

Electronic Supplementary Information (ESI)

Chain-based fluorescent Tb^{III} metal-organic framework with good stability as a blue-shift and turn-on sensor toward H₂PO₄[−]

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Table S1. Crystal data and structure refinements for **JXUST-52**.

Identification code	JXUST-52
Empirical formula	C ₆₀ H ₃₂ N ₆ O ₁₆ S ₃ Tb ₃
Formula weight	1665.85
Temperature/K	293.15
Crystal system	Triclinic
Space group	$P\bar{1}$
$a/\text{\AA}$	8.7857(5)
$b/\text{\AA}$	18.8668(11)
$c/\text{\AA}$	20.7919(13)
$\alpha/^\circ$	113.758(2)
$\beta/^\circ$	95.709(2)
$\gamma/^\circ$	102.086(2)
Volume/ $\text{\AA}^3$	3018.7(3)
Z	2
D _{calc} g/cm ³	1.833
μ/mm^{-1}	3.654
Reflections collected/unique	44107/13756
R_{int}	0.0625
$R_1^{\text{a}}/wR_2^{\text{b}}[I \geq 2\sigma(I)]$	0.0562/0.1401
$R_1^{\text{a}}/wR_2^{\text{b}}[\text{all data}]$	0.0786/0.1535
Goodness-of-fit on F^2	1.036

$$^{\text{a}}R_1 = \sum(|F_0| - |F_c|) / \sum|F_0|; \quad ^{\text{b}}wR_2 = [\sum w(|F_0|^2 - |F_c|^2)^2 / (\sum w|F_0|^2)^2]^{1/2}$$

Table S2. Selected bond lengths (Å) and angles (°) for **JXUST-52^a**.

