

## Electronic supplementary information

Syntheses, structures and magnetic properties of three dinuclear, one-dimensional and two-dimensional Dysprosium(III) complexes based on naphthalene diimide salicylic acid

*Jia-Ting Chen<sup>†</sup>, Xiao-Man Kuang<sup>†</sup>, You-Hong Li, Yun-Jing Zhong, Hui Zhang,<sup>a</sup> Jia-Yin Su,<sup>a</sup>*

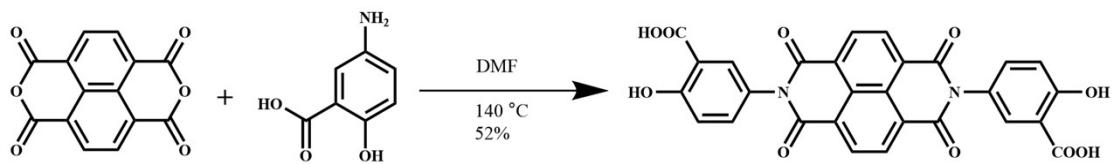
*Wen-Bin Chen\*, Wen Dong\**

Guangzhou Key Laboratory for Environmentally Functional Materials and Technology, School of Chemistry and Chemical Engineering, Guangzhou University, Guangzhou 510006, P. R. China.

E-mail: chenwb2018@gzhu.edu.cn; dw320@aliyun.com

<sup>†</sup>Jia-Ting Chen and Xiao-Man Kuang contributed equally.

<b>Scheme S1. Synthesis of H<sub>4</sub>NDISA.....</b>	<b>S1</b>
<b>Table S1. geometry analysis by using SHAPE 2.0 program (complex 1 and 3).....</b>	<b>S1</b>
<b>Table S2. geometry analysis by using SHAPE 2.0 program (complex 2) .....</b>	<b>S1</b>
<b>Figure S1. For 1, inter-molecular O19-H19B···O3 hydrogen bonding interaction and 1D face-to-face π-π stacking interaction.....</b>	<b>S1</b>
<b>Figure S2. For 2, crystal structure. ....</b>	<b>S2</b>
<b>Figure S3. For 3, 1D face-to-face π-π stacking interaction.. ..</b>	<b>S2</b>
<b>Figure S4. Powder X-ray diffraction (PXRD) patterns for complexes 1-3. ....</b>	<b>S3</b>
<b>Fig. S5 TG curves of complexes 1-3. ....</b>	<b>S3</b>
<b>Figure S6. M versus H/T curves for 1-3 (a-c). ....</b>	<b>S4</b>
<b>Figure S7. Temperature dependence of <math>\chi'_M</math> product and <math>\chi''_M</math> for complexes 1-3 (a-c) in a 1000 Oe dc field.....</b>	<b>S4</b>
<b>Table S3. Crystallographic data and redefinition parameters for complex 1-3.....</b>	<b>S5</b>
<b>Table S4. Selected bond lengths (Å) and angles (°) for complex 1 .....</b>	<b>S6</b>
<b>Table S5. Selected bond lengths (Å) and angles (°) for complex 2 .....</b>	<b>S7</b>
<b>Table S6. Selected bond lengths (Å) and angles (°) for complex 3 .....</b>	<b>S8</b>



Scheme S1. Synthesis of N,N'-bis(3-carboxy-4-hydroxyphenyl)-1,4,5,8-naphthalenetetradicarboximide (H<sub>4</sub>NDISA).

