

## Supporting information

### THEORETICAL INSIGHTS INTO THE ADSORPTION AND GAS SENSING PERFORMANCE OF Fe/Cu-ADSORBED GRAPHENE

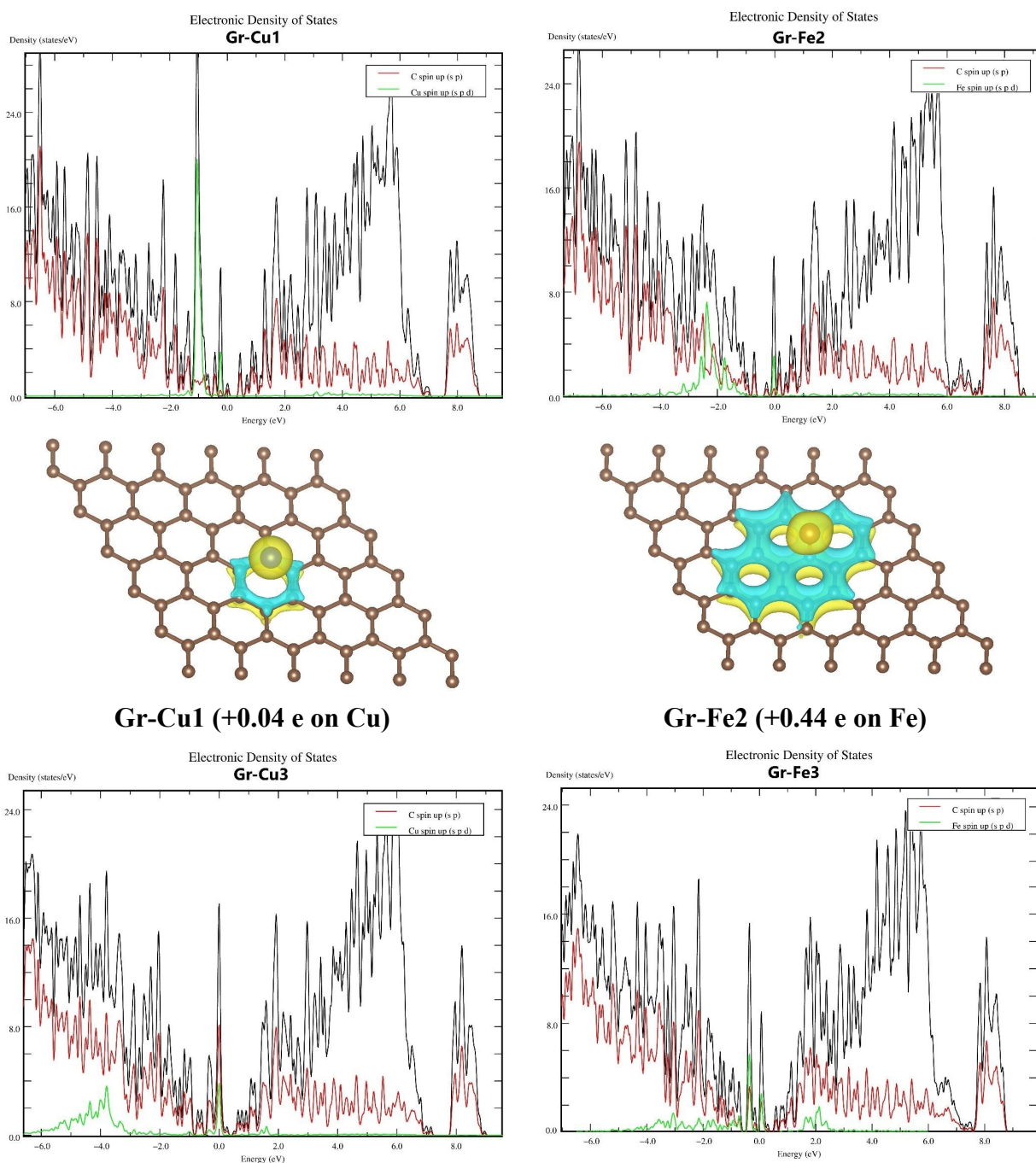
Nguyen Ngoc Tri<sup>1,\*</sup>, Dai Q. Ho<sup>1,2</sup>, Nguyen Tien Trung<sup>1,\*</sup>