Tb1—O1 ⁱ	2.363(6)	O13 ^v —Tb2—O9 ^{vi}	75.92(17)
Tb1—O6 ⁱⁱⁱ	2.410(6)	O13—Tb2—O9 ^{vi}	143.91(17)
Tb1—O7	2.397(6)	O13—Tb2—O13 ^v	68.06(19)
Tb1—O9 ^{iv}	2.515(6)	O13 ^v —Tb2—O14	81.37(16)
Tb1—O14 ⁱⁱ	2.413(5)	O13—Tb2—O14	103.33(16)
Tb1—O14	2.375(5)	O13—Tb2—O15	73.34(16)
Tb1—O15	2.369(5)	O13 ^v —Tb2—O15	123.32(16)
Tb1—O11 ^v	2.343(6)	O12—Tb2—O4	74.6(2)
O1 ⁱ —Tb1—O6 ⁱⁱⁱ	153.9(2)	O12—Tb2—O5 ^{vii}	91.7(2)
O1 ⁱ —Tb1—O7	80.3(2)	O12—Tb2—O9 ^{vi}	86.2(2)
O1 ⁱ —Tb1—O9 ^{iv}	86.43(19)	O12—Tb2—O13	86.41(19)
O1 ⁱ —Tb1—O14	104.83(19)	O12—Tb2—O13 ^v	74.63(18)
O1 ⁱ —Tb1—O14 ⁱⁱ	73.81(19)	O12—Tb2—O14	148.52(19)
O1 ⁱ —Tb1—O15	69.70(18)	O12—Tb2—O15	142.06(19)
O6 ⁱⁱⁱ —Tb1—O9 ^{iv}	71.49(19)	O14—Tb2—O4	136.85(18)
O6 ⁱⁱⁱ —Tb1—O14 ⁱⁱ	84.84(19)	O14—Tb2—O9 ^{vi}	68.15(17)
O7—Tb1—O6 ⁱⁱⁱ	108.0(2)	O14—Tb2—O15	68.89(16)
O7—Tb1—O9 ^{iv}	79.44(19)	O15—Tb2—O4	68.08(17)
O7—Tb1—O14 ⁱⁱ	138.88(17)	O15—Tb2—O9 ^{vi}	128.52(17)
O14—Tb1—O6 ⁱⁱⁱ	79.66(18)	Tb3—O3	2.421(6)
O14—Tb1—O7	151.87(18)	Tb3—O4	2.551(6)
O14—Tb1—O9 ^{iv}	128.01(17)	Tb3—O8	2.300(6)
O14 ⁱⁱ —Tb1—O9 ^{iv}	67.72(16)	Tb3—O10 ^{ix}	2.318(5)
O14—Tb1—O14 ⁱⁱ	67.41(18)	Tb3—O13	2.345(5)
O15—Tb1—O6 ⁱⁱⁱ	133.57(19)	Tb3—O15	2.369(5)
O15—Tb1—O7	88.58(18)	Tb3—O1W	2.408(7)
O15—Tb1—O9 ^{iv}	154.86(18)	O2 ^{viii} —Tb3—O3	91.5(2)
O15—Tb1—O14 ⁱⁱ	110.88(16)	O2 ^{viii} —Tb3—O4	75.1(2)
O15—Tb1—O14	62.28(17)	O2 ^{viii} —Tb3—O8	141.0(2)
O11 ^v —Tb1—O1 ⁱ	133.5(2)	O2 ^{viii} —Tb3—O10 ^{ix}	78.6(2)
O11 ^v —Tb1—O6 ⁱⁱⁱ	72.1(2)	O2 ^{viii} —Tb3—O13	79.8(2)
O11 ^v —Tb1—O7	73.5(2)	O2 ^{viii} —Tb3—O15	139.6(2)
O11 ^v —Tb1—O9 ^{iv}	124.3(2)	O2 ^{viii} —Tb3—O1W	71.7(3)
O11 ^v —Tb1—O14	83.85(19)	O3—Tb3—O4	52.36(19)
O11 ^v —Tb1—O14 ⁱⁱ	146.07(19)	O8—Tb3—O3	93.5(2)
O11 ^v —Tb1—O15	71.8(2)	O8—Tb3—O4	135.5(2)
Tb2—O4	2.472(5)	O8—Tb3—O10 ^{ix}	85.6(2)
Tb2—O5 ^{vi}	2.305(6)	O8—Tb3—O13	130.4(2)
Tb2—O9 ^{vii}	2.555(5)	O8—Tb3—O15	78.7(2)

Tb2—O13 ^v	2.333(5)	O8—Tb3—O1W	71.6(2)
Tb2—O13	2.313(5)	O10 ^{ix} —Tb3—O3	162.5(2)
Tb2—O12	2.260(5)	O10 ^{ix} —Tb3—O4	135.86(18)
Tb2—O14	2.337(5)	O10 ^{ix} —Tb3—O13	75.72(18)
Tb2—O15	2.370(5)	O10 ^{ix} —Tb3—O15	120.8(18)
O4—Tb2—O9 ^{vi}	143.19(18)	O10 ^{ix} —Tb3—O1W	84.9(2)
O5 ^{vii} —Tb2—O4	75.1(2)	O13—Tb3—O3	116.96(19)
O5 ^{vii} —Tb2—O9 ^{vi}	74.4(2)	O13—Tb3—O4	65.20(17)
O5 ^{vii} —Tb2—O13	141.10(19)	O13—Tb3—O15	72.78(17)
O5 ^{vii} —Tb2—O13 ^v	148.00(19)	O13—Tb3—O1W	148.2(2)
O5 ^{vii} —Tb2—O14	98.4(2)	O15—Tb3—O3	75.9(2)
O5 ^{vii} —Tb2—O15	85.2(2)	O15—Tb3—O1W	139.0(2)
O13—Tb2—O4	66.97(17)	O1W—Tb3—O3	78.3(3)
O13 ^v —Tb2—O4	126.18(17)	O1W—Tb3—O4	118.5(2)

