

Carbon-coated $\text{Ni}_{0.5}\text{Mg}_{0.5}\text{Fe}_{1.7}\text{Mn}_{0.3}\text{O}_4$ as novel anode material for high energy density lithium-ion batteries

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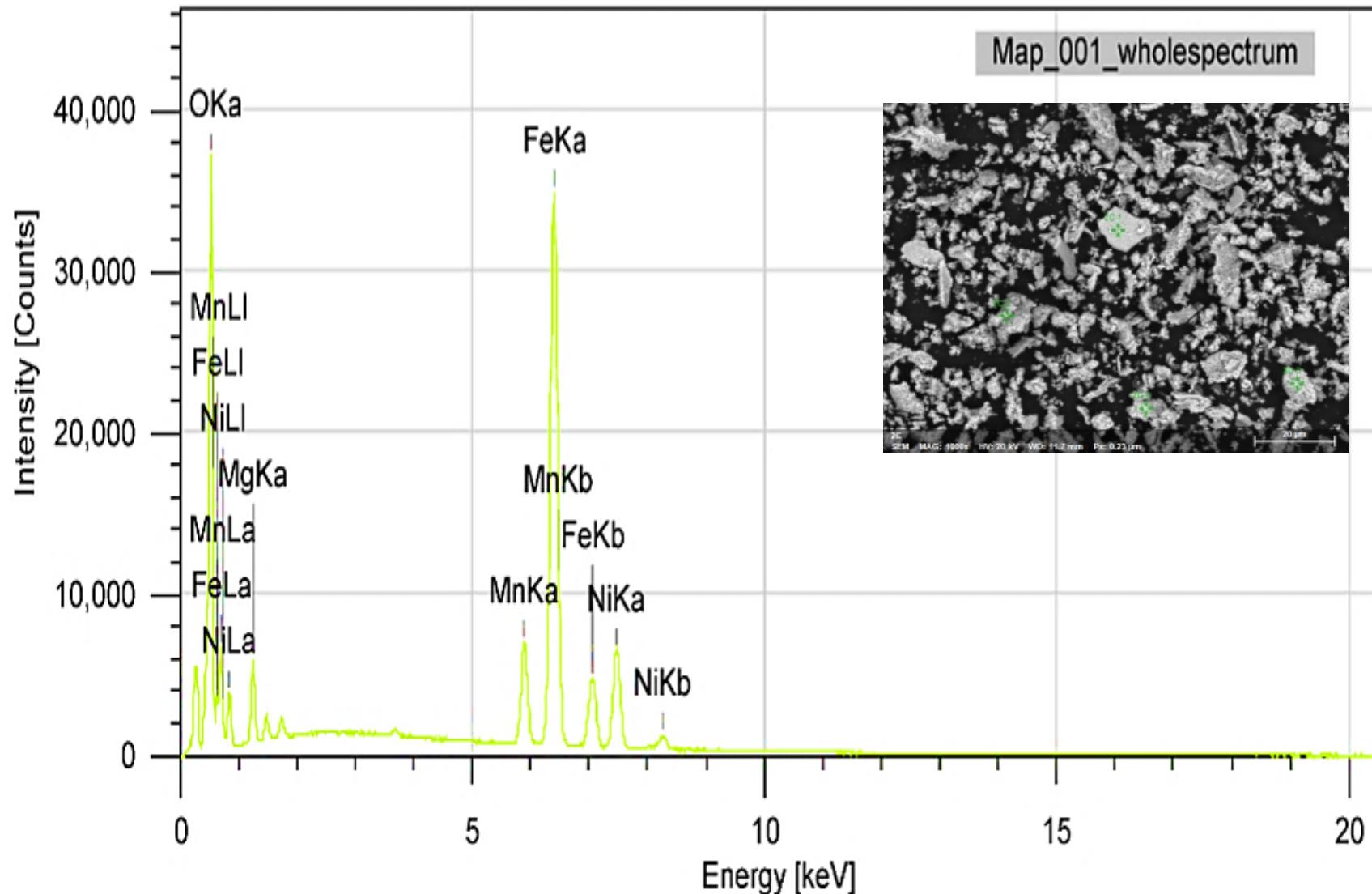


Fig. S1. EDX spectra of $\text{Ni}_{0.5}\text{Mg}_{0.5}\text{Fe}_{1.7}\text{Mn}_{0.3}\text{O}_4$ material

Table S1. The results of the Rietveld analysis for structural parameters, Chi-squared factor (χ^2), and various reliability R-factors for the synthesized $\text{Ni}_{0.5}\text{Mg}_{0.5}\text{Fe}_{1.7}\text{Mn}_{0.3}\text{O}_4$ material.

