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Complex	Formula	Conquest	CCDC	Reference
1	[RNH ₃][V ^V ₂ O ₂ (H ₁ TBC[8])]	-	-	8
2	[V ^V ₂ (<i>p</i> -tolylN) ₂ (H ₂ TBC[8])]	XENCOI	155489	9
3	[PPh ₄][V ^V ₂ O ₂ (H ₁ TBC[8])]	HEQJIX	610228	10
4	[Li(MeCN) ₄][V ^V ₂ O ₂ (H ₁ TBC[8])]	KUGJIH	1017371	11
5	[Na(MeCN) ₅][V ^V ₂ O ₂ (H ₁ TBC[8])]	KUGHIF	895365	11
6	[^t BuNH ₃][V ^V ₂ (<i>p</i> -tolylN) ₂ (H ₁ TBC[8])]	KUGJOL	895369	11
7	[Na(MeCN) ₆][V ^V ₈ Na ₇ O ₁₆ (TBC[8]) ₂ (MeCN) ₆]	KUGHOL	895366	11
8	[V ^V ₄ O ₆ (TBC[8])]	KUGHUR	895367	11
9	[V ^V ₄ Li ₆ O ₈ (TBC[8])(O ^t Bu) ₂ (THF) ₂ (Et ₂ O) ₂]	KUGJED	1017370	11
10	[V ^V ₂ Nb ^V ₂ O ₅ (O ⁱ Pr) ₂ (TBC[8])(MeCN) ₂]	YEKNAH	2156671	12
11	[V ^V ₂ Ta ^V ₂ O ₅ (O ⁱ Pr) ₂ (TBC[8])(MeCN) ₂]	YEKMUA	2156670	12
12	[V ^V ₃ V ^{IV} ₁ Nb ^V ₂ O ₉ (O ⁿ Pr) ₃ (TBC[8])(MeCN) ₂]	YEKNEL	2156672	12
13	[HNEt ₃] ₂ [V ^V O ₂ (H ₅ TBC[8])]	MIPXIV	1574244	13
14	[HNEt ₃] ₂ [Co ^{II} ₂ (H ₅ TBC[8]) ₂]	TIQFOP	631678	14
15a	[Sm ^{III} Co ^{II} ₈ (TBC[8]) ₂ O ₂ (OH) ₄ (CO ₃) ₂ (OAc) ₄ (HCOO) ₂ (dmf) ₈ (H ₂ O) ₆]	NEWCEZ	876031	15
15b	[Gd ^{III} Co ^{II} ₈ (TBC[8]) ₂ O ₂ (OH) ₄ (CO ₃) ₂ (OAc) ₄ (HCOO) ₂ (dmf) ₁₀ (H ₂ O) ₄]	DICZIA	876032	15
15c	[Dy ^{III} Co ^{II} ₈ (TBC[8]) ₂ O ₂ (OH) ₄ (CO ₃) ₂ (OAc) ₄ (HCOO) ₂ (dmf) ₁₀ (H ₂ O) ₄]	DICZEW	876033	15
16a	[Dy ^{III} ₄ Co ^{II} ₄ (TBC[8]) ₂ O ₂ (def) ₈ (H ₂ O) ₄]	PIBXUW	1548028	16
16b	[Er ^{III} ₄ Co ^{II} ₄ (TBC[8]) ₂ O ₂ (def) ₈ (H ₂ O) ₄]	PIBXOQ	1548027	16
16c	[Y ^{III} ₄ Co ^{II} ₄ (TBC[8]) ₂ O ₂ (def) ₆ (H ₂ O) ₆]	PIBYEH	1568708	16
17a	[HNEt ₃][Y ^{III} Co ^{II} ₆ (H ₁ TBC[8]) ₃ (OMe) ₆ (OAc) ₂ (CO ₃)(dmf) ₆ (MeOH)]	CABBEQ	2013507	17
17b	[HNEt ₃][Eu ^{III} Co ^{II} ₆ (H ₁ TBC[8]) ₃ (OMe) ₆ (OAc) ₂ (CO ₃)(dmf) ₆ (MeOH)]	CUZXAZ	2013506	17
17c	[HNEt ₃][Dy ^{III} Co ^{II} ₆ (H ₁ TBC[8]) ₃ (OMe) ₆ (OAc) ₂ (CO ₃)(dmf) ₆ (MeOH)]	CUZXED	2013508	17
18	[Co ^{II} ₂ Th ^{IV} ₄ O ₂ (OH) ₂ (H ₁ TBC[8]) ₂ (dmf) ₆]	JIYSIV	948469	18
19	[Co ^{II} ₂ Li ₈ (OH) ₂ (TBC[8])Br ₂ (MeCN) ₄ (THF) ₆]	GAZHUO	2090098	19
20	[Co ^{II} ₂ Li ₆ (OH) ₂ (TBC[8])Br ₂ (MeCN) ₄]	GAZJAW	2090100	19
21	[Co ^{II} ₄ Na(H ₄ TBC[8]) ₂ Br(MeCN) ₆]	GAZCUJ	2152002	19
22	[Ge ^{II} ₄ (TBC[8])][Fe ⁰ ₂ (CO) ₈]	XUGKOA	709989	20
23	[Ge ^{IV} ₄ (C[8])][Fe ^{-I} ₂ (CO) ₈] ₄	LIPHAU	665344	21
24	[Fe ^{III} ₄ K ₄ O ₂ (H ₂ O) ₂ (H ₂ TBC[8]) ₂ (MeCN) ₆]	GUQZOI	749033	22
25	[Mn ^{III} ₂ Na ₄ (H ₅ TBC[8]) ₂ (CO ₃) ₂ (dmf) ₆]	BAKDIC	824374	23
26	[Mn ^{III} Mn ^{IV} (H ₄ TBC[8])(Ph-sao)(μ-OMe)(dmf) ₂]	BEYHOE	914769	24
27	[Mn ^{II} ₅ (H ₄ TBC[8])(OH) ₂ (OMe) _{1.5} (HCOO)(OAc) _{0.5} (O ₂ CNMe ₂)(dmf) ₅]	XIPHOV	888020	25
28	[Ni ^{II} ₂ Th ^{IV} ₅ O ₄ (OH) ₂ (H ₁ TBC[8]) ₂ (dmf) ₅ (MeOH) ₂]	JIYSOB	948470	18
29a	[Cr ^{III} Gd ^{III} ₆ O ₂ (OH)(TBC[8]) ₂ (H ₂ O) ₄ (dmf) ₅ (MeOH) ₂]	HANWAX	1517454	27
29b	[Cr ^{III} Tb ^{III} ₆ O ₂ (OH)(TBC[8]) ₂ (H ₂ O) ₄ (dmf) ₅ (MeOH) ₂]	HANWOL	1517457	27
29c	[Cr ^{III} Dy ^{III} ₆ O ₂ (OH)(TBC[8]) ₂ (H ₂ O) ₄ (dmf) ₅ (MeOH) ₂]	HANWIF	1517456	27
29d	[Cr ^{III} Tm ^{III} ₆ O ₂ (OH)(TBC[8]) ₂ (H ₂ O) ₄ (dmf) ₅ (MeOH) ₂]	HANWEB	1517455	27
30	[Eu ^{III} ₂ (H ₂ TBC[8])(dmf) ₅]	FUVVEX	1161880	28
31	[Eu ^{III} (H ₆ TBC[8])(NO ₃)(dmf) ₄]	SOGDOH	1261238	33
32	[Gd ^{III} (H ₆ TBC[8])Cl(dmso) ₄]	VESVEW	892672	34
33	[Ce ^{IV} ₄ O ₂ (H ₂ TBC[8]) ₂ (dmf) ₄]	VESVIA	892673	34
34	[Tb ^{III} ₅ O(OH) ₄ (H ₃ TBC[8])Cl(dmso) ₈ (H ₂ O) ₃]Cl ₃	VESVOG	892674	34
35	[Ce ^{IV} ₆ O ₄ (H ₂ TBC[8]) ₂ (OMe) ₄ (dmf) ₆]	VESVUM	892675	34
36	[Dy ^{III} ₇ O ₂ (OH) ₄ (H ₁ TBC[8])(H ₂ TBC[8])(dmf) ₉]	VESWAT	892676	34
37	[NaDy ^{III} (H ₆ TBC[8])(OAc) ₂ (dmf) ₃]	JOJJAW	1902422	35
38	[Gd ^{III} ₈ (H ₁ TBC[8]) ₂ (CO ₃) ₄ (HCOO) ₂ (dmf) ₈]	VESWEX	892677	34
39a	[La ^{III} ₁₈ O ₃ (OH) ₁₂ (TBC[8]) ₃ (CO ₃) ₂ Cl ₆ (H ₂ O) ₆ (dmf) ₁₈ Cl]OH	KOFWEL	2299004	37
39b	[Nd ^{III} ₁₈ O ₃ (OH) ₁₂ (TBC[8]) ₃ (CO ₃) ₂ Cl ₆ (H ₂ O) ₆ (dmf) ₁₈ Cl]OH	-	-	37
39c	[Gd ^{III} ₁₈ O ₃ (OH) ₁₂ (TBC[8]) ₃ (CO ₃) ₂ Cl ₆ (H ₂ O) ₆ (dmf) ₁₈ Cl]OH	-	-	37

Table S1. A list of all known *p*-tert-butylcalix[8]arene-supported transition (3d limited to V – Cu) and lanthanide metal complexes in the literature. The table provides the number of the complex in this review, the formula of each complex, the relevant Conquest code, CCDC number and reference in the main text.