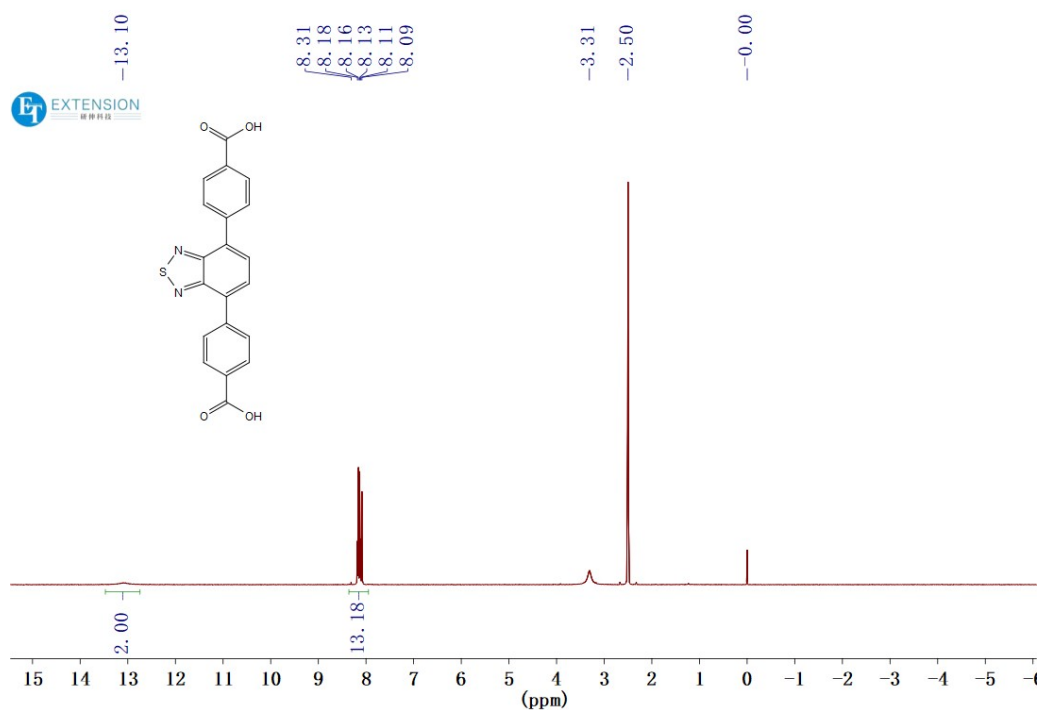
^aSymmetry codes: (i) 1-x, -y, 1-z; (ii) 1-x, -y, -z; (iii) 2-x, 1-y, 1-z; (iv) +x, 1+y, 1+z; (v) 2-x, -y, -z; (vi) -1+x, -1+y, -1+z; (vii) 1-x, -1-y, -1-z; (viii) 2-x, -y, -1-z; (ix) 1+x, 1+y, 1+z.

Table S3. The coordination configuration of Tb^{III} ions in **JXUST-52**.