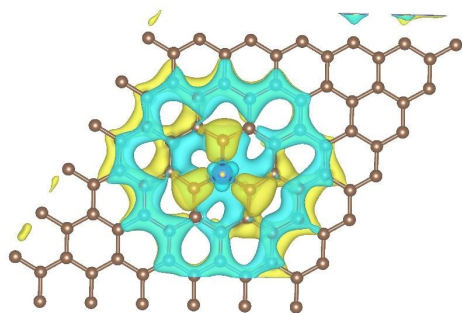
<sup>1</sup>Lab of Computational Chemistry and Modelling (LCCM), Faculty of Natural Sciences, Quy Nhon  
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<sup>2</sup>Department of Materials Science and Engineering, University of Delaware, USA

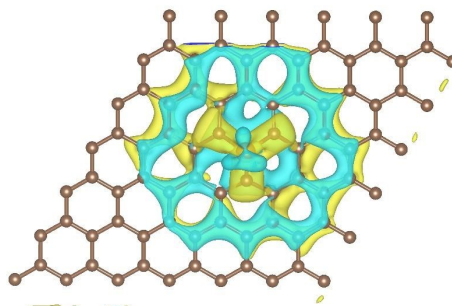
\*-Corresponding authors, Email: nguyenngoctri@qnu.edu.vn, nguyentientrung@qnu.edu.vn

#### Figures:



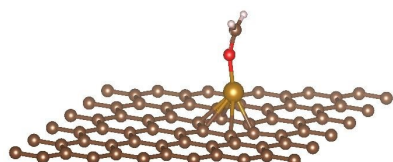


**Gr-Cu3 (+0.62 e on Cu)**

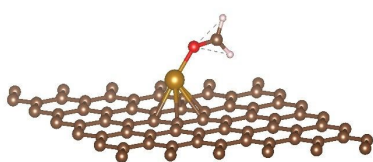


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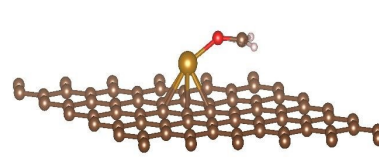
**Figure S1.** The density of states and charge density distribution maps of *Gr-Fe2(3)* and *Gr-Cu1(3)* surfaces