Table S1. geometry analysis by using SHAPE 2.0 program (complex **1** and **3**)

	SAPR-8 (D <sub>4d</sub> )	TDD-8 (D <sub>2h</sub> )	JBTPR-8 (C <sub>2v</sub> )	BTPR-8 (C <sub>2v</sub> )	JSD-8 (D <sub>2d</sub> )
<b>1</b>	<b>1.0003</b>	<b>0.9829</b>	<b>1.8019</b>	<b>1.4860</b>	<b>2.8753</b>
<b>3</b>	<b>3.7518</b>	<b>1.1756</b>	<b>4.0640</b>	<b>3.6774</b>	<b>4.4688</b>

Table S2. geometry analysis by using SHAPE 2.0 program (complex **2**)

	JCSAPR-9 (C <sub>4v</sub> )	CSAPR-9 (C <sub>4v</sub> )	JTCTPR-9 (D <sub>3h</sub> )	TCTPR-9 (D <sub>3h</sub> )	MFF-9 (C <sub>S</sub> )
<b>2</b>	<b>1.9197</b>	<b>0.8247</b>	<b>2.6546</b>	<b>1.1809</b>	<b>1.2309</b>

\*SAPR-8 = Square antiprism; TDD-8 = Triangular dodecahedron;  
JBTPR-8 = Biaugmented trigonal prism; BTPR-8 = Biaugmented trigonal prism;  
JSD-8 = Snub diphenoïd; JCSAPR-9 = Capped square antiprism;  
CSAPR-9 = Spherical capped square antiprism; JTCTPR-9 = Tricapped trigonal prism;  
TCTPR-9 = Spherical tricapped trigonal prism; MFF-9 = Muffin.

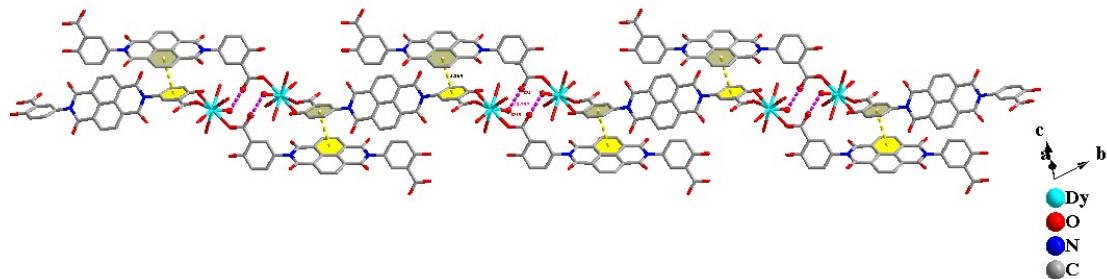


Figure S1. For **1**, inter-molecular O19-H19B...O3 hydrogen bonding interaction and 1D face-to-face  $\pi$ - $\pi$  stacking interaction. Hydrogen atoms and part solvent molecules were omitted for clarity.

Color code: Dy, green; O, red; N, blue; C, gray. Hydrogen atoms and uncoordinated solvent molecules were omitted for clarity.

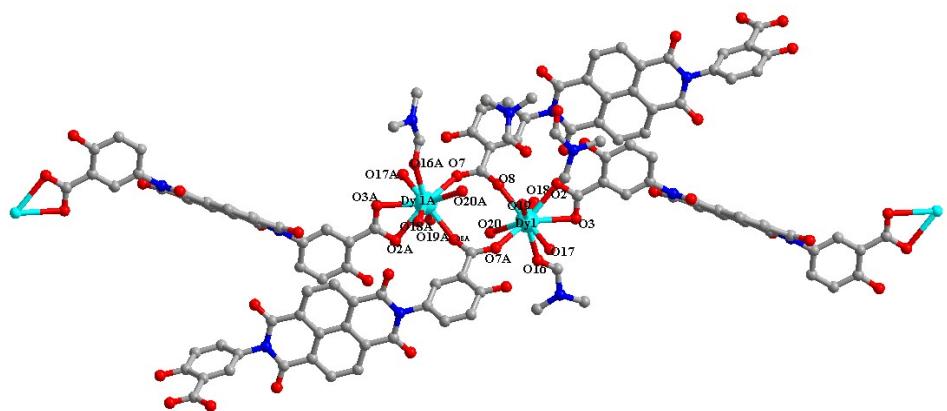


Figure S2. For **2**, crystal structure. Hydrogen atoms and part solvent molecules were omitted for clarity. Color code: Dy, green; O, red; N, blue; C, gray. Hydrogen atoms and uncoordinated solvent molecules were omitted for clarity.