Material	$\text{Ni}_{0.5}\text{Mg}_{0.5}\text{Fe}_{1.7}\text{Mn}_{0.3}\text{O}_4$										
Wavelengths (Å)	$\lambda_{\text{k}\alpha 1} = 1.54056$ $\lambda_{\text{k}\alpha 2} = 1.54439$										
Crystal structure	Cubic structure										
Space group	<table border="1"> <tr><td>Wavelengths (Å)</td><td>$\lambda_{\text{k}\alpha 1} = 1.54056$</td></tr> <tr><td>Crystal structure</td><td>Cubic structure</td></tr> <tr><td>Lattice parameters</td><td>$a = 8.36257 \text{ \AA}$</td></tr> <tr><td>R-factors</td><td>$R_p = 8.01$ $R_{wp} = 10.9$ $R_{exp} = 11.29$</td></tr> <tr><td>χ^2</td><td>2.47</td></tr> </table>	Wavelengths (Å)	$\lambda_{\text{k}\alpha 1} = 1.54056$	Crystal structure	Cubic structure	Lattice parameters	$a = 8.36257 \text{ \AA}$	R-factors	$R_p = 8.01$ $R_{wp} = 10.9$ $R_{exp} = 11.29$	χ^2	2.47
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Crystal structure	Cubic structure										
Lattice parameters	$a = 8.36257 \text{ \AA}$										
R-factors	$R_p = 8.01$ $R_{wp} = 10.9$ $R_{exp} = 11.29$										
χ^2	2.47										
Lattice parameters	$a = 8.36257 \text{ \AA}$ $V = 584.817 \text{ \AA}^3$										
R-factors	$R_p = 8.01$ $R_{wp} = 10.9$ $R_{exp} = 11.29$										
χ^2	2.47										

