Supporting Information

Organic Fluorine Mediated Intermolecular Interactions: Insights from Experimental and Theoretical Charge Density Analyses

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Figure S1: (a) Histogram for angle 1 (θ_1), (b) Histogram for angle 2 (θ_2), (c) Distance histogram, and (*d*) Heat plot for angle (θ_1 and θ_2 , for all the structures having C-F...F-C interaction along with other intermolecular interactions (search 1).



Figure S2: (a) Histogram for angle 1 (θ_1), (b) Histogram for angle 2 (θ_2), (c) Distance histogram, and (*d*) Heat plot for angle (θ_1 and θ_2), for all the structures having C-F···F-C interaction devoid of strong hydrogen bonding interactions (search 2).



Figure S3: (a) Histogram for angle 1 (θ_1), (b) Histogram for angle 2 (θ_2), (c) Distance histogram, and (*d*) Heat plot for angle (θ_1 and θ_2), for all the structures having C-F…F-C interaction devoid of strong and weak hydrogen bonding interactions but may contain C-H…F-C interaction (search 3).





(b)

1200 -



Figure S4: (a) Histogram for angle 1 (θ_1), (b) Histogram for angle 2 (θ_2), (c) Distance histogram, and (*d*) Heat plot for angle (θ_1 and θ_2), for all the structures having C-F…F-C interaction devoid of strong and weak hydrogen bonding interactions and not contain C-H…F-C interaction (search 4).

Statistical parameters	Search 1	Search 2	Search 3	Search 4	
No. of hits	3100	2529	1269	359	
No. of interactions	7550	6372	3576	1531	
Maximum range for θ ₁ (°)	120-135	120-135	120-135	120-160	
Maximum range for θ_2 (°)	120-135	120-135	120-135	120-160	
Maximum range for distance (Å)	2.90-2.94	2.90-2.94	2.90-2.94	2.90-2.94	
No. of type I interactions $[0^{\circ} \leq \theta_1 - \theta_2 \leq 15^{\circ}]$	3583	2958	1593	663	
No. of type II interactions $[30^{\circ} \le \theta_1 - \theta_2]$	2407	2050	1196	513	
No. of quasi type I/type II interactions $[15^{\circ} \leq \theta_1 - \theta_2 \leq 30^{\circ}]$	1560	1364	787	355	
Ratio of the number of hits of type I: type II: quasi type I/type II	2.3:1.54:1	2.2:1.50:1	2.02:1.5:1	1.9:1.45:1	

Table S1: Summary of CSD search for C-F \cdots F-C interactions among the structured reported at temperature 80 – 120 K.



(a)





(c)



Figure S5: (a) Molecular packing diagram depicting $C-F\cdots F-C$ interactions (blue dotted lines) and $C-H\cdots F-C$ hydrogen bonds (orange dotted lines), (b) type II $C-F\cdots F-C$ intermolecular interaction, (c) intermolecular C18–H18 \cdots F2–C12 and, (d) C2–H2A \cdots F3–C13 and C21–H21 \cdots F1–C11 interactions. The distances and angels are given in Å and degrees (°), respectively.



Figure S6: (a) The normal probability distribution plot, (b) Variation of the scale factor with respect to $\sin \theta / \lambda$.



Figure S7: (a) Residual electron density distribution plot, (b) fractal dimension vs residual electron density plot.

Table S2: Atomic charges after final multipole modelling (The values from the experimental and theoretical analyses are given in the first and second rows, respectively).