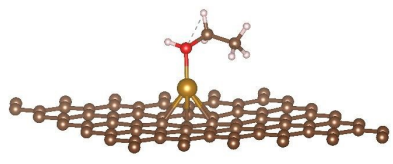
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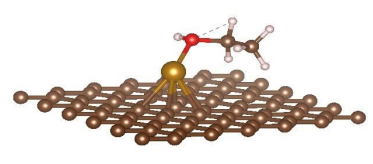
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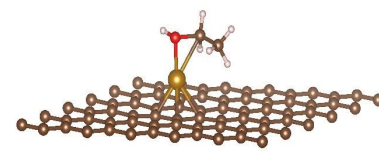
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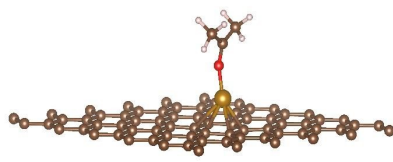
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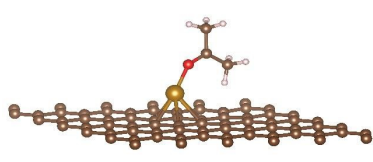
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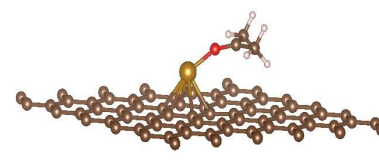
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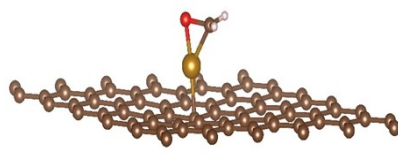
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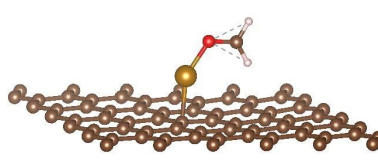
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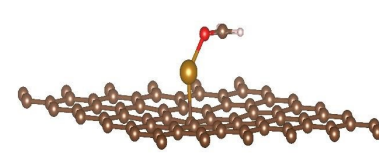
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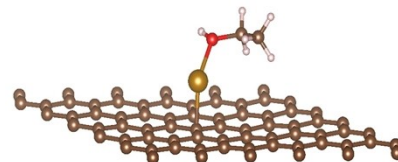
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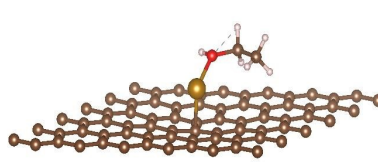
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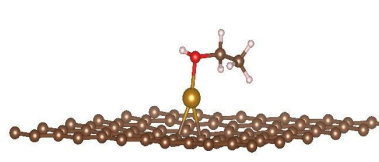
