

Structure, Magnetic Properties and Fluorescence Selectivity of 1D Chain Complexes Based on Pyrazolyl Nitronyl Nitroxide Radical

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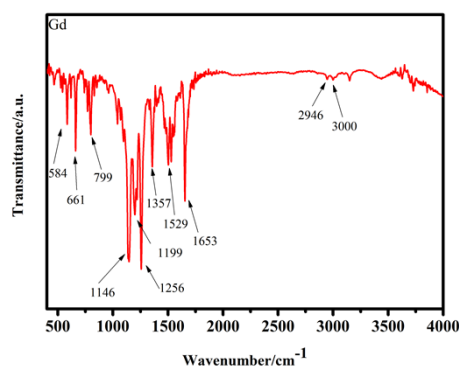


Fig. S1 The IR spectrum of complex 1.

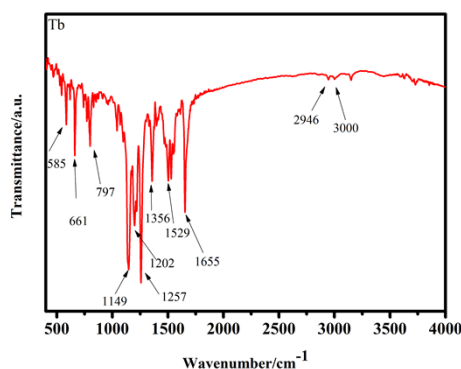


Fig. S2 The IR spectrum of complex 2.

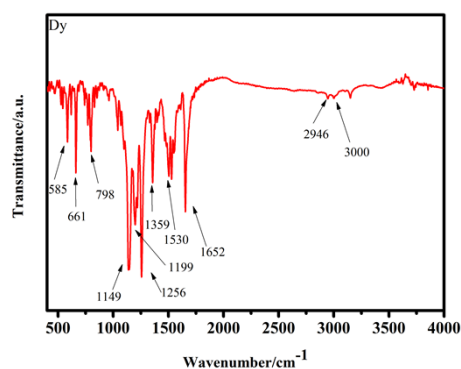


Fig. S3 The IR spectrum of complex 3.

Table S1 The infrared spectra for complexes **1-3**.

complex	Main IR absorption(cm^{-1})				
Gd	$\nu_{(\text{C-H})}=2946,3000$	$\nu_{(\text{C=O})}=1653$	$\nu_{(\text{N-O})}=1529$	$\nu_{(\text{N-O})}=1357$	$\nu_{(\text{C-N})}=1257$
	$\nu_{(\text{C-O})}=1199$	$\nu_{(\text{N-N})}=1146$	$\nu_{(\text{Benzene ring fingerprint region})}=799, 661, 584$		
Tb	$\nu_{(\text{C-H})}=2946,3000$	$\nu_{(\text{C=O})}=1655$	$\nu_{(\text{N-O})}=1529$	$\nu_{(\text{N-O})}=1356$	$\nu_{(\text{C-N})}=1256$
	$\nu_{(\text{C-O})}=1202$	$\nu_{(\text{N-N})}=1149$	$\nu_{(\text{Benzene ring fingerprint region})}=797, 661, 585$		
Dy	$\nu_{(\text{C-H})}=2946,3000$	$\nu_{(\text{C=O})}=1652$	$\nu_{(\text{N-O})}=1530$	$\nu_{(\text{N-O})}=1359$	$\nu_{(\text{C-N})}=1256$
	$\nu_{(\text{C-O})}=1199$	$\nu_{(\text{N-N})}=1149$	$\nu_{(\text{Benzene ring fingerprint region})}=798, 661, 585$		

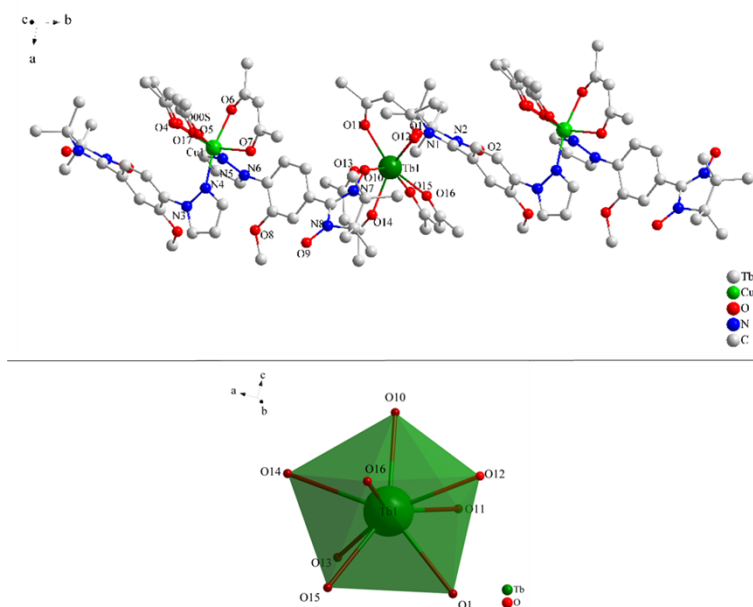


Fig. S4 Crystal structure of complex **2** (top) and coordination polyhedron of Tb (bottom)