Atom	Atomic charge	Карра	Kappa'
F(1)	-0.070	0.997	1.201
- (-)	-0.148	0.999	1.201
F(2)	-0.156	0.990	1.196
~ /	-0.159	0.999	1.196
F(3)	-0.171	0.991	1.193
(-)	-0.186	0.996	1.193
F(4)	-0.185	0.989	1.222
~ /	-0.165	0.998	1.222
O(1)	-0.221	0.984	1.211
	-0.143	0.996	1.123
N(1)	-0.137	0.990	0.936
~ /	-0.046	0.993	1.026
C(1)	-0.103	0.996	0.936
	-0.108	0.991	0.903
C(2)	-0.220	0.991	0.938
	-0.302	0.991	0.894
C(3)	-0.182	0.991	0.938
	-0.199	0.991	0.894
C(4)	-0.124	0.994	0.979
~ /	-0.063	0.995	0.852
C(5)	-0.052	0.993	0.933
	-0.076	0.994	0.885
C(6)	-0.143	0.994	0.979
	-0.127	0.995	0.852
C(7)	-0.151	0.994	0.979
	-0.106	0.995	0.852
C(8)	-0.076	0.997	0.948
	-0.097	0.994	0.839
C(9)	-0.030	0.997	0.948
	+0.006	0.994	0.839
C(10)	+0.020	1.006	1.029
	-0.027	0.992	0.833
C(11)	-0.000	1.007	0.987
	+0.003	1.000	0.903
C(12)	+0.116	1.007	0.987
	+0.069	1.000	0.903
C(13)	-0.122	0.994	0.979
	-0.198	0.995	0.852
C(14)	-0.102	0.994	0.979
	-0.218	0.995	0.852
C(15)	-0.192	0.994	0.979
	-0.135	0.995	0.852
C(16)	-0.023	1.001	0.999
	-0.094	0.995	0.885
C(17)	+0.043	1.007	0.987
	+0.017	1.000	0.903
C(18)	-0.197	0.994	0.979
	-0.166	0.995	0.852
C(19)	-0.168	0.994	0.979
- \ - /	-0.202	0.995	0.852
C(20)	0.202	1.007	0.097
C(20)	+0.102	1.007	0.90/
	-0.022	1.000	0.903
C(21)	-0.222	0.994	0.979
	-0.190	0.995	0.852
C(22)	-0.263	0.990	1.008

	-0.331	0.994	0.890	
H(1)	+0.167	1.230	1.230	
	+0.208	1.230	1.20	
H(2A)	+0.155	1.212	1.212	
	+0.193	1.212	1.212	
H(2B)	+0.150	1.212	1.212	
	+0.174	1.212	1.212	
H(3A)	+0.104	1.212	1.212	
	+0.172	1.212	1.212	
H(3B)	+0.149	1.212	1.212	
	+0.188	1.212	1.212	
H(4)	+0.187	1.233	1.233	
	+0.201	1.233	1.233	
H(6)	+0.207	1.233	1.233	
	+0.189	1.233	1.233	
H(7)	+0.194	1.233	1.233	
	+0.183	1.233	1.233	
H(13)	+0.199	1.233	1.233	
	+0.227	1.233	1.233	
H(14)	+0.201	1.233	1.233	
	+0.225	1.233	1.233	
H(15)	+0.176	1.233	1.233	
	+0.201	1.233	1.233	
H(18)	+0.237	1.233	1.233	
	+0.206	1.233	1.233	
H(19)	+0.212	1.233	1.233	
	+0.245	1.233	1.233	
H(21)	+0.221	1.233	1.233	
	+0.192	1.233	1.233	
H(22A)	+0.152	1.227	1.227	
	+0.222	1.227	1.227	
H(22B)	+0.137	1.227	1.227	
	+0.203	1.227	1.227	
H(22C)	+0.184	1.227	1.227	
	+0.186	1.227	1.227	



Figure S8: Residual density maps, drawn for the planes having atoms (a) F1, C10, and F2 (b) F3, C16, and F4 (c) C5, O1, and C22 and (d) N1, C2, and C3 using positive (blue, solid line) and negative (red, dotted line) with contour starting at ± 0.05 e Å⁻³ and with an interval of ± 0.1 e Å⁻³.



(d)

Figure S9: 2D Deformation electron density maps, drawn on the planes containing atoms (a) F1,C10, and F2, (b) F3, C16, and F4, (c) C5, O1, and C22, and (d) N1, C2, and C3 using positive (blue, solid line) and negative (red, dotted line) with contour starting at \pm 0.05 e Å⁻³ with an interval of \pm 0.1 e Å⁻³.



(b)

Figure S10: (a) experimental and (b) theoretical 3D static deformation density maps with positive (blue surface) and negative (red surface).



Figure S11: (a) experimental, and (b) theoretical 3D Deformation density maps for the weak C2–H2A…F3–C17 and C21–H21…F1–C11 hydrogen bonds.



