

Supporting Information
for

Theoretical insights into dopamine photochemistry adsorbed on graphene-type nanostructures

by

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1. UV-Vis adsorption and excited state relaxation for dopamine-o-quinone (DAQ)

Table S1. The excitation energy (in eV and nm in parenthesis), oscillator strength and the orbital transition with the largest coefficient computed for the first 14 electronic excited states of the dopamine-o-quinone at TDDFT/ ω B97X-D3BJ/def2-TZVPP level of theory (H = HOMO and L = LUMO).

	S_1	S_2	S_3	S_4	S_5	S_6	S_7
Energy	2.36 (526)	3.79 (327)	3.93 (315)	5.44 (228)	6.19 (200)	6.38 (194)	6.53 (190)
Osc. Str.	0.00003	0.00001	0.07752	0.00001	0.00019	0.17737	0.00000
Orb. Trans.	H \rightarrow L	H - 3 \rightarrow L	H - 1 \rightarrow L	H - 2 \rightarrow L	H \rightarrow L + 1	H - 4 \rightarrow L	H \rightarrow L + 2
	S_8	S_9	S_{10}	S_{11}	S_{12}	S_{13}	S_{14}
Energy	6.82 (182)	6.89 (180)	7.12 (174)	7.37 (168)	7.64 (162)	7.77 (160)	7.80 (159)
Osc. Str.	0.02025	0.00012	0.00038	0.27744	0.78035	0.00614	0.06234
Orb. Trans.	H - 2 \rightarrow L + 3	H - 6 \rightarrow L	H - 8 \rightarrow L	H - 1 \rightarrow L + 1	H - 1 \rightarrow L + 2	H - 12 \rightarrow L	H - 7 \rightarrow L

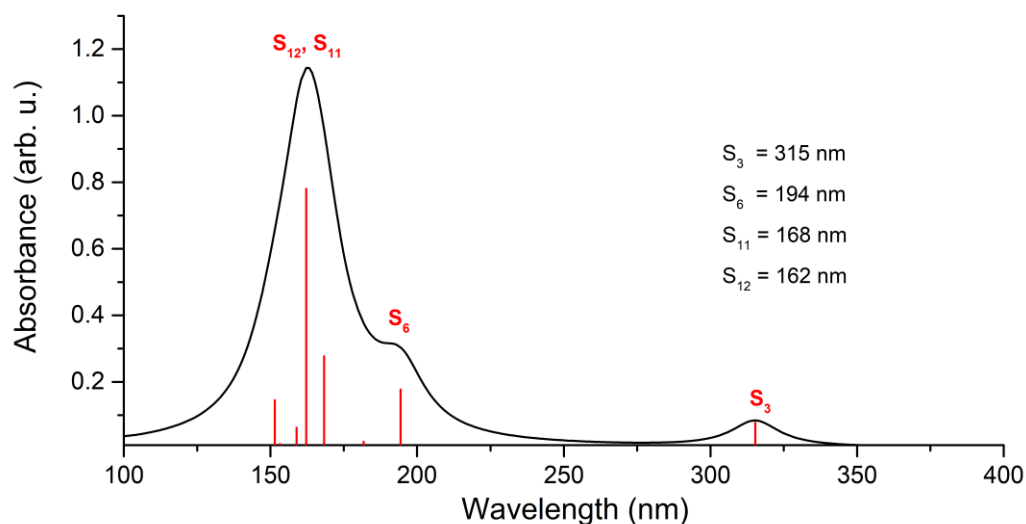


Figure S1. Theoretical UV-absorption spectra for DAQ, computed at TDDFT/ ω B97X-D3BJ/def2-TZVPP level of theory.

2. Electronic excited states and theoretical UV-absorption spectra for GrNP- and GrS-type graphene structures

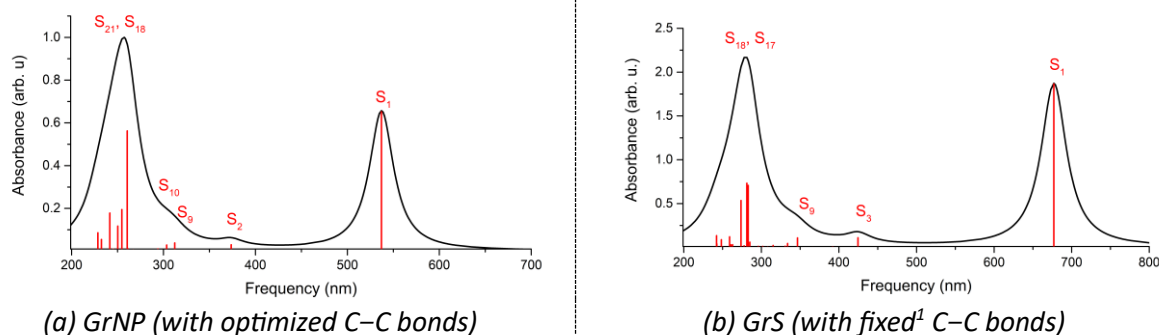


Figure S2. Theoretical UV-absorption spectra for GrNP- (a) and GrS-type (b) graphene nanoparticle geometries, computed at TDDFT/ ω B97X-D3BJ/def2-TZVPP level of theory.

Table S2. The excitation energy, oscillator strength and the largest orbital transition coefficients of the first 30 electronic excited states computed for the GrNP- and GrS-type graphene nanoparticle geometries at TDDFT/ ω B97X-D3BJ/def2-TZVPP level of theory (H = HOMO and L = LUMO). The most relevant electronic excited states are marked with red color.

