

*Supporting information*

**Gefitinib salt/cocrystal with phenolic acids as promising solid-state approach to improve solubility**

Yao Zou,<sup>a</sup> Xin Meng,<sup>a</sup> Baoxi Zhang,<sup>a</sup> Hongmei Yu,<sup>a</sup> Guorong He,<sup>b</sup> Ningbo Gong,<sup>\*a</sup> Yang Lu,<sup>\*a</sup> and Guanhua Du<sup>b</sup>

<sup>a</sup> Beijing Key Laboratory of Polymorphic Drugs, Institute of Materia Medica, Chinese Academy of Medical Sciences and Peking Union Medical College, Beijing 100050, China

<sup>b</sup> Beijing City Key Laboratory of Drug Target Identification and Drug Screening, Institute of Materia Medica, Chinese Academy of Medical Sciences and Peking Union Medical College, Beijing 100050, China.

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Table S1. Hydrogen bond geometrical parameters of crystal structures.

Fig. S1 The DSC thermograms of CCFs.

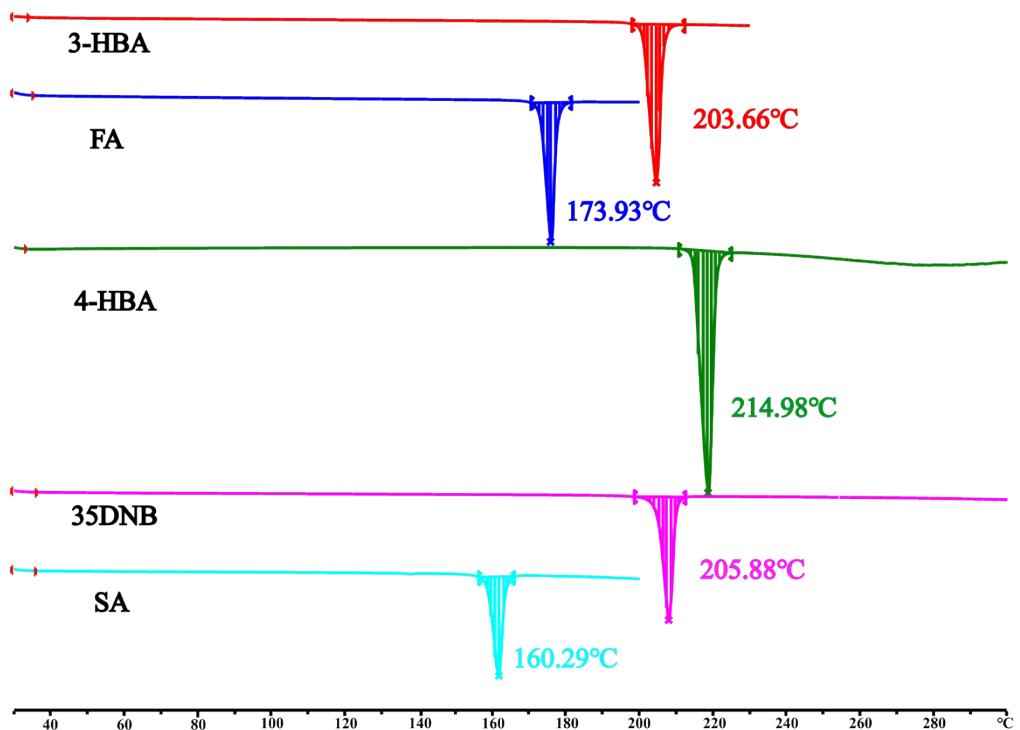
Fig. S2 FT-IR spectra of GEF, CCFs, and the corresponding synthesized salts/cocrystals.