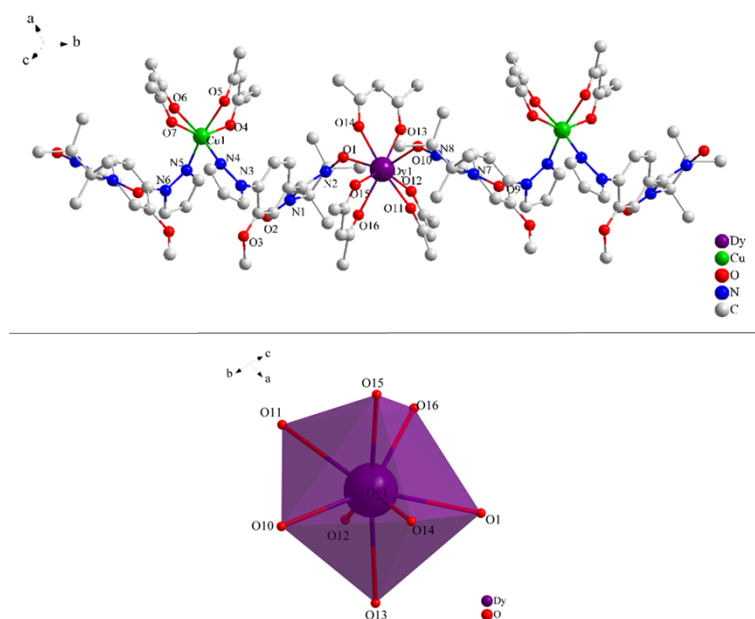


Fig. S5 Crystal structure of complex **3** (top) and coordination polyhedron of Dy (bottom)

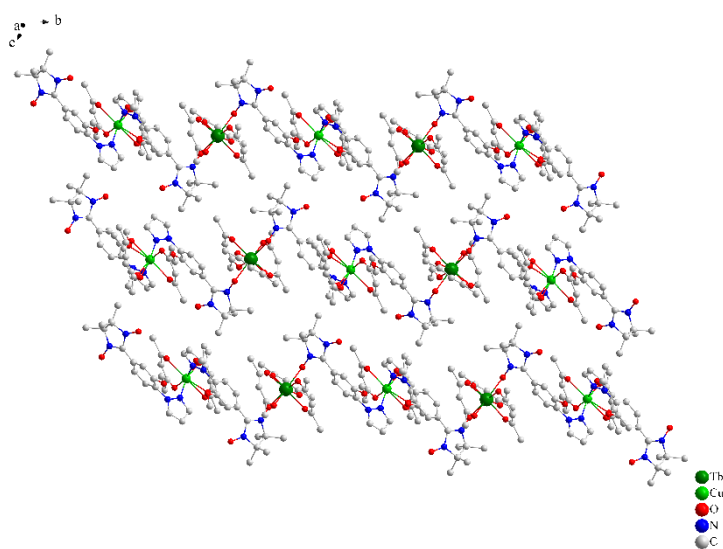


Fig. S6 Crystal packing diagram of complex **2**.

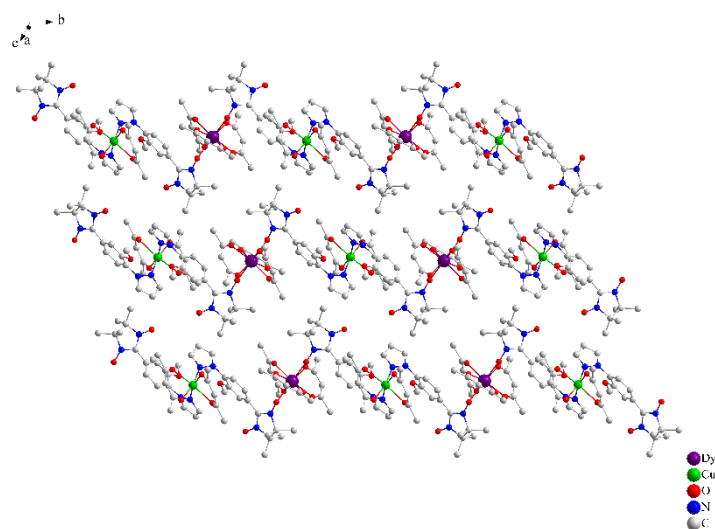


Fig. S7 Crystal packing diagram of complex **3**.

Table S2 Selected bond lengths (Å) and angles (°) for complex **1**.