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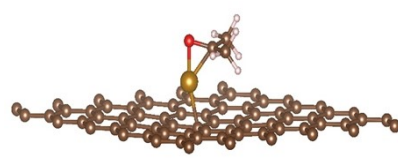
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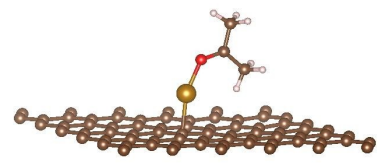
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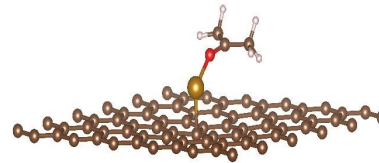
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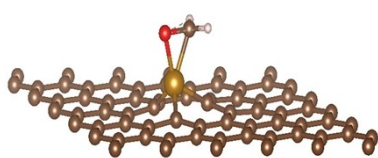
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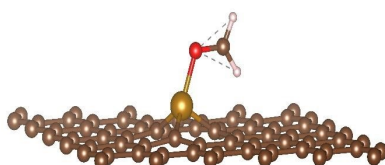
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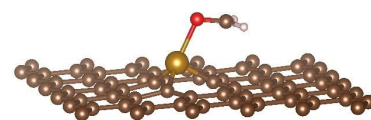
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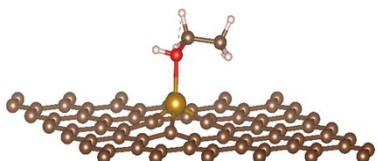
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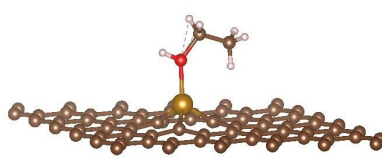
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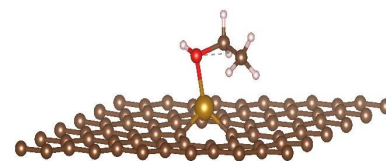
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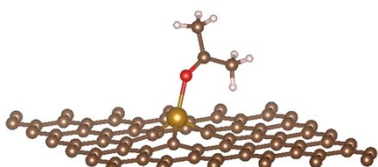
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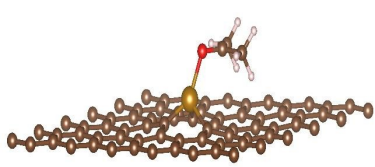