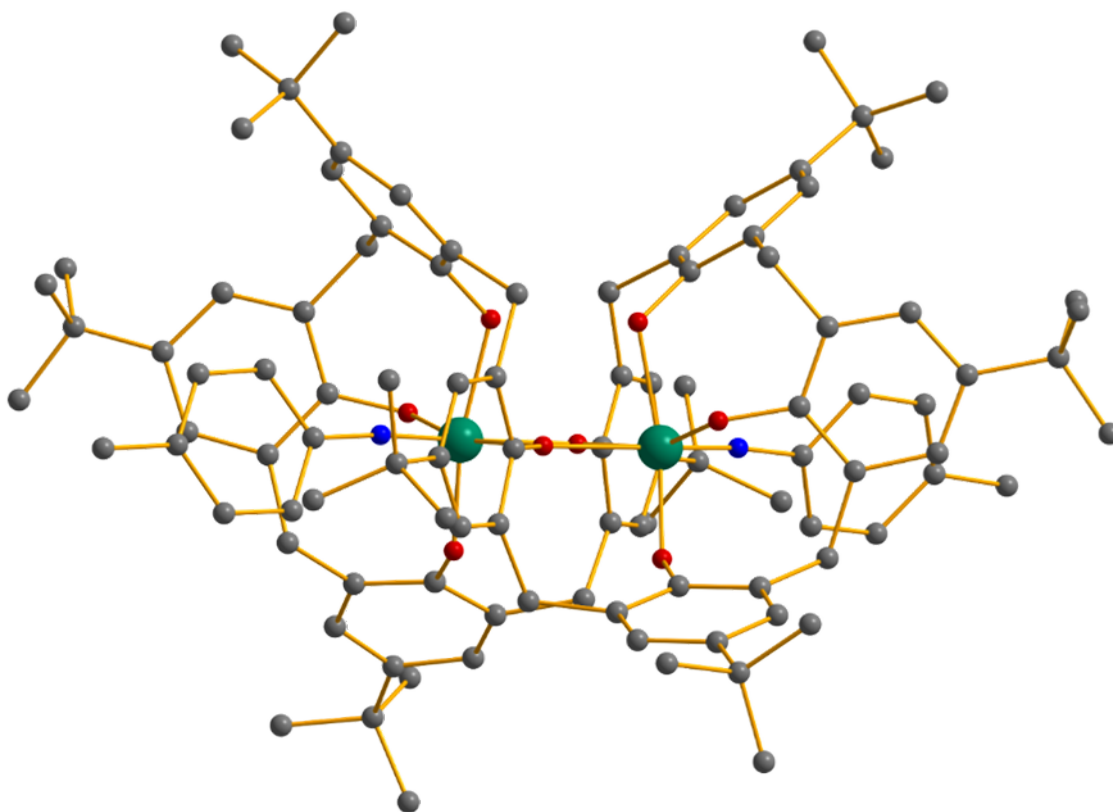


Figure S1. Structure of the [V₂] dimer in **2**. Colour code: V = dark green, O = red, N = blue, C = grey. H atoms, counter ions omitted for clarity.

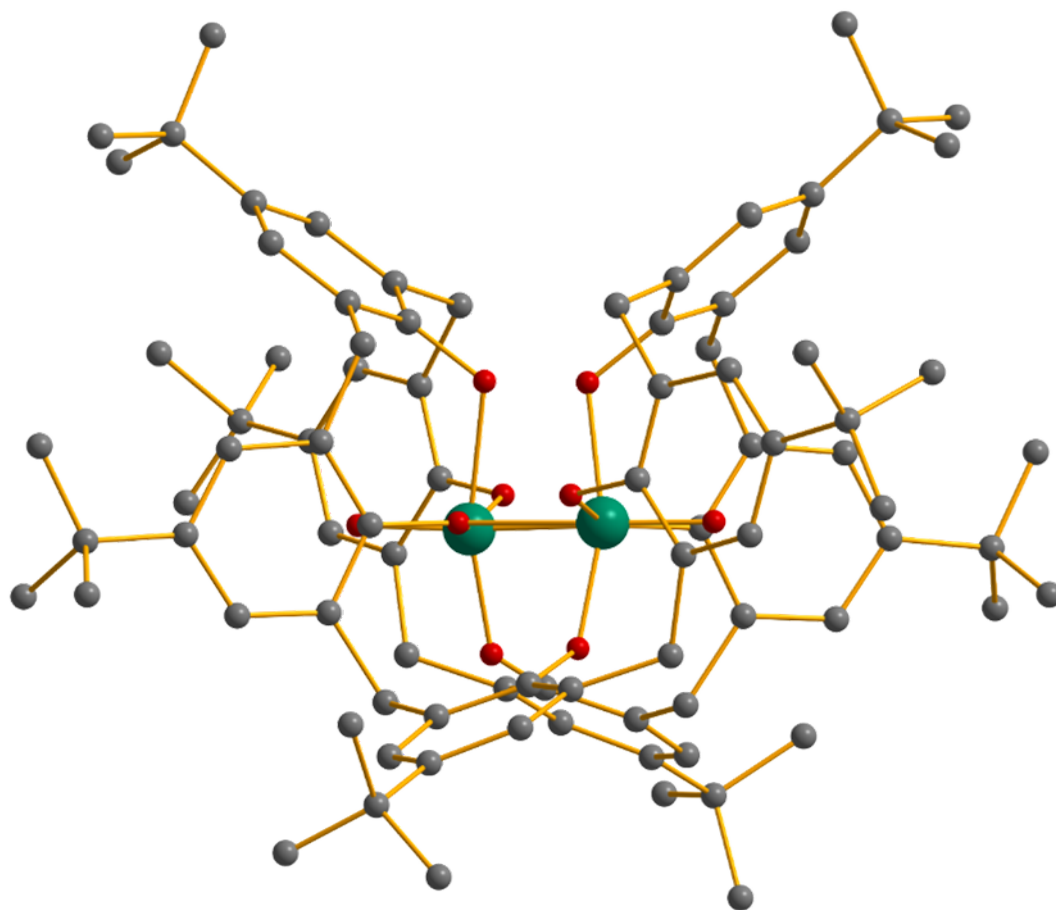


Figure S2. Structure of the [V₂] dimer present in **3-5**. Colour code: V = dark green, O = red, C = grey. H atoms, counter ions omitted for clarity.

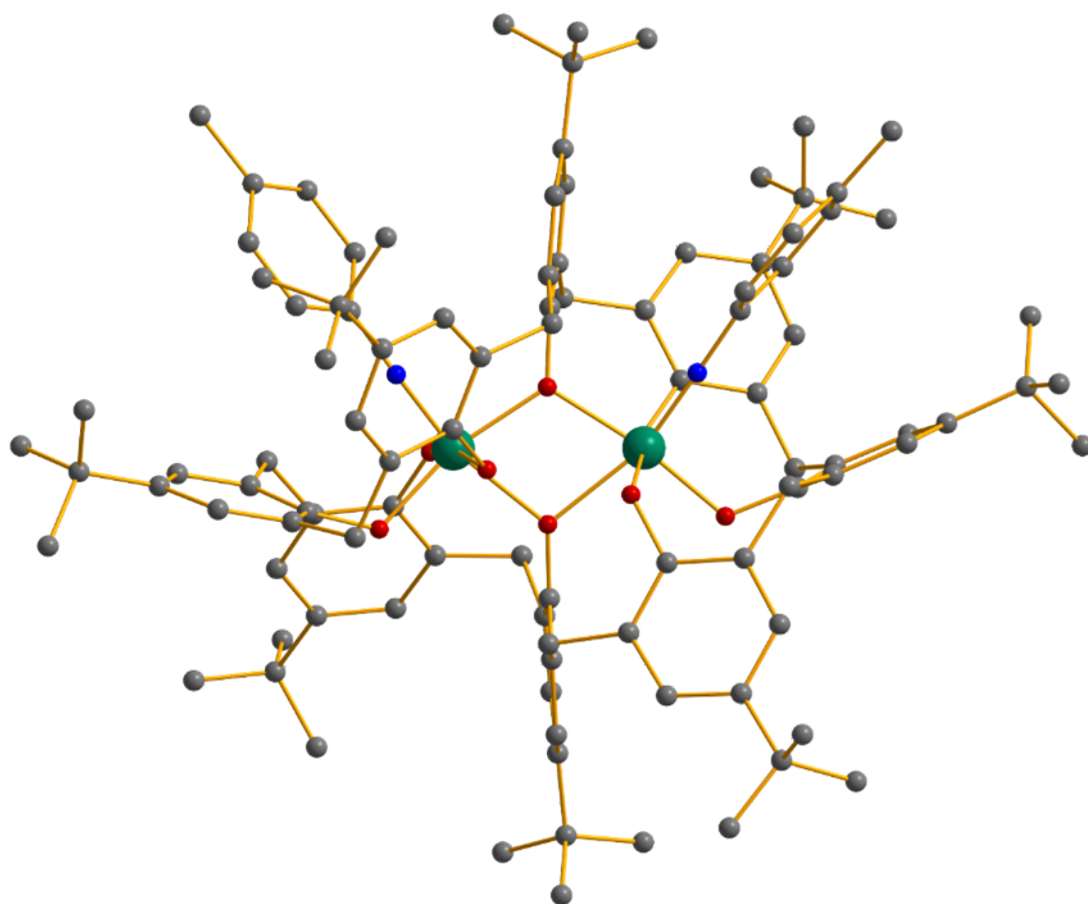


Figure S3. Structure of the [V₂] dimer present in **6**. Colour code: V = dark green, O = red, N = blue, C = grey. H atoms, counter ions omitted for clarity.

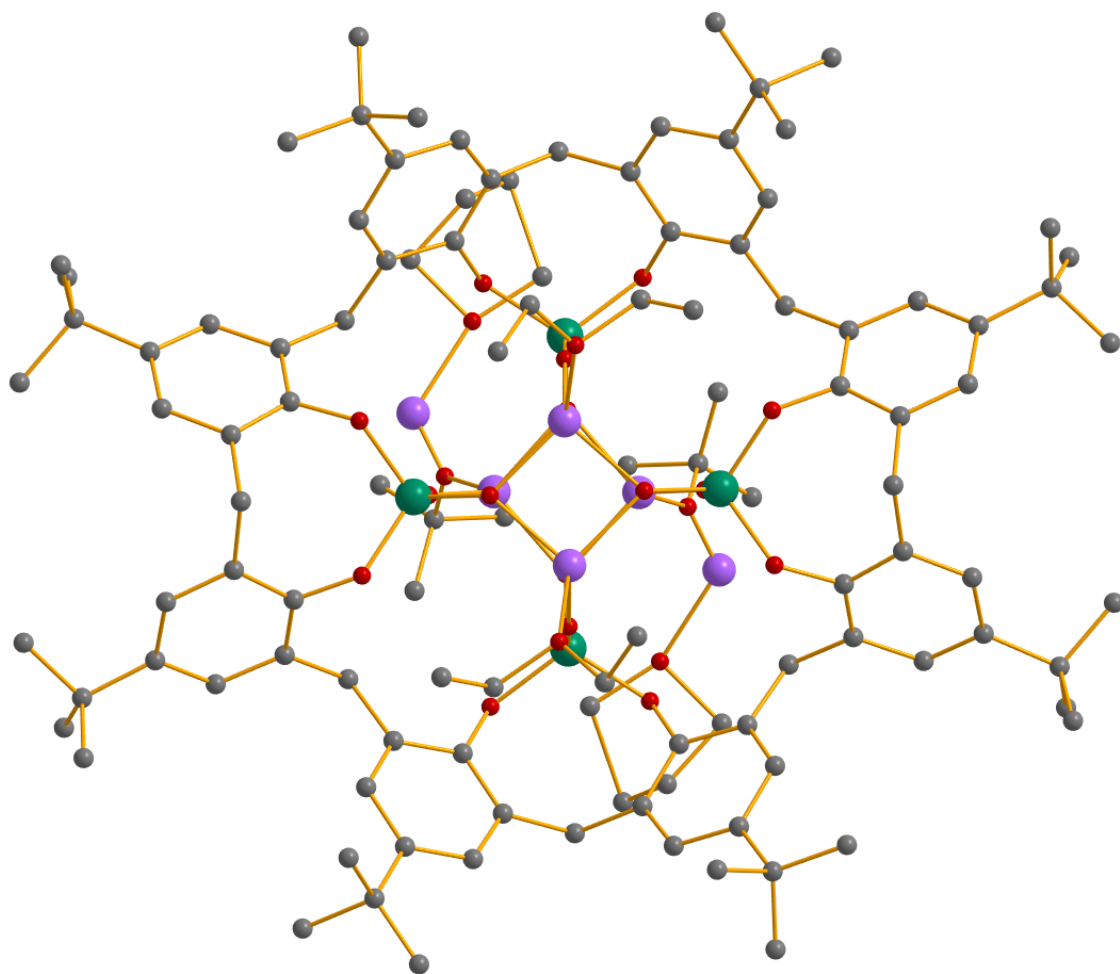


Figure S4. Structure of the [V₄Li₆] cluster in **9**. Colour code: V = dark green, Li = light purple, O = red, C = grey. H atoms omitted for clarity.