Figure S12: The molgraph view showing BCPs and bond paths for the covalent bonds and the ring critical points.



Figure S13: Laplacian ($\nabla^2 \rho$) maps on the planes containing atoms (a) F1, C10, and F2, (b) F3, C16, and F4, (c) C5, O1, and C22, and (d) N1, C2, and C3 drawn at the logarithmic interval of $-\nabla^2 \rho \, e \text{\AA}^{-5}$.

Table S3: Topological parameters of the covalent bonds in compound **1**. The values from the experimental and theoretical analyses are given in the first and second rows, respectively.

Atom A	Atom B	ρ (eÅ ⁻³)	$ \nabla^2 \rho $ (eÅ ⁻⁵)	R (Å)	d (Å)	d 2 (Å)	λ,	λ ₂	λ,	3
E(1)	C(11)	1.875(20)	-16.218(91)	1.347	0.8405	0.5067	-15.47	-14.1	13.35	0.1
F(1)	C(11)	1.899(5)	-15.796(23)	1.347	0.8332	0.5139	-14.85	-14.16	13.22	0.05
F(2)	C(12)	1.855(21)	-14.152(93)	1.341	0.8429	0.4988	-14.48	-13.6	13.93	0.06
		1.887(5)	-15.332(24) 12.912(90)	1.342	0.8359	0.5061	-14.41	-14.15	13.22	0.02
F(3)	C(17)	1.701(21)	-12.912(90)	1.557	0.8555	0.5018	-13.9	-12.19	13.17	0.14
		1.784(5)	10.776(22)	1.357	0.8361	0.5214	-12.55	-12.3	14.07	0.02
F(4)	C(20)	1.834(21)	-14.089(97)	1.351	0.8469	0.5044	-14.33	-13.37	13.61	0.07
		1.877(5)	-14.658(23)	1.352	0.8345	0.5176	-14.13	-13.89	13.36	0.02
O(1)	C(5)	1.985(18)	-17.136(88)	1.365	0.8212	0.5444	-15.26	-14.16	12.29	0.08
		2.022(5)	-18.001(23)	1.366	0.8207	0.5454	-15.48	-15.03	12.51	0.03
O(1)	C(22)	1.718(22)	-10.130(91)	1.422	0.0413	0.581	-12.1	-11.1	13.04	0.09
		1.792(5)	-13.524(16)	1.422	0.8352	0.5872	-13.87	7 -14.1 13.35 0.1 5 -14.16 13.22 0.05 8 -13.6 13.93 0.06 1 -14.15 13.22 0.02 -12.19 13.17 0.14 5 -12.3 14.07 0.02 3 -13.37 13.61 0.07 3 -13.89 13.36 0.02 6 -14.16 12.29 0.08 8 -15.03 12.51 0.03 -11.1 13.04 0.09 7 -12.94 13.29 0.07 6 -11.20 14.57 0.09 6 -11.35 13.09 0.06 5 -12.54 13.32 0.09 6 -11.35 13.09 0.06 5 -13.43 14.17 0.08 1 -13.80 13.81 0.17 8 -16.54 16.98 0.02 5 -16.34		
N(1)	C(1)	1.714(14)	-8.781(49)	1.484	0.8222	0.6623	-12.16	-11.20	14.57	0.09
		1.725(4)	-9.943(13)	1.485	0.8192	0.6657	-12.56	-11.56	13.35 0.1 13.22 0.05 13.93 0.06 13.93 0.02 13.17 0.14 14.07 0.02 13.61 0.07 13.61 0.02 13.61 0.07 13.61 0.02 12.29 0.08 12.51 0.03 13.04 0.09 13.29 0.07 14.57 0.09 14.18 0.09 13.04 0.09 13.05 0.02 13.32 0.09 14.17 0.08 13.32 0.09 14.17 0.08 13.81 0.17 16.87 0.02 10.68 0.06 10.07 0.05 17.45 0.03 15.99 0.04 15.74 0.03 15.93 0.02 16.31 0.01 10.28 0.34	0.09