<i>Bonds</i>			
Gd(1)-O(1)	2.369(5)	Cu(1)-N(5)	2.008(6)
Gd(1)-O(10)	2.330(5)	Cu(1)-O(4)	2.043(5)
Gd(1)-O(11)	2.366(5)	Cu(1)-O(5)	2.297(5)
Gd(1)-O(12)	2.366(5)	Cu(1)-O(6)	2.034(5)
Gd(1)-O(13)	2.416(5)	Cu(1)-O(7)	2.317(5)
Gd(1)-O(14)	2.353(4)	N(2)-O(2)	1.278(8)
Gd(1)-O(15)	2.427(5)	N(8)-O(9)	1.269(9)
Gd(1)-O(16)	2.353(5)	O(1)-N(1)	1.308(7)
Cu(1)-N(4)	2.036(6)	O(10)-N(7)	1.319(8)
<i>Angles</i>			
N(1)-O(1)-Gd(1)	135.1(4)	O(16)-Gd(1)-O(1)	94.57(18)
N(7)-O(10)-Gd(1)	134.9(4)	O(16)-Gd(1)-O(11)	75.40(18)
O(1)-Gd(1)-O(13)	73.42(17)	O(16)-Gd(1)-O(12)	147.45(18)
O(1)-Gd(1)-O(15)	146.11(16)	O(16)-Gd(1)-O(13)	73.65(17)
O(10)-Gd(1)-O(1)	140.01(18)	O(16)-Gd(1)-O(15)	72.15(17)
O(10)-Gd(1)-O(11)	72.77(18)	C(17)-N(4)-Cu(1)	125.6(5)
O(10)-Gd(1)-O(12)	75.71(18)	C(28)-N(5)-Cu(1)	125.4(5)

O(10)-Gd(1)-O(13)	146.25(18)	N(3)-N(4)-Cu(1)	128.4(4)
O(10)-Gd(1)-O(14)	92.45(17)	N(4)-Cu(1)-O(4)	90.4(2)
O(10)-Gd(1)-O(15)	73.87(17)	N(4)-Cu(1)-O(5)	116.0(2)
O(10)-Gd(1)-O(16)	101.66(17)	N(4)-Cu(1)-O(7)	83.8(2)
O(11)-Gd(1)-O(1)	76.46(18)	N(5)-Cu(1)-N(4)	92.8(2)
O(11)-Gd(1)-O(12)	72.86(19)	N(5)-Cu(1)-O(4)	166.1(2)
O(11)-Gd(1)-O(13)	134.33(17)	N(5)-Cu(1)-O(5)	85.1(2)
O(11)-Gd(1)-O(15)	126.78(18)	N(5)-Cu(1)-O(6)	89.5(2)
O(12)-Gd(1)-O(1)	71.21(18)	N(5)-Cu(1)-O(7)	115.2(2)
O(12)-Gd(1)-O(13)	126.01(17)	N(6)-N(5)-Cu(1)	128.7(4)
O(12)-Gd(1)-O(15)	134.63(17)	O(4)-Cu(1)-O(5)	81.4(2)
O(13)-Gd(1)-O(15)	72.93(16)	O(4)-Cu(1)-O(7)	78.6(2)
O(14)-Gd(1)-O(1)	99.88(17)	O(5)-Cu(1)-O(7)	151.9(2)
O(14)-Gd(1)-O(11)	146.93(18)	O(6)-Cu(1)-N(4)	164.3(2)
O(14)-Gd(1)-O(12)	74.88(17)	O(6)-Cu(1)-O(4)	91.0(2)
O(14)-Gd(1)-O(13)	72.56(16)	O(6)-Cu(1)-O(5)	79.6(2)
O(14)-Gd(1)-O(15)	73.65(17)	O(6)-Cu(1)-O(7)	81.2(2)
O(14)-Gd(1)-O(16)	137.41(17)		

Table S3 Selected bond lengths (Å) and angles (°) for complex **2**.

<i>Bonds</i>			
Tb(1)-O(1)	2.370(8)	Cu(1)-O(00S)	2.324(13)
Tb(1)-O(10)	2.320(8)	Cu(1)-O(4)	2.333(15)
Tb(1)-O(11)	2.352(9)	Cu(1)-O(5)	2.052(14)
Tb(1)-O(12)	2.345(8)	Cu(1)-O(6)	2.017(8)
Tb(1)-O(13)	2.348(7)	Cu(1)-O(7)	2.341(8)
Tb(1)-O(14)	2.407(8)	Cu(1)-O(17)	2.051(13)
Tb(1)-O(15)	2.372(8)	N(2)-O(2)	1.291(14)
Tb(1)-O(16)	2.346(7)	N(8)-O(9)	1.282(14)
Cu(1)-N(4)	2.024(10)	O(1)-N(1)	1.283(12)
Cu(1)-N(5)	1.986(10)	O(10)-N(7)	1.314(13)
<i>Angles</i>			