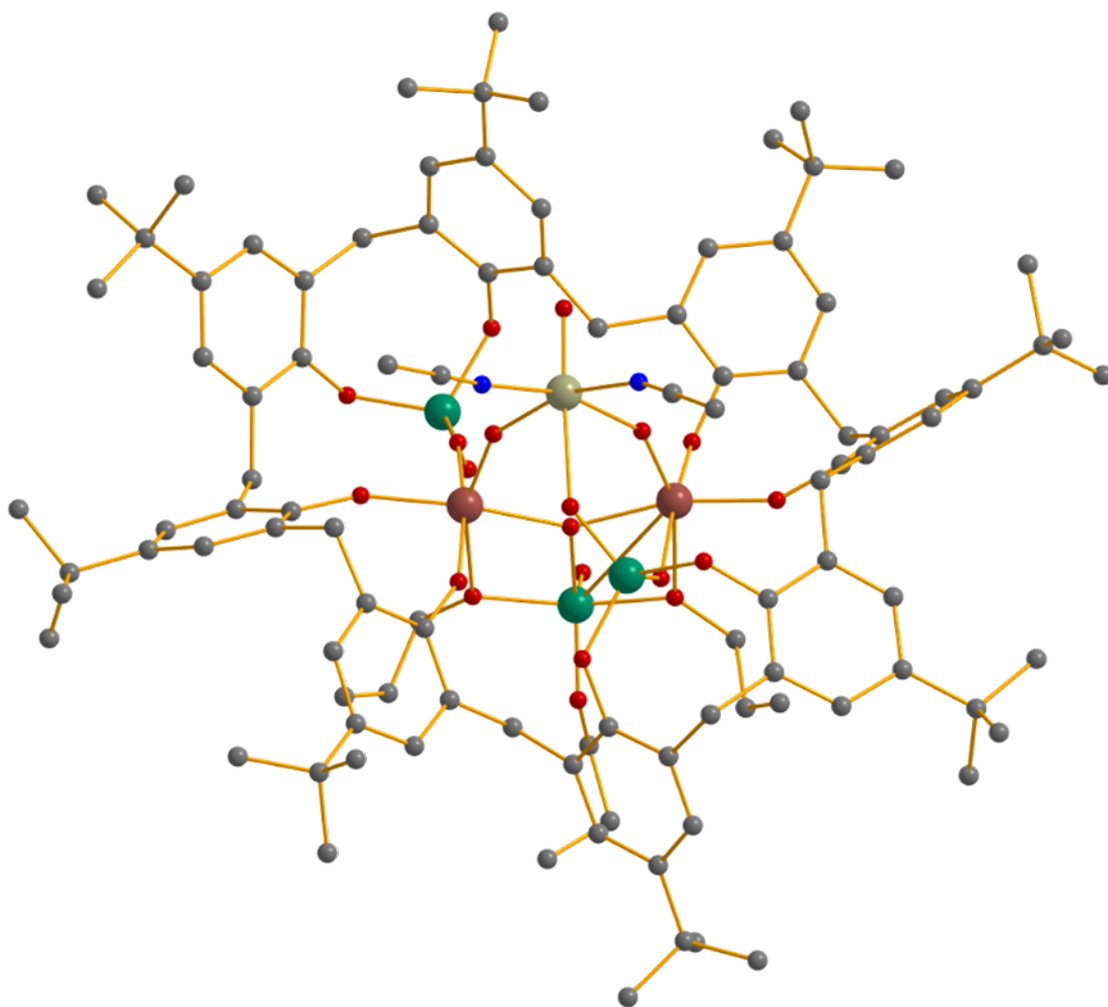


Figure S5. Structure of compound **12**. Colour code: V^V = dark green, V^{IV} = sage green, Nb = maroon, O = red, N = blue, C = grey. H atoms omitted for clarity.

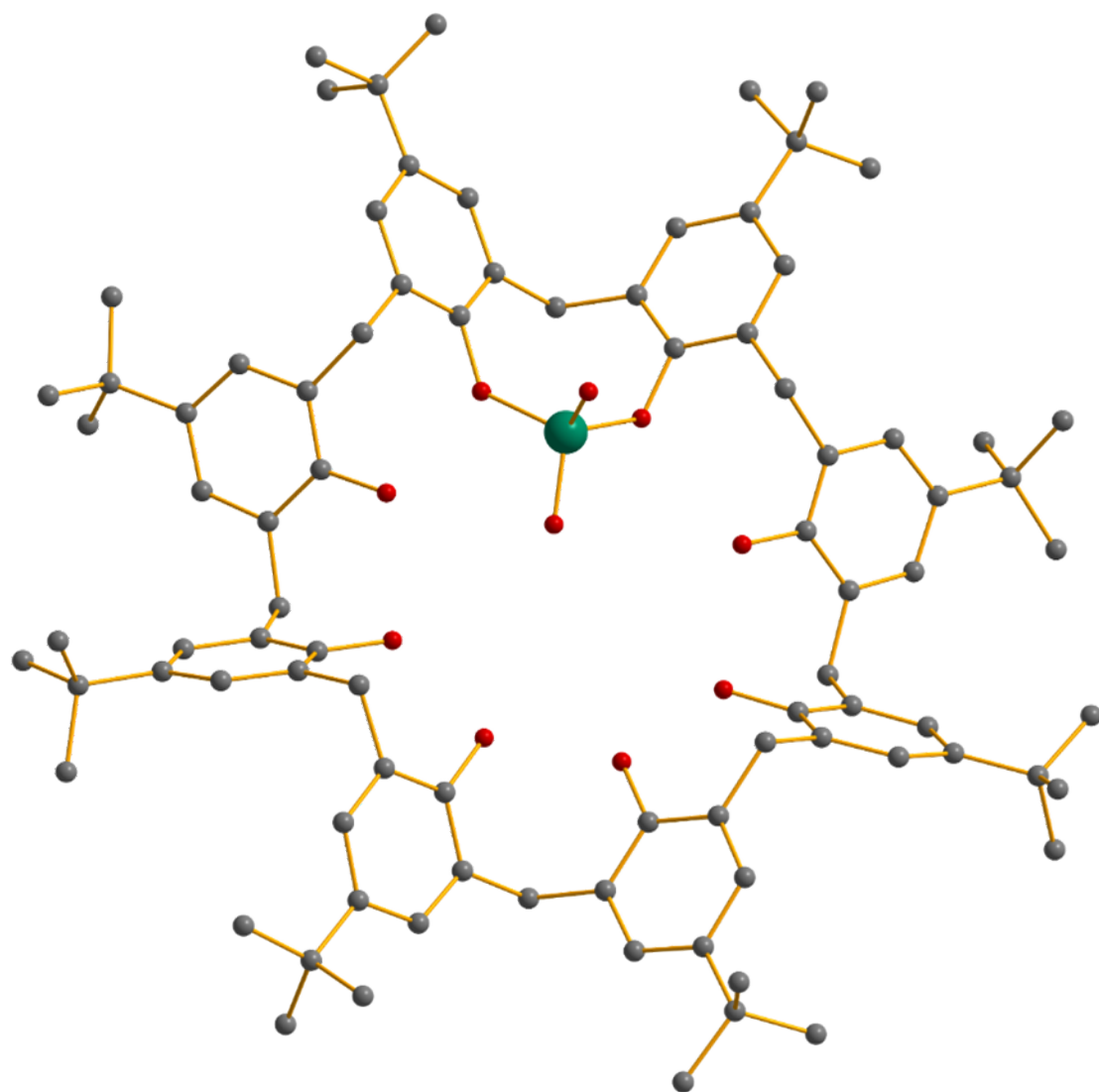


Figure S6. Structure of compound **13**. Colour code: V = dark green, O = red, C = grey. H atoms, counter ions omitted for clarity.

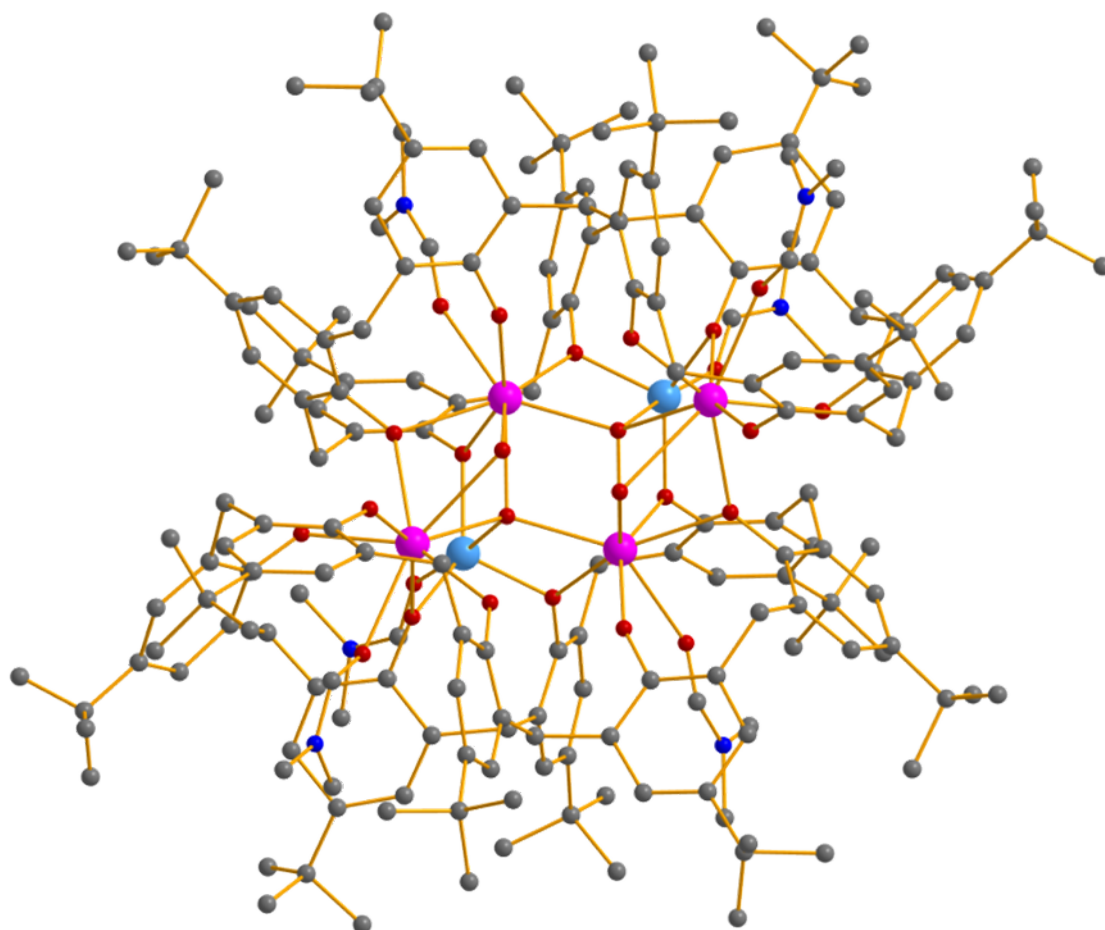


Figure S7. Structure of compound **18**. Colour code: Co = sky blue, Th = deep pink, O = red, N = blue, C = grey. H atoms omitted for clarity.