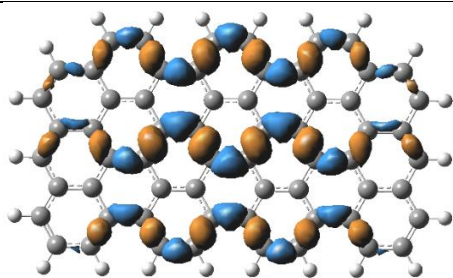
GrNP				GrS			
Nr.	Freq. (nm)	Strength	orb. trans.	Nr.	Freq. (nm)	Strength	orb. trans.
S_1	537.1	1.7945	H \rightarrow L	S_1	677.0	1.8644	H \rightarrow L
S_2	373.7	0.0823	H \rightarrow L+1 H-1 \rightarrow L	S_2	435.0	0.0007	H \rightarrow L+2
S_3	373.2	0.0002	H \rightarrow L+2	S_3	424.7	0.1096	H \rightarrow L+1
S_4	363.2	0.0009	H-3 \rightarrow L	S_4	399.9	0.0011	H-3 \rightarrow L H \rightarrow L+3
S_5	355.9	0.0000	H \rightarrow L+3	S_5	399.4	0.0001	H-2 \rightarrow L
S_6	339.3	0.0017	H \rightarrow L+5	S_6	391.9	0.0061	H-1 \rightarrow L
S_7	324.9	0.0006	H-1 \rightarrow L H \rightarrow L+1	S_7	366.0	0.0020	H \rightarrow L+4
S_8	321.6	0.0055	H-2 \rightarrow L	S_8	364.9	0.0002	H \rightarrow L+5
S_9	312.3	0.1052	H \rightarrow L+7	S_9	346.6	0.1074	H \rightarrow L+6 H \rightarrow L+7
S_{10}	303.5	0.0790	H-5 \rightarrow L	S_{10}	333.9	0.0446	H-6 \rightarrow L
S_{11}	302.7	0.0100	H \rightarrow L+7 H-2 \rightarrow L	S_{11}	330.6	0.0131	H \rightarrow L+7
S_{12}	294.0	0.0282	H-1 \rightarrow L+5 H \rightarrow L+8	S_{12}	315.2	0.0218	H-5 \rightarrow L H \rightarrow L+8
S_{13}	284.5	0.0150	H \rightarrow L+9	S_{13}	308.2	0.0051	H \rightarrow L+9
S_{14}	281.1	0.0032	H-4 \rightarrow L	S_{14}	301.8	0.0105	H-4 \rightarrow L
S_{15}	267.2	0.0201	H-6 \rightarrow L	S_{15}	296.2	0.0115	H \rightarrow L+8 H-8 \rightarrow L

¹ The length of each C–C bond was fixed to 1.42 Å.

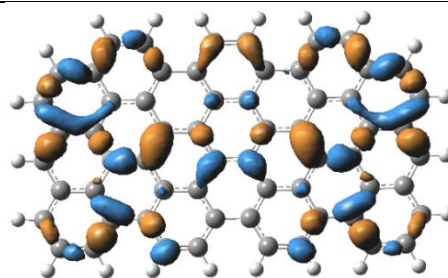
S ₁₆	265.6	0.0041	H-5 → L H-7 → L	S ₁₆	285.2	0.0585	H-5 → L H → L+7
S ₁₇	261.3	0.0310	H-9 → L	S ₁₇	283.2	0.7067	H-3 → L
S ₁₈	260.6	1.5476	H → L+4	S ₁₈	281.3	0.7330	H-7 → L
S ₁₉	258.7	0.0096	H-3 → L+2	S ₁₉	277.8	0.0177	H → L+11
S ₂₀	257.4	0.0000	H-9 → L	S ₂₀	273.8	0.5347	H → L+10
S ₂₁	254.8	0.5355	H → L+10	S ₂₁	271.5	0.0024	H-9 → L
S ₂₂	252.0	0.0046	H-2 → L+3	S ₂₂	266.3	0.0001	H-2 → L+4 H → L+13
S ₂₃	250.3	0.3219	H → L+11	S ₂₃	263.0	0.0302	H-3 → L+2 H-2 → L+3
S ₂₄	245.2	0.0256	H-10 → L	S ₂₄	260.9	0.0289	H-1 → L+1
S ₂₅	241.6	0.4908	H-2 → L+4	S ₂₅	259.1	0.1202	H-10 → L
S ₂₆	238.4	0.0014	H-2 → L+5	S ₂₆	258.6	0.0000	H-7 → L H → L+9
S ₂₇	232.5	0.1506	H-1 → L+4	S ₂₇	250.2	0.0052	H-10 → L H-2 → L+5
S ₂₈	232.3	0.0022	H-1 → L+2 H-2 → L+1	S ₂₈	248.5	0.0873	H-1 → L+4
S ₂₉	229.3	0.0204	H-1 → L+3	S ₂₉	242.6	0.0393	H-1 → L+4
S ₃₀	228.7	0.2366	H-3 → L+1 H-1 → L+4	S ₃₀	242.2	0.1338	H-1 → L+5

Table S3. The natural difference orbitals (NDO) between the ground and the given electronic excited state computed for the the GrNP- and GrS-type graphene geometries at TDDFT/ ω B97X-D3BJ/def2-TZVPP level of theory.

