Supporting Information

Al-induced fast phase transition constructing vanadium oxide cathode materials

for high-performance aqueous zinc ion batteries



S1. Morphological characterization of V₂O₃.



S2. HRTEM image of AlVO-0.2 and undoped- V_2O_3 .



S3. Crystal structure schematic of pure V₂O₃ and AlVO.



S4. (a) XPS full survey spectra, (b) Al 2p and (c) V 2p of V₂O₃, AlVO-0.17 and

AlVO-0.33 samples.



S5. (a)CV curves of V_2O_3 at 0.1 mV s⁻¹, (b) Comparison of CV curves of V_2O_3 and



AlVO-0.2 at $0.1 \text{mV} \text{ s}^{-1}$.

S6. (a) Pseudocapacitance contribution ratio at the current density of 1.0 mV s⁻¹. (b)



Capacitance contribution at different scan rates of V₂O₃ cathode.

S7. EIS plots of V_2O_3 cathode in different stages.



8. Ex situ XRD analysis of the V₂O₃ cathode at various states



S9. EIS plots of V_2O_3 and AlVO-0.2 cathode before and after cycle.

Sample name	Lattice parameters (Å)	
V_2O_3	a = 4.945	c = 14.003
AlVO-0.14	a = 4.993	c = 13.867
AlVO-0.2	a = 4.984	c = 13.949
AlVO-0.33	a = 4.989	c = 13.991
PDF # 65-9474	a = 4.951	c = 14.003

Table SI. Differences in lattice parameters of V_2O_3 and AlVO.

Table SII. Comparison of electrochemical properties of different vanadium oxide

Cathode materials	Specific capacity	Remain capacity	Reference
	(capacity @ current density	(current density, cycles)	
	$[mAh g^{-1} @ A g^{-1}])$		
AlVO-0.2	383.5@0.1	282.8 (3.0 A g ⁻¹ , 1000)	This work
VO ₂	280.9@0.1	220 (1.0 A g ⁻¹ , 200)	[1]
V ₂ O ₃	207@0.1	110 (3.0 A g ⁻¹ , 2500)	[2]
V ₂ O ₃ @C	392@1.0	207 (5.0 A g ⁻¹ , 2000)	[3]
V ₂ O ₃ @N–C	342.5@0.1	274.6 (5.0 A g ⁻¹ , 2000)	[4]
$Al_xV_2O_5\cdot nH_2O/rGO$	365@0.1	175(4.0 A g ⁻¹ , 1300)	[5]
Al-doped HV ₆ O ₁₃	351@0.1	224(10.0 A g ⁻¹ , 1000)	[6]

applied to AZIBs.

Reference

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