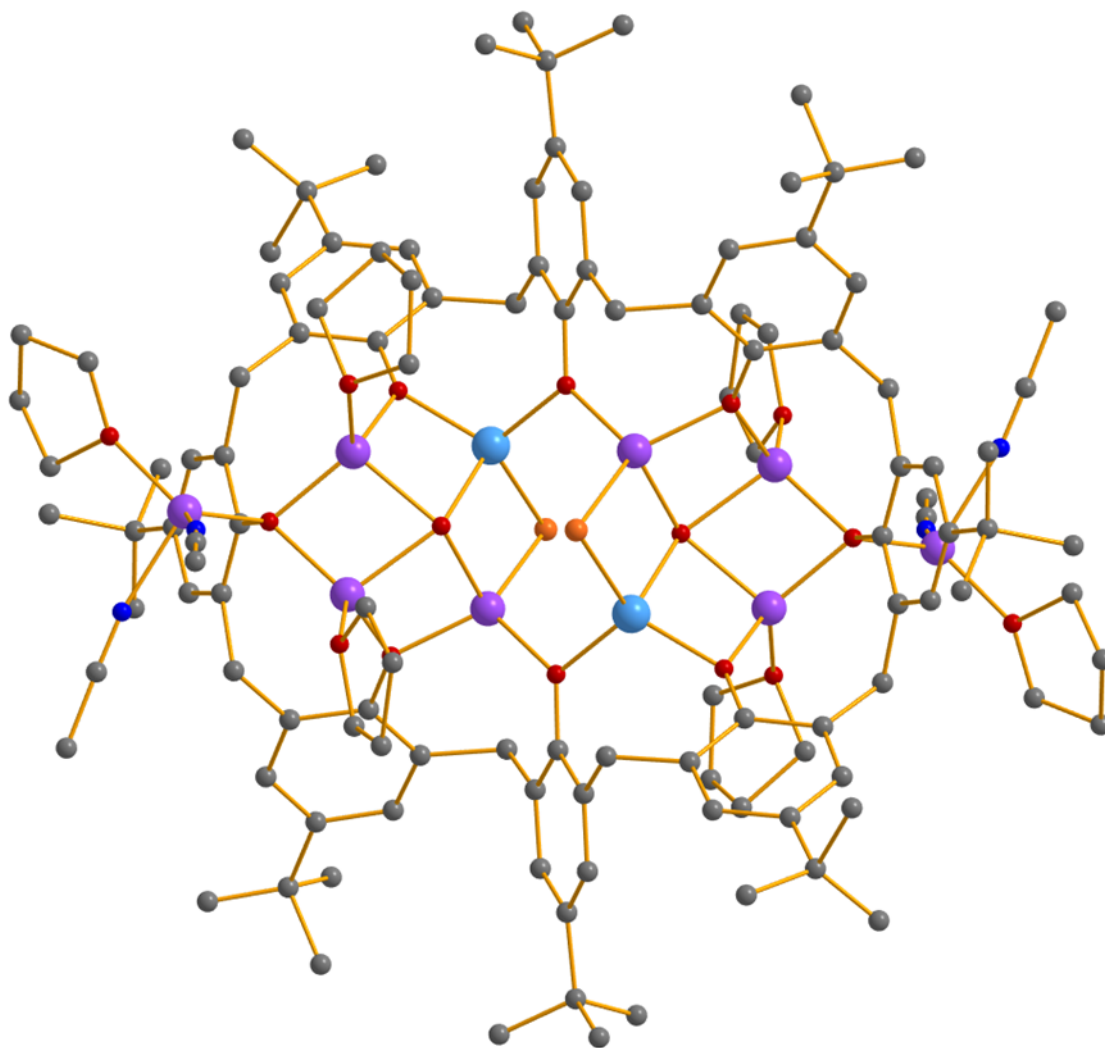


Figure S8. Structure of compound **19**. Colour code: Co = sky blue, Li = light purple, O = red, N = blue, C = grey, Br = orange. H atoms omitted for clarity.

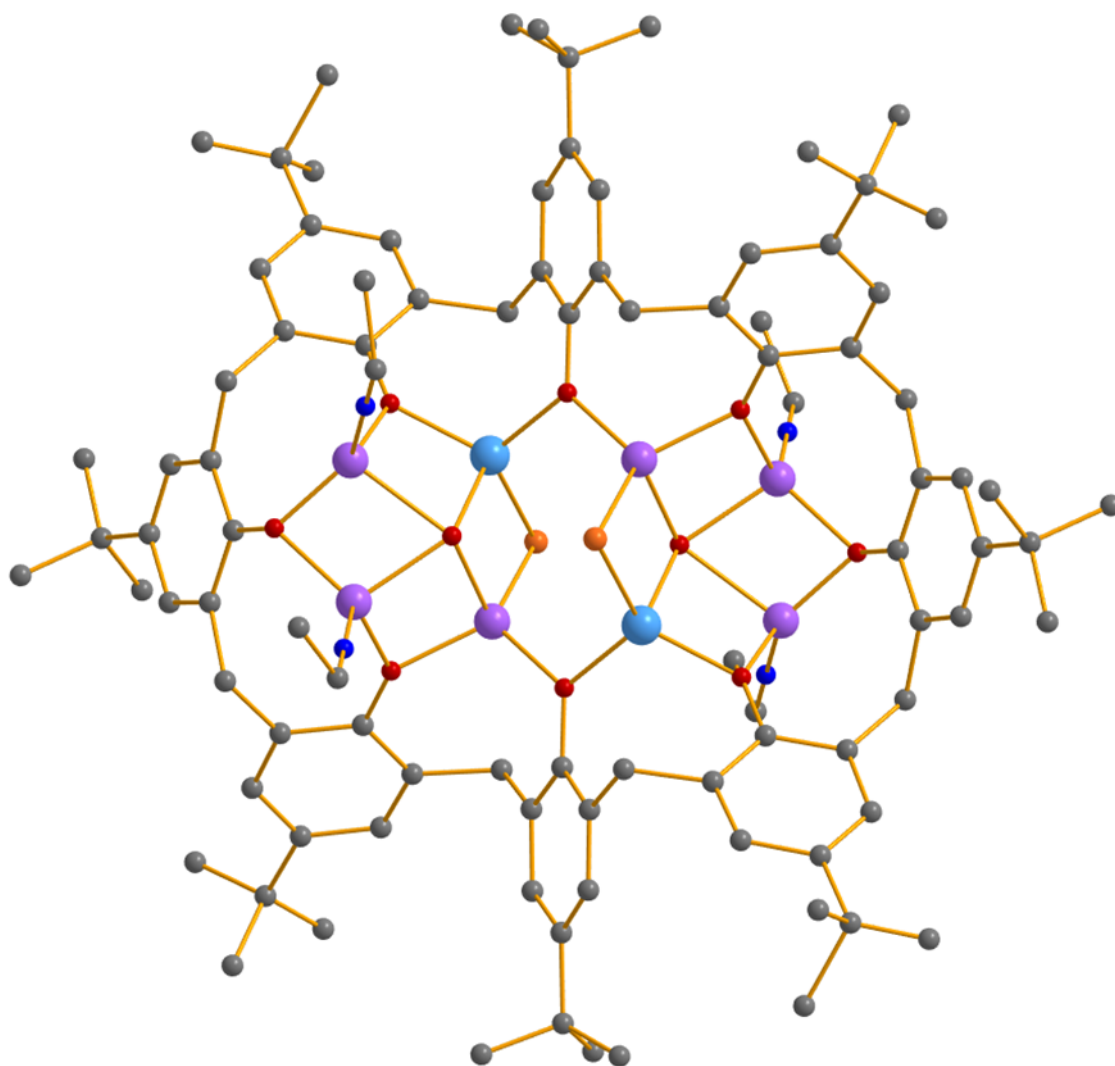


Figure S9. Structure of compound **20**. Colour code: Co = sky blue, Li = light purple, O = red, N = blue, C = grey, Br = orange. H atoms omitted for clarity.