N(1)-O(1)-Tb(1)	135.4(8)	O(16)-Tb(1)-O(15)	73.6(3)
N(7)-O(10)-Tb(1)	134.9(7)	C(17)-N(4)-Cu(1)	125.7(8)
O(1)-Tb(1)-O(14)	147.2(3)	C(28)-N(5)-Cu(1)	124.2(10)
O(1)-Tb(1)-O(15)	73.8(3)	N(3)-N(4)-Cu(1)	128.0(7)
O(10)-Tb(1)-O(1)	138.7(3)	N(4)-Cu(1)-O(00S)	116.5(6)
O(10)-Tb(1)-O(11)	72.7(3)	N(4)-Cu(1)-O(17)	91.1(10)
O(10)-Tb(1)-O(12)	74.8(3)	N(4)-Cu(1)-O(4)	114.6(13)
O(10)-Tb(1)-O(13)	102.3(3)	N(4)-Cu(1)-O(5)	88.8(13)
O(10)-Tb(1)-O(14)	74.1(3)	N(4)-Cu(1)-O(7)	84.1(3)
O(10)-Tb(1)-O(15)	147.2(3)	N(5)-Cu(1)-N(4)	93.4(4)
O(10)-Tb(1)-O(16)	91.9(3)	N(5)-Cu(1)-O(00S)	80.9(6)
O(11)-Tb(1)-O(1)	75.4(3)	N(5)-Cu(1)-O(17)	163.3(10)
O(11)-Tb(1)-O(14)	127.4(3)	N(5)-Cu(1)-O(4)	90.6(8)
O(11)-Tb(1)-O(15)	133.9(3)	N(5)-Cu(1)-O(5)	169.7(13)
O(12)-Tb(1)-O(1)	70.7(3)	N(5)-Cu(1)-O(6)	89.6(4)
O(12)-Tb(1)-O(11)	71.6(3)	N(5)-Cu(1)-O(7)	116.4(3)
O(12)-Tb(1)-O(13)	147.2(3)	N(6)-N(5)-Cu(1)	128.7(7)
O(12)-Tb(1)-O(14)	134.5(3)	O(00S)-Cu(1)-O(7)	153.3(6)
O(12)-Tb(1)-O(15)	126.6(3)	O(17)-Cu(1)-O(00S)	82.7(10)
O(12)-Tb(1)-O(16)	75.1(3)	O(17)-Cu(1)-O(7)	80.0(11)
O(13)-Tb(1)-O(1)	94.8(3)	O(4)-Cu(1)-O(7)	146.9(7)
O(13)-Tb(1)-O(11)	76.3(3)	O(5)-Cu(1)-O(4)	79.4(13)
O(13)-Tb(1)-O(14)	72.3(3)	O(5)-Cu(1)-O(7)	73.8(14)
O(13)-Tb(1)-O(15)	73.0(3)	O(6)-Cu(1)-N(4)	164.9(3)
O(15)-Tb(1)-O(14)	73.5(3)	O(6)-Cu(1)-O(00S)	78.7(10)
O(16)-Tb(1)-O(1)	100.4(3)	O(6)-Cu(1)-O(17)	90.1(12)
O(16)-Tb(1)-O(11)	146.0(3)	O(6)-Cu(1)-O(4)	80.1(12)
O(16)-Tb(1)-O(13)	137.5(3)	O(6)-Cu(1)-O(5)	90.8(15)
O(16)-Tb(1)-O(14)	73.5(2)	O(6)-Cu(1)-O(7)	81.3(3)

Table S4 Selected bond lengths (Å) and angles (°) for complex **3**.

<i>Bonds</i>			
Dy(1)-O(1)	2.345(5)	Cu(1)-N(5)	2.001(7)
Dy(1)-O(10)	2.321(5)	Cu(1)-O(4)	2.307(7)
Dy(1)-O(11)	2.396(5)	Cu(1)-O(5)	2.039(7)
Dy(1)-O(12)	2.345(5)	Cu(1)-O(6)	2.021(6)
Dy(1)-O(13)	2.337(6)	Cu(1)-O(7)	2.314(6)
Dy(1)-O(14)	2.352(5)	N(1)-O(2)	1.272(10)
Dy(1)-O(15)	2.348(5)	N(7)-O(9)	1.277(10)
Dy(1)-O(16)	2.386(5)	O(1)-N(2)	1.294(8)
Cu(1)-N(4)	2.050(7)	O(10)-N(8)	1.311(9)
<i>Angles</i>			
N(2)-O(1)-Dy(1)	135.5(5)	O(14)-Dy(1)-O(16)	126.15(18)
N(8)-O(10)-Dy(1)	134.9(5)	O(15)-Dy(1)-O(11)	73.69(18)
O(1)-Dy(1)-O(11)	146.5(2)	O(15)-Dy(1)-O(14)	74.6(2)
O(1)-Dy(1)-O(12)	94.23(19)	O(15)-Dy(1)-O(16)	72.54(18)
O(1)-Dy(1)-O(14)	71.61(19)	O(16)-Dy(1)-O(11)	73.15(19)
O(1)-Dy(1)-O(15)	100.08(19)	C(17)-N(4)-Cu(1)	125.2(6)
O(1)-Dy(1)-O(16)	73.61(19)	C(28)-N(5)-Cu(1)	126.5(6)
O(10)-Dy(1)-O(1)	139.2(2)	N(3)-N(4)-Cu(1)	128.2(5)
O(10)-Dy(1)-O(11)	73.6(2)	N(4)-Cu(1)-O(4)	115.5(2)
O(10)-Dy(1)-O(12)	101.95(19)	N(4)-Cu(1)-O(7)	84.0(2)
O(10)-Dy(1)-O(13)	73.1(2)	N(5)-Cu(1)-N(4)	93.0(3)
O(10)-Dy(1)-O(14)	75.42(19)	N(5)-Cu(1)-O(4)	84.9(3)
O(10)-Dy(1)-O(15)	92.5(2)	N(5)-Cu(1)-O(5)	165.2(3)
O(10)-Dy(1)-O(16)	146.2(2)	N(5)-Cu(1)-O(6)	90.4(3)
O(12)-Dy(1)-O(11)	72.37(19)	N(5)-Cu(1)-O(7)	116.3(3)
O(12)-Dy(1)-O(14)	147.8(2)	N(6)-N(5)-Cu(1)	128.2(5)
O(12)-Dy(1)-O(15)	137.30(19)	O(4)-Cu(1)-O(7)	151.3(2)
O(12)-Dy(1)-O(16)	73.40(19)	O(5)-Cu(1)-N(4)	90.5(3)
O(13)-Dy(1)-O(1)	76.0(2)	O(5)-Cu(1)-O(4)	80.6(3)