Ions	Label	Shape	Symmetry	Distortion (τ)
Tb1	OP-8	Octagon	D_{8h}	30.754
	HPY-8	Heptagonal pyramid	C_{7v}	23.092
	HBPY-8	Hexagonal bipyramid	D_{6h}	15.865
	CU-8	Cube	O_h	10.983
	SAPR-8	Square antiprism	D_{4d}	3.878
	TDD-8	Triangular dodecahedron	D_{2d}	1.218
	JGBF-8	Johnson gyrobifastigium J26	D_{2d}	11.966
	JETBPY-8	Johnson elongated triangular bipyramid J14	D_{3h}	26.460
	JBTPR-8	Biaugmented trigonal prism J50	C_{2v}	2.520
	BTPR-8	Biaugmented trigonal prism	C_{2v}	2.565
	JSD-8	Snub diphennoid J84	D_{2d}	2.830
	TT-8	Triakis tetrahedron	T_d	11.405
	ETBPY-8	Elongated trigonal bipyramid	D_{3h}	24.329
	OP-8	Octagon	D_{8h}	32.138
	HPY-8	Heptagonal pyramid	C_{7v}	23.410
Tb2	HBPY-8	Hexagonal bipyramid	D_{6h}	12.456
	CU-8	Cube	O_h	6.524
	SAPR-8	Square antiprism	D_{4d}	1.499
	TDD-8	Triangular dodecahedron	D_{2d}	1.890
	JGBF-8	Johnson gyrobifastigium J26	D_{2d}	14.119
	JETBPY-8	Johnson elongated triangular bipyramid J14	D_{3h}	27.074
	JBTPR-8	Biaugmented trigonal prism J50	C_{2v}	3.305
	BTPR-8	Biaugmented trigonal prism	C_{2v}	2.753
	JSD-8	Snub diphennoid J84	D_{2d}	5.405
	TT-8	Triakis tetrahedron	T_d	7.122
	ETBPY-8	Elongated trigonal bipyramid	D_{3h}	23.765
	OP-8	Octagon	D_{8h}	29.753
	HPY-8	Heptagonal pyramid	C_{7v}	22.116
	HBPY-8	Hexagonal bipyramid	D_{6h}	13.858
	CU-8	Cube	O_h	8.921
Tb3	SAPR-8	Square antiprism	D_{4d}	2.233
	TDD-8	Triangular dodecahedron	D_{2d}	2.811
	JGBF-8	Johnson gyrobifastigium J26	D_{2d}	12.051
	JETBPY-8	Johnson elongated triangular bipyramid J14	D_{3h}	26.651
	JBTPR-8	Biaugmented trigonal prism J50	C_{2v}	2.068
	BTPR-8	Biaugmented trigonal prism	C_{2v}	1.582
	JSD-8	Snub diphennoid J84	D_{2d}	4.542
	TT-8	Triakis tetrahedron	T_d	9.599
	ETBPY-8	Elongated trigonal bipyramid	D_{3h}	21.862

Table S4. The sensing properties of LMOF with other reported sensors for anions.

LMOF	Fluorescent Response	Anions	Detection Limit	Reference
JXUST-52	Turn-on response	H_2PO_4^-	0.016 mM	This work
Eu-MOF	Turn-off response	H_2PO_4^-	0.70 mM	S1
JXUST-13	Turn-on response	H_2PO_4^-	2.70 $\mu\text{mol/L}$	S2
UiO-66- NH_2	Turn-on response	PO_4^{3-}	1.25 mM	S3
Cd-MOF	Turn-off response	ClO^-	0.18 μM	S4
Eu-MOF	Turn-off response	$\text{Cr}_2\text{O}_7^{2-}$	$1.14 \times 10^{-4} \text{ mol L}^{-1}$	S5
Zn-MOF	Turn-off response	CrO_4^{2-}	5.25 μM	S6
CP-1	Turn-off response	MnO_4^-	1.291 μM	S7
Al-MOF	Turn-on response	F^-	0.31 μM	S8