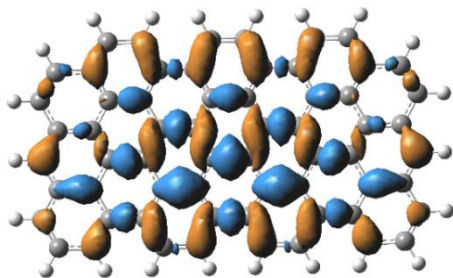
GrNP:



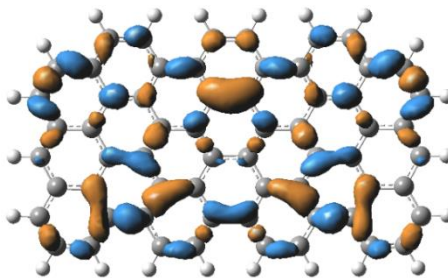
S₁



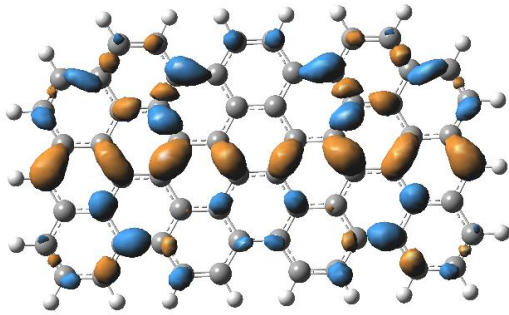
S₂



S₉

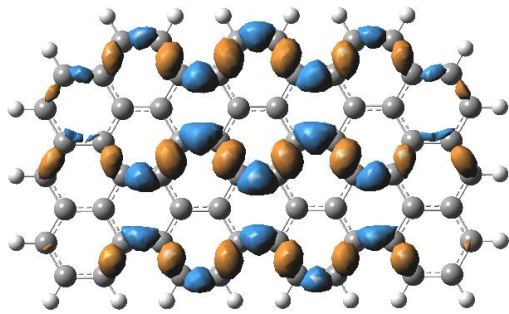


S₁₀

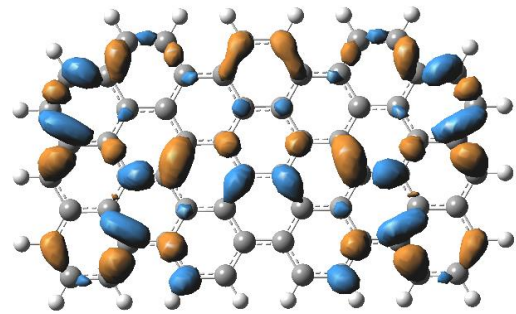


S_{18}

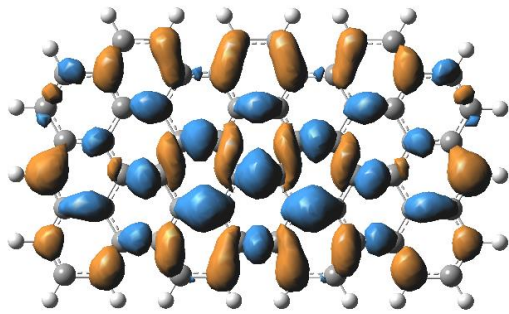
GrS:



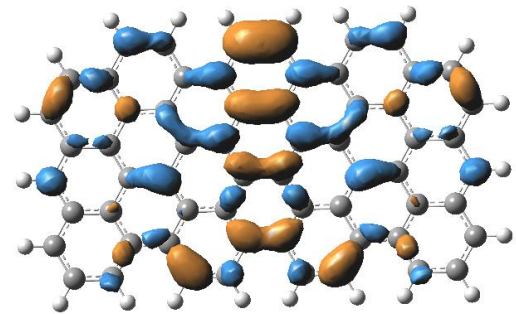
S_1



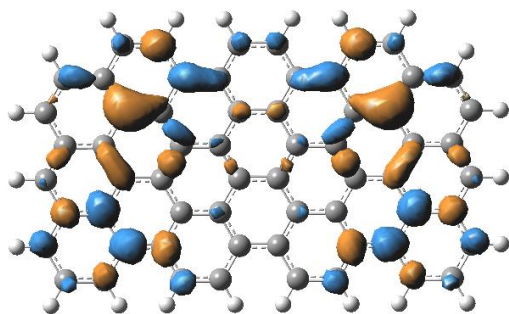
S_3



S_9



S_{17}



S_{18}

3. Electronic excited states and theoretical UV-absorption spectra for GrNP – DA binary complex.

Table S4. The frontier (HOMO and LUMO) orbitals of the isolated DA computed at ω B97X-D3BJ/def2-TZVPP level of theory.