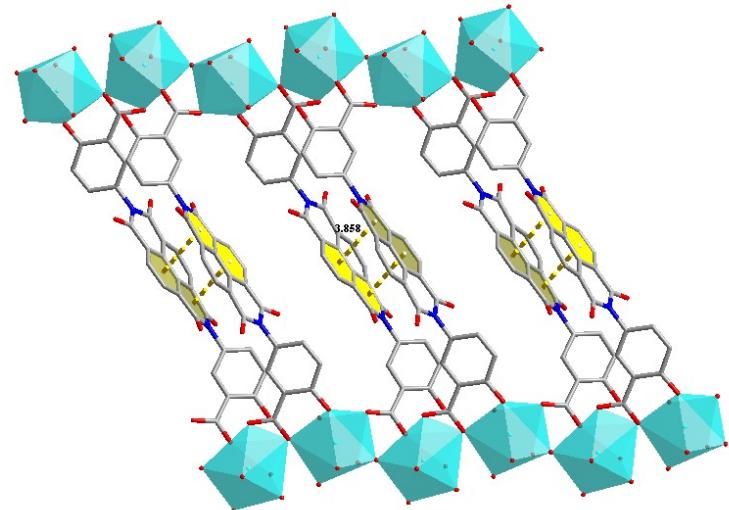


Figure S3. For **3**, 1D face-to-face  $\pi$ - $\pi$  stacking interaction. Hydrogen atoms and part solvent molecules were omitted for clarity. Color code: Dy, green; O, red; N, blue; C, gray. Hydrogen atoms and uncoordinated solvent molecules were omitted for clarity.

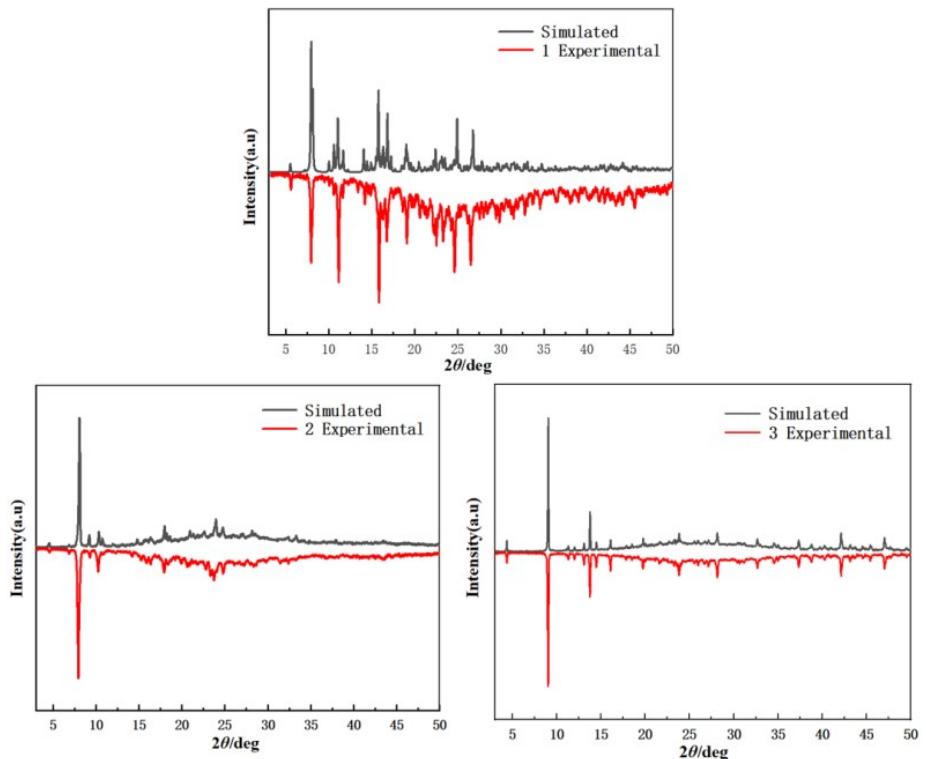


Figure S4. Powder X-ray diffraction (PXRD) patterns for complexes **1-3**.