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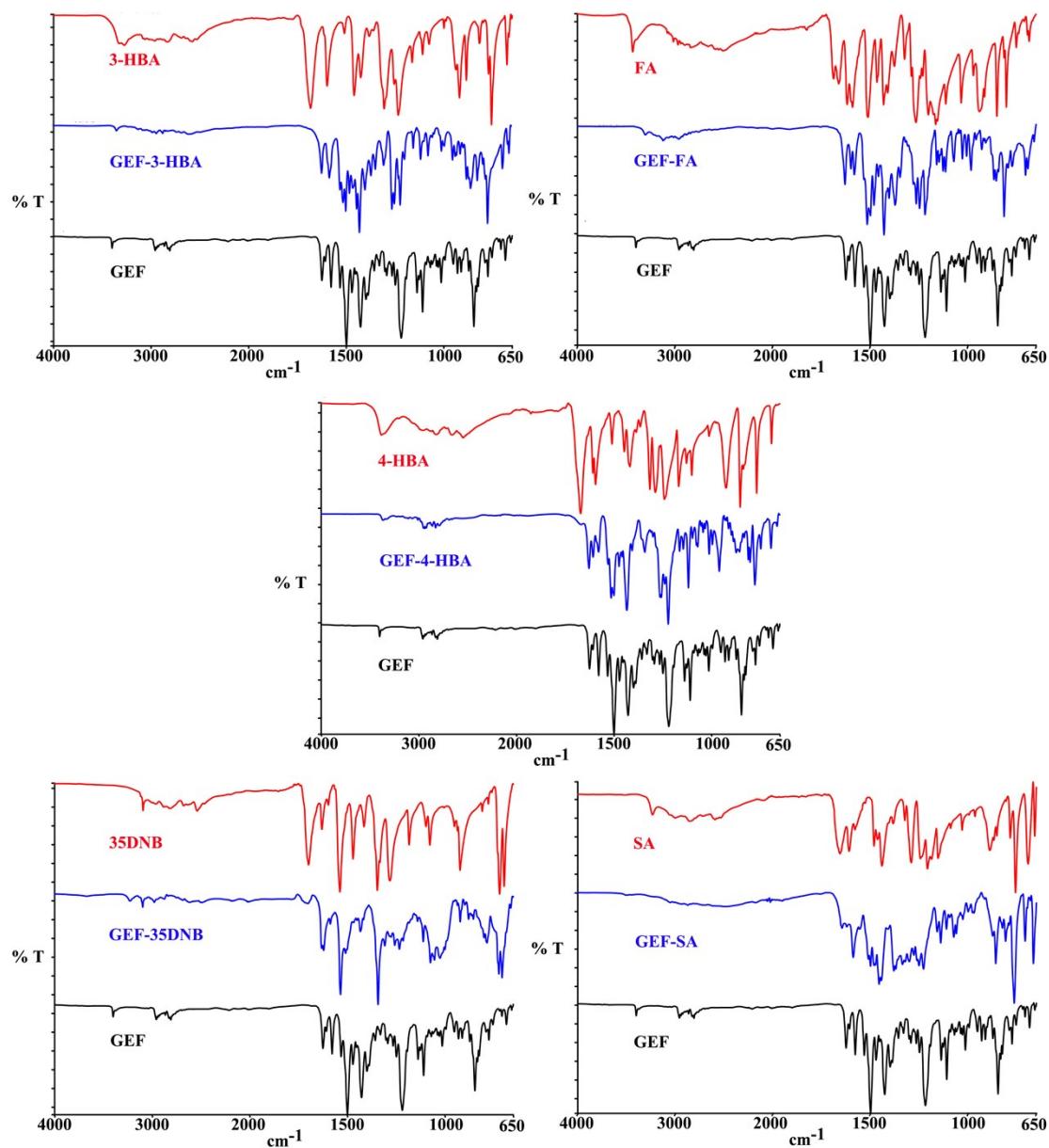
Fig. S4 The PRXD patterns of GEF, GEF-3-HBA, GEF-FA, GEF-4-HBA, GEF-35DNB and GEF-SA before and after dissolution experiment.