O(13)-Dy(1)-O(11)	126.64(18)	O(5)-Cu(1)-O(7)	78.4(2)
O(13)-Dy(1)-O(12)	75.2(2)	O(6)-Cu(1)-N(4)	164.7(3)
O(13)-Dy(1)-O(14)	73.4(2)	O(6)-Cu(1)-O(4)	79.6(2)
O(13)-Dy(1)-O(15)	147.3(2)	O(6)-Cu(1)-O(5)	90.0(3)
O(13)-Dy(1)-O(16)	133.9(2)	O(6)-Cu(1)-O(7)	81.1(2)
O(14)-Dy(1)-O(11)	134.0(2)		

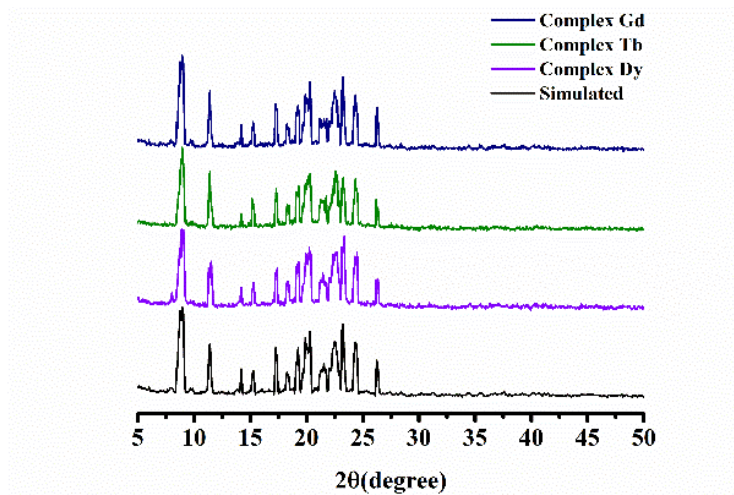


Fig. S8 Powder X-ray diffraction patterns of complexes **1-3**.