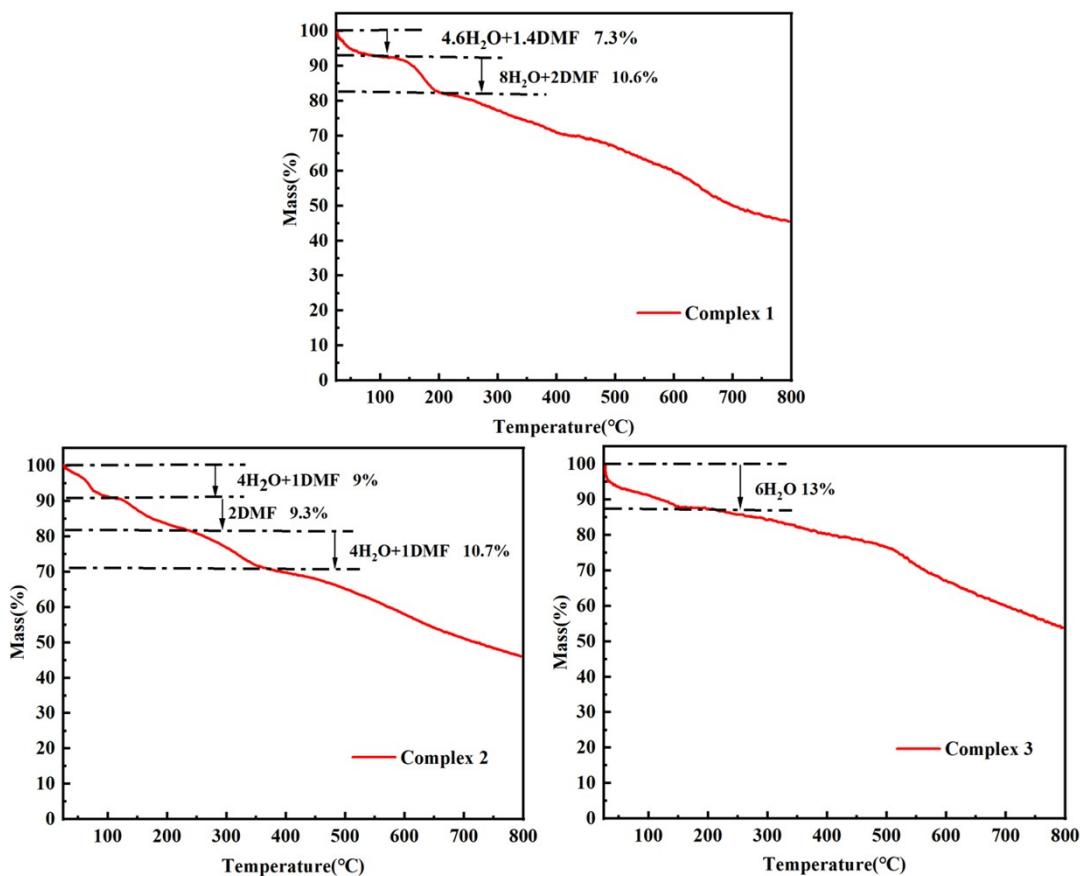


Figure S5 TG curves of complexes **1-3**.