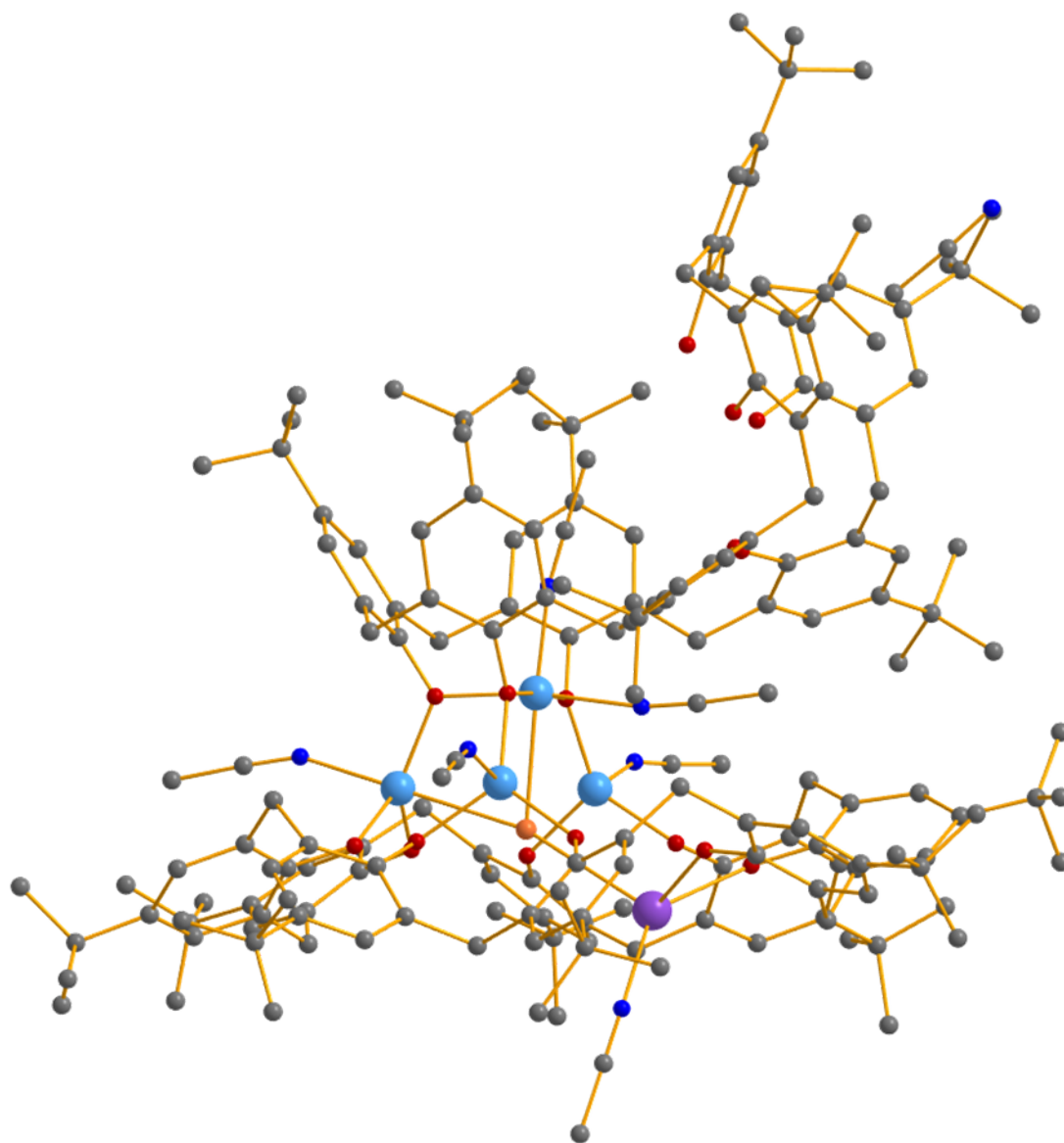


Figure S10. Structure of compound **21**. Colour code: Co = sky blue, Na = purple, O = red, N = blue, C = grey, Br = orange. H atoms omitted for clarity.

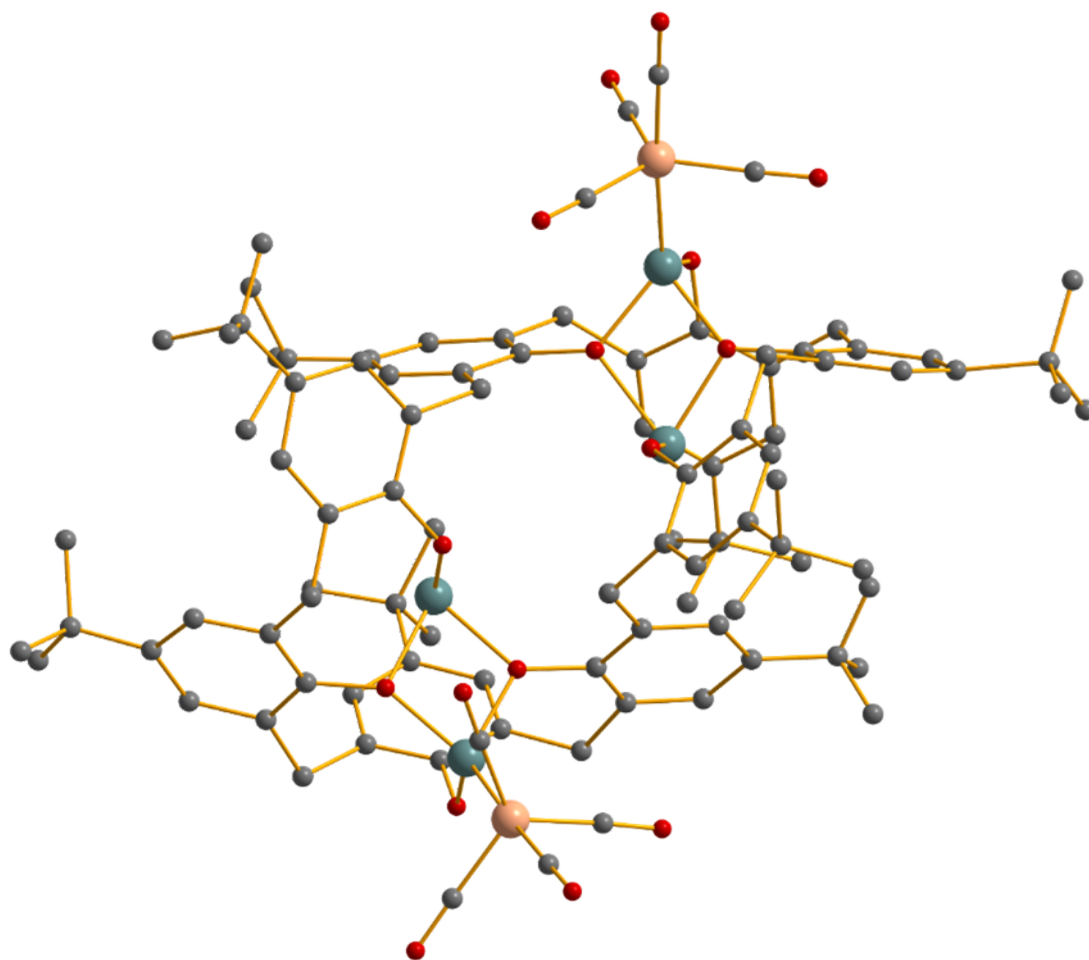


Figure S11. Structure of compound **22**. Colour code: Fe = pale orange, Ge = teal, O = red, C = grey. H atoms omitted for clarity.

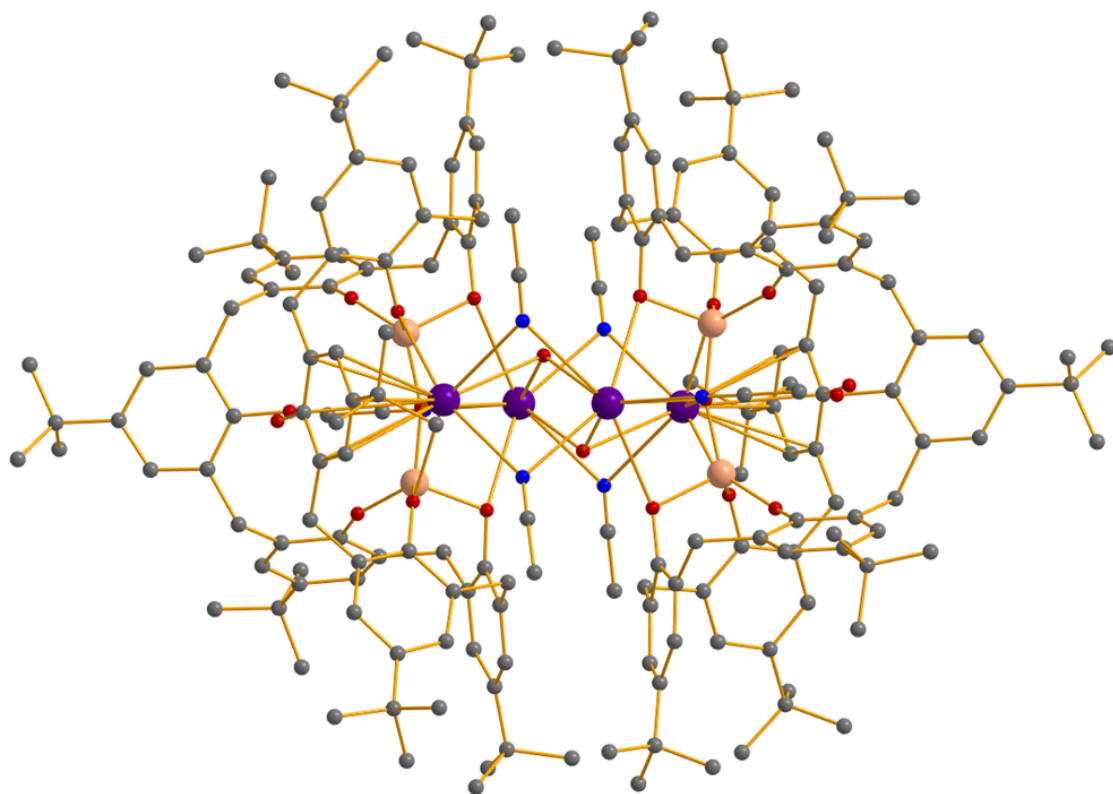


Figure S12. Structure of compound **24**. Colour code: Fe = pale orange, K = deep purple, O = red, N = blue, C = grey. H atoms omitted for clarity.

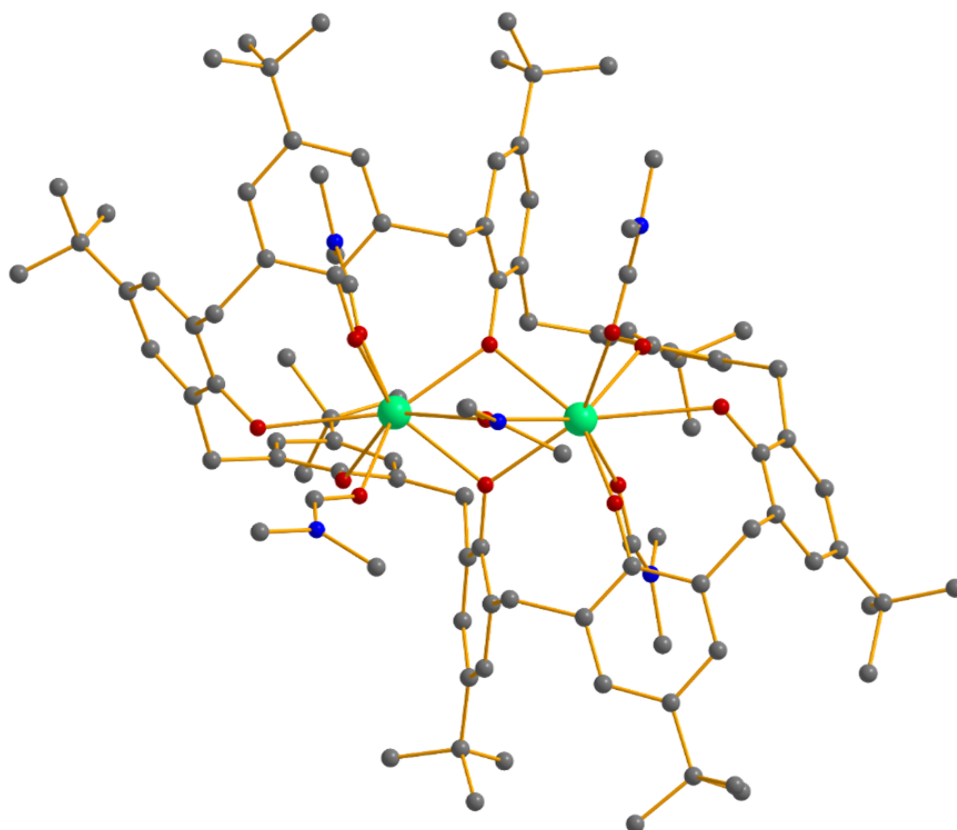


Figure S13. Structure of compound **30**. Colour code: Ln = light green, O = red, N = blue, C = grey. H atoms omitted for clarity.