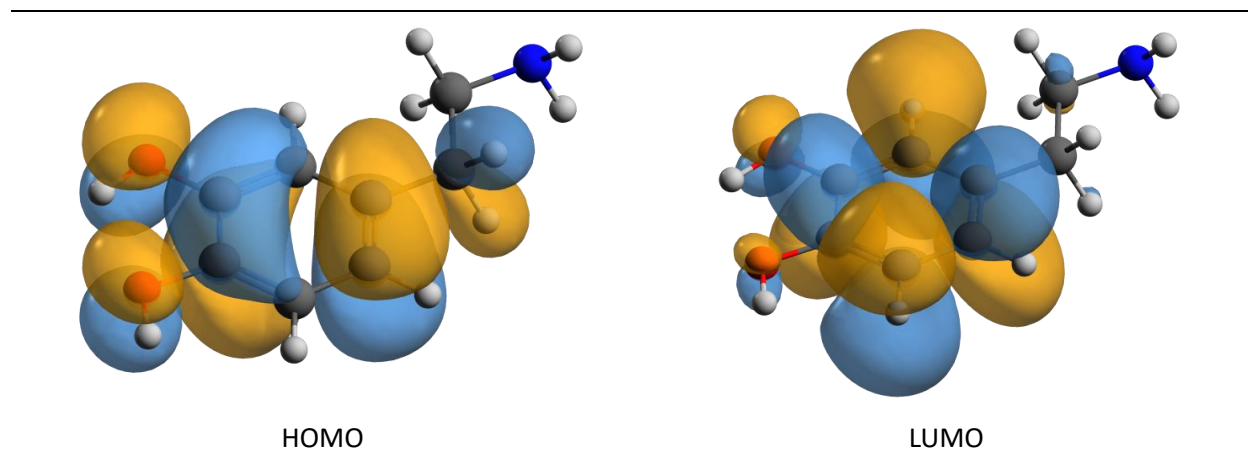
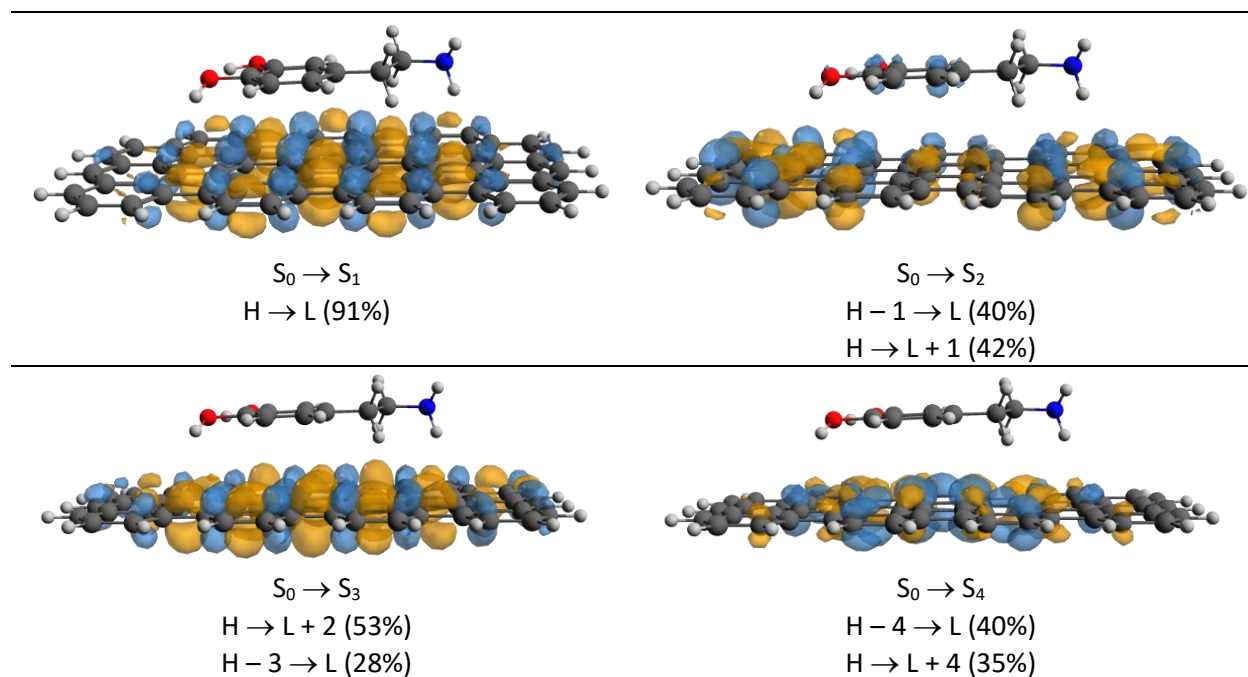
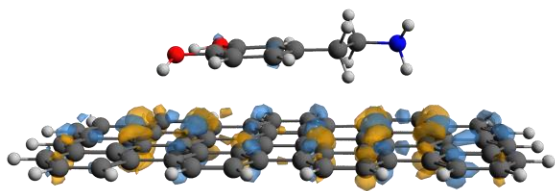
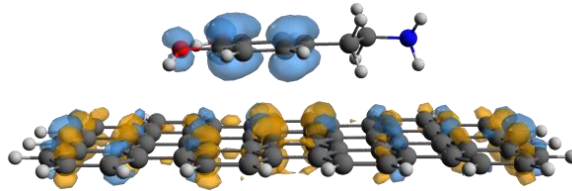


Table S5. The natural difference orbitals (NDO) between the ground and the given electronic excited state computed for the GrNP-DA binary complex at TDDFT/ ω B97X-D3BJ/def2-TZVPP level of theory (Blue = hole, Yellow = electron, orbital orders with red text color are specific to DA and blacks are graphene-types).

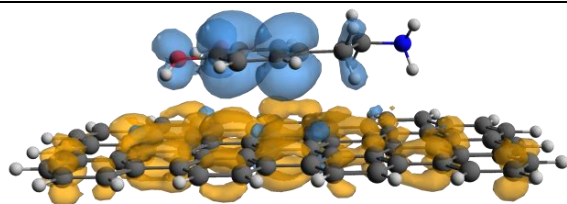




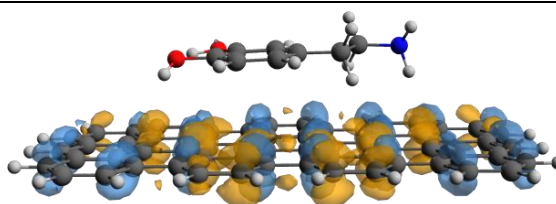
$S_0 \rightarrow S_5$
 $H \rightarrow L + 3$ (40%)
 $H - 5 \rightarrow L$ (17%)