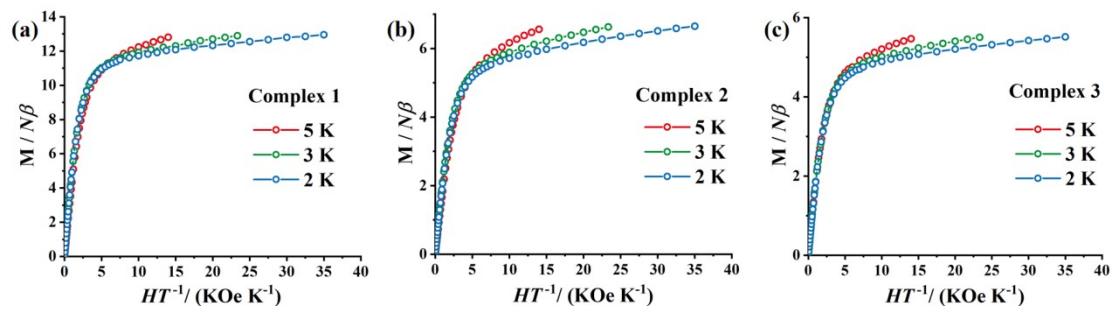


Figure S6.  $M$  versus  $H/T$  curves for **1-3** (a-c).

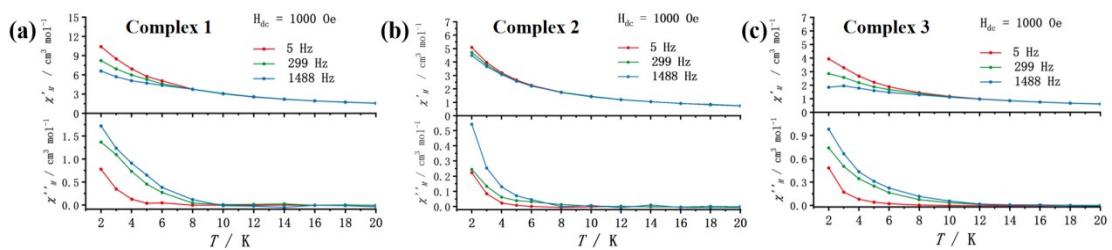


Figure S7. Temperature dependence of  $\chi'_M$  product and  $\chi''_M$  for complexes **1-3** (a-c) in a 1000 Oe dc field.

Table S3. Crystallographic data and redefinition parameters for complex **1-3**

Identification code	<b>1</b>	<b>2</b>	<b>3</b>
Empirical formula	C <sub>96</sub> H <sub>88</sub> Dy <sub>2</sub> N <sub>10</sub> O <sub>46</sub>	C <sub>54</sub> H <sub>58</sub> DyN <sub>7</sub> O <sub>25</sub>	C <sub>28</sub> H <sub>19</sub> DyN <sub>2</sub> O <sub>14</sub>
Formula weight	2442.76	1367.57	769.95
Crystal system	triclinic	triclinic	monoclinic
Space group	P-1	P-1	P2 <sub>1</sub> /n
a/Å	11.7197(2)	11.5540(2)	8.8770(3)
b/Å	12.7121(2)	12.8863(2)	37.7347(9)
c/Å	17.4599(3)	19.4414(3)	8.9024(4)
α/°	74.413(2)	83.0210(10)	90
β/°	71.436(2)	74.564(2)	119.789(5)
γ/°	86.5400(10)	84.6300(10)	90
Volume/Å <sup>3</sup>	2374.28(8)	2763.83(8)	2587.99(19)
Z	1	2	4
ρ <sub>calc</sub> g/cm <sup>3</sup>	1.708	1.643	1.976
μ/mm <sup>-1</sup>	1.670	8.062	2.972
F(000)	1234.0	1394.0	1516.0
Reflections collected	35944	40597	34502
Independent reflections	12043	10870	5955
Data/restraints/parameters	12043/46/793	10870/36/830	5955/0/407
Goodness-of-fit on F <sup>2</sup>	1.072	1.054	1.124
Final R indexes [I>=2σ (I)]	R <sub>1</sub> = 0.0398, wR <sub>2</sub> = 0.1018	R <sub>1</sub> = 0.0514, wR <sub>2</sub> = 0.1348	R <sub>1</sub> = 0.0382, wR <sub>2</sub> = 0.0738
Final R indexes [all data]	R <sub>1</sub> = 0.0489, wR <sub>2</sub> = 0.1059	R <sub>1</sub> = 0.0588, wR <sub>2</sub> = 0.1396	R <sub>1</sub> = 0.0483, wR <sub>2</sub> = 0.0768
Largest diff. peak/hole/ e Å <sup>-3</sup>	2.00/-1.16	1.32/-1.11	1.24/-1.92
CCDC	2369945	2369946	2369947