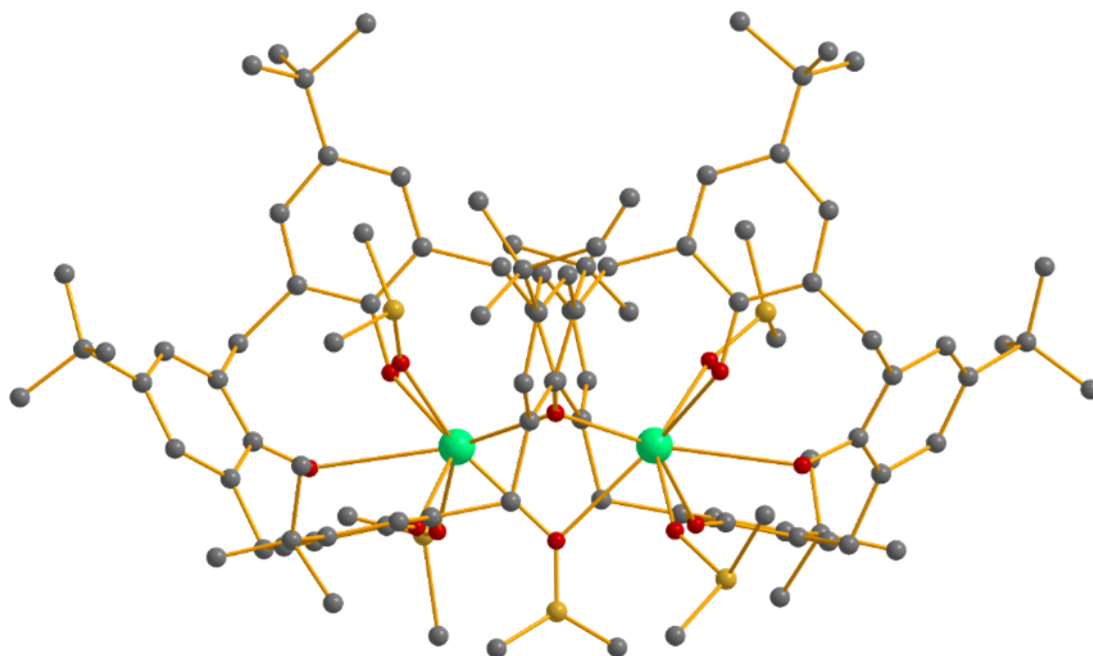


Figure S14. Structure of compound **30** with dmsO replacing dmF. Colour code: Ln = light green, O = red, C = grey, S = dark yellow. H atoms omitted for clarity.

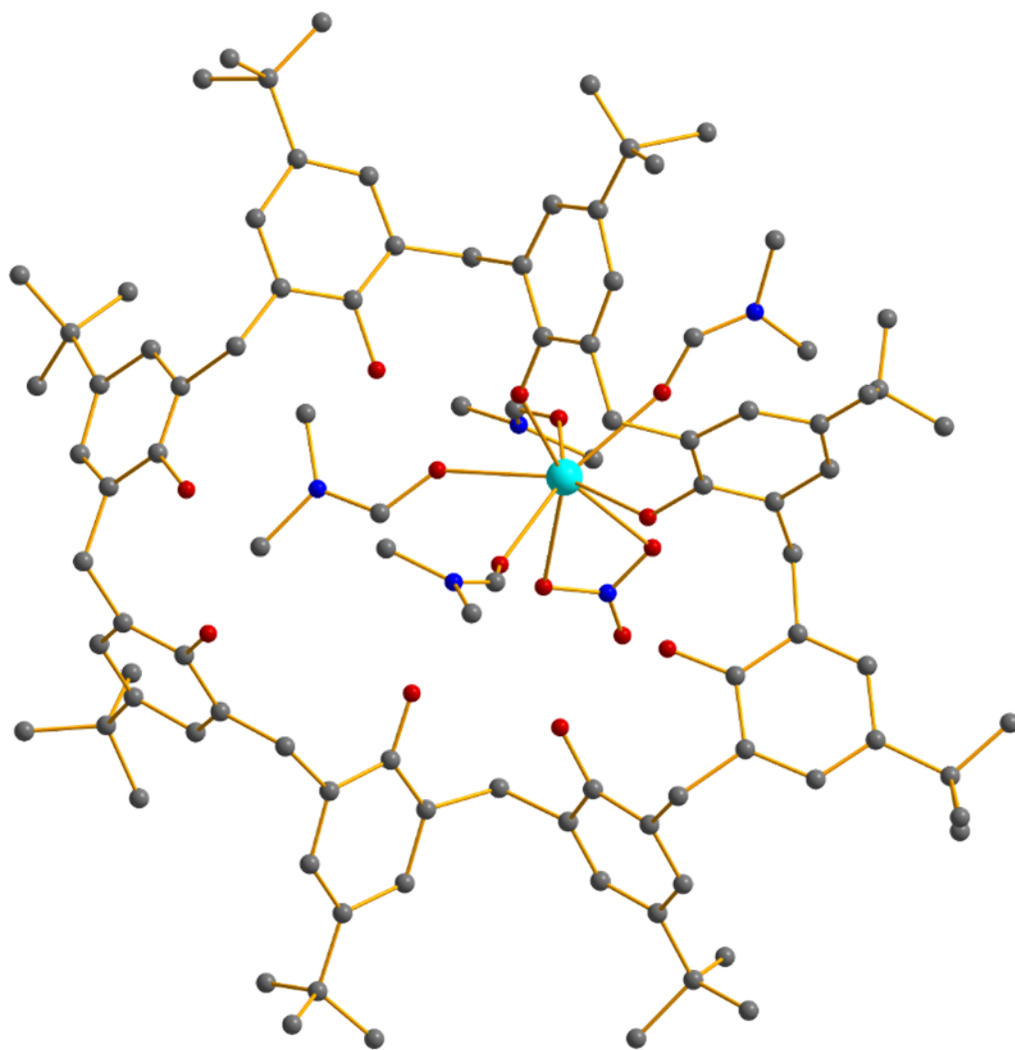


Figure S15. Structure of compound **31**. Colour code: Eu = light blue, O = red, N = blue, C = grey. H atoms omitted for clarity.

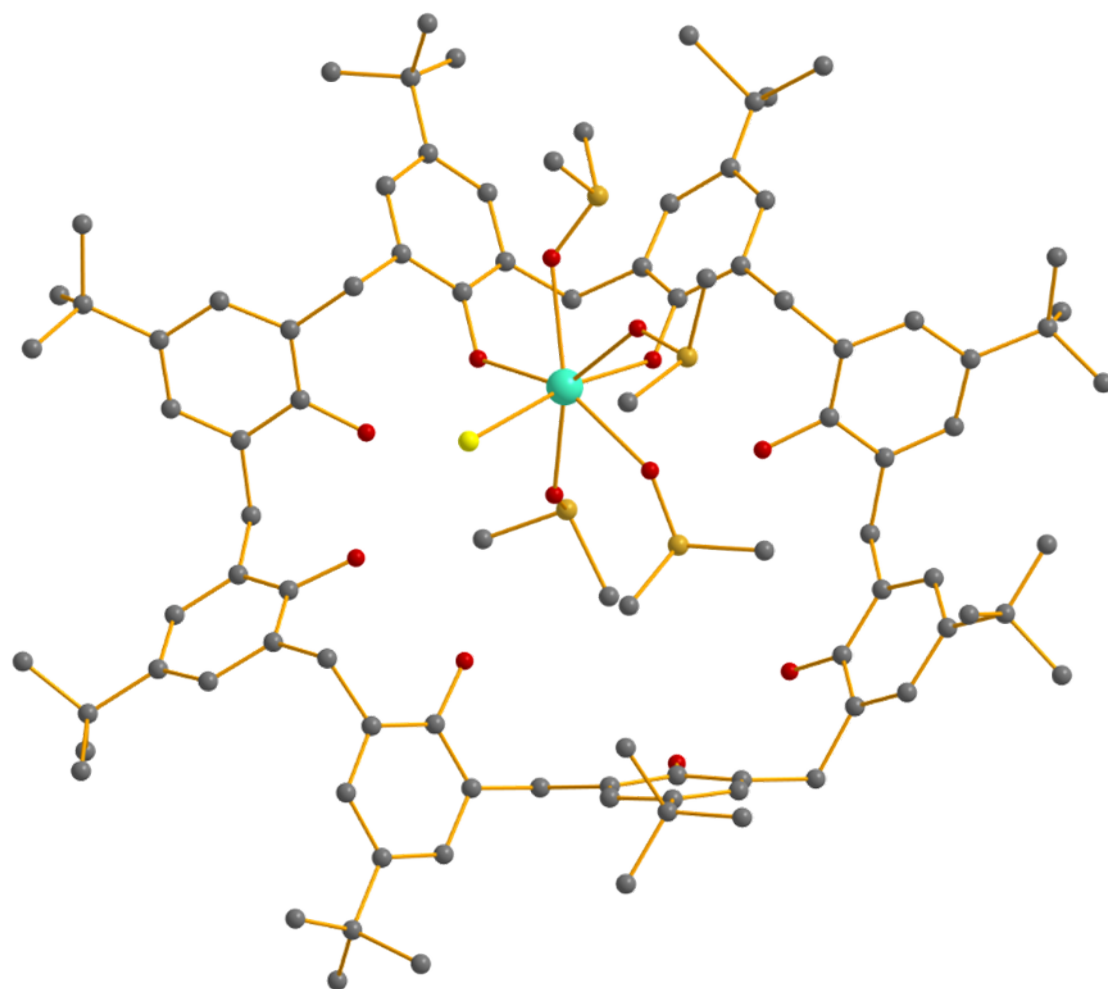


Figure S16. Structure of compound **32**. Colour code: Gd = cyan, O = red, C = grey, S = dark yellow, Cl = yellow. H atoms omitted for clarity.