**Table S1.** Hydrogen bond geometrical parameters of crystal structures.

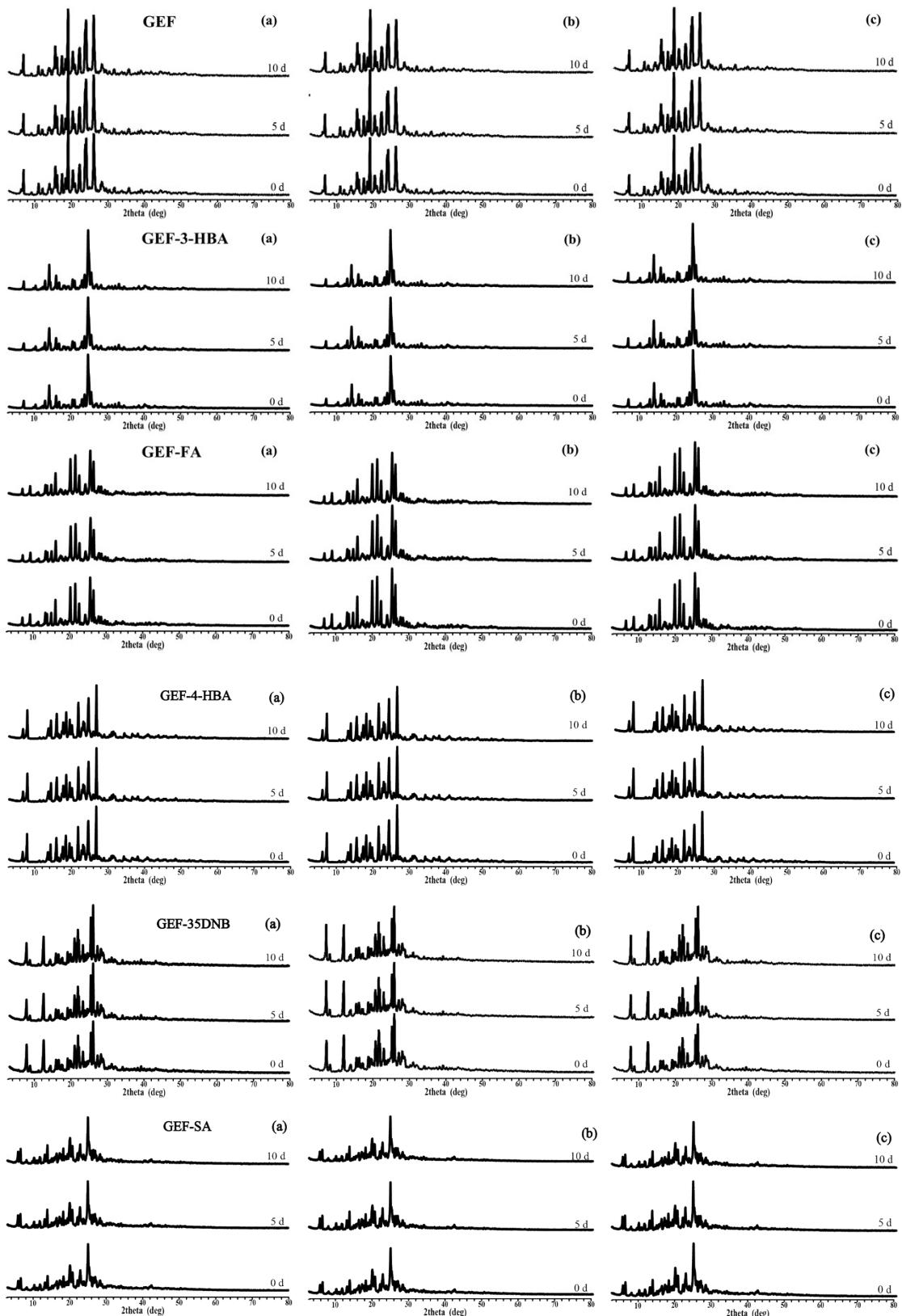
Connection	D···H/ Å	H···A/ Å	D···A/ Å	∠(DHA)/ °	Symmetry
<b>GEF-3-HBA</b>					
N <sub>1</sub> —H <sub>1</sub> ···O <sub>4</sub>	0.86	2.32	3.143	159	[x, y, z]
N <sub>4</sub> —H <sub>4</sub> ···O <sub>5</sub>	1.04	1.54	2.575	172	[x-1, y, z]
O <sub>6</sub> —H <sub>6</sub> ···N <sub>3</sub>	0.82	1.92	2.715	164	[x, -y+3/2, z+1/2]
<b>GEF-FA</b>					
N <sub>1</sub> —H <sub>1</sub> ···O <sub>6</sub>	0.86	2.20	3.011	157	[x, -y+1/2, z+1/2]
N <sub>4</sub> —H <sub>4</sub> ···O <sub>7</sub>	1.03	1.55	2.565	169	[x, y, z]
O <sub>4</sub> —H <sub>4B</sub> ···N <sub>3</sub>	0.82	2.07	2.874	165	[-x+1, -y+1, -z]
<b>GEF-4-HBA</b>					
N <sub>1</sub> —H <sub>1a</sub> ···O <sub>4</sub>	0.88	2.19	3.008	155	[1+x, y-1, z]
N <sub>1</sub> —H <sub>1b</sub> ···O <sub>6</sub>	0.88	2.15	2.888	141	[1-x, y-1, -z+1]
O <sub>5</sub> —H <sub>5b</sub> ···N <sub>3</sub>	0.84	1.68	2.492	161	[-x, -y+2, -z+1]
O <sub>6</sub> —H <sub>6</sub> ···N <sub>3</sub>	0.84	1.91	2.667	149	[x, y, z]
<b>GEF-35DNB</b>					
N <sub>1</sub> —H <sub>1</sub> ···O <sub>6</sub>	0.86	2.00	2.850	170	[x, y, z]
N <sub>4</sub> —H <sub>4</sub> ···O <sub>7</sub>	0.98	1.72	2.678	166	[x, y, z]
O <sub>11</sub> —H <sub>11A</sub> ···N <sub>3</sub>	0.82	1.72	2.505	158	[x, y, z]
<b>GEF-SA</b>					
N <sub>1A</sub> —H <sub>1A</sub> ···O <sub>4C</sub>	0.86	1.99	2.818	160	[x, y, z]
N <sub>1B</sub> —H <sub>1B</sub> ···O <sub>5C</sub>	0.86	2.00	2.832	162	[x, y, z]
N <sub>3B</sub> —H <sub>3B</sub> ···O <sub>4A</sub>	0.86	1.84	2.700	174	[1/2-x, 1-y, -1/2+z]
N <sub>4A</sub> —H <sub>4A</sub> ···O <sub>5A</sub>	0.98	1.81	2.778	170	[1-x, 1/2+y, 1/2-z]
N <sub>4B</sub> —H <sub>4B</sub> ···O <sub>4B</sub>	0.98	1.74	2.695	165	[x, y, z]
O <sub>5D</sub> —H <sub>5D</sub> ···N <sub>3A</sub>	0.82	1.95	2.732	159	[1/2+x, 3/2-y, -z]
O <sub>6A</sub> —H <sub>6A</sub> ···O <sub>5A</sub>	0.82	1.92	2.622	143	[x, y, z]
O <sub>6B</sub> —H <sub>6B</sub> ···O <sub>5B</sub>	0.82	1.79	2.502	145	[x, y, z]
O <sub>6C</sub> —H <sub>6C</sub> ···O <sub>5C</sub>	0.82	1.82	2.542	146	[x, y, z]
O <sub>6D</sub> —H <sub>6D</sub> ···O <sub>4D</sub>	0.82	1.78	2.505	146	[x, y, z]