Table S4. Selected bond lengths (Å) and angles (°) for complex 1

Complex 1			
Dy1-O16	2.290(3)	O16-Dy1-O2	80.80(10)
Dy1-O22	2.30(4)	O22-Dy1-O2	71.6(10)
Dy1-O12	2.320(2)	O12-Dy1-O2	144.85(9)
Dy1-O21	2.335(13)	O21-Dy1-O2	71.6(4)
Dy1-O19	2.347(2)	O19-Dy1-O2	74.73(9)
Dy1-O2	2.398(2)	O16-Dy1-O18	145.11(11)
Dy1-O18	2.411(3)	O22-Dy1-O18	69.9(9)
Dy1-O20	2.417(3)	O12-Dy1-O18	73.04(9)
Dy1-O17	2.423(3)	O21-Dy1-O18	78.2(3)
O16-Dy1-O22	96.6(8)	O19-Dy1-O18	69.09(9)
O16-Dy1-O12	104.83(10)	O2-Dy1-O18	121.57(8)
O22-Dy1-O12	139.6(10)	O16-Dy1-O20	73.39(10)
O16-Dy1-O21	85.6(3)	O22-Dy1-O20	144.0(10)
O12-Dy1-O21	142.6(4)	O12-Dy1-O20	75.81(9)
O16-Dy1-O19	145.79(10)	O21-Dy1-O20	140.9(4)
O22-Dy1-O19	97.9(8)	O19-Dy1-O20	76.70(9)
O12-Dy1-O19	83.24(9)	O2-Dy1-O20	72.71(9)
O21-Dy1-O19	108.3(3)	O18-Dy1-O20	135.70(10)
O16-Dy1-O17	73.52(12)	O19-Dy1-O17	140.18(11)
O22-Dy1-O17	76.4(9)	O2-Dy1-O17	135.89(11)
O12-Dy1-O17	77.44(11)	O18-Dy1-O17	72.06(11)
O21-Dy1-O17	71.2(4)	O20-Dy1-O17	129.77(11)

Table S5. Selected bond lengths (Å) and angles (°) for complex 2

Complex-2			
Dy1-O7 <sup>1</sup>	2.410(3)	O7 <sup>1</sup> -Dy1-O3	140.91(10)
Dy1-O2	2.475(3)	O7 <sup>1</sup> -Dy1-O20	68.74(10)
Dy1-O3	2.536(3)	O7 <sup>1</sup> -Dy1-O18	139.55(11)
Dy1-O8	2.358(3)	O7 <sup>1</sup> -Dy1-C7	149.29(11)
Dy1-O20	2.430(3)	O2-Dy1-O3	51.94(10)
Dy1-O19	2.408(3)	O2-Dy1-C7	26.39(12)
Dy1-O18	2.417(3)	O3-Dy1-C7	25.56(12)
Dy1-O17	2.405(3)	O8-Dy1-O7 <sup>1</sup>	93.04(10)
Dy1-O16	2.353(3)	O8-Dy1-O2	77.64(10)
O7 <sup>1</sup> -Dy1-O2	141.08(11)	O8-Dy1-O3	123.93(10)
O16--Dy1-O17	71.18(12)	O16-Dy1-C7	98.93(12)
O16-Dy1-O18	82.14(12)	O16-Dy1-O19	139.52(12)
O16-Dy1O20	71.74(11)	O16-Dy1-O3	73.42(11)
O16-Dy1-O8	142.99(11)	O16-Dy1-O2	125.15(11)
O16-Dy1-O7 <sup>1</sup>	84.37(11)	O17-Dy1-C7	81.47(12)
O17-Dy1-O18	137.75(11)	O17-Dy1-O3	71.87(11)
O17--Dy1-O19	70.40(12)	O17-Dy1-O2	93.72(11)
O17-Dy1-O20	126.62(11)	O17-Dy1-O7 <sup>1</sup>	70.65(11)
O18-Dy1-C7	70.79(12)	O18-Dy1-O20	70.82(10)
O18-Dy1-O2	75.30(11)	O18-Dy1-O3	69.40(11)
O8-Dy1-O20	72.97(11)	O8-Dy1-O19	72.05(11)
O19-Dy1-C7	87.59(12)	O19-Dy1-O3	105.78(10)
O19-Dy1-O18	136.68(11)	O19-Dy1-O2	69.81(10)
O19-Dy1-O20	124.31(10)	O19-Dy1-O7 <sup>1</sup>	71.37(10)
O20-Dy1-C7	141.38(11)	O20-Dy1-O3	129.66(10)
O20-Dy1-O2	139.31(11)	O8-Dy1-C7	101.64(12)
O8-Dy1-O17	142.15(11)	O8-Dy1-O18	76.13(11)