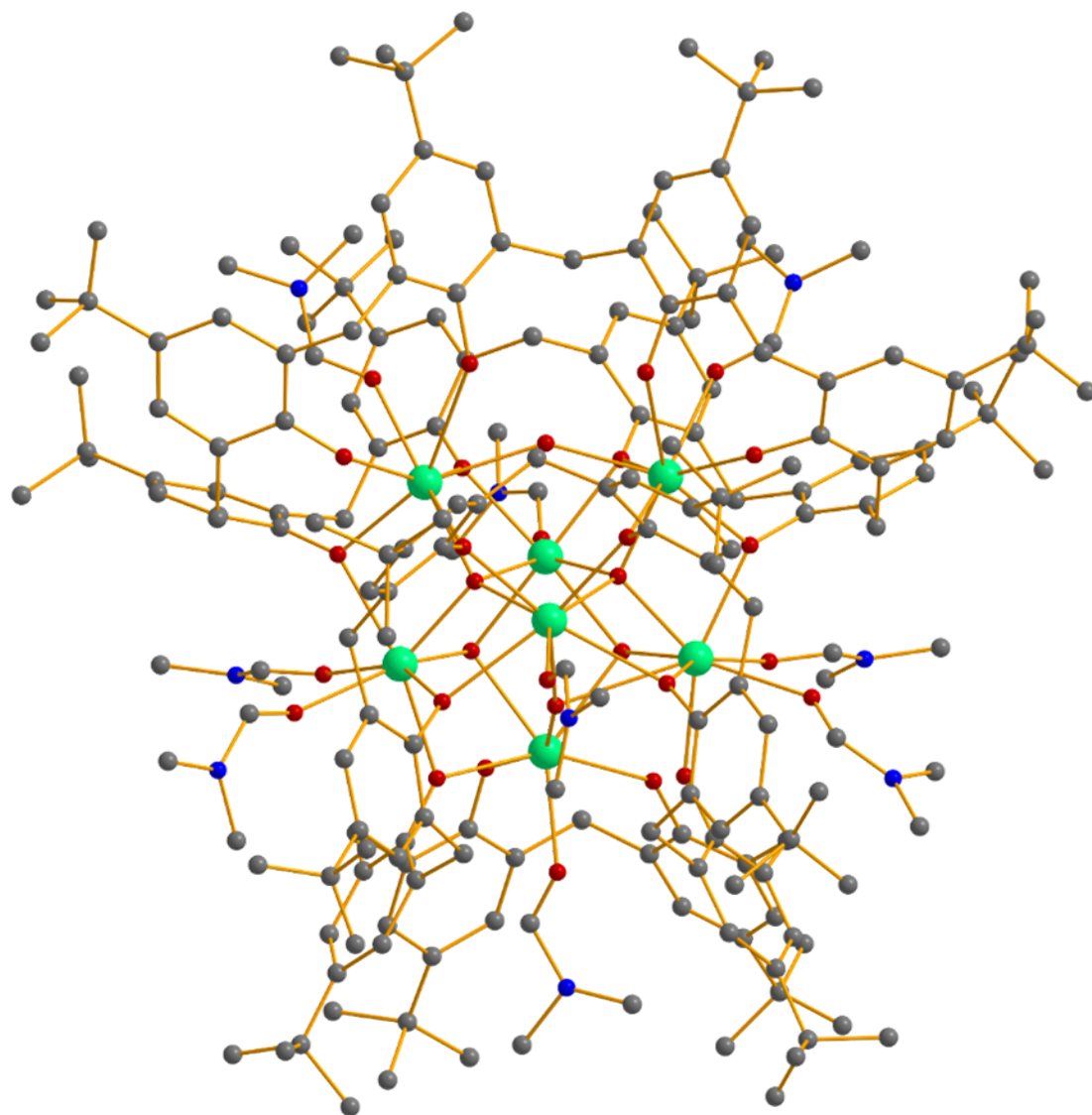


Figure S17. Structure of compound **36**. Colour code: Dy = green, O = red, N = blue, C = grey. H atoms omitted for clarity.

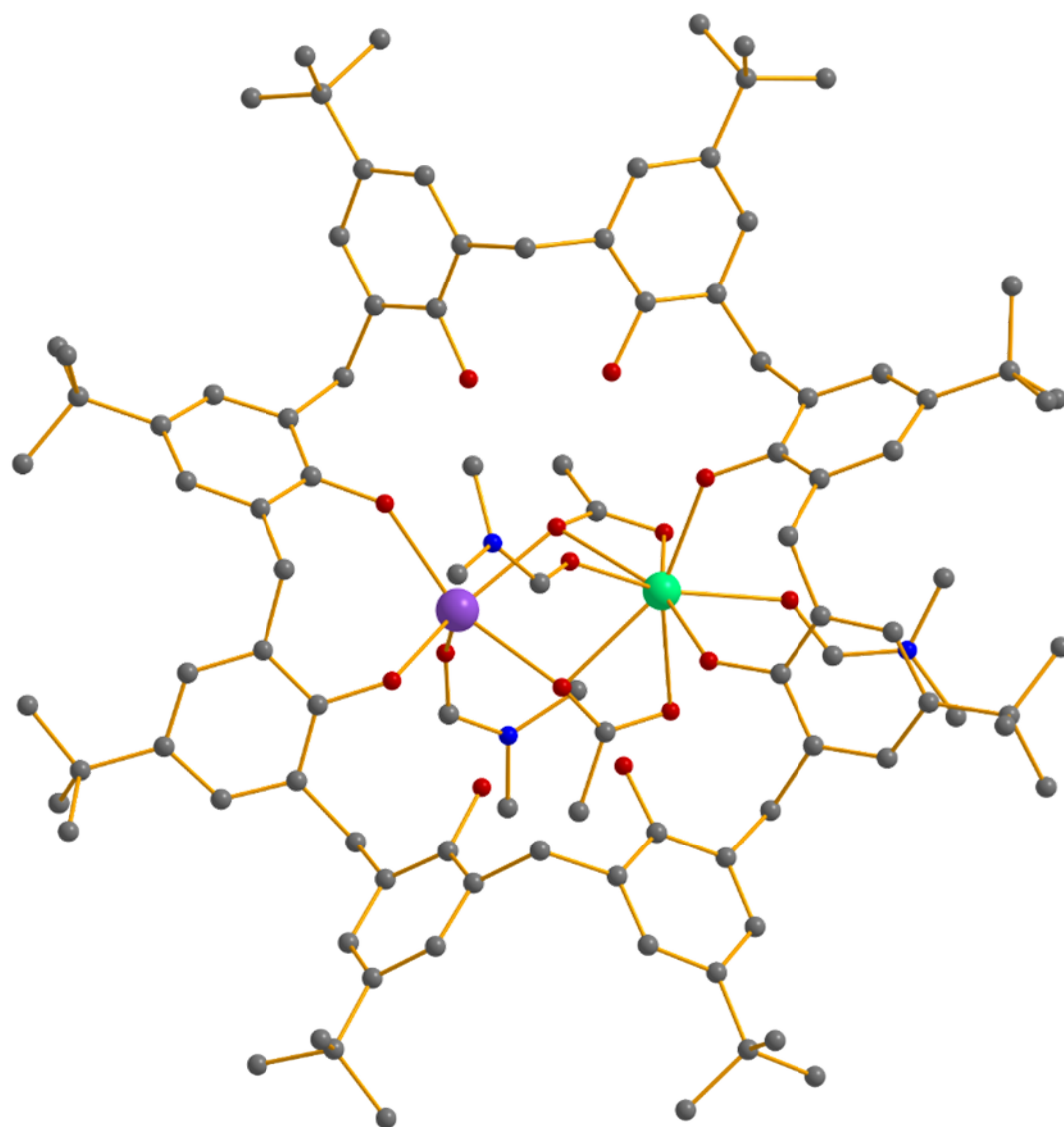


Figure S18. Structure of compound **37**. Colour code: Dy = green, Na = purple, O = red, N = blue, C = grey. H atoms omitted for clarity.

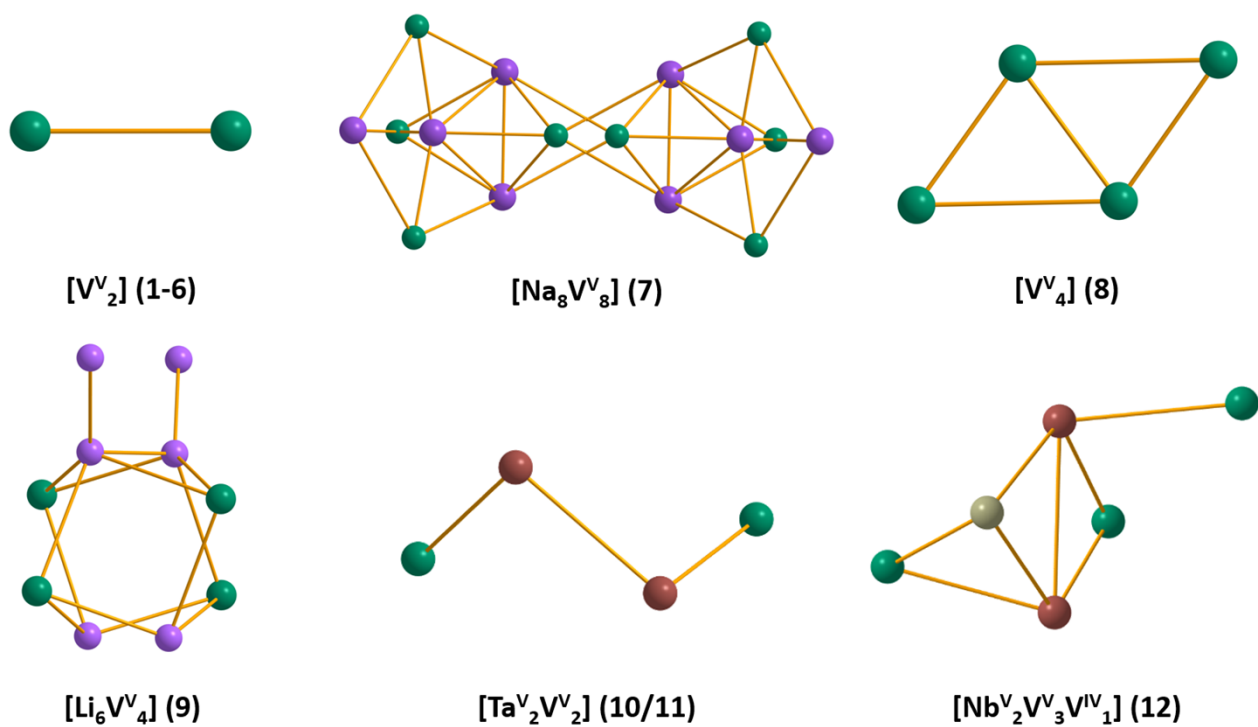


Figure S19. Metallic skeletons of the polymetallic V-based clusters in the review.