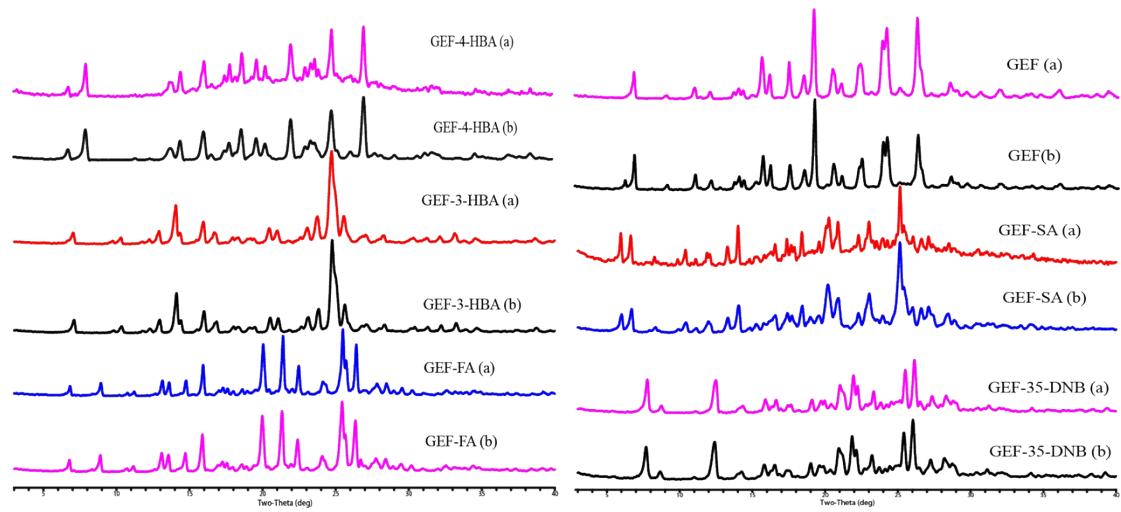
**Fig. S1** The DSC thermograms of CCFs used in this study.



**Fig. S2** FT-IR spectra of GEF, CCFs, and the corresponding salts/cocrystals.



**Fig. S3** Accelerated stability results of GEF, GEF-3-HBA, GEF-FA, GEF-4-HBA, GEF-35DNB and GEF-SA.  
 (a) high temperature ( $60 \pm 1$  °C); (b) high humidity ( $90 \pm 5\%$ , 25 °C); (c) illumination ( $4500 \pm 500$  lx, 25 °C)



**Fig. S4** The PRXD patterns of GEF, GEF-3-HBA, GEF-FA, GEF-4-HBA, GEF-35DNB and GEF-SA before and after dissolution experiment (a)the initial phases; (b) the bottom phases.