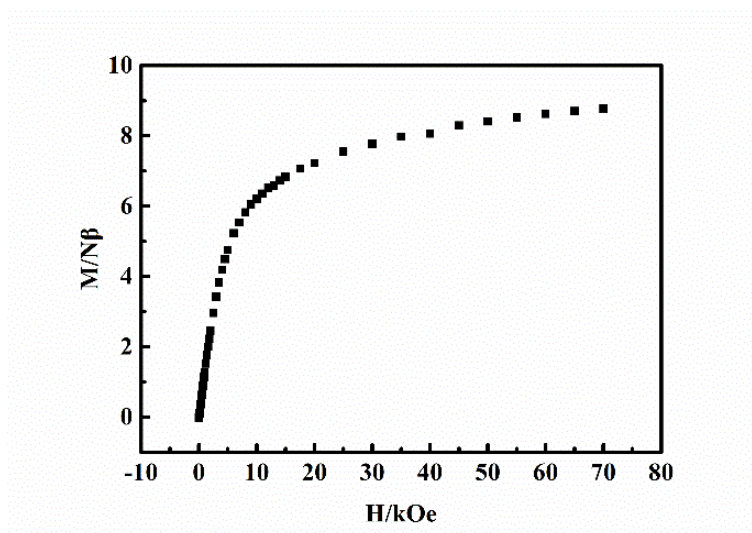


Fig. S9 The M vs. H plots of complex **2** at 2.0 K

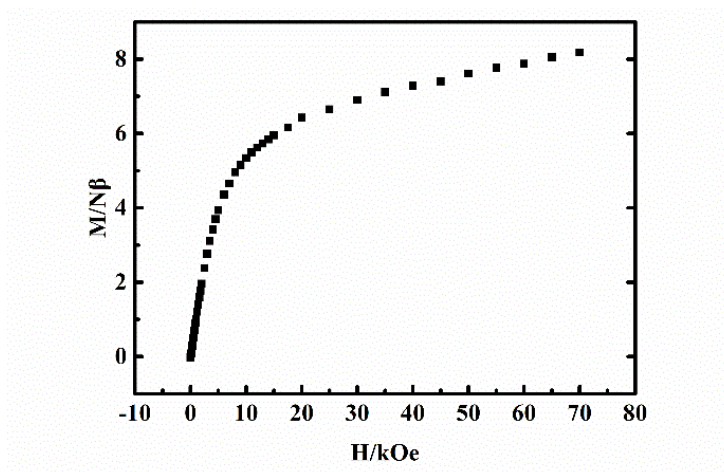


Fig. S10 The M vs. H plots of complex **3** at 2.0 K

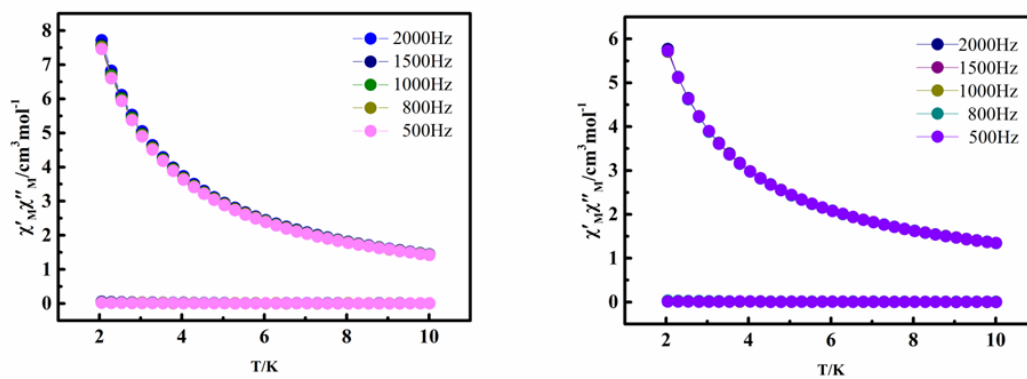


Fig. S11 Temperature dependence of the in-phase and out-of-phase components of the ac magnetic susceptibility for complex **2** (left) and **3** (right) in zero dc fields.

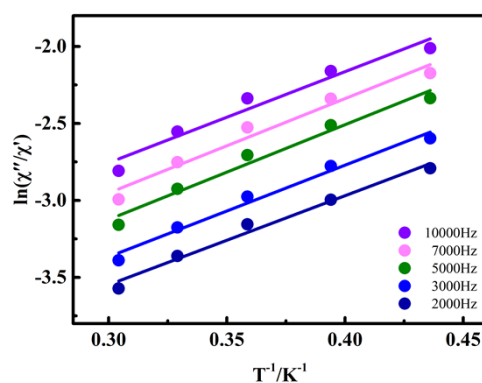


Fig. S12 Extractive $\ln(\chi''/\chi')$ vs $1/T$ plot for complex **2** (solid-lines: fitting curves).

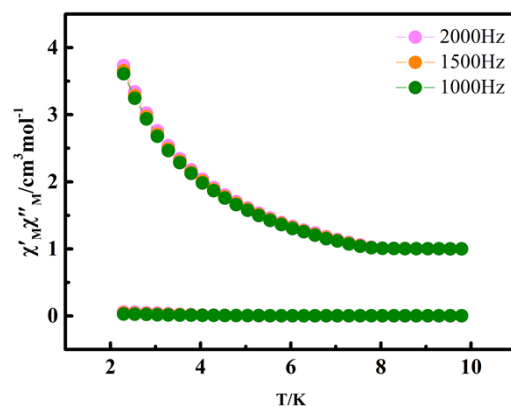


Fig. S13 Temperature-dependent ac signals for **3** at a 3kOe dc field.

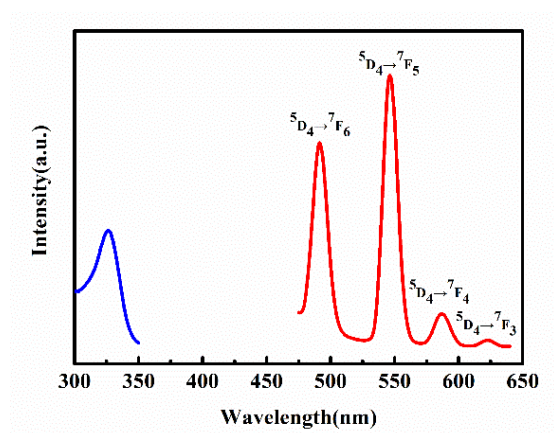


Fig. S14 Excitation (blue) and emission (red) spectra of complex **2** in aqueous solution.

