31G(d) level of theory.



Fig. S54 Frontier molecular orbitals and HOMO-LUMO energy gap of **2-c1** calculated at the B3LYP/6-31G(d) level of theory.



Fig. S55 Optimized structure of **2-c2** from (a) top view and (b) side view at the B3LYP/6-31G(d) level of theory.



Fig. S56 Frontier molecular orbitals and HOMO-LUMO energy gap of **2-c2** calculated at the B3LYP/6-31G(d) level of theory.



Fig. S57 Optimized structure of **2-c3** from (a) top view and (b) side view at the B3LYP/6-31G(d) level of theory.



Fig. S58 Frontier molecular orbitals and HOMO-LUMO energy gap of **2-c3** calculated at the B3LYP/6-31G(d) level of theory.



Fig. S59 Optimized structure of C_{60} (a) top view and (b) side view at the B3LYP/6-31G(d) level of theory.



Fig. S60 Frontier molecular orbitals and HOMO-LUMO energy gap of C_{60} (a) calculated at the B3LYP/6-31G(d) level of theory.



Fig. S61 Optimized structure of C_{70} @2 from (a) top view and (b) side view at the B3LYP/6-31G(d) level of theory.



Fig. S62 Frontier molecular orbitals and HOMO-LUMO energy gap of C_{70} (*a*) 2 calculated at the B3LYP/6-31G(d) level of theory.

Electrostatic potential (ESP) calculations



Fig. S63 Electrostatic potential (ESP) map of 2-c1 from (a) top view and (b) side view.



Fig. S64 Electrostatic potential (ESP) map of 2-c2 from (a) top view and (b) side view.



Fig. S65 Electrostatic potential (ESP) map of 2-c3 from (a) top view and (b) side view.



Fig. S66 Electrostatic potential (ESP) map of C_{60} (*a*) from (a) top view and (b) side view.



Fig. S67 Electrostatic potential (ESP) map of C_{70} (*a*) top view and (*b*) side view.

Independent gradient model (IGM) analysis



Fig. S68 Independent gradient model (IGM) analysis of C_{60} (*i*) from (a) top view and (b) side view.



Fig. S69 Independent gradient model (IGM) analysis of C_{70} (a) top view and (b) side view.

12. Cartesian coordinates of the optimized structure

0	8.2653	9.1036	1.2926
0	7.7543	7.7748	3.6502
0	5.679	12.5317	11.5993
Ν	9.418	13.8778	0.8561
Ν	7.5009	9.5121	8.2539
С	8.0725	10.4908	1.5241
Ν	0.7379	11.902	14.7376
С	6.6797	10.5338	8.7886

2-c1

С	4.8822	12.6237	9.2806
С	4.014	13.8778	9.4098
Η	3.3215	13.8778	8.7023
Н	3.5576	13.8778	10.2883
С	8.7198	12.7609	1.2571
С	7.3024	8.6699	4.6399
С	6.8773	10.8527	2.1835
С	10.7694	13.8778	0.3751