N(1)	C(2)	1.728(15)	-10.255(48)	1.463	0.8315	0.6323	-12.00	-11.35	13.09	0.06
		1.803(4)	-12.873(12)	1.464	0.8254	0.6386	-13.65	-12.54	13.32	0.09
N(1)	C(16)	1.932(21)	-13.811(88)	1.418	0.803	0.6151	-14.55	-13.43	14.17	0.08
~ /	~ /	2.012(5)	-16.196(18)	1.417	0.7961	0.6218	-16.21	-13.80	13.81	0.17
C(1)	H(1)	1.842(29)	-16.446(81)	1.099	0.7159	0.3832	-16.88	-16.54	16.98	0.02
0(1)	(1) H(1)	1.847(10)	-17.890(33)	1.099	0.7321	0.3668	-17.55	-17.22	16.87	0.02
C(2) C(3)	C(3)	1.647(15)	-10.587(42)	1.520	0.7639	0.757	-10.96	-10.3	10.68	0.06
	0(3)	1.716(4)	-13.232(9)	1.520	0.7726	0.7483	-11.91	-11.39	10.07	0.05
C(2)	H(2A)	1.812(29)	-15.744(86)	1.093	0.7238	0.3692	-16.85	-16.34	17.45	0.03
	~ /	1.871(8)	-19.092(24)	1.092	0.7231	0.3689	-17.87	-17.21	15.99	0.04
C(2)	H(2B)	1.776(30)	-14.764(88)	1.093	0.7234	0.3704	-16.47	-15.88	17.58	0.04
		1.866(8)	-18.254(23)	1.092	0.7253	0.3667	-17.81	-17.09	16.64	0.04
C(3)	H(3A)	1.760(27)	-14.523(71)	1.092	0.6913	0.4008	-15.36	-14.90	15.74	0.03
		1.832(8)	-17.672(24)	1.092	0.7152	0.3768	-16.93	-16.66	15.93	0.02
C(3)	H(3B)	1.736(30)	-14.039(86)	1.092	0.7127	0.3794	-15.63	-15.18	16.77	0.03
		1.859(7)	-19.098(21)	1.092	0.7135	0.3785	-17.40	-17.06	15.35	0.02
C(4)	I(1) C(1) I(1) C(2) I(1) C(16) I(1) C(16) I(1) H(1) I(2) C(3) I(2) H(2A) I(2) H(2B) I(3) H(3A) I(3) H(3B) I(4) H(4) I(5) C(4) I(5) C(6) I(6) H(6)	1.772(28)	-15.261(86)	1.083	0.7134	0.3699	-17.13	-15.94	17.82	0.07
		1.855(7)	-17.466(21)	1.083	0.757	0.3773	-17.00	-16.78	16.31	0.01
C(5)	C(4)	2.162(19)	-20.195(53)	1.390	0.715	0.6755	-17.42	-13.05	10.28	0.34
		2.138(4)	-18.955(12)	1.390	0.7406	0.6501	-15.57	-12.83	9.44	0.21
C(5)	C(6)	2.135(19)	-19.278(52)	1.404	0.7239	0.6803	-16.87	-13.00	10.59	0.30
0(0)	C(0)	2.100(4)	-18.793(11)	1.404	0.7261	0.6780	-15.40	-13.09	9.70	0.18
C(6)	H(6)	1.801(29)	-16.374(89)	1.083	0.7180	0.3653	-17.53	-16.59	17.75	0.06
		1.854(6)	-17.519(18)	1.083	0.6923	0.3097	-16.74	-16.37	15.59	0.02
C(7)	C(6)	2.122(18)	-18.518(51)	1.387	0.6937	0.6942	-16.22	-13.13	10.83	0.24
0(1)	0(0)	2.103(3)	-17.632(9)	1.388	0.6769	0.7111	-14.93	-12.45	9.75	0.20
C(7)	H(7)	1.797(28)	-16.023(85)	1.08	0.7152	0.3681	-17.26	-16.57	17.80	0.04
-(/)	(/)	1.843(7)	-17.017(20)	1.083	0.6961	0.3870	-16.58	-16.42	15.98	0.01
C(8)	C(9)	2.139(22)	-18.375(67)	1.394	0.7029	0.6915	-16.46	-12.91	11.00	0.27
- */	~ /	2.101(4)	-17.824(12)	1.394	0.7077	0.6868	-14.75	-12.62	9.55	0.17