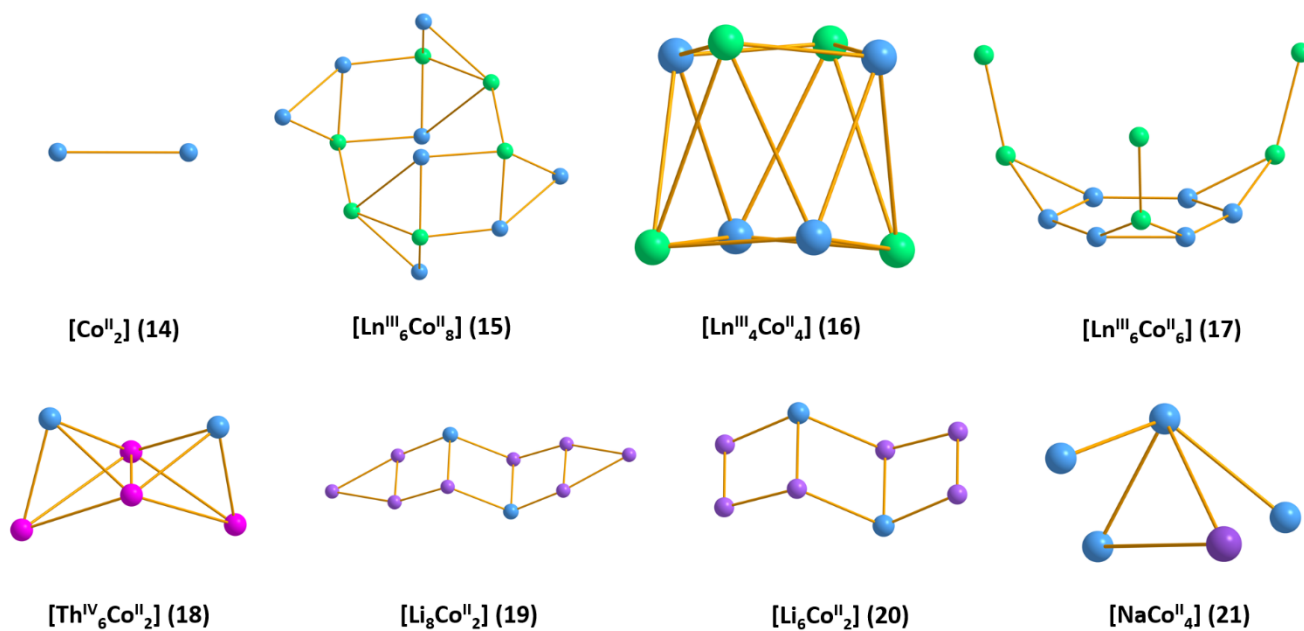


Figure S20. Metallic skeletons of the Co-based clusters in the review.

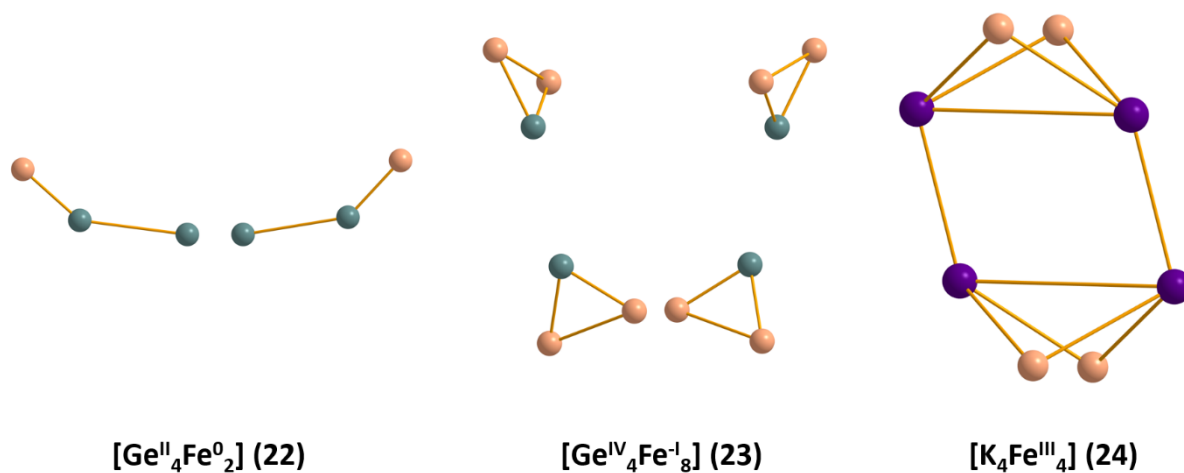


Figure S21. Metallic skeletons of the Fe-based clusters in the review.

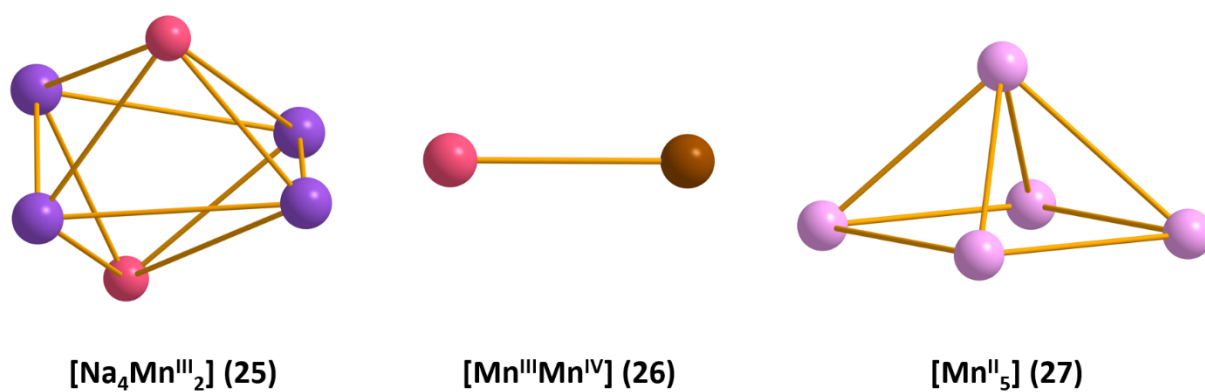


Figure S22. Metallic skeletons of the Mn-based clusters in the review.

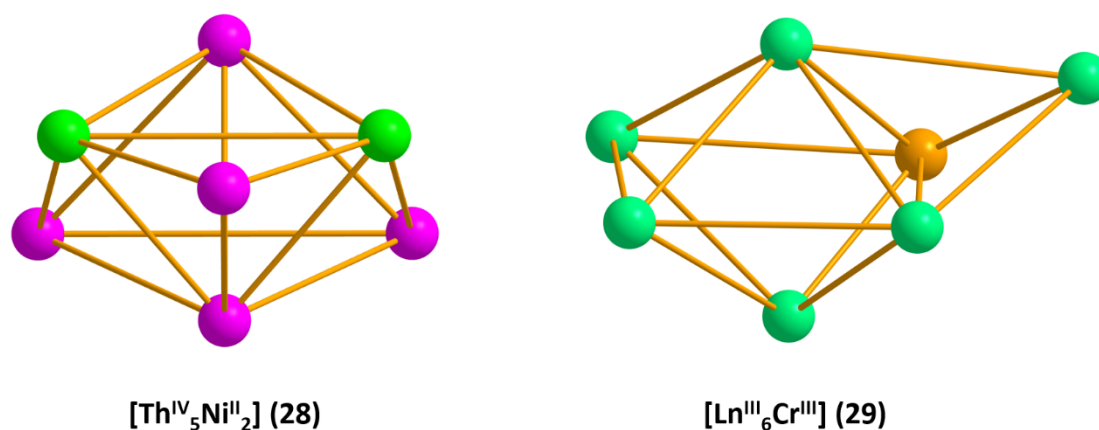


Figure S23. Metallic skeletons of the polymetallic Ni- and Cr-based clusters in the review.

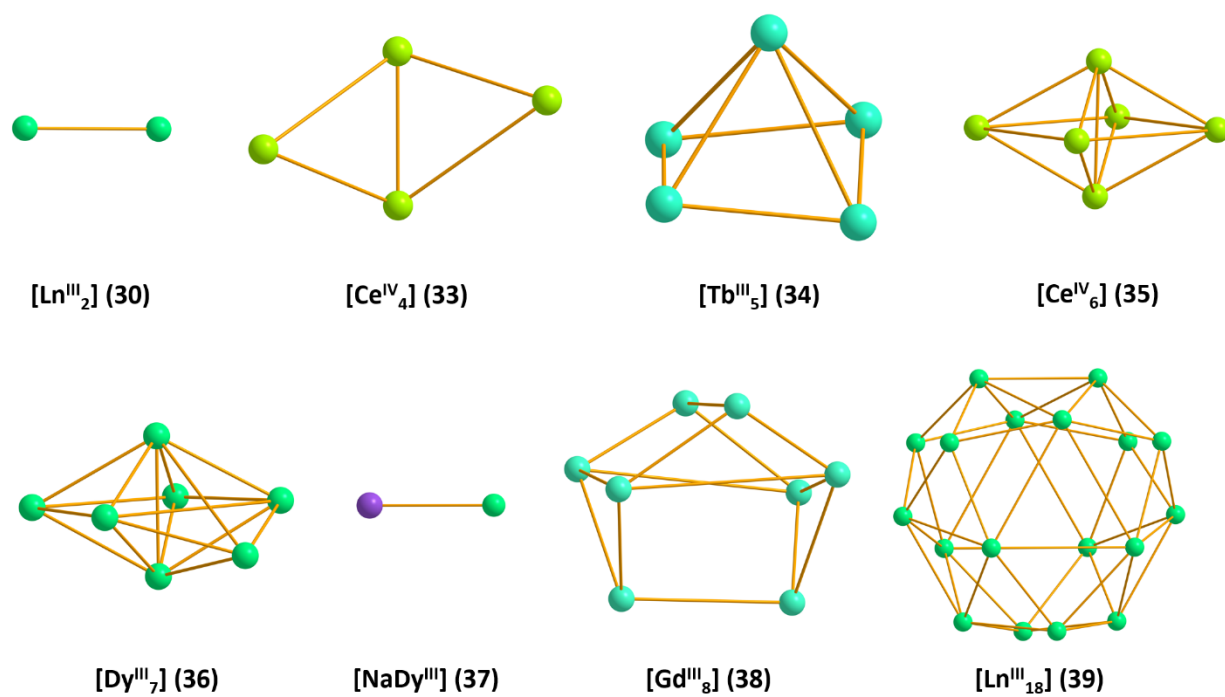


Figure S24. Metallic skeletons of the polynuclear Ln-based clusters in the review.