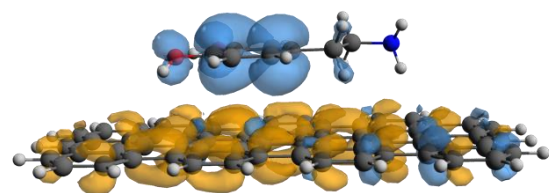
$S_0 \rightarrow S_6$
 $H \rightarrow L + 5$ (37%)
 $H - 1 \rightarrow L$ (17%)



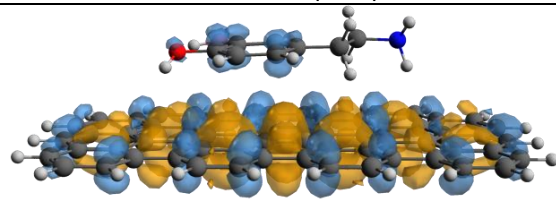
$S_0 \rightarrow S_7$
 $H - 2 \rightarrow L$ (26%)
 $H \rightarrow L + 5$ (21%)



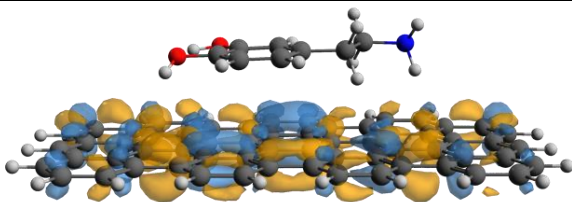
$S_0 \rightarrow S_8$
 $H - 3 \rightarrow L$ (22%)
 $H \rightarrow L + 2$ (17%)



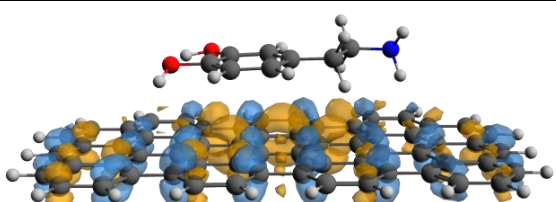
$S_0 \rightarrow S_9$
 $H - 2 \rightarrow L$ (43%)



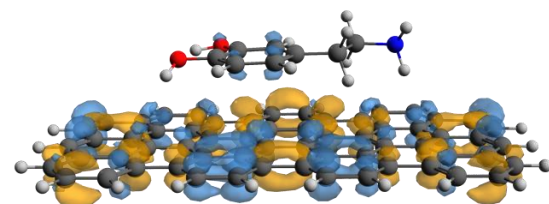
$S_0 \rightarrow S_{10}$
 $H \rightarrow L + 7$ (43%); $H \rightarrow L + 6$ (21%)



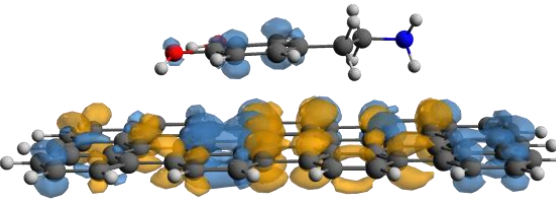
$S_0 \rightarrow S_{11}$
 $H - 7 \rightarrow L$ (18%)
 $H \rightarrow L + 9$ (17%)



$S_0 \rightarrow S_{12}$
 $H \rightarrow L + 7$ (18%)



$S_0 \rightarrow S_{13}$
 $H \rightarrow L + 8$ (23%)
 $H - 1 \rightarrow L + 1$ (17%)



$S_0 \rightarrow S_{14}$
 $H \rightarrow L + 9$ (21%)

Table S5 (cont.). The natural difference orbitals (NDO) between the ground and the given electronic excited state computed for the GrNP-DA binary complex at TDDFT/ ω B97X-D3BJ/def2-TZVPP level of theory (Blue = hole, Yellow = electron, orbital orders with red text color are specific to DA and blacks are graphenetypes).