13.9952	8.0245	1.116
6.251	10.344	6.5372
4.6109	13.18	12.1445
5.659	12.0437	10.2866
5.9639	9.8161	2.8008
5.025	10.1257	2.7488
6.037	8.966	2.2987
8.9826	11.3921	1.0346
9.7578	11.1071	0.5653
5.8951	11.0354	7.7555
7.2458	9.4246	6.8845
6.6361	12.1994	2.3812
5.8493	12.4697	2.8399
7.5239	13.1657	1.921
5.0145	12.0945	8.0024
4.5017	12.4554	7.2888
6.3437	9.5895	4.2386
9.293	8.518	2.0022
2.571	13.1756	13.4047
7.776	8.5442	5.931
8.4287	7.8936	6.1616
6.5764	11.0198	10.086
7.1031	10.6718	10.7958
10.5825	8.5259	1.4678
10.7468	8.9412	0.6293
5.793	10.4456	5.2325
5.1211	11.0777	5.0055
13.389	13.8778	-0.5165
14.2862	13.8778	-0.8288
11.6252	7.9265	2.1623
9.0462	7.8879	3.2113
3.6078	12.4696	12.7948
3.6283	11.5203	12.8231
10.0976	7.2605	3.8939
9.9229	6.8049	4.7089
11.0253	13.8778	-0.9808
10.3095	13.8778	-1.6054
8.5643	8.8302	8.9446
11.391	7.2975	3.3922
1.5465	12.4631	14.1432
12.9516	7.9897	1.5874
	13.9952 6.251 4.6109 5.659 5.9639 5.025 6.037 8.9826 9.7578 5.8951 7.2458 6.6361 5.8493 7.5239 5.0145 4.5017 6.3437 9.293 2.571 7.776 8.4287 6.5764 7.1031 10.5825 10.7468 5.793 5.1211 13.389 14.2862 11.6252 9.0462 3.6078 3.6283 10.0976 9.9229 11.0253 10.3095 8.5643 11.391 1.5465 12.9516	13.99528.02456.25110.3444.610913.185.65912.04375.96399.81615.02510.12576.0378.9668.982611.39219.757811.10715.895111.03547.24589.42466.636112.19945.849312.46977.523913.16575.014512.09454.501712.45546.34379.58959.2938.5182.57113.17567.7768.54428.42877.89366.576411.01987.103110.671810.58258.525910.74688.94125.79310.44565.121111.077713.38913.877814.286213.877811.62527.92659.04627.88793.607812.46963.628311.520310.09767.26059.92296.804911.025313.877810.309513.877810.309513.877810.309513.877810.309513.877810.309513.877810.309513.877810.309513.877810.309513.877810.309513.877810.309513.877810.309513.877810.309513.877810.309513.877810.309513.877810.309513.8778 <tr< td=""></tr<>

IN	13.2951	6.2886	4.8082
С	11.8087	13.8778	1.2757
Н	11.6224	13.8778	2.2073
С	8.3308	8.2217	10.1587
Н	7.4682	8.2394	10.5562
С	12.4621	6.7301	4.1486
С	9.8328	8.8046	8.3611
Н	9.9947	9.2499	7.5377
С	12.3392	13.8778	-1.4197
Н	12.5224	13.8778	-2.3518
С	9.4	7.5777	10.7887
Н	9.2582	7.1707	11.6353
С	10.8534	8.1172	9.0047
Н	11.7106	8.0604	8.5991
С	10.6351	7.516	10.2272
Н	11.342	7.0633	10.672
С	13.1276	13.8778	0.8398
Н	13.8412	13.8778	1.467
0	8.2653	18.652	1.2926
0	7.7543	19.981	3.6502
0	5.679	15.224	11.5993
Ν	7.5009	18.2436	8.2539
С	8.0725	17.2649	1.5241
Ν	0.7379	15.8537	14.7376
С	6.6797	17.2219	8.7886
С	4.8822	15.132	9.2806
С	8.7198	14.9948	1.2571
С	7.3024	19.0857	4.6399
С	6.8773	16.903	2.1835
Ν	13.995	19.7312	1.116
С	6.251	17.4117	6.5372
С	4.6109	14.5757	12.1445
С	5.659	15.712	10.2866
С	5.9639	17.9396	2.8008
Н	5.025	17.63	2.7488
Н	6.037	18.7897	2.2987
С	8.9826	16.3636	1.0346
Н	9.7578	16.6486	0.5653
С	5.8951	16.7203	7.7555
С	7.2458	18.3311	6.8845
С	6.6361	15.5563	2.3812

Η	5.8493	15.286	2.8399
С	7.5239	14.59	1.921
С	5.0145	15.6612	8.0024
Н	4.5017	15.3003	7.2888
С	6.3437	18.1662	4.2386
С	9.293	19.2377	2.0022
С	2.571	14.5801	13.4047
С	7.776	19.2114	5.931
Н	8.4287	19.8622	6.1616
С	6.5764	16.7359	10.086
Н	7.1031	17.0839	10.7958
С	10.5825	19.2297	1.4678
Н	10.7468	18.8145	0.6293
С	5.793	17.3101	5.2325
Н	5.1211	16.678	5.0055
С	11.6252	19.8292	2.1623
С	9.0462	19.8677	3.2113
С	3.6078	15.2861	12.7948
Н	3.6283	16.2354	12.8231

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C₇₀@2

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