**Fig. S1** ^1H NMR spectrum of H_2BTDB .

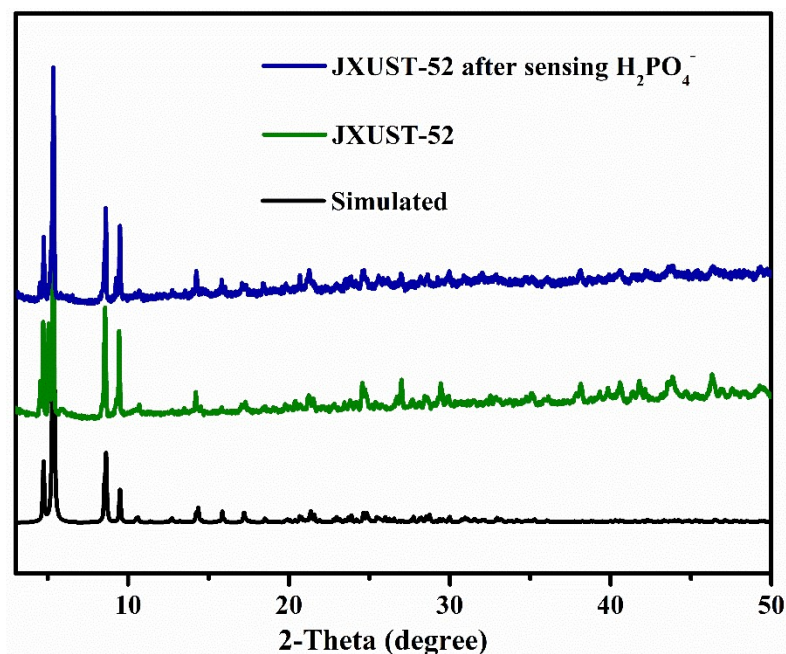


Fig. S2 The PXRD patterns of **JXUST-52** and **JXUST-52@ H_2PO_4^-** .

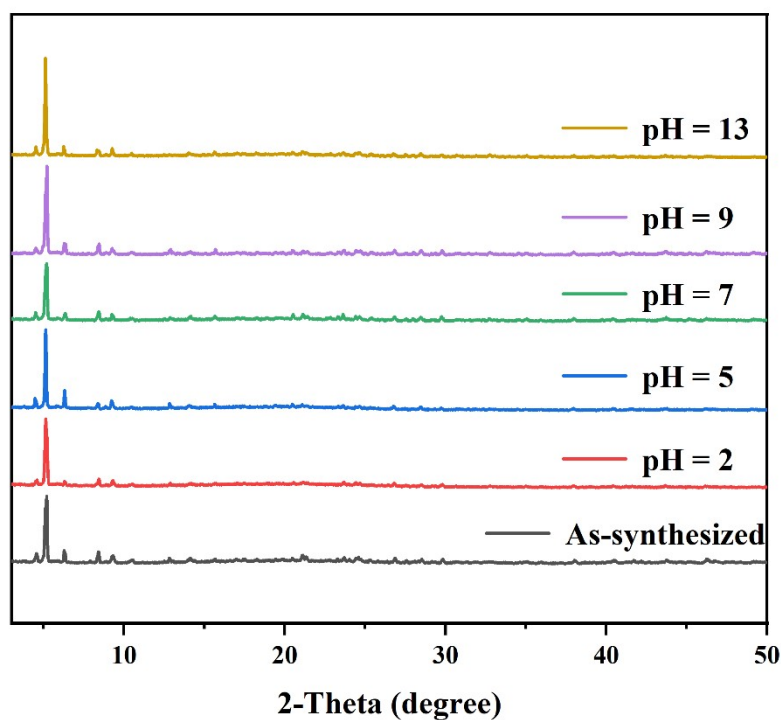


Fig. S3 The experimental PXRD patterns of **JXUST-52** immersed in aqueous solutions with different pH values for 24 hours.