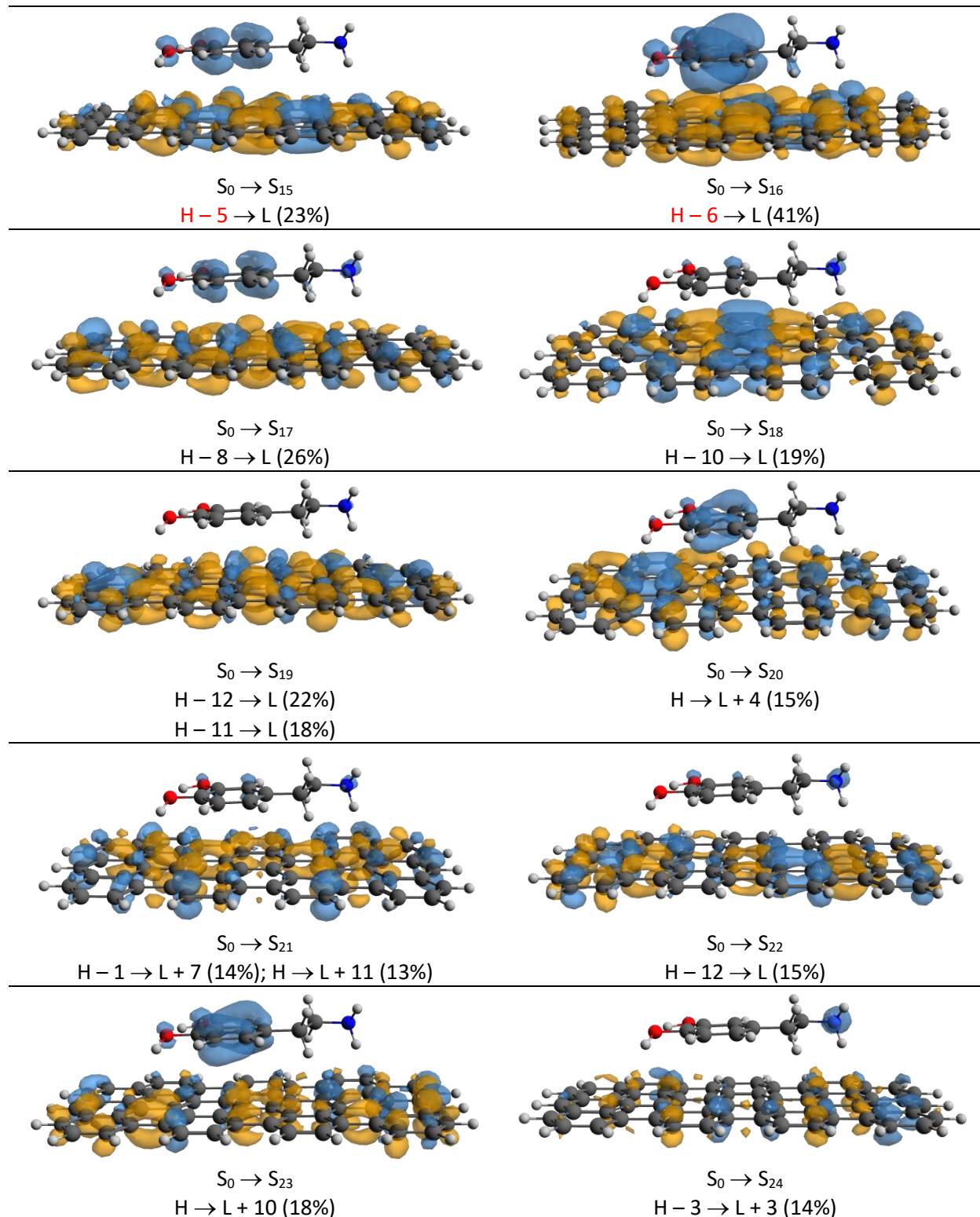
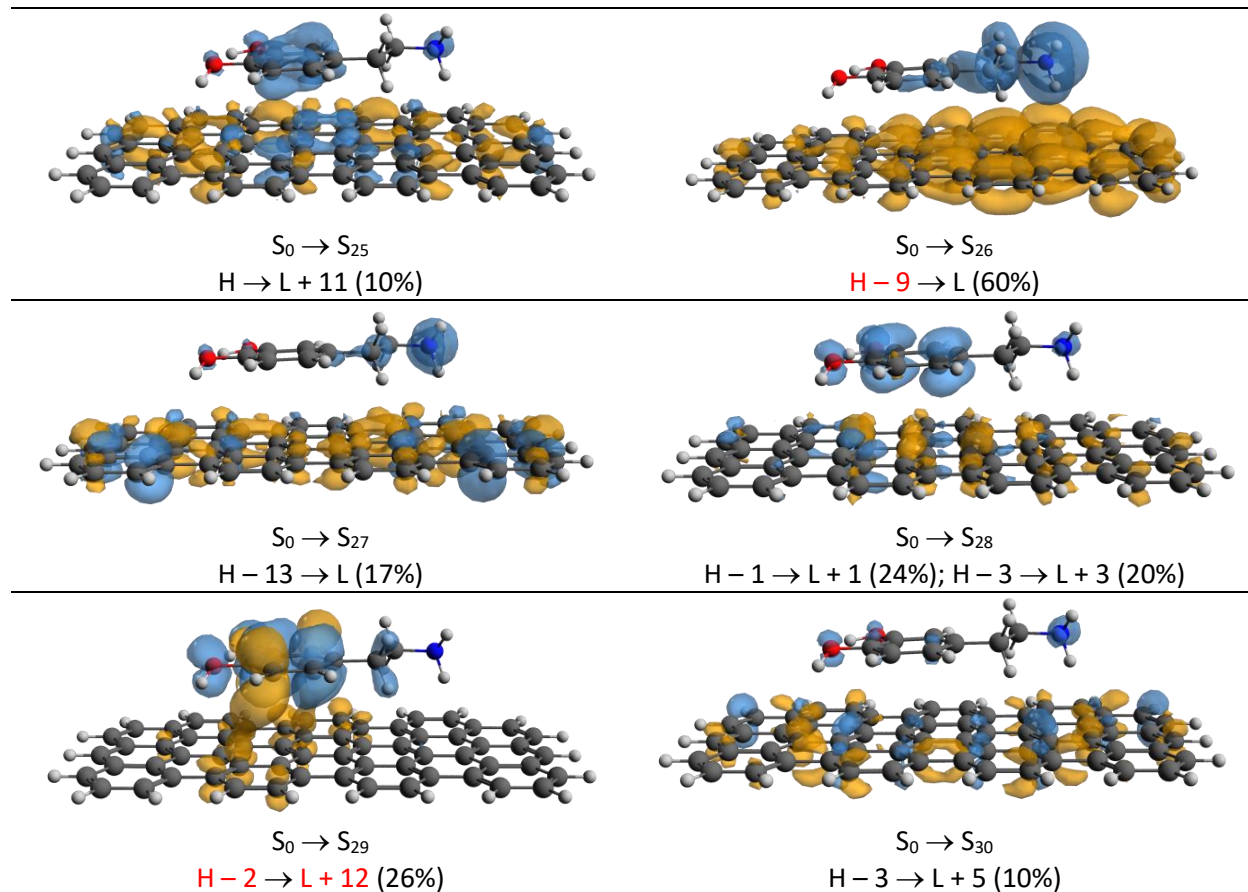


Table S5 (cont.). The natural difference orbitals (NDO) between the ground and the given electronic excited state computed for the GrNP-DA binary complex at TDDFT/ ω B97X-D3BJ/def2-TZVPP level of theory (Blue = hole, Yellow = electron, orbital orders with red text color are specific to DA and blacks are graphenetypes).