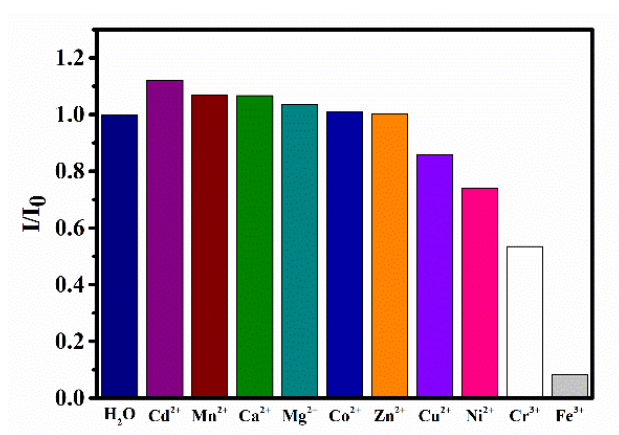


Fig. S15 Luminescence intensity of the transition (546 nm) of complex **2** in different cations suspension system. I and I_0 denote the fluorescence intensity of complex **2** in cation suspension system and water suspension system, respectively.

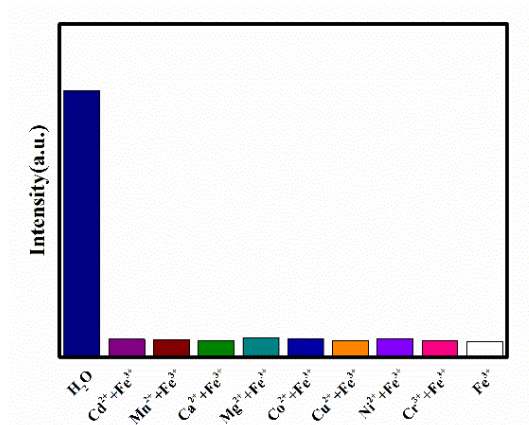


Fig. S16 Comparison of the luminescence intensity of **2** in the presence of mixed cations.

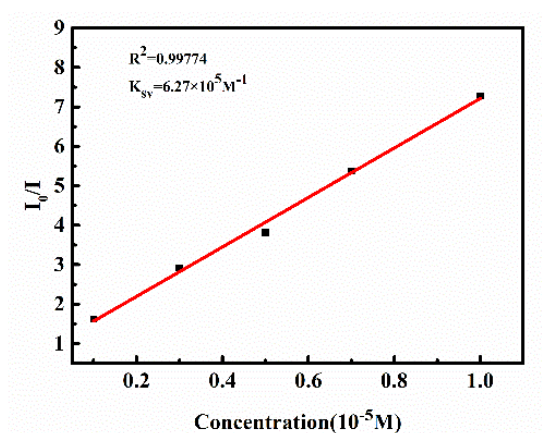


Fig. S17 The Stern-Volmer plot of **2** in the presence of $\text{Cr}_2\text{O}_7^{2-}$, the red line is linear fitting.

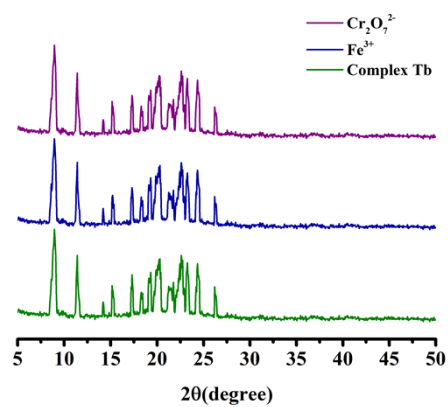


Fig. S18 PXRD patterns of complex **2** after soaking in $\text{Cr}_2\text{O}_7^{2-}$ and Fe^{3+} ions.

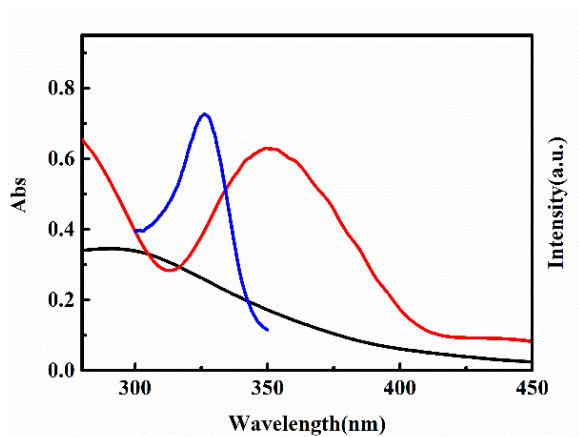


Fig. S19 The UV-vis absorption spectra of $\text{Cr}_2\text{O}_7^{2-}$ (red) and Fe^{3+} (black), and the excitation spectra of complex **2** (blue).

Table S5 Comparison of $\text{Cr}_2\text{O}_7^{2-}$ ion detection limits for Complex **2** and some fluorescent materials.