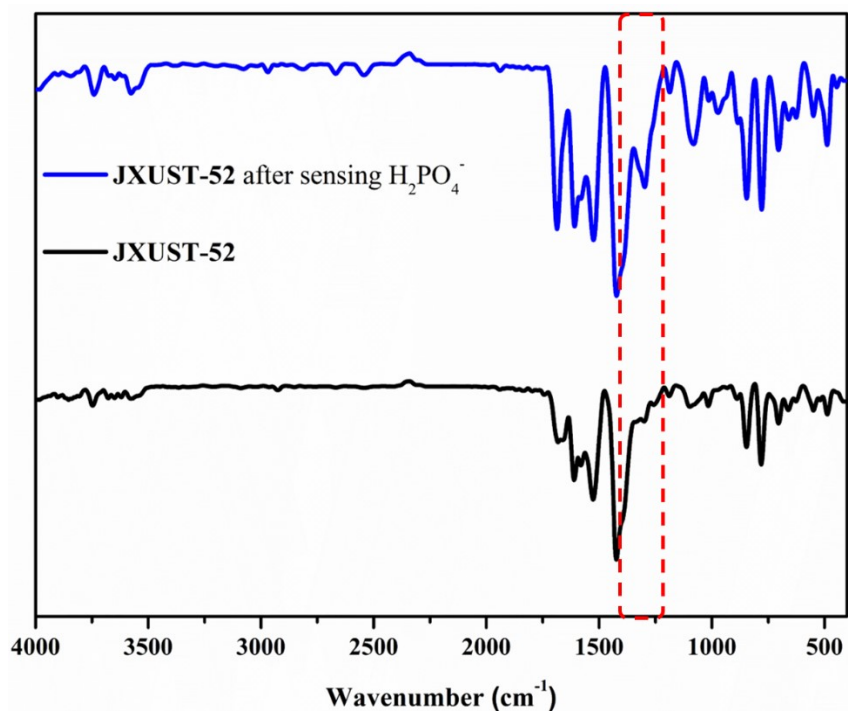


Fig. S4 IR spectra of JXUST-52 and JXUST-52@H₂PO₄⁻.

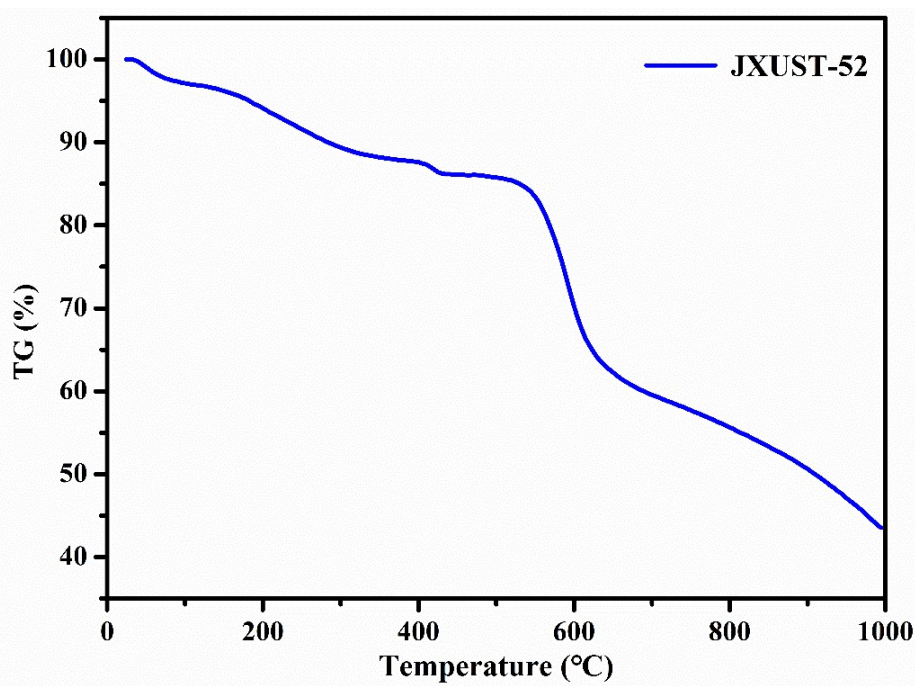


Fig. S5 The TGA curve of JXUST-52.