4. Electronic excited states and theoretical UV-absorption spectra for GrNP – DAQ binary complex.

Table S6. The frontier (HOMO and LUMO) orbitals of the isolated DAQ computed at ω B97X-D3BJ/def2-TZVPP level of theory.

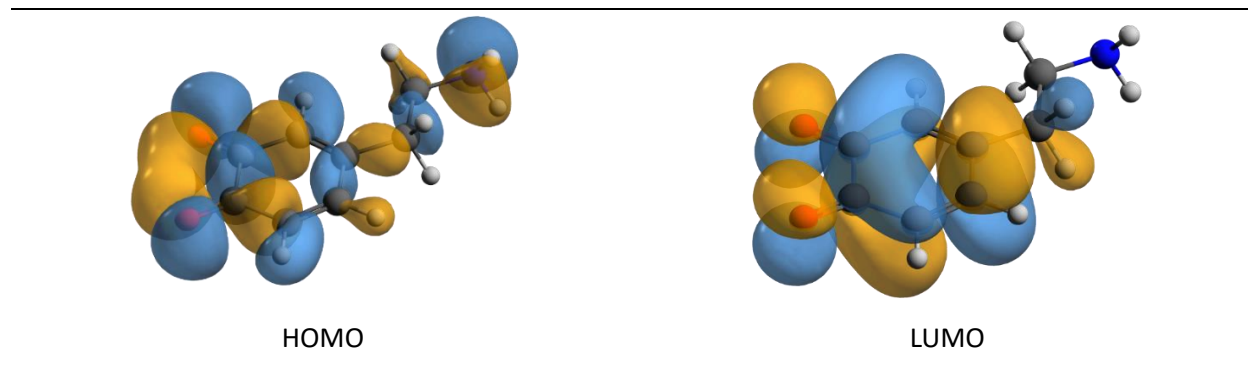


Table S7. The natural difference orbitals (NDO) between the ground and the given electronic excited state computed for the GrNP-DAQ binary complex at TDDFT/ ω B97X-D3BJ/def2-TZVPP level of theory (Blue = hole, Yellow = electron, orbital orders with red text color are specific to DA and blacks are graphene-types).