Complexes	$K_{\text{sv}}(\text{M}^{-1})$	LOD(M)	Ref.
In/Eu-CBDA ^[a]	1.08×10^4	2.15×10^{-4}	1
$[\text{Eu}(\text{L})(\text{HCOO})(\text{H}_2\text{O})]_n$ ^[b]	2762.6	1×10^{-5}	2
$[\text{Eu}(\text{HL})(\text{H}_2\text{O})_2(\text{NO}_3)] \cdot \text{NO}_3$ ^[c]	7.52×10^4	1.7×10^{-5}	3
$\{[\text{Eu}(\text{BCEbpy})(\text{H}_2\text{O})_4] [\text{Co}^{\text{III}}(\text{CN})_6] \cdot 4\text{H}_2\text{O}\}_n$ ^[d]	4.31×10^3	9.6×10^{-6}	4
$[\text{Eu}_2(\text{phen})_2(\text{bpdc})_3(\text{H}_2\text{O})_2]_n \cdot 6\text{nH}_2\text{O}$ ^[e]	5.85×10^4	6.76×10^{-6}	5
$\{[\text{Eu}_2\text{L}_2(\text{DMF})_4]\}_n$ ^[f]	1.13×10^5	4.01×10^{-6}	6
$[\text{Eu}(\text{hfac})_3(\text{NITPh-DOMe})_2]$ ^[g]	2.339×10^4	1×10^{-7}	7
$[\text{Tb}(\text{HL})(\text{H}_2\text{O})_2(\text{NO}_3)] \cdot \text{NO}_3$ ^[c]	4.99×10^4	2.5×10^{-5}	3
$[\text{Tb}(\text{L})(\text{HCOO})(\text{H}_2\text{O})]_n$ ^[b]	2133.5	2.1×10^{-5}	2
In/Tb-CBDA ^[a]	1.72×10^4	8.72×10^{-6}	1
$[\text{Tb}_2(\text{phen})_2(\text{bpdc})_3(\text{H}_2\text{O})_2]_n \cdot 6\text{nH}_2\text{O}$ ^[e]	6.97×10^4	6.73×10^{-6}	5
$\{[\text{Tb}_3(\text{bcbp})_3(\text{NO}_3)_7] \cdot \text{NO}_3 \cdot \text{ClO}_4\}_n$ ^[h]	1.40×10^4	5.6×10^{-6}	8
$[\text{Tb}(\text{Hbptc})(\text{H}_2\text{O})_4] \cdot \text{H}_2\text{O}$ ^[i]	1.27×10^4	2.36×10^{-6}	9
$[\text{Tb}(\text{hfac})_3(\text{NIT-Ph-3-COOMe})_2] \cdot 0.5\text{C}_7\text{H}_{16}$ ^[j]	6.13×10^5	1.01×10^{-7}	10
$[\text{Tb}(\text{hfac})_3(\text{NITPh-(OCH}_3)_2)_2]$ ^[k]	2×10^4	1×10^{-7}	11
$[\text{Tb}(\text{hfac})_3(\text{NITPh-Pa})_2][0.5\text{CH}_3(\text{CH}_2)_5\text{CH}_3]$ ^[l]	19860.68	1×10^{-8}	12
$[\text{TbCu}(\text{hfac})_5(\text{NIT-4-OMe-3PyzPh})_2]_n$	6.27×10^5	2.92×10^{-7}	Our work

- [a] CBDA = 5,5'-(carbonylbis(azanediyl))-diisophthalic acid
- [b] H₂L = 5-((2'-cyano-[1,1'-biphenyl]-4-yl)methoxy)isophthalic acid
- [c] H₂L = 4-(3,5-dicarboxylphenyl)-2-methylpyridine
- [d] BCEbpy = N, N'-bis(carboxymethyl)- 4,4'-bipyridinium dibromide
- [e] phen = π -conjugated 1,10-phenanthroline, H₂bpdc = 2,2' -bipyridine-3,3' -dicarboxylate
- [f] H₃L = 5-(4-carboxybenzyloxy)isophthalic acid
- [g] NITPh-DOMe = 2-(3, 4 -dioxylmethylene-phenyl)-4,4,5,5-tetramethyl-imidazoline-1-oxyl-3-oxide
- [h] H₂bcbpCl₂ = 1,1'-bis(4-carboxyphenyl)(4,4'-bipyridinium) dichloride
- [i] H₄bptc = 2, 3, 3', 4'-biphenyl tetracarboxylic acid
- [j] NITPh-(OCH₃)₂ = 2-(3',4'-dimethoxy-phenyl)-4,4,5,5-tetramethyl-imidazoline-1-oxyl-3-oxide
- [k] NIT-Ph-3-COOMe = 2-(3-(methoxycarbonyl)-phenyl)-4,4,5,5-tetramethyl-imidazoline-1-oxyl-3-oxide
- [l] NITPh-Pa = 2-(3',4'-dioxylmethylene-phenyl)-4,4,5,5-tetramethyl-imidazoline-1-oxyl-3-oxide

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