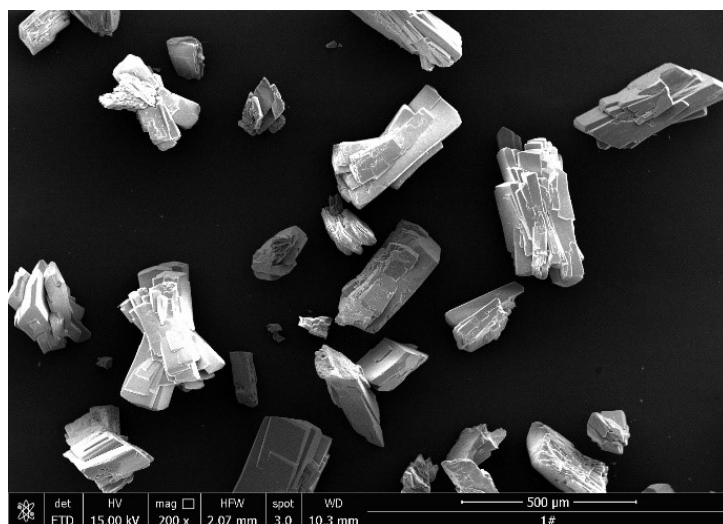


Fig. S6 The SEM image of the crystalline sample for **JXUST-52**.

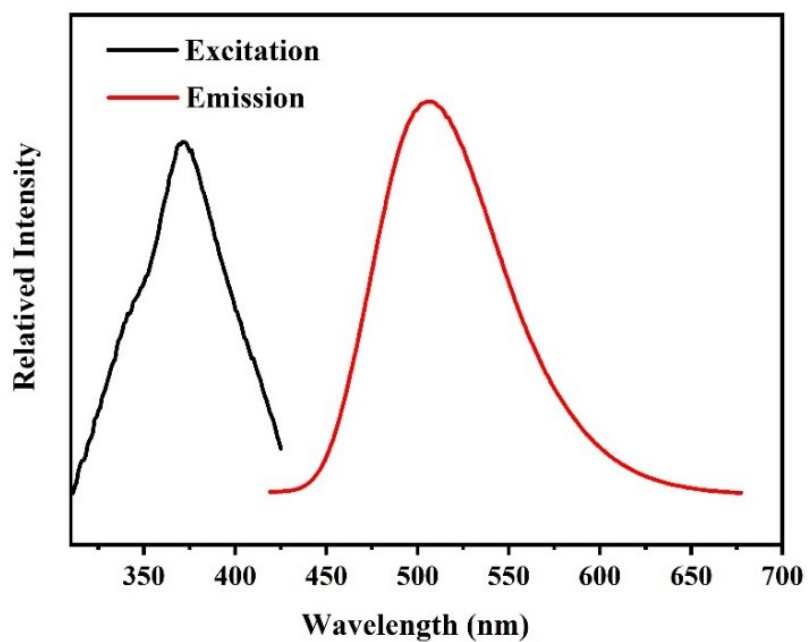


Fig. S7 The excitation and emission spectra of H_2BTDB .

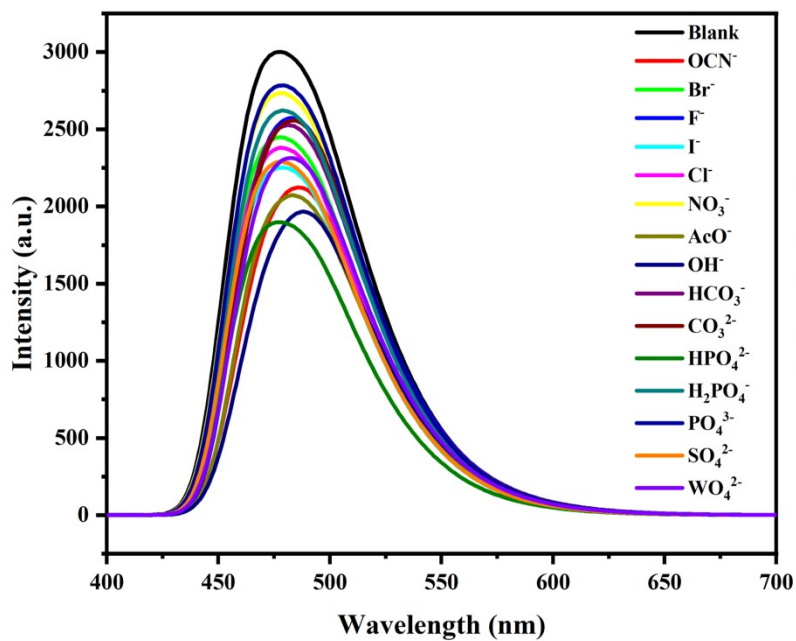


Fig. S8 The fluorescence intensities of H₂BTDB upon addition of different anions in DMF solutions.