C(8) $C(1)$	C(1)	1.680(15)	-11.124(40)	1.517	0.7664	0.7506	-11.71	-10.52	11.11	0.11
-(0)	-(-)	1.702(3)	-12.244(8)	1.517	0.7567	0.7604	-11.78	-10.72	10.25	0.10
C(8)	C(7)	2.077(18)	-17.718(47)	1.406	0.7063	0.6999	-15.95	-12.82	11.06	0.24
C(0)	C(/)	2.023(3)	-15.757(9)	1.406	0.6959	0.7104	-13.98	-11.89	10.11	0.18
C(0)	C(4)	2.045(17)	-16.878(46)	1.404	0.6992	0.7053	-15.20	-12.60	10.92	0.21
C(9)	C(4)	2.082(3)	-18.293(9)	1.404	0.7087	0.6958	-14.88	-12.96	9.54	0.15
C(0)	C(2)	1.703(15)	-11.171(37)	1.507	0.7552	0.7520	-11.61	-10.76	11.19	0.08
C(9)	C(3)	1.723(3)	-12.700(7)	1.507	0.7557	0.7513	-11.44	-11.42	10.16	0.00
		2.150(24)	-18.532(78)	1.392	0.6581	0.7339	-17.08	-13.42	11.97	0.27
C(10)	C(11)	2.127(4)	-19.080(12)	1.392	0.6583	0.7338	-15.29	-13.22	9.43	0.16
G(10)	C(1)	1.634(17)	-9.421(47)	1.529	0.7724	0.7568	-10.82	-10.58	11.98	0.02
C(10)	C(1)	1.665(3)	-11.563(9)	1.529	0.7634	0.7657	-11.28	-10.51	10.23	0.07
G(10)	C (15)	2.101(22)	-17.179(65)	1.402	0.6973	0.7055	-16.06	-13.28	12.16	0.21
C(10)	C(15)	2.070(4)	-17.604(10)	1.402	0.6946	0.7082	-14.82	-12.38	9.60	0.20
C(11)	C(12)	2.196(21)	-20.308(60)	1.389	0.7027	0.687	-18.46	-13.64	11.79	0.35
C(11)	C(12)	2.246(4)	-22.737(13)	1.390	0.7051	0.6855	-18.07	-18.07	9.77	0.25
G(10)	6(10)	2.223(19)	-21.111(52)	1.380	0.7102	0.6707	-18.01	-14.19	11.09	0.27
C(12)	C(13)	2.194(4)	-19.597(9)	1.381	0.7066	0.6745	-16.30	-13.20	9.900	0.23
C(12)	U(12)	1.813(30)	-16.529(91)	1.083	0.7138	0.3700	-17.54	-16.55	17.57	0.06
C(15)	п(15)	1.837(7)	-17.054(20)	1.083	0.7155	0.3675	-17.31	-16.45	16.71	0.05
C(12)	C(14)	2.075(18)	-17.833(51)	1.396	0.6985	0.6977	-15.86	-12.91	10.94	0.23
C(15)	C(14)	2.080(3)	-17.337(9)	1.396	0.6951	0.7012	-14.70	-12.60	9.96	0.17
C(14)	H(14)	1.793(30)	-15.844(91)	1.083	0.7155	0.3676	-17.38	-16.28	17.81	0.07
C(14)	11(14)	1.887(7)	-18.706(22)	1.083	0.7168	0.3663	-17.83	-17.42	16.42	0.02
C(15)	H(15)	1.824(28)	-16.843(81)	1.083	0.7044	0.3789	-17.34	-16.61	17.10	0.04
C(15)	11(13)	1.883(7)	-18.343(20)	1.083	0.7047	0.3783	-17.77	-16.72	16.15	0.06
C(15)	C(14)	2.093(19)	-18.101(51)	1.394	0.7071	0.6872	-15.98	-13.03	10.90	0.23
0(15)	0(11)	2.100(3)	-17.454(8)	1.394	0.6911	0.7033	-14.80	-12.57	9.91	0.18
0(10)	0(17)	2.171(22)	-19.510(64)	1.400	0.701	0.6994	-18.1	-13.44	12.03	0.35
C(16)	C(17)	2.130(4)	-19.046(13)	1.400	0.6894	0.7114	-16.19	-12.96	10.10	0.25
C(16)	C(21)	2.064(21)	-16.881(58)	1.402	0.7219	0.6802	-16.20	-12.27	11.58	0.32
C(10)	C(21)	2.051(4)	-16.385(12)	1.402	0.7162	0.686	-14.31	-12.32	10.25	0.16
		2.196(18)	-20.541(52)	1.381	0.7268	0.6551	-17.59	-13.68	10.73	0.29
C(17)	C(18)	2.184(4)	-20.391(10)	1.382	0.7173	0.6648	-16.51	-13.44	9.56	0.23
G (10)	G (10)	2.132(18)	-19.126(54)	1.394	0.7059	0.6886	-16.74	-13.37	10.98	0.25
C(18)	C(19)	2.071(3)	-17.005(8)	1.394	0.7073	0.6873	-14.45	-12.58	10.03	0.15
Q(19)	11(10)	1.834(30)	-17.382(100)	1.083	0.7337	0.3496	-18.33	-17.54	18.49	0.04
C(18)	H(18)	1.864(7)	-17.577(22)	1.083	0.7079	0.3751	-17.21	-16.77	16.41	0.03
C(19)	H(19)	1.773(28)	-16.178(84)	1.083	0.7092	0.3739	-17.09	-15.96	16.87	0.07
		1.880(7)	-18.800(22)	1.803	0.7225	0.3605	-17.98	-17.45	16.62	0.03
C(20)	C(19)	2.250(19)	-21.230(53)	1.385	0.6703	0.7147	-18.69	-13.95	11.41	0.34
		2.160(4)	-18.587(10)	1.385	0.7217	0.6636	-15.38	-13.25	10.04	0.16