Table S6. Selected bond lengths (Å) and angles (°) for complex 3

Complex 3			
Dy1-O1	2.239(3)	O1-Dy1-O2	71.95(10)
Dy1-O9 <sup>2</sup>	2.354(3)	O1-Dy1-O3 <sup>1</sup>	83.09(11)
Dy1-O10 <sup>3</sup>	2.318(3)	O1-Dy1-O9 <sup>2</sup>	139.18(10)
Dy1-O11	2.361(3)	O1-Dy1-O10 <sup>3</sup>	76.80(11)
Dy1-O12	2.384(3)	O1-Dy1-O11	92.96(11)
Dy1-O2 <sup>1</sup>	2.541(3)	O1-Dy1-O12	152.89(11)
Dy1-O2	2.428(3)	O1-Dy1-C7 <sup>1</sup>	89.86(11)
Dy1-O3 <sup>1</sup>	2.416(3)	O2-Dy1-O2 <sup>1</sup>	139.42(9)
O1-Dy1-O2 <sup>1</sup>	95.83(10)	O2-Dy1-C7 <sup>1</sup>	113.03(11)
O12-Dy1-C7 <sup>1</sup>	95.99(11)	O11-Dy1-O2 <sup>1</sup>	145.40(10)
O12-Dy1-O3 <sup>1</sup>	112.59(11)	O11-Dy1-O2	75.01(10)
O12-Dy1-O2 <sup>1</sup>	78.90(10)	O10 <sup>3</sup> -Dy1-C7 <sup>1</sup>	100.30(11)
O12-Dy1-O2	128.35(10)	O10 <sup>3</sup> -Dy1-O12	76.11(11)
O11-Dy1-C7 <sup>1</sup>	171.96(11)	O10 <sup>3</sup> -Dy1-O11	73.08(11)
O11-Dy1-O12	78.16(11)	O10 <sup>3</sup> -Dy1-O9 <sup>2</sup>	143.96(10)
O11-Dy1-O3 <sup>1</sup>	162.33(10)	O10 <sup>3</sup> -Dy1-O3 <sup>1</sup>	122.21(11)
O9 <sup>2</sup> -Dy1-C7 <sup>1</sup>	81.54(11)	O10 <sup>3</sup> -Dy1-O2 <sup>1</sup>	76.50(10)
O9 <sup>2</sup> -Dy1-O12	67.93(10)	O10 <sup>3</sup> -Dy1-O2	133.46(10)
O9 <sup>2</sup> -Dy1-O11	101.05(11)	O9 <sup>2</sup> -Dy1-O3 <sup>1</sup>	72.04(11)
O3 <sup>1</sup> -Dy1-C7 <sup>1</sup>	25.67(11)	O9 <sup>2</sup> -Dy1-O2	75.01(10)
O3 <sup>1</sup> -Dy1-O2	87.41(10)	O9 <sup>2</sup> -Dy1-O2 <sup>1</sup>	93.87(10)
O3 <sup>1</sup> -Dy1-O2 <sup>1</sup>	52.27(9)	O21-Dy1-C7 <sup>1</sup>	26.62(10)