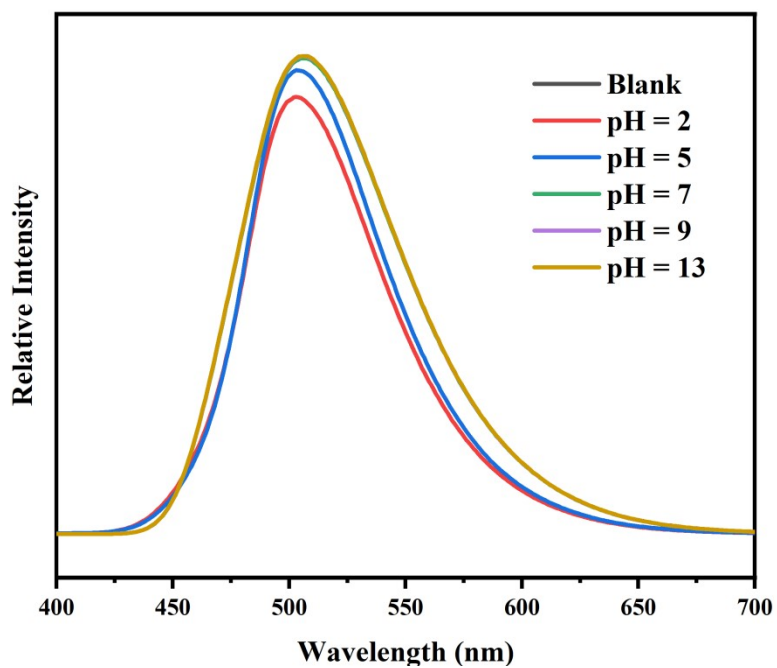


Fig. S9 The fluorescence intensities of JXUST-52 immersed in aqueous solution with different pH values.

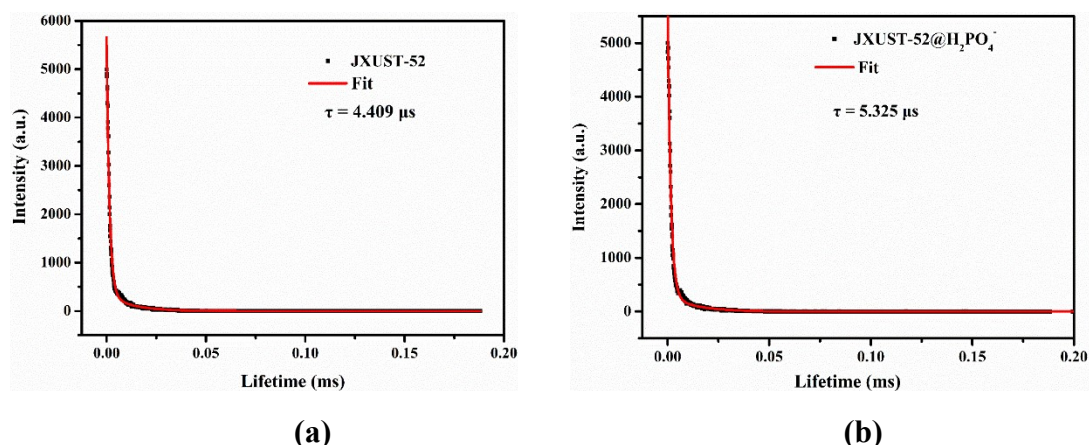


Fig. S10 The fluorescence lifetime decay diagrams of (a) **JXUST-52** and (b) **JXUST-52@H₂PO₄⁻**.

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