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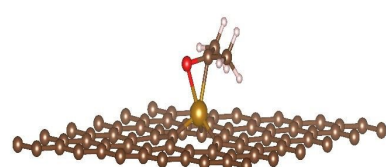
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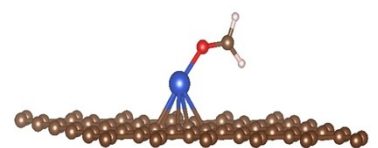
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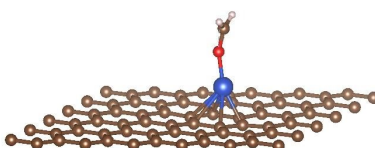
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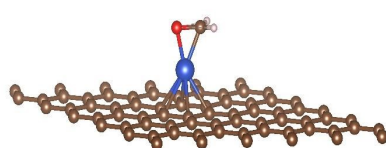
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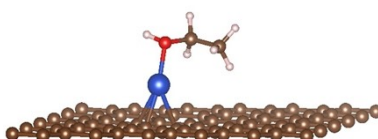
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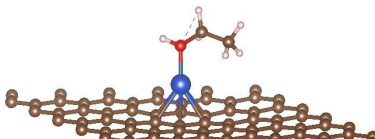
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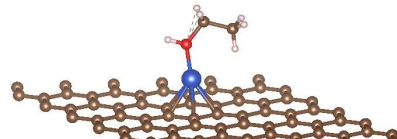
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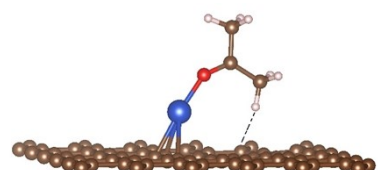
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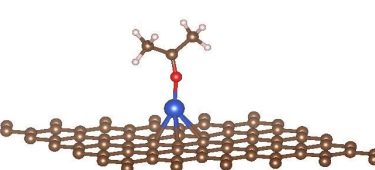
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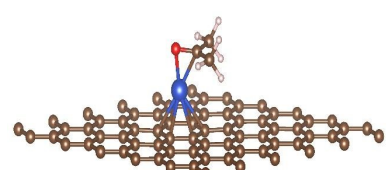
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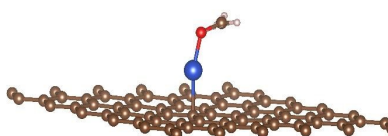