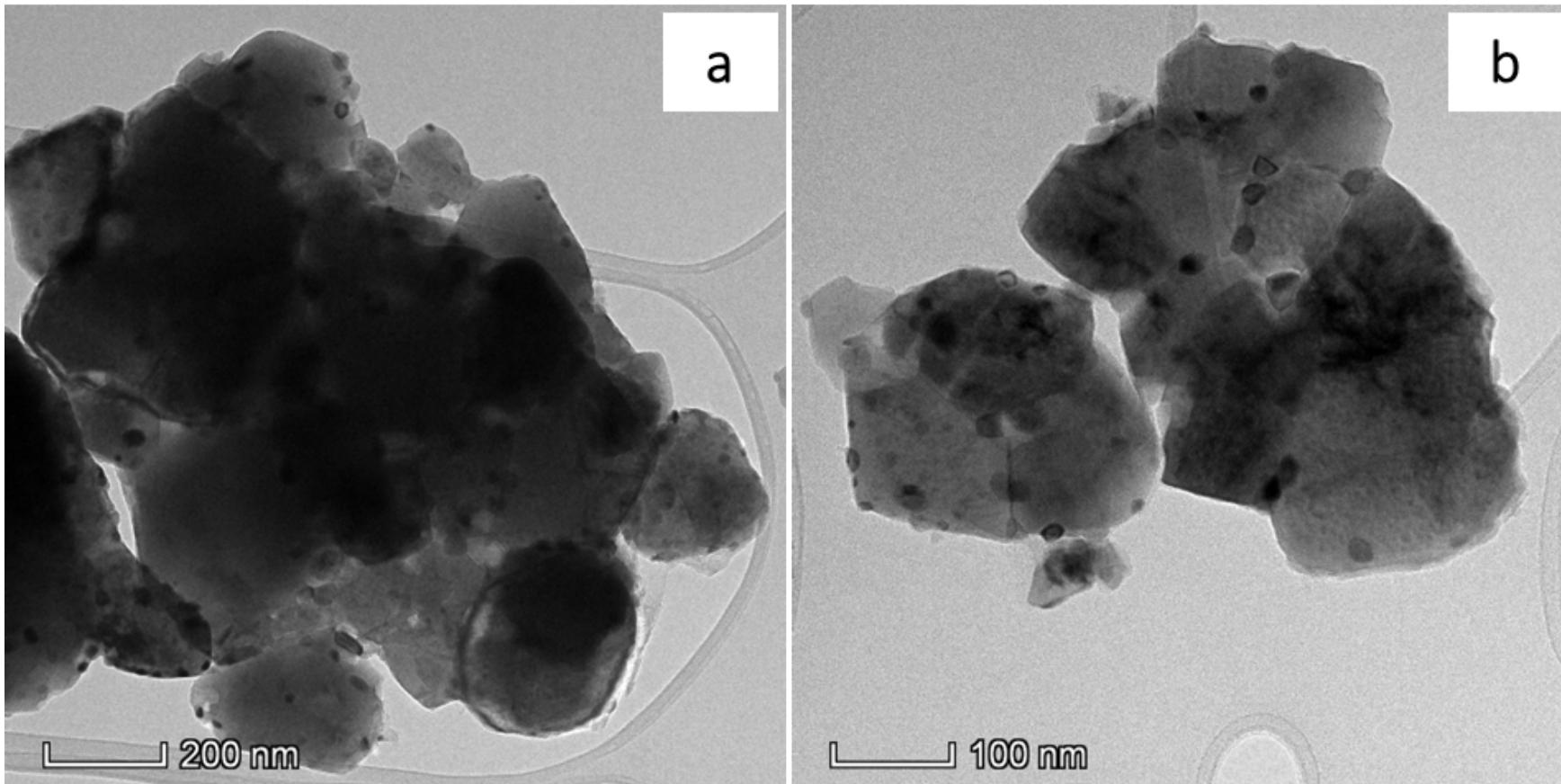


Fig. S2. TEM images of the uncoated $\text{Ni}_{0.5}\text{Mg}_{0.5}\text{Fe}_{1.7}\text{Mn}_{0.3}\text{O}_4$ sample.

Table S2. Comparison of the capacities for the 75th cycles at 0.125C, 100th cycles at 2C (100 mA. g⁻¹ = 0.125C), and rate capability performance of uncoated and carbon-coated Ni_{0.5}Mg_{0.5}Fe_{1.7}Mn_{0.3}O₄ (the unit of capacity: mAh. g⁻¹).

Anode material	Capacity of the 75 th cycle at 0.125 C	Capacity of the 100 th cycle at 2C	Capacities at various C-rates						
			0.03C	0.06C	0.125C	0.25C	0.625C	1.25C	2C
Ni _{0.5} Mg _{0.5} Fe _{1.7} Mn _{0.3} O ₄	744.3	36	809	739	710	681	637	593	192
Ni _{0.5} Mg _{0.5} Fe _{1.7} Mn _{0.3} O ₄ /C	819.1	343	720	778	760	731	690	651	372

Table S3. EIS fitting data of uncoated $\text{Ni}_{0.5}\text{Mg}_{0.5}\text{Fe}_{1.7}\text{Mn}_{0.3}\text{O}_4$ and coated $\text{Ni}_{0.5}\text{Mg}_{0.5}\text{Fe}_{1.7}\text{Mn}_{0.3}\text{O}_4/\text{C}$ electrodes before and after 100 cycles at 2C (100 mA. g⁻¹ = 0.125C).

Anode material		R_s (Ω)	R_f (Ω)	R_{ct} (Ω)	R_{total}
$\text{Ni}_{0.5}\text{Mg}_{0.5}\text{Fe}_{1.7}\text{Mn}_{0.3}\text{O}_4$	After rest	4.45	1012	18 646	19 662.5
	After 100 cycles	25.8	944.9	2145	3115.7
$\text{Ni}_{0.5}\text{Mg}_{0.5}\text{Fe}_{1.7}\text{Mn}_{0.3}\text{O}_4/\text{C}$	After rest	3.98	162.4	10 237	10 403.4
	After 100 cycles	3.59	51.68	400	455.3