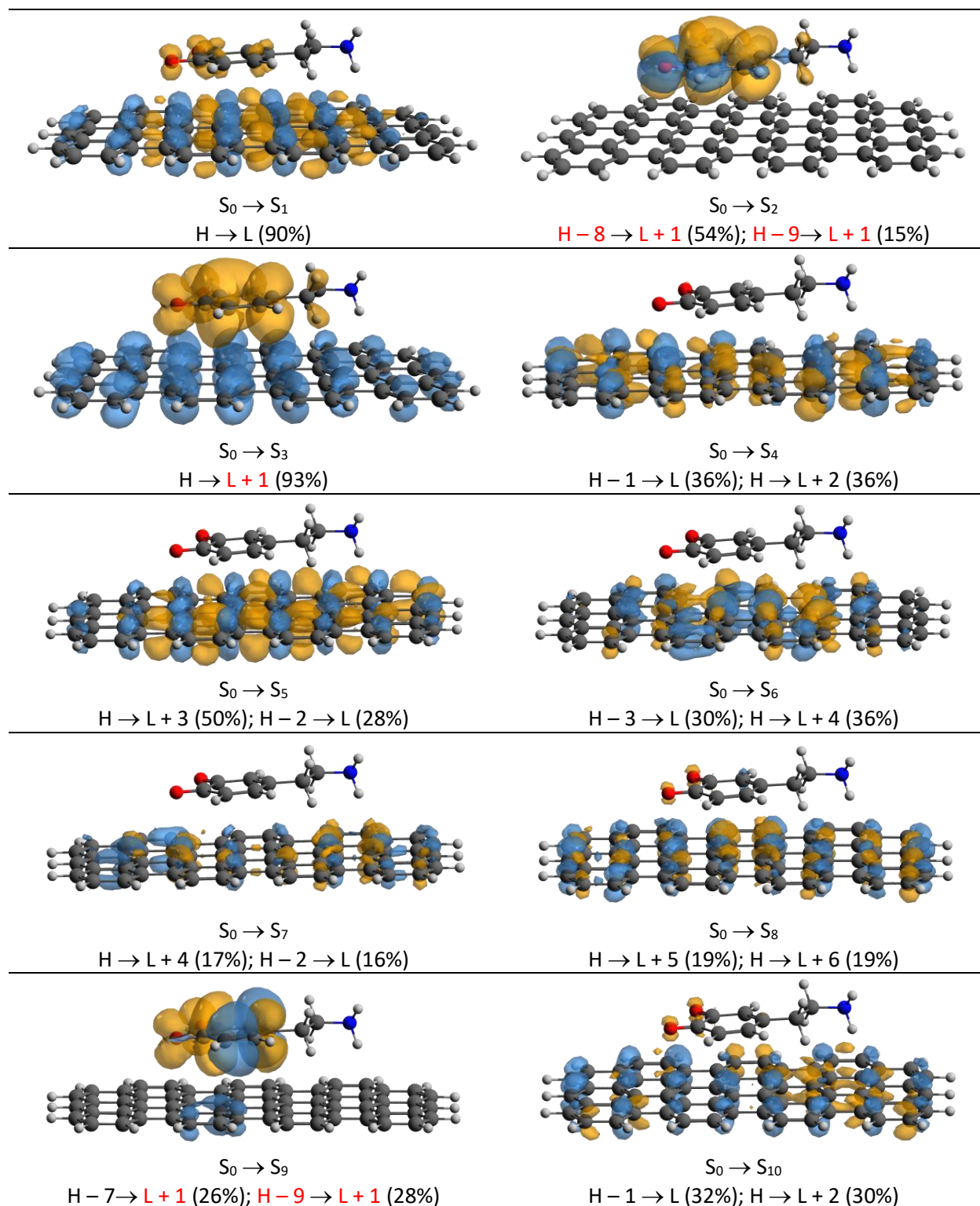


Table S7 (cont.). The natural difference orbitals (NDO) between the ground and the given electronic excited state computed for the GrNP-DAQ binary complex at TDDFT/ ω B97X-D3BJ/def2-TZVPP level of theory (Blue = hole, Yellow = electron, orbital orders with red text color are specific to DA and blacks are graphene-types).

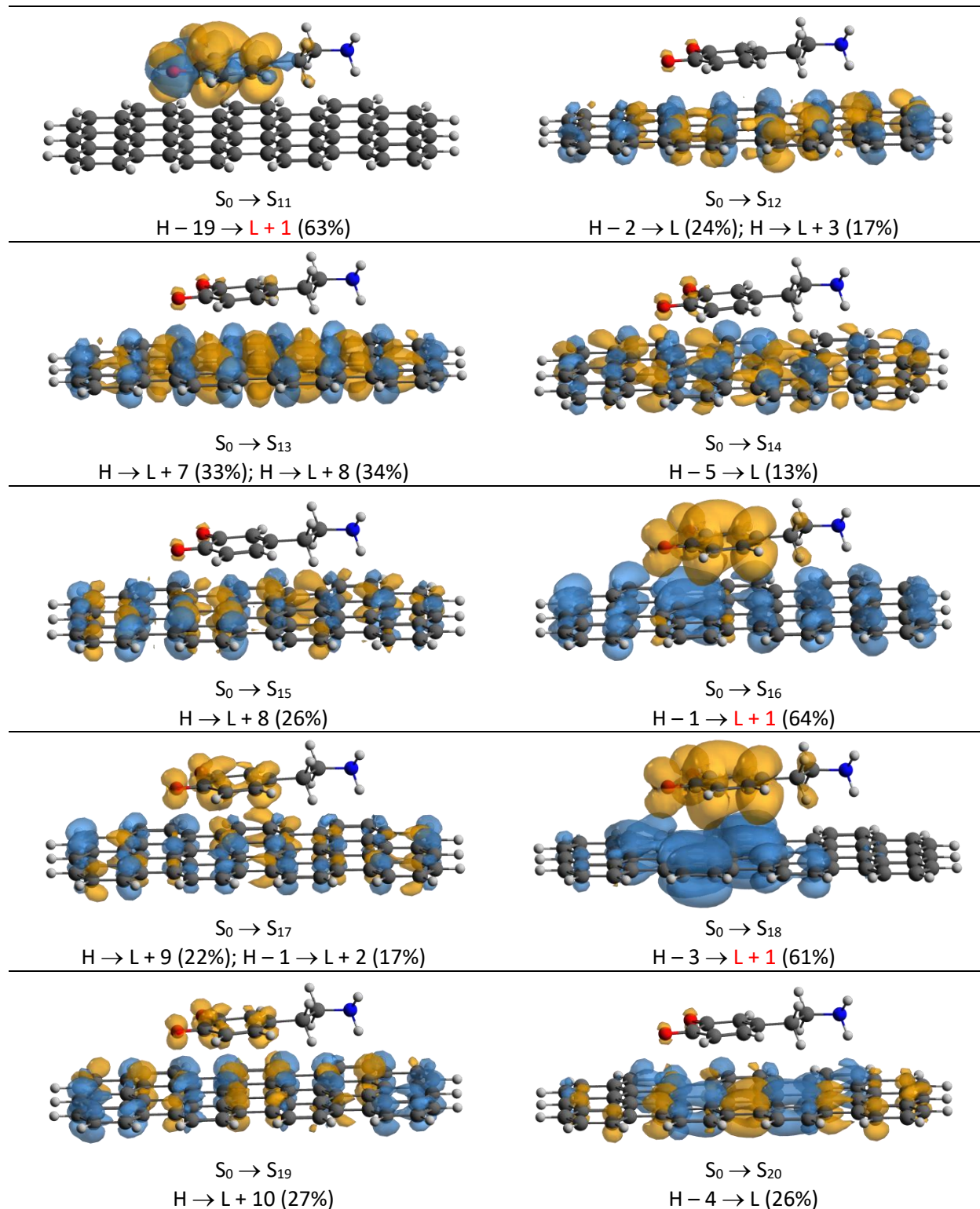


Table S7 (cont.). The natural difference orbitals (NDO) between the ground and the given electronic excited state computed for the GrNP-DAQ binary complex at TDDFT/ ω B97X-D3BJ/def2-TZVPP level of theory (Blue = hole, Yellow = electron, orbital orders with red text color are specific to DA and blacks are graphenetypes).