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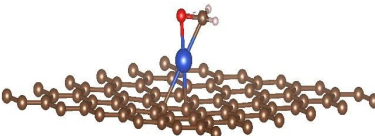
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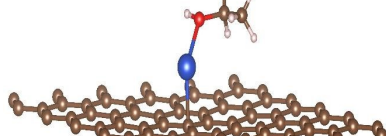
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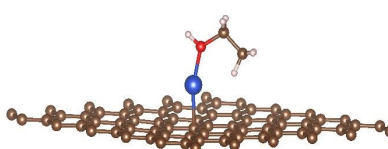
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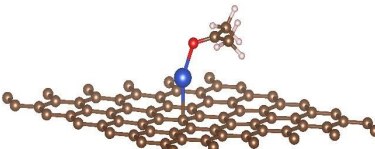
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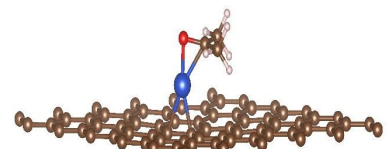
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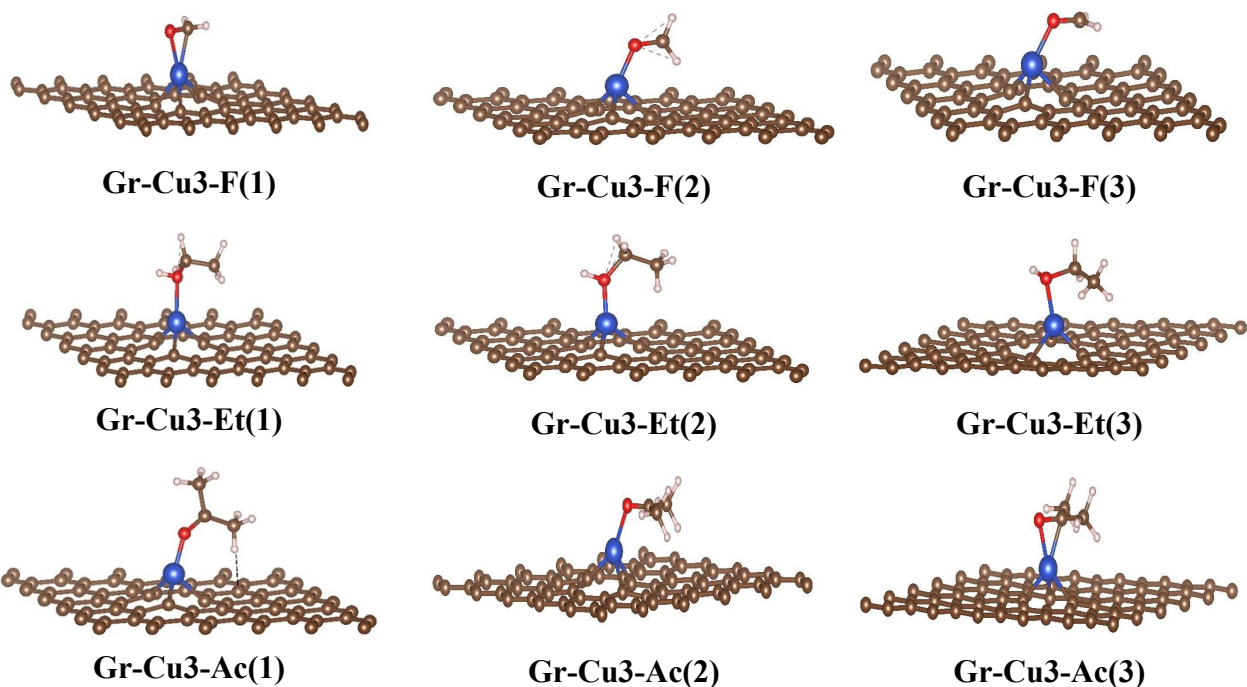
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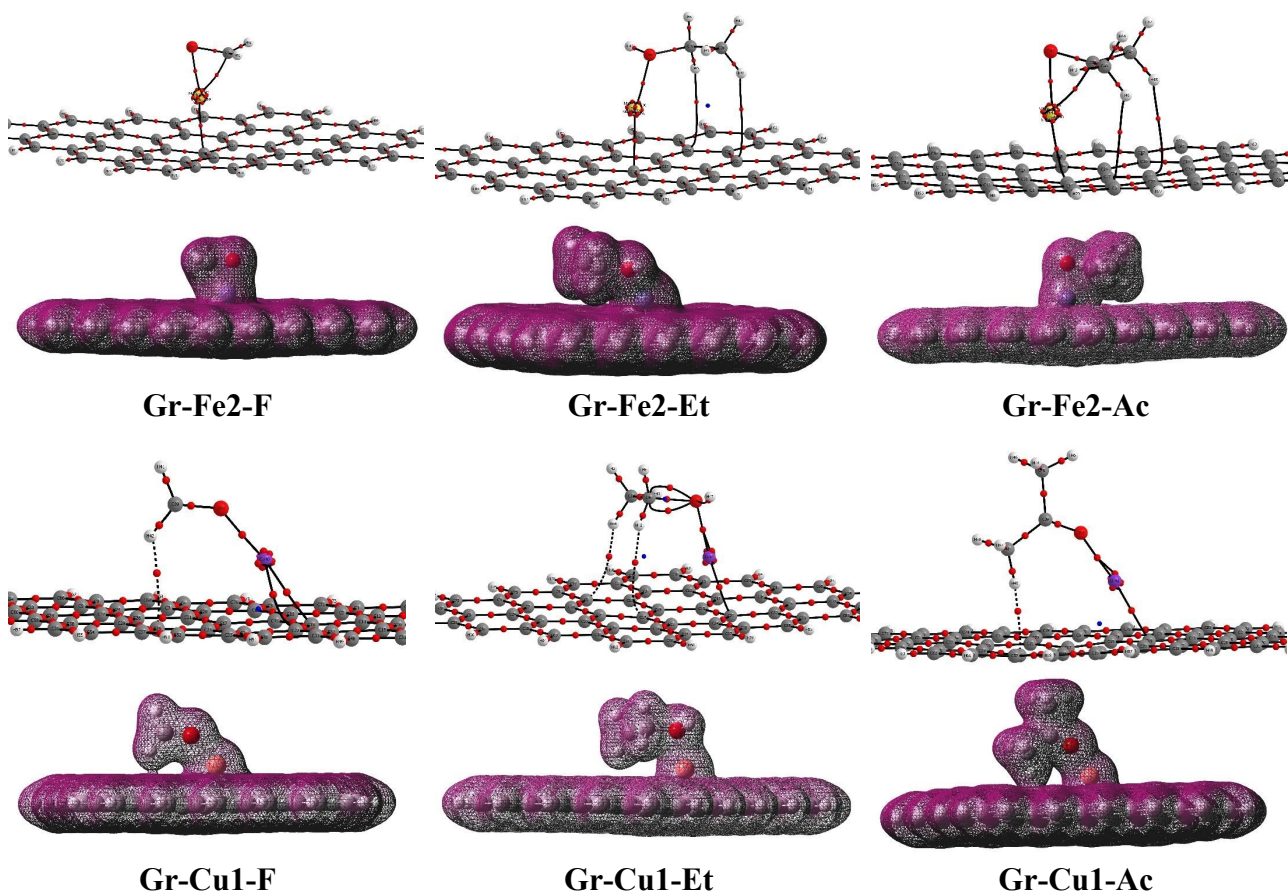
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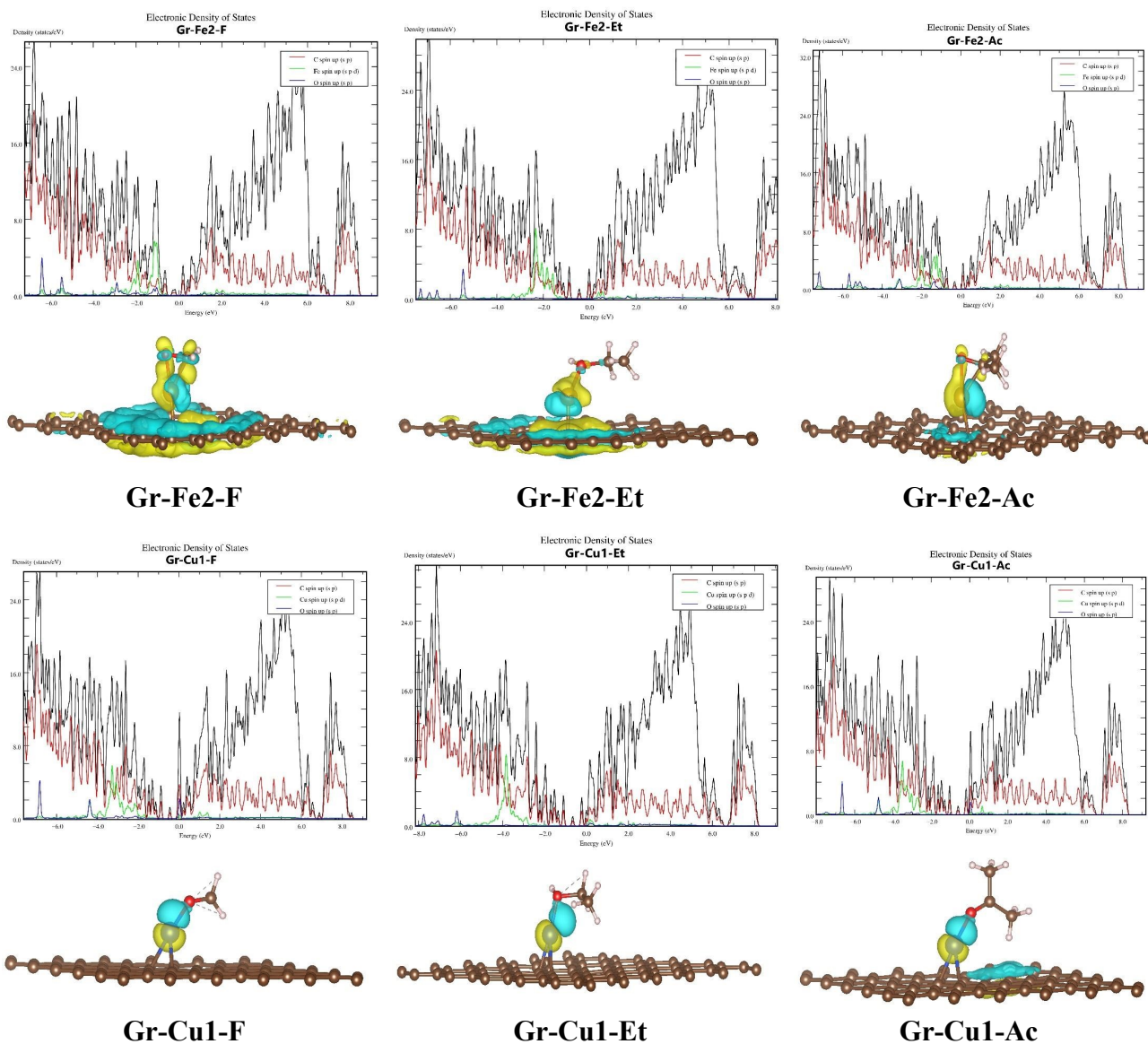
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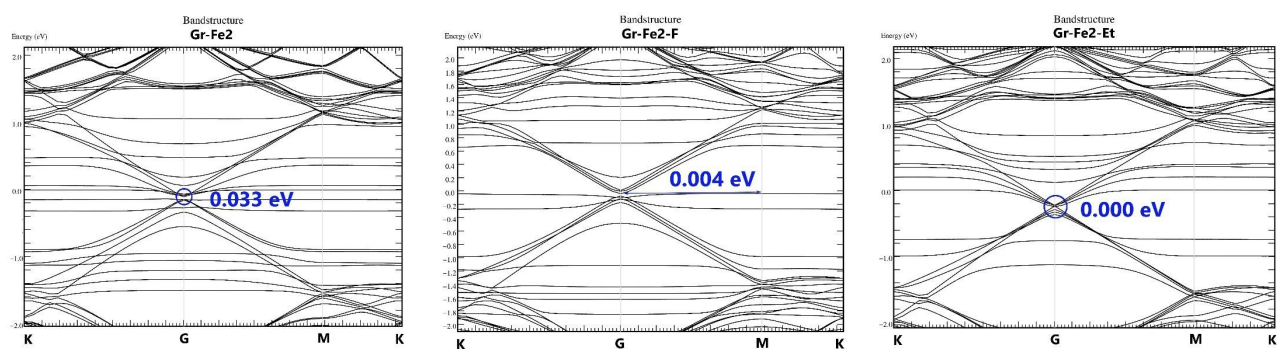
**Figure S2.** The other stable complexes for **Gr-Mi--VOC** systems ( $M = \text{Fe}, \text{Cu}; i = 1, 2, 3; \text{VOC} = \text{F}$  ( $\text{HCHO}$ ),  $\text{Et}$  ( $\text{C}_2\text{H}_5\text{OH}$ ),  $\text{Ac}$  ( $\text{CH}_3\text{COCH}_3$ ))

