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Modulation of Luminescence Properties of Circularly Polarized Thermally Activated Delayed Fluorescence Molecules with Axial Chirality by Donor Engineering

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Figure S1. The energy levels of molecules of Cz-Ax-CN, DMAC-Ax-CN, PTZ-Ax-CN and PXZ-Ax-CN in toluene.



Figure S2. Natural transition orbitals (NTOs) of the S₁ and T₁ states for (a) Cz-Ax-CN and (b) DMAC-Ax-CN in toluene, respectively.



Figure S3. Natural transition orbitals (NTOs) of the S₁ and T₁ states for (a) PTZ-Ax-CN and (b) PXZ-Ax-CN in toluene, respectively.



Figure S4. Rates constants of Radiative and Non-Radiative from S_1 to S_0 as well as the ISC and RISC rates between S_1 and T_1 for all studied molecules in toluene.

HOMO (eV)	LUMO (eV)	$\epsilon_{gap}\left(eV\right)$	Overlap
-6.01	-1.25	4.75	28.31%
-5.70	-1.18	4.52	26.43%
-5.48	-1.31	4.18	25.58%
-5.46	-1.24	4.23	25.60%
	HOMO (eV) -6.01 -5.70 -5.48 -5.46	HOMO (eV)LUMO (eV)-6.01-1.25-5.70-1.18-5.48-1.31-5.46-1.24	HOMO (eV)LUMO (eV)εgap (eV)-6.01-1.254.75-5.70-1.184.52-5.48-1.314.18-5.46-1.244.23

Table S1. The overlap and orbital energies of HOMO and LUMO are listed, the ϵ_{gap} is the energy gap between the HOMO and LUMO.

Table S2. The data of absorption asymmetry factor (g_{abs}) for Cz-Ax-CN, DMAC-Ax-CN, PTZ-Ax-CN and PXZ-Ax-CN in toluene.

	$ \mu $ (×10 ⁻¹⁸ esu·cm)	m (×10 ⁻²⁰ erg·G ⁻¹)	COS O	$g_{\rm abs}$ (×10 ⁻²)
Cz-Ax-CN	1.450	0.532	-1	-1.47
DMAC-Ax-CN	1.108	0.190	-1	-6.87
PTZ-Ax-CN	0.703	0.392	-1	-2.23
PXZ-Ax-CN	0.980	0.296	-1	-1.21