C(21)	C(21) C(20)	2.179(18)	-19.743(50)	1.388	0.6594	0.7292	-17.12	-13.63	11.00	0.26
		2.151(4)	-19.025(9)	1.388	0.6683	0.7206	-15.97	-12.95	9.89	0.23
C(21)	H(21)	1.813(30)	-16.225(96)	1.083	0.7282	0.3551	-17.76	-16.95	18.49	0.05
		1.856(7)	-17.247(21)	1.083	0.7033	0.3797	-17.14	-16.51	16.40	0.04
C(22)	H(22A)	1.856(33)	-17.029(97)	1.080	0.7069	0.3731	-17.78	-17.07	17.82	0.04
		1.944(9)	-21.325(28)	1.077	0.7154	0.3616	-19.20	-18.36	16.23	0.05
C(22)	H(22B)	1.856(36)	-16.763(112)	1.077	0.7070	0.3701	-17.80	-17.16	18.20	0.04
		1.945(9)	-20.835(26)	1.077	0.7086	0.3684	-19.10	-17.89	16.15	0.07
C(22)	H(22C)	1.820(35)	-16.501(113)	1.078	0.7222	0.3558	-17.78	-17.32	-18.6	0.03
		1.950(10)	-20.800(30)	1.077	0.7044	0.3727	-18.96	-17.95	16.11	0.06



Figure S14: BCPs (red spheres), RCPs (yellow spheres), and bond path (golden line) for the, C21–H21…F1–C11, and C2–H2A…F3–C17 hydrogen bonds based on experimental multipole model.





(b)



(c)

Figure S15: BCPs (red spheres), RCPs (yellow spheres), and bond path (golden line) for the (a) C12–F2…F4–C20 interaction, (b) C18–H18…F2–C12, (c) C2–H2A…F3–C17, and C21–H21…F1–C11 hydrogen bonds, plotted based on theoretical multipole model.



(a)



(b)

Figure S16: Experimental (left) and theoretical (right) Laplacian maps for (a) C2–H2A····F3–C17 and (b) C21–H21····F1–C11 hydrogen bonds, drawn at the logarithmic interval of $-\nabla^2 \rho \, e \text{\AA}^{-5}$.



Figure S17: Experimental (left) and theoretical (right) 3D Molecular Electrostatic potential isosurfaces plotted using a same scale.



Figure S18: Experimental (left) and theoretical (right) 3D ESP maps for the $C21-H21\cdots F1-C11$ and $C2-H2A\cdots F3-C17$ hydrogen bonds.