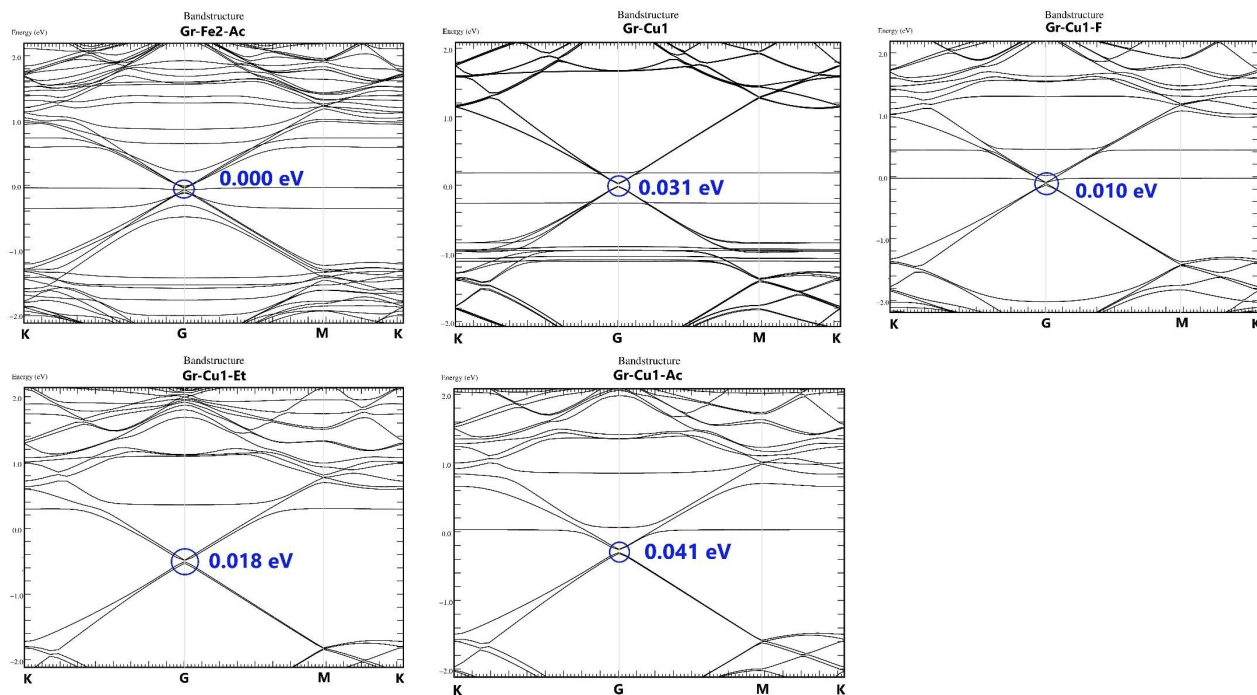


**Figure S3.** The topological geometries and total electron density transfer maps for the selected configurations in **Gr-Fe2/Gr-Cu1--VOC** systems



**Figure S4.** The density of states and charge density difference maps of the most stable configurations for *Gr-Fe2*, *Gr-Cu1* surfaces





**Figure S5.** The band structures of the most stable configurations in Gr-Fe<sub>2</sub>/Gr-Cu<sub>1</sub>--VOC systems

**Tables:**

**Table S1.** Adsorption energy of the other stable configurations for Fe/Cu-adsorbed graphene surfaces (Fig. S2) at the vdW-DF2 functional (in eV)

	HCHO (F-1/2/3)	C <sub>2</sub> H <sub>5</sub> OH (Et-1/2/3)	CH <sub>3</sub> COCH <sub>3</sub> (Ac-1/2/3)
<b>Gr-Fe1</b>	-1.59/-1.48/-1.23	-1.29/-1.23/-0.91	-1.48/-1.27/-1.43
<b>Gr-Fe2</b>	-2.33/-1.80/-1.70	-1.70/-1.52/-1.32	-2.17/-1.82/-1.84
<b>Gr-Fe3</b>	-1.16/-0.82/-0.65	-1.21/-1.16/-0.76	-1.12/-1.00/-0.68
<b>Gr-Cu1</b>	-0.70/-0.42/-0.44	-0.92/-0.69/-0.58	-0.82/-0.42/-0.53
<b>Gr-Cu2</b>	-0.61/-0.58	-0.71/-0.80	-0.77/-0.71
<b>Gr-Cu3</b>	-0.87/-0.87/-0.50	-0.72/-0.71/-0.49	-0.95/-0.79/-0.46

**Table S2.** Selected parameters of AIM and NBO analyses for the stable configurations ( $\rho(r)$ ;  $\nabla^2(\rho(r))$ ,  $H(r)$  in au, EDT in e, WBI-Wiberg bond index)

	BCP	$\rho(r)$	$\nabla^2(\rho(r))$	$H(r)$	WBI	EDT
<b>Gr-Fe2-F</b>	C...Fe	0.100	0.251	-0.028	0.842	0.088
<b>Gr-Fe2-Et</b>	O...Fe	0.118	0.590	-0.042	0.842	
	O...Fe	0.064	0.446	-0.005	0.426	0.088
<b>Gr-Fe2-Ac</b>	O...Fe	0.116	0.565	-0.041	0.755	0.216
<b>Gr-Cu1-F</b>	Cu...O	0.081	0.830	-0.005	0.210	-0.162
	H...C/ $\pi$	0.008	0.035	0.002	---	
<b>Gr-Cu1-Et</b>	Cu...O	0.074	0.718	-0.003	0.200	0.067
	H...C/ $\pi$	0.005	0.018	0.001	---	
<b>Gr-Cu1-Ac</b>	Cu...O	0.079	0.813	-0.004	0.202	0.056
	H...C/ $\pi$	0.016	0.064	0.001	---	

**Table S3.** The characteristics of gas sensing performance of **Gr-Fe2**, **Gr-Cu1** surfaces

	$E_g$ (eV)	S (%)	$\tau$ (s)		
			298K	400K	500K
<b>Gr-Fe2-F</b>	0.004	43.1	$2.4 \times 10^{27}$	$2.2 \times 10^{17}$	$3.0 \times 10^{11}$
<b>Gr-Fe2-Et</b>	0.000	47.4	$5.5 \times 10^{16}$	$2.6 \times 10^9$	$1.3 \times 10^5$
<b>Gr-Fe2-Ac</b>	0.000	47.4	$4.8 \times 10^{14}$	$2.1 \times 10^{15}$	$7.3 \times 10^9$
<b>Gr-Cu1-F</b>	0.010	33.6	$9.4 \times 10^{-3}$	$2.7 \times 10^{-5}$	$8.8 \times 10^{-7}$
<b>Gr-Cu1-Et</b>	0.018	22.4	$3.6 \times 10^3$	0.4	$1.9 \times 10^{-3}$
<b>Gr-Cu1-Ac</b>	0.041	21.5	72.8	$2.1 \times 10^{-2}$	$1.8 \times 10^{-4}$